Studies on the Structure/Reactivity Relationships of Bicyclic Epoxonium Ions and Tethered Nucleophiles. Efforts towards the Total Synthesis of (+)-Lactodehydrothyrsiferol and its Analogs. Multicomponent Approach to the Synthesis of Oxidized Amides through

Nitrile Hydrozirconation

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A systematic study on the structure/reactivity relationships of bicyclic epoxonium ions towards tethered nucleophiles has been conducted. The cyclization results show that bicyclo[3.1.0] epoxonium ions have a significant to exclusive preference for *exo*-cyclizations while bicyclo[4.1.0] epoxonium ions have a strong preference for *endo*-cyclizations.

A convergent approach towards the total synthesis of polycyclic ether natural product (+)lactodehydrothyrsiferol and its analogs is currently being pursued. This route includes the stereoselective reduction of the bicyclo[3.2.1] ketal which could be prepared from coupling of the functionalized aldehyde and vinyl iodide. Both enantiopure fragments can be obtained from cyclizations of the diepoxide and the monoepoxide, respectively. Key transformations involve two asymmetric epoxidations, a cascade cyclization of diepoxide, a Cr/Ni-mediated coupling reaction and a stereoselective reduction of bicyclo[3.2.1] ketal.

An efficient one-pot synthesis of oxidized amides from nitrile hydrozirconation has been developed. From the common acylimine intermediates, acyl aminals can be accessed through alcohol addition, acyl hemiaminals can be accessed through water addition and enamides can be accessed through tautomerization.

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1.0 STUDIES ON THE STRUCTURE/REACTIVITY RELATIONSHIPS OF BICYCLIC EPOXONIUM IONS AND TETHERED NUCLEOPHILES

1.1 INTRODUCTION

Polycyclic ether structures (Figure 1), which have been discovered in a number of marine natural products, have gained considerable attention from synthetic community due not only to their intrinsically complex structures but also to their interesting biological activities.^{1,2} From a biosynthetic view of point, these compounds have been proposed to arise from cascade cyclizations from the requisite polyepoxide precursors.³⁻⁵ A key issue associated in this process is the strict regiochemical control, i.e., *exo-* vs *endo-*cyclization (Figure 2). As can be envisioned in Figure 1, hemibrevetoxin B⁵ could be prepared from the polyepoxide through all *endo-*cyclization, bullatacin⁶ could be prepared from the polyepoxide precursor through all *exo-*cyclization and lactodehydrothyrsiferol⁷ could be prepared from the polyepoxide through a combination of *exo-* and *endo-*cyclizations.



Figure 1. Representative polyether natural products



LA = Lewis acid, Nu = nucleophile

Figure 2. Exo- vs endo-cyclization

The *exo*-pathway is well known and commonly observed from studies on intramolecular cyclizations through epoxide opening. For example, epoxy alcohol **1.1**, under acidic conditions, afforded tetrahydrofuran derivative **1.3** predominantly through *exo*-pathway transition state **1.2** (Figure 3).⁸ However, in the presence of the catalytic antibody IgG26D9 elicited from amine oxide antigen **1.4**, tetrahydropyran **1.6** was isolated as the sole product from *endo*-cyclization,⁸ which formally violated the Baldwin's rules⁹ for ring closure reactions.



Figure 3. Cyclizations of epoxy alcohol under acid- and antibody-catalyzed conditions

Ab initio calculations performed by Houk^{10,11} and $Coxon^{12,13}$ showed that under acidcatalyzed conditions, the *exo*-transition structure (TS) has a nearly ideal trajectory for the attack of the hydroxyl oxygen atom via an S_N^2 manner and therefore has lower energy than *endo*-TS, resulting in the formation of **1.3** as the major product through a kinetically and chemically favored process. The exclusive formation of **1.6** under antibody-catalyzed (enzymatic) conditions was attributed to the similarity between *endo*-TS **1.5** and *N*-oxide **1.4**. Calculations showed that *endo*-cyclization proceeds through an S_N 1-type transition structure.

Due to the difficulty in the polycyclic ether formation through *endo*-pathway and the high efficiency in increasing complexity in cascade cyclizations of polyepoxides, a number of researchers have been involved in investigating the epoxide opening cascades. Murai and co-workers¹⁴ first reported *endo*-selective cascade cyclizations of polyepoxides (Figure 4). Diepoxy alcohol **1.7**, when treated with La(OTf)₃, provided fused bicyclic product **1.8** in 52% yield. Unfortunately, when the conditions were applied to triepoxy alcohol **1.9**, the cyclization efficiency decreased drastically and only 9.3% of the desired tricycle **1.10** was obtained from all *endo*-cyclization. In addition, the proposed mechanism requires the presence of chelation of La(III) ion with the epoxide oxygen and the pendent methoxy oxygen to promote the *endo*-selectivity, which also limits its potential applications in natural product synthesis.



Figure 4. Endo-selective cascade cyclizations of polyepoxides in Murai group

At nearly the same time, McDonald and co-workers^{15,16} studied epoxide opening cascades under Lewis-acid-promoted conditions, and demonstrated that *endo*-cyclizations could be achieved in high efficiency with rationally designed substrates. Some representative examples are shown in Figure 5.



Figure 5. Representative endo-selective cyclizations of polyepoxides in McDonald group

McDonald postulated that bicyclic epoxonium ions are key intermediates in cascade cyclizations of epoxides to generate polycyclic ethers and that *endo*-cyclizations were favored over *exo*-cyclizations due to the increased ring strain in *exo*-TSs (bicyclo[3.1.0] epoxonium ions).¹⁵ As illustrated in Figure 6, bicyclo[4.1.0] epoxonium ion **1.20** (from *endo*-cyclization) is presumed to be energetically lower than bicyclo[3.1.0] intermediate **1.19** (from *exo*-cyclization) in the initial cyclization as well as a similar comparison between bicyclo[4.1.0] ion **1.22** and bicyclo[3.1.0] ion **1.21** in the ensuing cyclization, leading to the formation of *trans*-fused tricycle **1.23** with excellent *endo*-selectivity.



Figure 6. Hypothesis for regioselective cascade cyclizations

McDonald also observed that terminal nucleophiles can also affect the stereochemical outcomes and/or regioselectivity (Figure 7).¹⁷ Cyclization of diepoxide **1.24** with a *t*-butyl carbonate group as the terminal nucleophile provided *cis*-fused bicycle **1.28** which arose from the addition of the carbonate carbonyl to the tertiary carbocation **1.26** via an S_N l fashion as the predominant product. Alternatively, the cyclization of diepoxide **1.29** with the better terminal nucleophile dimethyl carbamate gave *trans*-fused bicycle **1.27** as the major product which came from direct attack of the carbamate carbonyl to the epoxonium ion **1.25** through an S_N 2 pathway. Interestingly, triepoxy *t*-butyl carbonate **1.30**, upon cyclization, afforded mainly tricyclic structure **1.32** which came from a combination of *endo*-cyclization and *exo*-cyclization and has a *cis*-ring fusion between the six- and five-membered cyclic ethers. In this case, only trace amount of all-fused *trans,trans*-tricycle **1.31** was obtained. On the other hand, reaction of dimethylcarbamate triepoxide **1.33** produced desired tricycle **1.31** in good yield, with **1.32** not being observed.



Figure 7. Impact of terminal nucleophiles on stereospecificity and regioselectivity

Besides acidic conditions, cyclizations of epoxides can also be performed under basic¹⁸, neutral¹⁹ and oxidative^{20,21} conditions. Jamison and co-workers demonstrated that diepoxide **1.34A** and triepoxide **1.35A**, under basic conditions in the protic solvent MeOH, provided THP triad **1.34B** and tetrad **1.35B**, respectively, in good yields, with the TMS group as a "disappearing" directing group.¹⁸ Without the "disappearing directing groups", the diepoxide **1.36A** and triepoxide **1.37A**, under essentially neutral conditions, delivered triad **1.36B** and tetrad **1.37B**, respectively, in excellent yields.¹⁹ In the total synthesis of hemibrevetoxin B, Holton developed a novel intramolecular epoxide opening cascade initiated by oxidation of the alkene with *N*-(phenylseleno)phthalimide to effect the formation of **1.39** in excellent yield, with

the B and C rings assembled in a single operation (Figure 9).²⁰ In this process, the highly polar solvent hexafluoroisopropanol was selected for the *endo*-selective epoxide opening through the S_N 1-type transition state, which is consistent with the computational analysis.^{10,11}



Figure 8. Regioselective epoxide opening cascades under basic and neutral conditions



Figure 9. Key transformation in Holton's total synthesis of hemibrevetoxin B

Cascade cyclizations of epoxides/polyepoxides were also studied in my laboratory²² under electron transfer initiated cyclization ($\text{ETIC}^{23,24}$) conditions. These photochemical conditions use medium-pressure mercury lamp as the excitation source, catalytic amount of Nmethylquinolinium hexafluorophosphate as the sensitizer, O_2 as the ultimate oxidant, 4 Å molecular sieves as moisture scavenger, NaOAc as the base, Na₂S₂O₃ as the peroxide remover, 1,2-dichloroethane as the solvent and toluene as the co-sensitizer. Two typical examples are depicted in Figure 10. Under ETIC conditions, homobenzylic ether 1.40 was oxidized to radical cation 1.41A which fragmented to form the benzyl radical and oxocarbenium ion 1.41B in a reversible manner. The tethered epoxide attacked the electrophile 1.41B to generate bicyclic epoxonium ion **1.41C**. The terminal nucleophile THP ether opened the bicyclo[3.1.0] epoxonium ion through *exo*-pathway to deliver bis-THF product **1.42** irreversibly in good yield. Similarly, reaction of diepoxide 1.44 gave rise to the consecutive exo, exo-cyclization product 1.45 in high efficiency. After removal of the anomeric center with Jones reagent,²⁵ lactones 1.43 and 1.46 were obtained as single diastereomers, respectively. The regioselectivity observed herein is in accord with Houk and Coxon's computational studies with 5-exo-pathway being preferred over 6-endo-pathway.



 $Conditions: (a) \ hv, O_2, \ NMQPF_6 \ (cat.), \ NaOAc, \ Na_2S_2O_3, \ 1,2-dichloroethane \ (DCE)/PhMe \ (6:1, \ v/v). \ (b) \ Jones \ reagent. \ (c) \ Ac_2O.$

Figure 10. Epoxide opening cascades under electron transfer initiated cyclization conditions

The oxocarbenium ion (Lewis acid) generated under ETIC conditions has the merit that it is able to activate the proximal epoxide specifically so that the complication from possible random activation of epoxides by a Brønsted or Lewis acid can be eliminated, making this method ideal for investigating the reactivity of epoxonium ions with specific structures. In order to examine the factors that can affect the reaction pathways in epoxide opening cascades, a systematic study on the cascade cyclizations of monoepoxides/diepoxides under ETIC conditions has been conducted²⁶ and details will be discussed in the following context.

1.2 STUDIES ON EPOXIDE OPENING CASCADES UNDER ETIC CONDITIONS

As previously described, regioselectivity in epoxide opening cascades could be influenced by ring strain in the forming bicyclic epoxonium ions, nucleophiles and solvents. Besides these, I was interested in whether other factors such as bicyclic epoxonium structure and Lewis acid selection can also affect the reaction pathways. Towards this end, we prepared substrates shown in Figure 11.

From monoepoxides **1.47-1.50**, oxocarbenium-activated bicyclic epoxonium ion intermediates (bicyclo[3.1.0] epoxonium and bicyclo[4.1.0] epoxonium) with different substitution patterns (disubstituted and trisubstituted) will be formed and compared in terms of their regioselectivity towards terminal nucleophiles. Diepoxides **1.51-1.55** were designed to study the impact of different Lewis acid-activating groups (oxocarbenium ion and non-stabilized carbenium ion) on the regiochemical outcomes as well as the effect of relative stereochemical orientations of epoxides on the cyclization efficiency. Figure 12 shows the corresponding bicyclic epoxonium ion intermediates that will be compared in this study. Also of note is that the more reactive diphenylmethyl group was employed in the monoepoxide substrates **1.47-1.50** instead of benzyl group (*cf.* Figure 10) as the electroauxiliary to initiate the oxocarbenium ion formation due to the reduced nucleophilicity of epoxides by the proximal *t*-butyl carbonate groups. For consistency throughout the studies, the diphenylmethyl group was also incorporated in the diepoxide substrates **1.51-1.55** though benzyl group is sufficiently reactive for this purpose.



Figure 11. Substrates for cyclizations



Figure 12. Bicyclic epoxonium ion intermediates to be investigated

1.2.1 Synthesis of epoxide substrates

The synthesis of disubstituted monoepoxide **1.47** proceeded from commercially available 4-pentenal in a straightforward manner (Figure 13). Addition of diphenylmethyllithium to 4-pentenal followed by methylation of the secondary alcohol and cleavage of the terminal alkene afforded aldehyde **1.56**. A sequence of Horner-Emmons olefination, ester reduction, allylic

alcohol epoxidation and primary hydroxyl group protection with Boc_2O^{27} provided carbonate **1.47**. Likewise, epoxide **1.48** was prepared from 5-hexenal which was obtained from Swern oxidation of 5-hexen-1-ol.



Figure 13. Synthesis of disubstituted monoepoxides 1.47 and 1.48

The synthesis of trisubstituted monoepoxides **1.49** and **1.50** is illustrated in Figure 14. Opening of epoxide **1.59** using Yamamoto's aluminum-amide promoted protocol^{28} followed by Johnson-Claisen rearrangement²⁹ of the allylic alcohol and reduction of the ethyl ester provided aldehyde **1.60**, which was converted into **1.49** through a sequence of diphenylmethyllithium addition, methylation, silyl ether deprotection, epoxidation and protection of the primary alcohol with Boc₂O. For the synthesis of homologous substrate, aldehyde **1.60** was homologated through Wittig olefination and mercury-mediated enol ether hydrolysis³⁰ to give the corresponding aldehyde which was further transformed into **1.50** in a similar manner.



Figure 14. Synthesis of trisubstituted monoepoxides 1.49 and 1.50

The trisubstituted diepoxides **1.51-1.52** were prepared similar to **1.49** and **1.50**. Monoepoxide **1.61** was converted into aldehyde **1.62** in excellent yields through epoxide opening, Johnson-Claisen rearrangement and reduction (Figure 15). Dienol **1.63** was obtained in good yields after a three-step sequence of diphenylmethyllithium addition, methylation and silyl group removal. To ensure the high enantiomeric and diastereomeric control in the epoxidations, asymmetric epoxidation methods were utilized. Dienol **1.63** was converted into diepoxy carbonate **1.51** through double Shi epoxidation³¹ and protection of the primary hydroxyl group with Boc₂O in excellent yields. A sequence of Sharpless epoxidation, ³² Shi epoxidation and the primary hydroxyl group protection of **1.63** efficiently provided diastereomeric counterpart **1.52**. The stereochemical orientations of the epoxides in **1.51** and **1.52** were given based on the mechanistic analysis.



Figure 15. Synthesis of diepoxides 1.51 and 1.52

Aldehyde **1.62** was homologated to aldehyde **1.64** to prepare diepoxides **1.53** and **1.54** through olefination and mercury-mediated enol ether hydrolysis in 91% yield (Figure 16). Dienol **1.65** was obtained in a similar manner to **1.63** through diphenylmethyllithium addition, methylation and deprotection. Subsequently, dienol **1.65** was converted into carbonate **1.53** through double Shi epoxidation and carbonate formation, or carbonate **1.54** through a sequence of Sharpless epoxidation, Shi epoxidation and carbonate formation.



Figure 16. Synthesis of diepoxides 1.53 and 1.54

Diepoxide **1.55** was prepared in a convergent manner (Figure 17). Reduction of δ -lactone followed by diphenylmethyllithium addition to the crude lactol provided the diol in 82% yield over the two steps, which was converted to sulfone **1.66** through a sequence of Mitsunobu reaction, methylation and oxidation³³ of the resulting sulfide with *m*CPBA. A Kocienski-modified Julia olefination³⁴ between sulfone **1.66** and aldehyde **1.60** afforded the desired diene in 63% yield. Further operations similar to the synthesis of **1.51** and **1.53** provided carbonate **1.55** in excellent yields.



Figure 17. Synthesis of diepoxide 1.55

1.2.2 Cyclizations of epoxide substrates under ETIC conditions

With these epoxides in hand, I examined their reactions under ETIC conditions to explore the factors that could affect the regioselectivity in the opening of bicyclic epoxonium ions by pendent nucleophiles. First, I carried out the reactions of disubstituted monoepoxides and the cyclization results are illustrated in Figure 18. The reaction of **1.47** exclusively provided 5-*exo*-product **1.68** through a disubstituted bicyclo[3.1.0] epoxonium ion intermediate, which is consistent with the previous observations that disubstituted bicyclo[3.1.0] ions prefer 5-*exo*-cyclization (*cf.* Figure 10).³⁵ Interestingly, cyclization of the homologated epoxide **1.48** gave a mixture of 5-*exo*- and 6-*endo*-products **1.71** and **1.72**, respectively, with *exo*-pathway being slightly favored. The difference between these two reactions is that the initial cyclization of **1.47** forms a bicyclo[3.1.0] intermediate while the cyclization of **1.48** forms a bicyclo[4.1.0] ion. The *exo*-products were fully characterized by oxidation to the corresponding lactones.^{25,36,37}



Figure 18. Cyclizations of disubstituted monoepoxides 1.47 and 1.48

Cyclization of trisubstituted epoxide **1.49** produced a complex mixture of *exo-* and *endo*products from trisubstituted bicyclo[3.1.0] ion **1.74**. The *trans-*fused *endo-*product **1.77** arose from the S_N^2 attack of the carbonate to the epoxonium ion **1.74** while the *syn-*fused *endo-*product **1.78** was from the addition of the carbonate carbonyl oxygen to the tertiary carbocation **1.75** in an S_N^1 manner. The formation of **1.78** is in accord with McDonald's observation¹⁷ of *cis-*fused bicycle **1.28** from **1.24** (*cf.* Figure 7). On the contrary, the reaction of **1.50** cleanly afforded *trans-*fused bicycle **1.81** as the only isolable product in excellent yield. These cyclization results clearly demonstrate that the combination of both the methyl substitution and the bicyclo[4.1.0] epoxonium ion can reverse the regiochemical outcomes from complete *exo*-pathway to exclusive *endo*-pathway.



Figure 19. Cyclizations of disubstituted monoepoxides 1.49 and 1.50

Having obtained the general information on the regioselectivity of monoepoxide opening/cyclization with carbonates as the nucleophile, I next investigated the cyclizations of diepoxides. Diepoxide **1.51**, under ETIC conditions, provided a mixture of consecutive *exo,exo*-and *endo,endo*-products **1.85** and **1.86**, respectively, in a combined 40% yield and with a 5.5:1 ratio (Figure 20). Mechanistic analysis suggests that initial cyclization formed bicyclo[3.1.0] epoxonium intermediate **1.82**, which could be opened by the epoxide either in an *exo*-mode to form a second bicyclo[3.1.0] epoxonium **1.83**, or in an *endo*-mode to form bicyclo[4.1.0] epoxonium **1.84**. It is noteworthy that no *endo*-cyclization product from **1.83** was isolated, indicating that bicyclo[3.1.0] epoxonium ions activated by non-stabilized carbeniums (stronger

Lewis acids) prefer *exo*-pathway towards tethered nucleophiles. Also of note is that no *cis*-fused *endo*-product was observed from **1.84**, indicating that epoxides are better nucleophiles than carbonates or that the addition of carbonate to the epoxonium ion is reversible before the loss of the *t*-butyl cation. A similar regioselectivity was observed and a better overall yield was obtained when diastereomeric counterpart **1.52** was exposed to the ETIC conditions.



Figure 20. Cyclizations of diepoxides 1.51 and 1.52

Following the completely *endo*-selective opening of bicyclo[4.1.0] epoxonium ions **1.80** and **1.84**, reaction of diepoxides **1.53** and **1.54** afforded *trans,syn,trans*-fused tricycles **1.91** and **1.93** in all *endo*-modes as expected since the intermediates were trisubstituted bicyclo[4.1.0]

epoxonium ions. The cyclization of **1.54** is highly efficient in consideration of the product complexity. The higher yield observed in cyclization of **1.54** compared with **1.53** is probably due to the diminished steric interactions in the cyclization processes. To simplify characterizations, acetals **1.91** and **1.93** were oxidized by treatment with *m*CPBA and BF3•OEt₂ followed by addition of $Et_3N^{36,37}$ to form lactones **1.92** and **1.94**. The structure of **1.92** was unambiguously confirmed through single crystal X-ray analysis (Figure 22).³⁸



Figure 21. Cyclizations of diepoxides 1.53 and 1.54



Figure 22. ORTEP structure of lactone 1.92

Since no selectivity was observed in the cyclization of **1.42**, I was not surprised to observe that when diepoxide **1.55** was subjected to ETIC conditions, consecutive *exo,exo-* and *endo,endo-*products **1.97** and **1.98** were obtained in comparable yields, with the disubstituted bicyclo[4.1.0] epoxonium **1.95** from initial cyclization being non-regioselective (Figure 23). Tricycle **1.98** was converted into lactone **1.99** and its stereochemical outcomes were established through single crystal X-ray analysis (Figure 24).³⁹



Figure 23. Cyclization of diepoxide 1.55



Figure 24. ORTEP structure of lactone 1.99

1.3 COMPUTATIONAL ANALYSIS

The cyclization results of the epoxide opening cascades under ETIC conditions show that the regioselectivity is highly dependent on the bicyclic epoxonium ion structures. Trisubstituted bicyclo[4.1.0] epoxonium ions prefer exclusive *endo*-pathways while disubstituted bicyclo[4.1.0] epoxonium ions essentially show no preference towards *exo*- or *endo*-pathways. I also found that trisubstituted bicyclo[3.1.0] epoxonium ions, when formed from attack of epoxides to non-stabilized carbenium ions, favor *exo*-selectivity exclusively; when formed from attack of epoxides to oxocarbenium ions, give lower selectivity towards *exo*-cyclization. In order to better understand the origin of the regioselectivity, especially *endo*-selectivity in cascade cyclizations of epoxides, computational analysis was initiated using the B3LYP/6-31G(d) method^{40,41} to mimic the cyclization transition structures in the gas phase. This was performed by using Gaussian03⁴² program in the Houk group at UCLA.

Initial study was carried out on a model reaction of bimolecular nucleophilic addition of dimethyl carbonate to 1,2,2,3-tetramethyloxiranium ion (Figure 25). The dimethyl carbonate can add to either the tertiary or secondary center of the epoxonium ion, with no geometrical constraints in either case. As shown in Figure 25, **TS1** corresponds to the transition structure for nucleophilic addition to the secondary center and **TS2** corresponds to the transition structure for nucleophilic addition to the tertiary center. In both TSs, the breaking and forming C-O bond distances are within the range of 2.0 and 2.2 Å, indicating that both reactions proceed through S_N1 -like transition states. The observation of the longer partial C-O bonds in **TS2** is an indicative of a looser S_N1 transition state, resulting from the formation of a partial tertiary carbocation. The partial tertiary carbocation formation can better stabilize the transition state, making **TS2** lower in energy than **TS1** by 4.3 kcal/mol. The dihedral angles in the absence of geometrical

constraints in **TS1** (O_{ep} - C_{ep} -H-O) and **TS2** (O_{ep} - C_{ep} -C-O) are 141.3° and 147.4°, respectively. The bond distances in **TS1** and **TS2** will be used as the references in the following context, with shorter bond distances being S_N 2-like transition structures.



Figure 25. Transition structures for the addition of dimethyl carbonate to the 1,2,2,3-tetramethyloxiranium ion. The distances are given in Å.

A model reaction was studied to elucidate the origin of the exclusive *endo*-selectivity from trisubstituted bicyclo[4.1.0] epoxonium ions. As shown in Figure 26, the difference between the model reaction and the real reaction $(1.50 \rightarrow 1.81)$, Figure 19) is the replacement of the anomeric methoxy group with a hydrogen atom and the *tert*-butyl carbonate with a methyl carbonate. This model reaction can also be employed to account for the formation of 1.86, 1.88, 1.91, 1.93 and 1.98 in the terminal cyclizations of the corresponding substrates.

The methyl carbonate carbonyl can add either to the secondary or tertiary center of bicyclo[4.1.0] epoxonium ion to form 5-*exo*- or 6-*endo*-cyclization products. The transition structure for *endo*-cyclization (**TS3_endo**) has a lower energy than that of *exo*-cyclization (**TS3_exo**) by 4.5 kcal/mol, which is nearly identical to the energy difference in the unconstrained system. The longer forming C-O bond in **TS3_endo** means a looser S_N1 -like transition state presumably due to the partial formation of the tertiary carbocation. However, **TS3_exo** has an S_N2 -like transition state as evidenced by the shorter breaking and forming C-O bond distances. The dihedral angle in **TS3_exo** has a greater distortion from the unconstrained system (161.1° vs 147.4°) than that in **TS3_endo** (135.5° vs 141.3°). The higher energy and greater distortion of the dihedral angle of **TS3_exo** than those of **TS3_endo** result in the complete preference towards *endo*-cyclization pathway.



Figure 26. Transition structures for the 5-*exo*- and 6-*endo*-cyclizations from trisubstituted bicyclo[4.1.0] epoxonium ion. The distances are given in Å.
The effect of the anomeric methoxy group was investigated using the model reaction illustrated in Figure 27. The methoxy group can be either *cis* or *trans* to the epoxonium ion ring, with the *cis*-isomer being more stable than the *trans*-isomer by 0.3 kcal/mol in the gas phase. It is clear that the incorporation of the methoxy group has negligible effect on the geometries of the transition structures, though **TS4_trans_endo** shows a slightly better leaving group departure and enhanced bond formation compared with **TS3_endo**. The *trans-* and *cis*-methoxy isomers favor the *endo*-cyclization pathway by 4.6 and 4.8 kcal/mol relative to the corresponding *exo*-pathway, respectively, which is similar to the energy difference (4.5 kcal/mol) observed between **TS3_endo** and **TS3_exo**.



Figure 27. Transition structures for 5-*exo*- and 6-*endo*-cyclizations of *cis*- and *trans*-methoxy trisubstituted bicyclo[4.1.0] epoxonium ions. The distances are given in Å.

As depicted in Figure 18, almost no regioselectivity was obtained from the cyclization of disubstituted epoxide **1.48** which does not have the methyl group on the epoxide. To explain the role of the methyl group, an additional model was employed with the angular methyl group being replaced by a hydrogen atom, in which both 5-*exo*- and 6-*endo*-cyclization modes would have partially formed secondary carbocations (Figure 28).

The calculations in the gas phase revealed that the absence of the methyl group has negligible influence on the 5-*exo*-transition structures. However, the two 6-*endo*-transition structures are perturbed substantially, with shorter breaking and forming bond distances being observed, indicating more S_N2 -like character in the transition states. The corresponding dihedral angles are also distorted significantly from 137.8° to 154.3°, and from 136.3° to 154.2° for the *trans*- and *cis*-isomers, respectively. The transition structures of *endo*-modes are energetically lower than the corresponding transition structures of *exo*-modes by 1.4 kcal/mol for both *cis*- and *trans*-isomers, meaning that *endo*-cyclization is slightly favored over *exo*-cyclization for disubstituted bicyclo[4.1.0] epoxonium ions. While experimental results show that *exo*- and *endo*-cyclizations are two competitive pathways as illustrated in Figure 18, computational studies still support that bicyclo[4.1.0] epoxonium ions have a strong tendency towards *endo*-cyclizations.



Figure 28. Transition structures for 5-*exo*- and 6-*endo*-cyclizations of *cis*- and *trans*-methoxy disubstituted bicyclo[4.1.0] epoxonium ions. The distances are given in Å.

In the terminal annulations of diepoxides **1.51** and **1.52**, the bicyclo[3.1.0] epoxonium ions (activated by non-stabilized carbenium ions) preferentially generated 5-*exo*-cyclization products, which is in accord with Baldwin's ring closure rules. This was mimicked by a model reaction depicted in Figure 29. The transition structure for *exo*-cyclization (**TS6_exo**) is similar to the transition structure for the bicyclo[4.1.0] epoxonium ion (**TS3_exo**) in terms of the forming and breaking bond distances and the dihedral angle. The transition structure for *endo*-cyclization (**TS6_endo**), though having similar bond breaking and forming features to the corresponding transition structure for the bicyclo[4.1.0] epoxonium ion (**TS3_endo**), shows a significant decrease in the dihedral angle (123.6° vs 135.5°). As a result, this perturbation leads to the

increase of the energy for **TS3_endo**, making 5-*exo*-pathway energetically favored over 6-*endo*-pathway by 2.7 kcal/mol.



Figure 29. Transition structures for 5-*exo*- and 6-*endo*-cyclizations of trisubstituted bicyclo[3.1.0] epoxonium ion. The distances are given in Å.

When the epoxide was activated by the oxocarbenium ion instead of a non-stabilized carbenium ion, a small amount of *trans*-fused *endo*-cyclization product **1.78** was also isolated from trisubstituted bicyclo[3.1.0] epoxonium ion **1.74**. To elucidate the impact of the anomeric methoxy group on the regiochemical outcomes, another model reaction was employed, as shown in Figure 30. Similar to the transition structures in Figure 27, the methoxy group could be *cis* or *trans* to the epoxonium ion ring. The *trans*-isomer 1.0 kcal/mol more stable than the *cis*-isomer

because of the stabilization between the lone pair of the epoxide oxygen and the anti-bonding orbital of the C-O (methoxy) bond when they are antiperiplanarly oriented.

For the transition structures for the two *cis*-isomers, **TS7_cis_exo** shows a greater leaving group departure in the presence of the anomeric methoxy group while the bond forming character and the dihedral angle remain approximately the same. **TS7_cis_endo**, however, has enhanced bond formation and a widened dihedral angle which is close to those in the transition structures (**TS3_endo**, **TS4_cis_endo** and **TS4_trans_endo**) of trisubstituted bicyclo[4.1.0] epoxonium ions. For the *trans*-isomers, the geometry of **TS7_trans_exo** is essentially unaffected by the incorporation of the methoxy group. Although the breaking and forming bond distances remain nearly the same in **TS7_trans_endo**, the dihedral angle is distorted from 123.6° to 132.9°, which is similar to that in **TS7_cis_endo**. From an energetic view of point, **TS7_trans_endo** has the lowest energy which is attributed to the stabilization of the anomeric effect when the electronegative methoxy group assumes a pseudoaxial position in the forming tetrahydropyran ring. The two transition structures for *exo*-cyclizations have slightly higher energy. In addition, **TS7_cis_endo**, without benefiting from the developing anomeric effect, is the highest in energy.



Figure 30. Transition structures for 5-*exo*- and 6-*endo*-cyclizations of *cis*- and *trans*-methoxy trisubstituted bicyclo[3.1.0] epoxonium ions. The distances are given in Å.

The role of the methoxy group is consistent with the previous observations that the 5-*exo*-regioselectivity decreased when bicyclo[3.1.0] epoxonium ions were generated from combination of epoxides with oxocarbenium ions (**1.74** and **1.82**) rather than non-stabilized carbenium ions (**1.83** and **1.96A**). Also of note is the isolation of *trans*-fused bicycle **1.77** as a single anomer with the methoxy group adopting an axial position.

1.4 CONCLUSIONS

A systematic study has been carried out on the oxocarbenium ion-initiated cascade cyclizations of epoxides under ETIC conditions, in which the impact of the epoxide substitution pattern, ring size of the bicyclic epoxonium ions and the Lewis acidic carbocation structures on regiochemical outcomes was fully investigated. These results clearly revealed that ring size is an important determinant on the regioselectivity of bicyclic epoxonium ion opened by tethered nucleophiles. That is, bicyclo[3.1.0] epoxonium ions show significant to exclusive preference towards exo-cyclization pathways while bicyclo[4.1.0] epoxonium ions show a strong tendency towards endo-cyclizations. This observation could be explained from the computational studies that larger rings can adopt a looser transition state with more S_N1 character, thereby favoring endo-cyclizations. As for smaller rings, endo-TSs are more distorted than exo-TSs, making endo-TSs energetically higher and exo-pathways more favorable. In addition, epoxide substitution pattern also has significant influence on the regioselectivity, especially when bicyclo[4.1.0] epoxonium ions serve as the key intermediates. Trisubstituted bicyclo[4.1.0] epoxonium ions prefer exclusive endo-cyclization pathways with the cyclization proceeding through an S_N1-like transition state due to the better stabilization from a partially formed tertiary carbocation. However, disubstituted bicyclo[4.1.0] epoxonium ions show almost no preference towards exoor endo-cyclizations. Though Lewis acid selection has negligible effect on the regiochemical outcomes of bicyclo[4.1.0] intermediates, it can affect the reaction pathways in a subtle manner when reactions proceed through bicyclo[3.1.0] epoxonium ions. That is, when bicyclo[3.1.0] epoxonium ions are formed from combination of epoxides and non-stabilized carbenium ions, exclusive exo-selectivity is observed; when bicyclo[3.1.0] epoxonium ions are formed from combination of epoxides and oxocarbenium ions, exo-selectivity decreases to some extent. The *endo*-cyclization in this process arises from the anomeric effect that generates from the forming tetrahydropyran ring through *endo*-pathway when the methoxy group adopts a pseudoaxial position, which makes *endo*-transition structure lower in energy. This delicate effect undoubtedly demonstrates that *endo*-cyclization can be achieved to generate tetrahydropyran rings through modification of bicyclic epoxonium ions. The current studies definitely provide a solid base for designing new epoxide substrates to provide polycyclic ether structures efficiently.

2.0 EFFORTS TOWARDS THE TOTAL SYNTHESIS OF (+)-LACTODEHYDROTHYRSIFEROL AND ITS ANALOGS

2.1 INTRODUCTION

(+)-Lactodehydrothyrsiferol (**2.1**), a marine polycyclic ether natural product, was isolated as an amorphous white solid by Fernandez and co-workers in 2002 from seeds of *Laurencia viridis* around the Canary Islands (Figure 31).⁷ Spectroscopic analysis shows that it has a central *trans*-fused pyranopyran structure with a pendent 5-membered lactone ring and an aliphatic side chain that connects the central unit to a *trans*-tetrahydrofuran ring. It is interesting to note that the B ring assumes a chair conformation while the C ring adopts a twist-boat-like conformation. This has also been observed in structurally related natural products through X-ray and NMR spectroscopic analyses.^{7,43-45}

Biological assay shows that **2.1** has modest inhibitory effect on serine/threonine protein phosphatase (PP2A) with IC₅₀ value of 100 μ M.⁴⁶ The structurally related natural product thyrsiferyl-23-acetate (**TA**) is more potent toward the inhibition of PP2A with IC₅₀ values of 4-16 μ M depending on the enzyme concentration. It is worth noting that **TA** exhibits specific inhibitory effect on PP2A, and has no effect on protein phosphatase 1 (PP1), 2B (PP2B), 2C (PP2C), or protein tyrosine phosphatases (PTP).⁴⁷ Though other structurally diverse natural products, such as polyether okadaic acid, polyketide tautomycin, and terpenoid cantharidin, are

much more potent inhibitors of PP2A (IC₅₀ = $0.2 \sim 40$ nM) than **2.1** and **TA**, they also show inhibitory effect on PP1 and/or PP2B (Figure 31).^{48,49} The exclusive selectivity of **TA** is presumably due to its unique structural features and makes itself an ideal tool to study the cellular processes mediated by PP2A. However, it is unclear whether **2.1** can affect the activity of other protein phosphatases besides PP2A. Up to now, no total synthesis of **2.1** has been reported. I am currently pursuing a convergent approach toward the total synthesis of **2.1** and its analogs to further explore their biological activity.



Figure 31. Biologically active natural products and the conformation of the B and C rings in 2.1

and TA

2.2 SYNTHETIC PROGRESS

2.2.1 Preliminary results

Previous studies on the cascade cyclizations of epoxides showed that the reaction of diepoxide **2.2** under ETIC conditions gave a mixture of *exo,exo*-product **2.3** and *endo,endo-*product **2.4** in a combined 61% yield and with a 6.1:1 ratio. In this reaction, tricycle **2.5** from an *exo*-cyclization followed by an *endo*-cyclization was not observed (Figure 32).²⁶ Therefore, construction of the A, B and C rings of **2.1** in a single operation from the diepoxide similar to **2.2** is difficult and a new strategy is required for this purpose.



Figure 32. Cyclization of diepoxide 2.2

As depicted in Figure 33, an alternative approach was proposed. The reaction of diepoxide **2.6** under ETIC conditions is expected to give bicyclic epoxonium ion **2.7** and the terminal cyclization will proceed through addition of carbonate carbonyl to the proximal tertiary center in a kinetically favored fashion to generate tricycle **2.8**. With **2.8** in hand, further elaborations will provide tetracyclic compound **2.9** with the A, B and C rings being installed.



Figure 33. Proposed cyclization of diepoxide 2.6

To validate the transformation from 2.7 to 2.8, I prepared epoxy carbonate 2.14, which will form an epoxonium ion intermediate similar to 2.7 upon cyclization (Figure 34). Oxidation of known alkene 2.10²⁶ with KMnO₄ afforded α -hydroxyl ketone 2.11 in 57% yield.⁵⁰ Olefination of the ketone under Lebel's Rh-catalyzed conditions provided allylic alcohol 2.12 in 72% yield.⁵¹ Epoxidation⁵² of 2.12 catalyzed by VO(acac)₂ in the presence of *tert*-butyl hydroperoxide followed by protection of the primary hydroxyl group with Boc₂O²⁷ gave carbonate 2.14 in excellent yield.



Figure 34. Synthesis of epoxide 2.14

Reaction of **2.14** under ETIC conditions cleanly afforded the desired spiro product **2.17** in excellent yield through addition of carbonate to the tertiary center of bicyclic epoxonium ion **2.16**, which was completely consistent with my expectations (Figure 35).



Figure 35. Cyclization of epoxide 2.14

2.2.2 Current progress

Following the smooth conversion of **2.14** into bicycle **2.17**, I proposed a retrosynthetic approach toward **2.1**. As shown in Figure 36, **2.1** can be obtained from a stereoselective

reduction of ketal **2.18** which will be prepared from coupling of aldehyde **2.19** and vinyl iodide **2.20**.⁵³ These two coupling components can be accessed from cyclizations of diepoxide **2.21** and **2.22**, respectively.



Figure 36. Retrosynthetic analysis of 2.1

Due to the availability of the chiral reagents for asymmetric epoxidations, *ent-2.21* was initially prepared to explore the feasibility of the above route (Figure 37). Coupling of known vinyl bromide 2.23^{54} and aldehyde 2.24 under Fürstner-modified Nozaki-Hiyama-Kishi conditions⁵³ provided allylic alcohol 2.25 in 85% yield. It was subsequently converted into ethyl ester 2.26 through a Johnson-Claisen rearrangement.⁵⁴ Trost⁵⁴ emphasized that the rearrangement efficiency of a similar allylic alcohol is highly dependent on the reaction temperature, with higher temperatures leading to decreased *Z/E* stereoselectivity and lower temperatures leading to decreased yields. Subsequently, 2.26 was converted into allylic alcohol 2.27 through reduction with DIBAL-H and addition with isopropenylmagnesium bromide. Another Johnson-Claisen rearrangement of 2.27 followed by reduction/addition gave homobenzylic alcohol 2.29 in 65%

yield over two steps. Methylation of the secondary alcohol followed by removal of the two silyl groups provided diol **2.30** in good yield. A sequence of Sharpless asymmetric epoxidation,³² Shi asymmetric epoxidation,³¹ and protection of the two hydroxyl groups with Boc₂O afforded the cyclization substrate *ent*-**2.21**. It is worth noting that the first epoxidation gave modest yield due to the unexpected cyclization through addition of the distal hydroxyl group to the epoxide and the diastereoselectivity after the two epoxidations was low (dr ~ 2:1) based on NMR analysis.



Figure 37. Synthesis of *ent*-2.21

Subjecting diepoxide *ent*-**2.21** to ETIC conditions produced desired tricycle **2.31** in 43% yield which contained small amounts of unknown materials (Figure 38). Also isolated from this reaction was tricycle **2.32**. Both **2.31** and **2.32** were oxidized by Jones reagent²⁵ to the corresponding lactones **2.33** and **2.34**, respectively. The relative stereochemical outcomes of the central *cis*-tetrahydropyran and the orientation of the 5-membered carbonate ring in lactone **2.33** were fully confirmed through 2D NMR NOESY studies. Subsequently, the Boc group was removed with TMSOTf in the presence of 2,6-lutidine⁵⁵ to give primary alcohol **2.35** in nearly quantitative yield. Oxidation of the hydroxyl group with Dess-Martin periodinane^{56,57} provided aldehyde *ent*-**2.19** in 61% yield.



Figure 38. Synthesis of advanced intermediate ent-2.19

The above results clearly revealed that aldehyde **2.19** can be prepared from diepoxide **2.21**. As previously mentioned, there are two problems in this sequence. One is the low yield in Sharpless epoxidation and the other is low diastereoselectivity in the epoxidations of **2.30** presumably because of the interference of the hydroxyl group in Sharpless and/or Shi epoxidations. In order to circumvent these two problems, I proposed an alternative route to **2.19**

with replacement of one of the *tert*-butyl carbonates with a terminal alkene (Figure 39). I envisioned that oxidative cleavage of the terminal alkene would yield the desired aldehyde.



Figure 39. An alternative approach to 2.19

The synthesis of **2.36** began with known dienol **2.37**, which was prepared from methyl acrylate and 4-pentenal through a Morita-Baylis-Hillman reaction (Figure 40).^{58,59} Reduction of the methyl ester followed by selective protection of the primary hydroxyl group with TBDPSC1 provided silyl ether **2.38**. Conversion of **2.38** into trienol **2.40** was achieved through a sequence similar to the synthesis of **2.30**. After Sharpless epoxidation, Shi epoxidation and protection of the hydroxyl group with Boc₂O, diepoxide **2.36** was obtained in higher efficiency compared to *ent-2.21*. Additionally, in this case, the diastereoselectivity in the epoxidations is about 4.6:1 with regard to the stereochemical orientations of the two epoxide functionalities.

Under ETIC conditions, diepoxide **2.36** underwent a cascade cyclization to afforded tricyclic product **2.37** (Figure 40). After removal of the anomeric center with Jones reagent, the corresponding lactone was obtained in 17% yield over two steps as a single diastereomer. The lower efficiency in the cascade cyclization is attributed to the intervention of the nucleophilic terminal alkene in the final cyclization process, suggesting that the terminal alkene must be replaced by non-nucleophilic groups. The ensuing cleavage of the terminal olefin under ozonolytic conditions smoothly afforded aldehyde **2.19** in 83% yield.



Figure 40. New approach to aldehyde 2.19

With aldehyde **2.19** in hand, I next addressed to the preparation of vinyl iodide **2.45** (Figure 41). Sharpless asymmetric epoxidation of geraniol followed by Sharpless asymmetric dihydroxylation⁶⁰ gave a mixture of triol **2.22** and tetrahydrofuran **2.42**, which was converted into **2.42** completely with the promotion by pyridinium 10-camphorsulfonate complex.⁶¹ The stereochemical outcome in **2.42** was established through mechanistic analysis. Activation of the primary hydroxyl group with TsCl followed by elimination under basic conditions provided the epoxide⁶² whose tertiary hydroxyl group was protected as silyl ether **2.43** in excellent yield. The epoxide was opened by 1,3-dilithiopropyne⁶³ and the nascent secondary hydroxyl group was protected with TESCl to give bis-silyl ether **2.44**. The terminal alkyne was transformed into the

vinylstannane in the presence of the $Bu_3Sn-AlEt_2$ complex and CuCN in 29% (59% brsm) yield, which was further converted into vinyl iodide **2.45** in good yield.^{64,65}



Figure 41. Synthesis of vinyl iodide 2.45

2.3 FUTURE WORK

Though aldehyde **2.19** could be produced from diepoxides **2.21** and **2.36**, the efficiency for the substrate preparation or cascade cyclization was still low. Therefore, a better substrate is necessary for the total synthesis. I am intended to prepare a new diepoxide **2.49** from known vinyl bromide **2.46**^{66,67} to differentiate the protecting groups for the two primary hydroxyl groups (Figure 42). Coupling of **2.46** with aldehyde **2.24** will afford the allylic alcohol, which will be converted into ethyl ester **2.47** through a Johnson-Claisen rearrangement. Similarly, **2.48** can be obtained through a repeated ester reduction/nucleophilic addition protocol followed by methylation of the secondary alcohol. Removal of the PMB group with DDQ followed by

epoxidations and carbonate formation will deliver diepoxide **2.49**. Since no extra hydroxyl group in the two epoxidations, a better yield from Sharpless epoxidation can be expected and the diastereoselectivity can be retained at a level of 4~5:1. Cyclization of **2.49** under ETIC conditions will give a comparable yield to that of *ent*-**2.21** due to the bulky and non-nucleophilic *tert*-butyldiphenylsilyloxy group.



Figure 42. New approach to aldehyde 2.19

With sufficient amounts of aldehyde **2.19** and vinyl iodide **2.42** in hand, I will next investigate the coupling of these two fragments under Fürstner-modified Nozaki-Hiyama-Kishi conditions (Figure 43).⁵³ Once **2.50** is formed, the allylic alcohol can be selectively oxidized with MnO_2 and the rest two hydroxyl group will be appropriately protected to give **2.51**. Opening of the 5-member carbonate under basic conditions⁶⁸ followed by BiBr₃-promoted ketal formation⁶⁹ will afford pentacycle **2.18**.



Figure 43. Future plan for completion of the total synthesis

From the retrosynthetic analysis, a challenging stereoselective reduction is required to complete the total synthesis. In the presence of suitable Lewis acids, ketal **2.18** will be opened to form oxocarbenium ion **2.52**, and various hydride sources will be examined in hope that bulkier hydrides will provide a better diastereoselectivity by favoring approach from the bottom face in order to avoid the severe steric repulsion from the axial substitution groups when approaching from the top face.⁷⁰ Subsequently, the hydroxyl group can be removed through sulfonate formation and reduction^{71,72} to give desired angular methyl group. Removal of the two hydroxyl groups will furnish the natural product **2.1**. Modification of C-12 stereochemical orientation or at other positions will generate a number of analogs. Biological activity of **2.1** and these analogs

will be investigated toward a series of protein phosphatases and the structure-activity relationship pattern can be established accordingly.

2.4 SUMMARY

I am currently pursuing a convergent approach to the total synthesis of (+)lactodehydrothyrsiferol and its analogs. This route includes the coupling of two functionalized intermediates aldehyde **2.19** and vinyl iodide **2.42**, both of which result from cyclizations of chiral epoxides with the former being obtained through oxocarbenium ion-initiated cascade cyclizations under ETIC conditions and the latter being obtained through a Brønsted acidmediated cyclization. All the stereocenters in **2.1** will be ultimately derived from chiral reagents except the stereochemical outcome at the C-12 center which will be formed through a stereoselective reduction controlled by both substrate and the reducing agent. This remains a challenge to be explored.

3.0 MULTICOMPONENT APPROACH TO THE SYNTHESIS OF OXIDIZED AMIDES THROUGH NITRILE HYDROZIRCONATION

3.1 BACKGROUND

Oxidized amides, in which the carbon atom connected to the nitrogen has a higher oxidation state than the normal (+1) valence, have been discovered in a number of natural products. These compounds usually possess acyl aminal, acyl hemiaminal or enamide functionalities, as exemplified by protein synthesis inhibitors pederin^{73,74} and psymberin,^{75,76} cytotoxin zampanolide,⁷⁷ and cytotoxins apicularen A⁷⁸ and salicylihalamide A,⁷⁹ respectively (Figure 44).



Figure 44. Representative natural products containing oxidized amides

Structure-activity relationship studies have shown that the oxidized amide moieties are closely related to the biological acitivities of these complex compounds (Figure 45).⁸⁰⁻⁸⁴ For example, apicularen A shows strong growth inhibitory effect against the human melanoma cell line SK-MEL-5 with GI₅₀ value of 6 nM. However, its synthetic analogs, with the enamide side chain being replace by simple alkenes or other enamides, have significantly reduced activity.⁸² Due to the structural complexity and interesting biological activities, these natural products have attracted considerable attention from synthetic organic community.



Figure 45. Apicularen A analogs

As for the synthesis of acyl aminals, a common method is through a Curtius rearrangement of acyl azide followed by nucleophilic addition to the isocyanate intermediate, as evidenced in the synthetic efforts towards mycalamides A and B (Figure 46, A).⁸⁵ A direct coupling of the activated carboxylic acid with α -alkoxy amine provides acyl aminal and this strategy has been successfully applied to the total synthesis of mycalamides A and B by Kishi (Figure 46, B).⁸⁶ Alternatively, coupling of carboxylic acid chlorides with alkyl imidates followed by reduction of the newly-formed acyl imines could also deliver acyl aminals (Figure 46, C), whereas the diastereoselectivity in the reduction varies with the substrates.⁸⁷ Our group also developed an efficient approach to the acyl aminals through addition of oxygen-containing nucleophiles to oxidatively generated acyl iminium ions under very mild conditions (Figure 46, D).⁸⁸



Figure 46. Preparation of acyl aminals

Acyl hemiaminals are relatively more difficult to prepare than acyl aminals. In the total synthesis of (+)-zampanolide, Smith employed a Curtius rearrangement of the acyl azide to set up the acyl aminal functionality (Figure 47, A).⁸⁹ After installation of the side chain, the PMB group was removed with DDQ and the desired acyl hemiaminal was obtained as a 1.3:1 mixture

of the two epimers, with the desired antipode of the natural product being slightly favored. Subsequently, Hoye developed a unique approach to the total synthesis of naturally occurring (-)-zampanolide through an aluminum-mediated aza-aldol reaction of the aluminum imidate with (-) -dactylolide and a mixture of 1:1 diastereomers was obtained (Figure 47, B).⁹⁰ In an effort toward the model synthesis of the zampanolide side chain, Porco⁹¹ studied the oxidative decarboxylation of the amino acid derivative and obtained α -acyloxy amide (Figure 47, C). Hydrolysis of the acetate yielded the desired acyl hemiaminal in good yield.



Figure 47. Preparation of acyl hemiaminals

Enamide synthesis has been extensively investigated. A Curtius rearrangement has also been employed to generate isocyanate intermediate from α,β -unsaturated acyl azides, which can be added by alkyllithium or Grignard reagents to afford enamides (Figure 48, A).⁹² This strategy has been utilized in the total synthesis of salicylihalamide A^{93,94} and palmeralide.⁹⁵ Fürstner developed an approach to either E-enamides or Z-enamides from E- or Z-alkenylsilanes stereospecifically (Figure 48, B).⁹⁶ Starting from alkenylsilanes, a sequence of epoxidation, epoxide opening with NaN₃ and reduction gives α -silyl amines which are further converted into the corresponding enamides through acylation and Peterson olefination. Coupling of amides with vinyl iodides, cyclic enol triflates or tosylates under Cu (I)- or Pd (0)-catalyzed conditions delivers enamides (Figure 48, C and D).97,98 Recently, Goossen reported a stereoselective enamide formation via a Ru(II)-catalyzed hydroamination of terminal alkynes, with E-enamides being favored in the presence of tributylphosphine and Z-enamides being preferred when bis(dicyclohexylphosphino)methane was used as the ligand (Figure 48, E).⁹⁹ A traditional Wittig olefination of N-acyl formamides with phosphonium ylides have also been utilized to generate Eenamides (Figure 48, F).¹⁰⁰

While most of the aforementioned methods provide entries into oxidized amides specifically, a general and mild route to all these three types of oxidized amides from a common intermediate needed to be developed. Details will be followed in the subsequent section.



Figure 48. Preparation of enamides

3.2 RESEARCH DESIGN AND RESULTS

Based on the structural features, I proposed that oxidized amides can be accessed from common acylimine intermediates. As depicted in Figure 49, acyl aminals can be prepared from acylimines through alcohol addition,¹⁰¹ acyl hemiaminals can be prepared from acylimines through water addition and enamides can be prepared from acylimines through a tautomerization.



Figure 49. Oxidized amides from acylimines

For the preparation of the key intermediates acylimines, the common known method is condensation of aldehyde with amide (Figure 50).¹⁰²⁻¹⁰⁴ However, when enolizable aldehydes are used, tautomerization of the forming acylimines could be a problem. Though this problem can be tackled by addition of sulfinic acids or sulfinate salts to the reaction system to form α -amido sulfones,^{105,106} reforming and isolating the acylimine is still inefficient, although it is possible.¹⁰⁷



Figure 50. Generation of acylimines

Metalloimines could be acylated with acid chlorides or carboxylic acid anhydrides to give acylimines.^{108,109} Majoral¹¹⁰ reported that when sterically hindered nitriles were treated with Schwartz' reagent (Cp₂Zr(H)Cl)^{111,112} followed by acylation with sterically hindered acid chlorides, acylimines were obtained in excellent yields (Figure 51).¹¹³⁻¹¹⁵ My approach to acylimines begins with simple nitriles as well. Hydrometallation of nitriles will be expected to give metalloimines which will react with acid chlorides to afford desired acylimines (Figure 52). Once acylimines are successfully prepared, investigations of the formation of acyl aminals, acyl hemiaminals and enamides under different conditions can be performed.



Figure 51. Hydrozirconation of nitriles



Figure 52. Proposed acylimine formation from nitriles through hydrometallation and acylation

I initiated my study by using the known α -methoxy nitrile 1^{116} as the substrate, which was prepared from addition of TMSCN to the corresponding dimethyl acetal mediated by BiBr₃ (Figure 53).¹¹⁷ Subjecting **3.1** to Schwartz' reagent in CH₂Cl₂ followed by sequential addition of PhOC(O)Cl and MeOH provided acyl aminals **3.2** and **3.3** in combined 55% yield and with a 2.4:1 diastereomeric ratio (see below for stereochemical assignment). The observation confirmed the formation of acylimine intermediate **3.4** from acylation of *N*-zircono-imine that arose from the nitrile reduction. From a mechanistic point of view, the major product **3.2** resulted from chelation-controlled MeOH addition while the minor product **3.3** was from addition of MeOH through a Felkin pathway.



Figure 53. Acyl aminal formation in initial studies

To assign the relative stereochemical outcomes of the two products, I prepared the α , β dimethoxy carboxylic acid **3.8** from α , β -unsaturated ester **3.7** through Sharpless asymmetric dihydroxylation,⁶⁰ double methylation¹¹⁸ and hydrolysis of the ethyl ester (Figure 54). Subsequently, **3.8** was converted into acyl azide **3.9** which underwent a spontaneous Curtius rearrangement to form isocyanate **3.10**.¹¹⁹ The phenoxide anion, which arose from hydrolysis of diphenylphosphoryl azide with adventitious moisture, added to the isocyanate to form (-)-**3.3** as a single enantiomer which showed identical spectroscopic features to racemate **3.3**.



Figure 54. Confirmation of stereochemical outcomes

Having established the reactivity pattern, I next addressed the formation of acyl aminals and acyl hemiaminals extensively. Ethoxy nitrile **3.11** (prepared in a similar manner to **3.1**) was used as the substrate for exploring the diastereocontrol in the alcohol addition and the acyl aminal formation with different acylating reagents and nucleophiles.

When **3.11** was subjected to hydrozirconation, acylation with isobutyryl chloride and MeOH addition at 0 °C, both CH_2Cl_2 and THF were suitable solvents, with the products **3.12** and **3.13** being isolated in combined 75% and 64% yields, respectively (entries 1 and 2, Figure 55). It is worth noting that chelation control was preferred in CH_2Cl_2 while Felkin-pathway was slightly

favored in THF. The reversed stereoselectivity may be explained by the formation of hydrogen bonds between THF and MeOH, which results in the weakening of chelation control. Since CH_2Cl_2 is a good solvent for chelation control, several conditions were tested to improve it. Simply lowering down the temperature for MeOH addition to -78 °C gave a slightly better result (entry 3). When proper chelating Lewis acids were employed, chelation control could be enhanced to a synthetically useful level. As shown in entries 4 and 5, when MeOH addition was carried out in the presence of a stoichiometric amount of $Zn(OTf)_2$ or $Mg(ClO_4)_2$, a decent diastereocontrol (dr = 5.0:1 or 5.7:1) was accomplished with the reaction efficiency being retained.

NC4 OEt 3.11		Cp ₂ Zr(H)Cl,solvent then ^{<i>i</i>} PrC(O)Cl then Lewis acid then MeOH, temperature		O OMe	+ H H OEt 3.13	
				N H OEt		
				3.12		
	entry	solvent	Lewis acid	temperature	yield (3.12:3.13)	
	1	CH ₂ Cl ₂	N/A	0°C	75% (2.3:1)	
	2	THF	N/A	0°C	64% (1:1.4)	
	3	CH_2CI_2	N/A	0°C	N/A (3.5:1)	
	4	CH_2CI_2	Zn(OTf) ₂	-78 °C	70% (5.0:1)	
	5	CH ₂ Cl ₂	Mg(ClO ₄) ₂	-78 °C	71% (5.7:1)	

Figure 55. Optimization of chelation control

Electrophiles other than isobutyryl chloride were also investigated (Figure 56). Acylation of the metalloimine with α -methoxyacetyl chloride give a mixture of **3.14** and **3.15** in a combined 69% yield and 1.7:1 diastereomeric ratio. The products are electronically similar to the acyl aminals in pederin and psymberin. CbzCl is also a suitable acylating reagent, with *N*,*O*-

acetals **3.16** and **3.17** being isolated in 64% overall yield. In this case, the Cbz group can serve as a protecting group and can be removed readily. Unfortunately, when the metalloimine was acylated with methanesulfonic anhydride, only modest yield of sulfonyl aminals **3.18** and **3.19** were obtained. From this reaction, considerable amounts of the aldehyde from hydrolysis of the metalloimine were also isolated.



Figure 56. Acyl aminal formation from 3.11 with various electrophiles

Besides MeOH, other heteronucleophiles also afforded satisfactory results when acylimine **3.20** was utilized as the common intermediate (Figure 57). Sterically hindered ^{*t*}BuOH had no influence on the reactivity, with the desired acyl aminals being isolated in 71% yield and the Felkin-pathway being favored. PhOH and PhSH are also suitable for this reaction, providing 69% and 72% yields, respectively. It is noteworthy that the chelation-controlled product **3.23** was dominant in the case of PhOH while the product **3.26** from Felkin-type pathway was

predominantly formed in the case of PhSH as the nucleophile. This observation is consistent with the hydrogen-bond-forming abilities of these two nucleophiles.



Figure 57. Acyl aminal formation from 3.11 with various nucleophiles

Nitriles with different substitutions are good substrates for the synthesis of acyl aminals and acyl hemiaminals (Figure 58). Subjecting α -benzoyloxy nitrile **3.27**¹²⁰ to hydrozirconation, acylation with isobutyryl chloride and MeOH addition, acyl aminal **3.28** was isolated in 64% yield (contaminated with 4% BnOH), indicating that ester groups can be tolerated in hydrozirconation.¹²¹ Instead of MeOH addition, a simple aqueous workup after the acylimine formation provided acyl hemiaminal **3.29** in 52% yield, which is structurally relevant to the zampanolide side chain. Also isolated from this reaction was **3.29A** in 13% yield which resulted from addition of BnOH to the acylimine intermediate. It is worth noting that from the reactions of **3.27**, the side product from migration of the benzoyl group to the metalloimine nitrogen was not observed. Octyl cyanide **3.30**, with no branching at the α -carbon, afforded the desired acyl
aminal **3.31** and acyl hemiaminal **3.32** in good yields as well. It was found that THF is a better solvent than CH_2Cl_2 to suppress the acylimine tautomerization to the corresponding enamide. Aromatic nitriles also proved to be excellent substrates, with acyl aminal **3.34** being obtained in 73% isolated yield from phenyl cyanide **3.33** though it underwent a much slower hydrozirconation than aliphatic nitriles.



Figure 58. Acyl aminal and acyl hemiaminal synthesis from various nitriles

Having achieved smooth transformations from nitriles to acyl aminals and acyl hemiaminals, I next examined the enamide synthesis. Octyl cyanide **3.30** and isobutyryl chloride were used to explore the optimum reaction conditions (Figure 59). In CH_2Cl_2 , 22% of the desired *E*-enamide **3.36** was obtained when the metalloimine was acylated in the presence of Et_3N

followed by addition of Lewis acid BF₃•OEt₂ (entry 1, Table 1). In the absence of BF₃•OEt₂, a mixture of **3.36** and **3.37** in ~1:1 ratio resulted from acylimine **3.35** (entry 2); without Et₃N base, no product (**3.36** or **3.37**) was observed (entry 3). When the reaction was conducted in THF, acylimine **3.35** was successfully generated in the presence of Et₃N and was smoothly tautomerized to afford *E*-enamide **3.36** in 57% yield in the presence of BF₃•OEt₂ (entry 4). From this reaction, only minimum amount of *Z*-enamide **3.37** was observed. When the tautomerization was performed in the absence of BF₃•OEt₂, only trace amount of **3.36** was observed (entry 5); and when the metalloimine was acylated in the absence of Et₃N, less than 10% yield of **3.36** was isolated (entry 6). These results convincingly demonstrated that both Et₃N and BF₃•OEt₂ are crucially important in this reaction, presumably due to their synergistic effect in converting **3.35** to **3.36**. Also of note is that use of more than 1 equiv. of isobutyryl chloride would result in the formation of significant amount of diacylation product **3.38**. Following the established conditions, allylic nitrile **3.40**, prepared in four steps from methacrolein and 1-dodecene, gave rise to *E*,*E*-dienamide **3.41** in 62% yield (Figure 60).



Figure 59. Synthesis of enamide 3.36 from octyl cyanide 3.30



Table 1. Optimization of reaction conditions for enamide formation



Figure 60. Synthesis of *E*,*E*-dienamide 3.41 from allylic nitrile 3.40

With successful synthesis of oxidized amides from simple nitrile substrates, I next applied this methodology to the synthesis of a more complex model compound that is related to pederin and psymberin. For this purpose, tetrahydropyranyl nitrile **3.43** was prepared (by Michael Green in the Floreancig group at the University of Pittsburgh) in its racemic form from known ketone **3.42**⁸⁸ through methylation, aluminum-mediated reduction of ketone,¹²² cleavage of the terminal alkene, acylation and displacement of the anomeric acetate group with cyanide (Figure 61).

Reaction of **3.43** through a sequence of hydrozirconation, acylation with isobutyryl chloride and MeOH addition at 0 °C provided desired acyl aminal **3.44** and its diastereomer **3.45** in a combined 75% yield and with a 1.9:1 ratio favoring the chelation control (entry 1, Table 2). Also isolated for this reaction was amide **3.46** in 8% yield, which resulted from direct reduction of the acylimine intermediate by the slightly excess amount of Cp₂Zr(H)Cl. Conducting the MeOH addition at lower temperature (-78 °C) was found to slightly improve the diastereoselectivity (entry 2). Employment of Mg(ClO₄)₂, which promoted chelation control to a considerable extent (*cf.* Figure 55), did not provide a better diastereocontrol, with acyl aminals being obtained in 77% yield and 2.3:1 ratio (entry 3). Although only moderate diastereocontrol was obtained under the conditions studied here, the high yield of the acyl aminal together with the desired stereochemical orientation still makes this method attractive for the synthesis of acyl aminals with similar structures in natural products and their analogues.



Figure 61. Synthesis of tetrahydropyranyl acyl aminals

Table 2.	Reaction	of 3.43	under	various	conditions

entry	temperature	Lewis acid (1 eq.)	yield (3.44+3.45)	dr (3.44/3.45)	yield (3.46)
1	0 °C	N/A	75%	1.9:1	8%
2	-78 °C	N/A	71%	2.3:1	10%
3	-78 °C	Mg(ClO ₄) ₂	77%	2.3:1	10%

3.3 CONCLUSIONS

An efficient one-pot approach to the synthesis of oxidized amides from nitriles was developed. In this process, the acylimines, which are generated from hydrozirconation of nitriles followed by acylation with suitable electrophiles, serve as the common intermediates to deliver acyl aminals, acyl hemiaminals or enamides through nucleophilic addition or tautomerization. In the acyl aminal formation, moderate to good diastereocontrol could be achieved with the assistance of Lewis acids through chelation-controlled nucleophilic addition. It was also found that the base and Lewis acid played a synergistic role in the *E*-enamide syntheses. In the absence of base or Lewis acid, low efficiency for *E*-enamide formation was observed. With the known methods for the synthesis of nitriles, along with various available electrophiles and nucleophiles, this method will provide a convenient and effective access to oxidized amides with diverse structures.

APPENDIX A

STUDIES ON THE STRUCTURE/REACTIVITY REALTIONSHIPS OF BICYCLIC EPOXONIUM IONS AND TETHERED NUCLEOPHILES (SUPPORTING INFORMATION)

General Experimental Proton (¹H NMR) and carbon (¹³C NMR) nuclear magnetic resonance spectra were recorded at ambient temperatures on Bruker Avance 300 spectrometer at 300 MHz and 75 MHz, respectively, Bruker Avance 500 spectrometer at 500 MHz and 125 MHz, or at Bruker Avance 600 spectrometer at 600 MHz and 151 MHz if specified. The chemical shifts are given in parts per million (ppm) on the delta (δ) scale. The solvent peak was used as a reference value, for ¹H NMR: CDCl₃ = 7.27 ppm, C₆D₆ = 7.15 ppm, for ¹³C NMR: CDCl₃ = 77.23. Data are reported as follows: (s = singlet; d = doublet; t = triplet; q = quartet; dd = doublet of doublets; dt = doublet of triplets; br = broad; app = apparently). High resolution mass spectra were recorded on a MICROMASS AUTOSPEC (for EI) or WATERS Q-TOF API-US (for ESI) spectrometer. Infrared (IR) spectra were collected on a Mattson Cygnus 100 spectrometer. Samples for IR were prepared as a thin film on a NaCl plate by dissolving the compound in CH₂Cl₂ and then evaporating the CH₂Cl₂. Optical rotations were measured on a Perkin-Elmer 241 polarimeter. Activated 4 Å molecular sieves were obtained through drying in oven at 150 °C overnight. Tetrahydrofuran and diethyl ether were dried by passage through an activated alumina column under positive N₂ pressure. Methylene chloride and benzene were distilled under N₂ from CaH₂. Diphenylmethane was purchased from Aldrich and used without further purification. Anhydrous DMF and MeI were purchased from Acros. *m*CPBA was purchased from Acros and purified according to the standard procedure (*cf.* Purification of Laboratory Chemicals, 4th Ed., by Armarego, W. L. F. and Perrin, D. D.). Boc₂O and *N*-methylimidole were purchased from Aldrich and used as received. Toluene and 1,2-dichloroethane were purchased from Fisher Scientific and dried with 4 Å molecular sieves overnight prior to use. Analytical TLC was performed on E. Merck pre-coated (0.25 mm) silica gel 60F-254 plates. Visualization was done under UV (254 nm). Flash chromatography was done using ICN SiliTech 32-63 60 Å silica gel. Reagent grade ethyl acetate, diethyl ether, and hexanes (commercial mixture) were purchased from EM Science and used as is for chromatography. All reactions were performed in oven or flame-dried glassware under a positive pressure of N₂ with magnetic stirring unless otherwise noted.

1,1-Diphenylhex-5-en-2-ol (A1)

Ph₂CH A solution of diphenylmethane (1.262 g, 7.50 mmol) in THF (7.5 ml) in a two-necked round-bottom flask was treated dropwise *n*-BuLi (1.6 M in hexanes, 4.7 mL, 7.5 mmol). The resulting deep orange solution was refluxed for 1 h and then cooled to 0 °C. A solution of 4-pentenal (0.252g, 3.00 mmol) in THF (3.0 mL) was added dropwise and the flask formerly containing 4-pentenal was rinsed with THF (2 x 1.0 mL). The reaction mixture was stirred at 0 °C for 1 h, then warmed to room temperature and quenched by slow addition of saturated NH₄Cl solution (10 mL). The mixture was poured onto water (20 mL)

and extracted with Et₂O (3 x 30 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (3% - 12% Et₂O in hexanes) to give the secondary alcohol **A1** (0.628 g, 83.0%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.42-7.17 (m, 10H), 5.80 (ddt, *J* = 17.0, 13.4, 6.8 Hz, 1H), 5.06-4.94 (m, 2H), 4.42-4.35 (m, 1H), 3.90 (d, *J* = 8.4 Hz, 1H), 2.35-2.12 (m, 2H), 1.67-1.41 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 142.5, 141.6, 138.7, 129.0, 129.0, 128.9 128.4, 127.1, 126.8, 115.1, 73.4, 59.1, 34.4, 30.4; IR (neat) 3560, 3062, 2916, 1640, 1598, 1494, 1451, 1080, 913, 745, 703 cm⁻¹; HRMS (EI): *m/z* calcd for C₁₈H₂₀O (M⁺⁺) 252.1514, found 252.1523.

5-Methoxy-6,6-diphenylhex-1-ene (A2)

Ph₂CH OMe The secondary alcohol A1 (0.585 g, 2.32 mmol) in anhydrous DMF (14.0 mL) at 0 °C was treated with NaH (60% dispersion in mineral oil, 0.232 g,

5.80 mmol) and the suspension was stirred at 0 °C for 30 min. MeI (0.58 mL, 9.28 mmol) was added dropwise and the cold bath was then removed. After stirred for 2 h at room temperature, the reaction was quenched with water (20 mL) cautiously and extracted with Et₂O (3 x 40 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The resulting residue was purified by flash chromatography (3% - 6% Et₂O in hexanes) to give the secondary alcohol **A2** (0.598 g, 96.8%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.44-7.16 (m, 10H), 5.80 (ddt, *J* = 17.0, 13.3, 6.7 Hz, 1H), 5.06-4.95 (m, 2H), 4.05 (d, *J* = 8.2 Hz, 1H), 3.98 (m, 1H), 3.19 (s, 3H), 2.25-2.15 (m, 2H), 1.70-1.54 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 142.9, 142.5, 138.7, 129.1, 128.7, 128.4, 126.5, 126.4, 114.9, 83.2, 58.1, 56.4, 31.8, 29.6; IR (neat) 3027, 2928, 1640, 1599, 1495, 1451, 1101, 911, 745, 701 cm⁻¹; HRMS (EI): *m/z* calcd for C₁₅H₁₅O (M-C₄H₇)⁺⁺ 211.1123, found 211.1125.

4-Methoxy-5,5-diphenylpentanal (1.56)

H At -78 °C, the terminal olefin A2 (193 mg, 0.724 mmol) in CH₂Cl₂ (7.5 mL) was bubbled gently with ozone until the solution retained a deep blue color and then PPh₃ (570 mg, 2.17 mmol) was added in one portion. The mixture was warmed to room temperature, stirred for 3 h and then concentrated. The residue was purified by column chromatography (20% - 30% Et₂O in hexanes) to give the aldehyde **1.56** (186 mg, 95.8%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 9.74 (t, J = 1.7 Hz, 1H), 7.44-7.21 (m, 10H), 4.04-3.96 (m, 2H), 3.12 (s, 3H), 2.60-2.45 (m, 2H), 2.05-1.95 (m, 1H), 1.80-1.68 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 202.3, 142.3, 142.2, 128.8, 128.6, 128.5, 126.8, 126.6, 83.0, 58.3, 56.6, 39.8, 25.3; IR (neat) 2928, 2827, 2726, 1722, 1495, 1451, 1113, 747, 704 cm⁻¹; HRMS (EI): *m/z* calcd for C₁₅H₁₅O (M-C₃H₅O) ⁺⁺ 211.1123, found 211.1124.

(*E*)-Ethyl 6-methoxy-7,7-diphenylhept-2-enoate (A3)

 Ph_2CH_{OMe} At 0 °C, triethyl phosphonoacetate (0.25 mL, 1.27 mmol) was added dropwise to a suspension of NaH (60% dispersion in mineral oil, 51.0 mg, 1.27 mmol) in THF (4.0 mL) and the resulting solution was stirred at 0 °C for 30 min. The aldehyde **1.56** (171 mg, 0.637 mmol, dissolved in 1.0 mL THF) was introduced dropwise and the flask formerly containing the aldehyde was rinsed with THF (2 x 0.5 mL). The reaction mixture was stirred at 0 °C for 1 h, then quenched with saturated NH₄Cl (5 mL) and poured onto water (10 mL). The mixture was extracted with CH₂Cl₂ (3 x 25 mL) and the organic extracts were dried (MgSO₄) and concentrated. The residue was purified by column chromatography (10% - 15% EtOAc in hexanes) to give the ethyl ester **A3** (202 mg, 93.8%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.50-7.17 (m, 10H), 6.96 (dt, *J* = 15.6, 6.9 Hz, 1H), 5.82 (br d, *J* = 15.6 Hz, 1H), 4.22 (q, *J* = 7.1 Hz, 2H), 4.04 (d, *J* = 8.5 Hz, 1H), 3.99-3.94 (m, 1H), 3.19 (s, 3H), 2.44-2.25 (m, 2H), 1.75-1.50 (m, 2H), 1.32 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.8, 149.0, 142.5, 142.2, 128.9, 128.8, 128.5, 128.5, 126.7, 126.5, 121.6, 83.0, 60.3, 58.2, 56.4, 30.8, 27.9, 14.4; IR (neat) 2980, 2931, 1717, 1653, 1495, 1451, 1267, 1202, 1109, 1043, 746, 704 cm⁻¹; HRMS (EI): *m/z* calcd for C₂₀H₂₁O₂ (M-C₂H₅O) ^{+•} 293.1542, found 293.1538.

(3-(3-Methoxy-4,4-diphenylbutyl)oxiran-2-yl)methanol (A4)

At -78 °C, DIBAL-H (1 M in hexanes, 1.5 mL, 1.5 mmol) was Ph₂CH、 added dropwise to a solution of the ethyl ester A3 (202 mg, 0.598 OMe mmol) in THF (6.0 mL). The mixture was stirred at -78 °C for 30 min, then quenched with saturated sodium tartrate (6.0 mL) and diluted with water (5 mL). The mixture was warmed to room temperature and stirred vigorously for 2 h. After that time, the mixture was extracted with CH_2Cl_2 (3 x 30 mL) and the extracts were dried (MgSO₄) and concentrated. The residue was dissolved in CH₂Cl₂ (6.0 mL) and cooled to 0 °C. NaHCO₃ (100 mg, 1.20 mmol) and mCPBA (pure, 134 mg, 0.777 mmol) were added sequentially. The suspension was stirred at 0 °C for 1.5 h, then quenched with saturated Na₂S₂O₃ (3 mL) and water (10 mL). The mixture was stirred at room temperature for 30 min, and extracted with CH₂Cl₂ (3 x 40 mL). The extracts were dried (MgSO₄), and concentrated and the residue was purified by column chromatography (40% - 70%EtOAc in hexanes containing 0.5% Et₃N) to give the epoxy alcohol A4 (132 mg, 70.4%, dr ~ 1:1) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.39-7.17 (m, 10H), 4.02-3.94 (m, 2H), 3.91-3.82 (m, 1H), 3.66-3.56 (m, 1H), 3.17/3.16 (s, 3H), 2.93-2.86 (m, 2H), 1.81-1.47 (m, 4H);

¹³C NMR (75 MHz, CDCl₃) δ 142.6, 142.3, 142.2, 128.9, 128.8, 128.8, 128.6, 128.5, 126.7, 126.5, 83.2, 83.1, 61.9, 61.8, 58.6, 58.6, 58.2, 57.9, 56.5, 56.1, 56.0, 28.8, 28.0, 27.4, 27.1; IR (neat) 3428, 2928, 1495, 1451, 1095, 746, 704 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₀H₂₄O₃Na [M+Na]⁺ 335.1623, found 335.1636.

tert-Butyl (3-(3-methoxy-4,4-diphenylbutyl)oxiran-2-yl)methyl carbonate (1.47)

The epoxy alcohol A4 (124.0 mg, 0.397 mmol) in dry toluene Ph₂CH (4.0 mL) at 0 °C was treated with 1-methylimidazole (32 $\mu L,$ ÓMe 0.397 mmol) followed by Boc₂O (173 mg, 0.794 mmol). The mixture was stirred at 0 °C for 2 h, then at room temperature for 1 h. After that time, the reaction was guenched with water (15 mL) and extracted with CH₂Cl₂ (3 x 20 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was azeotroped with hexanes (3 x 10 mL) and then purified by flash chromatography (15% - 20% EtOAc in hexanes containing 0.5% Et₃N) to give the *t*-butyl carbonate 1.47 (146.8 mg, 89.7%, dr ~ 1:1) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.38-7.16 (m, 10H), 4.23-4.18 (m, 1H), 3.99-3.92 (m, 3H), 3.14/3.13 (s, 3H), 2.94-2.91 (m, 1H), 2.81-2.77 (m, 1H), 1.79-1.42 (m, 4H), 1.48 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 153.4, 142.6, 142.3, 142.2, 128.9, 128.9, 128.8, 128.6, 128.5, 126.7, 126.5, 83.1, 83.0, 82.7, 67.2, 67.1, 58.3, 57.9, 56.8, 56.6, 56.5, 56.2, 55.2, 28.8, 28.0, 27.9, 27.4, 27.1; IR (neat) 2981, 2933, 1743, 1495, 1452, 1370, 1281, 1163, 1101, 912, 733, 704 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₅H₃₂O₅Na $[M+Na]^+$ 435.2147, found 435.2140.

1,1-Diphenylhept-6-en-2-ol (A5)

Ph₂CH \longrightarrow At -78 °C, a solution of DMSO (1.1 mL, 15.0 mmol) in CH₂Cl₂ (2 mL) OH

was added dropwise to a mixture of $(COCl)_2$ (0.65 mL, 7.5 mmol) in CH₂Cl₂ (20 mL). After 10 min, a solution of 5-hexen-1-ol (0.60 mL, 5.0 mmol) in CH₂Cl₂ (5 mL) was introduced. The white suspension was stirred at -78 °C for 30 min, then Et₃N (4.2 mL, 30.0 mmol) was added and the suspension was stirred for 30 min. After that time, the reaction was warmed to room temperature, diluted with CH₂Cl₂ (80 mL) and washed with saturated NaHCO₃ (80 mL). The organic layer was dried (Na₂SO₄), filtered and concentrated to give crude 5-hexenal.

A solution of diphenylmethane (2.52 g, 15.0 mmol) in THF (15 ml) in a two-necked roundbottom flask was treated dropwise n-BuLi (1.6 M in hexanes, 9.4 mL, 15.0 mmol). The resulting deep orange solution was refluxed for 1 h and then cooled to 0 °C. A solution of as-prepared crude 5-hexenal in THF (3.0 mL) was added dropwise and the flask formerly containing 4pentenal was rinsed with THF (2 x 1.0 mL). The reaction mixture was stirred at 0 °C for 1 h, then warmed to room temperature and quenched by slow addition of saturated NH₄Cl solution (15 mL). The mixture was poured onto water (40 mL) and extracted with Et₂O (3 x 50 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (3% - 15% Et₂O in hexanes) to give the secondary alcohol A5 (0.928 g, 70%, two steps) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.43-7.19 (m, 10H), 5.80 (tdd, J = 16.9, 13.2, 6.6 Hz, 1H), 5.02-4.93 (m, 2H), 4.40-4.36 (m, 1H), 3.92 (d, J = 8.3 Hz, 1H), 2.12-1.96 (m, 2H), 1.76-1.37 (m, 5H); ¹³C NMR (75 MHz, CDCl₃) δ 142.6, 141.6, 138.9, 129.0, 128.8, 128.4, 127.0, 126.7, 114.7, 73.8, 59.0, 34.6, 33.8, 25.3; IR (neat) 3561, 3453, 3026, 2918, 1640, 1598, 1494, 1451, 1080, 911, 746, 703 cm⁻¹; HRMS (ESI): *m/z* calcd for C₁₉H₂₂ONa $[M+Na]^+$ 289.1568, found 289.1600.

6-Methoxy-7,7-diphenylhept-1-ene (A6)

Ph₂CH OMe The secondary alcohol **A5** (0.918 g, 3.45 mmol) in anhydrous DMF (20 mL) at 0 °C was treated with NaH (60% dispersion in mineral oil, 0.345

g, 8.62 mmol) and the suspension was stirred at 0 °C for 30 min. MeI (0.86 mL, 13.8 mmol) was added dropwise and the cold bath was then removed. After stirred overnight, the reaction was quenched with water (40 mL) cautiously and extracted with Et₂O (3 x 60 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The resulting residue was purified by flash chromatography (1% - 5% Et₂O in hexanes) to give the secondary alcohol **A6** (0.921 g, 95.4%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.45-7.20 (m, 10H), 5.81 (tdd, *J* = 16.9, 13.1, 6.7 Hz, 1H), 5.05-4.96 (m, 2H), 4.06 (d, *J* = 8.3 Hz, 1H), 3.98-3.94 (m, 1H), 3.22 (s, 3H), 2.15-1.95 (m, 2H), 1.70-1.43 (m, 4H); ¹³C NMR (75 MHz, CDCl₃) δ 143.1, 142.6, 129.0, 128.7, 128.4, 126.5, 126.4, 114.7, 83.6, 58.0, 56.3, 34.0, 31.7, 24.5; IR (neat) 2934, 1640, 1495, 1451, 1101, 910, 737, 701 cm⁻¹; HRMS (EI): *m/z* calcd for C₂₀H₂₄O (M⁺⁺) 280.1827, found 280.1823.

5-Methoxy-6,6-diphenylhexanal (1.58)

Ph₂CH \rightarrow OMe \rightarrow At -78 °C, the terminal olefin A6 (400 mg, 1.43 mmol) in CH₂Cl₂ (145 mL) was bubbled gently with ozone until the solution retained a deep blue color and then PPh₃ (750 mg, 2.86 mmol) was added in one portion. The mixture was warmed to room temperature, stirred overnight and then concentrated. The residue was purified by column chromatography (20% - 30% Et₂O in hexanes) to give the aldehyde **1.58** (386 mg, 95.9%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ) δ 9.71 (t, *J* = 1.6 Hz, 1H), 7.39-7.16 (m, 10H), 4.02 (d, *J* = 8.4 Hz, 1H), 3.94-3.88 (m, 1H), 3.12 (s, 3H), 2.38 (dt, *J* = 7.2, 1.5 Hz, 2H), 1.79-1.70 (m, 2H), 1.58-1.53 (m, 1H), 1.48-1.42 (m, 1H),;¹³C NMR (75 MHz, CDCl₃) δ

202.2, 142.7, 142.3, 128.9, 128.7, 128.6, 128.4, 126.6, 126.5, 83.5, 58.1, 56.2, 44.0, 31.7, 17.9; IR (neat) 2929, 2825, 2721, 1722, 1495, 1451, 1112, 747, 704 cm⁻¹; HRMS (ESI): *m/z* calcd for C₁₉H₂₂ONa [M+Na]⁺ 305.1517, found 305.1564.

(E)-Ethyl 7-methoxy-8,8-diphenyloct-2-enoate (A7)

OEt At 0 °C, triethyl phosphonoacetate (0.52 mL, 2.60 mmol) was Ph₂CH. || 0 ÓMe added dropwise to a suspension of NaH (60% dispersion in mineral oil, 104 mg, 2.60 mmol) in THF (9.0 mL) and the resulting solution was stirred at 0 °C for 30 min. The aldehyde **1.58** (367 mg, 1.30 mmol, dissolved in 3.0 mL THF) was introduced dropwise and the flask formerly containing the aldehyde was rinsed with THF (2 x 0.5 mL). The reaction mixture was stirred at 0 °C for 1 h, then quenched with saturated NH₄Cl (10 mL) and poured onto water (20 mL). The mixture was extracted with CH₂Cl₂ (3 x 30 mL) and the organic extracts were dried (MgSO₄) and concentrated. The residue was purified by column chromatography (10% - 15% EtOAc in hexanes) to give the ethyl ester A7 (426 mg, 93%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.44-7.20 (m, 10H), 6.96 (td, J = 15.6, 6.8 Hz, 1H), 5.82 (d, J = 15.6 Hz, 1H), 4.24 (q, J = 7.1 Hz, 2H), 4.05 (d, J = 8.4 Hz, 1H), 3.98-3.92 (m, 1H), 3.21 (s, 3H), 2.25-2.10 (m, 2H), 1.71-1.44 (m, 4H), 1.34 (t, J = 7.1 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) § 166.8, 149.0, 142.8, 142.4, 128.9, 128.7, 128.5, 128.4, 126.6, 126.4, 121.6, 83.5, 60.3, 58.1, 56.3, 32.3, 31.8, 23.6, 14.4; IR (neat) 2979, 2934, 1718, 1654, 1495, 1451, 1368, 1269, 1186, 1098, 1043, 746, 703 cm⁻¹; HRMS (EI): m/z calcd for C₂₁H₂₃O₃ (M-C₂H₅O) ^{+•} 307.1698, found 307.1684.

(3-(4-Methoxy-5,5-diphenylpentyl)oxiran-2-yl)methanol (A8)

Ph₂CH OH At -78 °C, DIBAL-H (1 M in hexanes, 2.9 mL, 2.9 mmol) was added dropwise to a solution of the ethyl ester A7(413 mg, 1.17

mmol) in THF (12 mL). The mixture was stirred at -78 °C for 30 min, then guenched with saturated sodium tartrate (12 mL) and diluted with water (10 mL). The mixture was warmed to room temperature and stirred vigorously for 2 h. After that time, the mixture was extracted with CH₂Cl₂ (3 x 40 mL) and the extracts were dried (MgSO₄) and concentrated. The residue was passed through a short silica gel column and eluted with 50% EtOAc in hexanes. The product was concentrated and dissolved in CH₂Cl₂ (12 mL) and cooled to 0 °C. NaHCO₃ (196 mg, 2.34 mmol) and mCPBA (pure, 262 mg, 1.52 mmol) were added sequentially. The suspension was stirred at 0 °C for 3.5 h, then quenched with saturated Na₂S₂O₃ (10 mL) and water (20 mL). The mixture was stirred at room temperature for 20 min, and extracted with CH₂Cl₂ (3 x 30 mL). The extracts were dried (MgSO₄), and concentrated and the residue was purified by column chromatography (50% - 70% EtOAc in hexanes containing 0.5% Et₃N) to give the epoxy alcohol A8 (348 mg, 91%, dr ~ 1:1) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.39-7.17 (m, 10H), 4.00 (d, J = 8.4 Hz, 1H), 3.93-3.81 (m, 2H), 3.61-3.53 (m, 1H), 3.17/3.16 (s, 3H), 2.89-2.84 (m, 2H), 1.66-1.40 (m, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 142.8, 142.4, 128.9, 128.6, 128.5, 128.4, 126.5, 126.4, 83.5, 61.8, 58.6, 58.5, 58.1, 58.0, 56.2, 56.0, 55.9, 32.0, 31.7, 21.6; IR (neat) 3435, 2929, 1599, 1495, 1451, 1098, 910, 732, 703 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₁H₂₆O₃Na [M+Na]⁺ 349.1780, found 349.1792.

tert-Butyl (3-(4-methoxy-5,5-diphenylpentyl)oxiran-2-yl)methyl carbonate (1.48)



The epoxy alcohol **A8** (336 mg, 1.03 mmol) in anhydrous toluene (10 mL) at 0 °C was treated with 1-methylimidazole (82 μ L, 1.03 mmol) followed by Boc₂O (449 mg, 2.06 mmol). The mixture was stirred at 0 °C for 30 min, then at room temperature for 1.5 h. After that time, the reaction was quenched with water (15 mL) and extracted with CH₂Cl₂ (3 x 30 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was azeotroped with hexanes (3 x 10 mL) and then purified by flash chromatography (15% - 20% EtOAc in hexanes containing 0.5% Et₃N) to give the *t*-butyl carbonate **1.48** (369 mg, 84%, dr ~ 1:1) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.39-7.16 (m, 10H), 4.23-4.18 (m, 1H), 4.00 (d, *J* = 8.4 Hz, 1H), 3.97-3.88 (m, 2H), 3.16/3.16 (s, 3H), 2.96-2.91 (m, 1H), 2.81-2.79 (m, 1H), 1.59-1.43 (m, 6H), 1.50 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 153.4, 142.8, 142.4, 129.0, 128.7, 128.6, 128.4, 126.6, 126.4, 83.6, 82.7, 67.2, 58.2, 58.1, 56.6, 56.5, 56.3, 55.2, 55.2, 32.0, 32.0, 31.7, 27.9, 21.6, 21.6; IR (neat) 2980, 2936, 1742, 1495, 1452, 1370, 1280, 1163, 1099, 858, 733, 704 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₆H₃₄O₅Na [M+Na]⁺ 449.2304, found 449.2308.

((3,3-Dimethyloxiran-2-yl)methoxy)(*tert*-butyl)dimethylsilane (1.59)

A solution of (3-methylbut-2-enyloxy)(*tert*-butyl)dimethylsilane (6.012 g, 30.0 mmol) in CH₂Cl₂ (300 mL) at 0 °C was treated with NaHCO₃ powder (5.04 g, 60.0 mmol) followed by *m*-chloroperbenzoic acid (70-75%, 7.25 g, 31.5 mmol). The reaction mixture was stirred at 0 °C for 1.5 h and then quenched with saturated Na₂S₂O₃ solution (100 mL). After warmed to room temperature, the biphasic mixture was poured into water (100 mL) and the two layers were separated. The aqueous layer was extracted with CH₂Cl₂ (2 x 150 mL) and the combined organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (15% Et₂O in hexanes) to give the epoxide **1.59**

(6.312 g, 97.2%) as a colorless liquid: ¹H NMR (300 MHz, CDCl₃) δ 3.75 (d, J = 5.3 Hz, 1H),
3.74 (d, J = 5.4 Hz, 1H), 2.91 (t, J = 5.4 Hz, 1H), 1.34 (s, 3H), 1.29 (s, 3H), 0.92 (s, 9H), 0.10 (s, 3H), 0.09 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 64.2, 62.5, 58.3, 26.1, 24.9, 19.0, 18.5, -5.0, 5.2; IR (neat) 2958, 2930, 2886, 2858, 1472, 1379, 1256, 1140, 1086, 838, 778 cm⁻¹.

1-(tert-Butyldimethylsilanyloxy)-3-methylbut-3-en-2-ol (A9)

A solution of 2,2,6,6-tetramethylpiperidine (2.825 g, 20.0 mmol) in anhydrous OTBS benzene (12.0 mL) at 0 °C was treated dropwise with n-BuLi (1.6 M in ÔH hexanes, 12.5 mL, 20.0 mmol). After 10 min, diethylaluminum chloride (1.0 M in heptanes, 20.0 mL, 20.0 mmol) was added dropwise and the resulting white suspension was stirred at 0 °C for 30 min. The epoxide 1.59 (1.731 g, 8.00 mmol, dissolved in 8.0 mL of anhydrous benzene) was added dropwise and the flask formerly containing the epoxide was rinsed twice with benzene (2 x 4.0 mL). The reaction mixture was stirred further for 1.5 h at 0 °C and then quenched with saturated sodium tartrate solution (50 mL). The biphasic mixture was poured onto water (100 mL) and extracted with $E_{2}O$ (3 x 50 mL). The organic extracts were dried over MgSO₄, filtered and concentrated in vacuo. The resulting residue was purified by column chromatography (15% Et₂O in hexanes) to give allylic alcohol A9 (1.5581 g, 90.0%) as a colorless liquid: ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3) \delta 5.04 \text{ (m, 1H)}, 4.92 \text{ (m, 1H)}, 4.12 \text{ (m, 1H)}, 3.71 \text{ (dd, } J = 9.9, 3.6 \text{ Hz}, 1\text{H)},$ 3.48 (dd, J = 9.9, 8.0 Hz, 1H), 2.66 (d, J = 3.0 Hz, 1H), 1.75 (s, 3H), 0.92 (s, 9H), 0.09 (s, 6H);¹³C NMR (75 MHz, CDCl₃) δ 144.0, 112.0, 75.4, 66.4, 26.0, 19.0, 18.4, -5.2; IR (neat) 3446, 2955, 2929, 2858, 1472, 1256, 1113, 899, 836, 777 cm⁻¹; HRMS (EI): *m/z* calcd for C₇H₁₅O₂Si $(M-C_4H_9)^{+}$ 159.0841, found 159.0811.

Ethyl (E)-6-(tert-butyldimethylsilanyloxy)-4-methylhex-4-enoate (A10)

 EtO_{-} (46.4 mg, 0.626 mmol) in a round-bottom flask was equipped with a fractional distillation apparatus to allow for removal of ethanol. The mixture was heated to 145 °C for 4 h. (During this period of time, a lot of volatiles were distilled out.) The unreacted triethyl orthoacetate was removed by simple distillation and the residue was purified by column chromatography (5% EtOAc in hexanes) to give ethyl ester **A10** (3.433 g, 95.7%) as a colorless liquid: ¹H NMR (300 MHz, CDCl₃) δ 5.32 (qt, *J* = 6.3, 1.2 Hz, 1H), 4.20 (d, *J* = 6.3 Hz, 2H), 4.13 (q, *J* = 7.1 Hz, 2H), 2.46-2.40 (m, 2H), 2.37-2.30 (m, 2H), 1.64 (s, 3H), 1.26 (t, *J* = 7.1 Hz, 3H), 0.90 (s, 9H), 0.07 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 173.5, 135.3, 125.3, 60.5, 60.4, 34.6, 33.0, 26.2, 18.6, 16.6, 14.5, -4.9; IR (neat) 2956, 2930, 2857, 1739, 1472, 1255, 1158, 1110, 1068, 836, 776 cm⁻¹; HRMS (EI): *m/z* calcd for C₁₅H₂₉O₃Si (M-H)⁺⁺ 285.1886, found 285.1840.

(E)-7-(tert-Butyldimethylsilanyloxy)-5-methyl-1,1-diphenylhept-5-en-2-ol (A11)

Ph₂CH CTBS Ethyl ester A10 (1.000 g, 3.49 mmol) in CH₂Cl₂ (10.0 mL) at -78 °C was treated dropwise with DIBAL-H (1.0 M in hexanes, 3.7

mL, 3.7 mmol). The reaction mixture was stirred at -78 °C for 1 h and then quenched with saturated sodium tartrate solution (15 mL). After warmed up to room temperature, the mixture was stirred vigorously for 30 min and extracted with CH_2Cl_2 (3 x 30 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The crude aldehyde **1.60** was used in the next step without further purification.

In a separate two-necked round-bottom flask, a solution of diphenylmethane (1.76 g, 10.5 mmol) in THF (10.0 ml) was treated dropwise with n-BuLi (1.6 M in hexanes, 6.5 mL, 10.5 mmol). The resulting deep orange solution was refluxed for 1 h, and then cooled to 0 °C. The as-prepared crude aldehyde (dissolved in 2.0 mL of THF) was added dropwise and the flask formerly containing the aldehyde was rinsed with THF (2 x 0.5 mL). The reaction mixture was stirred at 0 °C for 1 h, then warmed to room temperature and quenched by slow addition of saturated NaHCO₃ solution (10 mL). The mixture was poured onto water (20 mL) and extracted with Et₂O (3 x 40 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (3% - 12% EtOAc in hexanes) to give the secondary alcohol A11 (0.997 g, 69.6%, two steps) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.42-7.17 (m, 10H), 5.32 (qt, J = 6.4, 1.2 Hz, 1H), 4.40-4.29 (m, 1H), 4.18 (d, J = 6.2 Hz, 2H), 3.92 (d, J = 8.4 Hz, 1H), 2.30-2.20 (m, 1H), 2.17-2.07 (m, 1H), 1.66 (d, J = 3.4 Hz, 1H), 1.70-1.44(m, 2H), 1.56 (s, 3H), 0.92 (s, 9H), 0.08 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 142.6, 141.7, 136.9, 129.0, 128.8, 128.4, 127.0, 126.7, 125.0, 73.5, 60.4, 59.0, 35.9, 33.1, 26.2, 18.6, 16.4, -4.8; IR (neat) 3458, 2954, 2928, 2856, 1599, 1494, 1451, 1386, 1254, 1112, 1067, 835, 776, 702 cm⁻¹; HRMS (ESI): m/z calcd for C₂₆H₃₈O₂SiK [M+K]⁺ 449.2278, found 449.2287.

(E)-6-Methoxy-3-methyl-7,7-diphenylhept-2-en-1-ol (A12)

The secondary alcohol A11 (0.908 g, 2.21 mmol) in anhydrous DMF (10.0 mL) at 0 °C was treated with NaH (60% dispersion in mineral oil, 0.221 g, 5.52 mmol) and the suspension was stirred at 0 °C for 30 min. MeI (0.55 mL, 8.84 mmol) was added dropwise and the reaction mixture was allowed to warm up to room temperature. After stirred for 3 h, the reaction was quenched with water (30 mL) cautiously and extracted with Et₂O (3 x 50 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The resulting residue was dissolved in THF (11.0 mL) and TBAF monohydrate (0.693 g, 5.730 mmol) was added in one portion. The yellow solution was stirred for 1.5 h and then concentrated. The residue was purified by column chromatography (30% - 40% EtOAc in hexanes) to give allylic alcohol **A12** (0.684 g, 99.7%, two steps) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.40-7.08 (m, 10H), 5.37 (qt, *J* = 6.9, 1.2 Hz, 1H), 4.12 (d, *J* = 6.8 Hz, 2H), 4.02 (d, *J* = 8.4 Hz, 1H), 3.92 (ddd, *J* = 8.2, 6.4, 4.1 Hz, 1H), 3.17 (s, 3H), 2.22-2.03 (m, 2H), 1.73-1.48 (m, 2H), 1.58 (s, 3H), 1.30 (br s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 142.9, 142.5, 139.7, 129.0, 128.7, 128.4, 126.6, 126.4, 123.8, 83.4, 59.5, 58.1, 56.4, 35.1, 30.5, 16.4; IR (neat) 3396, 3026, 2929, 1599, 1494, 1451, 1374, 1241, 1102, 1002, 756, 703 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₁H₂₆O₂Na [M+Na]⁺ 333.1831, found 333.1817.

(3-(3-Methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)methanol (A13)

 4H), 1.19/1.17 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 142.7, 142.4, 142.3, 128.9, 128.8, 128.6, 128.4, 126.7, 126.5, 83.4, 83.3, 63.0, 62.7, 61.5, 61.5, 61.4, 61.4, 58.1, 57.9, 56.4, 56.2, 33.9, 33.7, 27.5, 27.2, 17.0, 16.7; IR (neat) 3418, 2931, 1599, 1495, 1452, 1385, 1099, 1032, 747, 704 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₁H₂₆O₃Na [M+Na]⁺ 349.1780, found 349.1766.

tert-Butyl (3-(3-methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)methyl carbonate (1.49)

Epoxy alcohol A13 (192 mg, 0.588 mmol) in anhvdrous Ph₂CH、 toluene (5.8 mL) at 0 °C was treated with 1-methylimidazole ÓMe (46.9 µL, 0.588 mmol) followed by Boc₂O (321 mg, 1.47 mmol). The reaction mixture was stirred at 0 °C for 4 h, then diluted with CH₂Cl₂ (20 mL) and poured onto water (10 mL). The biphasic mixture was separated and the aqueous layer was washed with CH₂Cl₂ (3 x 15 mL). The combined organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (10% - 12.5% EtOAc in hexanes containing 0.5% Et₃N) to give the *t*-butyl carbonate **1.49** (216 mg, 86.3%, dr \sim 1:1) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.39-7.17 (m, 10H), 4.20-4.14 (m, 1H), 4.09 (dd, J = 11.8, 6.1 Hz, 1H), 3.99-3.91 (m, 2H), 3.15/3.14 (s, 3H), 2.97-2.91 (m, 1H), 1.83-1.42 (m, 4H), 1.50 (s, 9H), 1.20/1.18 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 153.5, 142.6, 142.4, 142.3, 128.9, 128.9, 128.7, 128.6, 128.4, 126.7, 126.5, 83.2, 83.0, 82.7, 65.7, 65.6, 60.8, 60.6, 59.6, 59.2, 58.1, 57.8, 56.4, 56.2, 33.5, 33.3, 27.9, 27.4, 27.1, 17.1, 16.7; IR (neat) 2980, 2933, 1743, 1495, 1453, 1370, 1327, 1279, 1163, 1098, 859, 738, 704 cm⁻¹; HRMS (ESI): m/z calcd for C₂₆H₃₄O₅Na [M+Na]⁺ 449.2304, found 449.2278.

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(E)-6-(tert-Butyldimethylsilanyloxy)-4-methylhex-4-enal (1.60)

Ethyl ester A10 (5.84 g, 20.4 mmol) in CH₂Cl₂ (58.0 mL) at -78 °C was treated dropwise with DIBAL-H (1.0 M in hexanes, 21.4 mL, 21.4 mmol). The reaction mixture was stirred at -78 °C for 2 h and then quenched with saturated sodium tartrate solution (120 mL). After warmed up to room temperature, the mixture was extracted with CH₂Cl₂ (3 x 70 mL) and the organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (6% - 8% EtOAc in hexanes) to give the aldehyde **1.60** (4.31 g, 87.4%) as a colorless liquid: ¹H NMR (300 MHz, CDCl₃) δ 9.79 (t, *J* = 1.7 Hz, 1H), 5.33 (qt, *J* = 6.2, 1.3 Hz, 1H), 4.19 (d, *J* = 6.2 Hz, 2H), 2.59-2.54 (m, 2H), 2.35 (app t, *J* = 7.7 Hz, 2H), 1.65 (s, 3H), 0.91 (s, 9H), 0.07 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 202.4, 135.0, 125.6, 60.3, 42.1, 31.7, 26.2, 18.6, 16.7, -4.9; IR (neat) 2955, 2929, 2857, 2714, 1728, 1472, 1255, 1114, 1074, 836, 776 cm⁻¹; HRMS (EI): *m/z* calcd for C₁₃H₂₅O₂Si (M-H)⁺⁺ 241.1624, found 241.1606.

(E)-7-(tert-Butyldimethylsilanyloxy)-5-methylhept-5-enal (A14)

H OTBS (Methoxymethyl)triphenylphosphonium chloride (2.587 g, 7.55 mmol) in a dry flask under high vacuum was heated with heat gun for 3 min to remove the moisture. The flask was cooled to 0 °C and THF (8.0 mL) was added in. NaHMDS (7.55 mL, 7.55 mmol) was added dropwise and the resulting deep orange suspension was stirred at 0 °C for 1 h. Aldehyde **1.60** (0.610 g, 2.52 mmol, dissolved in 1.0 mL of THF) was added dropwise and the flask formerly containing the aldehyde was rinsed twice with THF (2 x 0.5 mL). The reaction mixture was stirred at 0 °C for 1 h, then quenched with saturated NaHCO₃ (10 mL) and extracted with Et₂O (3 x 30 mL). The organic extracts were dried over MgSO₄,

filtered and concentrated. The residue was purified by flash chromatography (3.5% EtOAc in hexanes) to give the crude methyl vinyl ether.

The methyl vinyl ether in THF-H₂O (10:1, 40 mL) was treated with Hg(OAc)₂ (1.277 g, 4.01 mmol). The reaction mixture was stirred for 20 min and saturated KI (20 mL) was added. The resulting yellowish green mixture was stirred for 1 h and then diluted with Et₂O (50 mL). The two layers were separated and the organic layer was washed with saturated KI (40 mL), dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (4% - 6% EtOAc in hexanes) to give the title aldehyde **A14** (0.526 g, 81.6%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 9.78 (t, *J* = 1.5 Hz, 1H), 5.32 (qt, *J* = 6.2, 1.0 Hz, 1H), 4.19 (d, *J* = 6.3 Hz, 2H), 2.42 (dt, *J* = 7.3, 1.5 Hz, 2H), 2.04 (t, *J* = 7.3 Hz, 2H), 1.76 (pent, *J* = 7.6 Hz, 2H), 1.62 (s, 3H), 0.91 (s, 9H), 0.07 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 202.6, 136.0, 125.9, 60.5, 43.5, 39.0, 26.3, 20.3, 18.7, 16.4, -4.8; IR (neat) 2930, 2856, 2713, 1728, 1472, 1387, 1255, 1115, 1080, 836, 776 cm⁻¹; HRMS (ESI): *m/z* calcd for C₁₄H₂₈O₂SiNa [M+Na]⁺ 279.1756, found 279.1745.

(E)-8-(tert-Butyldimethylsilanyloxy)-6-methyl-1,1-diphenyl-oct-6-en-2-ol (A15)

Ph₂CH OTBS A solution of diphenylmethane (0.80 mL, 4.77 mmol) in THF (4.5 mL) was treated with *n*-BuLi (1.6 M in hexanes, 2.74 mL,

4.39 mmol) and the resulting deep-orange solution was refluxed for 2 h. After cooling to room temperature, the solution was cooled further to 0 °C and the aldehyde **A14** (0.489 g, 1.91 mmol, dissolved in 1.0 mL of THF) was added dropwise. The flask formerly containing the aldehyde was rinsed with THF (2 x 0.5 mL). The deep-orange solution was stirred at 0 °C for 1 h, then quenched by slow addition of saturated NaHCO₃ (10 mL). The biphasic mixture was diluted with

Et₂O (10 mL) and poured onto water (10 mL). The two layers were separated and the aqueous layer was extracted with Et₂O (3 x 20 mL). The combined organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (7% - 13% EtOAc in hexanes) to give the alcohol **A15** (0.638 g, 78.7%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.41-7.16 (m, 10H), 5.26 (qt, *J* = 6.4, 1.2 Hz, 1H), 4.36 (dt, *J* = 8.3, 3.0 Hz, 1H), 4.16 (d, *J* = 6.3 Hz, 2H), 3.88 (d, *J* = 8.3 Hz, 1H), 1.99-1.92 (m, 2H), 1.70-1.35 (m, 4H), 1.57 (s, 3H), 0.91 (s, 9H), 0.07 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 142.6, 141.6, 137.1, 129.0, 128.8, 128.4, 127.1, 126.7, 124.7, 73.8, 60.5, 59.0, 39.6, 34.8, 26.2, 24.0, 18.6, 16.4, -4.8; IR (neat) 3458, 2928, 2856, 1599, 1494, 1386, 1254, 1082, 835, 702 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₇H₄₀O₂SiNa [M+Na]⁺ 447.2695, found 447.2741.

(E)-7-Methoxy-3-methyl-8,8-diphenyloct-2-en-1-ol (A16)

mineral oil, 150 mg, 3.75 mmol) and the yellow suspension was stirred at 0 °C for 30 min. MeI (0.37 mL, 6.00 mmol) was added dropwise and the reaction mixture was stirred overnight at room temperature. The flask was cooled to 0 °C and the reaction was quenched with ice chips. The mixture was poured into water (20 mL) and extracted with Et₂O (3 x 30 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was dissolved in THF (7.5 mL) and TBAF monohydrate (0.471 g, 1.80 mmol) was added in. The yellow solution was stirred for 1.5 h and then concentrated in vacuo. The residue was purified by flash chromatography (25% - 35% EtOAc in hexanes) to give the allylic alcohol **A16** (0.475 g, 97.4%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.42-7.19 (m, 10H), 5.36 (qt, *J* = 6.9, 1.0 Hz,

1H), 4.11 (d, J = 6.9 Hz, 2H), 4.04 (d, J = 8.4 Hz, 1H), 3.97-3.92 (m, 1H), 3.19 (s, 3H), 1.98-1.96 (m, 2H), 1.62 (s, 3H), 1.58-1.41 (m, 4H); ¹³C NMR (75 MHz, CDCl₃) δ 142.9, 142.5, 139.4, 128.9, 128.6, 128.4, 126.5, 126.4, 123.7, 83.6, 59.3, 58.0, 56.2, 39.5, 31.6, 22.9, 16.2; IR (neat) 3386, 2934, 1667, 1599, 1495, 1451, 1380, 1186, 1100, 1002, 746, 703 cm⁻¹; HRMS (ESI): m/zcalcd for C₂₂H₂₈O₂Na [M+Na]⁺ 347.1987, found 347.1966.

(3-(4-Methoxy-5,5-diphenylpentyl)-3-methyloxiran-2-yl)methanol (A17)

Ph₂CH OH A solution of allylic alcohol A16 (85.0 mg, 0.262 mmol) in OMe CH₂Cl₂ (2.6 mL) at 0 °C was treated with NaHCO₃ powder (55.0

mg, 0.655 mmol) followed by *m*-chloroperbenzoic acid (pure, 47.5 mg, 0.275 mmol). The reaction mixture was stirred at 0 °C for 50 min and then quenched with saturated Na₂S₂O₃ solution (2.0 mL). After warmed to room temperature, the biphasic mixture was poured into water (5 mL) and the two layers were separated. The aqueous layer was extracted with CH₂Cl₂ (5 x 10 mL) and the combined organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (40% - 45% EtOAc in hexanes containing 0.5% Et₃N) to give the epoxy alcohol **A17** (88.7 mg, 99.4%, dr ~ 1:1) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.38-7.17 (m, 10H), 4.00 (d, *J* = 8.4 Hz, 1H), 3.91-3.89 (m, 1H), 3.80-3.74 (m, 1H), 3.65 (dd, *J* = 12.1, 6.6 Hz, 1H), 3.17/3.16 (s, 3H), 2.92-2.87 (m, 1H), 1.64-1.34 (m, 6H), 1.23 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 142.8, 142.4, 128.9, 128.9, 128.7, 128.6, 128.4, 126.6, 126.4, 83.6, 83.5, 63.1, 63.0, 61.5, 61.4, 58.1, 58.0, 56.3, 38.6, 32.1, 32.0, 20.7, 20.6, 16.8, 16.7; IR (neat) 3420, 2935, 1599, 1495, 1452, 1385, 1249, 1100, 1031, 862, 747, 704 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₂H₂₈O₃Na [M+Na]⁺ 363.1936, found 363.1947.

tert-Butyl (3-(4-methoxy-5,5-diphenylpentyl)-3-methyloxiran-2-yl)methyl carbonate (1.50)

Epoxy alcohol A17 (72.9 mg, 0.214 mmol) in anhydrous ∠O^tBu Ph₂CH toluene (2.0 mL) at 0 °C was treated with 1-OMe methylimidazole (22 µL, 0.278 mmol) followed by Boc₂O (187 mg, 0.856 mmol, dissolved in 0.5 mL of toluene). The reaction mixture was stirred at 0 °C for 4 h, then diluted with CH₂Cl₂ (5.0 mL) and poured into water (6 mL). The biphasic mixture was separated and the aqueous layer was washed with CH₂Cl₂ (3 x 5 mL). The combined organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (10% -15% EtOAc in hexanes containing 0.5% Et₃N) to give the *t*-butyl carbonate **1.50** (83.9 mg, 89.0%, dr ~ 1:1) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.39-7.17 (m, 10H), 4.18 (dd, J = 11.9, 4.8 Hz, 1H), 4.09 (dd, J = 11.9, 6.3 Hz, 1H), 4.00 (d, J = 8.4 Hz, 1H), 3.93-3.87 (m, 1H), 3.17 (br s, 3H), 2.98-2.94 (m, 1H), 1.64-1.41 (m, 6H), 1.52 (s, 9H), 1.24 (br s, 3H); ¹³C NMR (75 MHz, CDCl₃) & 153.4, 142.8, 142.4, 128.9, 128.6, 128.5, 128.4, 126.5, 126.4, 83.5, 82.6, 65.7, 60.6, 59.5, 59.4, 58.1, 58.0, 56.2, 38.3, 32.0, 27.9, 20.5, 20.5, 16.8, 16.8; IR (neat) 2934, 1742, 1495, 1452, 1369, 1279, 1255, 1163, 1098, 859, 704 cm⁻¹; HRMS (ESI): m/z calcd for C₂₇H₃₆O₅Na [M+Na]⁺ 463.2460, found 463.2462.

((E)-3-Methyl-5-(3,3-dimethyloxiran-2-yl)pent-2-enyloxy)(tert-butyl)dimethylsilane (1.61)

((E)-3,7-dimethylocta-2,6-dienyloxy)(tert-butyl)dimethylsilane ((E)-3,7-dimethylocta-2,6-dienyloxy)(tert-butyl)dimethylsilane (5.370 g, 20.0 mmol) in CHCl₃ (180 mL) at 0 °C was treated with *m*CPBA (70-75%, 5.621 g, 22.8 mmol) in small portions. The white suspension was stirred at 0 °C for 30 min, and then quenched with saturated Na₂S₂O₃ solution (20 mL) and saturated NaHCO₃ solution (100 mL). The mixture was warmed up to room temperature and the two layers were separated. The aqueous was washed with CH_2Cl_2 (2 x 100 mL) and the combination of the organic extracts were dried over MgSO₄ and evaporated. The residue was purified by column chromatography (8%-12% Et₂O in hexanes) to give the desired monoepoxide **1.61** (4.375 g, 76.9%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 5.36 (qt, *J* = 6.3, 1.2 Hz, 1H), 4.20 (d, *J* = 6.3, Hz, 1H), 4.20 (d, *J* = 6.3, Hz, 1H), 2.72 (t, *J* = 6.2 Hz, 1H), 2.25-2.06 (m, 2H), 1.65 (s, 3H), 1.71-1.61 (m, 2H), 1.31 (s, 3H), 1.27 (s, 3H), 0.91 (s, 9H), 0.08 (s, 6H).

(E)-8-(tert-Butyldimethylsilanyloxy)-2,6-dimethylocta-1,6-dien-3-ol (A18)

A solution of 2,2,6,6-tetramethylpiperidine (5.297 g, 37.5 mmol) in anhydrous benzene (25.0 mL) at 0 °C was treated dropwise with n-OTBS ÓН BuLi (1.6 M in hexanes, 23.4 mL, 37.5 mmol). After 10 min, diethylaluminum chloride (1.0 M in heptanes, 37.5 mL, 37.5 mmol) and the resulting white suspension was stirred at 0 °C for 30 min. Monoepoxide 1.61 (4.268 g, 15.0 mmol, dissolved in 5.0 mL of anhydrous benzene) was added dropwise and the flask formerly containing the epoxide was rinsed twice with benzene (2 x 2.5 mL). The reaction mixture was stirred further at 0 °C for 1.5 h and then quenched with saturated sodium tartrate solution (100 mL). The biphasic mixture was poured into water (100 mL) and extracted with Et_2O (3 x 150 mL). The organic extracts were dried over MgSO₄, filtered and concentrated in vacuo. The resulting residue was purified by column chromatography (12%-21% Et₂O in hexanes) to give the allylic alcohol A18 (3.924 g, 91.9%) as a colorless liquid: ¹H NMR (300 MHz, CDCl₃) δ 5.34 (qt, J = 6.3, 1.2 Hz, 1H), 4.94 (t, J = 0.8 Hz, 1H), 4.84 (t, J = 1.5Hz, 1H), 4.19 (d, J = 6.3 Hz, 2H), 4.05 (t, J = 6.2 Hz, 1H), 2.17-1.95 (m, 2H), 1.73 (s, 3H), 1.64 (s, 3H), 1.70-1.60 (m, 2H), 0.90 (s, 9H), 0.07 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 147.6, 136.8, 124.9, 111.3, 75.8, 60.4, 35.6, 33.0, 26.2, 18.6, 17.8, 16.6, -4.8; IR (neat) 3382, 2929, 2857, 1472, 1382, 1255, 1112, 1070, 836, 776 cm⁻¹; HRMS (EI): m/z calcd for C₁₂H₂₃O₂Si (M-C₄H₉)^{+•} 227.1467, found 227.1450.

(4E,8E)-Ethyl 10-(tert-butyldimethylsilanyloxy)-4,8-dimethyl-deca-4,8-dienoate (A19)

A mixture of allylic alcohol A18 (3.671 g, 12.90 mmol), EtO OTBS triethyl orthoacetate (freshly distilled, 10.46 g, 64.50 mmol) and propionic acid (47.8 mg, 0.645 mmol) in a round-bottom flask was equipped with a fractional distillation apparatus to allow for removal of ethanol. The mixture was heated to 145 °C for 1.5 h. (During this period of time, a lot of volatiles were distilled out.) The unreacted triethyl orthoacetate was removed by simple distillation and the residue was purified by column chromatography (3% EtOAc in hexanes) to give the ethyl ester A19 (4.281 g, 93.6%) as a colorless liquid: ¹H NMR (300 MHz, CDCl₃) δ 5.30 (qt, J = 6.3, 1.2 Hz, 1H), 5.14 (qt, J = 6.9, 1.2 Hz, 1H), 4.19 (d, J = 6.3 Hz, 1H), 4.19 (d, J = 6.3 Hz, 1H), 4.12 (q, J = 7.1 Hz, 2H), 2.42-2.35 (m, 2H), 2.31-2.26 (m, 2H), 2.14-2.06 (m, 2H), 2.02-1.97 (m, 2H), 1.62 (s, 3H), 1.61 (s, 3H), 1.25 (t, J = 7.1 Hz, 3H), 0.90 (s, 9H), 0.07 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 173.7, 136.9, 133.7, 125.0, 124.7, 60.5, 60.4, 39.6, 34.9, 33.5, 26.4, 26.2, 18.6, 16.5, 16.1, 14.5, -4.8; IR (neat) 2929, 2856, 1738, 1463, 1254, 1158, 1063, 836, 776 cm⁻¹; HRMS (ESI): *m/z* calcd for $C_{20}H_{38}O_3SiNa [M+Na]^+ 377.2488$, found 377.2513.

(4*E*,8*E*)-10-(*tert*-Butyldimethylsilanyloxy)-4,8-dimethyldeca-4,8-dienal (1.62)

Ethyl ester A19 (2.3345g, 6.58 mmol) in CH₂Cl₂ (20.0 mL) at -78 °C was treated dropwise with DIBAL-H (1.0 M in hexanes, 6.91 mL, 6.91 mmol). The reaction mixture was stirred at -78 °C for 40 min and DIBAL-H (1.0 M in hexanes, 0.66 mL, 0.66 mmol) were added. The mixture was stirred for 30 min more and then quenched with saturated sodium tartrate solution (30 mL). After warmed up to room temperature, the mixture was extracted with CH₂Cl₂ (3 x 40 mL) and the organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (4% - 20% EtOAc in hexanes) to give the aldehyde **1.62** (1.884 g, 92.1%) as a colorless liquid: ¹H NMR (300 MHz, CDCl₃) δ 9.75 (t, *J* = 1.9 Hz, 1H), 5.30 (qt, *J* = 6.3, 1.1 Hz, 1H), 5.15 (qt, *J* = 6.8, 1.0 Hz, 1H), 4.19 (d, *J* = 6.3 Hz, 2H), 2.52 (dt, *J* = 7.9, 1.7 Hz, 2H), 2.32 (t, *J* = 7.5 Hz, 2H), 2.15-2.08 (m, 2H), 2.03-1.98 (m, 2H), 1.62 (s, 6H), 0.91 (s, 9H), 0.08 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 202.8, 136.8, 133.4, 125.3, 124.8, 60.5, 42.4, 39.5, 32.0, 26.4, 26.2, 18.6, 16.5, 16.3, -4.8; IR (neat) 2928, 2856, 1728, 1472, 1386, 1254, 1110, 1066, 836, 776 cm⁻¹; HRMS (ESI): *m/z* calcd for C₁₈H₃₅O₂Si [M+H]⁺ 311.2406, found 311.2386.

(5*E*,9*E*)-11-(*tert*-Butyldimethylsilanyloxy)-5,9-dimethyl-1,1-diphenylundeca-5,9-dien-2-ol (A20)



in hexanes, 6.04 mL, 9.66 mmol). The resulting deep orange solution was stirred at 75 °C for 1 h and cooled to 0 °C. Dienal **1.62** (1.000 g, 3.22 mmol, dissolved in 2.0 mL of THF) was added dropwise and the flask formerly containing the aldehyde was rinsed with THF (2 x 0.5 mL). The reaction mixture was stirred at 0 °C for 1 h and quenched by slow addition of saturated NaHCO₃ solution (10 mL). The mixture was poured onto water (15 mL) and extracted with Et₂O (3 x 30 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (4% - 10% EtOAc in hexanes) to give the secondary alcohol

A20 (1.243 g, 80.6%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.41-7.16 (m, 10H), 5.30 (qt, *J* = 6.3, 1.1 Hz, 1H), 5.14 (app t, *J* = 6.2 Hz, 1H), 4.38-4.30 (m, 1H), 4.18 (d, *J* = 6.3 Hz, 2H), 3.91 (d, *J* = 8.4 Hz, 1H), 2.20-2.00 (m, 6H), 1.62 (s, 3H), 1.52 (s, 3H), 1.67-1.42 (m, 2H), 0.92 (s, 9H), 0.08 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 142.7, 141.8, 137.1, 135.2, 129.0, 129.0, 128.8, 128.5, 127.0, 126.7, 124. 8, 124.6, 73.6, 60.5, 58.9, 39.7, 36.1, 33.3, 26.5, 26.2, 18.6, 16.6, 16.1, -4.8; IR (neat) 3466, 2928, 2855, 1598, 1494, 1450, 1384, 1254, 1067, 835, 776, 702 cm⁻¹; HRMS (ESI): *m/z* calcd for C₃₁H₄₆O₂SiNa [M+Na]⁺ 501.3165, found 501.3150.

(2E,6E)-10-Methoxy-3,7-dimethyl-11,11-diphenylundeca-2,6-dien-1-ol (1.63)



The secondary alcohol A20 (1.200 g, 2.507 mmol) in anhydrous DMF (14.0 mL) at 0 $^{\circ}$ C was treated with NaH

(60% dispersion in mineral oil, 0.251 g, 6.268 mmol) and the suspension was stirred at 0 °C for 30 min. MeI (0.62 mL, 10.0 mmol) was added dropwise and the reaction mixture was allowed to warm up to room temperature. After stirred for 3 h, the reaction was quenched with water (25 mL) cautiously and extracted with Et₂O (3 x 35 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The resulting residue was dissolved in THF (12.5 mL) and TBAF monohydrate (0.787 g, 3.01 mmol) was added in one portion. The yellow solution was stirred for 1.3 h and then concentrated. The residue was purified by flash chromatography (20% - 30% EtOAc in hexanes) to give the allylic alcohol **1.63** (0.923 g, 97.2%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.40-7.16 (m, 10H), 5.40 (qt, *J* = 6.9, 1.1 Hz, 1H), 5.10 (t, *J* = 5.5 Hz, 1H), 4.14 (d, *J* = 6.7 Hz, 2H), 4.02 (d, *J* = 8.3 Hz, 1H), 3.92-3.87 (m, 1H), 3.16 (s, 3H), 2.11-2.01 (m, 6H), 1.68 (s, 3H), 1.51 (s, 3H), 1.62-1.46 (m, 2H), 0.92 (br s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 143.0, 142.5, 139.8, 135.3, 129.1, 128.7, 128.6, 128.4, 126.5, 126.4, 124.3, 123.6, 83.3,

59.6, 58.0, 56.2, 39.7, 35.3, 30.8, 26.4, 16.5, 16.1; IR (neat) 3388, 3026, 2925, 1599, 1494, 1451, 1382, 1189, 1102, 1002, 755, 703 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₆H₃₄O₂Na [M+Na]⁺ 401.2457, found 401.2477.

((2*R*,3*R*)-3-(2-((2*R*,3*R*)-3-(3-Methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)ethyl)-3methyloxiran-2-yl)methanol (A21)

To a solution of dienol 1.63 (100 mg, 0.264 mmol) in Ph₂CH CH₃CN/DMM (8.0 mL, 1:2, v/v) were added a 0.05 M OMe solution of Na₂B₄O₇ in 4x10⁻⁴ M Na₂(EDTA) (5.2 mL), Bu₄NHSO₄ (7.2 mg, 21.1 µmol) and Shi ketone (68.2 mg, 0.264 mmol) sequentially. The mixture was cooled to 0 °C, and the Oxone (448 mg, 0.729 mmol), dissolved in 4x10⁻⁴ M Na₂(EDTA) (3.4 mL), and K₂CO₃ (424 mg, 3.06 mmol), dissolved in water (3.4 mL), were added simultaneously via a syringe pump over 2.0 h. After the addition was completed, the slightly blue reaction mixture was stirred further for 15 min at 0 °C, then diluted with water (10 mL) and extracted with CH₂Cl₂ (3 x 30 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (40% - 80% EtOAc in hexanes) to give the diepoxy alcohol A21 (95.8 mg, 88.4%, dr ~ 1:1 regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.42-7.17 (m, 10H), 4.00-3.85 (m, 2H), 3.81-3.61 (m, 2H), 3.16/3.14 (s, 3H), 2.96 (t, J = 6.0 Hz, 1H), 2.64-2.60 (m, 1H), 2.01(br s, 1H), 1.86-1.73 (m, 2H), 1.66-1.42 (m, 6H), 1.31/1.30 (s, 3H), 1.15/1.13 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 142.7, 142.4, 142.2, 129.0, 128.9, 128.7, 128.7, 128.6, 128.4, 126.7, 126.5, 83.3, 63.1, 62.6, 62.5, 61.4, 61.2, 61.1, 61.0, 60.7, 58.2, 57.9, 56.4, 56.1, 35.2, 33.9, 33.8, 27.6, 27.3, 24.4, 17.1, 16.8, 16.4; IR (neat) 3435, 3026, 2929, 1495, 1452, 1386, 1100, 1032, 747, 704

cm⁻¹; HRMS (ESI): m/z calcd for C₂₆H₃₄O₄Na [M+Na]⁺ 433.2355, found 433.2348; [α]_D = +13.3 (CHCl₃, *c* 1.34).

tert-Butyl ((2*R*,3*R*)-3-(2-((2*R*,3*R*)-3-(3-methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)ethyl)-3-methyloxiran-2-yl)methyl carbonate (1.51)

To a solution of diepoxy alcohol A21 (112 mg, O O'Bu 0.273 mmol) in anhydrous toluene (2.7 mL) at 0 Ph₂CH OMe °C were added N-methylimidazole (22 µL, 0.273 mmol) and Boc₂O (119 mg, 0.546 mmol) and the reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. After that time, the reaction was quenched with water (10 mL) and extracted with CH₂Cl₂ (3 x 25 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was azeotroped with hexanes (3 x 10 mL) and then purified by column chromatography (20% - 25% EtOAc in hexanes containing 0.5% Et₃N) to give the *tert*-butyl carbonate 1.51 (129 mg, 92.7%, dr \sim 1:1 regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.39-7.17 (m, 10H), 4.27-4.10 (m, 2H), 4.00-3.88 (m, 2H), 3.17/3.16 (s, 3H), 3.02 (t, J = 5.8 Hz, 1H), 2.63-2.59 (m, 1H), 1.84-1.38 (m, 8H), 1.52 (s, 9H), 1.32 (s, 3H), 1.14/1.13 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 153.5, 142.7, 142.4, 142.3, 129.0, 128.9, 128.8, 128.7, 128.6, 128.4, 126.7, 126.5, 83.3, 82.8, 65.6, 61.1, 61.0, 60.3, 59.2, 58.2, 57.9, 56.4, 34.8, 33.9, 33.8, 27.9, 27.4, 24.3, 17.2, 16.9, 16.8, 16.4; IR (neat) 2978, 2932, 1743, 1495, 1453, 1370, 1279, 1256, 1163, 1098, 858, 756, 704 cm⁻¹; HRMS (ESI): m/z calcd for $C_{31}H_{42}O_6Na [M+Na]^+$ 533.2879, found 533.2859; $[\alpha]_{D} = +17.3$ (CHCl₃, *c* 1.49).

((2*S*,3*S*)-3-((*E*)-7-Methoxy-4-methyl-8,8-diphenyloct-3-enyl)-3-methyloxiran-2-yl)methanol (A22)

A suspension of the activated 4Å molecular sieves powder (95 mg) in CH₂Cl₂ (2.4 mL) was treated with (+)-

diisopropyl tartrate (8.0 μ L, 38.0 μ mol) and the mixture was cooled to -35 to -30 °C. The mixture was stirred for 10 min and Ti(O-*i*-Pr)₄ (9.5 μ L, 31.7 μ mol) was added and the mixture was stirred for 15 min more. After that time, t-butyl hydroperoxide (5.0-6.0 M in decane, 0.19 mL, ~0.951 mmol) was added dropwise and the mixture was stirred for 30 min. Dienol 1.63 (120.0 mg, dissolved in 0.5 mL of CH₂Cl₂) was added dropwise and the flask formerly containing the dienol was rinsed with CH₂Cl₂ (2 x 0.1 mL). The reaction mixture was stirred at -35 to -30 °C for 40 min and then water (0.5 mL) was added. The mixture was allowed to warm up to 0 °C and stirred for 1 h. A solution of 30% of NaOH saturated with NaCl (0.3 mL) was added and the mixture was warmed up to room temperature and stirred for 1.5 h. The suspension was filtered through a pad of Celite and the filtrate was dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (30% - 40% EtOAc in hexanes) to give the monoepoxy alcohol A22 (120.0 mg, 95.9%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.39-7.17 (m, 10H), 5.07 (app t, J = 6.7 Hz, 1H), 4.01 (d, J = 8.3 Hz, 1H), 3.92-3.86 (m, 1H), 3.81-3.75 (m, 1H), 3.71-3.63 (m, 1H), 3.15 (s, 3H), 2.94 (dd, J = 6.5, 4.5 Hz, 1H), 2.11-2.04 (m, 4H), 1.74-1.43 (m, 4H), 1.51 (s, 3H), 1.29 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 142.9, 142.5, 135.8, 135.7, 129.0, 128.6, 128.4, 126.5, 126.4, 123.7, 123.6, 83.3, 83.2, 63.1, 61.6, 61.2, 58.0, 57.8, 56.2, 38.6, 35.2, 30.7, 30.6, 23.8, 16.9, 16.0; IR (neat) 3424, 3026, 2930, 1598, 1494, 1451, 1384, 1103, 1032, 862, 746, 704 cm⁻¹; HRMS (ESI): m/z calcd for C₂₆H₃₄O₃Na [M+Na]⁺ 417.2406, found 417.2436; $[\alpha]_D = -3.8$ (CHCl₃, *c* 1.08).

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((2S,3S)-3-(2-((2R,3R)-3-(3-Methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)ethyl)-3-

methyloxiran-2-yl)methanol (A23)

To a solution of monoepoxy alcohol A22 (170.0 mg, , O Ph₂CH 0.431 mmol) in CH₃CN/DMM (6.5 mL, 1:2, v/v) were ÓMe added a 0.05 M solution of Na₂B₄O₇ in 4x10⁻⁴ M Na₂(EDTA) (4.3 mL), Bu₄NHSO₄ (5.8 mg, 17.2 µmol) and Shi ketone (55.6 mg, 0.216 mmol) sequentially. The mixture was cooled to 0 °C, and the Oxone (366 mg, 0.595 mmol), dissolved in 4x10⁻⁴ M Na₂(EDTA) (2.8 mL), and K₂CO₃ (346 mg, 2.50 mmol), dissolved in water (2.8 mL), were added simultaneously via a syringe pump over 2.0 h. After the addition was completed, the blue reaction mixture was stirred further for 15 min at 0 °C, then diluted with water (15 mL) and extracted with CH₂Cl₂ (3 x 30 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (50% - 80% EtOAc in hexanes) to give the diepoxy alcohol A23 (160.6 mg, 90.8%, dr \sim 1:1 regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups): ¹H NMR (300 MHz, CDCl₃) δ 7.38-7.17 (m, 10H), 3.99-3.95 (m, 1H), 3.93-3.87 (m, 1H), 3.83-3.73 (m, 1H), 3.71-3.60 (m, 1H), 3.15/3.14 (s, 3H), 2.98-2.90 (m, 1H), 2.67-2.60 (m, 1H), 2.35-2.28 (m, 1H), 1.91-1.86 (m, 1H), 1.78-1.71 (m, 2H), 1.69-1.55 (m, 3H), 1.52-1.44 (m, 3H), 1.30 (s, 3H), 1.16/1.14 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 142.6, 142.6, 142.3, 142.2, 128.9, 128.8, 128.7, 128.6, 128.4, 126.6, 126.5, 83.3, 83.1, 63.6, 63.0, 63.0, 61.4, 61.2, 61.0, 60.8, 58.2, 57.8, 56.2, 56.0, 36.1, 34.0, 33.7, 27.6, 27.1, 24.7, 16.9, 16.5, 16.4; IR (neat) 3438, 3026, 2929, 1495, 1452, 1385, 1100, 1032, 746, 704 cm⁻¹; HRMS (ESI): *m/z* calcd for $C_{26}H_{34}O_4Na [M+Na]^+ 433.2355$, found 433.2346; $[\alpha]_D = +19.3$ (CHCl₃, *c* 1.55).

tert-Butyl ((2*S*,3*S*)-3-(2-((2*R*,3*R*)-3-(3-methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)ethyl)-3-methyloxiran-2-yl)methyl carbonate (1.52)

To a solution of diepoxy alcohol A23 (145.1 mg, O^tBu 0.353 mmol) in anhydrous toluene (3.5 mL) at 0 Ph₂CH OMe °C were added 1-methylimidazole (28 µL, 0.353 mmol) and Boc₂O (154 mg, 0.706 mmol). The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. The reaction was guenched with water (15 mL) and extracted with CH₂Cl₂ (3 x 25 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (20% - 25% EtOAc in hexanes containing 0.5% Et₃N) to give the tertbutyl carbonate 1.52 (154.8 mg, 85.8%, dr \sim 1:1 regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.41-7.18 (m, 10H), 4.29-4.21 (m, 1H), 4.19-4.11 (m, 1H), 4.02-3.92 (m, 2H), 3.18/3.17 (s, 3H), 3.04 (dd, J = 6.2, 4.8 Hz, 1H), 2.63-2.59 (m, 1H), 1.82-1.40 (m, 8H), 1.53 (s, 9H), 1.32 (s, 3H), 1.17/1.15 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 153.5, 142.7, 142.4, 142.2, 129.0, 128.7, 128.6, 128.4, 126.6, 126.4, 83.3, 82.7, 65.6, 63.1, 62.7, 60.9, 60.3, 59.8, 58.1, 57.8, 56.3, 56.1, 35.2, 33.9, 33.8, 27.9, 27.6, 27.3, 24.5, 16.9, 16.8, 16.4; IR (neat) 2977, 2932, 1742, 1495, 1453, 1370, 1279, 1256, 1163, 1097, 858, 704 cm⁻¹; HRMS (ESI): m/z calcd for C₃₁H₄₂O₆Na [M+Na]⁺ 533.2879, found 533.2857; $[\alpha]_D = +1.3$ (CHCl₃, *c* 2.15).

(5E,9E)-11-(tert-Butyldimethylsilanyloxy)-5,9-dimethylundeca-5,9-dienal (1.64)

(Methoxymethyl)triphenylphosphonium chloride (1.656 H \rightarrow OTBS g, 4.830 mmol) in a dry flask under high vacuum was heated with heat gun for 3 min to remove the moisture. After the flask was cooled to room temperature, THF (8.0 mL) was added in and the suspension was cooled to 0 °C. NaHMDS (1 M in THF, 4.83 mL, 4.83 mmol) was added dropwise and the resulting deep orange suspension was stirred at 0 °C for 1 h. Dienal **1.62** (0.500 g, 1.61 mmol, dissolved in 2.0 mL of THF) was added dropwise and the flask formerly containing the aldehyde was rinsed with THF (2 x 1.0 mL). The reaction mixture was stirred at 0 °C for 1 h, then quenched with saturated NaHCO₃ (10 mL) and poured into water (20 mL). The mixture was extracted with Et_2O (3 x 30 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (3.5% EtOAc in hexanes) to give the methyl vinyl ether.

The methyl vinyl ether in THF–H₂O (10:1, 16.5 mL) was treated with Hg(OAc)₂ (0.558 g, 1.65 mmol). The reaction mixture was stirred for 30 min and saturated KI (30 mL) was added. The resulting yellowish green mixture was stirred for 1 h, then diluted with Et₂O (30 mL). The two layers were separated and the aqueous layer was extracted with Et₂O (2 x 40 mL). The combination of the organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (3% - 5% EtOAc in hexanes) to give the titled aldehyde (0.474 g, 90.7%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 9.77 (t, *J* = 1.7 Hz, 1H), 5.30 (qt, *J* = 6.3, 1.1 Hz, 1H), 5.12 (qt, *J* = 6.9, 1.1 Hz, 1H), 4.20 (d, *J* = 6.3 Hz, 2H), 2.39 (dt, *J* = 7.3, 1.7 Hz, 2H), 2.17-2.08 (m, 2H), 2.01 (app t, *J* = 7.9 Hz, 4H), 1.74 (pent, *J* = 7.3 Hz, 2H), 1.62 (s, 3H), 1.59 (s, 3H), 0.91 (s, 9H), 0.07 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 202.8, 136.9, 134.2, 125.5, 124.8, 60.5, 43.4, 39.6, 39.0, 26.4, 26.2, 20.4, 18.6, 16.5, 15.9, -4.8; IR (neat) 2929, 2856, 2712, 1728, 1472, 1386, 1254, 1110, 1068, 836, 776 cm⁻¹; HRMS (ESI): *m/z* calcd for C₁₉H₃₆O₂SiNa [M+Na]⁺ 347.2382, found 347.2402.
(6*E*,10*E*)-12-(*tert*-Butyldimethylsilanyloxy)-6,10-dimethyl-1,1-diphenyldodeca-6,10-dien-2ol (A24)

Ph₂CH OH OH OH OH OTBS A solution of diphenylmethane (0.66 mL, 3.93 mmol) in THF (3.9 mL) was treated with *n*-BuLi

(1.6 M in hexanes, 2.46 mL, 3.93 mmol) and the resulting deep-orange solution was refluxed for 1 h. After cooling to room temperature, the solution was cooled further to 0 °C and the above aldehyde 1.64 (0.426 g, 1.31 mmol, dissolved in 1.0 mL of THF) was added dropwise. The flask formerly containing the aldehyde was rinsed with THF (2 x 0.5 mL). The deep-orange solution was stirred at 0 °C for 2 h, then guenched with saturated NaHCO₃ (5 mL). The biphasic mixture was poured into water (25 mL) and extracted with Et₂O (3 x 25 mL). The combined organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (3% - 9% EtOAc in hexanes) to give the secondary alcohol A24 (0.519 g, 80.3%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.38-7.13 (m, 10H), 5.28 (qt, J = 6.3, 1.1 Hz, 1H), 5.03 (qt, J = 6.9, 1.0 Hz, 1H), 4.37-4.29 (m, 1H), 4.18 (d, J = 6.3 Hz, 2H), 3.86 (d, J= 8.3 Hz, 1H), 2.08-2.00 (m, 2H), 1.96-1.85 (m, 4H), 1.60 (s, 3H), 1.52 (s, 3H), 1.69-1.30 (m, 4H), 0.89 (s, 9H), 0.06 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 142.8, 141.7, 137.1, 135.2, 129.0, 129.0, 128.8, 128.5, 127.0, 126.7, 124.6, 124.3, 73.9, 60.6, 59.0, 39.7, 39.6, 34.8, 26.5, 26.2, 24.2, 18.6, 16.6, 16.0, -4.8; IR (neat) 3467, 2928, 2856, 1599, 1494, 1451, 1384, 1253, 1107, 1065, 1005, 835, 775, 702 cm⁻¹; HRMS (ESI): m/z calcd for C₃₂H₄₈O₂SiNa [M+Na]⁺ 515.3321, found 515.3317.

(2E,6E)-11-Methoxy-3,7-dimethyl-12,12-diphenyldodeca-2,6-dien-1-ol (1.65)

Ph₂CH OH The secondary alcohol A24 (0.473 g, 0.960 mmol) in OMe

anhydrous DMF (5.0 mL) at 0 °C was treated with NaH (60% dispersion in mineral oil, 96.0 mg, 2.40 mmol) and the yellow suspension was stirred at 0 °C for 30 min. MeI (0.24 mL, 3.84 mmol) was added dropwise and the reaction mixture was stirred for 2.5 h at room temperature. The reaction was quenched with water (10 mL) and extracted with Et₂O (3 x 30 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was dissolved in THF (5.0 mL) and TBAF monohydrate (0.301 g, 1.15 mmol) was added in. The yellow solution was stirred for 2 h and then concentrated in vacuo. The residue was purified by flash chromatography (20% - 30% EtOAc in hexanes) to give the dienol **1.65** (0.368 g, 97.5%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.40-7.16 (m, 10H), 5.40 (qt, J = 6.9, 1.2 Hz, 1H), 5.05 (qt, J = 6.8, 1.0 Hz, 1H), 4.16 (d, J = 6.8 Hz, 2H), 4.02 (d, J = 8.3 Hz, 1H), 3.94-3.88 (m, 1H), 3.18 (s, 3H), 2.13-1.98 (m, 4H), 1.93-1.90 (m, 2H), 1.68 (s, 3H), 1.54 (s, 3H), 1.58-1.30 (m, 4H); ¹³C NMR (75 MHz, CDCl₃) δ 143.1, 142.6, 139.7, 135.4, 129.1, 128.7, 128.6, 128.4, 126.5, 126.4, 124.2, 123.7, 83.8, 59.6, 58.0, 56.3, 39.8, 39.7, 31.7, 26.4, 23.4, 16.4, 16.0; IR (neat) 3388, 3026, 2931, 1667, 1599, 1495, 1451, 1382, 1186, 1102, 1003, 745, 703 cm⁻¹; HRMS (ESI): *m/z* calcd for $C_{27}H_{36}O_2Na [M+Na]^+ 415.2613$, found 415.2607.

((2*R*,3*R*)-3-(2-((2*R*,3*R*)-3-(4-methoxy-5,5-diphenylpentyl)-3-methyloxiran-2-yl)ethyl)-3methyloxiran-2-yl)methanol (A25)

Ph₂CH $(112 \text{ mg}, 0^{+})$ (70 m) (70 m)

(460 mg, 3.32 mmol), dissolved in water (3.7 mL), were added simultaneously via a syringe pump over 1.5 h. After the addition was completed, the blue reaction mixture was stirred further for 15 min at 0 °C, then diluted with water (20 mL) and extracted with CH_2Cl_2 (3 x 30 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (35% - 85% EtOAc in hexanes containing 0.5% Et₃N) to give diepoxy alcohol **A25** (78 mg, 64%, dr ~ 1:1 regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.62-7.16 (m, 10H), 4.00 (*J* = 8.4 Hz, 1H), 3.92-3.83 (m, 1H), 3.82-3.79 (m, 1H), 3.75-3.64 (m, 1H), 3.17/3.16 (s, 3H), 3.01-2.96 (m, 1H), 2.68-2.63 (m, 1H), 1.96-1.70 (m, 2H), 1.64-1.40 (m, 8H), 1.33 (s, 3H), 1.20/1.19 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 142.8, 142.4, 128.9, 128.9, 128.7, 128.6, 128.4, 126.6, 126.4, 83.6, 63.0, 62.9, 62.6, 61.4, 61.2, 61.1, 60.7, 58.1, 58.0, 56.2, 38.8, 38.7, 35.2, 32.1, 24.4, 20.8, 20.7, 17.0, 16.5, 16.5; IR (neat) 3433, 2930, 1599, 1495, 1452, 1386, 1102, 1032, 734, 704 cm⁻¹.

tert-Butyl ((2*R*,3*R*)-3-(2-((2*R*,3*R*)-3-(4-methoxy-5,5-diphenylpentyl)-3-methyloxiran-2-yl)ethyl)-3-methyloxiran-2-yl)methyl carbonate (1.53)

Ph₂CH (1.7 mL) The diepoxy alcohol A25 (74 mg, 0.174 mmol) was dissolved in dry toluene (1.7 mL) and cooled to 0 °C. *N*-Methylimidazole (18 µL, 0.226 mmol) and Boc₂O (152 mg, 0.696 mmol) were added sequentially. The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. The reaction was quenched with water (10 mL) and extracted with CH₂Cl₂ (3 x 20 mL). The extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by column chromatography (20% - 25% EtOAc in hexanes containing 0.5%)

Et₃N) to give the desired product **1.53** (72 mg, 78%, dr ~ 1:1 regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.41-7.18 (m, 10H), 4.30-4.14 (m, 2H), 4.02 (d, *J* = 8.4 Hz, 1H), 3.98-3.85 (m, 1H), 3.19/3.18 (s, 3H), 3.05 (t, *J* = 5.7 Hz, 1H), 2.66-2.63 (m, 1H), 1.81-1.39 (m, 8H), 1.52 (s, 9H), 1.35 (s, 3H), 1.20/1.19 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 153.5, 142.9, 142.5, 142.4, 129.0, 128.9, 128.7, 128.6, 128.4, 126.6, 126.4, 83.7, 83.6, 82.7, 65.6, 62.9, 62.8, 61.0, 60.3, 59.3, 58.1, 58.0, 56.3, 38.8, 38.8, 34.8, 32.2, 27.9, 24.4, 20.9, 20.8, 17.1, 16.5, 16.5; IR (neat) 3026, 2934, 1743, 1495, 1453, 1370, 1279, 1256, 1163, 1098, 859, 747, 705 cm⁻¹; HRMS (ESI): *m/z* calcd for C₃₂H₄₄O₆Na [M+Na]⁺ 547.3036, found 547.3002; [α]_D = +11.0 (CHCl₃, *c* 2.01).

((2*S*,3*S*)-3-((*E*)-8-Methoxy-4-methyl-9,9-diphenylnon-3-enyl)-3-methyloxiran-2-yl)methanol (A26)



with (+)-diisopropyl tartrate (9.6 μ L, 45.8 μ mol) and the mixture was cooled to -35 to -30 °C. The mixture was stirred for 10 min and Ti(O-*i*-Pr)₄ (11.4 μ L, 38.2 μ mol) was added and the mixture was stirred for 15 min more. After that time, *t*-butyl hydroperoxide (5.0-6.0 M in decane, 0.23 mL, ~1.15 mmol) was added dropwise and the mixture was stirred for 30 min. Dienol **1.65** (150.0 mg, 0.382 mmol, dissolved in 1.0 mL of CH₂Cl₂) was added dropwise and the flask formerly containing the allylic alcohol was rinsed with CH₂Cl₂ (2 x 0.5 mL). The reaction mixture was stirred at -35 to -30 °C for 1.5 h and then water (0.6 mL) was added. The mixture was allowed to warm up to 0 °C and stirred for 1 h. A solution of 30% of NaOH saturated with NaCl (0.2 mL) was added and the mixture was warmed up to room temperature and stirred for

1.5 h. The suspension was filtered through a pad of Celite and the filtrate was dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (30% - 40% EtOAc in hexanes) to give the monoepoxy alcohol **A26** (156.0 mg, 97.3%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.44-7.23 (m, 10H), 5.09 (t, *J* = 6.6 Hz, 1H), 4.06 (d, *J* = 8.3 Hz, 1H), 3.99-3.93 (m, 1H), 3.86-3.83 (m, 1H), 3.70 (dd, *J* = 12.0, 6.6 Hz, 1H), 3.22 (s, 3H), 3.02-2.99 (m, 1H), 2.46 (br s, 1H), 2.10 (q, *J* = 7.7 Hz, 2H), 1.96-1.91 (m, 2H), 1.72-1.43 (m, 6H), 1.59 (s, 3H), 1.33 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 143.0, 142.5, 135.7, 128.9, 128.6, 128.3, 126.4, 126.3, 123.5, 83.6, 63.2, 61.5, 61.2, 57.9, 56.2, 39.6, 38.6, 31.6, 23.6, 23.2, 16.9, 15.9; IR (neat) 3425, 3027, 2934, 1599, 1495, 1452, 1385, 1103, 1032, 910, 733, 703 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₇H₃₆O₃Na [M+Na]⁺ 431.2562, found 431.2556; [α]_D = -3.8 (CHCl₃, *c* 1.17).

tert-Butyl ((2*S*,3*S*)-3-(2-((2*R*,3*R*)-3-(4-methoxy-5,5-diphenylpentyl)-3-methyloxiran-2-yl)ethyl)-3-methyloxiran-2-yl)methyl carbonate (1.54)

Ph₂CH \rightarrow O'Bu A solution of monoepoxy alcohol A26 (145.0 mg, 0.3549 mmol) in CH₃CN/DMM (5.3 mL, 1.2, v/v) was treated a 0.05 M solution of Na₂B₄O₇ in 4x10⁻⁴ M Na₂(EDTA) (3.5 mL), Bu₄NHSO₄ (4.8 mg, 14.2 µmol) and Shi ketone (45.8 mg, 0.177 mmol) sequentially. The mixture was cooled to 0 °C, and the Oxone (301 mg, 0.490 mmol), dissolved in 4x10⁻⁴ M Na₂(EDTA) (2.3 mL), and K₂CO₃ (284 mg, 2.06 mmol), dissolved in water (2.3 mL), were added simultaneously via a syringe pump over 2.0 h. After the addition was completed, the blue reaction mixture was stirred further for 15 min at 0 °C, then diluted with water (5 mL) and extracted with CH₂Cl₂ (3 x 25 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was dissolved in dry toluene (3.5 mL) and cooled to 0 °C. *N*-

Methylimidazole (36.8 µL, 0.4614 mmol) and Boc₂O (232 mg, 1.06 mmol) were added sequentially. The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. The reaction was quenched with water (5 mL) and extracted with CH₂Cl₂ (3 x 20 mL). The extracts were dried over MgSO₄, filtered and concentrated. The residue was azeotroped with hexanes (3 x 20 mL) to removed t-BuOH and the residue was purified by column chromatography (12% - 24% EtOAc in hexanes containing 0.5% Et₃N) to give the desired product 1.54 (153.5 mg, 82.4%, dr ~ 1:1 regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.43-7.20 (m, 10H), 4.29 (dd, J = 11.9, 4.8 Hz, 1H), 4.17 (dd, J = 11.9, 6.3 Hz, 1H), 4.04 (d, J= 8.4 Hz, 1H), 3.99-3.90 (m, 1H), 3.20/3.19 (s, 3H), 3.07 (t, J = 5.3 Hz, 1H), 2.67-2.62 (m, 1H), 1.77-1.42 (m, 10H), 1.54 (s, 9H), 1.35 (s, 3H), 1.22 (br s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 153.4, 142.8, 142.4, 142.4, 128.9, 128.8, 128.6, 128.5, 128.3, 126.5, 126.4, 83.5, 83.5, 82.6, 65.6, 63.0, 62.9, 60.8, 60.8, 60.3, 59.7, 58.0, 57.9, 56.2, 38.7, 38.7, 35.1, 32.1, 27.8, 24.4, 20.8, 20.6, 16.8, 16.5, 16.4; IR (neat) 2979, 2935, 1743, 1495, 1453, 1370, 1279, 1163, 1098, 912, 859, 733, 704 cm⁻¹; HRMS (ESI): m/z calcd for C₃₂H₄₄O₆Na [M+Na]⁺ 547.3036, found 547.3031; [α]_D = +0.5 (CHCl₃, c 1.21).

6,6-Diphenylhexane-1,5-diol (A27)

Ph₂CH OH In a round-bottomed flask, the δ-valerolactone (1.82 mL, 20.0 mmol) in CH₂Cl₂ (20.0 mL) at -78 °C was treated dropwise with DIBAL-H (1 M in hexanes, 22.0 mL, 22.0 mmol). The mixture was stirred at -78 °C for 1h, then quenched with saturated sodium tartrate (80 mL) and extracted with Et₂O (3 x 100 mL). The extracts were

dried over MgSO₄, filtered and concentrated. The crude lactol was used in the next step without further purification.

In another two-necked round-bottomed flask, diphenylmethane (13.4 mL, 80.0 mmol) in THF (50.0 mL) was treated dropwise with *n*-BuLi (1.6 M in hexanes, 50.0 mL, 80.0 mmol). The resulting deep orange solution was refluxed for 1 h and then cooled to 0 °C. The as-prepared crude lactol (dissolved in 5 mL THF) was added dropwise and the flask formerly containing the lactol was rinsed with THF (2 x 1.5 mL). The deep-orange mixture was stirred at 0 °C for 0.5 h, then quenched with saturated NH₄Cl (80 mL) and extracted with Et₂O (3 x 100 mL). The combined extracts were dried over MgSO₄ and evaporated. The resulting residue was purified by column chromatography (40% - 80% EtOAc in hexanes) to give the diol **A27** (4.42 g, 81.8%) as a white solid: ¹H NMR (300 MHz, CDCl₃) δ 7.57-7.18 (m, 10H), 4.38 (app t, *J* = 8.0 Hz, 1H), 3.90 (d, *J* = 8.5 Hz, 1H), 3.59 (t, *J* = 5.7 Hz, 2H), 1.88 (br s, 1H), 1.81 (br s, 1H), 1.67-1.42 (m, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 142.6, 141.7, 129.0, 129.0, 128.8, 128.4, 127.0, 126.7, 73.9, 62.7, 59.0, 34.6, 32.6, 22.0; IR (neat) 3358, 3023, 2948, 1596, 1493, 1450, 1345, 1039, 696 cm⁻¹; HRMS (ESI): *m/z* calcd for C₁₈H₂₂O₂Na [M+Na]⁺ 293.1517, found 293.1537.

6-(1-Phenyl-1*H*-tetrazol-5-ylthio)-1,1-diphenylhexan-2-ol (A28)

Ph₂CH (1.000 g, 3.699 mmol), 1-Ph₂CH (1.000 g, 3.699 mmol), 1-OH (1.164 g, 4.44 mmol) in THF (30.0 mL) was added dropwise diisopropyl azodicarboxylate (0.82 mL, 4.07 mmol). The mixture was warmed to room temperature and stirred for 20 min. The reaction was quenched with water (60 mL) and extracted with Et₂O (3 x 70 mL). The organic extracts were dried over MgSO₄ and evaporated. The resulting residue was purified by column chromatography (25% - 40% EtOAc in hexanes) to give the crude product which was further purified by column chromatography (3.5% Et₂O in CH₂Cl₂) to give **A28** (1.542 g, 96.8%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.60-7.50 (m, 5H), 7.40-7.15 (m, 10H), 4.39-4.33 (m, 1H), 3.86 (d, *J* = 8.5 Hz, 1H), 3.35 (t, *J* = 7.1 Hz, 2H), 1.86-1.38 (m, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 154.6, 142.4, 141.5, 133.8, 130.2, 129.9, 129.0, 128.9, 128.8, 128.3, 127.0, 126.8, 124.0, 73.6, 59.1, 34.4, 33.4, 29.1, 25.0; IR (neat) 3439, 3026, 2942, 1597, 1499, 1451, 1387, 1243, 1088, 1015, 910, 760, 733, 704 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₅H₂₆N₄OSNa [M+Na]⁺ 453.1725, found 453.1705.

5-(5-Methoxy-6,6-diphenylhexylthio)-1-phenyl-1*H*-tetrazole (A29)

5-(5-Methoxy-6,6-diphenylhexylsulfonyl)-1-phenyl-1*H*-tetrazole (1.66)

Ph₂CH $\xrightarrow{O_2}$ $\xrightarrow{N_1}$ $\xrightarrow{N_1}$ of the thioether **A29** (330 mg, 0.742 mmol) was added to a solution of the thioether **A29** (330 mg, 0.742 mmol) in CH₂Cl₂ (10.0 mL) and then *m*CPBA (pure, 435 mg, 2.52 mmol) was added in small portions. The mixture was stirred at 0 °C for 15 min, then at room temperature overnight. The reaction was quenched with saturated Na₂S₂O₃ solution (20 mL), stirred for 30 min and extracted with CH₂Cl₂ (3 x 25 mL). The extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (16% - 24% EtOAc in hexanes) to give the desired sulfone **1.66** (336 mg, 95.0%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.70-7.60 (m, 5H), 7.38-7.17 (m, 10H), 3.99 (d, *J* = 8.5 Hz, 1H), 3.94-3.88 (m, 1H), 3.72-3.64 (m, 2H), 3.15 (s, 3H), 1.96-1.84 (m, 2H), 1.68-1.53 (m, 3H), 1.49-1.43 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 153.7, 142.6, 142.3, 133.2, 131.6, 129.9, 128.9, 128.8, 128.6, 128.5, 126.7, 126.6, 125.3, 83.4, 58.2, 56.4, 56.1, 31.7, 23.9, 22.3; IR (neat) 3027, 2934, 2828, 1597, 1496, 1452, 1342, 1153, 1103, 762, 705 cm⁻¹; HRMS (ESI): *m/z* calcd for C₂₆H₂₈N₄O₃SNa [M+Na]⁺ 499.1780, found 499.1787.

((2*E*,6*E*)-11-Methoxy-3-methyl-12,12-diphenyldodeca-2,6-dienyloxy)(*tert*-butyl)dimethylsilane (A30)

Ph₂CH OMe OTBS A solution of sulfone **1.66** (300 mg, 0.630 mmol, azeotropically dried with benzene) in anhydrous 1,2-

dimethoxyethane (3.8 mL) at -78 °C was treated dropwise with KHMDS (0.5 M in 1,2dimethoxyethane, 1.51 mL, 0.755 mmol) and the resulting yellow mixture was stirred at this temperature for 1 h. After that time, aldehyde **1.60** (183 mg, 0.755 mmol, dissolved in 0.5 mL of 1,2-dimethoxyethane) was added dropwise. The reaction mixture was stirred at -78 °C for 1.5 h, then at room temperature overnight. The reaction was quenched with saturated NH₄Cl solution (5 mL), poured onto water (10 mL) and extracted with Et₂O (3 x 30 mL). The extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (2% - 3% EtOAc in hexanes) to give the desired diene **A30** (196.4 mg, 63.3%) as a pale yellow oil: ¹H NMR (500 MHz, CDCl₃) δ 7.38-7.17 (m, 10H), 5.35-5.33 (m, 2H), 5.30 (qt, *J* = 6.3, 1.0 Hz, 1H), 4.20 (d, *J* = 6.3 Hz, 1H), 4.00 (d, *J* = 8.2 Hz, 1H), 3.91-3.88 (m, 1H), 3.17 (s, 3H), 2.09-2.05 (m, 2H), 2.20-1.99 (m, 2H), 1.92-1.91 (m, 2H), 1.61 (s, 3H), 1.52-1.48 (m, 2H), 1.45-1.40 (m, 2H), 0.92 (s, 9H), 0.08 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 143.2, 142.6, 136.9, 130.4, 130.3, 129.1, 128.7, 128.6, 128.4, 126.5, 126.4, 124.7, 83.8, 60.5, 58.0, 56.3, 39.8, 32.7, 31.7, 31.1, 26.2, 25.2, 18.6, 16.6, -4.8; IR (neat) 3027, 2928, 2855, 1599, 1495, 1451, 1381, 1254, 1105, 1062, 836, 775, 701 cm⁻¹; HRMS (EI): *m/z* calcd for C₂₈H₃₉OSi (M-C₄H₉) ⁺⁺ 435.2719, found 435.2706.

(2E,6E)-11-Methoxy-3-methyl-12,12-diphenyldodeca-2,6-dien-1-ol (A31)

monohydrate (125 mg, 0.478 mmol). The yellow solution was stirred for 1.5 h and the concentrated. The resulting residue was purified by flash chromatography (20% - 30% EtOAc in hexanes) to give the allylic alcohol **A31** (143.1 mg, 94.8%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.49-7.18 (m, 10H), 5.44-5.36 (m, 3H), 4.15 (d, *J* = 6.8 Hz, 2H), 4.04 (d, *J* = 8.3 Hz, 1H), 3.96-3.91 (m, 1H), 3.19 (s, 3H), 2.14-2.02 (m, 4H), 2.00-1.90 (m, 2H), 1.75 (br s, 1H), 1.68 (s, 3H), 1.60-1.39 (m, 4H); ¹³C NMR (75 MHz, CDCl₃) δ 143.0, 142.5, 139.1, 130.4, 130.0,

129.0, 128.6, 128.3, 126.4, 126.3, 123.8, 83.7, 59.3, 57.9, 56.2, 39.6, 32.6, 31.6, 30.9, 25.0, 16.4; IR (neat) 3390, 3026, 2930, 1599, 1495, 1451, 1101, 1003, 969, 745, 703 cm⁻¹; HRMS (ESI): m/z calcd for C₂₆H₃₄O₂Na [M+Na]⁺ 401.2457, found 401.2464.

((2*R*,3*R*)-3-(2-((2*R*,3*R*)-3-(4-Methoxy-5,5-diphenylpentyl)oxiran-2-yl)ethyl)-3-methyloxiran-2-yl)methanol (A32)

Ph₂CH OMe $V_{\bar{O}}$ $V_{\bar{O}}$ $V_{\bar{O}}$ $V_{\bar{O}}$ $V_{\bar{O}}$ To a solution of dienol A31(100.0 mg, 0.264 mmol) in CH₃CN/DMM (7.9 mL, 1:2, v/v) were added a 0.05 M

solution of Na₂B₄O₇ in 4x10⁻⁴ M Na₂(EDTA) (5.3 mL), Bu₄NHSO₄ (7.2 mg, 21.1 µmol) and Shi ketone (102 mg, 0.396 mmol) sequentially. The mixture was cooled to -5 °C, and the Oxone (672 mg, 1.09 mmol), dissolved in 4x10⁻⁴ M Na₂(EDTA) (3.4 mL), and K₂CO₃ (635 mg, 4.59 mmol), dissolved in water (3.4 mL), were added simultaneously via a syringe pump over 2.0 h. After the addition was completed, the slightly blue reaction mixture was stirred further for 15 min at 0 °C, then diluted with water (10 mL) and extracted with CH₂Cl₂ (4 x 20 mL). The organic extracts were dried over MgSO₄, filtered and concentrated. The residue was purified by flash chromatography (50% - 70% EtOAc in hexanes) to give the diepoxy alcohol **A32** (101.9 mg, 94.0%, dr ~ 1:1 regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.39-7.15 (m, 10H), 4.01 (d, *J* = 8.4 Hz, 1H), 3.94-3.89 (m, 1H), 3.79 (dd, *J* = 12.1, 4.7 Hz, 1H), 3.69 (dd, *J* = 12.0, 6.3 Hz, 1H), 3.17/3.16 (s, 3H), 2.98-2.95 (m, 1H), 2.66-2.60 (m, 2H), 1.86-1.46 (m, 10H), 1.3 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 142.8, 142.4, 128.9, 128.7, 128.6, 128.4, 126.5, 126.4, 83.6, 62.6, 61.4, 60.7, 58.9, 58.8, 58.3, 58.2, 58.1, 58.0, 56.2, 34.5, 32.1, 32.0, 27.6, 21.6, 17.1; IR (neat)

3426, 2934, 1495, 1452, 1097, 1032, 732, 704 cm⁻¹; HRMS (EI): m/z calcd for C₂₆H₃₄O₄ (M⁺⁺) 410.2457, found 410.2447; [α]_D = +14.3 (CHCl₃, *c* 1.20).

tert-Butyl ((2*R*,3*R*)-3-(2-((2*R*,3*R*)-3-(4-methoxy-5,5-diphenylpentyl)oxiran-2-yl)ethyl)-3-methyloxiran-2-yl)methyl carbonate (1.55)

A solution of diepoxy alcohol A32 (99.2 mg, \tilde{O} O'Bu Ph₂CH ,, <u>,</u> ÓMe added 1-Methylimidazole (19 µL, 0.24 mmol) and Boc₂O (106 mg, 0.484 mmol) sequentially. The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. After that time, the reaction was quenched with water (10 mL) and extracted with CH₂Cl₂ (3 x 25 mL). The extracts were dried over MgSO₄, filtered and concentrated. The residue was azeotroped with hexanes (2 x 20 mL) to remove t-BuOH and the residue was purified by column chromatography (16% - 20% EtOAc in hexanes containing 0.5% Et₃N) to give the desired product 1.55 (102.0 mg, 82.6%, dr ~ 1:1 regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.39-7.17 (m, 10H), 4.22 (dd, J = 11.8, 4.7 Hz, 1H), 4.14 (dd, J = 11.9, 6.0 Hz, 1H), 4.00 (d, J = 8.3 Hz, 1H), 3.93-3.88 (m, 1H), 3.17/3.16 (s, 3H), 3.02 (t, J = 5.4 Hz, 1H), 2.64-2.60 (m, 2H), 1.75-1.44 (m, 10H), 1.50 (s, 9H), 1.32 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 153.5, 142.9, 142.4, 129.0, 128.7, 128.6, 128.4, 126.6, 126.4, 83.7, 82.8, 65.6, 60.2, 59.3, 58.8, 58.8, 58.2, 58.1, 56.3, 34.2, 32.2, 27.9, 27.6, 21.7, 17.1; IR (neat) 2978, 2934, 1743, 1495, 1453, 1370, 1279, 1256, 1163, 1097, 859, 746, 704 cm⁻¹; HRMS (ESI): m/z calcd for C₃₁H₄₂O₆Na [M+Na]⁺ 533.2879, found 533.2866; $[\alpha]_D = +19.1$ (CHCl₃, *c* 1.08).

(S)-4-((R)-Tetrahydro-5-methoxyfuran-2-yl)-1,3-dioxolan-2-one (1.68)

To *tert*-butyl carbonate **1.47** (92.0 mg, 0.223 mmol) in MeO dichloroethane/toluene (8.6 mL, 5:1, v/v) in borosilicate flask at room temperature were added the activated 4Å molecular sieves (184 mg), anhydrous Na₂S₂O₃ (184 mg), NaOAc (184 mg) and N-methylquinolinium hexafluorophosphate (6.4 mg, 22.3 µmol). The mixture was photoirradiated with gentle air bubbling for 3 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with Et₂O (40 mL). The filtrate was concentrated and the resulting residue was purified by flash chromatography (45% - 55% EtOAc in hexanes) to give the product 1.68 (24.8 mg, 59.0%) in a 1.9:1 ratio as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 5.04 (dd, J = 4.5, 1.8 Hz, 66% of 1H), 5.01-4.99 (m, 34% of 1H), 4.67-4.46 (m, 2.4H), 4.39-4.20 (m, 1.6H), 3.33/3.32 (s, 3H), 2.26-1.93 (m, 3H), 1.74-1.66 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 155.1 (minor), 154.9 (major), 105.7 (minor), 105.6 (major), 79.4 (minor), 77.8 (minor), 77.1 (major), 66.8 (major), 66.4 (minor), 55.1 (major), 32.7 (minor), 31.7 (major), 25.9 (minor), 25.5 (major); IR (neat) 2920, 1807, 1464, 1376, 1170, 1088, 1031, 955 cm⁻¹; HRMS (EI): m/z calcd for C₇H₉O₄ (M-CH₃O)^{+•} 157.0501, found 157.0499.

(S)-4-((R)-Tetrahydro-5-oxofuran-2-yl)-1,3-dioxolan-2-one (1.69)

At 0 °C, the acetal **1.68** (6.4 mg, 34.0 μ mol) in acetone (1.0 mL) was treated with Jones reagent (2.67M, 60 μ L, 0.160 mmol). The mixture was stirred at 0 °C for 1 h, and Jones reagent (2.67M, 60 μ L, 0.160 mmol) was added. The mixture was stirred at 0 °C fro 30 min, then at room temperature for 2 h. After that time, the mixture was purified by column chromatography (10% - 20% EtOAc in CH₂Cl₂) to give the lactone **1.69** (4.3 mg, 74.1%) as a pale yellow solid: ¹H NMR (300 MHz, CDCl₃) δ 4.74 (ddd, J = 8.0, 6.7, 5.8 Hz, 1H), 4.64 (t, J = 8.9 Hz, 1H), 4.65-4.58 (m, 1H), 4.40 (dd, J = 8.9, 5.6 Hz, 1H), 2.68-2.62 (m, 2H), 2.60-2.50 (m, 1H), 2.20-2.10 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 175.2, 154.0, 78.1, 76.2, 66.7, 27.5, 24.0; IR (neat) 2919, 1778, 1462, 1401, 1328, 1173, 1087, 1048 cm⁻¹; HRMS (EI): m/z calcd for C₇H₈O₅ [M+H]⁺ 173.0450, found 173.0455.

To monoepoxide **1.48** (102.0 mg, 0.239 mmol) in dichloroethane/toluene (9.2 mL, 5:1, v/v) in borosilicate flask at room temperature were added the activated 4Å molecular sieves (204 mg), anhydrous Na₂S₂O₃ (204 mg), NaOAc (204 mg) and *N*-methylquinolinium hexafluorophosphate (6.9 mg, 23.9 µmol). The mixture was photoirradiated with gentle air bubbling for 3 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with Et₂O (50 mL). The filtrate was concentrated and the resulting yellowish-green residue was dissolved in CH₂Cl₂ (2.0 mL). To this solution were added Et₃N (0.22 mL, 1.6 mmol), Ac₂O (57 µL, 0.6 mmol) and DMAP (2.4 mg, 20 µmol) sequentially. The mixture was stirred at room temperature for 3 h, then concentrated and purified by column chromatography (20% - 50% EtOAc in hexanes) to provide the cyclization products, which were further purified by column chromatography (4% - 10% EtOAc in CH₂Cl₂) to give *exo*-product **1.71** (14.6 mg, 30%, dr = 2:1) and *endo*-product **1.72** (10.6 mg, 22%, dr = 3.4:1) as colorless oils.

(S)-4-((R)-Tetrahydro-6-methoxy-2H-pyran-2-yl)-1,3-dioxolan-2-one (1.71)

3H), 3.36 (s, 67% of 3H), 1.98-1.16 (m, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 155.1 (major), 155.0 (minor), 103.4 (minor), 98.4 (major), 78.0 (major), 77.4 (minor), 75.3 (minor), 68.2 (major), 66.4 (minor), 66.1 (major), 56.4 (minor), 55.0 (major), 30.9 (minor), 29.5 (major), 26.6 (minor), 26.5 (major), 21.2 (minor), 17.2 (major); IR (neat) 2952, 2851, 1799, 1389, 1174, 1078, 1031 cm⁻¹; HRMS (EI): *m/z* calcd for C₈H₁₁O₄ (M⁺⁺) 171.0657, found 171.0650.

(4aR,9aS)-Hexahydro-6-methoxy-4H-[1,3]dioxino[5,4-b]oxepin-2-one (1.72)

$$\underset{MeO}{\overset{}}{\overset{}}_{H} \overset{H}{\overset{}}_{H} \overset{H}{\overset{H}} \overset{H}{\overset{H}}$$

of 1H), 3.42 (s, 23% of 3H), 3.36 (s, 77% of 3H), 2.37-2.14 (m, 3H), 1.96-1.93 (m, 23% of 1H), 1.77-1.42 (m, 5H); 13 C NMR (75 MHz, CDCl₃) δ 148.5 (minor), 148.1 (major), 104.9 (minor), 102.8 (major), 81.2 (minor), 80.8 (major), 69.3 (major), 69.2 (minor), 68.2 (minor), 61.5 (major), 56.4 (minor), 55.8 (major), 35.5 (major), 34.2 (major), 33.5 (minor), 27.9 (minor), 17.7 (major), 16.6 (minor); IR (neat) 2943, 1760, 1403, 1382, 1224, 1140, 1057 cm⁻¹; HRMS (EI): *m/z* calcd for C₈H₁₁O₄ (M⁺⁺) 171.0657, found 171.0651; an analytical sample of the major diastereomer was obtained through purifying the above mixture by column chromatography (35% - 45% EtOAc in hexanes): ¹H NMR (300 MHz, C₆D₆) δ 4.02 (dd, *J* = 8.9, 5.8 Hz, 1H), 3.60 (dd, *J* = 10.1, 5.5 Hz, 1H), 3.42 (t, *J* = 10.2 Hz, 1H), 3.34-3.27 (m, 2H), 2.85 (s, 3H), 1.76-1.67 (m, 1H), 1.57-1.47 (m, 1H), 1.10 (dddd, *J* = 15.3, 11.6, 8.9, 1.0 Hz, 1H), 0.95-0.85 (m, 2H), 0.82-0.71 (m, 1H).

(R)-Tetrahydro-6-((S)-2-oxo-1,3-dioxolan-4-yl)pyran-2-one (1.73)

A solution of acetal **1.71** (6.0 mg, 29.7 µmol) in CH₂Cl₂ (0.6 mL) at 0 °C was treated with *m*CPBA (pure, 6.7 mg, 38.6 µmol) and BF₃•OEt₂ (4.5 µL, 35.6 µmol) sequentially. After stirred at 0 °C for 10 min and then at room temperature for 1.5 h, the mixture was cooled to 0 °C and Et₃N (20.7 µL, 148 µmol) was added dropwise. The mixture was stirred at 0 °C for 1 h, then concentrated, and the resulting residue was purified by column chromatography (15% - 25% EtOAc in CH₂Cl₂) to give the desired lactone **1.73** (4.6 mg, 83.6%) as a colorless liquid: ¹H NMR (300 MHz, CDCl₃) δ 4.69-4.59 (m, 2H), 4.56-4.41 (m, 2H), 2.69 (dddd, *J* = 18.0, 6.8, 4.8, 1.1 Hz, 1H), 2.54 (ddd, *J* = 17.9, 9.3, 7.0 Hz, 1H), 2.24-2.16 (m, 1H), 2.08-1.90 (m, 2H), 1.64 (dtd, *J* = 13.8, 11.0, 5.2 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 169.3, 154.2, 79.0, 76.3, 66.9, 29.8, 24.6, 18.2; IR (neat) 2919, 1790, 1732, 1376, 1239, 1166, 1056 cm⁻¹; HRMS (EI): *m/z* calcd for C₈H₁₀O₅ (M⁺⁺) 186.0528, found 186.0536.

4-(Tetrahydro-5-methoxy-2-methylfuran-2-yl)-1,3-dioxolan-2-one (1.76), Hexahydro-6methoxy-8a-methylpyrano[3,2-d][1,3]dioxin-2-one (1.77) and (4a*R*,6*S*,8a*R*)-Hexahydro-6methoxy-8a-methylpyrano[3,2-d][1,3]dioxin-2-one (1.78)



and the residue was washed with EtOAc (30 mL). The filtrate was concentrated and the resulting residue was purified by flash chromatography (30% - 45% EtOAc in hexanes) to provide a mixture of **1.76** and **1.77** (19.7 mg, 33.2%, a pale yellow oil) with a molar ratio of 4.8:1 and *cis*-fused *endo*-product **1.78** (3.8 mg, 8.0%) as a white solid. For the mixture of **1.76** and **1.77**: IR (neat) 2928, 2835, 1791, 1755, 1463, 1375, 1170, 1084, 1034, 951 cm⁻¹. For *cis*-fused *endo*-product **1.78**: ¹H NMR (300 MHz, CDCl₃) δ 4.78 (app d, *J* = 2.0 Hz, 1H), 4.66 (dd, *J* = 12.1, 2.7 Hz, 1H), 4.34 (dd, *J* = 12.1, 0.4 Hz, 1H), 3.86 (app d, *J* = 2.0 Hz, 1H), 3.41 (s, 3H), 2.12-2.00 (m, 1H), 1.94 (dd, *J* = 12.8, 4.0 Hz, 1H), 1.87-1.81 (m, 1H), 1.67-1.61 (m, 1H), 1.45 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 148.5, 98.3, 78.6, 69.3, 63.2, 55.4, 29.4, 25.3, 25.1; IR (neat) 2932, 1748, 1212, 1178, 1130, 1060, 1024 cm⁻¹; HRMS (EI): *m/z* calcd for C₉H₁₅O₅ [M+H]⁺ 203.0919, found 203.0929.

(S)-4-((R)-Tetrahydro-2-methyl-5-oxofuran-2-yl)-1,3-dioxolan-2-one (1.79)

To a solution of the mixture of **1.76** and **1.77** (18.9 mg, 93.5 µmol) in acetone (3.0 mL) at 0 °C was added Jones reagent (0.3 mL). The mixture was stirred at 0 °C for 15 min and then at room temperature for 3 h. After that time, the reaction was quenched with isopropyl alcohol (1 drop), concentrated and purified by column chromatography (2% - 20% EtOAc in CH₂Cl₂) to give the unreacted acetal **1.77** (2.9 mg, 15.3%) as a white solid and the title lactone **1.79** (11.0 mg, ~74.8% based on unreacted acetal): ¹H NMR (300 MHz, CDCl₃) δ 4.72 (dd, J = 8.4, 6.2 Hz, 1H), 4.57 (t, J = 9.0 Hz, 1H), 4.36 (dd, J = 9.2, 6.1 Hz, 1H), 2.70 (t. J = 8.8 Hz, 2H), 2.34-2.24 (m, 1H), 2.19-2.09 (m, 1H), 1.47 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 175.0, 154.1, 83.9, 78.7, 65.4, 30.0, 28.2, 20.8; IR (neat) 2920, 1789, 1463, 1267, 1167, 1082 cm⁻¹; HRMS (EI): m/z calcd for C₈H₁₁O₅ [M+H]⁺ 187.0606, found 187.0612.

(4aR,6S,8aS)-Hexahydro-6-methoxy-8a-methylpyrano[3,2-d][1,3]dioxin-2-one (1.77)

 $\underbrace{\mathsf{MeO}}_{\mathsf{H}} \underbrace{\mathsf{O}}_{\mathsf{H}} \underbrace{\mathsf{O}} \underbrace{\mathsf{O}}$

(4aR,9aS)-Hexahydro-6-methoxy-9a-methyl-4H-[1,3]dioxino[5,4-b]oxepin-2-one (1.81)



To *tert*-butyl carbonate **1.50** (65.8 mg, 149 μ mol) in dichloroethane/toluene (5.7 mL, 5:1, v/v) in borosilicate flask at room temperature were added the activated 4Å molecular sieves (132 mg),

anhydrous Na₂S₂O₃ (132 mg), NaOAc (132 mg) and *N*-methylquinolinium hexafluorophosphate (4.3 mg, 14.9 µmol). The mixture was photoirradiated with gentle air bubbling for 2 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc (20 mL). The filtrate was concentrated and the resulting residue was purified by flash chromatography (5% - 15% EtOAc in CH₂Cl₂) to provide the desired compound **1.81** (23.7 mg, 73.4%) as a mixture of two diastereomers in a 1.2:1 ratio: ¹H NMR (300 MHz, CDCl₃) δ 4.76 (t, *J* = 3.8 Hz, 46% of 1H), 4.66 (dd, *J* = 8.8, 5.8 Hz, 54% of 1H), 4.34 (dd, *J* = 10.8, 6.4 Hz, 46% of 1H), 4.29-4.18 (m, 54% of 2H), 4.19 (t, *J* = 10.8 Hz, 46% of 1H), 3.42 (s, 46% of 3H), 3.35 (s, 54% of 3H),

2.23-2.01 (m, ~1.5H), 1.96-1.92 (m, 46% of 1H), 1.75-1.58 (m, ~3.5H), 1.51 (s, 46% of 3H), 1.48 (s, 54% of 3H), 1.45-1.35 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 148.2, 148.0, 104.3, 102.8, 84.2, 83.0, 68.1, 66. 7, 66.3, 62.7, 56.3, 55.7, 43.2, 41.6, 34.5, 34.4, 19.5, 19.3, 18.3, 16.7; IR (neat) 2941, 1755, 1464, 1384, 1252, 1199, 1128, 1091, 1050, 969 cm⁻¹; HRMS (EI): *m/z* calcd for C₉H₁₃O₄Na (M-CH₃O)⁺⁺ 185.0814, found 185.0811. An analytical sample of the slightly major diastereomer was obtained through purifying the above mixture by column chromatography (35% - 40% EtOAc in hexanes): ¹H NMR (300 MHz, C₆D₆) δ 4.00 (dd, *J* = 8.8, 5.9 Hz, 1H), 3.60 (d, *J* = 10.4 Hz, 1H), 3.58 (d, *J* = 6.8 Hz, 1H), 3.44 (dd, *J* = 10.4, 6.8 Hz, 1H), 2.85 (s, 3H), 1.56-1.48 (m, 2H), 1.19-1.10 (m, 2H), 0.98-0.88 (m, 1H), 0.94 (s, 3H), 0.75-0.66 (m, 1H).

(*S*)-4-((2*R*,5*S*)-Tetrahydro-5-((*R*)-tetrahydro-5-methoxy-2-methylfuran-2-yl)-2methylfuran-2-yl)-1,3-dioxolan-2-one (1.85) and (4a*R*,5a*S*,9a*R*,11a*S*)-8-Methoxy-5a,11adimethyldecahydro-1,3,5,9-tetraoxadibenzo[a,d]cyclohepten-2-one (1.86)

mL, 5:1, v/v) in borosilicate flask at room temperature were added the activated 4Å molecular sieves (258 mg), anhydrous Na₂S₂O₃ (258 mg), NaOAc (258 mg) and *N*-methylquinolinium hexafluorophosphate (7.3 mg, 25.3 μ mol). The mixture was photoirradiated with gentle air bubbling for 4.5 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc (40 mL). The filtrate was concentrated and the resulting residue was purified by flash chromatography (35% - 50% EtOAc

in hexanes) to provide a mixture of the above two products (28.7 mg, 39.6%) as a colorless oil: IR (neat) 2926, 1796, 1754, 1460, 1374, 1166, 1085, 1036, 1006, 952 cm⁻¹.

(S)-4-((2R,5S)-Tetrahydro-5-((R)-tetrahydro-2-methyl-5-oxofuran-2-yl)-2-methylfuran-2yl)-1,3-dioxolan-2-one (A33)

A mixture of acetals **1.85** and **1.86** (20.8 mg, 72.6 µmol) in acetone (2.1 mL) at 0 °C was treated dropwise with Jones reagent (0.2 mL). The mixture was stirred at 0 °C for 10 min, then at room temperature for 1.5 h and purified without workup by column chromatography (50% - 90% EtOAc in hexanes) to give the unreacted acetal **1.86** (3.2 mg, 15.4%, nearly pure) and lactone **A33** (13.8 mg, ~80%). For lactone **A33**: ¹H NMR (300 MHz, CDCl₃) δ 4.60 (dd, J = 8.4, 6.2 Hz, 1H), 4.50 (t, J = 8.6 Hz, 1H), 4.38 (dd, J = 8.7, 6.2 Hz, 1H), 4.08 (dd, J = 8.6, 6.3 Hz, 1H), 2.64-2.58 (m, 2H), 2.31-2.19 (m, 1H), 2.10-2.04 (m, 2H), 1.94-1.73 (m, 3H), 1.39 (s, 3H), 1.27 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 176.7, 155.1, 86.9, 83.5, 83.2, 79.5, 66.0, 34.1, 29.3, 29.2, 26.7, 23.8, 21.0; IR (neat) 2958, 2924, 2853, 1790, 1770, 1456, 1382, 1248, 1166, 1085, 1020, 944, 770, 728 cm⁻¹; HRMS (EI): m/z calcd for C₁₃H₁₈O₆ (M⁺⁺) 270.1103, found 270.1094; [α]_D = +4.5 (CHCl₃, *c* 0.24).

(4a*R*,5a*S*,9a*R*,11a*S*)-5a,11a-Dimethyloctahydro-1,3,5,9-tetraoxadibenzo[a,d]cycloheptene-2,8-dione (A34)



A solution of acetal **1.86** (2.9 mg, 11.2 μ mol) in CH₂Cl₂ (0.5 mL) at 0 °C was treated with *m*CPBA (pure, 2.5 mg, 14.6 μ mol) and BF₃•OEt₂ (1.9 μ L, 13.4 μ mol) sequentially. After stirred at 0 °C for 10 min and

then at room temperature for 30 min, the mixture was cooled to 0 °C and Et₃N (7.8 µL, 56.0

μmol) was added dropwise. The mixture was stirred at 0 °C for 30 min, and purified by column chromatography (10% - 20% EtOAc in CH₂Cl₂) to give the desired lactone **A34** (1.8 mg, 66.7%) as colorless needles: ¹H NMR (300 MHz, CDCl₃) δ 4.28 (dd, J = 8.6, 5.1 Hz, 1H), 4.21-4.14 (m, 1H), 4.09 (dd, J = 10.1, 8.6 Hz, 1H), 4.06 (dd, J = 11.0, 2.9 Hz, 1H), 2.80 (ddd, J = 18.3, 9.4, 5.5 Hz, 1H), 2.64 (ddd, J = 18.3, 8.7, 7.4 Hz, 1H), 2.35-2.28 (m, 1H), 2.17-1.88 (m, 5H), 1.50 (s, 3H), 1.38 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 170.4, 148.2, 83.1, 82.7, 77.4, 75.7, 66.5, 65.2, 37.2, 34.3, 27.5, 24.8, 22.4, 16.0; IR (neat) 2923, 1747, 1463, 1408, 1229, 1124, 1068 cm⁻¹; ¹H NMR (300 MHz, C₆D₆) δ 3.44 (d, J = 7.7 Hz, 1H), 3.43 (d, J = 9.6 Hz, 1H), 3.18 (dd, J = 9.6, 7.7 Hz, 1H), 2.93 (dd, J = 10.9, 3.0 Hz, 1H), 2.02-1.97 (m, 2H), 1.65 (td, J = 15.3, 4.8 Hz, 1H), 1.43-1.26 (m, 3H), 1.17-1.04 (m, 2H), 0.81 (s, 3H), 0.48 (s, 3H); HRMS (ESI): *m/z* calcd for C₁₃H₁₈O₆Na [M+Na]⁺ 293.1001, found 293.1020; [α]_D = +101 (CHCl₃, *c* 0.15).

(*R*)-4-((2*S*,5*S*)-tetrahydro-5-((*R*)-tetrahydro-5-methoxy-2-methylfuran-2-yl)-2methylfuran-2-yl)-1,3-dioxolan-2-one (1.87) and (4a*S*,5a*S*,9a*R*,11a*R*)-8-Methoxy-5a,11adimethyldecahydro-1,3,5,9-tetraoxadibenzo[a,d]cvclohepten-2-one (1.88)

mL, 5:1, v/v) in borosilicate flask at room temperature were added the activated 4Å molecular sieves (300 mg), anhydrous Na₂S₂O₃ (300 mg), NaOAc (300 mg) and *N*-methylquinolinium hexafluorophosphate (8.5 mg, 29.4 μ mol). The mixture was photoirradiated with gentle air bubbling for 4.5 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc (40 mL). The filtrate was

concentrated and the resulting residue was purified by flash chromatography (35% - 50% EtOAc in hexanes) to provide a mixture of the above two products (51.2 mg, 60.9%): IR (neat) 2925, 1797, 1750, 1462, 1384, 1259, 1167, 1120 cm⁻¹.

(*R*)-4-((2*S*,5*S*)-tetrahydro-5-((*R*)-tetrahydro-2-methyl-5-oxofuran-2-yl)-2-methylfuran-2yl)-1,3-dioxolan-2-one (A35)

A mixture of acetals **1.87** and **1.88** (34.9 mg, 122 µmol) in acetone (1.8 mL) at 0 °C was treated dropwise with Jones reagent (0.3 mL). The mixture was stirred at 0 °C for10 min, then at room temperature for 1.5 h and purified without workup by column chromatography (50% - 90% EtOAc in hexanes) to give the unreacted acetal **1.88** (4.9 mg, nearly pure) and lactone **A35** (22.1 mg, ~81%). For lactone **A35**: ¹H NMR (300 MHz, CDCl₃) δ 4.58 (dd, *J* = 8.3, 6.1 Hz, 1H), 4.48 (t, *J* = 8.4 Hz, 1H), 4.32 (dd, *J* = 8.8, 6.1 Hz, 1H), 4.08 (dd, *J* = 8.8, 5.6 Hz, 1H), 2.65-2.59 (m, 2H), 2.24 (ddd, *J* = 12.9, 9.6, 6.9 Hz, 1H), 2.07-1.81 (m, 5H), 1.38 (s, 3H), 1.27 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 176.7, 155.0, 86.9, 85.1, 83.1, 80.3, 66.2, 34.5, 29.4, 29.1, 27.0, 23.4, 21.3; IR (neat) 2979, 2880, 1790, 1767, 1454, 1382, 1170, 1111, 1085, 944 cm⁻¹; HRMS (EI): *m/z* calcd for C₁₃H₁₈O₆ (M⁺⁺) 270.1103, found 270.1095; [α]_D = -11.3 (CHCl₃, *c* 1.03).

(4aS,5aS,9aR,11aR)-5a,11a-Dimethyloctahydro-1,3,5,9-tetraoxadibenzo[a,d]cycloheptene-

2,8-dione (A36)



A solution of acetal **1.88** (4.5 mg, 15.7 μ mol) in CH₂Cl₂ (0.5 mL) at 0 °C was treated with *m*CPBA (pure, 3.5 mg, 20.4 μ mol) and BF₃•OEt₂ (2.7 μ L, 18.8 μ mol) sequentially. After stirred at 0 °C for 10 min and

at room temperature for 20 min, the mixture was cooled to 0 °C and Et₃N (10.9 µL, 78.5 µmol) was added dropwise. The mixture was stirred at 0 °C for 30 min and purified by column chromatography (15% - 25% EtOAc in CH₂Cl₂) to give the desired lactone **A36** (3.0 mg, 71.4%) as colorless needles: ¹H NMR (300 MHz, CDCl₃) δ 4.40 (app dd, *J* = 10.0, 2.2 Hz, 1H), 4.23 (dd, *J* = 10.5, 6.1 Hz, 1H), 4.09 (t, *J* = 10.2, Hz, 1H), 4.00 (dd, *J* = 10.1, 6.1 Hz, 1H), 2.88 (ddd, *J* = 18.3, 11.2, 4.7 Hz, 1H), 2.72 (ddd, *J* = 18.3, 9.6, 5.5 Hz, 1H); 2.21-1.76 (m, 6H), 1.49 (s, 3H), 1.24 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 170.7, 147.4, 83.8, 81.4, 77.4, 66.6, 64.9, 39.1, 30.8, 28.1, 24.4, 20.5, 19.5; IR (neat) 2924, 1748, 1463, 1408, 1229, 1124, 1068 cm⁻¹; HRMS (ESI): *m/z* calcd for C₁₃H₁₈O₆Na [M+Na]⁺ 293.1001, found 293.0988; [α]_D = +48.7 (CHCl₃, *c* 0.23).

(4a*R*,5a*S*,10a*R*,12a*S*)-9-Methoxy-5a,12a-dimethyldecahydro-1,3,5,10-tetraoxabenzo[b]heptalen-2-one (1.91)



To diepoxide **1.53** (145 mg, 276 μmol) in dichloroethane/toluene (10.6 mL, 5:1, v/v) in borosilicate flask at room temperature were added the activated 4Å molecular sieves (290 mg), anhydrous

Na₂S₂O₃ (290 mg), NaOAc (290 mg) and *N*-methylquinolinium hexafluorophosphate (8.0 mg, 27.6 µmol). The mixture was photoirradiated with gentle air bubbling for 2 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc (40 mL). The filtrate was concentrated and the resulting residue was purified by flash chromatography (25% - 35% EtOAc in hexanes) to provide the product **1.91** (44.8 mg, 54.0%, pale yellow liquid) as two diastereomers in about 1:1 ratio: ¹H NMR (300 MHz, CDCl₃) δ 4.54-4.48 (m, 1H), 4.25-3.98 (m, 3H), 3.92-3.85 (m, 0.5H), 3.56 (dd, *J* = 11.2, 2.4 Hz, 0.5H), 3.40/3.37 (s, 3H), 3.24 (dd, *J* = 10.8, 3.8 Hz, 0.5H), 2.26-1.94 (m, 2.5H), 1.91-

1.72 (m, 3.5H), 1.65-1.52 (m, 3.5H), 1.47/1.44 (s, 3H), 1.33/1.29 (s, 3H), 1.21-1.18 (m, 0.5H); ¹³C NMR (75 MHz, CDCl₃) δ 149.0, 148.6, 105.2, 102.6, 83.3, 83.2, 81.4, 80.3, 79.7, 75.6, 67.0, 67.0, 65.2, 63.8, 56.1, 55.9, 44.6, 43.3, 37.5, 37.0, 35.3, 33.6, 27.6, 26.2, 22.3, 21.6, 19.2, 17.7, 17.0, 16.7; IR (neat) 2940, 1759, 1454, 1384, 1209, 1111, 1053, 1008, 921 cm⁻¹; HRMS (ESI): *m/z* calcd for C₁₅H₂₄O₆Na [M+Na]⁺ 323.1471, found 323.1500; [α]_D = +31.5 (CHCl₃, *c* 1.45).

(4a*R*,5a*S*,10a*R*,12a*S*)-5a,12a-Dimethyldecahydro-1,3,5,10-tetraoxabenzo[b]heptalene-2,9dione (1.92)



A solution of acetal **1.91** (15.6 mg, 51.9 μ mol) in CH₂Cl₂ (0.5 mL) at 0 °C was treated with *m*CPBA (pure, 11.6 mg, 67.5 μ mol) and BF₃•OEt₂ (7.2 μ L, 57.1 μ mol) sequentially. After stirred at 0 °C for 10

min, then at room temperature for 1 h, the mixture was cooled to 0 °C and Et₃N (36.2 µL, 256 µmol) was added dropwise. The mixture was stirred at 0 °C for 1.5 h, the quenched with a mixture of saturated NaHCO₃/saturated Na₂S₂O₃ (4 mL, 1:1, v/v). The mixture was poured onto water (5 mL) and extracted with Et₂O (3 x 25 mL). The extracts were dried over MgSO₄, filtered and concentrated, and the resulting residue was purified by column chromatography (10% - 20% EtOAc in CH₂Cl₂) to give the desired lactone **1.92** (9.9 mg, 66.9%) as a white crystalline solid: ¹H NMR (300 MHz, CDCl₃) δ 4.26-4.20 (m, 2H), 4.14-4.06 (m, 2H), 2.70-2.55 (m, 2H), 2.35-2.23 (m, 2H), 1.99-1.81 (m, 4H), 1.77-1.68 (m, 2H), 1.46 (s, 3H), 1.31 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 173.3, 148.6, 84.3, 82.2, 78.9, 66.8, 64.3, 43.2, 36.2, 33.6, 26.6, 22.0, 20.0, 15.8; IR (neat) 2989, 2941, 2871, 1748, 1727, 1501, 1454, 1365, 1328, 1272, 1212, 1098, 1040 cm⁻¹; HRMS (ESI): *m/z* calcd for C₁₄H₂₀O₆Na [M+Na]⁺ 307.1158, found 307.1158. [α]_D = +50.4 (CHCl₃, *c* 0.42).

(4a*S*,5a*S*,10a*R*,12a*R*)-9-Methoxy-5a,12a-dimethyldecahydro-1,3,5,10-tetraoxabenzo[b]heptalen-2-one (1.93)



To diepoxide **1.54** (48.2 mg, 91.9 μ mol) in dichloroethane/toluene (3.5 mL, 5:1, v/v) in borosilicate flask at room temperature were added the activated 4Å molecular sieves (96 mg), anhydrous

Na₂S₂O₃ (96 mg), NaOAc (96 mg) and *N*-methylquinolinium hexafluorophosphate (2.6 mg, 9.2 μmol). The mixture was photoirradiated with gentle air bubbling for 3 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc (30 mL). The filtrate was concentrated and the resulting residue was purified by flash chromatography (25% - 35% EtOAc in hexanes) to provide the product **1.93** (21.7 mg, 78.6%, pale yellow solid) as two diastereomers in about 1:1 ratio: ¹H NMR (300 MHz, CDCl₃) δ 4.69 (dd, J = 3.8, 2.2 Hz, 0.5H), 4.54 (dd, J = 8.9, 5.7 Hz, 0.5H), 4.17 (dd, J = 10.7, 6.6 Hz, 1H), 4.02 (t, J = 10.7 Hz, 1H), 3.90 (dd, J = 10.7, 6.6 Hz, 1H), 3.90-3.85 (m, 0.5H), 3.52 (dd, J = 10.1, 0.8 Hz, 0.5H), 3.40/3.37 (s, 3H), 2.08-2.00 (m, 2H), 1.89-1.53 (m, 8H), 1.44 (s, 3H), 1.21/1.17 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 151.6, 148.0, 102.7, 102.4, 83.6, 83.6, 81.3, 80.6, 78.8, 74.1, 67.0 (2C), 64.0, 63.9, 56.0, 55.8, 40.5, 40.2, 39.8, 39.5, 33.7, 33.4, 27.3 (2C), 20.8, 20.3, 19.4, 19.3, 19.3, 17.5; IR (neat) 2940, 1755, 1461, 1382, 1246, 1223, 1116, 1051, 913 cm⁻¹; HRMS (ESI): *m/z* calcd for C₁₅H₂₄O₆Na [M+Na]⁺ 323.1471, found 323.1462; [α]_D = +26.6 (CHCl₃, *c* 0.55).

(4a*S*,5a*S*,10a*R*,12a*R*)-5a,12a-Dimethyldecahydro-1,3,5,10-tetraoxabenzo[b]heptalene-2,9dione (1.94)



A solution of acetal **1.93** (19.0 mg, 63.2 μ mol) in CH₂Cl₂ (2.0 mL) at 0 °C was treated with *m*CPBA (pure, 14.2 mg, 82.2 μ mol) and BF₃•OEt₂ (9.5 μ L, 75.8 μ mol) sequentially. The mixture was stirred at

0 °C for 10 min, and then at room temperature for 1 h. After that time, the mixture was cooled to 0 °C and Et₃N (44.0 μL, 316 μmol) was added dropwise. The mixture was stirred at 0 °C for 30 min, then concentrated, and the resulting residue was purified by column chromatography (15% -25% EtOAc in CH₂Cl₂) to give the desired lactone **1.94** (14.4 mg, 80.0%) as a white solid: ¹H NMR (600 MHz, CDCl₃) δ 4.47 (dd, *J* = 10.4 Hz, 1H), 4.20 (dd, *J* = 10.7, 6.4 Hz, 1H), 4.05 (t, *J* = 10.7 Hz, 1H), 3.98 (dd, *J* = 10.5, 6.4 Hz, 1H), 2.70 (dt, *J* = 14.1, 2.2 Hz, 1H), 2.64 (ddd, *J* = 14.1, 5.8, 1.3 Hz, 1H), 2.12 (ddd, *J* = 13.6, 5.9, 2.0 Hz, 1H), 2.07-1.98 (m, 3H), 1.90 (dddd, *J* = 14.7, 5.8, 2.6, 1.0 Hz, 1H), 1.84 (app dt, *J* = 13.6, 1.7 Hz, 1H), 1.78-1.70 (m, 2H), 1.48 (s, 3H), 1.14 (s, 3H); ¹³C NMR (151 MHz, CDCl₃) δ 173.4, 147.8, 83.6, 82.7, 79.0, 66.8, 64.6, 39.3, 38.3, 33.4, 26.4, 20.4, 19.14, 19.10; IR (neat) 2984, 2941, 1747, 1732, 1444, 1388, 1274, 1252, 1200, 1116, 1100, 1070, 1049 cm⁻¹; HRMS (EI): *m/z* calcd for C₁₄H₂₀O₆ (M⁺⁺) 284.1260, found 284.1254; [α]_D = +17.2 (CHCl₃, *c* 0.52).

Key NOESY enhancements observed in lactone 1.94:



(*S*)-4-((2*R*,5*S*)-tetrahydro-5-((*R*)-tetrahydro-6-methoxy-2*H*-pyran-2-yl)-2-methylfuran-2yl)-1,3-dioxolan-2-one (1.97) and (4a*R*,5a*S*,10a*R*,12a*S*)-9-Methoxy-12a-methyldecahydro-1,3,5,10-tetraoxabenzo[b]heptalen-2-one (1.98)

mL, 5:1, v/v) in borosilicate flask at room temperature were added the activated 4Å molecular sieves (106 mg), anhydrous Na₂S₂O₃ (106 mg,), NaOAc (106 mg) and N-methylquinolinium hexafluorophosphate (3.0 mg, 10.3 µmol). The mixture was photoirradiated with gentle air bubbling for 4 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc (20 mL). The filtrate was concentrated and the resulting residue was purified by flash chromatography (5% - 20% EtOAc in CH₂Cl₂) to provide the exo, exo-product **1.97** (7.3 mg, 24.7%) as a white solid and endo, endoproduct **1.98** (8.8 mg, 29.7%) as a colorless oil. For *exo*, *exo*-product **1.97** (dr = 2:1): ¹H NMR (600 MHz, CDCl₃) δ 4.71 (br s, 67% of 1H), 4.61-4.56 (m, 1H), 4.54-4.50 (m, 1H), 4.44-4.41 (m, 1H), 4.32 (dd, J = 9.5, 2.0 Hz, 33% of 1H), 4.02 (dd, J = 7.1, 4.9 Hz, 33% of 1H), 3.98 (dd, J =7.4, 4.4 Hz, 67% of 1H), 3.72 (ddd, J = 11.6, 4.2, 2.0 Hz, 67% of 1H), 3.48 (s, 33% of 3H), 3.40 (dd, J = 11.3, 4.7, 1.9 Hz, 33% of 1H), 3.33 (s, 67% of 3H), 2.05-1.95 (m, 4H), 1.90-1.78 (m,3H), 1.73-1.65 (m, 1H), 1.60 (s, 3H), 1.55-1.48 (m, 1H), 1.31-1.22 (m, 1H), 1.28 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) & 155.3 (major), 151.1 (minor), 103.7 (minor), 98.7 (major), 82.2 (major, 2C), 81.9 (minor), 79.6 (minor), 79.2 (major), 69.7 (major), 66.0 (major), 56.3 (minor), 54.7 (major), 35.1 (major), 34.7 (minor), 31.3 (minor), 29.9 (major), 27.6 (minor), 27.3 (major), 26.7

(minor), 26.3 (major), 22.0 (minor), 20.9 (minor), 20.5 (major), 17.8 (major); IR (neat) 2943, 1798, 1455, 1374, 1166, 1033, 949 cm⁻¹; HRMS (EI): m/z calcd for C₁₄H₂₂O₆ (M⁺⁺) 286.1416, found 286.1419; $[\alpha]_D = -24.1$ (CHCl₃, c 0.71). For endo, endo-product **1.98** (dr = 2.3:1): ¹H NMR (600 MHz, CDCl₃) δ 4.56 (dd, J = 8.8, 5.8 Hz, 70% of 1H), 4.49-4.46 (m, 30% of 1H), 4.39-4.34 (m, 1H), 4.11 (t, J = 10.6 Hz, 70% of 1H), 4.10 (t, J = 10.6 Hz, 30% of 1H), 3.94 (dd, J = 11.3, 6.5 Hz, 70% of 1H), 3.84 (dd, J = 11.0, 6.3 Hz, 30% of 1H), 3.68 (dt, J = 8.5, 4.5 Hz, 70% of 1H), 3.62-3.59 (m, 30% of 1H), 3.49-3.47 (m, 30% of 1H), 3.42 (s, 30% of 3H), 3.38 (s, 70% of 3H), 3.37-3.33 (70% of 1H), 2.22-1.97 (m, 4H), 1.92-1.78 (m, 2H), 1.65-1.59 (m, 2H), 1.46 (s, 30% of 3H), 1.43 (s, 70% of 3H), 1.38-1.33 (m, 1H), 1.28-1.25 (m, 1H); ¹³C NMR (151 MHz, CDCl₃) δ 149.0 (minor), 148.9 (major), 106.9 (minor), 102.6 (major), 86.4 (major), 83.3 (minor), 82.6 (minor), 82.4 (major), 79.5 (minor), 75.2 (minor), 74.0 (major), 73.2 (minor), 73.1 (major), 66.5 (major), 56.1 (minor), 55.9 (major), 39.5 (minor), 36.7 (major), 35.9 (minor), 35.6 (major), 34.9 (minor), 33.5 (major), 29.7 (major), 28.6 (minor), 28.0 (minor), 21.0 (major), 18.9 (major),17.9 (minor); IR (neat) 2939, 1755, 1455, 1384, 1255, 1205, 1109, 1042, 999 cm⁻¹; HRMS (EI): m/z calcd for $C_{14}H_{22}O_6$ (M^{+•}) 286.1416, found 286.1414; $[\alpha]_D = +11.8$ (CHCl₃, c 0.85).

(4a*R*,5a*S*,10a*R*,12a*S*)-12a-Methyldecahydro-1,3,5,10-tetraoxa-benzo[b]heptalene-2,9-dione (1.99)



To a solution of acetal **1.98** (8.0 mg, 27.9 μ mol) in CH₂Cl₂ (0.5 mL) at 0 °C were added *m*CPBA (pure, 6.2 mg, 36.3 μ mol) and BF₃•OEt₂ (4.2 μ L, 33.5 μ mol) sequentially. After stirred at room temperature for

30 min, the mixture was cooled to 0 °C and Et₃N (19.4 µL, 140 µmol) was added dropwise. The

mixture was stirred at 0 °C for 30 min, then concentrated, and the resulting residue was purified by column chromatography (10% - 20% EtOAc in CH₂Cl₂) to give lactone **1.99** (5.3 mg, 70.2%) as a white crystalline solid: ¹H NMR (600 MHz, CDCl₃) δ 4.43-4.39 (m, 1H), 4.40 (dd, J = 10.4, 6.5 Hz, 1H), 4.13 (dd, J = 11.2, 10.5 Hz, 1H), 3.86 (dd, J = 11.3, 6.5 Hz, 1H), 3.53 (ddd, J = 10.6, 8.0, 3.4 Hz, 1H), 2.70-2.61 (m, 2H), 2.22-2.17 (m, 3H), 2.07-2.01 (m, 2H), 1.92 (ddd, J = 15.4, 9.6, 2.2 Hz, 1H),1.77-1.73 (m, 2H), 1.48 (s, 3H); ¹³C NMR (151 MHz, CDCl₃) δ 173.9, 148.2, 85.5, 81.9, 81.2, 78.4, 66.5, 35.7, 34.5, 33.6, 27.5, 21.0, 19.2; ¹H NMR (500 MHz, C₆D₆) δ 3.58 (dd, J = 10.0, 6.6 Hz, 1H), 3.46 (dd, J = 11.2, 10.2 Hz, 1H), 3.38-3.34 (m, 1H), 2.66 (dd, J =11.2, 6.6 Hz, 1H), 2.58 (ddd, J = 11.2, 7.8, 3.3 Hz, 1H), 2.22-2.18 (m, 1H), 1.73-1.65 (m, 2H), 1.52 (dddd, J = 15.8, 8.8, 3.8, 1.4 Hz, 1H), 1.42-1.38 (m, 1H), 1.30 (ddd, J = 14.4, 8.8, 1.4 Hz, 1H), 1.22 (dddd, J = 17.1, 11.6, 5.2, 1.5 Hz, 1H), 1.16-1.11 (m, 2H), 0.92-0.84 (m, 1H), 0.80 (s, 3H); IR (neat) 2922, 2850, 1747, 1453, 1387, 1273, 1204, 1106, 1058, 1015 cm⁻¹; HRMS (EI): *m/z* calcd for C₁₃H₁₈O₆ (M⁺⁺) 270.1103, found 270.1111; [α]_D = +12.7 (CHCl₃, *c* 0.26).

(*R*)-Tetrahydro-6-((2*S*,5*R*)-tetrahydro-5-methyl-5-((*S*)-2-oxo-1,3-dioxolan-4-yl)furan-2-yl)pyran-2-one (A37)

To a solution of acetal **1.97** (6.8 mg, 23.7 µmol) in CH₂Cl₂ (0.5 $G = (4.0 \ \mu L, 28.4 \ \mu mol)$ at 0 °C were added *m*CPBA acid (pure, 5.3 mg, 30.8 µmol) and BF₃•OEt₂ (4.0 µL, 28.4 µmol) sequentially. After stirred at room temperature for 30 min, the mixture was cooled to 0 °C and Et₃N (16.5 µL, 118 µmol) was added dropwise. The mixture was stirred at 0 °C for 30 min, then concentrated, and the resulting residue was purified by column chromatography (15% - 25% EtOAc in CH₂Cl₂) to give the desired lactone **A37** (5.2 mg, 81.2%) as a white solid: ¹H NMR (500 MHz, CDCl₃) δ 4.64 (dd, J = 8.4, 6.0 Hz, 1H), 4.52 (t, J = 8.8 Hz, 1H), 4.45 (dd, J = 8.8, 6.0 Hz, 1H), 4.30 (dd, J = 11.4, 4.6, 3.0 Hz, 1H), 4.10 (dt, J = 7.2, 4.6 Hz, 1H), 2.62 (dddd, J = 17.8, 6.6, 4.8, 1.4 Hz, 1H), 2.46 (ddd, J = 17.8, 9.3, 7.0 Hz, 1H), 2.19-2.12 (m, 1H), 2.07-2.02 (m, 1H), 2.01-1.93 (m, 3H), 1.90-1.85 (m, 1H), 1.84-1.79 (m, 1H), 1.50-1.44 (m, 1H), 1.28 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 171.1, 155.2, 83.0, 81.1, 80.7, 79.0, 66.1, 34.5, 29.9, 26.4, 24.8, 20.7, 18.5; IR (neat) 2957, 2929, 1789, 1731, 1242, 1173, 1084, 1049, 1018, 771 cm⁻¹; HRMS (EI): *m*/*z* calcd for C₁₃H₁₈O₆ (M⁺⁺) 270.1103, found 270.1104; [α]_D = -42.8 (CHCl₃, *c* 0.50).

































































































































APPENDIX B

EFFORTS TOWARDS THE TOTAL SYNTHESIS OF (+)-LACTODEHYDROTHYRSIFEROL AND ITS ANALOGS (SUPPORTING INFORMATION)

General Experimental Proton (¹H NMR) and carbon (¹³C NMR) nuclear magnetic resonance spectra were recorded at ambient temperature on Bruker Avance 300 spectrometer at 300 MHz and 75 MHz or Bruker Avance 500 spectrometer at 500 MHz and 125 MHz if specified. The chemical shifts are given in parts per million (ppm) on the delta (δ) scale. The solvent peak was used as a reference value, for ¹H NMR: CDCl₃ = 7.27 ppm, for ¹³C NMR: CDCl₃ = 77.23. Data are reported as follows: (s = singlet; d = doublet; t = triplet; q = quartet; dd = doublet of doublets; ddd = doublet of doublet of doublets; dt = doublet of triplets; td = triplet of doublets; br = broad). High resolution mass spectra were recorded on a MICROMASS AUTOSPEC (for EI) or WATERS Q-TOF API-US (for ESI) spectrometer. Infrared (IR) spectra were collected on a Mattson Cygnus 100 spectrometer. Samples for IR were prepared as a thin film on a NaCl plate by dissolving the compound in CH₂Cl₂ and then evaporating the CH₂Cl₂. Tetrahydrofuran was distilled from sodium and benzophenone. Activated 4 Å molecular sieves were obtained through drying in oven at 150 °C overnight. Boc₂O and *N*-methylimidole were purchased from Acros and used without further purification. Anhydrous Na₂S₂O₃ was purchased from Aldrich and used as received. Toluene and 1,2-dichloroethane were purchased from Fisher Scientific and dried with 4 Å molecular sieves overnight prior to use. Anhydrous DMF and MeI were purchased from Acros. NiCl₂, CrCl₃, Mn, and anhydrous LiCl were purchased from Aldrich and used without further purification. TMSCl (purchased from Aldrich) was distilled from anhydrous K₂CO₃ before use. Analytical TLC was performed on E. Merck pre-coated (25 mm) silica gel 60F-254 plates. Visualization was done under UV (254 nm). Flash chromatography was done using ICN SiliTech 32-63 60 Å silica gel. Reagent grade ethyl acetate, diethyl ether and hexanes (commercial mixture) were purchased from EM Science and used as is for chromatography. All reactions were performed in flame-dried glassware under nitrogen with magnetic stirring unless otherwise noted.

1-Hydroxy-5-methoxy-6,6-diphenylhexan-2-one (2.11)

(6.9 mL). The mixture was stirred for 5 min and EtOH (0.8 mL) was added. After 20 min, the mixture was filtered through Celite and the residue was washed with acetone (50 mL). The filtrate was concentrated and the residue was azeotroped with acetone (3 x) and purified by column chromatography (35% - 40% EtOAc in hexanes) to give hydroxyl ketone **2.11** (258 mg, 57%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.38-7.18 (m, 10H), 4.29-4.14 (m, 2H), 4.02-3.95 (m, 2H), 3.10 (s, 3H), 3.08-3.06 (m, 1H), 2.58-2.38 (m, 2H), 2.07-1.95 (m, 1H), 1.73-1.64 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 209.7, 142.2, 128.9, 128.8, 128.6, 126.9, 126.7, 82.6,

68.2, 58.1, 56.4, 33.9, 26.3; IR (neat) 3435, 2931, 1718, 1494, 1451, 1102, 748; HRMS (ESI): *m/z* calcd for C₁₉H₂₂O₃K [M+K]⁺ 337.1206, found 337.1176.

5-Methoxy-2-methylene-6,6-diphenylhexan-1-ol (2.12)

To a solution of hydroxyl ketone 2.11 (235 mg, 0.788 mmol) in THF Ph₂CH (7 mL) were added PPh₃ (227 mg, 0.867 mmol), (Ph₃P)₃RhCl (18.2 OMe mg, 19.7 µmol) and ⁱPrOH (0.6 mL, 7.88 mmol) sequentially. After 5 min, TMSCHN₂ (0.63 mL, 1.26 mmol) was added. The yellow mixture was stirred for 18 h, PPh₃ (200 mg, 0.788 mmol) was added followed by TMSCHN₂ (0.63 mL, 1.26 mmol). The reaction was stirred for another 18 h and TBAF (1 M in THF, 3.0 mL, 3.0 mmol) was added. The mixture was stirred for 30 min, then treated with saturated NH₄Cl (15 mL) and extracted with Et₂O (4 x 20 mL). The extract was dried (MgSO₄), filtered and concentrated. The residue was purified column chromatography (20% - 45% EtOAc in hexanes) to give allylic alcohol 2.12 (168 mg, 72%) and unreacted starting material **2.11** (23 mg, 10%). For allylic alcohol **2.12**: ¹H NMR (300 MHz, CDCl₃) δ 7.40-7.18 (m, 10H), 5.00 (br s, 1H), 4.82 (d, J = 0.9 Hz, 1H), 4.04 (d, J = 8.4 Hz, 1H), 4.00 (br s, 2H), 3.97-3.90 (m, 1H), 3.16 (s, 3H), 2.27-2.09 (m, 2H), 1.78-1.52 (m, 3H); ¹³C NMR (75 MHz, CDCl₃) § 149.0, 142.8, 142.4, 129.0, 128.8, 128.7, 128.5, 126.7, 126.5, 109.8, 83.4, 66.1, 58.1, 56.4, 30.6, 28.7; IR (neat) 3397, 2928, 1599, 1494, 1451, 1103, 1030, 898, 745; HRMS (ESI): m/z calcd for C₂₀H₂₄O₂K [M+K]⁺ 335.1413, found 335.1404.

(2-(3-Methoxy-4,4-diphenylbutyl)oxiran-2-yl)methanol (2.13)

Ph₂CH OH (3.0 mL) was added VO(acac)₂ (2.8 mg, 10.6 µmol) followed by

dropwise addition of 'BuOOH (5-6 M in decane, 0.12 mL, ~0.65 mmol). The mixture was heated to 80 °C for 20 min, then cooled to room temperature and saturated Na₂SO₃ (2 mL) was added. The mixture was stirred for 10 min, the treated with saturated NaHCO₃ (10 mL) and extracted with Et₂O (4 x 20 mL). The extract was dried (MgSO₄), filtered and concentrated. The residue was purified column chromatography (30% - 60% EtOAc in hexanes containing 0.5% Et₃N) to give epoxy alcohol **2.13** (148 mg, 90%, dr ~ 1:1) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.39-7.19 (m, 10H), 3.98 (d, *J* = 8.5 Hz, 1H), 3.96-3.90 (m, 1H), 3.65 (ddd, *J* = 12.3, 6.6, 4.4 Hz, 1H), 3.53 (dd, *J* = 12.2, 8.4 Hz, 1H), 3.16/3.15 (s, 3H), 2.81 (dd, *J* = 4.6, 2.2 Hz, 1H), 2.56 (dd, *J* = 4.6, 1.4 Hz, 1H), 1.98-1.56 (m, 4H), 1.56-1.38 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 142.6, 142.3, 128.9, 128.8, 128.6, 128.5, 126.8, 126.6, 83.4, 83.3, 63.2, 62.9, 59.8, 59.7, 58.2, 58.1, 56.3, 56.3, 50.2, 49.7, 27.5, 27.2, 27.0, 26.8; IR (neat) 3432, 2929, 1642, 1600, 1494, 1452, 1101, 747; HRMS (ESI): *m/z* calcd for C₂₀H₂₄O₃Na [M+Na]⁺ 335.1623, found 335.1607.

tert-Butyl 2-(3-methoxy-4,4-diphenylbutyl)oxiranylmethyl carbonate (2.14)



To a solution of epoxy alcohol **2.13** (135 mg, 0.432 mmol) in dry toluene (4.3 mL) at 0 °C were added 1-methylimidazole (34 μ L, 0.432 mmol) and di-*tert*-butyl dicarbonate (188 mg, 0.864 mmol).

The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. After that time, water (2 drops) was added. After 10 min, the mixture was concentrated and the residue was azeotroped with acetone (5 mL), then hexanes (2 x 5 mL) and purified by flash chromatography (10% - 20% EtOAc in hexanes containing 0.5% Et₃N) to give the *tert*-butyl carbonate **2.14** (162 mg, 91%, dr ~ 1:1) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.41-7.19 (m, 10H), 4.12-3.94 (m, 4H), 3.18/3.17 (s, 3H), 2.72 (t, *J* = 4.2 Hz, 1H), 2.58

(dd, J = 4.5, 2.0 Hz, 1H), 1.97-1.58 (m, 4H), 1.51 (s, 9H); ¹³C NMR (75 MHz, CDCl₃) δ 153.4, 153.4, 142.6, 142.3, 142.3, 129.0, 128.9, 128.8, 128.8, 128.6, 128.5, 126.7, 126.7, 126.5, 83.2, 82.6, 68.5, 68.1, 58.0, 57.3, 57.2, 56.2, 56.2, 50.8, 50.3, 27.9, 27.4, 26.9, 26.8, 26.6; IR (neat) 2979, 2933, 1742, 1494, 1453, 1279, 1161, 1102, 743; HRMS (ESI): m/z calcd for $C_{25}H_{32}O_{5}Na [M+Na]^{+} 435.2147$, found 435.2150.

(5S)-8-Methoxy-1,3,7-trioxaspiro[4.5]decan-2-one (2.17)



To epoxide **2.14** (70 mg, 0.17 mmol) in dichloroethane/toluene (5.6 mL, 5:1, v/v) in borosilicate flask at room temperature were added the activated 4Å molecular sieves (140 mg), anhydrous Na₂S₂O₃ (140 mg),

NaOAc (140 mg) and *N*-methylquinolinium hexafluorophosphate (24.6 mg, 85 µmol). The mixture was photoirradiated with gentle air bubbling for 8 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with Et₂O (30 mL). The filtrate was concentrated and the resulting residue was purified by flash chromatography (30% - 60% EtOAc in hexanes) to produce the bicycle **2.17** (25.2 mg, 79%, dr 1.1:1) as a pale yellow oil: ¹H NMR (300 MHz, CDCl₃) δ 4.70 (t, *J* = 2.3 Hz, 0.5H), 4.60 (app dd, *J* = 3.0, 2.3 Hz, 0.5H), 4.50 (d, *J* = 8.6 Hz, 0.5H), 4.20 (dd, *J* = 8.6, 1.4 Hz, 0.5H), 4.10 (d, *J* = 8.8 Hz, 0.5H), 4.06 (dd, *J* = 8.8 Hz, 0.5H), 3.88 (dd, *J* = 11.0, 1.0 Hz, 0.5H), 3.72 (d, *J* = 12.4 Hz, 0.5H), 3.66 (dd, *J* = 12.4, 2.2 Hz, 0.5H), 2.15-2.00 (m, 1H), 1.99-1.90 (m, 1H), 1.84 (dtd, *J* = 12.4, 4.1, 2.4 Hz, 0.5H), 1.76-1.57 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 154.2, 154.2, 97.6, 96.9, 79.4, 78.6, 72.8, 71.7, 64.5, 63.3, 55.5, 55.3, 28.2, 28.1, 27.4, 26.1; IR (neat) 2921, 1804,

1389, 1183, 1125, 1050, 772; HRMS (EI): m/z calcd for C₈H₁₁O₅ (M^{+•}) 187.0606, found 187.0606.

6-(tert-Butyldiphenylsilanyloxy)-2-(tert-butyldiphenylsilanyloxymethyl)hex-1-en-3-ol (2.25)

OTBDPS To a mixture of (2-bromoallyloxy)(*tert*-butyl)diphenylsilane **2.23** (2.764 g, 7.36 mmol), aldehyde **2.24** (1.200 g, 3.68 mmol), CrCl₃ (116 mg, 0.736 mmol), NiCl₂ (95 mg, 0.736 mmol) and Mn (1.011 g, 18.4

mmol) in anhydrous DMF (6 mL, degassed with argon prior to use) was added TMSCl (1.12 mL, 8.83 mmol). After stirring overnight, the reaction was quenched with water (15 mL) and transferred to a beaker. HCl (1 N) was added until all the manganese metal was completely consumed. The mixture was extracted with Et₂O (4 x 40 mL), and the extract was dried (MgSO4) and concentrated. The residue was purified by column chromatography (2% - 10% EtOAc in hexanes) to give allylic alcohol **2.25** (1.948 g, 85%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.71-7.64 (m, 8H), 7.46-7.34 (m, 12H), 5.20 (q, *J* = 1.5 Hz, 1H), 5.12 (s, 1H), 4.30 (d, *J* = 13.6 Hz, 1H), 4.23-4.19 (m, 2H), 3.65 (app t, *J* = 6.0 Hz, 2H), 2.63 (d, *J* = 4.9 Hz, 1H), 1.70-1.51 (m, 4H), 1.07 (s, 9H), 1.04 (s, 9H).

Ethyl (*Z*)-8-(*tert*-butyldiphenylsilanyloxy)-4-(*tert*-butyldiphenylsilanyloxymethyl)oct-4-enoate (2.26)



unreacted triethyl orthoacetate was distilled out under reduced pressure. The residue was purified

by column chromatography (1% - 3% EtOAc in hexanes) to give the ethyl ester **2.26** (1.724 g, 83%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.69-7.60 (m, 8H), 7.43-7.32 (m, 12H), 5.19 (t, *J* =7.3 Hz, 1H), 4.20 (s, 2H), 4.12 (q, *J* = 7.1 Hz, 2H), 3.53 (t, *J* = 6.4 Hz, 2H), 2.51-2.41 (m, 4H), 1.86 (q, *J* = 7.3 Hz, 2H), 1.52-1.43 (m, 2H), 1.23 (t, *J* = 7.4 Hz, 3H), 1.04 (s, 9H), 1.00 (s, 9H).

(Z)-10-(*tert*-Butyldiphenylsilanyloxy)-6-(*tert*-butyldiphenylsilanyloxymethyl)-2-methyldeca-1,6-dien-3-ol (2.27)



DIBAL-H (0.36 mL, 0.36 mmol) was added over 2 min. The mixture was stirred for 15 min and isopropenylmagnesium bromide (0.5 M in THF, 9.8 mL, 4.9 mmol) was added dropwise. The reaction was stirred at -78 °C for 1 h, then warmed to room temperature and quenched with saturated NH₄Cl solution (10 mL) and saturated sodium tartrate solution (25 mL). The mixture was stirred vigorously for 1 h and extracted with Et₂O (3 x 30 mL). The organic extract was dried (MgSO₄) and concentrated. The residue was purified by column chromatography (3% - 12% EtOAc in hexanes) to give secondary alcohol **2.27** (1.333 g, 79%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.70-7.63 (m, 8H), 7.45-7.34 (m, 12H), 5.23 (t, *J* = 7.3 Hz, 1H), 4.97-4.96 (m, 1H), 4.87-4.86 (m, 1H), 4.22 (d, *J* = 2.5 Hz, 2H), 4.08-4.05 (m, 1H), 3.56 (t, *J* = 6.5 Hz, 2H), 2.28-2.21 (m, 2H), 1.91 (q, *J* = 7.2 Hz, 2H), 1.78-1.67 (m, 5H), 1.56-1.49 (m, 2H), 1.06 (s, 9H), 1.02 (s, 9H).

Ethyl (4*E*,8*Z*)-12-(*tert*-butyldiphenylsilanyloxy)-8-(*tert*-butyldiphenylsilanyloxymethyl)-4methyldodeca-4,8-dienoate (2.28)



A mixture of allylic alcohol **2.27** (1.333 g, 1.93 mmol), triethyl orthoacetate (1.4 mL, 7.72 mmol) and propionic acid (7 μ L, 96 μ mol) was heated to

115 °C for 2 h and the unreacted triethyl orthoacetate was distilled out under reduced pressure. The residue was purified by column chromatography (2% - 4% EtOAc in hexanes) to give the ethyl ester **2.28** (1.216 g, 83%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.69-7.60 (m, 8H), 7.41-7.32 (m, 12H), 5.16 (t, *J* = 6.8 Hz, 2H), 4.18 (s, 2H), 4.12 (q, *J* = 7.1 Hz, 2H), 3.55 (t, *J* = 6.4 Hz, 2H), 2.42-2.37 (m, 2H), 2.32-2.28 (m, 2H), 2.19-2.06 (m, 4H), 1.89 (q, *J* = 7.4 Hz, 2H), 1.60 (br s, 3H), 1.54-1.47 (m, 2H), 1.25 (t, *J* = 7.1 Hz, 3H), 1.04 (s, 9H), 1.00 (s, 9H).

(5*E*,9*Z*)-13-(*tert*-Butyldiphenylsilanyloxy)-9-(*tert*-butyldiphenylsilanyloxymethyl)-5-methyl-1-phenyltrideca-5,9-dien-2-ol (2.29)



A solution of ethyl ester **2.28** (600 mg, 0.788 mmol) in CH_2Cl_2 (4 mL) at -78 °C was treated with DIBAL-H (1 M in hexanes, 0.83 mL, 0.83

mmol) over 10 min. After 30 min, DIBAL-H (0.12 mL, 0.12 mmol) was added and the mixture was stirred for 30 min.

In a separate round-bottom flask, a mixture of CuCN (282 mg, 3.15 mmol) and LiCl (294 mg, 6.93 mmol) in THF (10 mL) at -78 °C was treated with BnMgCl (2 M in THF, 1.6 mmol, 3.2 mmol) dropwise. After 1 h, BF₃•OEt₂ (0.28 mL, 2.26 mmol) was added and the mixture was stirred for 5 min. The reaction mixture from the first reaction was cannulated into the second

flask followed by rinse (2 x 0.5 mL THF). The mixture was stirred at -78 °C for 1 h and then at room temperature for 1 h. After that time, the reaction was quenched with saturated NH₄Cl solution (10 mL) /saturated sodium tartrate solution (5 mL) and stirred vigorously for 1 h. The mixture was extracted with Et₂O (4 x 20 mL) and the organic extract was dried (MgSO₄) and concentrated. The residue was purified by column chromatography (3% - 12% EtOAc in hexanes) to give secondary alcohol **2.29** (496 mg, 78%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.70-7.62 (m, 8H), 7.44-7.30 (m, 12H), 7.30-7.20 (m, 5H), 5.21-5.15 (m, 2H), 4.19 (s, 2H), 3.86-3.76 (m, 1H), 3.55 (t, *J* = 6.5 Hz, 2H), 2.84 (dd, *J* = 13.6, 4.4 Hz, 1H), 2.67 (dd, *J* = 13.5, 8.3 Hz, 1H), 2.23-1.98 (m, 2H), 1.60 (s, 3H), 1.66-1.47 (m, 6H), 1.05 (s, 9H), 1.01 (s, 9H).

(Z)-6-((*E*)-7-Methoxy-4-methyl-8-phenyloct-3-enyl)-2,2-13,13-tetramethyl-3,3,12,12tetraphenyl-4,11-dioxa-3,12-disilatetradec-6-ene (B1)



A solution of alcohol **2.29** (486 mg, 0.600 mmol) in DMF (5 mL) at 0 °C was treated with NaH (60% weight in mineral oil, 60 mg, 1.50 mmol).

After 20 min, MeI (0.15 mL, 2.4 mmol) was added and the mixture was stirred overnight at room temperature. After that time, the reaction was quenched with water (20 mL) cautiously and extracted with Et₂O (3 x 20 mL). The organic extract was dried (MgSO₄) and concentrated. The residue was purified by column chromatography (2% - 5% EtOAc in hexanes) to give methyl ether **B1** (431 mg, 87%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.69-7.61 (m, 8H), 7.43-7.30 (m, 12H), 7.29-7.26 (m, 2H), 7.22-7.18 (m, 3H), 5.20-5.10 (m, 2H), 4.19 (s, 2H), 3.55 (t, *J* = 6.5 Hz, 2H), 3.36-3.30 (m, 1H), 3.31 (s, 3H), 2.84 (dd, *J* = 13.7, 6.2 Hz, 1H), 2.72 (dd, *J* =

13.8, 6.2 Hz, 1H), 2.22-1.96 (m, 6H), 1.89 (q, J = 7.3 Hz, 2H), 1.58-1.48 (m, 7H), 1.04 (s, 9H), 1.00 (s, 9H).

(2Z)-2-((E)-7-Methoxy-4-methyl-8-phenyloct-3-enyl)hex-2-ene-1,6-diol (2.30)



To a solution of silvl ether **B1** (409 mg, 0.497 mmol) in THF (5 mL) was added TBAF·H₂O (312 mg, 1.19 ОΗ mmol). The reaction was stirred for 4 h, then concentrated and the residue was purified by column chromatography (50% - 60% EtOAc in hexanes) to give the diol 2.30 (144 mg, 84%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.32-7.19 (m, 5H), 5.28 (t, J = 7.8 Hz, 1H), 5.16-4.99 (m, 1H), 4.17 (s, 2H), 3.63 (t, J = 5.9 Hz, 2H), 3.39-3.35 (m, 1H), 3.32 (s, 3H), 2.86 (dd, J = 13.7, 6.2 Hz, 1H), 2.71 (dd, J = 13.7, 6.3 Hz, 1H), 2.24 (q, J = 7.1 Hz, 2H), 2.15-1.97 (m, 8H), 1.68-1.60 (m, 2H), 1.58-1.49 (m, 5H).

3-((2S,3R)-3-(Hydroxymethyl)-3-((E)-7-methoxy-4-methyl-8-phenyloct-3-enyl)oxiran-2yl)propan-1-ol (B2)



To a mixture of diol **2.30** (138 mg, 0.398 mmol) and activated 4 Å molecular sieves (120 mg) in CH₂Cl₂ (4 mL) at -20 °C was added D--diisopropyl tartrate (8 µL,

48 μmol). After 10 min, Ti(O'Pr)₄ (12 μL, 40μmol) was introduced. The mixture was stirred for 30 min, and ¹BuOOH (5-6 M in decane, 0.22 mL, ~1.2 mmol) was added dropwise. The reaction was stirred for 5 h at -20 °C, and then stored in at -20 °C overnight. Water (0.5 mL) was added the mixture was stirred at 0 °C for 1 h. After that time, 30% NaOH saturated with NaCl (0.5 mL) was added and the mixture was stirred at room temperature for 2.5 h. The mixture was filtered through a 1:1 mixture of MgSO₄/Celite and the residue was washed with CH₂Cl₂ (20 mL). The combined filtrates were concentrated and the residue was purified by column chromatography (60% - 80% EtOAc in hexanes containing 1% Et₃N) to give monoepoxy diol **B2** (47 mg, 33%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.32-7.19 (m, 5H), 5.11 (t, *J* = 6.8 Hz, 1H), 3.78-3.65 (m, 4H), 3.39-3.34 (m, 1H), 3.32 (s, 3H), 2.90-2.83 (m, 2H), 2.70 (ddd, *J*=13.8, 6.3, 2.5 Hz, 1H), 2.38 (br s, 1H), 2.16-1.95 (m, 6H), 1.86-1.49 (m, 10H).

3-((2*S*,3*R*)-3-(Hydroxymethyl)-3-(2-((2*R*,3*R*)-3-(3-methoxy-4-phenylbutyl)-3-methyloxiran-2-yl)ethyl)oxiran-2-yl)propan-1-ol (B3)



A solution of monoepoxide **B2** (45 mg, 0.124 mmol) in CH₃CN/DMM (1.8 mL, 1:2, v/v) was treated with 0.05 M solution of Na₂B₄O₇ in $4x10^{-4}$ M Na₂(EDTA) (1.2

mL), Bu₄NHSO₄ (1.7 mg, 4.5 μ mol) and Shi ketone (16.0 mg, 62 μ mol) sequentially. The mixture was cooled to -5 °C. Oxone (122 mg, 0.198 mmol), dissolved in 4x10⁻⁴ M Na₂(EDTA) (0.8 mL), and K₂CO₃ (115 mg, 0.831 mmol), dissolved in water (0.8 mL), were added simultaneously via a syringe pump over 2.0 h. After the addition was completed, the blue mixture was stirred further for 10 min, and anhydrous Na₂SO₄ was added in portions until all the water disappeared. The mixture was filtered and the residues was washed with CH₂Cl₂ (30 mL). The combined filtrates were concentrated and the residue was purified by flash chromatography (80% - 100% EtOAc in hexanes containing 1% Et₃N) to give diepoxide **B3** (42 mg, 89%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.30-7.18 (m, 5H), 3.90-3.49 (m, 4H), 3.38-3.34 (m, 1H), 3.32 (s, 3H), 2.91-2.82 (m, 2H), 2.76-2.64 (m, 2H), 2.20-1.41 (m, 14H), 1.22/1.22/1.21 (s, 3H).

3-((2S,3R)-3-tert-butoxycarbonyloxymethyl-3-(2-[3-(3-methoxy-4-phenylbutyl)*tert*-Butyl (2R,3R)-3-methyloxiranyl]ethyl)oxiranyl)propyl carbonate (ent-2.21)



A solution of the diepoxide **B3** (39 mg, 0.103 mmol) in dry toluene (1.0 mL) at 0 °C was treated with Nmethylimidazole (16 µL, 0.206 mmol) followed by di-

tert-butyl dicarbonate (90 mg, 0.412 mmol). The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. After that time, the reaction was quenched with water (5 mL) and extracted with CH₂Cl₂ (3 x 15 mL). The organic extracts were dried (MgSO₄), filtered and concentrated. The residue was purified by flash chromatography (15% - 25% EtOAc in hexanes containing 1% Et₃N) to give the tert-butyl carbonate ent-2.21 (31 mg, 52%, dr \sim 2:1 regarding the stereochemical outcomes of the two epoxide groups) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.32-7.18 (m, 5H), 4.20-4.05 (m, 4H), 3.36-3.34 (m, 1H), 3.31 (s, 3H), 2.90-2.81 (m, 2H), 2.72-2.63 (m,, 2H), 1.90-1.70 (m, 6H), 1.66-1.56 (m, 5H), 1.54-1.43 (m, 1H), 1.49 (s, 9H), 1.20/1.20/1.19 (s, 3H).

tert-Butyl 3-[8-((*R*)-2-methyl-5-oxotetrahydrofuran-2-yl)-(5*S*,6*S*,8*S*)-2-oxo-1,3,7-trioxaspiro[4.5]dec-6-yl]propyl carbonate (2.32)



51.8 Diepoxide ent-2.21 (30 mg, µmol) in dichloroethane/toluene (1.7 mL, 5:1, v/v) in borosilicate flask at room temperature was treated with the activated 4Å molecular sieves (60 mg), anhydrous Na₂S₂O₃ (60 mg), NaOAc (60 mg) and Nmethylquinolinium hexafluorophosphate (7.5 mg, 26 µmol). The mixture was photoirradiated with gentle air bubbling for 3.5 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with Et₂O (25 mL). The filtrate was concentrated and the resulting residue was purified by column chromatography (25% - 40% EtOAc in hexanes) to provide a pale yellow oil **2.31** (9.5 mg, 43%, containing small amounts of unknown materials) and another pale yellow oil (3.0 mg, 13%). For the major product **2.31**: IR (neat) 2976, 1809, 1739, 1280, 1163, 1067.

The acetal **2.31** (3.2 mg) was dissolved in acetone (0.3 mL) at 0 °C and Jones reagent (16 μ L) was added. The mixture was stirred for 1 h, then purified by column chromatography (50% - 80% EtOAc in hexanes) to give title lactone **2.33** (2.6 mg, 84%) as a colorless oil: ¹H NMR (500 MHz, CDCl₃) δ 4.54 (d, *J* = 8.9 Hz, 1H), 3.41 (d, *J* = 9.0 Hz, 1H), 4.08 (t, *J* = 5.8 Hz, 2H), 3.50 (dd, *J* = 10.6, 1.4 Hz, 1H), 3.46 (dd, *J* = 11.6, 2.0 Hz, 1H), 2.62-2.58 (m, 2H), 2.24 (ddd, *J* = 13.1, 9.6, 6.9Hz, 1H), 2.18 (td, *J* = 12.8, 3.8 Hz, 1H), 2.10 (dt, *J* = 13.4, 4.4 Hz, 1H), 1.96-1.91 (m, 2H), 1.90-1.86 (m, 1H), 1.74-1.70 (m, 2H), 1.53-1.46 (m, 1H), 1.49 (s, 9H), 1.39-1.33 (m, 1H), 1.35 (s, 3H).

(5*S*,6*S*,8*S*)-6-(3-Hydroxypropyl)-8-((*R*)-2-methyl-5-oxotetrahydrofuran-2-yl)-1,3,7trioxaspiro[4.5]decan-2-one (2.35)



A solution of lactone **2.33** (7.2 mg, 17.4 μ mol) in CH₂Cl₂ (0.8 mL) at 0 °C was treated with 2,6-lutidine (7.1 μ L, 61 μ mol) followed by TMSOTf (10 μ L, 52 μ mol). The reaction was stirred

at 0 °C for 1 h, and then quenched with saturated NaHCO₃ (0.5 mL). Anhydrous Na₂SO₄ was added and the mixture was filtered. The residue was washed with EtOAc (40 mL) and the filtrate was concentrated. The resulting residue was purified by column chromatography (70% - 100%)

EtOAc in hexanes) to give the alcohol **2.35** (5.4 mg, 98%) as a white solid: ¹H NMR (500 MHz, CDCl₃) δ 4.54 (d, *J* = 8.8 Hz, 1H), 4.11 (d, *J* = 8.8Hz, 1H), 3.67 (br s, 2H), 3.54 (dd, *J* = 11.9, 1.8 Hz, 1H), 3.50 (dd, *J* = 11.8, 2.2 Hz, 1H), 2.68-2.54 (m, 2H), 2.24 (ddd, *J* = 13.1, 10.2, 5.5 Hz, 1H), 2.19 (ddd, *J* = 3.0, 4.2, 3.0 Hz, 1H), 2.11 (dt, *J* = 13.4, 4.6 Hz, 1H), 1.94-1.88 (m, 2H), 1.78-1.69 (m, 2H), 1.66-1.60 (m, 2H), 1.66-1.60 (m, 1H), 1.53-1.48 (m, 1H), 1.40-1.32 (m, 1H), 1.37 (s, 3H).

3-(5*S*,6*S*,8*S*)-[8-((*R*)-2-Methyl-5-oxotetrahydrofuran-2-yl)-2-oxo-1,3,7-trioxaspiro[4.5]dec-6-yl]propionaldehyde (*ent*-2.19)



A solution of alcohol **2.35** (5.2 mg, 16 μ mol) in CH₂Cl₂ (0.5 mL) was treated with NaHCO₃ (5.5 mg, 66 μ mol) and Dess-Martin periodinane (10.5 mg, 25 μ mol) sequentially. The mixture was stirred for 30 min, then loaded onto column and purified (60% -

80% EtOAc in hexanes) to give the aldehyde *ent*-**2.19** (3.2 mg, 61%) as a colorless oil: ¹H NMR (500 MHz, CDCl₃) δ 9.77 (s, 1H), 4.57 (d, J = 8.9 Hz, 1H), 4.13 (d, J = 9.0 Hz, 1H), 3.52 (dd, J = 11.1, 2.4 Hz, 1H), 3.47 (dd, J = 11.8, 2.2 Hz, 1H), 2.67-2.54 (m, 4H), 2.23-2.17 (m, 2H), 2.10 (dt, J = 8.6, 4.6 Hz, 1H), 1.99-1.86 (m, 3H), 1.84-1.76 (m, 1H), 1.39-1.32 (m, 2H), 1.34 (s, 3H).

2-(tert-Butyldiphenylsilanyloxymethyl)hepta-1,6-dien-3-ol (2.38)

OTBDPS A solution of methyl ester 2.37 (2.050 g, 12.0 mmol) in Et₂O (24 mL) at 0 °C was treated with LiAlH₄ (1 M in Et₂O, 12.0 mL) dropwise over 20 min. The reaction was stirred at 0 °C for 1 h and then quenched with saturated NH₄Cl

(20 mL) cautiously. Saturated sodium tartrate solution (40 mL) was added and the mixture was
stirred vigorously for 2 h. The mixture was extracted with EtOAc (3 x 60 mL) and the organic extract was dried (MgSO₄), filtered and concentrated. The crude product was purified by column chromatography (40% - 60% EtOAc in hexanes) to give a colorless oil (1.033 g). This oil was separated into two parts (325 mg and 708 mg). The first part (325 mg) was dissolved in CH₂Cl₂ (10 mL) and treated with TBDPSCl (0.47 mL, 1.83 mmol) and imidazole (155 mg, 2.28 mmol) at 0 °C. The mixture was stirred at 0 °C for 15 min, then quenched with water (20 mL) and extracted with CH₂Cl₂ (3 x 25 mL). The organic extract was dried (MgSO₄) and concentrated. The crude product was purified by column chromatography (5% - 10% EtOAc in hexanes) to give desired silyl ether. The second part (708 mg) was treated in a similar manner with TBDPSCl (1.15 mL, 4.48 mmol) and imidazole (373 mg, 5.48 mmol) in CH₂Cl₂ (10 mL). The combined silyl ether **2.38** (1.180g, 26%, two steps) was obtained as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.71-7.69 (m, 4H), 7.48-7.38 (m, 6H), 5.89-5.75 (m, 1H), 5.18 (q, *J* = 1.4 Hz, 1H), 5.12 (s, 1H), 5.06-4.96 (m, 2H), 4.32 (d, *J* = 13.5 Hz, 1H), 4.24-4.18 (m, 2H), 2.25 (d, *J* = 4.9 Hz, 1H), 2.20-2.01 (m, 2H), 1.71-1.63 (m, 2H), 1.08 (s, 9H).

Ethyl (Z)-4-(*tert*-butyldiphenylsilanyloxymethyl)nona-4,8-dienoate (B4)



A mixture of allylic alcohol **2.38** (1.180 g, 3.10 mmol), triethyl orthoacetate (2.3 mL, 12.4 mmol) and propionic acid (11 μ L, 0.16 mmol) was heated to 100 °C for 6 h and the unreacted triethyl

orthoacetate was removed by distillation under reduced pressure. The residue was purified by column chromatography (1% - 4% EtOAc in hexanes) to give ethyl ester **B4** (0.903 g, 65%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.71-7.68 (m, 4H), 7.47-7.36 (m, 6H), 5.76-5.63 (m,

1H), 5.24 (t, *J* = 7.1 Hz, 1H), 4.97-4.89 (m, 2H), 4.21 (s, 2H), 4.14 (q, *J* = 7.1 Hz, 2H), 2.56-2.45 (m, 4H), 2.01-1.85 (m, 4H), 1.26 (t, *J* = 7.1 Hz, 3H), 1.05 (s, 9H).

(Z)-6-(tert-Butyldiphenylsilanyloxymethyl)-2-methylundeca-1,6,10-trien-3-ol (2.39)



A solution of the ethyl ester **B4** (897 mg, 1.99 mmol) in CH₂Cl₂ (6 mL) at -78 °C was treated with DIBAL-H (1 M in hexanes, 2.10 mL, 2.1 mmol) over 30 min. After 40 min, DIBAL-H (0.30 mL,

0.30 mmol) was added over 5 min. The mixture was stirred for 15 min and isopropenylmagnesium bromide (0.5 M in THF, 6.0 mL, 3.0 mmol) was added dropwise. The reaction was stirred at -78 °C for 15 min, then warmed to room temperature and quenched with saturated NH₄Cl solution (10 mL) and saturated sodium tartrate solution (8 mL). The mixture was stirred vigorously for 1 h and extracted with Et₂O (3 x 30 mL). The organic extract was dried (MgSO₄) and concentrated. The residue was purified by column chromatography (3% - 10% EtOAc in hexanes) to give secondary alcohol **2.39** (678 mg, 76%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.71-7.68 (m, 4H), 7.47-7.36 (m, 6H), 5.76-5.65 (m, 1H), 5.26 (t, *J* = 7.0 Hz, 1H), 4.99-4.90 (m, 3H), 4.87-4.85 (m, 1H), 4.24-4.16 (m, 2H), 4.11-4.06 (m, 1H), 2.29-2.22 (m, 2H), 2.04-1.89 (m, 4H), 1.79-1.66 (m, 2H), 1.74 (s, 3H), 1.62 (d, *J* = 3.9 Hz, 1H), 1.06 (s, 9H).

Ethyl (4*E*,8*Z*)-8-(*tert*-butyldiphenylsilanyloxymethyl)-4-methyltrideca-4,8,12-trienoate (B5)



A mixture of allylic alcohol **2.39** (670 mg, 1.49 mmol), triethyl orthoacetate (1.1 mL, 6.0 mmol) and propionic acid (5.5 μ L, 74 μ mol) was heated to 135 °C for 2 h and then purified by column chromatography (1% - 3% EtOAc in hexanes) to give ethyl ester **B5** (594 mg, 77%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.71-7.68 (m, 4H), 7.46-7.36 (m, 6H), 5.76-5.65 (m, 1H), 5.23-5.15 (m, 2H), 5.01-4.91 (m, 2H), 4.19 (s, 2H), 4.13 (q, *J* = 7.2 Hz, 2H), 2.43-2.37 (m, 2H), 2.35-2.28 (m, 2H), 2.22-2.10 (m, 4H), 2.03-1.88 (m, 4H), 1.61 (s, 3H), 1.26 (t, *J* = 7.1 Hz, 3H), 1.05 (s, 9H).

(5*E*,9*Z*)-9-(*tert*-Butyldiphenylsilanyloxymethyl)-5-methyl-1-phenyltetradeca-5,9,13-trien-2ol (B6)



A solution of the ethyl ester **B5** (588 mg, 1.13 mmol) in CH_2Cl_2 (5.6 mL) at - 78 °C was added DIBAL-H (1 M in hexanes, 1.2 mL, 1.2 mmol) over 15 min. After 30 min,

DIBAL-H (0.20 mL, 0.20 mmol) was added over 5 min. The mixture was stirred for 20 min.

In a separate round-bottom flask, to a mixture of CuCN (405 mg, 4.52 mmol) and LiCl (422 mg, 9.94 mmol) in THF (14 mL) at -78 °C was added BnMgCl (2 M in THF, 2.3 mmol, 4.6 mmol) dropwise over 15 min. After 1 h, BF₃•OEt₂ (0.28 mL, 2.26 mmol) was added and the mixture was stirred for 5 min. The reaction mixture from the first reaction was cannulated to the second flask followed by rinse (2 x 1 mL THF). The mixture was stirred at -78 °C for 1 h and then at room temperature for 2 h. After that time, the reaction was quenched with saturated NH₄Cl solution (10 mL)/saturated sodium tartrate solution (8 mL) and stirred vigorously for 1 h. The mixture was extracted with Et₂O (3 x 30 mL) and the organic extract was dried (MgSO₄) and concentrated. The residue was purified by column chromatography (3% - 12% EtOAc in hexanes) to give secondary alcohol **B6** (520 mg, 81%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.73-7.70 (m, 4H), 7.48-7.37 (m, 6H), 7.35-7.22 (m, 5H), 5.77-5.66 (m, 1H), 5.24-5.19

(m, 2H), 4.99-4.91 (m, 2H), 4.20 (s, 2H), 3.87-3.78 (m, 1H), 2.84 (dd, *J* = 13.5, 4.4 Hz, 1H), 2.68 (dd, *J* = 13.5, 8.3 Hz, 1H), 2.24-2.08 (m, 6H), 2.06-1.88 (m, 4H), 1.62 (s, 3H), 1.66-1.58 (m, 2H), 1.07 (s, 9H).

((2Z,5E) - 9 - Methoxy - 6 - methyl - 2 - (pent - 4 - enylidene) - 10 - phenyldec - 5 - enyloxy) (tert - 1

butyl)diphenylsilane (B7)



A solution of the secondary alcohol **B6** (510 mg, 0.900 mmol) in DMF (8 mL) at 0 °C was treated with NaH (60% weight in mineral oil, 90 mg, 2.25 mmol). After 30 min,

MeI (0.22 mL, 3.60 mmol) was added and the mixture was stirred overnight at room temperature. After that time, the reaction was quenched with water (20 mL) cautiously and extracted with Et₂O (3 x 30 mL) and the organic extract was dried (MgSO₄) and concentrated. The residue was purified by column chromatography (1% - 4% EtOAc in hexanes) to give the methyl ether **B7** (474 mg, 91%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.72-7.66 (m, 4H), 7.47-7.36 (m, 6H), 7.32-7.25 (m, 2H), 7.23-7.18 (m, 3H), 5.79-5.66 (m, 1H), 5.21 (app t, *J* = 7.0 Hz, 1H), 5.14 (app qt, *J* = 6.8, 1.0 Hz, 1H), 4.99-4.91 (m, 2H), 4.20 (s, 2H), 3.39-3.32 (m, 1H), 3.32 (s, 3H), 2.86 (dd, *J* = 13.7, 6.1 Hz, 1H), 2.72 (dd, *J* = 13.7, 6.2 Hz, 1H), 2.25-2.07 (m, 5H), 2.04-1.88 (m, 5H), 1.59-1.51 (m, 2H), 1.54 (br s, 3H), 1.06 (s, 9H).

(2Z,5E)-9-Methoxy-6-methyl-2-(pent-4-enylidene)-10-phenyldec-5-en-1-ol (2.40)



A solution of the silyl ether **B7** (464 mg, 0.799 mmol) in THF (5 mL) was treated with TBAF (1 M in THF, 1.9 mL, 1.9 mmol). The reaction was stirred for 4 h, then concentrated and the residue was purified by column chromatography (10% - 22% EtOAc in hexanes) to give the trienol **2.40** (230 mg, 84%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.32-7.28 (m, 2H), 7.23-7.18 (m, 3H), 5.82 (tdd, J = 16.8, 10.2, 6.5 Hz, 1H), 5.31 (t, J = 6.8 Hz, 1H), 5.14-5.10 (m, 1H), 5.06-4.98 (m, 2H), 4.12 (d, J = 5.6 Hz, 2H), 3.38-3.30 (m, 1H), 3.32 (s, 3H), 2.86 (dd, J = 13.7, 6.1 Hz, 1H), 2.72 (dd, J = 13.7, 6.2 Hz, 1H), 2.23-1.96 (m, 10H), 1.57-1.50 (m, 2H), 1.55 (s, 3H), 1.15 (t, J = 5.7 Hz, 1H).

((2S,3R)-3-(But-3-enyl)-2-((E)-7-methoxy-4-methyl-8-phenyloct-3-enyl)oxiran-2-

yl)methanol (B8)



A mixture of trienol **2.40** (223 mg, 0.651 mmol) and activated 4 Å molecular sieves (195 mg) in CH_2Cl_2 (6 mL) at -20 °C was treated with L-diisopropyl tartrate (16 μ L,

78 µmol). After 15 min, Ti(O^{*i*}Pr)₄ (20 µL, 65µmol) was introduced. The mixture was stirred for 30 min, and 'BuOOH (5-6 M in decane, 0.36 mL, ~1.9 mmol) was added dropwise. The reaction was stirred for 40 min at -20 °C, and then water (0.3 mL) was added. The mixture was stirred at 0 °C for 1 h. After that time, 30% NaOH saturated with NaCl (0.3 mL) was added and the mixture was stirred at room temperature for 4 h. The mixture was filtered through a 1:1 mixture of MgSO₄/Celite and the residue was washed with CH₂Cl₂ (30 mL). The combined filtrates were concentrated and the residue was purified by column chromatography (20% - 25% EtOAc in hexanes containing 0.5% Et₃N) to give a colorless oil (150 mg, containing 7 mol% (+)-DIPT) which was treated with a mixture of 30% NaOH saturated with NaCl (0.3 mL) and CH₂Cl₂ (3 mL) for 5 h. The mixture was concentrated and the residue was purified by column chromatography (20% - 25% EtOAc in hexanes containing 0.5% Et₃N) to give a colorless oil (150 mg, containing 7 mol% (+)-DIPT) which was treated with a mixture of 30% NaOH saturated with NaCl (0.3 mL) and CH₂Cl₂ (3 mL) for 5 h. The mixture was concentrated and the residue was purified by column

B8 (134 mg, 58%) as a colorless oil: ¹H NMR (500 MHz, CDCl₃) δ 7.31-7.20 (m, 5H), 5.84 (tdd, *J* = 16.9, 10.2, 6.6 Hz, 1H), 5.10-5.02 (m, 3H), 3.76 (dd, *J* = 11.8, 6.8 Hz, 1H), 3.67 (dd, *J* = 11.8, 5.3 Hz, 1H), 3.36-3.34 (m, 1H), 3.32 (s, 3H), 2.90-2.84 (m, 2H), 2.71 (dd, *J* = 13.7, 6.2 Hz, 1H), 2.28-1.99 (m, 6H), 1.94-1.90 (m, 1H), 1.78-1.45 (m, 6H), 1.55 (s, 3H).

((2S,3R)-3-(But-3-enyl)-2-(2-((2S,3S)-3-(3-methoxy-4-phenylbutyl)-3-methyloxiran-2-

yl)ethyl)oxiran-2-yl)methanol (B9)



A solution of **B8** (125 mg, 0.349 mmol) in CH₃CN/DMM (5.2 mL, 1:2, v/v) was treated with a 0.05 M solution of Na₂B₄O₇ in $4x10^{-4}$ M Na₂(EDTA) (3.5 mL), Bu₄NHSO₄

(4.7 mg, 14.0 µmol) and *ent*-Shi ketone (27.0 mg, 0.105 mmol) sequentially. The mixture was cooled to - 5 °C. Oxone (268 mg, 0.436 mmol), dissolved in 4x10⁻⁴ M Na₂(EDTA) (2.0 mL), and K₂CO₃ (253 mg, 1.83 mmol), dissolved in water (2.0 mL), were added simultaneously via a syringe pump over 2.0 h. After the addition was completed, the slightly blue reaction mixture was stirred further for 15 min, then diluted with water (10 mL) and extracted with CH₂Cl₂ (4 x 15 mL). The organic extracts were dried (MgSO₄), filtered and concentrated. The residue was purified by flash chromatography (25% - 38% EtOAc in hexanes) to give unreacted starting material **B8** (37.4 mg, 30%) and the diepoxide **B9** (79.7 mg, 61%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.32-7.19 (m, 5H), 5.91-5.78 (m, 1H), 5.11-5.01 (m, 2H), 3.73-3.71 (m, 2H), 3.38-3.34 (m, 1H), 3.32 (s, 3H), 2.92-2.83 (m, 2H), 2.71-2.65 (m, 1H), 2.32-2.18 (m, 2H), 2.03-1.91 (m, 1H), 1.87 (dt, *J* = 5.9, 2.2 Hz, 1H), 1.81-1.41 (m, 8H), 1.22 (s, 3H).

tert-Butyl ((2*S*,3*R*)-3-(but-3-enyl)-2-(2-((2*S*,3*S*)-3-(3-methoxy-4-phenylbutyl)-3-methyloxiran-2-yl)ethyl)oxiran-2-yl)methyl carbonate (2.36)



A solution of the diepoxy alcohol **B9** (78 mg, 0.208 mmol) in dry toluene (2.0 mL) at 0 °C was treated with di*tert*-butyl dicarbonate (91 mg, 0.416 mmol) and 1-

methylimidazole (16 µL, 0.208 mmol). The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. After that time, the reaction was quenched with water (10 mL) and extracted with CH₂Cl₂ (3 x 15 mL). The organic extracts were dried (MgSO₄), filtered and concentrated. The residue was purified by flash chromatography (15% - 20% EtOAc in hexanes containing 0.5% Et₃N) to give the *tert*-butyl carbonate **2.36** (90 mg, 91%, dr ~ 4.6:1 regarding the stereochemical outcomes of the two epoxide groups) as a colorless oil: ¹H NMR (500 MHz, CDCl₃) δ 7.31-7.28 (m, 2H), 7.23-7.19 (m, 3H), 5.85-5.80 (m, 1H), 5.08 (td, *J* = 17.1, 1.6 Hz, 1H), 5.02 (dd, *J* = 10.4, 1.2 Hz, 1H), 4.21-4.12 (m, 2H), 3.38-3.33 (m, 1H), 3.31 (s, 3H), 2.89-2.83 (m, 2H), 2.70-2.65 (m, 2H), 2.29-2.21 (m, 2H), 1.85-1.81 (m, 1H), 1.78-1.60 (m, 7H), 1.54-1.42 (m, 2H), 1.45 (s, 9H), 1.21/1.20 (s, 3H).

(5R,6R,8R)-6-But-3-enyl-8-((S)-2-methyl-5-oxo-tetrahydrofuran-2-yl)-1,3,7-

trioxaspiro[4.5]decan-2-one (B10)



Diepoxide **2.36** (44 mg, 92.7 μ mol) in dichloroethane/toluene (3.1 mL, 5:1, v/v) in borosilicate flask at room temperature was treated with activated 4Å molecular sieves (88 mg), anhydrous Na₂S₂O₃

(88 mg), NaOAc (88 mg) and *N*-methylquinolinium hexafluorophosphate (13.4 mg, 46.4 μ mol). The mixture was photoirradiated with gentle air bubbling for 6.5 h while stirring at room

temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc (20 mL). The filtrate was concentrated and the resulting residue was purified by flash chromatography (25% - 35% EtOAc in hexanes) to provide a pale yellow oil (10.5 mg), which was dissolved in acetone (1.0 mL) at 0 °C. Jones reagent (2.67 M, 6 drops) was added dropwise (1 drop/5 min). After completion of addition, the mixture was stirred for 1 h and then concentrated. The residues was purified by column chromatography (50% - 70% EtOAc in hexanes) to give the lactone **B10** (4.6 mg, 17%, two steps) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 5.83-5.69 (m, 1H), 5.11-5.01 (m, 2H), 4.54 (d, *J* = 8.8 Hz, 1H), 4.12 (d, *J* = 9.0 Hz, 1H), 3.50 (dd, *J* = 10.8, 2.0 Hz, 1H), 3.46 (dd, *J* = 11.7, 2.3 Hz, 1H), 2.64-2.57 (m, 2H), 2.32-2.05 (m, 5H), 1.97-1.87 (m, 2H), 1.69 (dtd, *J* = 13.6, 8.1, 2.2 Hz, 1H), 1.49 (dtd, *J* = 13.6, 5.7, 2.9 Hz, 1H), 1.40-1.32 (m, 1H), 1.36 (s, 3H).

3-((5*R*,6*R*,8*R*)-[8-((*S*)-2-Methyl-5-oxotetrahydrofuran-2-yl)-2-oxo-1,3,7-trioxaspiro[4.5]dec-6-yl])propionaldehyde (2.19)

A solution of the lactone **B10** (3.3 mg, 10.6 µmol) in CH₂Cl₂ (0.8 mL) at - 78 °C was treated dropwise with a saturated solution of O₃ in CH₂Cl₂ at -78 °C until all the starting material disappeared. PPh₃ (a crystal) was added and the cold bath was removed. The reaction was stirred for 10 h, then concentrated and the residues was purified by column chromatography (60% - 80% EtOAc in hexanes) to give the aldehyde **2.19** (3.6 mg (containing ~17% Ph₃P=O, 83%) as a colorless oil, which has identical spectral data to *ent*-**2.19**.

(S)-1-((2R,5R)-Tetrahydro-5-(2-hydroxypropan-2-yl)-2-methylfuran-2-yl)ethane-1,2-diol (2.42)

A mixture of AD-mix- β (12.35 g) in ^tBuOH/H₂O (88 mL, 1:1, v/v) at 0 $^{\circ}$ OH $^{\circ}$ C was treated with CH₃SO₂NH₂ (0.839 g, 8.82 mmol). After 15 min, HO the epoxy alcohol (1.50 g, 8.82 mmol, prepared from geraniol through Sharpless asymmetric epoxidation) was added dropwise and the flask formerly containing the epoxy alcohol was rinsed with ^tBuOH/H₂O (2 x 2 mL, 1:1, v/v). The mixture was stirred at 0 °C for 2 days, then concentrated to ~50 mL and extracted with EtOAc (25 x 30 mL). The combined organic extracts were dried (Na₂SO₄), filtered and concentrated. The residue was purified by column chromatography (90% - 100% EtOAc in hexanes followed by 5% - 15% MeOH in EtOAc) to give a mixture of 2.22 and 2.42 (1.720 g). This mixture was dissolved in PhMe (200 mL) at 0 °C and CSA pyridine (262 mg, 0.842 mmol) was added. The reaction was stirred for 1.5 h, and Et₃N (1 mL) was added. The solvent was removed under reduced pressure and the residue was purified by column chromatography (90% - 100% EtOAc in hexanes followed by 5% - 15% MeOH in EtOAc) to give triol 2.42 (1.627 g, 90%, two steps) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 3.79 (t, J = 8.2 Hz, 1H), 3.75-3.70 (m, 2H), 3.58-3.52 (m, 1H), 2.88 (br s, 1H), 2.53-2.51 (m, 1H), 2.18 (s, 1H), 2.10 (td, J = 11.3, 10.2 Hz, 1H), 1.90-1.81 (m, 2H), 1.65-1.57(m, 1H), 1.22 (s, 3H), 1.18 (s, 3H), 1.14 (s, 3H); HRMS (ESI): m/z calcd for $C_{10}H_{20}O_4Na$ (M + Na⁺) 227.1259, found 227.1246.

yl]ethyl toluene-4-sulfonate (B11)

A solution of triol **2.42** (1.627 g, 7.96 mmol) in CH₂Cl₂ (20 mL) was treated with pyridine (1.3 mL, 15.9 mmol), TsCl (1.670 g, 8.76 mmol) and DMAP (49 mg, 0.40 mmol) sequentially. The reaction was stirred overnight, then concentrated and the residue was purified by column chromatography (30% - 60% EtOAc in hexanes) to give the tosylate **B11** (1.284 g, 45%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 7.82 (td, *J* = 8.4, 1.9 Hz, 2H), 7.38-7.35 (m, 2H), 4.26 (dd, *J* = 10.4, 2.6 Hz, 1H), 4.02 (dd, *J* = 10.5, 7.8 Hz, 1H), 3.78 (td, *J* = 7.6, 2.9 Hz, 1H), 3.72 (dd, *J* = 8.9, 6.8 Hz, 1H), 2.52 (d, *J* = 3.2 Hz, 1H), 2.46 (s, 3H), 2.06 (td, *J* = 11.9, 9.2 Hz, 1H), 1.96 (s, 1H), 1.89-1.81 (m, 2H), 1.66 (ddd, *J* = 11.9, 6.2, 3.5 Hz, 1H), 1.18 (s, 3H), 1.13 (s, 3H), 1.11 (s, 3H); HRMS (ESI): *m/z* calcd for C₁₇H₂₆O₆NaS [M+Na]⁺ 381.1348, found 381.1340.

2-((2R,5R)-Tetrahydro-5-methyl-5-((S)-oxiran-2-yl)furan-2-yl)propan-2-ol (B12)

A solution of the tosylate **B11** (270 mg, 0.753 mmol) in dry MeOH (21 OH mL) was treated with anhydrous K₂CO₃ (104 mg, 0.753 mmol). The mixture was stirred for 1.5 h, then concentrated and the residue was purified by column chromatography (50% EtOAc in hexanes containing 0.5% Et₃N) to give the epoxide **B12** (134 mg, 96%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 3.81-3.76 (m, 1H), 3.04 (dd, J = 4.1, 2.8 Hz, 1H), 2.74 (dd, J = 5.0, 4.2 Hz, 1H), 2.58 (dd, J = 5.0, 2.8 Hz, 1H), 2.11 (s, 1H), 1.90-1.78 (m, 2H), 1.68-1.58 (m, 1H), 1.28 (s, 3H), 1.23 (s, 3H), 1.13 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 86.8, 81.4, 70.7, 57.2, 43.9, 32.8, 27.5, 26.3, 24.3, 24.3.

(2-((2*R*,5*R*)-Tetrahydro-5-methyl-5-((*S*)-oxiran-2-yl)furan-2-yl)propan-2-yloxy)triethylsilane (2.43)

A solution of the tertiary alcohol **B12** (178 mg, 0.956 mmol) in DMF O_{H}^{+} (OSiEt₃ (5.8 mL) at 0 °C were treated with imidazole (130 mg, 1.91 mmol), TESCI (0.24 mL, 1.43 mmol) and DMAP sequentially. The mixture was stirred at 0 °C for 2 h and then at room temperature overnight. After that time, the reaction was quenched with water (15 mL) at 0 °C and extracted with Et₂O (3 x 20 mL). The organic extract was dried (MgSO₄), filtered and concentrated. The residue was purified by column chromatography (2% - 8% Et₂O in hexanes containing 0.5% Et₃N) to give the silyl ether **2.43** (261 mg, 91%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 3.72 (t, *J* = 7.0 Hz, 1H), 2.99 (dd, *J* = 4.1, 2.8 Hz, 1H), 2.73 (dd, *J* = 4.9, 4.2 Hz, 1H), 2.58 (dd, *J* = 5.1, 2.8 Hz, 1H), 1.98-1.73 (m, 3H), 1.58 (ddd, *J* = 13.7, 8.2, 5.2 Hz, 1H), 1.24 (s, 3H), 1.20 (s, 3H), 1.19 (s, 3H), 0.95 (t, *J* = 8.1 Hz, 9H), 0.58 (q, *J* = 7.8 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 87.7, 81.6, 74.3, 57.4, 44.2, 33.1, 28.1, 26.6, 25.5, 24.0, 7.3, 6.9; IR (neat) 2958, 2876, 1240, 1174, 1044, 726. HRMS (ESI): *m/z* calcd for C₁₆H₃₂O₃NaSi [M+Na]⁺ 323.2018, found 323.1993.

(*R*)-2-Methyl-(5*R*)-5-(1-methyl-1-triethylsilanyloxyethyl)-2-((*S*)-1-triethylsilanyloxypent-4ynyl)tetrahydrofuran (B13)

A mixture of *n*-BuLi (1.6 M in hexanes, 23.2 mL, 37.2 mmol) in H OSiEt₃ Et₂O (8 mL) at -78 °C was treated with tetramethylethylenediamine (1.4 mL, 9.3 mmol) followed by dropwise addition of propargyl bromide (80% weight in PhMe, 2.1 mL, 18.6 mmol). The resulting yellow suspension was stirred at -78 °C for 20 min, then transferred to a solution of epoxide **2.43** (180 mg, 0.6 mmol) in anhydrous Et₂O (1.0 mL) at -78 °C quickly via syringe. The mixture was stirred at -78 °C for 1 h, then quenched with saturated NH₄Cl (20 mL)/water (10 mL) and extracted with Et₂O (3 x 20 mL). The ether solution was dried (MgSO₄), filtered and concentrated. The residue was purified by column chromatography (4% - 12% Et₂O in hexanes) to give starting material **2.43** (65 mg, 36%) and the terminal alkyne **B13** (128 mg, 63%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 3.71-3.61 (m, 2H), 2.49-2.40 (m, 2H), 2.33 (ddd, *J* = 16.8, 8.0, 2.6 Hz, 1H), 2.04 (td, *J* = 11.2, 8.2 Hz, 1H), 1.97 (t, *J* = 2.7 Hz, 1H), 1.96-1.83 (m, 2H), 1.74-1.65 (m, 1H), 1.56-1.44 (m, 2H), 1.22 (s, 3H), 1.18 (s, 3H), 1.13 (s, 3H), 0.96 (t, *J* = 7.7 Hz, 9H), 0.59 (q, *J* = 8.0 Hz, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 88.9, 85.9, 84.7, 75.6, 74.1, 68.6, 31.1, 30.9, 27.7, 26.9, 26.1, 24.0, 16.1, 7.3, 7.0.

(S)-1-[(2R)-2-Methyl-(5R)-5-(1-methyl-1-triethylsilanyloxy-ethyl)-tetrahydro-furan-2-yl]pent-4-yn-1-ol (2.44)

A solution of the secondary alcohol **B13** (120 mg, 0.352 mmol) in CH₂Cl₂ (3.5 mL) at 0 °C was treated with imidazole (27 mg, 0.396 mmol), TESCl (65 μ L, 0.387 mmol) and DMAP (2.1 mg, 17 μ mol) sequentially, and the cold bath was removed. After 1 h, the reaction mixture was concentrated and the residue was purified by column chromatography (1% - 3% Et₂O in hexanes) to give the bis-silyl ether **2.44** (145 mg, 91%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 3.67-3.63 (m, 2H), 2.38-2.29 (m, 1H), 2.23 (ddd, *J* = 16.9, 7.9, 2.6 Hz, 1H), 1.99-1.74 (m, 5H), 1.59-1.47 (m, 2H), 1.19 (s, 3H), 1.17 (s, 3H), 1.09 (s, 3H), 1.00-0.93 (m, 18H), 0.68-0.55 (m, 12H); ¹³C NMR (75 MHz, CDCl₃) δ 87.4, 85.8, 85.2, 76.3, 74.3, 68.4, 34.6, 32.8, 28.1, 26.5, 25.7, 22.7, 15.9, 7.3, 7.3, 7.0, 5.7.

(2*R*)-2-Methyl-(5*R*)-5-(1-methyl-1-triethylsilanyloxyethyl)-2-((1*S*)-4-tributylstannanyl-1-triethylsilanyloxypent-4-enyl)tetrahydrofuran (B14)

A solution of ${}^{i}Pr_{2}NH$ (0.57 mL, 4.05 mmol) in THF (3.5 Bu₃Sn H OSiEt₃ mL) at 0 °C was treated with *n*-BuLi (1.6 M in hexanes, 2.3 mL, 3.6 mmol). After 30 min, the flask was cooled to -30 °C and a solution of *n*-Bu₃SnH (0.97 mL, 3.6 mmol) in THF (2.1 mL) was added dropwise. The pale yellow solution was stirred for 1 h and Et₂AlCl (1 M in heptane, 3.0 mL, 3.0 mmol) was added. The mixture was stirred at -30 °C for 5 h to form an *n*-Bu₃SnAlEt₂ solution (~0.244 M).

A solution of terminal alkyne **2.44** (140 mg, 0.308 mmol) in THF (11 mL) at -30 °C was treated with *n*-Bu₃SnAlEt₂ solution (6.2 mL, ~1.5 mmol) followed by CuCN (8.3 mg, 92.4 µmol). After 1 h, the *n*-Bu₃SnAlEt₂ solution (2.0 mL, ~0.5 mmol) and CuCN (8.0 mg, 89 µmol) were added. The mixture was stirred at -30 °C for 5 h, and then stored at -20 °C overnight. The reaction was quenched with saturated NH₄Cl (20 mL) and extracted with Et₂O (3 x 20 mL). The ether solution was dried (MgSO₄), filtered and concentrated. The residue was purified by column chromatography (2% - 2.5% Et₂O in hexanes) to give the desired vinyl stannane **B14** (67.8 mg, 29%) and unreacted starting material **2.44** (71.9 mg, 51%). For vinyl stannane: ¹H NMR (500 MHz, CDCl₃) δ 5.70 (*J* = 84.6, 0.8 Hz, 1H), 5.10 (*J* = 38.7, 1.5 Hz, 1H), 3.64 (dd, *J* = 5.5, 3.7 Hz, 1H), 3.52 (dd, *J* = 5.0, 1.9 Hz, 1H), 2.42 (dt, *J* = 8.2, 2.8 Hz, 1H), 2.22 (dt, *J* = 7.5, 2.8 Hz, 1H), 1.97 (dt, *J* = 6.5, 4.8 Hz, 1H), 1.92-1.84 (m, 1H), 1.82-1.76 (m, 1H), 1.66 (tt, *J* = 8.0, 2.0 Hz, 1H), 1.57-1.29 (m, 14H), 1.19 (s, 3H), 1.17 (s, 3H), 1.08 (s, 3H), 0.99-0.88 (m, 33H), 0.66-0.56 (m, 12H).

2-((1S)-4-Iodo-1-triethylsilanyloxypent-4-enyl)-(2R)-2-methyl-(5R)-5-(1-methyl-1-

triethylsilanyloxy-ethyl)tetrahydrofuran (2.45)

A solution of the vinyl stannane **B14** (62 mg, 83 µmol) in I = 0 H OSiEt₃ CH₂Cl₂ (1.6 mL) was treated with I₂ (23 mg, 91µmol). The slightly purple solution was stirred for 10 min, then quenched with saturated Na₂S₂O₃ solution (5 mL) and extracted with CH₂Cl₂ (3 x 15 mL). The organic extract was dried (Na₂SO₄), filtered and concentrated. The residue was purified by column chromatography (2% Et₂O in hexanes containing 0.5% Et₃N) to give the vinyl iodide **2.45** (41.7 mg, 86%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) δ 6.04 (q, *J* = 1.4 Hz, 1H), 5.69 (d, *J* = 0.8 Hz, 1H), 3.66-3.61 (m, 1H), 3.54 (dd, *J* = 7.5, 4.1 Hz, 1H), 2.60-2.41 (m, 2H), 1.99-1.73 (m, 4H), 1.60-1.50 (m, 2H), 1.19 (s, 3H), 1.17 (s, 3H), 1.10 (s, 3H), 1.00-0.93 (m, 18H), 0.67-0.54 (m, 12H); ¹³C NMR (75 MHz, CDCl₃) δ 125.2, 113.1, 87.4, 85.9, 76.8, 74.2, 42.9, 34.9, 33.9, 28.0, 26.5, 25.8, 22.6, 7.4, 7.0, 5.7; IR (neat) 2956, 2876, 1459, 1238, 1173, 1099, 1068. HRMS (ESI): *m/z* calcd for C₂₅H₅₁IO₃NaSi [M+Na]⁺ 605.2319, found 605.2341.















GRAME SW08090701 EXPNO PROCNO 1	F2 - Acquisition Parameters Date 20070809 Time 8.46 INSTRUM 8.46 INSTRUM SPECT PULPROG 5 mm QNP 1H/1 PULPROG 65536 SOLVENT CDC13	NS 17985.611 Hz 25 SWH 17985.611 Hz 75 FIDRES 0.274439 Hz 0.274439 Hz 72600 usec 27.800 usec 27.800 usec 27.800 usec 200 usec 200 usec 200 0.0000 0.0000 0.000 0.0000 0.0000 0.000 0.0000 0.0000 0.000 0.000 0.000 0.000	CHANNEL fl ====== NUC1 13C P1 7.00 usec P1 0.00 dB F1 75.4639789 MHz	CPDPRG2 CHANNEL f2 CHANNEL f2 CPDPRG2 waltz16 NUC22 100.00 usec PL22 0.00 dB PL12 18.24 dB PL13 18.24 dB PL13 300.0862003 MHz	P2 - Processing parameters SI 32768 MHz SF 75,4564198 MHz WDW EM SSB 1.00 Hz CB 1.00 Hz CB 1.40 PC 1.40
Carbonate Carbon		OMe 0. 2.14	58		200 180 160 140 120 100 80 60 40 20 0 p


























































Current Data Parameters NAME SW02130804 EXEND NAME SW02130804 EXEND PROCNO 1 F2 - Acquisition Parameters Date 20080213 Time 16.55 INSTRUM 5 mm QNP 1H/1 F012P806 5 mm QNP 1H/1 F012P806 5 mm QNP 1H/1 F17985.611 Hz CDC13 NS D1 17985.611 Hz CDC13 SWH 0.02739508 sec 6 5533 SWH 0.027308 sec 6 5533 SWH 0.000000 sec D1 10.00000000 sec D1 10.0000000 sec D1 0.03000000 sec D1 10.0000000 sec D1 0.03000000 sec D1 10.0000000 sec D1 0.03000000 sec D1 10.0000000 sec	===== CHANNEL fl ======= NUCL 7.00 usec Pl 0.00 dB PL 7.5.4639789 MHZ FPL 7.6.4639789 MHZ EPD 0.00 dB PL 7.5.4639789 MHZ EPD 0.00 dB NUCZ NUCZ NUCZ 100.00 dB PL13 100.00 dB PL13 300.065203 MHZ PL13 300.0652003 MHZ F2 Processing parameters SF02 75.45444 MHZ MDM SF SF 75.456424 MHZ MDM 1.00 MZ D 0 PL13 1.40
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APPENDIX C

MULTICOMPONENT APPROACH TO THE SYNTHESIS OF OXIDIZED AMIDES THROUGH NITRILE HYDROZIRCONATION (SUPPORTING INFORMATION)

General Experimental Proton (¹H NMR) and carbon (¹³C NMR) nuclear magnetic resonance spectra were recorded at ambient temperature on Bruker Avance 300 spectrometer at 300 MHz and 75 MHz or Bruker Avance 500 spectrometer at 500 MHz and 125 MHz if specified. The chemical shifts are given in parts per million (ppm) on the delta (δ) scale. The solvent peak was used as a reference value, for ¹H NMR: CDCl₃ = 7.27 ppm, CD₃OD = 3.31, for ¹³C NMR: CDCl₃ = 77.23, CD₃OD = 49.00. Data are reported as follows: (s = singlet; d = doublet; t = triplet; q = quartet; sept = septet; sext = sextet; dd = doublet of doublets; ddd = doublet of doublet of doublets; dt = doublet of triplets; td = triplet of doublets; dtd = doublet of triplet of doublets; br = broad; app = apparently). High resolution mass spectra were recorded on a MICROMASS AUTOSPEC (for EI) or WATERS Q-TOF API-US (for ESI) spectrometer. Infrared (IR) spectra were collected on a Mattson Cygnus 100 spectrometer. Samples for IR were prepared as a thin film on a NaCl plate by dissolving the compound in CH₂Cl₂ and then evaporating the CH₂Cl₂. Optical rotations were measured on a Perkin-Elmer 241 polarimeter at ambient temperature. Tetrahydrofuran and diethyl ether were distilled from sodium and benzophenone. Methylene chloride and benzene was distilled under N₂ from CaH₂. Anhydrous CH₂Cl₂ was obtained through distillation from CaH₂. PhOCOCl, 'PrCOCl, MeOCH₂COCl, CbzCl, PhSH, Et₃N, and BF₃•OEt₂ were distilled prior to use. MeOH and 'BuOH were distilled from Mg and stored over 4 Å molecular sieves prior to use. PhOH was azeotroped with toluene and dried under high vacuum before use. Methanesulfonic anhydride was purchased from Aldrich and used without further purification. Anhydrous Mg(ClO₄)₂ and Zn(OTf)₂ were purchased from Aldrich and Fluka, respectively, stored in dessicator, and used as received. Analytical TLC was performed on E. Merck pre-coated (25 mm) silica gel 60F-254 plates. Visualization was done under UV (254 nm). Flash chromatography was done using ICN SiliTech 32-63 60 Å silica gel. Reagent grade ethyl acetate, diethyl ether and hexanes (commercial mixture) were purchased from EM Science and used as is for chromatography. All reactions were performed in oven or flame-dried glassware under Ar with magnetic stirring unless otherwise noted. Schwartz' reagent, though commercially available, was prepared according to the literature.¹¹² All the compounds in this work were prepared in their racemic form unless otherwise noted.

Phenyl (1R,2R)-1,2-dimethoxyoctylcarbamate (3.2) and phenyl (1S,2R)-1,2-dimethoxy-octylcarbamate (3.3)

stirred at room temperature for 15 min and then cooled to 0 °C. A solution of MeOH (0.36 ml, 9.02 mmol) in CH₂Cl₂ (0.6 mL) was added dropwise. The reaction was stirred for 15 min at 0 °C and then guenched with saturated NaHCO₃ (25 mL). The mixture was extracted with CH₂Cl₂ (4 x 20 mL) and the combined organic extracts were dried (Na_2SO_4) and concentrated. The residue was purified by column chromatography (6% - 120% EtOAc in hexanes containing 0.5% Et₃N) to give the desired product (77.2 mg, 55.3%) as a colorless oil in a 2.4:1.0 diastereomeric ratio. Further purification by column chromatography (8% - 14% EtOAc in hexanes containing 0.5% Et₃N) yielded analytically pure samples. For faster eluting *anti*-product **3.2**: ¹H NMR (300 MHz, $CDCl_3$) 7.38 (app t, J = 7.7 Hz, 2H), 7.24-7.12 (m, 3H), 5.90 (d, J = 9.8 Hz, 1H), 4.88 (d, J =10.0 Hz, 1H), 3.59-3.49 (m, 1H), 3.52 (s, 3H), 3.44 (s, 3H), 1.55-1.27 (m, 10H), 0.90 (t, J = 6.9Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 155.2, 151.0, 129.5, 125.6, 121.7, 85.6, 82.4, 59.7, 56.0, 31.9, 31.4, 29.5, 25.6, 22.8, 14.3; IR (neat) 3322, 2930, 2857, 1747, 1515, 1487, 1334, 1206, 1103, 1025, 952, 738; HRMS (ESI): m/z calcd for $C_{17}H_{27}NO_4Na$ [M+Na]⁺ 332.1838, found 332.1830. For slower eluting syn-product **3.3**: ¹H NMR (300 MHz, CDCl₃) 7.40-7.34 (m, 2H), 7.29-7.22 (m, 3H), 5.82 (d, J = 9.7 Hz, 1H), 5.00 (dd, J = 10.0, 2.9 Hz, 1H), 3.41 (s, 3H), 3.40 (s, 3H), 3.18 (dt, J = 6.8, 2.9 Hz, 1H), 1.62-1.55 (m, 2H), 1.40-1.24 (m, 8H), 0.90 (t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 154.9, 151.0, 129.5, 125.7, 121.7, 82.9, 82.6, 58.4, 56.5, 31.9, 29.7, 29.0, 25.6, 22.8, 14.3; IR (neat) 3324, 2928, 2857, 1747, 1523, 1488, 1356, 1209, 1086, 954; HRMS (ESI): m/z calcd for C₁₇H₂₇NO₄Na [M+Na]⁺ 332.1838, found 332.1841.

(*E*)-Ethyl non-2-enoate (3.7)

EtO H NMR (300 MHz, CDCl₃) 6.97 (td, J = 15.6, 7.0 Hz, 1H), 5.81 (td, J = 15.7, 1.5 Hz, 1H), 4.19 (q, J = 7.1 Hz, 2H), 2.20 (qd, J = 7.0, 1.5 Hz, 2H),

1.50-1.41 (m, 2H), 1.36-1.27 (m, 9H), 0.89 (t, J = 6.9 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 167.0, 149.7, 121.4, 60.3, 32.4, 31.8, 29.0, 28.2, 22.8, 14.5, 14.3.

(2S,3R)-Ethyl 2,3-dihydroxynonanoate (C1)

A mixture of AD-mix-β in ¹BuOH/H₂O (20 mL, 1:1, v/v) at 0 °C was $EtO_{OH} = O_{OH} =$

(2S,3R)-Ethyl 2,3-dimethoxynonanoate (C2)

A solution of the diol C1 (170.0 mg, 0.779 mmol) in CH₂Cl₂ (4.0 mL) were $EtO \rightarrow OMe$ treated with Ag₂O (271 mg, 1.17 mmol) and MeI (0.22 mL, 3.50 mmol). The reaction was refluxed for 10 h, and Ag₂O (271 mg, 1.17 mmol) and MeI (0.22 mL, 3.50 mmol) were added sequentially. After 12 h, Ag₂O (271 mg, 1.17 mmol) and MeI (0.22 mL, 3.50 mmol) were added sequentially. mmol) were added. The mixture was refluxed for another 6 h, then filter through Celite and the residue was washed with CH₂Cl₂ (30 mL). The combined filtrate was concentrated and the resulting residue was purified by column chromatography (5% - 15% EtOAc in hexanes) to give the desired product **C2** (59.4 mg, 31.0%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) 4.32-4.21 (m, 2H), 3.78 (d, J = 4.1 Hz, 1H), 3.51 (dt, J = 6.5, 4.1 Hz, 1H), 3.44 (s, 3H), 3.39 (s, 3H), 1.61-1.54 (m, 2H), 1.34-1.26 (m, 11H), 0.89 (t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 171.4, 82.7, 82.0, 61.1, 59.2, 58.6, 32.0, 30.1, 29.6, 25.8, 22.8, 14.5, 14.3; IR (neat) 2927, 1747, 1464, 1261, 1190, 1143, 1105, 1031; HRMS (ESI): m/z calcd for C₁₃H₂₆O₄Na [M+Na]⁺ 269.1729, found 269.1713; [α]_D = -29.7 (CHCl₃, *c* 0.63).

(2S,3R)-2,3-Dimethoxynonanoic acid (3.8)



A solution of the ethyl ester C2 (40.0 mg, 0.162 mmol) in 1,2dimethoxyethane/H₂O (2.8 mL, 4:1, v/v) was treated with LiOH·H₂O (13.6 mg, 0.324 mmol). After 3 and 4 h, LiOH·H₂O (6.8 mg, 0.162 mmol) was

added, respectively. The reaction was stirred for another 3 h, then quenched with HCl (0.5 N, ~1.0 mL) to pH~1.5 and extracted with Et₂O (5 x 10 mL). The combined organic extracts were dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (30% EtOAc in hexanes followed by 50% MeOH in EtOAc) to give the unreacted ester (7.6 mg, 19.0%) and carboxylic acid **3.8** (28.1 mg, 79.4%) as a white sticky solid: ¹H NMR (300 MHz, CD₃OD) 3.66 (d, J = 3.0 Hz, 1H), 3.54 (dt, J = 6.7, 3.1 Hz, 1H), 3.42 (s, 3H), 3.41 (s, 3H), 1.69-1.57 (m, 2H), 1.46-1.29 (m, 8H), 0.91 (t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CD₃OD) 178.2, 84.9, 83.9, 59.4, 58.9, 32.9, 31.2, 30.5, 26.8, 23.7, 14.4; IR (neat) 3401, 2926, 2856, 1618, 1418,

1194, 1091; HRMS (ESI): m/z calcd for C₁₁H₂₂O₄Na [M+Na]⁺ 241.1416, found 241.1407; [α]_D = -26.0 (CH₃OH, *c* 0.77).

Phenyl (1*S*,2*R*)-1,2-dimethoxyoctylcarbamate ((-)-3.3)

A stirred solution of the carboxylic acid **3.8** (15.4 mg, 70.5 µmol) in PhO H H H H H benzene (2.0 mL) was treated with Et₃N (0.12 ml, 0.846 mmol) and diphenyl phosphoryl azide (61 µL, 0.282 mmol). After 2 h, diphenylphosphoryl azide (30 µL, 0.140 mmol) was added. The reaction was stirred for 2 h, then quenched with water (10 mL) and extracted with Et₂O (3 x 20 mL). The organic extracts were dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (5% - 15% EtOAc in hexanes) to give the carbamate (-)-**3.3** (11.4 mg, 52.3%) as a colorless oil: $[\alpha]_D = -3.8$ (CHCl₃, *c* 0.52). No other diastereomer was observed.

2-Ethoxyoctanenitrile (3.11)

 NC_{VET} A mixture of heptanal (4.00 g, 35.0 mmol), absolute EtOH (80 ml), (EtO)₃CH (5.8 mL, 35.0 mmol) and the activated 4Å molecular sieves (4.00 g) at 0 °C was treated dropwise with concentrated H₂SO₄ (2.0 ml) and the mixture was stirred at room temperature overnight. After that time, the reaction mixture was concentrated to ~30 mL and slowly poured onto a cold saturated NaHCO₃ solution (80 mL) at 0 °C. The resulting mixture was filtered through Celite. The filtrate was extracted with CH₂Cl₂ (3 x 80mL) and the extracts were dried (Na₂SO₄) and concentrated. The resulting residue was dissolved in CH₂Cl₂ (70 mL), and BiBr₃ (1.57 g, 3.50 mmol) and TMSCN (5.60 ml, 42.0 mmol) were added sequentially. The reaction was stirred overnight, then quenched with saturated NaHCO₃ solution (50 mL)/water

(20 mL) and extracted with CH₂Cl₂ (3 x 100mL). The combined extracts were dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (2% - 5% EtOAc in hexanes) to give the ethoxy nitrile **3.11** (4.51 g, 76.0%) as a colorless liquid. ¹H NMR (300 MHz, CDCl₃) 4.11 (t, J = 6.6 Hz, 1H), 3.82 (qd, J = 8.8, 6.9 Hz, 1H), 3.51 (qd, J = 8.9, 7.0 Hz, 1H), 1.87-1.80 (m, 2H), 1.52-1.44 (m, 2H), 1.38-1.30 (m, 6H), 1.26 (t, J = 7.0 Hz, 3H), 0.89 (t, J = 6.7 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 118.9, 69.0, 66.4, 33.8, 31.7, 28.9, 24.9, 22.7, 15.0, 14.2; IR (neat) 2957, 2930, 2860, 1468, 1335, 1126, 1108, 735; HRMS (EI): *m/z* calcd for C₁₀H₁₉NO (M⁺⁺) 169.1467, found 169.1474.

Representative procedure for the preparation of acyl aminals:

N-((1R,2R)-2-Ethoxy-1-methoxyoctyl)isobutyramide (3.12) and N-((1S,2R)-2-ethoxy-1-methoxyoctyl)isobutyramide (3.13)



0.886 mmol). The reaction was stirred for 15 min, then cooled to 0 °C and isobutyryl chloride (94 μ L, 0.886 mmol) was added dropwise. The mixture was stirred for 15 min at 0 °C and MeOH (1.0 mL, 23.6 mmol) was added dropwise. The reaction was stirred for 15 min at 0 °C and quenched with AcOH (2.0 mL)/water (20 mL). The mixture was extracted with CH₂Cl₂ (3 x 25 mL) and the combined organic extracts were washed with saturated NaHCO₃ (15 mL), dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (15% - 30% EtOAc in hexanes) to give the desired product (121.3 mg, 75.1%) as a white solid in a 2.3:1.0 diastereomeric ratio. Further purification (15% - 30% EtOAc in hexanes) yielded analytically

pure samples. For faster eluting *anti*-product **3.12**: ¹H NMR (300 MHz, CDCl₃) 6.20 (d, J = 9.5 Hz, 1H), 5.01 (dd, J = 9.7, 1.4 Hz, 1H), 3.74 (qd, J = 9.4, 7.0 Hz, 1H), 3.54 (qd, 9.4, 7.1 Hz, 1H), 3.47-3.43 (m, 1H), 3.29 (s, 3H), 2.40 (sept, J = 6.9 Hz, 1H), 1.37-1.23 (m, 10H), 1.18-1.14 (m, 9H), 0.84 (app t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 177.9, 82.6, 80.6, 67.3, 55.8, 36.1, 31.9, 31.8, 29.4, 25.6, 22.7, 19.8, 19.7, 15.8, 14.2; IR (neat) 3271, 2965, 2920, 1653, 1540, 1467, 1233, 1113, 1101; HRMS (EI): m/z calcd for C₁₄H₂₈NO₂ (M-CH₃O)⁺⁺ 242.2120, found 242.2123. For slower eluting *syn*-product **3.13**: ¹H NMR (300 MHz, CDCl₃) 6.20 (d, J = 9.7 Hz, 1H), 5.17 (dd, J = 9.8, 2.9 Hz, 1H), 3.66 (qd, J = 9.4, 7.0 Hz, 1H), 3.46 (qd, J = 9.3, 7.0 Hz, 1H), 3.36 (s, 3H), 3.24 (dt, J = 6.8, 2.9 Hz, 1H), 2.42 (sept, J = 6.9 Hz, 1H), 1.63-1.55 (m, 2H), 1.40-1.25 (m, 10H), 1.23-1.17 (m, 9H), 0.88 (t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 177.5, 81.0, 80.0, 66.1, 56.4, 36.2, 31.9, 29.8, 29.7, 25.6, 22.8, 19.9, 19.7, 15.8, 14.2; IR (neat) 3273, 2971, 2921, 1651, 1538, 1467, 1154, 1103, 1072; HRMS (EI): m/z calcd for C₁₄H₂₈NO₂ (M-CH₃O)⁺⁺ 242.2120, found 242.2119.

N-((1*R*,2*R*)-2-Ethoxy-1-methoxyoctyl)-2-methoxyacetamide (3.14) and *N*-((1*S*,2*R*)-2-ethoxy-1-methoxyoctyl)-2-methoxyacetamide (3.15)

 MeO_{I} $MeO_$

purified by column chromatography (20% - 40% EtOAc in hexanes) to give the desired product (111.8 mg, 68.7%) as a colorless oil in a 1.7:1.0 diastereomeric ratio. Further purification (20% -40% EtOAc in hexanes) yielded analytically pure samples. For the faster eluting anti-product **3.14**: ¹H NMR (300 MHz, CDCl₃) 7.18 (d, J = 9.8 Hz, 1H), 5.06 (dd, J = 10.0, 1.5 Hz, 1H), 3.98 (d, J = 15.2 Hz, 1H), 3.90 (d, J = 15.2 Hz, 1H), 3.76 (qd, J = 9.4, 7.0 Hz, 1H), 3.56 (qd, J = 9.3)7.0 Hz, 1H), 3.49-3.45 (m, 1H), 3.43 (s, 3H), 3.33 (s, 3H), 1.45-1.25 (m, 10H), 1.18 (t, J = 7.0Hz, 3H), 0.86 (t, J = 6.9 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 170.8, 82.4, 80.5, 72.0, 67.3, 59.4, 56.1, 31.9, 31.8, 29.4, 25.7, 22.8, 15.8, 14.2; IR (neat) 3413, 2930, 2858, 1695, 1506, 1113; HRMS (EI): m/z calcd for C₁₃H₂₆NO₃ (M-CH₃O)^{+•} 244.1913, found 244.1925. For the slower eluting syn-product 3.15: ¹H NMR (300 MHz, CDCl₃) 7.22 (d, J = 10.0 Hz, 1H), 5.18 (dd, J =10.1, 3.0 Hz, 1H), 3.98 (d, J = 15.3 Hz, 1H), 3.92 (d, J = 15.3 Hz, 1H), 3.66 (qd, J = 9.2, 7.0 Hz, 1H), 3.50 (qd, J = 9.2, 7.0 Hz, 1H), 3.43 (s, 3H), 3.36 (s, 3H), 3.26 (dt, J = 6.8, 3.0 Hz, 1H), 1.63-1.55 (m, 2H), 1.41-1.29 (m, 8H), 1.21 (t, J = 7.0 Hz, 3H), 0.88 (t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 170.4, 81.0, 79.8, 72.0, 66.5, 59.4, 56.5, 31.9, 29.9, 29.6, 25.7, 22.8, 15.7, 14.3; IR (neat) 3417, 2928, 2858, 1686, 1510, 1112, 1078; HRMS (EI): m/z calcd for C₁₃H₂₆NO₃ (M-CH₃O)^{+•} 244.1913, found 244.1917.

Benzyl (1R,2R)-2-ethoxy-1-methoxyoctylcarbamate (3.16) and benzyl (1S,2R)-2-ethoxy-1-methoxyoctylcarbamate (3.17)

 $O_{H} = O_{OEt} = O_{H} = O_{H}$

the reaction mixture was cooled to 0 °C and benzyl chloroformate (71 µL, 0.500 mmol) was added dropwise. The cold bath was removed and the mixture was stirred for 10 min. After that time, the flask was cooled to 0 °C and benzyl chloroformate (50 µL, 0.354 mmol) was added. The mixture was stirred at room temperature for 30 min and then cooled to 0 °C. A solution of MeOH (0.28 ml, 7.08 mmol) in CH₂Cl₂ (0.5 mL) was added dropwise. The reaction was stirred for 10 min at 0 °C and then quenched with saturated NaHCO₃ (20 mL). The mixture was extracted with CH₂Cl₂ (3 x 20 mL) and the combined organic extracts were dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (5% - 20% EtOAc in hexanes containing 0.5% Et₃N) to give the desired product (76.3 mg, 63.8%) as a colorless oil in a 1.5:1.0 diastereomeric ratio. Further purification (10% - 13% EtOAc in hexanes containing 0.5% Et₃N) yielded analytically pure materials. For faster eluting anti-product 3.16: ¹H NMR (300 MHz, $CDCl_3$) 7.39-7.30 (m, 5H), 5.66 (d, J = 9.8 Hz, 1H), 5.14 (s, 2H), 4.82 (dd, J = 9.9, 1.0 Hz, 1H), 3.74 (qd, J = 9.3, 7.0 Hz, 1H), 3.56 (qd, J = 9.2, 7.0 Hz, 1H), 3.48-3.44 (m, 1H), 3.37 (s, 3H),1.46-1.28 (m, 10H), 1.17 (t, J = 7.0 Hz, 3H), 0.89 (t, J = 7.0 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 156.8, 136.6, 128.7, 128.4, 128.2, 85.6, 80.6, 67.4, 67.1, 55.7, 31.9, 29.5, 25.7, 22.8, 15.9, 14.3; IR (neat) 3337, 2929, 2858, 1731, 1497, 1456, 1326, 1216, 1107, 966, 735; HRMS (ESI): m/z calcd for C₁₉H₃₁NO₄Na [M+Na]⁺ 360.2151, found 360.2148. For slower eluting synproduct **3.17**: ¹H NMR (300 MHz, CDCl₃) 7.40-7.31 (m, 5H), 5.54 (d, J = 10.0 Hz, 1H), 5.15/5.14 (two s, 2H), 4.94 (dd, J = 10.1, 2.9 Hz, 1H), 3.64 (qd, J = 9.2, 7.0 Hz, 1H), 3.48 (qd, J= 9.2, 7.0 Hz, 1H), 3.38 (s, 3H), 3.28 (dt, J = 6.8, 2.9 Hz, 1H), 1.63-1.52 (m, 2H), 1.41-1.26 (m, 8H), 1.20 (t, J = 7.0 Hz, 3H), 0.89 (t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 156.6, 136.5, 128.8, 128.5, 128.4, 82.9, 81.1, 77.4, 67.2, 66.2, 56.3, 31.9, 29.7, 29.6, 25.7, 22.8, 15.8, 14.3; IR

(neat) 3334, 2928, 2858, 1729, 1501, 1455, 1232, 1097, 737; HRMS (ESI): m/z calcd for $C_{19}H_{31}NO_4Na [M+Na]^+$ 360.2151, found 360.2149.

N-(2-Ethoxy-1-methoxyoctyl)methanesulfonamide (3.18 and 3.19)

 $O_2 \xrightarrow{O_2} N \xrightarrow{O_1} V_4$ + $O_2 \xrightarrow{O_2} N \xrightarrow{O_1} V_4$ The title compounds were prepared by following the representative procedure with The title compounds were prepared by the following amounts of reagents: ethoxynitrile 3.11 (100.0 mg, 0.591 mmol), CH₂Cl₂ (4.5 mL), Cp₂Zr(H)Cl (228 mg, 0.886 mmol). After addition of methanesulfonic anhydride (144 mg, 0.827 mmol), The mixture was stirred for 2 min at 0 °C and MeOH (1.0 mL, 23.6 mmol) was added dropwise. The reaction was stirred for 10 min at 0 °C and quenched with saturated NaHCO₃ (15 mL). The mixture was extracted with CH₂Cl₂ (3 x 20 mL) and the combined organic extracts were dried (Na_2SO_4) and concentrated. The residue was purified by column chromatography (20% - 30% EtOAc in hexanes) to give the desired product (40.8 mg, 24.5%) as a colorless oil in a 2.4:1.0 diastereomeric ratio: ¹H NMR (300 MHz, CDCl₃) 5.42 (d, J = 9.4 Hz, 71% of 1H), 5.22 (d, J = 9.5 Hz, 29% of 1H), 4.64 (dd, J = 9.5, 3.1 Hz, 29% of 1H), 4.48 (dd, J = 9.4, 2.4 Hz, 71%of 1H), 3.72-3.52 (m, 2H), 3.45 (s, 29% of 3H), 3.46-3.42 (m, 71% of 1H), 3.40 (s, 71% of 3H), 3.32 (ddd, *J* = 7.2, 5.6, 3.1 Hz, 29% of 1H), 3.06 (s, 29% of 3H), 3.05 (s, 71% of 3H), 1.56-1.25 (m, 10H), 1.21 (t, J = 6.9 Hz, 29% of 3H), 1.18 (t, J = 7.0 Hz, 71% of 3H), 0.88 (app t, J = 6.8Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 88.1 (major), 86.0 (minor), 81.2 (minor), 79.5 (major), 67.2 (major), 66.4 (minor), 56.5, (minor), 55.7 (major), 43.3 (major), 43.2 (minor), 31.9, 31.6 (major), 29.6 (minor), 29.5 (major), 29.3 (minor), 25.8 (minor), 25.4 (major), 22.8, 15.8 (major), 15.7 (minor), 14.2; IR (neat) 3286, 2926, 2858, 1458, 1328, 1161, 1110, 978, 766; HRMS (EI): m/z calcd for C₁₁H₂₄NO₃S (M-CH₃O)^{+•} 250.1477, found 250.1466.
N-((1R,2R)-2-Ethoxy-1-*tert*-butoxyoctyl)isobutyramide (3.21) and N-((1S,2R)-2-ethoxy-1*tert*-butoxyoctyl)isobutyramide (3.22)

 $V_{H}^{0} \xrightarrow{O_{H}} (f)_{4} + V_{H}^{0} \xrightarrow{O_$ procedure with the following amounts of reagents: ethoxynitrile 3.11 (60.0 mg, 0.354 mmol), CH₂Cl₂ (3.5 mL), Cp₂Zr(H)Cl (110.0 mg, 0.425 mmol). After addition of isobutyryl chloride (52 µL, 0.500 mmol), the cold bath was removed and the mixture was stirred for 10 min. After that time, the flask was cooled to 0 °C and a solution of 'BuOH (0.67 ml, 7.08 mmol) in CH₂Cl₂ (0.5 mL) was added dropwise to the reaction mixture over 3 min. The reaction was stirred for 10 min at 0 °C, then diluted with CH₂Cl₂ (10 mL) and guenched with saturated NaHCO₃ (20 mL). The organic layer was separated and the aqueous layer was extracted with CH₂Cl₂ (3 x 15 mL). The combined organic extracts were dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (10% - 20% EtOAc in hexanes containing 0.5% Et₃N) to give the desired product (79.2 mg, 70.8%) as a white solid in a 1.0:2.0 diastereometric ratio. Further purification (12% - 18% EtOAc in hexanes containing 0.5% Et₃N) yielded analytically pure samples. For faster eluting *anti*-product **3.21**: ¹H NMR (300 MHz, CDCl₃) 6.10 (d, J = 9.2 Hz, 1H), 5.32 (dd, *J* = 9.4, 2.0 Hz, 1H), 3.84 (qd, *J* = 9.6, 7.1 Hz, 1H), 3.58 (qd, *J* = 9.6, 7.0 Hz, 1H), 3.30-3.26 (m, 1H), 2.33 (sept, J = 6.9 Hz, 1H), 1.38-1.27 (m, 10H), 1.22 (s, 9H), 1.19-1.12 (m, 9H), 0.87 (t, J = 6.8 Hz, 3H), ¹³C NMR (75 MHz, CDCl₃) 175.8, 83.0, 75.5, 74.6, 67.6, 36.1, 32.0, 31.6, 29.5, 28.6, 25.9, 22.8, 19.7, 19.4, 15.9, 14.3; IR (neat) 3246, 2969, 2922, 2858, 1648, 1552, 1466, 1109, 1069; HRMS (ESI): m/z calcd for C₁₈H₃₇NO₃Na [M+Na]⁺ 338.2671, found 338.2663. For slower eluting *syn*-product **3.22**: ¹H NMR (300 MHz, CDCl₃) 6.00 (d, J = 9.2 Hz,

1H), 5.39 (dd, J = 9.4, 4.0 Hz, 1H), 3.64-3.54 (m, 2H), 3.14-3.09 (m, 1H), 2.30 (sept, J = 6.9 Hz, 1H), 1.51-1.38 (m, 2H), 1.35-1.24 (m, 8H), 1.20 (s, 9H), 1.19-1.11 (m, 9H), 0.87 (t, J = 6.9 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 175.3, 82.1, 74.9, 74.3, 66.8, 36.1, 32.0, 30.2, 29.6, 28.5, 26.0, 22.8, 19.5, 19.4, 15.8, 14.3; IR (neat) 3254, 2960, 2920, 2856, 1646, 1544, 1459, 1365, 1193, 1109, 1072, 731; HRMS (ESI): m/z calcd for C₁₈H₃₇NO₃Na [M+Na]⁺ 338.2671, found 338.2666.

N-((1R,2R)-2-Ethoxy-1-phenoxyoctyl)isobutyramide (3.23) and N-((1S,2R)-2-ethoxy-1phenoxyoctyl)isobutyramide (3.24)



The title compounds were prepared following procedure with the

amounts of reagents: ethoxynitrile 3.11 (60.0 mg, 0.354 mmol), CH₂Cl₂ (3.5 mL), Cp₂Zr(H)Cl (110.0 mg, 0.425 mmol). After addition of isobutyryl chloride (52 µL, 0.500 mmol), the cold bath was removed and the mixture was stirred for 10 min. The mixture was cooled to °C and a solution of PhOH (333 mg, 3.54 mmol) in CH₂Cl₂ (0.5 mL) was added dropwise. The reaction was stirred at °C for 40 min, then quenched with saturated NaHCO₃ solution (20 mL) and extracted with CH₂Cl₂ (4 x 15 mL). The organic extracts were washed with saturated Na₂CO₃ solution (20 mL), dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (7% - 10% EtOAc in hexanes containing 0.5% Et₃N) to give the desired product (81.7 mg, 68.7%) as a white solid in a 5.6:1.0 diastereomeric ratio. Further purification (7% -10% EtOAc in hexanes containing 0.5% Et₃N) yielded analytically pure samples. For faster eluting *anti*-product **3.23**: ¹H NMR (300 MHz, CDCl₃) 7.30-7.23 (m, 2H), 7.03 (app td, J = 7.8, 1.0 Hz, 2H), 6.96 (app tt, J = 7.3, 0.9 Hz, 1H), 6.36 (d, J = 9.9 Hz, 1H), 5.92 (dd, J = 9.9, 1.4 Hz,

1H), 3.96 (qd, J = 9.4, 7.0 Hz, 1H), 3.72 (qd, J = 9.5, 7.0 Hz, 1H), 3.68-3.64 (m, 1H), 2.37 (sept, J = 6.9 Hz, 1H), 1.47-1.37 (m, 4H), 1.33-1.24 (m, 9H), 1.15 (d, J = 6.9 Hz, 3H), 1.08 (d, J = 6.9 Hz, 3H), 0.88 (t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 177.1, 156.4, 129.7, 121.8, 116.5, 80.7, 80.0, 68.1, 36.0, 31.9, 29.4, 25.7, 22.8, 19.6, 19.5, 16.0, 14.2; IR (neat) 3290, 2964, 2929, 2859, 1657, 1595, 1534, 1495, 1222, 1107, 753; HRMS (ESI): m/z calcd for C₂₀H₃₃NO₃Na [M+Na]⁺ 358.2358, found 358.2359. For slower eluting *syn*-product **3.24** (containing trace amount of unknown impurity): ¹H NMR (300 MHz, CDCl₃) 7.30-7.24 (m, 2H), 7.07-7.04 (m, 2H), 6.97 (app tt, J = 7.3, 0.9 Hz, 1H), 6.24 (d, J = 9.8 Hz, 1H), 6.02 (dd, J = 9.8, 3.4 Hz, 1H), 3.72 (qd, J = 9.4, 7.0 Hz, 1H), 3.62 (qd, J = 9.4, 7.0 Hz, 1H), 3.44 (dt, J = 7.0, 3.3 Hz, 1H), 2.36 (sept, J = 6.9 Hz, 1H), 1.71-1.62 (m, 2H), 1.47-1.21 (m, 11H), 1.15 (d, J = 6.9 Hz, 3H), 1.10 (d, J = 6.9 Hz, 3H), 0.87 (app t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 176.5, 156.8, 129.7, 121.9, 116.2, 81.2, 77.8, 66.8, 36.0, 31.9, 30.1, 29.6, 25.7, 22.8, 19.6, 19.5, 15.9, 14.3; IR (neat) 3288, 2963, 2926, 2857, 1653, 1535, 1495, 1220, 1109, 1042, 752; HRMS (ESI): m/z calcd for C₂₀H₃₃NO₃Na [M+Na]⁺ 358.2358, found 358.2358.

N-((1R,2R)-2-Ethoxy-1-(phenylthio)octyl)isobutyramide (3.25) and N-((1S,2R)-2-ethoxy-1-(phenylthio)octyl)isobutyramide (3.26)

 V_{H} V_{OEt} V_{H} by following the representative procedure with the following amounts of reagents: ethoxynitrile **3.11** (60.0 mg, 0.354 mmol), CH₂Cl₂ (3.5 mL), Cp₂Zr(H)Cl (110.0 mg, 0.425 mmol). After addition of isobutyryl chloride (52 μ L, 0.500 mmol), the cold bath was removed and the mixture was stirred for 10 min. The mixture was cooled to °C and a solution of PhSH (117 mg, 1.06 mmol) in CH₂Cl₂ (0.3 mL) was

added dropwise. The reaction was stirred at °C for 10 min, then quenched with saturated NaHCO₃ solution (20 mL) and extracted with CH₂Cl₂ (4 x 15 mL). The organic extracts were dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (7% -13% EtOAc in hexanes containing 0.5% Et_3N) to give the desired product (89.4 mg, 71.7%) as a white solid in a 1.0:7.1 diastereometric ratio. Further purification (10% - 16% EtOAc in hexanes containing 0.5% Et₃N) yielded analytically pure samples. For faster eluting *anti*-product **3.25**: ¹H NMR (300 MHz, CDCl₃) 7.49-7.45 (m, 2H), 7.31-7.20 (m, 3H), 6.00 (d, J = 9.9 Hz, 1H), 5.58 (dd, J = 10.0, 1.8 Hz, 1H), 3.76-3.63 (m, 2H), 3.56 (dt, J = 6.5, 1.7 Hz, 1H), 2.29 (sept, J = 6.9)Hz, 1H), 1.60-1.48 (m, 1H), 1.40-1.14 (m, 12H), 1.08 (d, J = 6.9 Hz, 3H), 1.00 (d, J = 6.9 Hz, 3H), 0.87 (t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 176.3, 133.9, 132.3, 129.1, 127.5, 81.7, 67.3, 60.0, 35.9, 32.5, 31.9, 29.4, 25.7, 22.8, 19.7, 19.6, 15.9, 14.3; IR (neat) 3302, 2962, 2928, 2859, 1652, 1497, 1440, 1379, 1223, 1098, 739; HRMS (ESI): m/z calcd for C₂₀H₃₃NO₂SNa [M+Na]⁺ 374.2130, found 374.2130. For slower eluting *syn*-product **3.26**: ¹H NMR (300 MHz, CDCl₃) 7.47-7.44 (m, 2H), 7.30-7.18 (m, 3H), 6.00 (d, J = 9.5 Hz, 1H), 5.63 (dd, J = 9.7, 3.2 Hz, 1H), 3.69-3.49 (m, 3H), 2.25 (sept, J = 6.9 Hz, 1H), 1.73-1.66 (m, 2H),1.42-1.28 (m, 8H), 1.21 (t, J = 7.0 Hz, 3H), 1.06 (d, J = 6.9 Hz, 3H), 0.98 (d, J = 6.9 Hz, 3H), 0.88 (t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 176.0, 133.6, 132.0, 129.1, 127.3, 82.0, 66.0, 59.8, 35.8, 31.9, 31.6, 29.5, 25.8, 22.7, 19.6, 19.5, 15.7, 14.2; IR (neat) 3293, 2962, 2927, 2858, 1650, 1526, 1223, 1100, 736; HRMS (ESI): m/z calcd for C₂₀H₃₃NO₂SNa [M+Na]⁺ 374.2130, found 374.2115.

1-Cyanoheptyl benzoate (3.27)

 NC_{OBz} A solution of 2-hydroxyoctanenitrile (0.600 g, 4.25 mmol) in CH₂Cl₂ (14 mL) was treated with Et₃N (1.2 mL, 8.50 mmol), DMAP (5.2 mg, 42.5 µmol) and benzoyl chloride (0.60 mL, 5.10 mmol). The reaction was stirred for 1 h, then quenched with water (30 mL) and extracted with CH₂Cl₂ (3 x 30 mL). The organic extracts were dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (5%- 10% Et₂O in hexanes) to give the benzoate **3.27** (1.042 g, 94.1%) as a colorless oil. For spectral data, see ref. 119.

1-(Isobutyramido)-1-methoxyoctan-2-yl benzoate (3.28)

OMe The title compound was prepared by following the representative H_{OBz} The title compound was prepared by following the representative procedure with the following amounts of reagents: benzoate **3.27** (100.0 mg, 0.408 mmol), CH₂Cl₂ (4.0 mL), Cp₂Zr(H)Cl (158 mg, 0.612 mmol), isobutyryl chloride (52 μ L, 0.490 mmol), MeOH (0.7 mL, 17.3 mmol). After the reaction was complete, it was quenched with 1 N HCl (1.5 mL) and water (15 mL) and extracted with CH₂Cl₂ (3 x 25 mL). The organic extracts were washed with saturated NaHCO₃ (15 mL), dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (15% - 30% EtOAc in hexanes) to give the product **3.28** (90.4 mg, 63.5%, containing 4% BnOH) as a white solid in a 1.4:1.0 diastereomeric ratio. Further purification (15% - 30% EtOAc in hexanes) yielded analytically pure materials. For faster eluting product: ¹H NMR (500 MHz, CDCl₃) 8.03-8.01 (m, 2H), 7.60-7.56 (m, 1H), 7.48-7.44 (m, 2H), 5.97 (d, *J* = 9.5 Hz, 1H), 5.24 (dd, *J* = 9.6, 6.6 Hz, 1H), 5.19 (ddd, *J* = 8.6, 6.6, 3.8 Hz, 1H), 3.38 (s, 3H), 2.32 (sept, *J* = 7.0 Hz, 1H), 1.86-1.80 (m, 1H), 1.78-1.70 (m, 1H), 1.44-1.23 (m, 7H), 1.21-1.15 (m, 1H), 1.10 (d, *J* = 7.0 Hz, 3H), 1.00 (d, *J* = 7.0 Hz, 3H), 0.86 (t,

J = 7.0 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) 178.0, 167.0, 133.5, 130.0, 129.9, 128.7, 81.9, 74.5, 56.3, 36.1, 31.8, 31.3, 29.3, 25.3, 22.8, 19.6, 19.5, 14.2; IR (neat) 3295, 2959, 2929, 2858, 1721, 1663, 1529, 1452, 1273, 1113, 712; HRMS (ESI): *m*/*z* calcd for C₂₀H₃₁NO₄Na [M+Na]⁺ 372.2151, found 372.2123. For slower eluting product: ¹H NMR (500 MHz, CDCl₃) 8.08 (app d, J = 7.3 Hz, 2H), 7.59 (app t, J = 7.4 Hz, 1H), 7.47 (app t, J = 7.8 Hz, 2H), 6.01 (d, J = 9.6 Hz, 1H), 5.33 (dd, J = 9.8, 4.0 Hz, 1H), 5.12 (td, J = 8.6, 4.4 Hz, 1H), 3.38 (s, 3H), 2.40 (sept, J = 6.9 Hz, 1H), 1.82-1.73 (m, 2H), 1.44-1.26 (m, 8H), 1.18 (d, J = 7.0 Hz, 3H), 1.17 (d, J = 7.0 Hz, 3H), 0.87 (t, J = 6.7 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) 177.6, 166.7, 133.4, 130.1, 130.0, 128.7, 81.0, 75.8, 56.7, 36.1, 31.8, 30.6, 29.3, 25.4, 22.7, 19.8, 19.7, 14.2; IR (neat) 3299, 2929, 2858, 1722, 1661, 1527, 1453, 1273, 1113, 712; HRMS (ESI): *m*/*z* calcd for C₂₀H₃₁NO₄Na [M+Na]⁺ 372.2151, found 372.2133.

1-(Isobutyramido)-1-hydroxyoctan-2-yl benzoate (3.29)

The title compound was prepared by following the representative procedure with the following amounts of reagents: benzoate **3.27** (100.0 mg, 0.408 mmol), CH₂Cl₂ (4.0 mL), Cp₂Zr(H)Cl (158 mg, 0.612 mmol),

isobutyryl chloride (52 µL, 0.490 mmol). The reaction was quenched with water (15 mL) and extraction of the mixture with EtOAc (3 x 25 mL). After evaporation of the solvent, the crude product was purified by column chromatography (20% - 60% EtOAc in hexanes containing 0.5% Et₃N) gave the product **3.29** (71.6 mg, 52.4%, containing trace amount of impurity) as a colorless oil in a 3.0:1.0 diastereomeric ratio. ¹H NMR (300 MHz, CDCl₃) 8.10-8.07 (m, 1.5H), 8.03-7.99 (m, 0.5H), 7.61-7.56 (m, 1H), 7.48-7.42 (m, 2H), 7.11 (d, J = 7.5 Hz, 0.75H), 6.74 (d, J = 8.3 Hz, 0.25H), 5.53-5.36 (m, 1H), 5.22-5.17 (m, 0.25H), 5.15-5.10 (m, 0.75H), 4.74 (br s, 0.25H), 4.52

(br s, 0.75H), 2.47-2.22 (m, 1H), 1.92-1.78 (m, 2H), 1.44-1.20 (m, 8H), 1.13 (d, J = 6.8 Hz, 2.25H), 1.11 (d, J = 6.8 Hz, 2.25H), 1.04 (d, J = 6.9 Hz, 0.75H), 0.99 (d, J = 6.9 Hz, 0.75H), 0.85 (t, J = 7.0 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) (for major diastereomer) 178.4, 167.7, 133.6, 130.1, 128.6, 76.2, 75.0, 35.6, 31.8, 30.7, 29.2, 25.5, 22.7, 19.5, 19.3, 14.2; IR (neat) 3338, 2959, 2928, 2858, 1720, 1657, 1530, 1451, 1274, 1119, 1070, 711; HRMS (EI): m/z calcd for C₁₉H₂₈NO₃ (M-OH)⁺⁺ 318.2069, found 318.2064.

N-(1-Methoxynonyl)isobutyramide (3.31)

The title compound was prepared by following the representative OMe Ο \swarrow_5 procedure with the following amounts of reagents: octyl cyanide 3.30 (84.0 mg, 0.603 mmol), THF (6.0 mL), Cp₂Zr(H)Cl (194 mg, 0.754 mmol). The hydrozirconation reaction was stirred for 30 min, then cooled to 0 °C and isobutyryl chloride (95 µL, 0.904 mmol) was added dropwise. The reaction was stirred for 10 min at 0 °C and MeOH (0.73 mL, 18.1 mmol) was added dropwise. The reaction was stirred at 0 °C for 15 min, then quenched with a solution of Et₃N (0.25 mL) in water (15 mL) and extracted with CH₂Cl₂ (4 x 20 mL). The combined organic extracts were dried (Na_2SO_4) and concentrated in vacuo. The residue was purified by column chromatography (15% - 25% EtOAc in hexanes containing 0.5% Et₃N) to gave the title product (91.7 mg, 62.3%) as a white solid: ¹H NMR (300 MHz, CDCl₃) 5.66 (d, J = 9.5 Hz, 1H), 5.10 (td, J = 9.8, 6.1 Hz, 1H), 3.31 (s, 3H), 2.37 (sept, J = 6.9 Hz, 1H), 1.66-1.59 (m, 1H), 1.52-1.43 (m, 1H), 1.38-1.24 (m, 12H), 1.18 (app d, J = 6.4 Hz, 3H), 1.16 (app d, J = 6.4 Hz, 3H), 0.86 (t, J = 6.9 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 177.4, 81.1, 55.9, 36.1, 35.8, 32.0, 29.6, 29.5, 29.4, 25.0, 22.8, 19.9, 19.7, 14.2; IR (neat) 3281, 2920, 2853, 1651,

1538, 1466, 1377, 1236, 1081, 929, 720; HRMS (EI): m/z calcd for C₁₃H₂₆NO (M- CH₃O)^{+•} 212.2014 found 212.2010.

N-(1-Hydroxynonyl)isobutyramide (3.32)

The title compound was prepared by following the representative \swarrow_{5} procedure with the following amounts of reagents: octyl cyanide 3.30 (84.0 mg, 0.603 mmol), CH₂Cl₂ (4.5 mL), Cp₂Zr(H)Cl (171 mg, 0.663 mmol). After hydrozirconation was complete, a solution of isobutyryl chloride (76 μ L, 0.724 mmol) and Et₃N (0.25 mL, 1.81 mmol) in CH₂Cl₂ (0.5 mL) was added dropwise at 0 °C. The reaction was stirred for 15 min at 0 °C and quenched with water (20 mL). The mixture was acidified by adding 1 N HCl to pH~1.0 and extracted with CH₂Cl₂ (3 x 25 mL). The combined organic extracts were washed with saturated NaHCO₃ (20 mL), dried (Na₂SO₄) and concentrated in vacuo. The residue was purified by column chromatography (10% - 70% EtOAc in hexanes) to give acyl hemiaminal **3.32** (74.7 mg, 54.0%) as white solids: ¹H NMR (300 MHz, CDCl₃) 6.17 (br s, 1H), 5.30 (q, J = 6.6 Hz, 1H), 4.27 (br s, 1H), 2.35 (sept, J = 6.9 Hz, 1H), 1.73-1.61 (m, 1H), 1.59-1.48 (m, 1H), 1.40-1.26 (m, 12H), 1.15 (d, J = 6.9 Hz, 3H), 1.14 (d, J = 6.9 Hz, 3H), 0.87 (t, J =6.9 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 178.4, 74.5, 35.7, 35.3, 32.0, 29.6, 29.5, 29.4, 25.1, 22.8, 19.6, 19.4, 14.3; IR (neat) 3298, 2934, 2854, 1653, 1540, 1462, 1231, 1095; HRMS (EI): m/z calcd for C₁₃H₂₆NO (M-OH)^{+•} 212.2014 found 212.2015.

N-(Methoxy(phenyl)methyl)isobutyramide (3.34)

By following the representative procedure, reaction of benzonitrile **3.33** (60.0 mg, 0.582 mmol) with $Cp_2Zr(H)Cl$ (240 mg, 0.931 mmol) in THF (5.8

mL) for 2.5 h followed by acylation with isobutyryl chloride (92 μ L, 0.873 mmol) and addition of MeOH (0.71 mL, 17.5 mmol) in CH₂Cl₂ (0.5 mL) gave the title product **3.34** (87.9 mg, 72.9%) as a white solid: ¹H NMR (300 MHz, CDCl₃) 7.42-7.30 (m, 5H), 6.14 (d, J = 9.4 Hz, 1H), 6.02 (d, J = 8.8 Hz, 1H), 3.45 (s, 3H), 2.40 (sept, J = 6.9 Hz, 1H), 1.21 (d, J = 6.9 Hz, 3H), 1.18 (d, J = 6.9 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 177.4, 139.6, 128.8, 128.6, 126.0, 81.3, 56.1, 36.0, 19.8, 19.6; IR (neat) 3286, 2967, 1653, 1535, 1451, 1230, 1099, 1046, 951, 746; HRMS (EI): *m/z* calcd for C₁₁H₁₄NO₂ (M-CH₃O)^{+•} 192.1024, found 192.1031.

N-((*E*)-Non-1-enyl)isobutyramide (3.36)

A solution of octyl cyanide **3.30** (84.0 mg, 0.603 mmol) in THF (6.0 mL) \downarrow \downarrow H \downarrow \downarrow H \downarrow \downarrow H \downarrow \downarrow H was treated with Cp₂Zr(H)Cl (171 mg, 0.663 mmol). The reaction was stirred for 20 min, then cooled to 0 °C and a solution of isobutyryl chloride (60 µL, 0.573 mmol) and Et₃N (0.25 mL, 1.81 mmol) in THF (4.0 mL) was added dropwise. The flask formerly containing the isobutyryl chloride and Et₃N was rinsed with THF (2 x 1 mL). The reaction was stirred for 10 min at 0 °C and BF₃•OEt₂ (98 µL, 0.784 mmol) was added dropwise. The cold bath was removed and the mixture was stirred overnight. After that time, the reaction was quenched with water (30 mL) and extracted with EtOAc (4 x 30 mL). The combined organic extracts were washed with water (30 mL), dried (Na₂SO₄) and concentrated in vacuo. The residue was purified by column chromatography (10% - 20% EtOAc in hexanes) to gave the title product **3.36** (73.1 mg, 57.3%) as a white solid: ¹H NMR (300 MHz, CDCl₃) 7.24 (d, *J* = 9.6 Hz, 1H), 6.74 (app dd, *J* = 14.2, 10.5 Hz, 1H), 5.15 (td, *J* = 14.2, 7.1 Hz, 1H), 2.37 (sept, *J* = 6.9 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 174.2, 122.7, 113.3, 35.7, 32.0, 30.1, 29.9, 29.3, 29.2, 22.8, 19.6, 14.3; IR (neat) 3283, 2967, 2921, 2851, 1680, 1647, 1526, 1467, 1238, 950, 723; HRMS (EI): *m/z* calcd for C₁₃H₂₅NO (M⁺⁺) 211.1936, found 211.1938.

N-(Isobutyryl)-N-((E)-non-1-enyl)isobutyramide (3.38)

¹H NMR (500 MHz, CDCl₃) 6.22 (td,
$$J = 13.9, 1.4$$
 Hz, 1H), 5.47 (td, $J = 14.2, 7.2$ Hz, 1H), 3.21 (sept, $J = 6.8$ Hz, 1H), 2.16 (dq, $J = 7.3, 1.4$ Hz, 2H), 1.44 (pent, 2H), 1.34-1.26 (m, 8H), 1.17 (d, $J = 6.8$ Hz, 12H), 0.89 (t, $J = 6.8$ Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) 181.0, 132.6, 125.8, 35.0, 32.0, 30.1, 29.4, 29.3, 29.1, 22.8, 19.7, 14.3; IR (neat) 2962, 2928, 1706, 1466, 1383, 1187, 1162, 1092; HRMS (ESI): m/z calcd for C₁₇H₃₁NO₂Na [M+Na]⁺ 304.2252, found 304.2246.

(*E*)-2-Methyltridec-2-en-1-ol (3.39)

HO ++++++++++= A solution of 1-dodecene (freshly distilled, 0.842 g, 5.00 mmol) and methacrolein (90%, 3.89 g, 50.0 mmol) in CH₂Cl₂ (50 mL) was treated with Grubbs' 2nd generation catalyst (64.0 mg, 75 µmol). The reaction was refluxed for 1.5 h and then concentrated. The residue was purified by column chromatography (1% - 5% EtOAc in hexanes) to give the desired product (1.222 g, contaminated with unknown impurities). This product was dissolved in MeOH (29 mL) and cooled to 0 °C. NaBH₄ (219 mg, 5.80 mmol) was added. The reaction was stirred for 30 min and quenched with saturated NH₄Cl solution (1 mL). The mixture was stirred for 10 min while warming to room temperature and then concentrated. The residue was purified by column chromatography (10% - 20% EtOAc in hexanes) to give the allylic alcohol **3.39** (0.621 g, 58.5%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) 5.41 (app t, J = 7.1 Hz, 1H), 4.00 (s, 2H), 2.06-1.99 (m, 2H), 1.66 (s, 3H), 1.45 -1.27 (m, 16H), 0.88 (t, J = 7.0 Hz,

3H); ¹³C NMR (75 MHz, CDCl₃) 134.7, 126.9, 69.3, 32.1, 29.9, 29.8, 29.7, 29.6, 27.8, 22.9, 14.3, 13.8; IR (neat) 3335, 2924, 2854, 1464, 1378, 1012; HRMS (EI): *m/z* calcd for C₁₄H₂₈O (M⁺⁺) 212.2140, found 212.2150.

(E)-3-Methyltetradec-3-enenitrile (3.40)

A solution of allylic alcohol 3.39 (505 mg, 2.38 mmol) and Et₃N (0.66 mL, NC 4.78 mmol) in CH₂Cl₂ (12 mL)/THF (6 mL) was cooled to -42 °C and methanesulfonyl chloride (0.24 mL, 3.09 mmol) was added dropwise. The mixture was stirred for 30 min and anhydrous LiBr (620 mg, 7.14 mmol) was added followed by THF (18 mL). The mixture was warmed to 0 °C and stirred for 1.5 h. After that time, the reaction was diluted with hexanes (150 mL), washed with water (80 mL) and brine (50 mL), dried (Na₂SO₄) and concentrated. The crude allylic bromide was dissolved in DMF (4.5 mL) and CuCN (213 mg, 2.38 mmol) was added in one portion. The reaction was stirred overnight, then quenched with water (30 mL) and extracted with EtOAc (3 x 25 mL). The organic extract was dried (Na₂SO₄) and concentrated. The resulting residue was purified by column chromatography (3% - 5% EtOAc in hexanes) to give allylic nitrile **3.40** (295 mg, 56.0%) as a colorless oil: ¹H NMR (300 MHz, CDCl₃) 5.49 (sext of t, J = 7.2, 1.4 Hz, 1H), 3.03 (s, 2H), 2.04 (q, J = 7.0 Hz, 2H), 1.73 (s, 3H), 1.38-1.27 (m, 16H), 0.89 (t, J = 6.9 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 130.2, 124.0, 118.0, 32.0, 29.8, 29.6, 29.5, 29.4, 29.3, 28.2, 27.4, 22.8, 16.1, 14.2; IR (neat) 2925, 2854, 2249, 1464, 1412, 1114, 721; HRMS (EI): m/z calcd for C₁₅H₂₇N (M⁺⁺) 221.2144, found 221.2152.

N-((1*E*,3*E*)-3-Methyltetradeca-1,3-dienyl)isobutyramide (3.41)

A solution of allylic nitrile 3.40 (90.0 mg, 0.406 mmol) in THF (5.0 mL) was treated with Cp₂Zr(H)Cl (157 mg, 0.609 mmol), The reaction was stirred for 30 min, then cooled to 0 °C and a solution of isobutyryl chloride (51 µL, 0.487 mmol) and Et₃N (0.18 mL, 1.26 mmol) in THF (2.0 mL) was added dropwise. The flask formerly containing the isobutyryl chloride and Et_3N was rinsed with THF (0.5 mL). The reaction was stirred for 2 min at 0 °C and BF₃•OEt₂ (76 µL, 0.609 mmol) was added dropwise. The cold bath was removed and the mixture was stirred for 2 h. After that time, the reaction was diluted with Et₂O (3 mL) and filtered through a small plug of silica gel. The residue was washed with Et₂O (30 mL) and the combined filtrate was dried (Na₂SO₄) and concentrated in vacuo. The residue was purified by column chromatography (5% - 17% EtOAc in hexanes containing 0.5% Et₃N) to gave the title product **3.41** (74.1 mg, 62.1%) as a colorless oil: ¹H NMR (300 MHz, $CDCl_3$) 7.10 (d, J = 10.3 Hz, 1H), 6.91 (dd, J = 14.2, 10.7 Hz, 1H), 5.83 (d, J = 14.3Hz, 1H), 5.34 (t, J = 7.2 Hz, 1H), 2.40 (sept, J = 6.9 Hz, 1H), 2.10 (g, J = 7.0 Hz, 2H), 1.75 (s, 3H), 1.40-1.27 (m, 16H), 1.20 (d, J = 6.9 Hz, 6H), 0.88 (t, J = 6.8 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) 174.3, 131.8, 130.8, 120.4, 118.5, 35.9, 32.1, 30.0, 29.9, 29.8, 29.6, 28.4, 22.9, 19.7, 14.3, 12.7; IR (neat) 3276, 2924, 2854, 1644, 1531, 1467, 1253, 950; HRMS (EI): m/z calcd for C₁₉H₃₅NO (M^{+•}) 293.2719, found 293.2717.

(2S,4R,6R)-Tetrahydro-4-methoxy-5,5,6-trimethyl-2H-pyran-2-carbonitrile (3.43)

¹H NMR (300 MHz, CDCl₃) 4.90 (dd, J = 6.0, 1.2 Hz, 1H), 3.63 (q, J = 6.3Hz, 1H), 3.37 (s, 3H), 3.17 (dd, J = 11.7, 4.5 Hz, 1H), 2.06 (ddd, J = 13.5, 4.5, 1.4 Hz, 1H), 1.84 (ddd, J = 13.4, 11.8, 6.1 Hz, 1H), 1.12 (d, J = 6.3 Hz, 3H), 0.96 (s, 3H), 0.83 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) 118.0, 81.4, 78.0, 64.2, 57.8, 39.4, 29.3, 22.6, 14.6, 12.2; IR (neat) 2980, 2941, 2874, 1470, 1450, 1391, 1164, 1104, 954, 867, 718; HRMS (EI): m/z calcd for C₁₀H₁₇NO₂ (M⁺⁺) 183.1259, found 183.1255.

N-((S)-((2S,4R,6R)-Tetrahydro-4-methoxy-5,5,6-trimethyl-2H-pyran-2-

yl)(methoxy)methyl)isobutyramide (3.44), N-((R)-((2S,4R,6R)-Tetrahydro-4-methoxy-5,5,6-trimethyl-2H-pyran-2-yl)(methoxy)methyl)isobutyramide (3.45) and N-(((2S,4R,6R)-Tetrahydro-4-methoxy-5,5,6-trimethyl-2H-pyran-2-yl)methyl)isobutyramide (3.46)



tetrahydropyranyl cyanide **3.43** (50.0 mg, 0.273 mmol) in CH₂Cl₂ (2.7 mL) was treated with Schwartz reagent (84.5 mg, 0.328 mmol). The mixture was stirred for 15 min, then cooled to 0 °C and isobutyryl chloride (40 μ L, 0.382 mmol) was added dropwise. The cold bath was removed and the mixture was stirred for 10 min. After that time, the flask was cooled to -78 °C and Mg(ClO₄)₂ (61 mg, 0.273 mmol) was added in one portion. After 30 min, a pre-cooled solution (-78 °C) of MeOH (0.22 ml, 5.46 mmol) in CH₂Cl₂ (0.5 mL) was cannulated dropwise to the reaction mixture over 5 min. After completion of addition, the reaction was stirred at -78 °C for 15 min, then quenched with saturated NaHCO₃ solution (15 mL) and warmed to room temperature. The biphasic mixture was extracted with CH₂Cl₂ (3 x 20 mL) and the combined organic extracts were dried (Na₂SO₄) and concentrated. The residue was purified by column chromatography (20% - 70% EtOAc in hexanes containing 0.5% Et₃N) to give the desired products **3.44** and **3.45** (60.2 mg, 76.8%) in a 2.3:1.0 diastereomeric ratio as a colorless oil and

the over-reduction product **3.46** (6.8 mg, 9.7%) as a colorless oil. For **3.46**: ¹H NMR (300 MHz, $CDCl_3$) 5.77 (br s, 1H), 4.05-3.97 (m, 1H), 3.48-3.39 (m, 3H), 3.32 (s, 3H), 3.03 (t, J = 6.4 Hz, 1H), 2.38 (sept, J = 6.9 Hz, 1H), 1.74-1.69 (m, 2H), 1.18-1.14 (m, 9H), 0.96 (s, 3H), 0.88 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) 177.3, 82.0, 74.5, 69.0, 57.7, 41.0, 38.6, 35.9, 27.6, 24.5, 19.9, 19.8, 15.6, 15.5; IR (neat) 3305, 2970, 2933, 2874, 1651, 1548, 1468, 1386, 1243, 1103; HRMS (ESI): m/z calcd for C₁₄H₂₇NO₃Na [M+Na]⁺ 280.1889, found 280.1899. Further purification (20% - 40% EtOAc in hexanes containing 0.5% Et₃N) of the mixture of 3.44 and 3.45 yielded analytically pure diastereomers. For the faster eluting product **3.44** (major, white solid): ¹H NMR $(500 \text{ MHz}, \text{CDCl}_3) 6.01 \text{ (d, } J = 9.0 \text{ Hz}, 1\text{H}), 5.26 \text{ (dd, } J = 9.5, 6.5 \text{ Hz}, 1\text{H}), 3.84-3.80 \text{ (m, 1H)},$ 3.39 (s, 3H), 3.36 (q, J = 6.5 Hz, 1H), 3.33 (s, 3H), 3.04 (dd, J = 8.6, 4.1 Hz, 1H), 2.44 (sept, J =6.9 Hz, 1H), 1.92 (td, J = 13.7, 4.5 Hz, 1H), 1.66 (ddd, J = 13.8, 8.6, 5.2 Hz, 1H), 1.20 (d, J = 6.9Hz, 3H), 1.19 (d, J = 6.9 Hz, 3H), 1.10 (d, J = 6.6 Hz, 3H), 0.94 (s, 3H), 0.86 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) 178.2, 82.0, 80.0, 76.2, 70.9, 57.7, 56.4, 38.2, 36.2, 26.0, 24.5, 19.9, 19.8, 16.1, 15.3; IR (neat) 3300, 2972, 2938, 1659, 1536, 1468, 1387, 1103; HRMS (ESI): m/z calcd for $C_{15}H_{29}NO_4Na$ [M+Na]⁺ 310.1994, found 310.1985. For the slower eluting product 3.45 (minor, colorless oil): ¹H NMR (300 MHz, CDCl₃) 6.26 (d, J = 9.3 Hz, 1H), 5.16 (dd, J = 9.6, 3.9 Hz, 1H, 3.83-3.78 (m, 1H), 3.67 (q, J = 6.6 Hz, 1H), 3.38 (s, 3H), 3.31 (s, 3H), 3.18 (dd, J = 10.08 Hz) 7.0, 3.8 Hz, 1H), 2.42 (sept, J = 6.9 Hz, 1H), 1.96 (ddd, J = 13.7, 6.6, 3.9 Hz, 1H), 1.71-1.63 (m, 1H), 1.22-1.17 (m, 9H), 1.00 (s, 3H), 0.88 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) 177.5, 82.4, 81.7, 70.3, 57.8, 56.7, 38.0, 36.1, 26.3, 25.4, 19.9, 19.7, 17.6, 15.6; IR (neat) 3293, 2970, 2934, 1658, 1531, 1468, 1387, 1170, 1102; HRMS (ESI): m/z calcd for C₁₅H₂₉NO₄Na [M+Na]⁺ 310.1994, found 310.2002.



























































































































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