Studies on the Structure/Reactivity Relationships of Bicyclic Epoxonium Ions and Tethered Nucleophiles. Efforts towards the Total Synthesis of (+)-Lactodehydrothyrsiferol and its Analogs. Multicomponent Approach to the Synthesis of Oxidized Amides through Nitrile Hydrozirconation

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A systematic study on the structure/reactivity relationships of bicyclic epoxonium ions towards tethered nucleophiles has been conducted. The cyclization results show that bicyclo[3.1.0] epoxonium ions have a significant to exclusive preference for exo-cyclizations while bicyclo[4.1.0] epoxonium ions have a strong preference for endo-cyclizations.

A convergent approach towards the total synthesis of polycyclic ether natural product $(+)$ lactodehydrothyrsiferol and its analogs is currently being pursued. This route includes the stereoselective reduction of the bicyclo[3.2.1] ketal which could be prepared from coupling of the functionalized aldehyde and vinyl iodide. Both enantiopure fragments can be obtained from cyclizations of the diepoxide and the monoepoxide, respectively. Key transformations involve two asymmetric epoxidations, a cascade cyclization of diepoxide, a $\mathrm{Cr} / \mathrm{Ni}$-mediated coupling reaction and a stereoselective reduction of bicyclo[3.2.1] ketal.

An efficient one-pot synthesis of oxidized amides from nitrile hydrozirconation has been developed. From the common acylimine intermediates, acyl aminals can be accessed through alcohol addition, acyl hemiaminals can be accessed through water addition and enamides can be accessed through tautomerization.

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# 1.0 STUDIES ON THE STRUCTURE/REACTIVITY RELATIONSHIPS OF BICYCLIC EPOXONIUM IONS AND TETHERED NUCLEOPHILES 

### 1.1 INTRODUCTION

Polycyclic ether structures (Figure 1), which have been discovered in a number of marine natural products, have gained considerable attention from synthetic community due not only to their intrinsically complex structures but also to their interesting biological activities. ${ }^{1,2}$ From a biosynthetic view of point, these compounds have been proposed to arise from cascade cyclizations from the requisite polyepoxide precursors. ${ }^{3-5}$ A key issue associated in this process is the strict regiochemical control, i.e., exo-vs endo-cyclization (Figure 2). As can be envisioned in Figure 1, hemibrevetoxin $\mathrm{B}^{5}$ could be prepared from the polyepoxide through all endocyclization, bullatacin ${ }^{6}$ could be prepared from the polyepoxide precursor through all exocyclization and lactodehydrothyrsiferol ${ }^{7}$ could be prepared from the polyepoxide through a combination of exo- and endo-cyclizations.


Hemibrevetoxin B


Bullatacin


Lactodehydrothyrsiferol

Figure 1. Representative polyether natural products

$\mathrm{LA}=$ Lewis acid, $\mathrm{Nu}=$ nucleophile
Figure 2. Exo- vs endo-cyclization

The exo-pathway is well known and commonly observed from studies on intramolecular cyclizations through epoxide opening. For example, epoxy alcohol 1.1, under acidic conditions, afforded tetrahydrofuran derivative $\mathbf{1 . 3}$ predominantly through exo-pathway transition state $\mathbf{1 . 2}$ (Figure 3). ${ }^{8}$ However, in the presence of the catalytic antibody IgG26D9 elicited from amine oxide antigen 1.4, tetrahydropyran 1.6 was isolated as the sole product from endo-cyclization, ${ }^{8}$ which formally violated the Baldwin's rules ${ }^{9}$ for ring closure reactions.


Figure 3. Cyclizations of epoxy alcohol under acid- and antibody-catalyzed conditions

Ab initio calculations performed by Houk ${ }^{10,11}$ and Coxon ${ }^{12,13}$ showed that under acidcatalyzed conditions, the exo-transition structure (TS) has a nearly ideal trajectory for the attack of the hydroxyl oxygen atom via an $\mathrm{S}_{\mathrm{N}} 2$ manner and therefore has lower energy than endo-TS, resulting in the formation of $\mathbf{1 . 3}$ as the major product through a kinetically and chemically
favored process. The exclusive formation of $\mathbf{1 . 6}$ under antibody-catalyzed (enzymatic) conditions was attributed to the similarity between endo-TS 1.5 and $N$-oxide 1.4. Calculations showed that endo-cyclization proceeds through an $\mathrm{S}_{\mathrm{N}} 1$-type transition structure.

Due to the difficulty in the polycyclic ether formation through endo-pathway and the high efficiency in increasing complexity in cascade cyclizations of polyepoxides, a number of researchers have been involved in investigating the epoxide opening cascades. Murai and coworkers ${ }^{14}$ first reported endo-selective cascade cyclizations of polyepoxides (Figure 4). Diepoxy alcohol 1.7, when treated with $\mathrm{La}(\mathrm{OTf})_{3}$, provided fused bicyclic product 1.8 in $52 \%$ yield. Unfortunately, when the conditions were applied to triepoxy alcohol 1.9, the cyclization efficiency decreased drastically and only $9.3 \%$ of the desired tricycle $\mathbf{1 . 1 0}$ was obtained from all endo-cyclization. In addition, the proposed mechanism requires the presence of chelation of $\mathrm{La}(\mathrm{III})$ ion with the epoxide oxygen and the pendent methoxy oxygen to promote the endoselectivity, which also limits its potential applications in natural product synthesis.


Figure 4. Endo-selective cascade cyclizations of polyepoxides in Murai group

At nearly the same time, McDonald and co-workers ${ }^{15,16}$ studied epoxide opening cascades under Lewis-acid-promoted conditions, and demonstrated that endo-cyclizations could be
achieved in high efficiency with rationally designed substrates. Some representative examples are shown in Figure 5.


Figure 5. Representative endo-selective cyclizations of polyepoxides in McDonald group

McDonald postulated that bicyclic epoxonium ions are key intermediates in cascade cyclizations of epoxides to generate polycyclic ethers and that endo-cyclizations were favored over exo-cyclizations due to the increased ring strain in exo-TSs (bicyclo[3.1.0] epoxonium ions). ${ }^{15}$ As illustrated in Figure 6, bicyclo[4.1.0] epoxonium ion $\mathbf{1 . 2 0}$ (from endo-cyclization) is presumed to be energetically lower than bicyclo[3.1.0] intermediate $\mathbf{1 . 1 9}$ (from exo-cyclization) in the initial cyclization as well as a similar comparison between bicyclo[4.1.0] ion $\mathbf{1 . 2 2}$ and bicyclo[3.1.0] ion $\mathbf{1 . 2 1}$ in the ensuing cyclization, leading to the formation of trans-fused tricycle 1.23 with excellent endo-selectivity.


Figure 6. Hypothesis for regioselective cascade cyclizations

McDonald also observed that terminal nucleophiles can also affect the stereochemical outcomes and/or regioselectivity (Figure 7). ${ }^{17}$ Cyclization of diepoxide 1.24 with a $t$-butyl carbonate group as the terminal nucleophile provided cis-fused bicycle $\mathbf{1 . 2 8}$ which arose from the addition of the carbonate carbonyl to the tertiary carbocation 1.26 via an $\mathrm{S}_{\mathrm{N}} 1$ fashion as the predominant product. Alternatively, the cyclization of diepoxide 1.29 with the better terminal nucleophile dimethyl carbamate gave trans-fused bicycle 1.27 as the major product which came from direct attack of the carbamate carbonyl to the epoxonium ion $\mathbf{1 . 2 5}$ through an $\mathrm{S}_{\mathrm{N}} 2$ pathway. Interestingly, triepoxy $t$-butyl carbonate 1.30, upon cyclization, afforded mainly tricyclic structure 1.32 which came from a combination of endo-cyclization and exo-cyclization and has a cis-ring fusion between the six- and five-membered cyclic ethers. In this case, only trace amount of all-fused trans,trans-tricycle $\mathbf{1 . 3 1}$ was obtained. On the other hand, reaction of dimethylcarbamate triepoxide $\mathbf{1 . 3 3}$ produced desired tricycle 1.31 in good yield, with 1.32 not being observed.

1.24
1.25

$1.27<4 \%$
$1.2735 \%$
1.29


1.30


1.31 4.5\%
1.31 31\%

1.32 20\%
1.32 not observed
1.33

Figure 7. Impact of terminal nucleophiles on stereospecificity and regioselectivity

Besides acidic conditions, cyclizations of epoxides can also be performed under basic ${ }^{18}$, neutral ${ }^{19}$ and oxidative ${ }^{20,21}$ conditions. Jamison and co-workers demonstrated that diepoxide 1.34A and triepoxide 1.35 A , under basic conditions in the protic solvent MeOH , provided THP triad 1.34B and tetrad 1.35B, respectively, in good yields, with the TMS group as a "disappearing" directing group. ${ }^{18}$ Without the "disappearing directing groups", the diepoxide 1.36A and triepoxide 1.37 A , under essentially neutral conditions, delivered triad 1.36 B and tetrad 1.37B, respectively, in excellent yields. ${ }^{19}$ In the total synthesis of hemibrevetoxin B, Holton developed a novel intramolecular epoxide opening cascade initiated by oxidation of the alkene with $N$-(phenylseleno)phthalimide to effect the formation of 1.39 in excellent yield, with
the B and C rings assembled in a single operation (Figure 9). ${ }^{20}$ In this process, the highly polar solvent hexafluoroisopropanol was selected for the endo-selective epoxide opening through the $\mathrm{S}_{\mathrm{N}} 1$-type transition state, which is consistent with the computational analysis. ${ }^{10,11}$



1.36A dr 4:1
1.36B 60\%


Figure 8. Regioselective epoxide opening cascades under basic and neutral conditions


Figure 9. Key transformation in Holton's total synthesis of hemibrevetoxin B

Cascade cyclizations of epoxides/polyepoxides were also studied in my laboratory ${ }^{22}$ under electron transfer initiated cyclization (ETIC ${ }^{23,24}$ ) conditions. These photochemical conditions use medium-pressure mercury lamp as the excitation source, catalytic amount of N methylquinolinium hexafluorophosphate as the sensitizer, $\mathrm{O}_{2}$ as the ultimate oxidant, $4 \AA$ molecular sieves as moisture scavenger, NaOAc as the base, $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ as the peroxide remover, 1,2-dichloroethane as the solvent and toluene as the co-sensitizer. Two typical examples are depicted in Figure 10. Under ETIC conditions, homobenzylic ether $\mathbf{1 . 4 0}$ was oxidized to radical cation 1.41A which fragmented to form the benzyl radical and oxocarbenium ion 1.41B in a reversible manner. The tethered epoxide attacked the electrophile 1.41B to generate bicyclic epoxonium ion 1.41C. The terminal nucleophile THP ether opened the bicyclo[3.1.0] epoxonium ion through exo-pathway to deliver bis-THF product 1.42 irreversibly in good yield. Similarly, reaction of diepoxide 1.44 gave rise to the consecutive exo,exo-cyclization product 1.45 in high efficiency. After removal of the anomeric center with Jones reagent, ${ }^{25}$ lactones 1.43 and 1.46 were obtained as single diastereomers, respectively. The regioselectivity observed herein is in accord with Houk and Coxon's computational studies with 5-exo-pathway being preferred over 6-endo-pathway.


Conditions: (a) $h v, \mathrm{O}_{2}, \mathrm{NMQPF}_{6}$ (cat.), $\mathrm{NaOAc}, \mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}, 1,2$-dichloroethane (DCE)/PhMe (6:1, v/v). (b) Jones reagent. (c) $\mathrm{Ac}_{2} \mathrm{O}$.
Figure 10. Epoxide opening cascades under electron transfer initiated cyclization conditions

The oxocarbenium ion (Lewis acid) generated under ETIC conditions has the merit that it is able to activate the proximal epoxide specifically so that the complication from possible random activation of epoxides by a Brønsted or Lewis acid can be eliminated, making this method ideal for investigating the reactivity of epoxonium ions with specific structures. In order to examine the factors that can affect the reaction pathways in epoxide opening cascades, a systematic study on the cascade cyclizations of monoepoxides/diepoxides under ETIC conditions has been conducted ${ }^{26}$ and details will be discussed in the following context.

### 1.2 STUDIES ON EPOXIDE OPENING CASCADES UNDER ETIC CONDITIONS

As previously described, regioselectivity in epoxide opening cascades could be influenced by ring strain in the forming bicyclic epoxonium ions, nucleophiles and solvents. Besides these, I was interested in whether other factors such as bicyclic epoxonium structure and Lewis acid selection can also affect the reaction pathways. Towards this end, we prepared substrates shown in Figure 11.

From monoepoxides 1.47-1.50, oxocarbenium-activated bicyclic epoxonium ion intermediates (bicyclo[3.1.0] epoxonium and bicyclo[4.1.0] epoxonium) with different substitution patterns (disubstituted and trisubstituted) will be formed and compared in terms of their regioselectivity towards terminal nucleophiles. Diepoxides $\mathbf{1 . 5 1 - 1 . 5 5}$ were designed to study the impact of different Lewis acid-activating groups (oxocarbenium ion and non-stabilized carbenium ion) on the regiochemical outcomes as well as the effect of relative stereochemical orientations of epoxides on the cyclization efficiency. Figure 12 shows the corresponding bicyclic epoxonium ion intermediates that will be compared in this study. Also of note is that the more reactive diphenylmethyl group was employed in the monoepoxide substrates 1.47-1.50 instead of benzyl group ( $c f$. Figure 10) as the electroauxiliary to initiate the oxocarbenium ion formation due to the reduced nucleophilicity of epoxides by the proximal $t$-butyl carbonate groups. For consistency throughout the studies, the diphenylmethyl group was also incorporated in the diepoxide substrates $\mathbf{1 . 5 1}$-1.55 though benzyl group is sufficiently reactive for this purpose.










Figure 11. Substrates for cyclizations

VS


vs


vs


Figure 12. Bicyclic epoxonium ion intermediates to be investigated

### 1.2.1 Synthesis of epoxide substrates

The synthesis of disubstituted monoepoxide 1.47 proceeded from commercially available 4-pentenal in a straightforward manner (Figure 13). Addition of diphenylmethyllithium to 4pentenal followed by methylation of the secondary alcohol and cleavage of the terminal alkene afforded aldehyde 1.56. A sequence of Horner-Emmons olefination, ester reduction, allylic
alcohol epoxidation and primary hydroxyl group protection with $\mathrm{Boc}_{2} \mathrm{O}^{27}$ provided carbonate 1.47. Likewise, epoxide 1.48 was prepared from 5 -hexenal which was obtained from Swern oxidation of 5-hexen-1-ol.
(a) $\mathrm{Ph}_{2} \mathrm{CH}_{2}, n$-BuLi, THF, $0^{\circ} \mathrm{C}, 83 \%$

(b) $\mathrm{NaH}, \mathrm{DMF}, 0^{\circ} \mathrm{C}$, then Mel, rt, $97 \%$
(c) $\mathrm{O}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2},-78^{\circ} \mathrm{C}$, then $\mathrm{PPh}_{3}, \mathrm{rt}, 96 \%$

1.56

(g) $\mathrm{Boc}_{2} \mathrm{O}, \mathrm{N}$-methylimidazole, $\mathrm{PhMe}, 0^{\circ} \mathrm{C}$ to rt, $90 \%$

1.47



Figure 13. Synthesis of disubstituted monoepoxides 1.47 and 1.48

The synthesis of trisubstituted monoepoxides $\mathbf{1 . 4 9}$ and $\mathbf{1 . 5 0}$ is illustrated in Figure 14. Opening of epoxide 1.59 using Yamamoto's aluminum-amide promoted protocol ${ }^{28}$ followed by Johnson-Claisen rearrangement ${ }^{29}$ of the allylic alcohol and reduction of the ethyl ester provided aldehyde 1.60 , which was converted into 1.49 through a sequence of diphenylmethyllithium addition, methylation, silyl ether deprotection, epoxidation and protection of the primary alcohol with $\mathrm{Boc}_{2} \mathrm{O}$. For the synthesis of homologous substrate, aldehyde $\mathbf{1 . 6 0}$ was homologated through Wittig olefination and mercury-mediated enol ether hydrolysis ${ }^{30}$ to give the corresponding aldehyde which was further transformed into $\mathbf{1 . 5 0}$ in a similar manner.
(a) 2,2,6,6-Tetramethylpiperidine, $n-\mathrm{BuLi}$

(d) $\mathrm{Ph}_{2} \mathrm{CH}_{2}, n$-BuLi, THF, $0^{\circ} \mathrm{C}, 70 \%$
(e) $\mathrm{NaH}, \mathrm{DMF}, 0^{\circ} \mathrm{C}$, them Mel, rt
(f) TBAF, THF, 100\%
(g) $m$-CPBA, $\mathrm{NaHCO}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}, 95 \%$
(h) $\mathrm{Boc}_{2} \mathrm{O}, \mathrm{N}$-methylimidazole, $\mathrm{PhMe}, 0^{\circ} \mathrm{C}$ to rt, $86 \%$

(i) $\mathrm{Ph}_{3} \stackrel{+}{\mathrm{P}} \mathrm{CH}_{2} \mathrm{OMeCl}, \mathrm{NaHMDS}, \mathrm{THF},-78^{\circ} \mathrm{C}$ then $\mathrm{Hg}(\mathrm{OAc})_{2}, \mathrm{THF}, \mathrm{H}_{2} \mathrm{O}, \mathrm{KI}, 82 \%$.
(j) $\mathrm{Ph}_{2} \mathrm{CH}_{2}, n$-BuLi, THF, $0^{\circ} \mathrm{C}, 79 \%$
(k) $\mathrm{NaH}, \mathrm{DMF}, 0^{\circ} \mathrm{C}$ them $\mathrm{Mel}, \mathrm{r}$
(I) TBAF, THF, 97\%
(m) $m$-CPBA, $\mathrm{NaHCO}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{O}^{\circ} \mathrm{C}, 99 \%$
(n) $\mathrm{Boc}_{2} \mathrm{O}, \mathrm{N}$-methylimidazole, $\mathrm{PhMe}, 0^{\circ} \mathrm{C}$ to $\mathrm{rt}, 89 \%$

1.50

Figure 14. Synthesis of trisubstituted monoepoxides 1.49 and 1.50

The trisubstituted diepoxides $\mathbf{1 . 5 1 - 1 . 5 2}$ were prepared similar to $\mathbf{1 . 4 9}$ and $\mathbf{1 . 5 0}$. Monoepoxide 1.61 was converted into aldehyde 1.62 in excellent yields through epoxide opening, Johnson-Claisen rearrangement and reduction (Figure 15). Dienol $\mathbf{1 . 6 3}$ was obtained in good yields after a three-step sequence of diphenylmethyllithium addition, methylation and silyl group removal. To ensure the high enantiomeric and diastereomeric control in the epoxidations, asymmetric epoxidation methods were utilized. Dienol 1.63 was converted into diepoxy carbonate $\mathbf{1 . 5 1}$ through double Shi epoxidation ${ }^{31}$ and protection of the primary hydroxyl group with $\mathrm{Boc}_{2} \mathrm{O}$ in excellent yields. A sequence of Sharpless epoxidation, ${ }^{32}$ Shi epoxidation and the primary hydroxyl group protection of 1.63 efficiently provided diastereomeric counterpart 1.52. The stereochemical orientations of the epoxides in $\mathbf{1 . 5 1}$ and $\mathbf{1 . 5 2}$ were given based on the mechanistic analysis.


Figure 15. Synthesis of diepoxides 1.51 and 1.52

Aldehyde 1.62 was homologated to aldehyde 1.64 to prepare diepoxides 1.53 and 1.54 through olefination and mercury-mediated enol ether hydrolysis in 91\% yield (Figure 16). Dienol 1.65 was obtained in a similar manner to $\mathbf{1 . 6 3}$ through diphenylmethyllithium addition, methylation and deprotection. Subsequently, dienol 1.65 was converted into carbonate 1.53 through double Shi epoxidation and carbonate formation, or carbonate $\mathbf{1 . 5 4}$ through a sequence of Sharpless epoxidation, Shi epoxidation and carbonate formation.



Figure 16. Synthesis of diepoxides 1.53 and 1.54

Diepoxide 1.55 was prepared in a convergent manner (Figure 17). Reduction of $\delta$-lactone followed by diphenylmethyllithium addition to the crude lactol provided the diol in $82 \%$ yield over the two steps, which was converted to sulfone 1.66 through a sequence of Mitsunobu reaction, methylation and oxidation ${ }^{33}$ of the resulting sulfide with $m$ CPBA. A Kocienskimodified Julia olefination ${ }^{34}$ between sulfone $\mathbf{1 . 6 6}$ and aldehyde $\mathbf{1 . 6 0}$ afforded the desired diene in $63 \%$ yield. Further operations similar to the synthesis of $\mathbf{1 . 5 1}$ and $\mathbf{1 . 5 3}$ provided carbonate 1.55 in excellent yields.
(a) DIBAL-H, $\mathrm{CH}_{2} \mathrm{Cl}_{2},-78{ }^{\circ} \mathrm{C}$

(b) $\mathrm{Ph}_{2} \mathrm{CH}_{2}$, BuLi, THF, $0^{\circ} \mathrm{C}, 82 \%$, two steps
(c) 1-Phenyl-1-H-tetrazole-5-thiol, $\mathrm{PPh}_{3}$, DIAD, THF, $97 \%$
(d) $\mathrm{NaH}, \mathrm{DMF}, 0^{\circ} \mathrm{C}$, then Mel, rt, $85 \%$
(e) $\mathrm{mCPBA}, \mathrm{NaHCO}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0^{\circ} \mathrm{C}, 95 \%$

1.66
(f) KHMDS, 1,2-dimethoxyethane, $-78^{\circ} \mathrm{C}$, then


Figure 17. Synthesis of diepoxide 1.55

### 1.2.2 Cyclizations of epoxide substrates under ETIC conditions

With these epoxides in hand, I examined their reactions under ETIC conditions to explore the factors that could affect the regioselectivity in the opening of bicyclic epoxonium ions by pendent nucleophiles. First, I carried out the reactions of disubstituted monoepoxides and the cyclization results are illustrated in Figure 18. The reaction of 1.47 exclusively provided 5-exoproduct 1.68 through a disubstituted bicyclo[3.1.0] epoxonium ion intermediate, which is consistent with the previous observations that disubstituted bicyclo[3.1.0] ions prefer 5-exocyclization ( $c f$. Figure 10). ${ }^{35}$ Interestingly, cyclization of the homologated epoxide 1.48 gave a mixture of 5-exo- and 6-endo-products 1.71 and 1.72, respectively, with exo-pathway being slightly favored. The difference between these two reactions is that the initial cyclization of $\mathbf{1 . 4 7}$ forms a bicyclo[3.1.0] intermediate while the cyclization of $\mathbf{1 . 4 8}$ forms a bicyclo[4.1.0] ion. The exo-products were fully characterized by oxidation to the corresponding lactones. ${ }^{25,36,37}$



Figure 18. Cyclizations of disubstituted monoepoxides 1.47 and 1.48

Cyclization of trisubstituted epoxide $\mathbf{1 . 4 9}$ produced a complex mixture of exo- and endoproducts from trisubstituted bicyclo[3.1.0] ion 1.74. The trans-fused endo-product $\mathbf{1 . 7 7}$ arose from the $\mathrm{S}_{\mathrm{N}} 2$ attack of the carbonate to the epoxonium ion 1.74 while the syn-fused endo-product 1.78 was from the addition of the carbonate carbonyl oxygen to the tertiary carbocation 1.75 in an $\mathrm{S}_{\mathrm{N}} 1$ manner. The formation of $\mathbf{1 . 7 8}$ is in accord with McDonald's observation ${ }^{17}$ of cis-fused bicycle $\mathbf{1 . 2 8}$ from 1.24 ( $c f$. Figure 7). On the contrary, the reaction of $\mathbf{1 . 5 0}$ cleanly afforded trans-fused bicycle $\mathbf{1 . 8 1}$ as the only isolable product in excellent yield. These cyclization results clearly demonstrate that the combination of both the methyl substitution and the bicyclo[4.1.0]
epoxonium ion can reverse the regiochemical outcomes from complete exo-pathway to exclusive endo-pathway.


Figure 19. Cyclizations of disubstituted monoepoxides 1.49 and 1.50

Having obtained the general information on the regioselectivity of monoepoxide opening/cyclization with carbonates as the nucleophile, I next investigated the cyclizations of diepoxides. Diepoxide 1.51, under ETIC conditions, provided a mixture of consecutive exo,exoand endo,endo-products 1.85 and 1.86 , respectively, in a combined $40 \%$ yield and with a $5.5: 1$ ratio (Figure 20). Mechanistic analysis suggests that initial cyclization formed bicyclo[3.1.0] epoxonium intermediate 1.82 , which could be opened by the epoxide either in an exo-mode to form a second bicyclo[3.1.0] epoxonium 1.83, or in an endo-mode to form bicyclo[4.1.0] epoxonium 1.84. It is noteworthy that no endo-cyclization product from 1.83 was isolated, indicating that bicyclo[3.1.0] epoxonium ions activated by non-stabilized carbeniums (stronger

Lewis acids) prefer exo-pathway towards tethered nucleophiles. Also of note is that no cis-fused endo-product was observed from 1.84, indicating that epoxides are better nucleophiles than carbonates or that the addition of carbonate to the epoxonium ion is reversible before the loss of the $t$-butyl cation. A similar regioselectivity was observed and a better overall yield was obtained when diastereomeric counterpart 1.52 was exposed to the ETIC conditions.





1.85
5.5


Figure 20. Cyclizations of diepoxides 1.51 and 1.52

Following the completely endo-selective opening of bicyclo[4.1.0] epoxonium ions $\mathbf{1 . 8 0}$ and 1.84, reaction of diepoxides 1.53 and 1.54 afforded trans,syn,trans-fused tricycles 1.91 and 1.93 in all endo-modes as expected since the intermediates were trisubstituted bicyclo[4.1.0]
epoxonium ions. The cyclization of $\mathbf{1 . 5 4}$ is highly efficient in consideration of the product complexity. The higher yield observed in cyclization of $\mathbf{1 . 5 4}$ compared with $\mathbf{1 . 5 3}$ is probably due to the diminished steric interactions in the cyclization processes. To simplify characterizations, acetals 1.91 and 1.93 were oxidized by treatment with $m \mathrm{CPBA}$ and $\mathrm{BF} 3 \cdot \mathrm{OEt}_{2}$ followed by addition of $\mathrm{Et}_{3} \mathrm{~N}^{36,37}$ to form lactones $\mathbf{1 . 9 2}$ and 1.94. The structure of $\mathbf{1 . 9 2}$ was unambiguously confirmed through single crystal X-ray analysis (Figure 22). ${ }^{38}$


Figure 21. Cyclizations of diepoxides 1.53 and 1.54


Figure 22. ORTEP structure of lactone 1.92

Since no selectivity was observed in the cyclization of 1.42, I was not surprised to observe that when diepoxide 1.55 was subjected to ETIC conditions, consecutive exo,exo- and endo,endo-products 1.97 and 1.98 were obtained in comparable yields, with the disubstituted bicyclo[4.1.0] epoxonium $\mathbf{1 . 9 5}$ from initial cyclization being non-regioselective (Figure 23). Tricycle 1.98 was converted into lactone $\mathbf{1 . 9 9}$ and its stereochemical outcomes were established through single crystal X-ray analysis (Figure 24). ${ }^{39}$


Figure 23. Cyclization of diepoxide 1.55


Figure 24. ORTEP structure of lactone 1.99

### 1.3 COMPUTATIONAL ANALYSIS

The cyclization results of the epoxide opening cascades under ETIC conditions show that the regioselectivity is highly dependent on the bicyclic epoxonium ion structures. Trisubstituted bicyclo[4.1.0] epoxonium ions prefer exclusive endo-pathways while disubstituted bicyclo[4.1.0] epoxonium ions essentially show no preference towards exo- or endo-pathways. I also found that trisubstituted bicyclo[3.1.0] epoxonium ions, when formed from attack of epoxides to nonstabilized carbenium ions, favor exo-selectivity exclusively; when formed from attack of epoxides to oxocarbenium ions, give lower selectivity towards exo-cyclization. In order to better understand the origin of the regioselectivity, especially endo-selectivity in cascade cyclizations of epoxides, computational analysis was initiated using the B3LYP/6-31G(d) method ${ }^{40,41}$ to mimic the cyclization transition structures in the gas phase. This was performed by using Gaussian $03^{42}$ program in the Houk group at UCLA.

Initial study was carried out on a model reaction of bimolecular nucleophilic addition of dimethyl carbonate to 1,2,2,3-tetramethyloxiranium ion (Figure 25). The dimethyl carbonate can add to either the tertiary or secondary center of the epoxonium ion, with no geometrical constraints in either case. As shown in Figure 25, TS1 corresponds to the transition structure for nucleophilic addition to the secondary center and TS2 corresponds to the transition structure for nucleophilic addition to the tertiary center. In both TSs, the breaking and forming $\mathrm{C}-\mathrm{O}$ bond distances are within the range of 2.0 and $2.2 \AA$, indicating that both reactions proceed through $\mathrm{S}_{\mathrm{N}} 1$-like transition states. The observation of the longer partial C-O bonds in TS2 is an indicative of a looser $\mathrm{S}_{\mathrm{N}} 1$ transition state, resulting from the formation of a partial tertiary carbocation. The partial tertiary carbocation formation can better stabilize the transition state, making TS2 lower in energy than TS1 by $4.3 \mathrm{kcal} / \mathrm{mol}$. The dihedral angles in the absence of geometrical
constraints in TS1 $\left(\mathrm{O}_{\mathrm{ep}}-\mathrm{C}_{\mathrm{ep}}-\mathrm{H}-\mathrm{O}\right)$ and TS2 $\left(\mathrm{O}_{\mathrm{ep}}-\mathrm{C}_{\mathrm{ep}}-\mathrm{C}-\mathrm{O}\right)$ are $141.3^{\circ}$ and $147.4^{\circ}$, respectively. The bond distances in TS1 and TS2 will be used as the references in the following context, with shorter bond distances being $\mathrm{S}_{\mathrm{N}} 2$-like transition structures.



Figure 25. Transition structures for the addition of dimethyl carbonate to the 1,2,2,3-tetramethyloxiranium ion. The distances are given in $\AA$.

A model reaction was studied to elucidate the origin of the exclusive endo-selectivity from trisubstituted bicyclo[4.1.0] epoxonium ions. As shown in Figure 26, the difference between the model reaction and the real reaction $(\mathbf{1 . 5 0} \boldsymbol{\rightarrow} \mathbf{1 . 8 1}$, Figure 19$)$ is the replacement of the anomeric methoxy group with a hydrogen atom and the tert-butyl carbonate with a methyl carbonate. This model reaction can also be employed to account for the formation of 1.86, 1.88, 1.91, 1.93 and 1.98 in the terminal cyclizations of the corresponding substrates.

The methyl carbonate carbonyl can add either to the secondary or tertiary center of bicyclo[4.1.0] epoxonium ion to form 5-exo- or 6-endo-cyclization products. The transition structure for endo-cyclization (TS3_endo) has a lower energy than that of exo-cyclization (TS3_exo) by $4.5 \mathrm{kcal} / \mathrm{mol}$, which is nearly identical to the energy difference in the unconstrained system. The longer forming C-O bond in TS3_endo means a looser $\mathrm{S}_{\mathrm{N}} 1$-like transition state presumably due to the partial formation of the tertiary carbocation. However, TS3_exo has an $\mathrm{S}_{\mathrm{N}} 2$-like transition state as evidenced by the shorter breaking and forming C-O bond distances. The dihedral angle in TS3_exo has a greater distortion from the unconstrained system ( $161.1^{\circ}$ vs $147.4^{\circ}$ ) than that in TS3_endo ( $135.5^{\circ}$ vs $141.3^{\circ}$ ). The higher energy and greater distortion of the dihedral angle of TS3_exo than those of TS3_endo result in the complete preference towards endo-cyclization pathway.



$\Delta \mathrm{H}_{\mathrm{rel}}(\mathrm{kcal} / \mathrm{mol})$
4.5
$\Delta \mathrm{G}_{\mathrm{rel}}(\mathrm{kcal} / \mathrm{mol}) \quad 4.6$

0.0

Figure 26. Transition structures for the 5-exo- and 6-endo-cyclizations from trisubstituted bicyclo[4.1.0] epoxonium ion. The distances are given in $\AA$.

The effect of the anomeric methoxy group was investigated using the model reaction illustrated in Figure 27. The methoxy group can be either cis or trans to the epoxonium ion ring, with the cis-isomer being more stable than the trans-isomer by $0.3 \mathrm{kcal} / \mathrm{mol}$ in the gas phase. It is clear that the incorporation of the methoxy group has negligible effect on the geometries of the transition structures, though TS4_trans_endo shows a slightly better leaving group departure and enhanced bond formation compared with TS3_endo. The trans- and cis-methoxy isomers favor the endo-cyclization pathway by 4.6 and $4.8 \mathrm{kcal} / \mathrm{mol}$ relative to the corresponding exopathway, respectively, which is similar to the energy difference ( $4.5 \mathrm{kcal} / \mathrm{mol}$ ) observed between TS3_endo and TS3_exo.



TS4_trans_endo

4.8
5.2

0.0
0.0

$\Delta \mathrm{H}_{\mathrm{rel}}(\mathrm{kcal} / \mathrm{mol}) 5.8$
$\Delta \mathrm{G}_{\mathrm{rel}}(\mathrm{kcal} / \mathrm{mol}) 5.9$

Figure 27. Transition structures for 5-exo- and 6-endo-cyclizations of cis- and trans-methoxy trisubstituted bicyclo[4.1.0] epoxonium ions. The distances are given in $\AA$.

As depicted in Figure 18, almost no regioselectivity was obtained from the cyclization of disubstituted epoxide 1.48 which does not have the methyl group on the epoxide. To explain the role of the methyl group, an additional model was employed with the angular methyl group being replaced by a hydrogen atom, in which both 5-exo- and 6-endo-cyclization modes would have partially formed secondary carbocations (Figure 28).

The calculations in the gas phase revealed that the absence of the methyl group has negligible influence on the 5-exo-transition structures. However, the two 6-endo-transition structures are perturbed substantially, with shorter breaking and forming bond distances being observed, indicating more $\mathrm{S}_{\mathrm{N}} 2$-like character in the transition states. The corresponding dihedral angles are also distorted significantly from $137.8^{\circ}$ to $154.3^{\circ}$, and from $136.3^{\circ}$ to $154.2^{\circ}$ for the trans- and cis-isomers, respectively. The transition structures of endo-modes are energetically lower than the corresponding transition structures of exo-modes by $1.4 \mathrm{kcal} / \mathrm{mol}$ for both cis- and trans-isomers, meaning that endo-cyclization is slightly favored over exo-cyclization for disubstituted bicyclo[4.1.0] epoxonium ions. While experimental results show that exo- and endo-cyclizations are two competitive pathways as illustrated in Figure 18, computational studies still support that bicyclo[4.1.0] epoxonium ions have a strong tendency towards endocyclizations.


$\Delta \mathrm{H}_{\mathrm{rel}}(\mathrm{kcal} / \mathrm{mol}) \quad 1.4$
$\Delta \mathrm{G}_{\text {rel }}(\mathrm{kcal} / \mathrm{mol}) \quad 0.8$

0.0
0.0

1.4
1.2

0.0

Figure 28. Transition structures for 5-exo- and 6-endo-cyclizations of cis- and trans-methoxy disubstituted bicyclo[4.1.0] epoxonium ions. The distances are given in $\AA$.

In the terminal annulations of diepoxides $\mathbf{1 . 5 1}$ and $\mathbf{1 . 5 2}$, the bicyclo[3.1.0] epoxonium ions (activated by non-stabilized carbenium ions) preferentially generated 5-exo-cyclization products, which is in accord with Baldwin's ring closure rules. This was mimicked by a model reaction depicted in Figure 29. The transition structure for exo-cyclization (TS6_exo) is similar to the transition structure for the bicyclo[4.1.0] epoxonium ion (TS3_exo) in terms of the forming and breaking bond distances and the dihedral angle. The transition structure for endo-cyclization (TS6_endo), though having similar bond breaking and forming features to the corresponding transition structure for the bicyclo[4.1.0] epoxonium ion (TS3_endo), shows a significant decrease in the dihedral angle $\left(123.6^{\circ}\right.$ vs $\left.135.5^{\circ}\right)$. As a result, this perturbation leads to the
increase of the energy for TS3_endo, making 5-exo-pathway energetically favored over 6-endopathway by $2.7 \mathrm{kcal} / \mathrm{mol}$.



2.7
2.4

Figure 29. Transition structures for 5-exo- and 6-endo-cyclizations of trisubstituted bicyclo[3.1.0] epoxonium ion. The distances are given in $\AA$.

When the epoxide was activated by the oxocarbenium ion instead of a non-stabilized carbenium ion, a small amount of trans-fused endo-cyclization product 1.78 was also isolated from trisubstituted bicyclo[3.1.0] epoxonium ion 1.74. To elucidate the impact of the anomeric methoxy group on the regiochemical outcomes, another model reaction was employed, as shown in Figure 30. Similar to the transition structures in Figure 27, the methoxy group could be cis or trans to the epoxonium ion ring. The trans-isomer $1.0 \mathrm{kcal} / \mathrm{mol}$ more stable than the cis-isomer
because of the stabilization between the lone pair of the epoxide oxygen and the anti-bonding orbital of the C-O (methoxy) bond when they are antiperiplanarly oriented.

For the transition structures for the two cis-isomers, TS7_cis_exo shows a greater leaving group departure in the presence of the anomeric methoxy group while the bond forming character and the dihedral angle remain approximately the same. TS7_cis_endo, however, has enhanced bond formation and a widened dihedral angle which is close to those in the transition structures (TS3_endo, TS4_cis_endo and TS4_trans_endo) of trisubstituted bicyclo[4.1.0] epoxonium ions. For the trans-isomers, the geometry of TS7_trans_exo is essentially unaffected by the incorporation of the methoxy group. Although the breaking and forming bond distances remain nearly the same in TS7_trans_endo, the dihedral angle is distorted from $123.6^{\circ}$ to $132.9^{\circ}$, which is similar to that in TS7_cis_endo. From an energetic view of point, TS7_trans_endo has the lowest energy which is attributed to the stabilization of the anomeric effect when the electronegative methoxy group assumes a pseudoaxial position in the forming tetrahydropyran ring. The two transition structures for exo-cyclizations have slightly higher energy. In addition, TS7_cis_endo, without benefiting from the developing anomeric effect, is the highest in energy.


$\Delta \mathrm{H}_{\mathrm{rel}}(\mathrm{kcal} / \mathrm{mol}) \quad 0.2$
$\Delta \mathrm{G}_{\text {rel }}(\mathrm{kcal} / \mathrm{mol}) \quad 0.6$

1.7
2.1


TS7_trans_exo
0.6
0.4

0.0

Figure 30. Transition structures for 5-exo- and 6-endo-cyclizations of cis- and trans-methoxy trisubstituted bicyclo[3.1.0] epoxonium ions. The distances are given in $\AA$.

The role of the methoxy group is consistent with the previous observations that the 5-exoregioselectivity decreased when bicyclo[3.1.0] epoxonium ions were generated from combination of epoxides with oxocarbenium ions (1.74 and 1.82) rather than non-stabilized carbenium ions ( 1.83 and 1.96 A ). Also of note is the isolation of trans-fused bicycle 1.77 as a single anomer with the methoxy group adopting an axial position.

### 1.4 CONCLUSIONS

A systematic study has been carried out on the oxocarbenium ion-initiated cascade cyclizations of epoxides under ETIC conditions, in which the impact of the epoxide substitution pattern, ring size of the bicyclic epoxonium ions and the Lewis acidic carbocation structures on regiochemical outcomes was fully investigated. These results clearly revealed that ring size is an important determinant on the regioselectivity of bicyclic epoxonium ion opened by tethered nucleophiles. That is, bicyclo[3.1.0] epoxonium ions show significant to exclusive preference towards exo-cyclization pathways while bicyclo[4.1.0] epoxonium ions show a strong tendency towards endo-cyclizations. This observation could be explained from the computational studies that larger rings can adopt a looser transition state with more $\mathrm{S}_{\mathrm{N}} 1$ character, thereby favoring endo-cyclizations. As for smaller rings, endo-TSs are more distorted than exo-TSs, making endoTSs energetically higher and exo-pathways more favorable. In addition, epoxide substitution pattern also has significant influence on the regioselectivity, especially when bicyclo[4.1.0] epoxonium ions serve as the key intermediates. Trisubstituted bicyclo[4.1.0] epoxonium ions prefer exclusive endo-cyclization pathways with the cyclization proceeding through an $\mathrm{S}_{\mathrm{N}} 1$-like transition state due to the better stabilization from a partially formed tertiary carbocation. However, disubstituted bicyclo[4.1.0] epoxonium ions show almost no preference towards exoor endo-cyclizations. Though Lewis acid selection has negligible effect on the regiochemical outcomes of bicyclo[4.1.0] intermediates, it can affect the reaction pathways in a subtle manner when reactions proceed through bicyclo[3.1.0] epoxonium ions. That is, when bicyclo[3.1.0] epoxonium ions are formed from combination of epoxides and non-stabilized carbenium ions, exclusive exo-selectivity is observed; when bicyclo[3.1.0] epoxonium ions are formed from combination of epoxides and oxocarbenium ions, exo-selectivity decreases to some extent. The
endo-cyclization in this process arises from the anomeric effect that generates from the forming tetrahydropyran ring through endo-pathway when the methoxy group adopts a pseudoaxial position, which makes endo-transition structure lower in energy. This delicate effect undoubtedly demonstrates that endo-cyclization can be achieved to generate tetrahydropyran rings through modification of bicyclic epoxonium ions. The current studies definitely provide a solid base for designing new epoxide substrates to provide polycyclic ether structures efficiently.

# 2.0 EFFORTS TOWARDS THE TOTAL SYNTHESIS OF (+)LACTODEHYDROTHYRSIFEROL AND ITS ANALOGS 

### 2.1 INTRODUCTION

(+)-Lactodehydrothyrsiferol (2.1), a marine polycyclic ether natural product, was isolated as an amorphous white solid by Fernandez and co-workers in 2002 from seeds of Laurencia viridis around the Canary Islands (Figure 31). ${ }^{7}$ Spectroscopic analysis shows that it has a central trans-fused pyranopyran structure with a pendent 5-membered lactone ring and an aliphatic side chain that connects the central unit to a trans-tetrahydrofuran ring. It is interesting to note that the B ring assumes a chair conformation while the C ring adopts a twist-boat-like conformation. This has also been observed in structurally related natural products through X-ray and NMR spectroscopic analyses. ${ }^{7,43-45}$

Biological assay shows that $\mathbf{2 . 1}$ has modest inhibitory effect on serine/threonine protein phosphatase (PP2A) with $\mathrm{IC}_{50}$ value of $100 \mu \mathrm{M} .^{46}$ The structurally related natural product thyrsiferyl-23-acetate (TA) is more potent toward the inhibition of PP2A with $\mathrm{IC}_{50}$ values of 4$16 \mu \mathrm{M}$ depending on the enzyme concentration. It is worth noting that TA exhibits specific inhibitory effect on PP2A, and has no effect on protein phosphatase 1 (PP1), 2B (PP2B), 2C (PP2C), or protein tyrosine phosphatases (PTP). ${ }^{47}$ Though other structurally diverse natural products, such as polyether okadaic acid, polyketide tautomycin, and terpenoid cantharidin, are
much more potent inhibitors of PP2A $\left(\mathrm{IC}_{50}=0.2 \sim 40 \mathrm{nM}\right)$ than 2.1 and TA, they also show inhibitory effect on PP1 and/or PP2B (Figure 31). ${ }^{48,49}$ The exclusive selectivity of TA is presumably due to its unique structural features and makes itself an ideal tool to study the cellular processes mediated by PP2A. However, it is unclear whether 2.1 can affect the activity of other protein phosphatases besides PP2A. Up to now, no total synthesis of $\mathbf{2 . 1}$ has been reported. I am currently pursuing a convergent approach toward the total synthesis of $\mathbf{2 . 1}$ and its analogs to further explore their biological activity.

2.1


Thyrsiferyl-23-acetate (TA)





Figure 31. Biologically active natural products and the conformation of the B and C rings in 2.1 and TA

### 2.2 SYNTHETIC PROGRESS

### 2.2.1 Preliminary results

Previous studies on the cascade cyclizations of epoxides showed that the reaction of diepoxide 2.2 under ETIC conditions gave a mixture of exo,exo-product 2.3 and endo,endoproduct 2.4 in a combined $61 \%$ yield and with a 6.1:1 ratio. In this reaction, tricycle 2.5 from an exo-cyclization followed by an endo-cyclization was not observed (Figure 32). ${ }^{26}$ Therefore, construction of the $\mathrm{A}, \mathrm{B}$ and C rings of 2.1 in a single operation from the diepoxide similar to 2.2 is difficult and a new strategy is required for this purpose.


Figure 32. Cyclization of diepoxide 2.2

As depicted in Figure 33, an alternative approach was proposed. The reaction of diepoxide 2.6 under ETIC conditions is expected to give bicyclic epoxonium ion 2.7 and the terminal cyclization will proceed through addition of carbonate carbonyl to the proximal tertiary center in a kinetically favored fashion to generate tricycle 2.8 . With 2.8 in hand, further elaborations will provide tetracyclic compound 2.9 with the $\mathrm{A}, \mathrm{B}$ and C rings being installed.

2.6

2.7

2.8

Figure 33. Proposed cyclization of diepoxide 2.6

To validate the transformation from 2.7 to $\mathbf{2 . 8}$, I prepared epoxy carbonate $\mathbf{2 . 1 4}$, which will form an epoxonium ion intermediate similar to 2.7 upon cyclization (Figure 34). Oxidation of known alkene $\mathbf{2 . 1 0}{ }^{26}$ with $\mathrm{KMnO}_{4}$ afforded $\alpha$-hydroxyl ketone $\mathbf{2 . 1 1}$ in $57 \%$ yield. ${ }^{50}$ Olefination of the ketone under Lebel's Rh-catalyzed conditions provided allylic alcohol 2.12 in $72 \%$ yield. ${ }^{51}$ Epoxidation ${ }^{52}$ of 2.12 catalyzed by $\mathrm{VO}(\mathrm{acac})_{2}$ in the presence of tert-butyl hydroperoxide followed by protection of the primary hydroxyl group with $\mathrm{Boc}_{2} \mathrm{O}^{27}$ gave carbonate 2.14 in excellent yield.


Figure 34. Synthesis of epoxide 2.14

Reaction of $\mathbf{2 . 1 4}$ under ETIC conditions cleanly afforded the desired spiro product $\mathbf{2 . 1 7}$ in excellent yield through addition of carbonate to the tertiary center of bicyclic epoxonium ion 2.16, which was completely consistent with my expectations (Figure 35).


Figure 35. Cyclization of epoxide 2.14

### 2.2.2 Current progress

Following the smooth conversion of 2.14 into bicycle 2.17, I proposed a retrosynthetic approach toward 2.1. As shown in Figure 36, 2.1 can be obtained from a stereoselective
reduction of ketal 2.18 which will be prepared from coupling of aldehyde 2.19 and vinyl iodide 2.20. ${ }^{53}$ These two coupling components can be accessed from cyclizations of diepoxide 2.21 and 2.22, respectively.



Figure 36. Retrosynthetic analysis of 2.1

Due to the availability of the chiral reagents for asymmetric epoxidations, ent-2.21 was initially prepared to explore the feasibility of the above route (Figure 37). Coupling of known vinyl bromide $2.23^{54}$ and aldehyde 2.24 under Fürstner-modified Nozaki-Hiyama-Kishi conditions ${ }^{53}$ provided allylic alcohol $\mathbf{2 . 2 5}$ in $85 \%$ yield. It was subsequently converted into ethyl ester 2.26 through a Johnson-Claisen rearrangement. ${ }^{54}$ Trost ${ }^{54}$ emphasized that the rearrangement efficiency of a similar allylic alcohol is highly dependent on the reaction temperature, with higher temperatures leading to decreased $Z / E$ stereoselectivity and lower temperatures leading to decreased yields. Subsequently, 2.26 was converted into allylic alcohol 2.27 through reduction with DIBAL-H and addition with isopropenylmagnesium bromide. Another Johnson-Claisen rearrangement of 2.27 followed by reduction/addition gave homobenzylic alcohol 2.29 in $65 \%$
yield over two steps. Methylation of the secondary alcohol followed by removal of the two silyl groups provided diol 2.30 in good yield. A sequence of Sharpless asymmetric epoxidation, ${ }^{32}$ Shi asymmetric epoxidation, ${ }^{31}$ and protection of the two hydroxyl groups with $\mathrm{Boc}_{2} \mathrm{O}$ afforded the cyclization substrate ent-2.21. It is worth noting that the first epoxidation gave modest yield due to the unexpected cyclization through addition of the distal hydroxyl group to the epoxide and the diastereoselectivity after the two epoxidations was low ( $\mathrm{dr} \sim 2: 1$ ) based on NMR analysis.




Figure 37. Synthesis of ent-2.21

Subjecting diepoxide ent-2.21 to ETIC conditions produced desired tricycle 2.31 in 43\% yield which contained small amounts of unknown materials (Figure 38). Also isolated from this reaction was tricycle 2.32 . Both 2.31 and 2.32 were oxidized by Jones reagent ${ }^{25}$ to the corresponding lactones 2.33 and $\mathbf{2 . 3 4}$, respectively. The relative stereochemical outcomes of the central cis-tetrahydropyran and the orientation of the 5-membered carbonate ring in lactone 2.33 were fully confirmed through 2D NMR NOESY studies. Subsequently, the Boc group was removed with TMSOTf in the presence of 2,6-lutidine ${ }^{55}$ to give primary alcohol 2.35 in nearly quantitative yield. Oxidation of the hydroxyl group with Dess-Martin periodinane ${ }^{56,57}$ provided aldehyde ent-2.19 in 61\% yield.

Key NOE enhancements observed in $\mathbf{2 . 3 3}$

Figure 38. Synthesis of advanced intermediate ent-2.19

The above results clearly revealed that aldehyde $\mathbf{2 . 1 9}$ can be prepared from diepoxide $\mathbf{2 . 2 1}$. As previously mentioned, there are two problems in this sequence. One is the low yield in Sharpless epoxidation and the other is low diastereoselectivity in the epoxidations of $\mathbf{2 . 3 0}$ presumably because of the interference of the hydroxyl group in Sharpless and/or Shi epoxidations. In order to circumvent these two problems, I proposed an alternative route to $\mathbf{2 . 1 9}$
with replacement of one of the tert-butyl carbonates with a terminal alkene (Figure 39). I envisioned that oxidative cleavage of the terminal alkene would yield the desired aldehyde.

2.19

2.36

Figure 39. An alternative approach to 2.19

The synthesis of 2.36 began with known dienol 2.37 , which was prepared from methyl acrylate and 4-pentenal through a Morita-Baylis-Hillman reaction (Figure 40). ${ }^{58,59}$ Reduction of the methyl ester followed by selective protection of the primary hydroxyl group with TBDPSCl provided silyl ether 2.38. Conversion of $\mathbf{2 . 3 8}$ into trienol 2.40 was achieved through a sequence similar to the synthesis of 2.30. After Sharpless epoxidation, Shi epoxidation and protection of the hydroxyl group with $\mathrm{Boc}_{2} \mathrm{O}$, diepoxide 2.36 was obtained in higher efficiency compared to ent-2.21. Additionally, in this case, the diastereoselectivity in the epoxidations is about $4.6: 1$ with regard to the stereochemical orientations of the two epoxide functionalities.

Under ETIC conditions, diepoxide 2.36 underwent a cascade cyclization to afforded tricyclic product 2.37 (Figure 40). After removal of the anomeric center with Jones reagent, the corresponding lactone was obtained in $17 \%$ yield over two steps as a single diastereomer. The lower efficiency in the cascade cyclization is attributed to the intervention of the nucleophilic terminal alkene in the final cyclization process, suggesting that the terminal alkene must be replaced by non-nucleophilic groups. The ensuing cleavage of the terminal olefin under ozonolytic conditions smoothly afforded aldehyde $\mathbf{2 . 1 9}$ in $83 \%$ yield.





Figure 40. New approach to aldehyde 2.19

With aldehyde 2.19 in hand, I next addressed to the preparation of vinyl iodide 2.45
(Figure 41). Sharpless asymmetric epoxidation of geraniol followed by Sharpless asymmetric dihydroxylation ${ }^{60}$ gave a mixture of triol 2.22 and tetrahydrofuran 2.42 , which was converted into 2.42 completely with the promotion by pyridinium 10 -camphorsulfonate complex. ${ }^{61}$ The stereochemical outcome in $\mathbf{2 . 4 2}$ was established through mechanistic analysis. Activation of the primary hydroxyl group with TsCl followed by elimination under basic conditions provided the epoxide ${ }^{62}$ whose tertiary hydroxyl group was protected as silyl ether $\mathbf{2 . 4 3}$ in excellent yield. The epoxide was opened by 1,3-dilithiopropyne ${ }^{63}$ and the nascent secondary hydroxyl group was protected with TESCl to give bis-silyl ether 2.44. The terminal alkyne was transformed into the
vinylstannane in the presence of the $\mathrm{Bu}_{3} \mathrm{Sn}-\mathrm{AlEt}_{2}$ complex and CuCN in $29 \%$ ( $59 \%$ brsm) yield, which was further converted into vinyl iodide 2.45 in good yield. ${ }^{64,65}$





Figure 41. Synthesis of vinyl iodide 2.45

### 2.3 FUTURE WORK

Though aldehyde 2.19 could be produced from diepoxides 2.21 and $\mathbf{2 . 3 6}$, the efficiency for the substrate preparation or cascade cyclization was still low. Therefore, a better substrate is necessary for the total synthesis. I am intended to prepare a new diepoxide 2.49 from known vinyl bromide $\mathbf{2 . 4 6}{ }^{66,67}$ to differentiate the protecting groups for the two primary hydroxyl groups (Figure 42). Coupling of 2.46 with aldehyde 2.24 will afford the allylic alcohol, which will be converted into ethyl ester $\mathbf{2 . 4 7}$ through a Johnson-Claisen rearrangement. Similarly, $\mathbf{2 . 4 8}$ can be obtained through a repeated ester reduction/nucleophilic addition protocol followed by methylation of the secondary alcohol. Removal of the PMB group with DDQ followed by
epoxidations and carbonate formation will deliver diepoxide 2.49. Since no extra hydroxyl group in the two epoxidations, a better yield from Sharpless epoxidation can be expected and the diastereoselectivity can be retained at a level of $4 \sim 5: 1$. Cyclization of $\mathbf{2 . 4 9}$ under ETIC conditions will give a comparable yield to that of ent-2.21 due to the bulky and non-nucleophilic tert-butyldiphenylsilyloxy group.

2.46
2.47



Figure 42. New approach to aldehyde 2.19

With sufficient amounts of aldehyde 2.19 and vinyl iodide 2.42 in hand, I will next investigate the coupling of these two fragments under Fürstner-modified Nozaki-Hiyama-Kishi conditions (Figure 43). ${ }^{53}$ Once 2.50 is formed, the allylic alcohol can be selectively oxidized with $\mathrm{MnO}_{2}$ and the rest two hydroxyl group will be appropriately protected to give 2.51 . Opening of the 5 -member carbonate under basic conditions ${ }^{68}$ followed by $\mathrm{BiBr}_{3}$-promoted ketal formation ${ }^{69}$ will afford pentacycle 2.18.




Figure 43. Future plan for completion of the total synthesis

From the retrosynthetic analysis, a challenging stereoselective reduction is required to complete the total synthesis. In the presence of suitable Lewis acids, ketal $\mathbf{2 . 1 8}$ will be opened to form oxocarbenium ion 2.52, and various hydride sources will be examined in hope that bulkier hydrides will provide a better diastereoselectivity by favoring approach from the bottom face in order to avoid the severe steric repulsion from the axial substitution groups when approaching from the top face. ${ }^{70}$ Subsequently, the hydroxyl group can be removed through sulfonate formation and reduction ${ }^{71,72}$ to give desired angular methyl group. Removal of the two hydroxyl groups will furnish the natural product 2.1. Modification of C-12 stereochemical orientation or at other positions will generate a number of analogs. Biological activity of $\mathbf{2 . 1}$ and these analogs
will be investigated toward a series of protein phosphatases and the structure-activity relationship pattern can be established accordingly.

### 2.4 SUMMARY

I am currently pursuing a convergent approach to the total synthesis of (+)lactodehydrothyrsiferol and its analogs. This route includes the coupling of two functionalized intermediates aldehyde 2.19 and vinyl iodide 2.42 , both of which result from cyclizations of chiral epoxides with the former being obtained through oxocarbenium ion-initiated cascade cyclizations under ETIC conditions and the latter being obtained through a Brønsted acidmediated cyclization. All the stereocenters in 2.1 will be ultimately derived from chiral reagents except the stereochemical outcome at the C-12 center which will be formed through a stereoselective reduction controlled by both substrate and the reducing agent. This remains a challenge to be explored.

### 3.0 MULTICOMPONENT APPROACH TO THE SYNTHESIS OF OXIDIZED AMIDES THROUGH NITRILE HYDROZIRCONATION

### 3.1 BACKGROUND

Oxidized amides, in which the carbon atom connected to the nitrogen has a higher oxidation state than the normal ( +1 ) valence, have been discovered in a number of natural products. These compounds usually possess acyl aminal, acyl hemiaminal or enamide functionalities, as exemplified by protein synthesis inhibitors pederin ${ }^{73,74}$ and psymberin, ${ }^{75,76}$ cytotoxin zampanolide, ${ }^{77}$ and cytotoxins apicularen $\mathrm{A}^{78}$ and salicylihalamide $\mathrm{A},{ }^{79}$ respectively (Figure 44).


Pederin



Zampanolide


Apicularen A


Salicylihalamide A

Figure 44. Representative natural products containing oxidized amides

Structure-activity relationship studies have shown that the oxidized amide moieties are closely related to the biological acitivities of these complex compounds (Figure 45). ${ }^{80-84}$ For example, apicularen A shows strong growth inhibitory effect against the human melanoma cell line SK-MEL-5 with $\mathrm{GI}_{50}$ value of 6 nM . However, its synthetic analogs, with the enamide side chain being replace by simple alkenes or other enamides, have significantly reduced activity. ${ }^{82}$ Due to the structural complexity and interesting biological activities, these natural products have attracted considerable attention from synthetic organic community.

$\mathrm{GI}_{50}($ SK-MEL-5) $>20000 \mathrm{nM}$


60 nM


900 nM

Figure 45. Apicularen A analogs

As for the synthesis of acyl aminals, a common method is through a Curtius rearrangement of acyl azide followed by nucleophilic addition to the isocyanate intermediate, as evidenced in the synthetic efforts towards mycalamides A and B (Figure 46, A). ${ }^{85}$ A direct coupling of the activated carboxylic acid with $\alpha$-alkoxy amine provides acyl aminal and this strategy has been successfully applied to the total synthesis of mycalamides A and B by Kishi (Figure 46, B). ${ }^{86}$ Alternatively, coupling of carboxylic acid chlorides with alkyl imidates followed by reduction of the newly-formed acyl imines could also deliver acyl aminals (Figure 46, C), whereas the diastereoselectivity in the reduction varies with the substrates. ${ }^{87}$ Our group also developed an
efficient approach to the acyl aminals through addition of oxygen-containing nucleophiles to oxidatively generated acyl iminium ions under very mild conditions (Figure 46, D). ${ }^{88}$


(B)
(C)


(D)


Figure 46. Preparation of acyl aminals

Acyl hemiaminals are relatively more difficult to prepare than acyl aminals. In the total synthesis of (+)-zampanolide, Smith employed a Curtius rearrangement of the acyl azide to set up the acyl aminal functionality (Figure 47, A). ${ }^{89}$ After installation of the side chain, the PMB group was removed with DDQ and the desired acyl hemiaminal was obtained as a 1.3:1 mixture
of the two epimers, with the desired antipode of the natural product being slightly favored. Subsequently, Hoye developed a unique approach to the total synthesis of naturally occurring (-)zampanolide through an aluminum-mediated aza-aldol reaction of the aluminum imidate with (-) -dactylolide and a mixture of 1:1 diastereomers was obtained (Figure 47, B). ${ }^{90}$ In an effort toward the model synthesis of the zampanolide side chain, Porco ${ }^{91}$ studied the oxidative decarboxylation of the amino acid derivative and obtained $\alpha$-acyloxy amide (Figure 47, C). Hydrolysis of the acetate yielded the desired acyl hemiaminal in good yield.
(A)




(B)

(C)


Figure 47. Preparation of acyl hemiaminals

Enamide synthesis has been extensively investigated. A Curtius rearrangement has also been employed to generate isocyanate intermediate from $\alpha, \beta$-unsaturated acyl azides, which can be added by alkyllithium or Grignard reagents to afford enamides (Figure 48, A). ${ }^{92}$ This strategy has been utilized in the total synthesis of salicylihalamide $\mathrm{A}^{93,94}$ and palmeralide. ${ }^{95}$ Fürstner developed an approach to either $E$-enamides or $Z$-enamides from $E$ - or $Z$-alkenylsilanes stereospecifically (Figure 48, B). ${ }^{96}$ Starting from alkenylsilanes, a sequence of epoxidation, epoxide opening with $\mathrm{NaN}_{3}$ and reduction gives $\alpha$-silyl amines which are further converted into the corresponding enamides through acylation and Peterson olefination. Coupling of amides with vinyl iodides, cyclic enol triflates or tosylates under $\mathrm{Cu}(\mathrm{I})$ - or $\mathrm{Pd}(0)$-catalyzed conditions delivers enamides (Figure 48, C and D). ${ }^{97,98}$ Recently, Goossen reported a stereoselective enamide formation via a $\mathrm{Ru}(\mathrm{II})$-catalyzed hydroamination of terminal alkynes, with $E$-enamides being favored in the presence of tributylphosphine and $Z$-enamides being preferred when bis(dicyclohexylphosphino)methane was used as the ligand (Figure 48, E). ${ }^{99}$ A traditional Wittig olefination of $N$-acyl formamides with phosphonium ylides have also been utilized to generate $E$ enamides (Figure 48, F). ${ }^{100}$

While most of the aforementioned methods provide entries into oxidized amides specifically, a general and mild route to all these three types of oxidized amides from a common intermediate needed to be developed. Details will be followed in the subsequent section.
(A)

(B)

or
(a) $m$-CPBA or DMDO
(b) $\mathrm{NaN}_{3}$
$\xrightarrow{\text { (c) } \mathrm{LiAlH}_{4}}$
 57-85\%
$\mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}, \mathrm{Ph} ; \mathrm{R}^{2}=$ alkyl, alkenyl, aryl
(C)

$\mathrm{R}^{1}=n-\mathrm{C}_{5} \mathrm{H}_{11}, \mathrm{Ph} ; \mathrm{R}^{2}=\mathrm{H}, \mathrm{Ph}$, alkenyl; $\mathrm{R}^{3}=\mathrm{H}, \mathrm{Me}$
(D)

$R^{2}=$ alkyl, aryl, alkoxy; $R^{1}=T f, T s$
(E)



(F)


Figure 48. Preparation of enamides

### 3.2 RESEARCH DESIGN AND RESULTS

Based on the structural features, I proposed that oxidized amides can be accessed from common acylimine intermediates. As depicted in Figure 49, acyl aminals can be prepared from acylimines through alcohol addition, ${ }^{101}$ acyl hemiaminals can be prepared from acylimines through water addition and enamides can be prepared from acylimines through a tautomerization.


Figure 49. Oxidized amides from acylimines

For the preparation of the key intermediates acylimines, the common known method is condensation of aldehyde with amide (Figure 50). ${ }^{102-104}$ However, when enolizable aldehydes are used, tautomerization of the forming acylimines could be a problem. Though this problem can be tackled by addition of sulfinic acids or sulfinate salts to the reaction system to form $\alpha$-amido sulfones, ${ }^{105,106}$ reforming and isolating the acylimine is still inefficient, although it is possible. ${ }^{107}$


Figure 50. Generation of acylimines

Metalloimines could be acylated with acid chlorides or carboxylic acid anhydrides to give acylimines. ${ }^{108,109}$ Majoral ${ }^{110}$ reported that when sterically hindered nitriles were treated with Schwartz' reagent $\left(\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}\right)^{111,112}$ followed by acylation with sterically hindered acid chlorides, acylimines were obtained in excellent yields (Figure 51). ${ }^{113-115} \mathrm{My}$ approach to acylimines begins with simple nitriles as well. Hydrometallation of nitriles will be expected to give metalloimines which will react with acid chlorides to afford desired acylimines (Figure 52). Once acylimines are successfully prepared, investigations of the formation of acyl aminals, acyl hemiaminals and enamides under different conditions can be performed.


Figure 51. Hydrozirconation of nitriles


Figure 52. Proposed acylimine formation from nitriles through hydrometallation and acylation

I initiated my study by using the known $\alpha$-methoxy nitrile $\mathbf{1}^{116}$ as the substrate, which was prepared from addition of TMSCN to the corresponding dimethyl acetal mediated by $\mathrm{BiBr}_{3}$ (Figure 53). ${ }^{117}$ Subjecting 3.1 to Schwartz' reagent in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ followed by sequential addition of $\mathrm{PhOC}(\mathrm{O}) \mathrm{Cl}$ and MeOH provided acyl aminals 3.2 and 3.3 in combined $55 \%$ yield and with a 2.4:1 diastereomeric ratio (see below for stereochemical assignment). The observation confirmed the formation of acylimine intermediate 3.4 from acylation of N -zircono-imine that arose from the nitrile reduction. From a mechanistic point of view, the major product 3.2 resulted from chelation-controlled MeOH addition while the minor product 3.3 was from addition of MeOH through a Felkin pathway.


Figure 53. Acyl aminal formation in initial studies

To assign the relative stereochemical outcomes of the two products, I prepared the $\alpha, \beta$ dimethoxy carboxylic acid 3.8 from $\alpha, \beta$-unsaturated ester 3.7 through Sharpless asymmetric dihydroxylation, ${ }^{60}$ double methylation ${ }^{118}$ and hydrolysis of the ethyl ester (Figure 54). Subsequently, 3.8 was converted into acyl azide 3.9 which underwent a spontaneous Curtius rearrangement to form isocyanate 3.10. ${ }^{119}$ The phenoxide anion, which arose from hydrolysis of diphenylphosphoryl azide with adventitious moisture, added to the isocyanate to form (-)-3.3 as a single enantiomer which showed identical spectroscopic features to racemate 3.3.
(a) AD-mix- $\beta, \mathrm{CH}_{3} \mathrm{SO}_{2} \mathrm{NH}_{2}$

3.7


Figure 54. Confirmation of stereochemical outcomes

Having established the reactivity pattern, I next addressed the formation of acyl aminals and acyl hemiaminals extensively. Ethoxy nitrile 3.11 (prepared in a similar manner to 3.1) was used as the substrate for exploring the diastereocontrol in the alcohol addition and the acyl aminal formation with different acylating reagents and nucleophiles.

When 3.11 was subjected to hydrozirconation, acylation with isobutyryl chloride and MeOH addition at $0{ }^{\circ} \mathrm{C}$, both $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and THF were suitable solvents, with the products $\mathbf{3 . 1 2}$ and 3.13 being isolated in combined $75 \%$ and $64 \%$ yields, respectively (entries 1 and 2, Figure 55). It is worth noting that chelation control was preferred in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ while Felkin-pathway was slightly
favored in THF. The reversed stereoselectivity may be explained by the formation of hydrogen bonds between THF and MeOH , which results in the weakening of chelation control. Since $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ is a good solvent for chelation control, several conditions were tested to improve it. Simply lowering down the temperature for MeOH addition to $-78^{\circ} \mathrm{C}$ gave a slightly better result (entry 3). When proper chelating Lewis acids were employed, chelation control could be enhanced to a synthetically useful level. As shown in entries 4 and 5, when MeOH addition was carried out in the presence of a stoichiometric amount of $\mathrm{Zn}(\mathrm{OTf})_{2}$ or $\mathrm{Mg}\left(\mathrm{ClO}_{4}\right)_{2}$, a decent diastereocontrol ( $\mathrm{dr}=5.0: 1$ or $5.7: 1$ ) was accomplished with the reaction efficiency being retained.


Figure 55. Optimization of chelation control

Electrophiles other than isobutyryl chloride were also investigated (Figure 56). Acylation of the metalloimine with $\alpha$-methoxyacetyl chloride give a mixture of 3.14 and 3.15 in a combined $69 \%$ yield and 1.7:1 diastereomeric ratio. The products are electronically similar to the acyl aminals in pederin and psymberin. CbzCl is also a suitable acylating reagent, with $\mathrm{N}, \mathrm{O}$ -
acetals 3.16 and 3.17 being isolated in $64 \%$ overall yield. In this case, the Cbz group can serve as a protecting group and can be removed readily. Unfortunately, when the metalloimine was acylated with methanesulfonic anhydride, only modest yield of sulfonyl aminals 3.18 and 3.19 were obtained. From this reaction, considerable amounts of the aldehyde from hydrolysis of the metalloimine were also isolated.



3.11

3.18
2.4

3.19
1

Figure 56. Acyl aminal formation from 3.11 with various electrophiles

Besides MeOH , other heteronucleophiles also afforded satisfactory results when acylimine 3.20 was utilized as the common intermediate (Figure 57). Sterically hindered ${ }^{t} \mathrm{BuOH}$ had no influence on the reactivity, with the desired acyl aminals being isolated in $71 \%$ yield and the Felkin-pathway being favored. PhOH and PhSH are also suitable for this reaction, providing $69 \%$ and $72 \%$ yields, respectively. It is noteworthy that the chelation-controlled product 3.23 was dominant in the case of PhOH while the product 3.26 from Felkin-type pathway was
predominantly formed in the case of PhSH as the nucleophile. This observation is consistent with the hydrogen-bond-forming abilities of these two nucleophiles.

3.20



Figure 57. Acyl aminal formation from 3.11 with various nucleophiles

Nitriles with different substitutions are good substrates for the synthesis of acyl aminals and acyl hemiaminals (Figure 58). Subjecting $\alpha$-benzoyloxy nitrile $3.27^{120}$ to hydrozirconation, acylation with isobutyryl chloride and MeOH addition, acyl aminal 3.28 was isolated in $64 \%$ yield (contaminated with $4 \% \mathrm{BnOH}$ ), indicating that ester groups can be tolerated in hydrozirconation. ${ }^{121}$ Instead of MeOH addition, a simple aqueous workup after the acylimine formation provided acyl hemiaminal 3.29 in $52 \%$ yield, which is structurally relevant to the zampanolide side chain. Also isolated from this reaction was 3.29 A in $13 \%$ yield which resulted from addition of BnOH to the acylimine intermediate. It is worth noting that from the reactions of 3.27 , the side product from migration of the benzoyl group to the metalloimine nitrogen was not observed. Octyl cyanide 3.30, with no branching at the $\alpha$-carbon, afforded the desired acyl
aminal 3.31 and acyl hemiaminal 3.32 in good yields as well. It was found that THF is a better solvent than $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ to suppress the acylimine tautomerization to the corresponding enamide. Aromatic nitriles also proved to be excellent substrates, with acyl aminal $\mathbf{3 . 3 4}$ being obtained in $73 \%$ isolated yield from phenyl cyanide 3.33 though it underwent a much slower hydrozirconation than aliphatic nitriles.


Figure 58. Acyl aminal and acyl hemiaminal synthesis from various nitriles

Having achieved smooth transformations from nitriles to acyl aminals and acyl hemiaminals, I next examined the enamide synthesis. Octyl cyanide $\mathbf{3 . 3 0}$ and isobutyryl chloride were used to explore the optimum reaction conditions (Figure 59). In $\mathrm{CH}_{2} \mathrm{Cl}_{2}, 22 \%$ of the desired $E$-enamide 3.36 was obtained when the metalloimine was acylated in the presence of $\mathrm{Et}_{3} \mathrm{~N}$
followed by addition of Lewis acid $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ (entry 1 , Table 1). In the absence of $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$, a mixture of 3.36 and 3.37 in $\sim 1: 1$ ratio resulted from acylimine 3.35 (entry 2); without $\mathrm{Et}_{3} \mathrm{~N}$ base, no product ( 3.36 or 3.37 ) was observed (entry 3 ). When the reaction was conducted in THF, acylimine 3.35 was successfully generated in the presence of $\mathrm{Et}_{3} \mathrm{~N}$ and was smoothly tautomerized to afford $E$-enamide 3.36 in $57 \%$ yield in the presence of $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ (entry 4). From this reaction, only minimum amount of $Z$-enamide 3.37 was observed. When the tautomerization was performed in the absence of $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$, only trace amount of 3.36 was observed (entry 5 ); and when the metalloimine was acylated in the absence of $\mathrm{Et}_{3} \mathrm{~N}$, less than $10 \%$ yield of 3.36 was isolated (entry 6). These results convincingly demonstrated that both $\mathrm{Et}_{3} \mathrm{~N}$ and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ are crucially important in this reaction, presumably due to their synergistic effect in converting $\mathbf{3 . 3 5}$ to 3.36. Also of note is that use of more than 1 equiv. of isobutyryl chloride would result in the formation of significant amount of diacylation product 3.38. Following the established conditions, allylic nitrile 3.40, prepared in four steps from methacrolein and 1-dodecene, gave rise to $E, E$-dienamide 3.41 in $62 \%$ yield (Figure 60 ).


Figure 59. Synthesis of enamide 3.36 from octyl cyanide 3.30

Table 1. Optimization of reaction conditions for enamide formation

| entry | solvent | base (3.0 eq.) | Lewis acid (1.3-1.5 eq.) | yield (3.36) |
| :---: | :---: | :---: | :---: | :---: |
| 1 | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $\mathrm{Et}_{3} \mathrm{~N}$ | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | $22 \%$ |
| 2 | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $\mathrm{Et}_{3} \mathrm{~N}$ | $\mathrm{~N} / \mathrm{A}$ | $E: Z \sim 1: 1$ |
| 3 | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | $\mathrm{~N} / \mathrm{A}$ | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | no product |
| 4 | THF | $\mathrm{Et}_{3} \mathrm{~N}$ | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | $57 \%$ |
| 5 | THF | $\mathrm{Et}_{3} \mathrm{~N}$ | $\mathrm{~N} / \mathrm{A}$ | trace |
| 6 | THF | $\mathrm{N} / \mathrm{A}$ | $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ | $<10 \%$ |



Figure 60. Synthesis of $E, E$-dienamide 3.41 from allylic nitrile 3.40

With successful synthesis of oxidized amides from simple nitrile substrates, I next applied this methodology to the synthesis of a more complex model compound that is related to pederin and psymberin. For this purpose, tetrahydropyranyl nitrile 3.43 was prepared (by Michael Green in the Floreancig group at the University of Pittsburgh) in its racemic form from known ketone $3.42^{88}$ through methylation, aluminum-mediated reduction of ketone, ${ }^{122}$ cleavage of the terminal alkene, acylation and displacement of the anomeric acetate group with cyanide (Figure 61).

Reaction of 3.43 through a sequence of hydrozirconation, acylation with isobutyryl chloride and MeOH addition at $0^{\circ} \mathrm{C}$ provided desired acyl aminal 3.44 and its diastereomer 3.45 in a combined $75 \%$ yield and with a 1.9:1 ratio favoring the chelation control (entry 1, Table 2). Also isolated for this reaction was amide 3.46 in $8 \%$ yield, which resulted from direct reduction of the acylimine intermediate by the slightly excess amount of $\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}$. Conducting the MeOH addition at lower temperature $\left(-78{ }^{\circ} \mathrm{C}\right)$ was found to slightly improve the diastereoselectivity (entry 2). Employment of $\mathrm{Mg}\left(\mathrm{ClO}_{4}\right)_{2}$, which promoted chelation control to a considerable extent (cf. Figure 55), did not provide a better diastereocontrol, with acyl aminals being obtained in 77\% yield and 2.3:1 ratio (entry 3). Although only moderate diastereocontrol was obtained under the conditions studied here, the high yield of the acyl aminal together with the desired stereochemical orientation still makes this method attractive for the synthesis of acyl aminals with similar structures in natural products and their analogues.

3.42
(a) 2,6-di-tert-butylpryidine, MeOTf, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, $81 \%$
(b) $\mathrm{Bu}_{3} \mathrm{SnH}, \mathrm{Me}_{2} \mathrm{AlCl}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$
(c) $\mathrm{O}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, then $\mathrm{PPh}_{3}, 48 \%$, two steps
(d) $\mathrm{Ac}_{2} \mathrm{O}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 75 \%$
(e) $\mathrm{TMSCN}, \mathrm{BF}_{3} \bullet \mathrm{OEt}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 85 \%$

3.43


3.44

3.46

Figure 61. Synthesis of tetrahydropyranyl acyl aminals

Table 2. Reaction of 3.43 under various conditions

| entry | temperature | Lewis acid (1 eq.) | yield (3.44+3.45) | dr (3.44/3.45) | yield (3.46) |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | $0{ }^{\circ} \mathrm{C}$ | N/A | $75 \%$ | $1.9: 1$ | $8 \%$ |
| 2 | $-78{ }^{\circ} \mathrm{C}$ | N/A | $71 \%$ | $2.3: 1$ | $10 \%$ |
| 3 | $-78{ }^{\circ} \mathrm{C}$ | $\mathrm{Mg}\left(\mathrm{ClO}_{4}\right)_{2}$ | $77 \%$ | $2.3: 1$ | $10 \%$ |

### 3.3 CONCLUSIONS

An efficient one-pot approach to the synthesis of oxidized amides from nitriles was developed. In this process, the acylimines, which are generated from hydrozirconation of nitriles followed by acylation with suitable electrophiles, serve as the common intermediates to deliver acyl aminals, acyl hemiaminals or enamides through nucleophilic addition or tautomerization. In the acyl aminal formation, moderate to good diastereocontrol could be achieved with the assistance of Lewis acids through chelation-controlled nucleophilic addition. It was also found that the base and Lewis acid played a synergistic role in the $E$-enamide syntheses. In the absence of base or Lewis acid, low efficiency for $E$-enamide formation was observed. With the known methods for the synthesis of nitriles, along with various available electrophiles and nucleophiles, this method will provide a convenient and effective access to oxidized amides with diverse structures.

## APPENDIX A

## STUDIES ON THE STRUCTURE/REACTIVITY REALTIONSHIPS OF BICYCLIC EPOXONIUM IONS AND TETHERED NUCLEOPHILES (SUPPORTING INFORMATION)

General Experimental Proton ( ${ }^{1} \mathrm{H}$ NMR) and carbon $\left({ }^{13} \mathrm{C}\right.$ NMR) nuclear magnetic resonance spectra were recorded at ambient temperatures on Bruker Avance 300 spectrometer at 300 MHz and 75 MHz , respectively, Bruker Avance 500 spectrometer at 500 MHz and 125 MHz , or at Bruker Avance 600 spectrometer at 600 MHz and 151 MHz if specified. The chemical shifts are given in parts per million ( ppm ) on the delta ( $\delta$ ) scale. The solvent peak was used as a reference value, for ${ }^{1} \mathrm{H}$ NMR: $\mathrm{CDCl}_{3}=7.27 \mathrm{ppm}, \mathrm{C}_{6} \mathrm{D}_{6}=7.15 \mathrm{ppm}$, for ${ }^{13} \mathrm{C}$ NMR: $\mathrm{CDCl}_{3}=77.23$. Data are reported as follows: $(\mathrm{s}=$ singlet; $\mathrm{d}=$ doublet; $\mathrm{t}=$ triplet; $\mathrm{q}=\mathrm{quartet} ; \mathrm{dd}=$ doublet of doublets; $\mathrm{dt}=$ doublet of triplets; $\mathrm{br}=$ broad; app $=$ apparently). High resolution mass spectra were recorded on a MICROMASS AUTOSPEC (for EI) or WATERS Q-TOF API-US (for ESI) spectrometer. Infrared (IR) spectra were collected on a Mattson Cygnus 100 spectrometer. Samples for IR were prepared as a thin film on a NaCl plate by dissolving the compound in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and then evaporating the $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. Optical rotations were measured on a Perkin-Elmer 241 polarimeter. Activated $4 \AA$ molecular sieves were obtained through drying in oven at $150{ }^{\circ} \mathrm{C}$
overnight. Tetrahydrofuran and diethyl ether were dried by passage through an activated alumina column under positive $\mathrm{N}_{2}$ pressure. Methylene chloride and benzene were distilled under $\mathrm{N}_{2}$ from $\mathrm{CaH}_{2}$. Diphenylmethane was purchased from Aldrich and used without further purification. Anhydrous DMF and MeI were purchased from Acros. mCPBA was purchased from Acros and purified according to the standard procedure (cf. Purification of Laboratory Chemicals, $4^{\text {th }}$ Ed., by Armarego, W. L. F. and Perrin, D. D.). $\mathrm{Boc}_{2} \mathrm{O}$ and N -methylimidole were purchased from Acros and used without further purification. Anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ was purchased from Aldrich and used as received. Toluene and 1,2-dichloroethane were purchased from Fisher Scientific and dried with $4 \AA$ molecular sieves overnight prior to use. Analytical TLC was performed on E. Merck pre-coated ( 0.25 mm ) silica gel 60F-254 plates. Visualization was done under UV (254 nm). Flash chromatography was done using ICN SiliTech 32-63 60 Å silica gel. Reagent grade ethyl acetate, diethyl ether, and hexanes (commercial mixture) were purchased from EM Science and used as is for chromatography. All reactions were performed in oven or flame-dried glassware under a positive pressure of $\mathrm{N}_{2}$ with magnetic stirring unless otherwise noted.

## 1,1-Diphenylhex-5-en-2-ol (A1)

Ph ${ }_{2} \mathrm{CH}$ A solution of diphenylmethane ( $1.262 \mathrm{~g}, 7.50 \mathrm{mmol}$ ) in THF ( 7.5 ml ) in a two-necked round-bottom flask was treated dropwise $n-\mathrm{BuLi}(1.6 \mathrm{M}$ in hexanes, $4.7 \mathrm{~mL}, 7.5 \mathrm{mmol}$ ). The resulting deep orange solution was refluxed for 1 h and then cooled to $0{ }^{\circ} \mathrm{C}$. A solution of 4-pentenal $(0.252 \mathrm{~g}, 3.00 \mathrm{mmol})$ in THF $(3.0 \mathrm{~mL})$ was added dropwise and the flask formerly containing 4-pentenal was rinsed with THF ( $2 \times 1.0 \mathrm{~mL}$ ). The reaction mixture was stirred at $0^{\circ} \mathrm{C}$ for 1 h , then warmed to room temperature and quenched by slow addition of saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution $(10 \mathrm{~mL})$. The mixture was poured onto water ( 20 mL )
and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $3 \%-12 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give the secondary alcohol $\mathbf{A 1}(0.628 \mathrm{~g}, 83.0 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 7.42-7.17 (m, 10H), $5.80(\mathrm{ddt}, J=17.0,13.4,6.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.06-4.94(\mathrm{~m}, 2 \mathrm{H}), 4.42-4.35(\mathrm{~m}, 1 \mathrm{H})$, $3.90(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.35-2.12(\mathrm{~m}, 2 \mathrm{H}), 1.67-1.41(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $142.5,141.6,138.7,129.0,129.0,128.9$ 128.4, 127.1, 126.8, 115.1, 73.4, 59.1, 34.4, 30.4; IR (neat) $3560,3062,2916,1640,1598,1494,1451,1080,913,745,703 \mathrm{~cm}^{-1}$; HRMS (EI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{18} \mathrm{H}_{20} \mathrm{O}\left(\mathrm{M}^{+\bullet}\right)$ 252.1514, found 252.1523.

## 5-Methoxy-6,6-diphenylhex-1-ene (A2)



The secondary alcohol A1 (0.585 g, 2.32 mmol ) in anhydrous DMF (14.0 $\mathrm{mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with $\mathrm{NaH}(60 \%$ dispersion in mineral oil, 0.232 g , $5.80 \mathrm{mmol})$ and the suspension was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min . MeI $(0.58 \mathrm{~mL}, 9.28 \mathrm{mmol})$ was added dropwise and the cold bath was then removed. After stirred for 2 h at room temperature, the reaction was quenched with water $(20 \mathrm{~mL})$ cautiously and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 40 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The resulting residue was purified by flash chromatography ( $3 \%-6 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give the secondary alcohol A2 ( $0.598 \mathrm{~g}, 96.8 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.44-7.16(\mathrm{~m}, 10 \mathrm{H}), 5.80$ (ddt, $J=17.0,13.3,6.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.06-4.95(\mathrm{~m}, 2 \mathrm{H}), 4.05(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.98(\mathrm{~m}, 1 \mathrm{H}), 3.19$ $(\mathrm{s}, 3 \mathrm{H}), 2.25-2.15(\mathrm{~m}, 2 \mathrm{H}), 1.70-1.54(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.9,142.5$, 138.7, 129.1, 128.7, 128.4, 126.5, 126.4, 114.9, 83.2, 58.1, 56.4, 31.8, 29.6; IR (neat) 3027, 2928, 1640, 1599, 1495, 1451, 1101, 911, 745, $701 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{O}$ (M$\left.\mathrm{C}_{4} \mathrm{H}_{7}\right)^{+\bullet} 211.1123$, found 211.1125 .

## 4-Methoxy-5,5-diphenylpentanal (1.56)



At $-78{ }^{\circ} \mathrm{C}$, the terminal olefin A2 $(193 \mathrm{mg}, 0.724 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(7.5$ mL ) was bubbled gently with ozone until the solution retained a deep blue color and then $\mathrm{PPh}_{3}(570 \mathrm{mg}, 2.17 \mathrm{mmol})$ was added in one portion. The mixture was warmed to room temperature, stirred for 3 h and then concentrated. The residue was purified by column chromatography ( $20 \%-30 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give the aldehyde $1.56(186 \mathrm{mg}, 95.8 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.74(\mathrm{t}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.44-7.21(\mathrm{~m}, 10 \mathrm{H}), 4.04-$ $3.96(\mathrm{~m}, 2 \mathrm{H}), 3.12(\mathrm{~s}, 3 \mathrm{H}), 2.60-2.45(\mathrm{~m}, 2 \mathrm{H}), 2.05-1.95(\mathrm{~m}, 1 \mathrm{H}), 1.80-1.68(\mathrm{~m}, 1 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 202.3,142.3,142.2,128.8,128.6,128.5,126.8,126.6,83.0,58.3,56.6,39.8$, 25.3; IR (neat) $2928,2827,2726,1722,1495,1451,1113,747,704 \mathrm{~cm}^{-1}$; HRMS (EI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{O}\left(\mathrm{M}-\mathrm{C}_{3} \mathrm{H}_{5} \mathrm{O}\right)^{+\bullet} 211.1123$, found 211.1124 .

## (E)-Ethyl 6-methoxy-7,7-diphenylhept-2-enoate (A3)



At $0{ }^{\circ} \mathrm{C}$, triethyl phosphonoacetate $(0.25 \mathrm{~mL}, 1.27 \mathrm{mmol})$ was added dropwise to a suspension of NaH ( $60 \%$ dispersion in mineral oil, $51.0 \mathrm{mg}, 1.27 \mathrm{mmol})$ in THF $(4.0 \mathrm{~mL})$ and the resulting solution was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min. The aldehyde 1.56 ( $171 \mathrm{mg}, 0.637 \mathrm{mmol}$, dissolved in 1.0 mL THF) was introduced dropwise and the flask formerly containing the aldehyde was rinsed with THF ( $2 \times 0.5 \mathrm{~mL}$ ). The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h , then quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}(5 \mathrm{~mL})$ and poured onto water $(10 \mathrm{~mL})$. The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 25 \mathrm{~mL})$ and the organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was purified by column chromatography (10\% - 15\% EtOAc in hexanes) to give the ethyl ester A3 (202 mg, 93.8\%) as a
colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.50-7.17(\mathrm{~m}, 10 \mathrm{H}), 6.96(\mathrm{dt}, J=15.6,6.9 \mathrm{~Hz}, 1 \mathrm{H})$, 5.82 ( br d, $J=15.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.22(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.04(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.99-3.94(\mathrm{~m}$, $1 \mathrm{H}), 3.19(\mathrm{~s}, 3 \mathrm{H}), 2.44-2.25(\mathrm{~m}, 2 \mathrm{H}), 1.75-1.50(\mathrm{~m}, 2 \mathrm{H}), 1.32(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 166.8,149.0,142.5,142.2,128.9,128.8,128.5,128.5,126.7,126.5,121.6,83.0$, 60.3, 58.2, 56.4, 30.8, 27.9, 14.4; IR (neat) 2980, 2931, 1717, 1653, 1495, 1451, 1267, 1202, 1109, 1043, 746, $704 \mathrm{~cm}^{-1}$; HRMS (EI): m/z calcd for $\mathrm{C}_{20} \mathrm{H}_{21} \mathrm{O}_{2}\left(\mathrm{M}-\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{O}\right)^{+\cdot} 293.1542$, found 293.1538.

## (3-(3-Methoxy-4,4-diphenylbutyl)oxiran-2-yl)methanol (A4)



At $-78{ }^{\circ} \mathrm{C}$, DIBAL-H ( 1 M in hexanes, $1.5 \mathrm{~mL}, 1.5 \mathrm{mmol}$ ) was added dropwise to a solution of the ethyl ester A3 (202 mg, 0.598 $\mathrm{mmol})$ in THF $(6.0 \mathrm{~mL})$. The mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for 30 min , then quenched with saturated sodium tartrate $(6.0 \mathrm{~mL})$ and diluted with water $(5 \mathrm{~mL})$. The mixture was warmed to room temperature and stirred vigorously for 2 h . After that time, the mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$ and the extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(6.0 \mathrm{~mL})$ and cooled to $0{ }^{\circ} \mathrm{C} . \mathrm{NaHCO}_{3}(100 \mathrm{mg}, 1.20 \mathrm{mmol})$ and $m \mathrm{CPBA}$ (pure, $134 \mathrm{mg}, 0.777 \mathrm{mmol}$ ) were added sequentially. The suspension was stirred at $0{ }^{\circ} \mathrm{C}$ for 1.5 $h$, then quenched with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(3 \mathrm{~mL})$ and water $(10 \mathrm{~mL})$. The mixture was stirred at room temperature for 30 min , and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 40 \mathrm{~mL})$. The extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated and the residue was purified by column chromatography ( $40 \%-70 \%$ EtOAc in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the epoxy alcohol $\mathbf{A 4}(132 \mathrm{mg}, 70.4 \%, \mathrm{dr} \sim$ 1:1) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.39-7.17(\mathrm{~m}, 10 \mathrm{H}), 4.02-3.94(\mathrm{~m}, 2 \mathrm{H})$, $3.91-3.82(\mathrm{~m}, 1 \mathrm{H}), 3.66-3.56(\mathrm{~m}, 1 \mathrm{H}), 3.17 / 3.16(\mathrm{~s}, 3 \mathrm{H}), 2.93-2.86(\mathrm{~m}, 2 \mathrm{H}), 1.81-1.47(\mathrm{~m}, 4 \mathrm{H}) ;$
${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 142.6,142.3,142.2,128.9,128.8,128.8,128.6,128.5,126.7$, $126.5,83.2,83.1,61.9,61.8,58.6,58.6,58.2,57.9,56.5,56.1,56.0,28.8,28.0,27.4,27.1 ;$ IR (neat) $3428,2928,1495,1451,1095,746,704 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{O}_{3} \mathrm{Na}$ $[\mathrm{M}+\mathrm{Na}]^{+} 335.1623$, found 335.1636.
tert-Butyl (3-(3-methoxy-4,4-diphenylbutyl)oxiran-2-yl)methyl carbonate (1.47)
 $0.397 \mathrm{mmol})$ followed by $\mathrm{Boc}_{2} \mathrm{O}(173 \mathrm{mg}, 0.794 \mathrm{mmol})$. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 2 h , then at room temperature for 1 h . After that time, the reaction was quenched with water ( 15 mL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( $3 \times 20 \mathrm{~mL}$ ). The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was azeotroped with hexanes ( $3 \times 10 \mathrm{~mL}$ ) and then purified by flash chromatography $\left(15 \%-20 \%\right.$ EtOAc in hexanes containing $\left.0.5 \% \mathrm{Et}_{3} \mathrm{~N}\right)$ to give the $t$-butyl carbonate $1.47(146.8 \mathrm{mg}, 89.7 \%$, $\mathrm{dr} \sim 1: 1)$ as a colorless oil: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 7.38-7.16 (m, 10H), 4.23-4.18 (m, 1H), 3.99-3.92 (m, 3H), 3.14/3.13 (s, 3H), 2.94-2.91 (m, 1H), 2.81-2.77 (m, 1H), 1.79-1.42 (m, 4H), $1.48(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 153.4,142.6$, $142.3,142.2,128.9,128.9,128.8,128.6,128.5,126.7,126.5,83.1,83.0,82.7,67.2,67.1,58.3$, $57.9,56.8,56.6,56.5,56.2,55.2,28.8,28.0,27.9,27.4,27.1$; IR (neat) 2981, 2933, 1743, 1495, 1452, 1370, 1281, 1163, 1101, 912, 733, $704 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{25} \mathrm{H}_{32} \mathrm{O}_{5} \mathrm{Na}$ $[\mathrm{M}+\mathrm{Na}]^{+} 435.2147$, found 435.2140 .

## 1,1-Diphenylhept-6-en-2-ol (A5)

Pht $-78{ }^{\circ} \mathrm{C}$, a solution of $\mathrm{DMSO}(1.1 \mathrm{~mL}, 15.0 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$
was added dropwise to a mixture of $(\mathrm{COCl})_{2}(0.65 \mathrm{~mL}, 7.5 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$. After 10 min, a solution of 5-hexen-1-ol $(0.60 \mathrm{~mL}, 5.0 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ was introduced. The white suspension was stirred at $-78{ }^{\circ} \mathrm{C}$ for 30 min , then $\mathrm{Et}_{3} \mathrm{~N}(4.2 \mathrm{~mL}, 30.0 \mathrm{mmol})$ was added and the suspension was stirred for 30 min . After that time, the reaction was warmed to room temperature, diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(80 \mathrm{~mL})$ and washed with saturated $\mathrm{NaHCO}_{3}(80 \mathrm{~mL})$. The organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated to give crude 5-hexenal.

A solution of diphenylmethane $(2.52 \mathrm{~g}, 15.0 \mathrm{mmol})$ in THF $(15 \mathrm{ml})$ in a two-necked roundbottom flask was treated dropwise $n-\mathrm{BuLi}(1.6 \mathrm{M}$ in hexanes, $9.4 \mathrm{~mL}, 15.0 \mathrm{mmol})$. The resulting deep orange solution was refluxed for 1 h and then cooled to $0^{\circ} \mathrm{C}$. A solution of as-prepared crude 5-hexenal in THF ( 3.0 mL ) was added dropwise and the flask formerly containing 4pentenal was rinsed with THF ( $2 \times 1.0 \mathrm{~mL}$ ). The reaction mixture was stirred at $0^{\circ} \mathrm{C}$ for 1 h , then warmed to room temperature and quenched by slow addition of saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution $(15 \mathrm{~mL})$. The mixture was poured onto water $(40 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 50 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $3 \%-15 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give the secondary alcohol $\mathbf{A 5}(0.928 \mathrm{~g}$, $70 \%$, two steps) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.43-7.19(\mathrm{~m}, 10 \mathrm{H}), 5.80$ (tdd, $J=16.9,13.2,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.02-4.93(\mathrm{~m}, 2 \mathrm{H}), 4.40-4.36(\mathrm{~m}, 1 \mathrm{H}), 3.92(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.12-$ $1.96(\mathrm{~m}, 2 \mathrm{H}), 1.76-1.37(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 142.6,141.6,138.9,129.0$, $128.8,128.4,127.0,126.7,114.7,73.8,59.0,34.6,33.8,25.3$; IR (neat) $3561,3453,3026,2918$, 1640, 1598, 1494, 1451, 1080, 911, 746, $703 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{ONa}$ $\left[^{M}+\mathrm{Na}\right]^{+} 289.1568$, found 289.1600.

## 6-Methoxy-7,7-diphenylhept-1-ene (A6)



The secondary alcohol A5 ( $0.918 \mathrm{~g}, 3.45 \mathrm{mmol}$ ) in anhydrous DMF (20 mL ) at $0{ }^{\circ} \mathrm{C}$ was treated with $\mathrm{NaH}(60 \%$ dispersion in mineral oil, 0.345 $\mathrm{g}, 8.62 \mathrm{mmol})$ and the suspension was stirred at $0^{\circ} \mathrm{C}$ for 30 min . MeI $(0.86 \mathrm{~mL}, 13.8 \mathrm{mmol})$ was added dropwise and the cold bath was then removed. After stirred overnight, the reaction was quenched with water ( 40 mL ) cautiously and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \mathrm{x} 60 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The resulting residue was purified by flash chromatography ( $1 \%-5 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give the secondary alcohol $\mathbf{A 6}(0.921 \mathrm{~g}$, 95.4\%) as a colorless oil: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.45-7.20(\mathrm{~m}, 10 \mathrm{H}), 5.81(\mathrm{tdd}, J=16.9$, $13.1,6.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.05-4.96(\mathrm{~m}, 2 \mathrm{H}), 4.06(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.98-3.94(\mathrm{~m}, 1 \mathrm{H}), 3.22(\mathrm{~s}, 3 \mathrm{H})$, 2.15-1.95 (m, 2H), 1.70-1.43 (m, 4H); ${ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 143.1, 142.6, 129.0, 128.7, $128.4,126.5,126.4,114.7,83.6,58.0,56.3,34.0,31.7,24.5$; IR (neat) $2934,1640,1495,1451$, 1101, 910, 737, $701 \mathrm{~cm}^{-1}$; HRMS (EI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{O}\left(\mathrm{M}^{+\cdot}\right)$ 280.1827, found 280.1823.

## 5-Methoxy-6,6-diphenylhexanal (1.58)



At $-78{ }^{\circ} \mathrm{C}$, the terminal olefin $\mathbf{A 6}(400 \mathrm{mg}, 1.43 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(145$ mL ) was bubbled gently with ozone until the solution retained a deep blue color and then $\mathrm{PPh}_{3}(750 \mathrm{mg}, 2.86 \mathrm{mmol})$ was added in one portion. The mixture was warmed to room temperature, stirred overnight and then concentrated. The residue was purified by column chromatography ( $20 \%-30 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give the aldehyde $1.58(386 \mathrm{mg}$, $95.9 \%$ ) as a colorless oil: $\left.{ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta\right) \delta 9.71(\mathrm{t}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.39-7.16$ $(\mathrm{m}, 10 \mathrm{H}), 4.02(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.94-3.88(\mathrm{~m}, 1 \mathrm{H}), 3.12(\mathrm{~s}, 3 \mathrm{H}), 2.38(\mathrm{dt}, J=7.2,1.5 \mathrm{~Hz}$, 2H), 1.79-1.70(m, 2H), 1.58-1.53(m, 1H), 1.48-1.42(m, 1H), ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$
$202.2,142.7,142.3,128.9,128.7,128.6,128.4,126.6,126.5,83.5,58.1,56.2,44.0,31.7,17.9 ;$ IR (neat) 2929, 2825, 2721, 1722, 1495, 1451, 1112, 747, $704 \mathrm{~cm}^{-1} ;$ HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{ONa}[\mathrm{M}+\mathrm{Na}]^{+}$305.1517, found 305.1564.

## (E)-Ethyl 7-methoxy-8,8-diphenyloct-2-enoate (A7)

 added dropwise to a suspension of $\mathrm{NaH}(60 \%$ dispersion in mineral oil, $104 \mathrm{mg}, 2.60 \mathrm{mmol}$ ) in THF $(9.0 \mathrm{~mL})$ and the resulting solution was stirred at $0^{\circ} \mathrm{C}$ for 30 min . The aldehyde $1.58(367 \mathrm{mg}, 1.30 \mathrm{mmol}$, dissolved in 3.0 mL THF) was introduced dropwise and the flask formerly containing the aldehyde was rinsed with THF ( $2 \times 0.5 \mathrm{~mL}$ ). The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h , then quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}(10 \mathrm{~mL})$ and poured onto water $(20 \mathrm{~mL})$. The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$ and the organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $10 \%-15 \%$ EtOAc in hexanes) to give the ethyl ester A7 (426 mg, 93\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.44-7.20(\mathrm{~m}, 10 \mathrm{H}), 6.96(\mathrm{td}, J=15.6,6.8 \mathrm{~Hz}, 1 \mathrm{H})$, $5.82(\mathrm{~d}, J=15.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.24(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.05(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.98-3.92(\mathrm{~m}, 1 \mathrm{H})$, $3.21(\mathrm{~s}, 3 \mathrm{H}), 2.25-2.10(\mathrm{~m}, 2 \mathrm{H}), 1.71-1.44(\mathrm{~m}, 4 \mathrm{H}), 1.34(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 166.8,149.0,142.8,142.4,128.9,128.7,128.5,128.4,126.6,126.4,121.6,83.5,60.3$, 58.1, 56.3, 32.3, 31.8, 23.6, 14.4; IR (neat) 2979, 2934, 1718, 1654, 1495, 1451, 1368, 1269, 1186, 1098, 1043, 746, $703 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{O}_{3}\left(\mathrm{M}_{\left.-\mathrm{C}_{2} \mathrm{H}_{5} \mathrm{O}\right)^{+\cdot} \text { 307.1698, }}\right.$ found 307.1684.

## (3-(4-Methoxy-5,5-diphenylpentyl)oxiran-2-yl)methanol (A8)



At $-78{ }^{\circ} \mathrm{C}$, DIBAL-H ( 1 M in hexanes, $2.9 \mathrm{~mL}, 2.9 \mathrm{mmol}$ ) was added dropwise to a solution of the ethyl ester A7(413 mg, 1.17
mmol) in THF ( 12 mL ). The mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for 30 min , then quenched with saturated sodium tartrate $(12 \mathrm{~mL})$ and diluted with water $(10 \mathrm{~mL})$. The mixture was warmed to room temperature and stirred vigorously for 2 h . After that time, the mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{x} 40 \mathrm{~mL})$ and the extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was passed through a short silica gel column and eluted with $50 \%$ EtOAc in hexanes. The product was concentrated and dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(12 \mathrm{~mL})$ and cooled to $0{ }^{\circ} \mathrm{C} . \mathrm{NaHCO}_{3}(196 \mathrm{mg}, 2.34$ mmol ) and $m$ CPBA (pure, $262 \mathrm{mg}, 1.52 \mathrm{mmol}$ ) were added sequentially. The suspension was stirred at $0{ }^{\circ} \mathrm{C}$ for 3.5 h , then quenched with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(10 \mathrm{~mL})$ and water $(20 \mathrm{~mL})$. The mixture was stirred at room temperature for 20 min , and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$. The extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated and the residue was purified by column chromatography ( $50 \%-70 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the epoxy alcohol A8 (348 mg, 91\%, dr ~1:1) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.39-7.17(\mathrm{~m}, 10 \mathrm{H})$, $4.00(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.93-3.81(\mathrm{~m}, 2 \mathrm{H}), 3.61-3.53(\mathrm{~m}, 1 \mathrm{H}), 3.17 / 3.16(\mathrm{~s}, 3 \mathrm{H}), 2.89-2.84(\mathrm{~m}$, 2H), 1.66-1.40 (m, 6H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.8,142.4,128.9,128.6,128.5,128.4$, $126.5,126.4,83.5,61.8,58.6,58.5,58.1,58.0,56.2,56.0,55.9,32.0,31.7,21.6$; IR (neat) 3435 , 2929, 1599, 1495, 1451, 1098, 910, 732, $703 \mathrm{~cm}^{-1}$; HRMS (ESI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{21} \mathrm{H}_{26} \mathrm{O}_{3} \mathrm{Na}$ $[\mathrm{M}+\mathrm{Na}]^{+} 349.1780$, found 349.1792.
tert-Butyl (3-(4-methoxy-5,5-diphenylpentyl)oxiran-2-yl)methyl carbonate (1.48)


The epoxy alcohol A8 (336 mg, 1.03 mmol$)$ in anhydrous toluene $(10 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with 1-methylimidazole ( $82 \mu \mathrm{~L}, 1.03 \mathrm{mmol}$ ) followed by $\mathrm{Boc}_{2} \mathrm{O}(449 \mathrm{mg}, 2.06 \mathrm{mmol}$ ). The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min , then at room temperature for 1.5 h . After that time, the reaction was quenched with water $(15 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was azeotroped with hexanes ( $3 \times 10 \mathrm{~mL}$ ) and then purified by flash chromatography ( $15 \%-20 \%$ EtOAc in hexanes containing $\left.0.5 \% \mathrm{Et}_{3} \mathrm{~N}\right)$ to give the $t$-butyl carbonate $1.48(369 \mathrm{mg}, 84 \%, \mathrm{dr} \sim 1: 1)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.39-7.16(\mathrm{~m}, 10 \mathrm{H}), 4.23-4.18(\mathrm{~m}, 1 \mathrm{H}), 4.00(\mathrm{~d}, J=8.4 \mathrm{~Hz}$, $1 \mathrm{H}), 3.97-3.88(\mathrm{~m}, 2 \mathrm{H}), 3.16 / 3.16(\mathrm{~s}, 3 \mathrm{H}), 2.96-2.91(\mathrm{~m}, 1 \mathrm{H}), 2.81-2.79(\mathrm{~m}, 1 \mathrm{H}), 1.59-1.43(\mathrm{~m}$, $6 \mathrm{H}), 1.50(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 153.4,142.8,142.4,129.0,128.7,128.6,128.4$, $126.6,126.4,83.6,82.7,67.2,58.2,58.1,56.6,56.5,56.3,55.2,55.2,32.0,32.0,31.7,27.9,21.6$, 21.6; IR (neat) $2980,2936,1742,1495,1452,1370,1280,1163,1099,858,733,704 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{5} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 449.2304$, found 449.2308 .

## ((3,3-Dimethyloxiran-2-yl)methoxy)(tert-butyl)dimethylsilane (1.59)

 60.0 mmol ) followed by $m$-chloroperbenzoic acid ( $70-75 \%, 7.25 \mathrm{~g}, 31.5 \mathrm{mmol}$ ). The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1.5 h and then quenched with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ solution (100 $\mathrm{mL})$. After warmed to room temperature, the biphasic mixture was poured into water ( 100 mL ) and the two layers were separated. The aqueous layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 150 \mathrm{~mL})$ and the combined organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography $\left(15 \% \mathrm{Et}_{2} \mathrm{O}\right.$ in hexanes) to give the epoxide $\mathbf{1 . 5 9}$
( $6.312 \mathrm{~g}, 97.2 \%$ ) as a colorless liquid: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.75(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H})$, $3.74(\mathrm{~d}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.91(\mathrm{t}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}), 1.29(\mathrm{~s}, 3 \mathrm{H}), 0.92(\mathrm{~s}, 9 \mathrm{H}), 0.10(\mathrm{~s}$, $3 \mathrm{H}), 0.09(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 64.2,62.5,58.3,26.1,24.9,19.0,18.5,-5.0,-$ 5.2; IR (neat) 2958, 2930, 2886, 2858, 1472, 1379, 1256, 1140, 1086, 838, $778 \mathrm{~cm}^{-1}$.

## 1-(tert-Butyldimethylsilanyloxy)-3-methylbut-3-en-2-ol (A9)



A solution of 2,2,6,6-tetramethylpiperidine ( $2.825 \mathrm{~g}, 20.0 \mathrm{mmol}$ ) in anhydrous benzene $(12.0 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated dropwise with $n-\mathrm{BuLi}(1.6 \mathrm{M}$ in hexanes, $12.5 \mathrm{~mL}, 20.0 \mathrm{mmol}$ ). After 10 min , diethylaluminum chloride ( 1.0 M in heptanes, 20.0 $\mathrm{mL}, 20.0 \mathrm{mmol}$ ) was added dropwise and the resulting white suspension was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min . The epoxide $1.59(1.731 \mathrm{~g}, 8.00 \mathrm{mmol}$, dissolved in 8.0 mL of anhydrous benzene) was added dropwise and the flask formerly containing the epoxide was rinsed twice with benzene (2 x 4.0 mL ). The reaction mixture was stirred further for 1.5 h at $0^{\circ} \mathrm{C}$ and then quenched with saturated sodium tartrate solution ( 50 mL ). The biphasic mixture was poured onto water (100 $\mathrm{mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 50 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated in vacuo. The resulting residue was purified by column chromatography ( $15 \%$ $\mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give allylic alcohol $\mathbf{A 9}(1.5581 \mathrm{~g}, 90.0 \%)$ as a colorless liquid: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.04(\mathrm{~m}, 1 \mathrm{H}), 4.92(\mathrm{~m}, 1 \mathrm{H}), 4.12(\mathrm{~m}, 1 \mathrm{H}), 3.71(\mathrm{dd}, J=9.9,3.6 \mathrm{~Hz}, 1 \mathrm{H})$, $3.48(\mathrm{dd}, J=9.9,8.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.66(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.75(\mathrm{~s}, 3 \mathrm{H}), 0.92(\mathrm{~s}, 9 \mathrm{H}), 0.09(\mathrm{~s}, 6 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 144.0,112.0,75.4,66.4,26.0,19.0,18.4,-5.2$; IR (neat) 3446, 2955, 2929, 2858, 1472, 1256, 1113, 899, 836, $777 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{7} \mathrm{H}_{15} \mathrm{O}_{2} \mathrm{Si}$ $\left(\mathrm{M}-\mathrm{C}_{4} \mathrm{H}_{9}\right)^{+\bullet} 159.0841$, found 159.0811 .

## Ethyl (E)-6-(tert-butyldimethylsilanyloxy)-4-methylhex-4-enoate (A10)

 A mixture of allylic alcohol A9 ( $2.7093 \mathrm{~g}, 12.52 \mathrm{mmol}$ ), triethyl orthoacetate (freshly distilled, $9.2 \mathrm{~mL}, 50.1 \mathrm{mmol}$ ) and propionic acid $(46.4 \mathrm{mg}, 0.626 \mathrm{mmol})$ in a round-bottom flask was equipped with a fractional distillation apparatus to allow for removal of ethanol. The mixture was heated to $145^{\circ} \mathrm{C}$ for 4 h . (During this period of time, a lot of volatiles were distilled out.) The unreacted triethyl orthoacetate was removed by simple distillation and the residue was purified by column chromatography ( $5 \%$ EtOAc in hexanes) to give ethyl ester $\mathbf{A 1 0}$ (3.433 g, 95.7\%) as a colorless liquid: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.32(\mathrm{qt}, J=6.3,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.20(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.13(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H})$, 2.46-2.40(m, 2H), 2.37-2.30(m, 2H), $1.64(\mathrm{~s}, 3 \mathrm{H}), 1.26(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}), 0.90(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}$, $6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 173.5,135.3,125.3,60.5,60.4,34.6,33.0,26.2,18.6,16.6$, $14.5,-4.9$; IR (neat) $2956,2930,2857,1739,1472,1255,1158,1110,1068,836,776 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{29} \mathrm{O}_{3} \mathrm{Si}(\mathrm{M}-\mathrm{H})^{+\bullet} 285.1886$, found 285.1840 .

## (E)-7-(tert-Butyldimethylsilanyloxy)-5-methyl-1,1-diphenylhept-5-en-2-ol (A11)

 $\mathrm{mL}, 3.7 \mathrm{mmol}$ ). The reaction mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h and then quenched with saturated sodium tartrate solution ( 15 mL ). After warmed up to room temperature, the mixture was stirred vigorously for 30 min and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The crude aldehyde $\mathbf{1 . 6 0}$ was used in the next step without further purification.

In a separate two-necked round-bottom flask, a solution of diphenylmethane $(1.76 \mathrm{~g}, 10.5 \mathrm{mmol})$ in THF ( 10.0 ml ) was treated dropwise with $n-\operatorname{BuLi}(1.6 \mathrm{M}$ in hexanes, $6.5 \mathrm{~mL}, 10.5 \mathrm{mmol})$. The resulting deep orange solution was refluxed for 1 h , and then cooled to $0^{\circ} \mathrm{C}$. The as-prepared crude aldehyde (dissolved in 2.0 mL of THF) was added dropwise and the flask formerly containing the aldehyde was rinsed with THF ( $2 \times 0.5 \mathrm{~mL}$ ). The reaction mixture was stirred at 0 ${ }^{\circ} \mathrm{C}$ for 1 h , then warmed to room temperature and quenched by slow addition of saturated $\mathrm{NaHCO}_{3}$ solution $(10 \mathrm{~mL})$. The mixture was poured onto water $(20 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}$ ( $3 \times 40 \mathrm{~mL}$ ). The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $3 \%-12 \%$ EtOAc in hexanes) to give the secondary alcohol A11 ( $0.997 \mathrm{~g}, 69.6 \%$, two steps) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.42-$ 7.17 (m, 10H), $5.32(\mathrm{qt}, J=6.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.40-4.29(\mathrm{~m}, 1 \mathrm{H}), 4.18(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.92$ $(\mathrm{d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.30-2.20(\mathrm{~m}, 1 \mathrm{H}), 2.17-2.07(\mathrm{~m}, 1 \mathrm{H}), 1.66(\mathrm{~d}, J=3.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.70-1.44$ $(\mathrm{m}, 2 \mathrm{H}), 1.56(\mathrm{~s}, 3 \mathrm{H}), 0.92(\mathrm{~s}, 9 \mathrm{H}), 0.08(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta$ 142.6, 141.7, $136.9,129.0,128.8,128.4,127.0,126.7,125.0,73.5,60.4,59.0,35.9,33.1,26.2,18.6,16.4,-$ 4.8; IR (neat) 3458, 2954, 2928, 2856, 1599, 1494, 1451, 1386, 1254, 1112, 1067, 835, 776, 702 $\mathrm{cm}^{-1} ;$ HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{38} \mathrm{O}_{2} \mathrm{SiK}[\mathrm{M}+\mathrm{K}]^{+} 449.2278$, found 449.2287.

## (E)-6-Methoxy-3-methyl-7,7-diphenylhept-2-en-1-ol (A12)



The secondary alcohol A11 ( $0.908 \mathrm{~g}, 2.21 \mathrm{mmol}$ ) in anhydrous DMF ( 10.0 mL ) at $0{ }^{\circ} \mathrm{C}$ was treated with $\mathrm{NaH}(60 \%$ dispersion in mineral oil, $0.221 \mathrm{~g}, 5.52 \mathrm{mmol})$ and the suspension was stirred at $0^{\circ} \mathrm{C}$ for 30 min . MeI $(0.55$ $\mathrm{mL}, 8.84 \mathrm{mmol}$ ) was added dropwise and the reaction mixture was allowed to warm up to room temperature. After stirred for 3 h , the reaction was quenched with water ( 30 mL ) cautiously and
extracted with $\mathrm{Et}_{2} \mathrm{O}$ ( $3 \times 50 \mathrm{~mL}$ ). The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The resulting residue was dissolved in THF ( 11.0 mL ) and TBAF monohydrate $(0.693 \mathrm{~g}, 5.730 \mathrm{mmol})$ was added in one portion. The yellow solution was stirred for 1.5 h and then concentrated. The residue was purified by column chromatography (30\%-40\% EtOAc in hexanes) to give allylic alcohol A12 ( $0.684 \mathrm{~g}, 99.7 \%$, two steps) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.40-7.08(\mathrm{~m}, 10 \mathrm{H}), 5.37(\mathrm{qt}, J=6.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.12(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.02$ (d, $J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.92(\mathrm{ddd}, J=8.2,6.4,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.17(\mathrm{~s}, 3 \mathrm{H}), 2.22-2.03(\mathrm{~m}, 2 \mathrm{H}), 1.73-$ $1.48(\mathrm{~m}, 2 \mathrm{H}), 1.58(\mathrm{~s}, 3 \mathrm{H}), 1.30(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.9,142.5,139.7$, $129.0,128.7,128.4,126.6,126.4,123.8,83.4,59.5,58.1,56.4,35.1,30.5,16.4$; IR (neat) 3396 , 3026, 2929, 1599, 1494, 1451, 1374, 1241, 1102, 1002, 756, $703 \mathrm{~cm}^{-1} ;$ HRMS (ESI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{21} \mathrm{H}_{26} \mathrm{O}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 333.1831$, found 333.1817.

## (3-(3-Methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)methanol (A13)



A solution of allylic alcohol A12 (252 mg, 0.812 mmol$)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(8.0 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with $\mathrm{NaHCO}_{3}$ powder $(136 \mathrm{mg}, 1.62$ mmol ) followed by $m \mathrm{CPBA}$ (pure, $147 \mathrm{mg}, 0.852 \mathrm{mmol}$ ). The reaction mixture was stirred at 0 ${ }^{\circ} \mathrm{C}$ for 1 h and then quenched with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ solution ( 2.0 mL ). After warmed to room temperature, the biphasic mixture was poured onto water ( 5 mL ) and the two layers were separated. The aqueous layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 20 \mathrm{~mL})$ and the combined organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $40 \%-50 \%$ EtOAc in hexanes) to give the epoxy alcohol A13 (251 mg, 94.9\%, $\mathrm{dr} \sim 1: 1$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.38-7.17(\mathrm{~m}, 10 \mathrm{H}), 4.00-3.91(\mathrm{~m}$, $2 H), 3.81-3.73(\mathrm{~m}, 1 \mathrm{H}), 3.68-3.59(\mathrm{~m}, 1 \mathrm{H}), 3.16 / 3.14(\mathrm{~s}, 3 \mathrm{H}), 2.91-2.85(\mathrm{~m}, 1 \mathrm{H}), 1.80-1.38(\mathrm{~m}$,

4H), 1.19/1.17 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 142.7, 142.4, 142.3, 128.9, 128.8, 128.6, $128.4,126.7,126.5,83.4,83.3,63.0,62.7,61.5,61.5,61.4,61.4,58.1,57.9,56.4,56.2,33.9$, 33.7, 27.5, 27.2, 17.0, 16.7; IR (neat) 3418, 2931, 1599, 1495, 1452, 1385, 1099, 1032, 747, 704 $\mathrm{cm}^{-1} ;$ HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{21} \mathrm{H}_{26} \mathrm{O}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 349.1780$, found 349.1766.
tert-Butyl (3-(3-methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)methyl carbonate (1.49)


Epoxy alcohol A13 (192 mg, 0.588 mmol ) in anhydrous $(46.9 \mu \mathrm{~L}, 0.588 \mathrm{mmol})$ followed by $\mathrm{Boc}_{2} \mathrm{O}(321 \mathrm{mg}, 1.47 \mathrm{mmol})$. The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 4 h , then diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ and poured onto water $(10 \mathrm{~mL})$. The biphasic mixture was separated and the aqueous layer was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 15 \mathrm{~mL})$. The combined organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $10 \%-12.5 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the $t$-butyl carbonate $1.49(216 \mathrm{mg}, 86.3 \%, \mathrm{dr} \sim 1: 1)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.39-7.17(\mathrm{~m}, 10 \mathrm{H}), 4.20-4.14(\mathrm{~m}, 1 \mathrm{H}), 4.09(\mathrm{dd}, J=11.8,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.99-3.91(\mathrm{~m}$, $2 \mathrm{H}), 3.15 / 3.14(\mathrm{~s}, 3 \mathrm{H}), 2.97-2.91(\mathrm{~m}, 1 \mathrm{H}), 1.83-1.42(\mathrm{~m}, 4 \mathrm{H}), 1.50(\mathrm{~s}, 9 \mathrm{H}), 1.20 / 1.18(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 153.5,142.6,142.4,142.3,128.9,128.9,128.7,128.6,128.4,126.7$, $126.5,83.2,83.0,82.7,65.7,65.6,60.8,60.6,59.6,59.2,58.1,57.8,56.4,56.2,33.5,33.3,27.9$, 27.4, 27.1, 17.1, 16.7; IR (neat) 2980, 2933, 1743, 1495, 1453, 1370, 1327, 1279, 1163, 1098, 859, 738, $704 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{5} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$449.2304, found 449.2278.

## (E)-6-(tert-Butyldimethylsilanyloxy)-4-methylhex-4-enal (1.60)



Ethyl ester A10 (5.84 g, 20.4 mmol$)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(58.0 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ was treated dropwise with DIBAL-H (1.0 M in hexanes, 21.4 mL , 21.4 mmol ). The reaction mixture was stirred at $-78^{\circ} \mathrm{C}$ for 2 h and then quenched with saturated sodium tartrate solution ( 120 mL ). After warmed up to room temperature, the mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 70 \mathrm{~mL})$ and the organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $6 \%-8 \% \mathrm{EtOAc}$ in hexanes) to give the aldehyde $\mathbf{1 . 6 0}(4.31 \mathrm{~g}, 87.4 \%)$ as a colorless liquid: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.79$ $(\mathrm{t}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.33(\mathrm{qt}, J=6.2,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.19(\mathrm{~d}, J=6.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.59-2.54(\mathrm{~m}, 2 \mathrm{H})$, $2.35(\operatorname{app} \mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 1.65(\mathrm{~s}, 3 \mathrm{H}), 0.91(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 6 \mathrm{H}),{ }^{13} \mathrm{C} \mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 202.4,135.0,125.6,60.3,42.1,31.7,26.2,18.6,16.7,-4.9$; IR (neat) 2955, 2929, 2857, 2714, 1728, 1472, 1255, 1114, 1074, 836, $776 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{25} \mathrm{O}_{2} \mathrm{Si}(\mathrm{M}-\mathrm{H})^{+\bullet}$ 241.1624, found 241.1606.

## (E)-7-(tert-Butyldimethylsilanyloxy)-5-methylhept-5-enal (A14)

(Methoxymethyl)triphenylphosphonium chloride (2.587 g, 7.55
mmol $)$ in a dry flask under high vacuum was heated with heat gun for 3 min to remove the moisture. The flask was cooled to $0^{\circ} \mathrm{C}$ and THF ( 8.0 mL ) was added in. NaHMDS ( $7.55 \mathrm{~mL}, 7.55 \mathrm{mmol}$ ) was added dropwise and the resulting deep orange suspension was stirred at $0^{\circ} \mathrm{C}$ for 1 h . Aldehyde $1.60(0.610 \mathrm{~g}, 2.52 \mathrm{mmol}$, dissolved in 1.0 mL of THF) was added dropwise and the flask formerly containing the aldehyde was rinsed twice with THF ( 2 x 0.5 mL ). The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h , then quenched with saturated $\mathrm{NaHCO}_{3}$ $(10 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$,
filtered and concentrated. The residue was purified by flash chromatography (3.5\% EtOAc in hexanes) to give the crude methyl vinyl ether.

The methyl vinyl ether in THF- $\mathrm{H}_{2} \mathrm{O}(10: 1,40 \mathrm{~mL})$ was treated with $\mathrm{Hg}(\mathrm{OAc})_{2}(1.277 \mathrm{~g}, 4.01$ $\mathrm{mmol})$. The reaction mixture was stirred for 20 min and saturated $\mathrm{KI}(20 \mathrm{~mL})$ was added. The resulting yellowish green mixture was stirred for 1 h and then diluted with $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{~mL})$. The two layers were separated and the organic layer was washed with saturated KI ( 40 mL ), dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $4 \%$ 6\% EtOAc in hexanes) to give the title aldehyde $\mathbf{A 1 4}(0.526 \mathrm{~g}, 81.6 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.78(\mathrm{t}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.32(\mathrm{qt}, J=6.2,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.19(\mathrm{~d}, J=$ $6.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.42(\mathrm{dt}, J=7.3,1.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.04(\mathrm{t}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 1.76$ (pent, $J=7.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), $1.62(\mathrm{~s}, 3 \mathrm{H}), 0.91(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 202.6, 136.0, 125.9, 60.5, 43.5, 39.0, 26.3, 20.3, 18.7, 16.4, -4.8; IR (neat) 2930, 2856, 2713, 1728, 1472, 1387, 1255, 1115, 1080, 836, $776 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{28} \mathrm{O}_{2} \mathrm{SiNa}[\mathrm{M}+\mathrm{Na}]^{+}$279.1756, found 279.1745.

## (E)-8-(tert-Butyldimethylsilanyloxy)-6-methyl-1,1-diphenyl-oct-6-en-2-ol (A15)

Ph ${ }_{2} \mathrm{CH}$ A solution of diphenylmethane $(0.80 \mathrm{~mL}, 4.77 \mathrm{mmol})$ in THF ( 4.5 mL ) was treated with $n-\mathrm{BuLi}(1.6 \mathrm{M}$ in hexanes, 2.74 mL , 4.39 mmol ) and the resulting deep-orange solution was refluxed for 2 h . After cooling to room temperature, the solution was cooled further to $0^{\circ} \mathrm{C}$ and the aldehyde $\mathbf{A 1 4}(0.489 \mathrm{~g}, 1.91 \mathrm{mmol}$, dissolved in 1.0 mL of THF) was added dropwise. The flask formerly containing the aldehyde was rinsed with THF ( $2 \times 0.5 \mathrm{~mL}$ ). The deep-orange solution was stirred at $0^{\circ} \mathrm{C}$ for 1 h , then quenched by slow addition of saturated $\mathrm{NaHCO}_{3}(10 \mathrm{~mL})$. The biphasic mixture was diluted with
$\mathrm{Et}_{2} \mathrm{O}(10 \mathrm{~mL})$ and poured onto water $(10 \mathrm{~mL})$. The two layers were separated and the aqueous layer was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. The combined organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $7 \%-13 \%$ EtOAc in hexanes) to give the alcohol A15 (0.638 g, 78.7\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.41-7.16(\mathrm{~m}, 10 \mathrm{H}), 5.26(\mathrm{qt}, J=6.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.36(\mathrm{dt}, J=8.3,3.0 \mathrm{~Hz}, 1 \mathrm{H})$, $4.16(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 3.88(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.99-1.92(\mathrm{~m}, 2 \mathrm{H}), 1.70-1.35(\mathrm{~m}, 4 \mathrm{H}), 1.57(\mathrm{~s}$, $3 \mathrm{H}), 0.91(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.6,141.6,137.1,129.0,128.8$, $128.4,127.1,126.7,124.7,73.8,60.5,59.0,39.6,34.8,26.2,24.0,18.6,16.4,-4.8$; IR (neat) 3458, 2928, 2856, 1599, 1494, 1386, 1254, 1082, 835, $702 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{27} \mathrm{H}_{40} \mathrm{O}_{2} \mathrm{SiNa}[\mathrm{M}+\mathrm{Na}]^{+} 447.2695$, found 447.2741 .

## (E)-7-Methoxy-3-methyl-8,8-diphenyloct-2-en-1-ol (A16)

 The secondary alcohol A15 ( $0.638 \mathrm{~g}, 1.50 \mathrm{mmol}$ ) in anhydrous DMF ( 10 mL ) at $0{ }^{\circ} \mathrm{C}$ was treated with $\mathrm{NaH}(60 \%$ dispersion in mineral oil, $150 \mathrm{mg}, 3.75 \mathrm{mmol}$ ) and the yellow suspension was stirred at $0^{\circ} \mathrm{C}$ for 30 min . MeI ( $0.37 \mathrm{~mL}, 6.00 \mathrm{mmol}$ ) was added dropwise and the reaction mixture was stirred overnight at room temperature. The flask was cooled to $0{ }^{\circ} \mathrm{C}$ and the reaction was quenched with ice chips. The mixture was poured into water $(20 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was dissolved in THF $(7.5 \mathrm{~mL})$ and TBAF monohydrate $(0.471 \mathrm{~g}, 1.80 \mathrm{mmol})$ was added in . The yellow solution was stirred for 1.5 h and then concentrated in vacuo. The residue was purified by flash chromatography ( $25 \%-35 \%$ EtOAc in hexanes) to give the allylic alcohol A16 ( $0.475 \mathrm{~g}, 97.4 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.42-7.19(\mathrm{~m}, 10 \mathrm{H}), 5.36(\mathrm{qt}, J=6.9,1.0 \mathrm{~Hz}$,
$1 \mathrm{H}), 4.11(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 2 \mathrm{H}), 4.04(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.97-3.92(\mathrm{~m}, 1 \mathrm{H}), 3.19(\mathrm{~s}, 3 \mathrm{H}), 1.98-$ $1.96(\mathrm{~m}, 2 \mathrm{H}), 1.62(\mathrm{~s}, 3 \mathrm{H}), 1.58-1.41(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.9,142.5,139.4$, $128.9,128.6,128.4,126.5,126.4,123.7,83.6,59.3,58.0,56.2,39.5,31.6,22.9,16.2$; IR (neat) 3386, 2934, 1667, 1599, 1495, 1451, 1380, 1186, 1100, 1002, 746, $703 \mathrm{~cm}^{-1}$; HRMS (ESI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{O}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$347.1987, found 347.1966.

## (3-(4-Methoxy-5,5-diphenylpentyl)-3-methyloxiran-2-yl)methanol (A17)



A solution of allylic alcohol A16 ( $85.0 \mathrm{mg}, 0.262 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2.6 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with $\mathrm{NaHCO}_{3}$ powder (55.0 $\mathrm{mg}, 0.655 \mathrm{mmol}$ ) followed by $m$-chloroperbenzoic acid (pure, $47.5 \mathrm{mg}, 0.275 \mathrm{mmol}$ ). The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 50 min and then quenched with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ solution ( 2.0 mL ). After warmed to room temperature, the biphasic mixture was poured into water ( 5 mL ) and the two layers were separated. The aqueous layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 5 x 10 mL ) and the combined organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography (40\% - 45\% EtOAc in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the epoxy alcohol $\mathbf{A 1 7}(88.7 \mathrm{mg}, 99.4 \%, \mathrm{dr} \sim 1: 1)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.38-7.17(\mathrm{~m}, 10 \mathrm{H}), 4.00(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.91-3.89(\mathrm{~m}, 1 \mathrm{H}), 3.80-3.74$ (m, 1H), $3.65(\mathrm{dd}, J=12.1,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.17 / 3.16(\mathrm{~s}, 3 \mathrm{H}), 2.92-2.87(\mathrm{~m}, 1 \mathrm{H}), 1.64-1.34(\mathrm{~m}$, $6 \mathrm{H}), 1.23(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.8,142.4,128.9,128.9,128.7$, 128.6, 128.4, $126.6,126.4,83.6,83.5,63.1,63.0,61.5,61.4,58.1,58.0,56.3,38.6,32.1,32.0,20.7,20.6,16.8$, 16.7; IR (neat) $3420,2935,1599,1495,1452,1385,1249,1100,1031,862,747,704 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{22} \mathrm{H}_{28} \mathrm{O}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 363.1936$, found 363.1947.
tert-Butyl (3-(4-methoxy-5,5-diphenylpentyl)-3-methyloxiran-2-yl)methyl carbonate (1.50)


Epoxy alcohol A17 ( $72.9 \mathrm{mg}, 0.214 \mathrm{mmol}$ ) in anhydrous toluene $(2.0 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with 1 methylimidazole ( $22 \mu \mathrm{~L}, 0.278 \mathrm{mmol}$ ) followed by $\mathrm{Boc}_{2} \mathrm{O}(187 \mathrm{mg}, 0.856 \mathrm{mmol}$, dissolved in 0.5 mL of toluene). The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 4 h , then diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(5.0 \mathrm{~mL})$ and poured into water ( 6 mL ). The biphasic mixture was separated and the aqueous layer was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{x} 5 \mathrm{~mL})$. The combined organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $10 \%$ $15 \%$ EtOAc in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the $t$-butyl carbonate $\mathbf{1 . 5 0}(83.9 \mathrm{mg}$, $89.0 \%, \mathrm{dr} \sim 1: 1$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.39-7.17(\mathrm{~m}, 10 \mathrm{H}), 4.18(\mathrm{dd}, J$ $=11.9,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.09(\mathrm{dd}, J=11.9,6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.00(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.93-3.87(\mathrm{~m}, 1 \mathrm{H})$, $3.17(\mathrm{br} \mathrm{s}, 3 \mathrm{H}), 2.98-2.94(\mathrm{~m}, 1 \mathrm{H}), 1.64-1.41(\mathrm{~m}, 6 \mathrm{H}), 1.52(\mathrm{~s}, 9 \mathrm{H}), 1.24(\mathrm{br} \mathrm{s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR (75 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 153.4,142.8,142.4,128.9,128.6,128.5,128.4,126.5,126.4,83.5,82.6,65.7$, $60.6,59.5,59.4,58.1,58.0,56.2,38.3,32.0,27.9,20.5,20.5,16.8,16.8$; IR (neat) 2934, 1742, 1495, 1452, 1369, 1279, 1255, 1163, 1098, 859, $704 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{27} \mathrm{H}_{36} \mathrm{O}_{5} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 463.2460$, found 463.2462 .

## ((E)-3-Methyl-5-(3,3-dimethyloxiran-2-yl)pent-2-enyloxy)(tert-butyl)dimethylsilane (1.61)


((E)-3,7-dimethylocta-2,6-dienyloxy)(tert-butyl)dimethylsilane $(5.370 \mathrm{~g}, 20.0 \mathrm{mmol})$ in $\mathrm{CHCl}_{3}(180 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with $m$ CPBA $(70-75 \%, 5.621 \mathrm{~g}, 22.8 \mathrm{mmol})$ in small portions. The white suspension was stirred at 0 ${ }^{\circ} \mathrm{C}$ for 30 min , and then quenched with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ solution ( 20 mL ) and saturated $\mathrm{NaHCO}_{3}$ solution ( 100 mL ). The mixture was warmed up to room temperature and the two
layers were separated. The aqueous was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 100 \mathrm{~mL})$ and the combination of the organic extracts were dried over $\mathrm{MgSO}_{4}$ and evaporated. The residue was purified by column chromatography ( $8 \%-12 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give the desired monoepoxide $\mathbf{1 . 6 1}$ $(4.375 \mathrm{~g}, 76.9 \%)$ as a colorless oil: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.36(\mathrm{qt}, J=6.3,1.2 \mathrm{~Hz}, 1 \mathrm{H})$, $4.20(\mathrm{~d}, J=6.3, \mathrm{~Hz}, 1 \mathrm{H}), 4.20(\mathrm{~d}, J=6.3, \mathrm{~Hz}, 1 \mathrm{H}), 2.72(\mathrm{t}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.25-2.06(\mathrm{~m}, 2 \mathrm{H})$, $1.65(\mathrm{~s}, 3 \mathrm{H}), 1.71-1.61(\mathrm{~m}, 2 \mathrm{H}), 1.31(\mathrm{~s}, 3 \mathrm{H}), 1.27(\mathrm{~s}, 3 \mathrm{H}), 0.91(\mathrm{~s}, 9 \mathrm{H}), 0.08(\mathrm{~s}, 6 \mathrm{H})$.

## (E)-8-(tert-Butyldimethylsilanyloxy)-2,6-dimethylocta-1,6-dien-3-ol (A18)



A solution of 2,2,6,6-tetramethylpiperidine $(5.297 \mathrm{~g}, 37.5 \mathrm{mmol})$ in anhydrous benzene $(25.0 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ was treated dropwise with $n$ BuLi (1.6 M in hexanes, $23.4 \mathrm{~mL}, 37.5 \mathrm{mmol}$ ). After 10 min , diethylaluminum chloride ( 1.0 M in heptanes, $37.5 \mathrm{~mL}, 37.5 \mathrm{mmol}$ ) and the resulting white suspension was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min. Monoepoxide 1.61 ( $4.268 \mathrm{~g}, 15.0 \mathrm{mmol}$, dissolved in 5.0 mL of anhydrous benzene) was added dropwise and the flask formerly containing the epoxide was rinsed twice with benzene ( 2 x 2.5 mL ). The reaction mixture was stirred further at $0{ }^{\circ} \mathrm{C}$ for 1.5 h and then quenched with saturated sodium tartrate solution $(100 \mathrm{~mL})$. The biphasic mixture was poured into water (100 $\mathrm{mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 150 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated in vacuo. The resulting residue was purified by column chromatography (12\%$21 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give the allylic alcohol $\mathbf{A 1 8}$ (3.924 g, 91.9\%) as a colorless liquid: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 5.34(\mathrm{qt}, J=6.3,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.94(\mathrm{t}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.84(\mathrm{t}, J=1.5$ $\mathrm{Hz}, 1 \mathrm{H}), 4.19(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 4.05(\mathrm{t}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.17-1.95(\mathrm{~m}, 2 \mathrm{H}), 1.73(\mathrm{~s}, 3 \mathrm{H}), 1.64$ $(\mathrm{s}, 3 \mathrm{H}), 1.70-1.60(\mathrm{~m}, 2 \mathrm{H}), 0.90(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 147.6, $136.8,124.9,111.3,75.8,60.4,35.6,33.0,26.2,18.6,17.8,16.6,-4.8$; IR (neat) 3382,2929 ,

2857, 1472, 1382, 1255, 1112, 1070, 836, $776 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{12} \mathrm{H}_{23} \mathrm{O}_{2} \mathrm{Si}$ (M$\left.\mathrm{C}_{4} \mathrm{H}_{9}\right)^{+\bullet} 227.1467$, found 227.1450 .
(4E,8E)-Ethyl 10-(tert-butyldimethylsilanyloxy)-4,8-dimethyl-deca-4,8-dienoate (A19)


A mixture of allylic alcohol A18 (3.671 g, 12.90 mmol ), triethyl orthoacetate (freshly distilled, $10.46 \mathrm{~g}, 64.50$ $\mathrm{mmol})$ and propionic acid $(47.8 \mathrm{mg}, 0.645 \mathrm{mmol})$ in a round-bottom flask was equipped with a fractional distillation apparatus to allow for removal of ethanol. The mixture was heated to 145 ${ }^{\circ} \mathrm{C}$ for 1.5 h . (During this period of time, a lot of volatiles were distilled out.) The unreacted triethyl orthoacetate was removed by simple distillation and the residue was purified by column chromatography (3\% EtOAc in hexanes) to give the ethyl ester A19 (4.281 g, 93.6\%) as a colorless liquid: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.30(\mathrm{qt}, J=6.3,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{qt}, J=6.9$, $1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.19(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.19(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.12(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.42-$ $2.35(\mathrm{~m}, 2 \mathrm{H}), 2.31-2.26(\mathrm{~m}, 2 \mathrm{H}), 2.14-2.06(\mathrm{~m}, 2 \mathrm{H}), 2.02-1.97(\mathrm{~m}, 2 \mathrm{H}), 1.62(\mathrm{~s}, 3 \mathrm{H}), 1.61(\mathrm{~s}$, $3 \mathrm{H}), 1.25(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}), 0.90(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 173.7, $136.9,133.7,125.0,124.7,60.5,60.4,39.6,34.9,33.5,26.4,26.2,18.6,16.5,16.1,14.5,-4.8 ;$ IR (neat) $2929,2856,1738,1463,1254,1158,1063,836,776 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{20} \mathrm{H}_{38} \mathrm{O}_{3} \mathrm{SiNa}[\mathrm{M}+\mathrm{Na}]^{+}$377.2488, found 377.2513.

## (4E,8E)-10-(tert-Butyldimethylsilanyloxy)-4,8-dimethyldeca-4,8-dienal (1.62)



Ethyl ester A19 (2.3345g, 6.58 mmol$)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20.0 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ was treated dropwise with DIBAL-H $(1.0 \mathrm{M}$ in hexanes, $6.91 \mathrm{~mL}, 6.91 \mathrm{mmol})$. The reaction mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for 40 min and

DIBAL-H (1.0 M in hexanes, $0.66 \mathrm{~mL}, 0.66 \mathrm{mmol}$ ) were added. The mixture was stirred for 30 min more and then quenched with saturated sodium tartrate solution $(30 \mathrm{~mL})$. After warmed up to room temperature, the mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{x} 40 \mathrm{~mL})$ and the organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $4 \%-20 \%$ EtOAc in hexanes) to give the aldehyde $1.62(1.884 \mathrm{~g}, 92.1 \%)$ as a colorless liquid: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.75(\mathrm{t}, J=1.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.30(\mathrm{qt}, J=6.3,1.1 \mathrm{~Hz}$, $1 \mathrm{H}), 5.15(\mathrm{qt}, J=6.8,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.19(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.52(\mathrm{dt}, J=7.9,1.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.32$ $(\mathrm{t}, J=7.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.15-2.08(\mathrm{~m}, 2 \mathrm{H}), 2.03-1.98(\mathrm{~m}, 2 \mathrm{H}), 1.62(\mathrm{~s}, 6 \mathrm{H}), 0.91(\mathrm{~s}, 9 \mathrm{H}), 0.08(\mathrm{~s}, 6 \mathrm{H}) ;$ ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 202.8,136.8,133.4,125.3,124.8,60.5,42.4,39.5,32.0,26.4$, 26.2, 18.6, 16.5, 16.3, -4.8; IR (neat) 2928, 2856, 1728, 1472, 1386, 1254, 1110, 1066, 836, 776 $\mathrm{cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{35} \mathrm{O}_{2} \mathrm{Si}[\mathrm{M}+\mathrm{H}]^{+} 311.2406$, found 311.2386 .

## (5E,9E)-11-(tert-Butyldimethylsilanyloxy)-5,9-dimethyl-1,1-diphenylundeca-5,9-dien-2-ol

(A20)


A solution of diphenylmethane ( $1.625 \mathrm{~g}, 9.66 \mathrm{mmol}$ ) in THF ( 9.0 ml ) was treated dropwise with $n-\operatorname{BuLi}(1.6 \mathrm{M}$ in hexanes, $6.04 \mathrm{~mL}, 9.66 \mathrm{mmol}$ ). The resulting deep orange solution was stirred at $75^{\circ} \mathrm{C}$ for 1 h and cooled to $0^{\circ} \mathrm{C}$. Dienal $1.62(1.000 \mathrm{~g}, 3.22 \mathrm{mmol}$, dissolved in 2.0 mL of THF) was added dropwise and the flask formerly containing the aldehyde was rinsed with THF ( $2 \times 0.5 \mathrm{~mL}$ ). The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h and quenched by slow addition of saturated $\mathrm{NaHCO}_{3}$ solution ( 10 mL ). The mixture was poured onto water ( 15 mL ) and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30$ $\mathrm{mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $4 \%-10 \%$ EtOAc in hexanes) to give the secondary alcohol

A20 (1.243 g, 80.6\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.41-7.16(\mathrm{~m}, 10 \mathrm{H}), 5.30$ $(\mathrm{qt}, J=6.3,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\operatorname{app} \mathrm{t}, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.38-4.30(\mathrm{~m}, 1 \mathrm{H}), 4.18(\mathrm{~d}, J=6.3 \mathrm{~Hz}$, $2 \mathrm{H}), 3.91(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.20-2.00(\mathrm{~m}, 6 \mathrm{H}), 1.62(\mathrm{~s}, 3 \mathrm{H}), 1.52(\mathrm{~s}, 3 \mathrm{H}), 1.67-1.42(\mathrm{~m}, 2 \mathrm{H})$, 0.92 ( $\mathrm{s}, 9 \mathrm{H}$ ), $0.08(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.7,141.8,137.1,135.2,129.0$, $129.0,128.8,128.5,127.0,126.7,124.8,124.6,73.6,60.5,58.9,39.7,36.1,33.3,26.5,26.2$, 18.6, 16.6, 16.1, -4.8; IR (neat) 3466, 2928, 2855, 1598, 1494, 1450, 1384, 1254, 1067, 835, 776, $702 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{31} \mathrm{H}_{46} \mathrm{O}_{2} \mathrm{SiNa}[\mathrm{M}+\mathrm{Na}]^{+} 501.3165$, found 501.3150 .

## (2E,6E)-10-Methoxy-3,7-dimethyl-11,11-diphenylundeca-2,6-dien-1-ol (1.63)



The secondary alcohol A20 (1.200 g, 2.507 mmol ) in anhydrous DMF ( 14.0 mL ) at $0{ }^{\circ} \mathrm{C}$ was treated with NaH ( $60 \%$ dispersion in mineral oil, $0.251 \mathrm{~g}, 6.268 \mathrm{mmol}$ ) and the suspension was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min . MeI ( $0.62 \mathrm{~mL}, 10.0 \mathrm{mmol}$ ) was added dropwise and the reaction mixture was allowed to warm up to room temperature. After stirred for 3 h , the reaction was quenched with water ( 25 $\mathrm{mL})$ cautiously and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 35 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The resulting residue was dissolved in THF ( 12.5 mL ) and TBAF monohydrate $(0.787 \mathrm{~g}, 3.01 \mathrm{mmol})$ was added in one portion. The yellow solution was stirred for 1.3 h and then concentrated. The residue was purified by flash chromatography ( $20 \%$ $30 \%$ EtOAc in hexanes) to give the allylic alcohol 1.63 ( $0.923 \mathrm{~g}, 97.2 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.40-7.16(\mathrm{~m}, 10 \mathrm{H}), 5.40(\mathrm{qt}, J=6.9,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{t}, J=5.5$ $\mathrm{Hz}, 1 \mathrm{H}), 4.14(\mathrm{~d}, J=6.7 \mathrm{~Hz}, 2 \mathrm{H}), 4.02(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.92-3.87(\mathrm{~m}, 1 \mathrm{H}), 3.16(\mathrm{~s}, 3 \mathrm{H}), 2.11-$ $2.01(\mathrm{~m}, 6 \mathrm{H}), 1.68(\mathrm{~s}, 3 \mathrm{H}), 1.51(\mathrm{~s}, 3 \mathrm{H}), 1.62-1.46(\mathrm{~m}, 2 \mathrm{H}), 0.92(\mathrm{br} \mathrm{s}, 1 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 143.0,142.5,139.8,135.3,129.1,128.7,128.6,128.4,126.5,126.4,124.3,123.6,83.3$,
59.6, 58.0, 56.2, 39.7, 35.3, 30.8, 26.4, 16.5, 16.1; IR (neat) 3388, 3026, 2925, 1599, 1494, 1451, 1382, 1189, 1102, 1002, 755, $703 \mathrm{~cm}^{-1}$; HRMS (ESI): m/z calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 401.2457, found 401.2477.

## ((2R,3R)-3-(2-((2R,3R)-3-(3-Methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)ethyl)-3-methyloxiran-2-yl)methanol (A21)



To a solution of dienol $1.63(100 \mathrm{mg}, 0.264 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{DMM}(8.0 \mathrm{~mL}, 1: 2, \mathrm{v} / \mathrm{v})$ were added a 0.05 M solution of $\mathrm{Na}_{2} \mathrm{~B}_{4} \mathrm{O}_{7}$ in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}($ EDTA $)(5.2 \mathrm{~mL}), \mathrm{Bu}_{4} \mathrm{NHSO}_{4}(7.2 \mathrm{mg}, 21.1 \mu \mathrm{~mol})$ and Shi ketone ( $68.2 \mathrm{mg}, 0.264 \mathrm{mmol}$ ) sequentially. The mixture was cooled to $0^{\circ} \mathrm{C}$, and the Oxone ( 448 $\mathrm{mg}, 0.729 \mathrm{mmol}$ ), dissolved in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}($ EDTA $)(3.4 \mathrm{~mL})$, and $\mathrm{K}_{2} \mathrm{CO}_{3}(424 \mathrm{mg}, 3.06$ mmol ), dissolved in water ( 3.4 mL ), were added simultaneously via a syringe pump over 2.0 h . After the addition was completed, the slightly blue reaction mixture was stirred further for 15 min at $0{ }^{\circ} \mathrm{C}$, then diluted with water $(10 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $40 \%-80 \%$ EtOAc in hexanes) to give the diepoxy alcohol A21 ( 95.8 mg , $88.4 \%, \mathrm{dr} \sim 1: 1$ regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.42-7.17(\mathrm{~m}, 10 \mathrm{H}), 4.00-3.85$ (m, 2H), 3.81-3.61 (m, 2H), 3.16/3.14 (s, 3H), $2.96(\mathrm{t}, J=6.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.64-2.60(\mathrm{~m}, 1 \mathrm{H}), 2.01$ (br s, 1H), 1.86-1.73 (m, 2H), 1.66-1.42 (m, 6H), 1.31/1.30 ( $\mathrm{s}, 3 \mathrm{H}), 1.15 / 1.13(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.7,142.4,142.2,129.0,128.9,128.7,128.7,128.6,128.4,126.7,126.5$, $83.3,63.1,62.6,62.5,61.4,61.2,61.1,61.0,60.7,58.2,57.9,56.4,56.1,35.2,33.9,33.8,27.6$, 27.3, 24.4, 17.1, 16.8, 16.4; IR (neat) 3435, 3026, 2929, 1495, 1452, 1386, 1100, 1032, 747, 704
$\mathrm{cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 433.2355$, found 433.2348; $[\alpha]_{\mathrm{D}}=+13.3$ $\left(\mathrm{CHCl}_{3}, c 1.34\right)$.
tert-Butyl ((2R,3R)-3-(2-((2R,3R)-3-(3-methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)eth-yl)-3-methyloxiran-2-yl)methyl carbonate (1.51)


To a solution of diepoxy alcohol A21 (112 mg, $0.273 \mathrm{mmol})$ in anhydrous toluene $(2.7 \mathrm{~mL})$ at 0 ${ }^{\circ} \mathrm{C}$ were added $N$-methylimidazole ( $22 \mu \mathrm{~L}, 0.273 \mathrm{mmol}$ ) and $\mathrm{Boc}_{2} \mathrm{O}(119 \mathrm{mg}, 0.546 \mathrm{mmol})$ and the reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. After that time, the reaction was quenched with water $(10 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{x} 25 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was azeotroped with hexanes $(3 \times 10 \mathrm{~mL})$ and then purified by column chromatography $\left(20 \%-25 \% \mathrm{EtOAc}\right.$ in hexanes containing $\left.0.5 \% \mathrm{Et}_{3} \mathrm{~N}\right)$ to give the tert-butyl carbonate 1.51 (129 mg, 92.7\%, dr $\sim 1: 1$ regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.39-7.17(\mathrm{~m}, 10 \mathrm{H}), 4.27-4.10(\mathrm{~m}, 2 \mathrm{H}), 4.00-3.88(\mathrm{~m}, 2 \mathrm{H}), 3.17 / 3.16(\mathrm{~s}, 3 \mathrm{H}), 3.02(\mathrm{t}, J=5.8$ $\mathrm{Hz}, 1 \mathrm{H}), 2.63-2.59(\mathrm{~m}, 1 \mathrm{H}), 1.84-1.38(\mathrm{~m}, 8 \mathrm{H}), 1.52(\mathrm{~s}, 9 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}), 1.14 / 1.13(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 153.5,142.7,142.4,142.3,129.0,128.9,128.8,128.7,128.6,128.4$, $126.7,126.5,83.3,82.8,65.6,61.1,61.0,60.3,59.2,58.2,57.9,56.4,34.8,33.9,33.8,27.9,27.4$, $24.3,17.2,16.9,16.8,16.4$; IR (neat) $2978,2932,1743,1495,1453,1370,1279,1256,1163$, 1098, 858, 756, $704 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{31} \mathrm{H}_{42} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 533.2879$, found $533.2859 ;[\alpha]_{\mathrm{D}}=+17.3\left(\mathrm{CHCl}_{3}, c 1.49\right)$.
 A suspension of the activated $4 \AA$ molecular sieves powder ( 95 mg ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2.4 \mathrm{~mL})$ was treated with $(+)$ diisopropyl tartrate $(8.0 \mu \mathrm{~L}, 38.0 \mu \mathrm{~mol})$ and the mixture was cooled to -35 to $-30^{\circ} \mathrm{C}$. The mixture was stirred for 10 min and $\mathrm{Ti}(\mathrm{O}-i-\mathrm{Pr})_{4}(9.5 \mu \mathrm{~L}, 31.7 \mu \mathrm{~mol})$ was added and the mixture was stirred for 15 min more. After that time, $t$-butyl hydroperoxide ( $5.0-6.0 \mathrm{M}$ in decane, $0.19 \mathrm{~mL}, \sim 0.951$ mmol ) was added dropwise and the mixture was stirred for 30 min . Dienol 1.63 ( 120.0 mg , dissolved in 0.5 mL of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) was added dropwise and the flask formerly containing the dienol was rinsed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 0.1 \mathrm{~mL})$. The reaction mixture was stirred at -35 to $-30{ }^{\circ} \mathrm{C}$ for 40 min and then water $(0.5 \mathrm{~mL})$ was added. The mixture was allowed to warm up to $0^{\circ} \mathrm{C}$ and stirred for 1 h . A solution of $30 \%$ of NaOH saturated with $\mathrm{NaCl}(0.3 \mathrm{~mL})$ was added and the mixture was warmed up to room temperature and stirred for 1.5 h . The suspension was filtered through a pad of Celite and the filtrate was dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $30 \%-40 \%$ EtOAc in hexanes) to give the monoepoxy alcohol A22 (120.0 mg, 95.9\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.39-7.17(\mathrm{~m}, 10 \mathrm{H})$, $5.07(\operatorname{app} \mathrm{t}, J=6.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.01(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.92-3.86(\mathrm{~m}, 1 \mathrm{H}), 3.81-3.75(\mathrm{~m}, 1 \mathrm{H})$, 3.71-3.63 (m, 1H), $3.15(\mathrm{~s}, 3 \mathrm{H}), 2.94(\mathrm{dd}, J=6.5,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.11-2.04(\mathrm{~m}, 4 \mathrm{H}), 1.74-1.43(\mathrm{~m}$, $4 \mathrm{H}), 1.51(\mathrm{~s}, 3 \mathrm{H}), 1.29(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.9,142.5,135.8,135.7,129.0$, $128.6,128.4,126.5,126.4,123.7,123.6,83.3,83.2,63.1,61.6,61.2,58.0,57.8,56.2,38.6,35.2$, 30.7, 30.6, 23.8, 16.9, 16.0; IR (neat) $3424,3026,2930,1598,1494,1451,1384,1103,1032$, 862, 746, $704 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$417.2406, found 417.2436; $[\alpha]_{\mathrm{D}}=-3.8\left(\mathrm{CHCl}_{3}, c 1.08\right)$.

## ((2S,3S)-3-(2-((2R,3R)-3-(3-Methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)ethyl)-3-

 methyloxiran-2-yl)methanol (A23)

To a solution of monoepoxy alcohol A22 (170.0 mg, 0.431 mmol ) in $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{DMM}(6.5 \mathrm{~mL}, 1: 2, \mathrm{v} / \mathrm{v}$ ) were added a 0.05 M solution of $\mathrm{Na}_{2} \mathrm{~B}_{4} \mathrm{O}_{7}$ in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}($ EDTA $)(4.3 \mathrm{~mL}), \mathrm{Bu}_{4} \mathrm{NHSO}_{4}(5.8 \mathrm{mg}$, $17.2 \mu \mathrm{~mol})$ and Shi ketone ( $55.6 \mathrm{mg}, 0.216 \mathrm{mmol}$ ) sequentially. The mixture was cooled to $0{ }^{\circ} \mathrm{C}$, and the Oxone ( $366 \mathrm{mg}, 0.595 \mathrm{mmol}$ ), dissolved in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}$ (EDTA) $(2.8 \mathrm{~mL})$, and $\mathrm{K}_{2} \mathrm{CO}_{3}$ ( $346 \mathrm{mg}, 2.50 \mathrm{mmol}$ ), dissolved in water ( 2.8 mL ), were added simultaneously via a syringe pump over 2.0 h . After the addition was completed, the blue reaction mixture was stirred further for 15 min at $0^{\circ} \mathrm{C}$, then diluted with water $(15 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography (50\% - 80\% EtOAc in hexanes) to give the diepoxy alcohol A23 (160.6 $\mathrm{mg}, 90.8 \%, \mathrm{dr} \sim 1: 1$ regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups): ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.38-7.17(\mathrm{~m}, 10 \mathrm{H}), 3.99-3.95(\mathrm{~m}, 1 \mathrm{H})$, 3.93-3.87 (m, 1H), 3.83-3.73(m, 1H), 3.71-3.60(m, 1H), 3.15/3.14(s, 3H), 2.98-2.90(m, 1H), 2.67-2.60 (m, 1H), 2.35-2.28 (m, 1H), 1.91-1.86(m, 1H), 1.78-1.71 (m, 2H), 1.69-1.55 (m, 3H), 1.52-1.44 (m, 3H), $1.30(\mathrm{~s}, 3 \mathrm{H}), 1.16 / 1.14(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 142.6, 142.6, $142.3,142.2,128.9,128.8,128.7,128.6,128.4,126.6,126.5,83.3,83.1,63.6,63.0,63.0,61.4$, $61.2,61.0,60.8,58.2,57.8,56.2,56.0,36.1,34.0,33.7,27.6,27.1,24.7,16.9,16.5,16.4$; IR (neat) $3438,3026,2929,1495,1452,1385,1100,1032,746,704 \mathrm{~cm}^{-1} ;$ HRMS (ESI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 433.2355$, found 433.2346; $[\alpha]_{\mathrm{D}}=+19.3\left(\mathrm{CHCl}_{3}, c 1.55\right)$.
tert-Butyl ((2S,3S)-3-(2-((2R,3R)-3-(3-methoxy-4,4-diphenylbutyl)-3-methyloxiran-2-yl)eth-yl)-3-methyloxiran-2-yl)methyl carbonate (1.52)


To a solution of diepoxy alcohol A23 (145.1 mg,
$0.353 \mathrm{mmol})$ in anhydrous toluene $(3.5 \mathrm{~mL})$ at 0 ${ }^{\circ} \mathrm{C}$ were added 1-methylimidazole ( $28 \mu \mathrm{~L}, 0.353 \mathrm{mmol}$ ) and $\mathrm{Boc}_{2} \mathrm{O}(154 \mathrm{mg}, 0.706 \mathrm{mmol})$. The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. The reaction was quenched with water ( 15 mL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 25 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $20 \%-25 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the tertbutyl carbonate $1.52(154.8 \mathrm{mg}, 85.8 \%, \mathrm{dr} \sim 1: 1$ regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.41-7.18(\mathrm{~m}, 10 \mathrm{H}), 4.29-4.21(\mathrm{~m}, 1 \mathrm{H}), 4.19-4.11(\mathrm{~m}, 1 \mathrm{H}), 4.02-3.92(\mathrm{~m}, 2 \mathrm{H}), 3.18 / 3.17(\mathrm{~s}$, $3 \mathrm{H}), 3.04(\mathrm{dd}, J=6.2,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.63-2.59(\mathrm{~m}, 1 \mathrm{H}), 1.82-1.40(\mathrm{~m}, 8 \mathrm{H}), 1.53(\mathrm{~s}, 9 \mathrm{H}), 1.32(\mathrm{~s}$, $3 \mathrm{H}), 1.17 / 1.15(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 153.5,142.7$, 142.4, 142.2, 129.0, 128.7, $128.6,128.4,126.6,126.4,83.3,82.7,65.6,63.1,62.7,60.9,60.3,59.8,58.1,57.8,56.3,56.1$, $35.2,33.9,33.8,27.9,27.6,27.3,24.5,16.9,16.8,16.4$; IR (neat) 2977, 2932, 1742, 1495, 1453, $1370,1279,1256,1163,1097,858,704 \mathrm{~cm}^{-1} ;$ HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{31} \mathrm{H}_{42} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 533.2879, found 533.2857; $[\alpha]_{\mathrm{D}}=+1.3\left(\mathrm{CHCl}_{3}\right.$, c 2.15).
(5E,9E)-11-(tert-Butyldimethylsilanyloxy)-5,9-dimethylundeca-5,9-dienal (1.64)

(Methoxymethyl)triphenylphosphonium chloride (1.656 OTBS $\mathrm{g}, 4.830 \mathrm{mmol}$ ) in a dry flask under high vacuum was heated with heat gun for 3 min to remove the moisture. After the flask was cooled to room
temperature, THF ( 8.0 mL ) was added in and the suspension was cooled to $0^{\circ} \mathrm{C}$. $\mathrm{NaHMDS}(1 \mathrm{M}$ in THF, $4.83 \mathrm{~mL}, 4.83 \mathrm{mmol}$ ) was added dropwise and the resulting deep orange suspension was stirred at $0^{\circ} \mathrm{C}$ for 1 h . Dienal $1.62(0.500 \mathrm{~g}, 1.61 \mathrm{mmol}$, dissolved in 2.0 mL of THF) was added dropwise and the flask formerly containing the aldehyde was rinsed with THF ( $2 \times 1.0 \mathrm{~mL}$ ). The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h , then quenched with saturated $\mathrm{NaHCO}_{3}(10 \mathrm{~mL})$ and poured into water ( 20 mL ). The mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \mathrm{x} 30 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $3.5 \%$ EtOAc in hexanes) to give the methyl vinyl ether.

The methyl vinyl ether in $\mathrm{THF}-\mathrm{H}_{2} \mathrm{O}(10: 1,16.5 \mathrm{~mL})$ was treated with $\mathrm{Hg}(\mathrm{OAc})_{2}(0.558 \mathrm{~g}, 1.65$ $\mathrm{mmol})$. The reaction mixture was stirred for 30 min and saturated $\mathrm{KI}(30 \mathrm{~mL})$ was added. The resulting yellowish green mixture was stirred for 1 h , then diluted with $\mathrm{Et}_{2} \mathrm{O}(30 \mathrm{~mL})$. The two layers were separated and the aqueous layer was extracted with $\mathrm{Et}_{2} \mathrm{O}(2 \times 40 \mathrm{~mL})$. The combination of the organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $3 \%-5 \%$ EtOAc in hexanes) to give the titled aldehyde $(0.474 \mathrm{~g}, 90.7 \%)$ as a colorless oil: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.77(\mathrm{t}, J=1.7 \mathrm{~Hz}$, $1 \mathrm{H}), 5.30(\mathrm{qt}, J=6.3,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.12(\mathrm{qt}, J=6.9,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.20(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 2.39$ (dt, $J=7.3,1.7 \mathrm{~Hz}, 2 \mathrm{H}), 2.17-2.08(\mathrm{~m}, 2 \mathrm{H}), 2.01(\operatorname{app} \mathrm{t}, J=7.9 \mathrm{~Hz}, 4 \mathrm{H}), 1.74$ (pent, $J=7.3 \mathrm{~Hz}$, $2 \mathrm{H}), 1.62(\mathrm{~s}, 3 \mathrm{H}), 1.59(\mathrm{~s}, 3 \mathrm{H}), 0.91(\mathrm{~s}, 9 \mathrm{H}), 0.07(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 202.8$, $136.9,134.2,125.5,124.8,60.5,43.4,39.6,39.0,26.4,26.2,20.4,18.6,16.5,15.9,-4.8 ;$ IR (neat) 2929, 2856, 2712, 1728, 1472, 1386, 1254, 1110, 1068, 836, $776 \mathrm{~cm}^{-1}$; HRMS (ESI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{19} \mathrm{H}_{36} \mathrm{O}_{2} \mathrm{SiNa}[\mathrm{M}+\mathrm{Na}]^{+} 347.2382$, found 347.2402.


A solution of diphenylmethane $(0.66 \mathrm{~mL}, 3.93$ mmol ) in THF ( 3.9 mL ) was treated with $n-\mathrm{BuLi}$ ( 1.6 M in hexanes, $2.46 \mathrm{~mL}, 3.93 \mathrm{mmol}$ ) and the resulting deep-orange solution was refluxed for 1 h . After cooling to room temperature, the solution was cooled further to $0^{\circ} \mathrm{C}$ and the above aldehyde $1.64(0.426 \mathrm{~g}, 1.31 \mathrm{mmol}$, dissolved in 1.0 mL of THF) was added dropwise. The flask formerly containing the aldehyde was rinsed with THF ( $2 \times 0.5 \mathrm{~mL}$ ). The deep-orange solution was stirred at $0{ }^{\circ} \mathrm{C}$ for 2 h , then quenched with saturated $\mathrm{NaHCO}_{3}(5 \mathrm{~mL})$. The biphasic mixture was poured into water $(25 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 25 \mathrm{~mL})$. The combined organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $3 \%-9 \%$ EtOAc in hexanes) to give the secondary alcohol A24 ( 0.519 g , $80.3 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.38-7.13(\mathrm{~m}, 10 \mathrm{H}), 5.28(\mathrm{qt}, J=6.3$, $1.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.03(\mathrm{qt}, J=6.9,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.37-4.29(\mathrm{~m}, 1 \mathrm{H}), 4.18(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 2 \mathrm{H}), 3.86(\mathrm{~d}, J$ $=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.08-2.00(\mathrm{~m}, 2 \mathrm{H}), 1.96-1.85(\mathrm{~m}, 4 \mathrm{H}), 1.60(\mathrm{~s}, 3 \mathrm{H}), 1.52(\mathrm{~s}, 3 \mathrm{H}), 1.69-1.30(\mathrm{~m}$, 4H), $0.89(\mathrm{~s}, 9 \mathrm{H}), 0.06(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.8,141.7,137.1,135.2,129.0$, $129.0,128.8,128.5,127.0,126.7,124.6,124.3,73.9,60.6,59.0,39.7,39.6,34.8,26.5,26.2$, 24.2, 18.6, 16.6, 16.0, -4.8; IR (neat) 3467, 2928, 2856, 1599, 1494, 1451, 1384, 1253, 1107, 1065, 1005, 835, 775, $702 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{32} \mathrm{H}_{48} \mathrm{O}_{2} \mathrm{SiNa}[\mathrm{M}+\mathrm{Na}]^{+}$515.3321, found 515.3317.

## (2E,6E)-11-Methoxy-3,7-dimethyl-12,12-diphenyldodeca-2,6-dien-1-ol (1.65)


anhydrous DMF $(5.0 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ was treated with $\mathrm{NaH}(60 \%$ dispersion in mineral oil, 96.0 mg , $2.40 \mathrm{mmol})$ and the yellow suspension was stirred at $0^{\circ} \mathrm{C}$ for 30 min . MeI $(0.24 \mathrm{~mL}, 3.84 \mathrm{mmol})$ was added dropwise and the reaction mixture was stirred for 2.5 h at room temperature. The reaction was quenched with water ( 10 mL ) and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was dissolved in THF $(5.0 \mathrm{~mL})$ and TBAF monohydrate $(0.301 \mathrm{~g}, 1.15 \mathrm{mmol})$ was added in . The yellow solution was stirred for 2 h and then concentrated in vacuo. The residue was purified by flash chromatography (20\%-30\% EtOAc in hexanes) to give the dienol $1.65(0.368 \mathrm{~g}, 97.5 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 7.40-7.16(\mathrm{~m}, 10 \mathrm{H}), 5.40(\mathrm{qt}, J=6.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.05(\mathrm{qt}, J=6.8$, $1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.16(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.02(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.94-3.88(\mathrm{~m}, 1 \mathrm{H}), 3.18(\mathrm{~s}, 3 \mathrm{H})$, 2.13-1.98 (m, 4H), 1.93-1.90 (m, 2H), $1.68(\mathrm{~s}, 3 \mathrm{H}), 1.54(\mathrm{~s}, 3 \mathrm{H}), 1.58-1.30(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 143.1,142.6,139.7,135.4,129.1,128.7,128.6,128.4,126.5,126.4,124.2$, $123.7,83.8,59.6,58.0,56.3,39.8,39.7,31.7,26.4,23.4,16.4,16.0$; IR (neat) $3388,3026,2931$, 1667, 1599, 1495, 1451, 1382, 1186, 1102, 1003, 745, $703 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{27} \mathrm{H}_{36} \mathrm{O}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 415.2613$, found 415.2607 .

## ((2R,3R)-3-(2-((2R,3R)-3-(4-methoxy-5,5-diphenylpentyl)-3-methyloxiran-2-yl)ethyl)-3-

 methyloxiran-2-yl)methanol (A25)

To a solution of monoepoxy alcohol A24 (112 mg, 0.287 mmol ) in $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{DMM}(4.3 \mathrm{~mL}, 1: 2, \mathrm{v} / \mathrm{v}$ ) were added a 0.05 M solution of $\mathrm{Na}_{2} \mathrm{~B}_{4} \mathrm{O}_{7}$ in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}$ (EDTA) ( 2.9 mL ), $\mathrm{Bu}_{4} \mathrm{NHSO}_{4}(7.8 \mathrm{mg}$, $22.9 \mu \mathrm{~mol})$ and Shi ketone ( $74 \mathrm{mg}, 0.287 \mathrm{mmol}$ ) sequentially. The mixture was cooled to $0^{\circ} \mathrm{C}$, and the Oxone ( $486 \mathrm{mg}, 0.791 \mathrm{mmol}$ ), dissolved in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}$ (EDTA) $(3.7 \mathrm{~mL})$, and $\mathrm{K}_{2} \mathrm{CO}_{3}$
( $460 \mathrm{mg}, 3.32 \mathrm{mmol}$ ), dissolved in water ( 3.7 mL ), were added simultaneously via a syringe pump over 1.5 h . After the addition was completed, the blue reaction mixture was stirred further for 15 min at $0{ }^{\circ} \mathrm{C}$, then diluted with water $(20 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $35 \%-85 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give diepoxy alcohol A25 (78 mg, 64\%, dr $\sim 1: 1$ regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 7.62-7.16 $(\mathrm{m}, 10 \mathrm{H}), 4.00(J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.92-3.83(\mathrm{~m}, 1 \mathrm{H}), 3.82-3.79(\mathrm{~m}, 1 \mathrm{H}), 3.75-3.64(\mathrm{~m}, 1 \mathrm{H})$, 3.17/3.16 (s, 3H), 3.01-2.96 (m, 1H), 2.68-2.63 (m, 1H), 1.96-1.70 (m, 2H), 1.64-1.40 (m, 8H), $1.33(\mathrm{~s}, 3 \mathrm{H}), 1.20 / 1.19(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.8,142.4,128.9,128.9,128.7$, $128.6,128.4,126.6,126.4,83.6,63.0,62.9,62.6,61.4,61.2,61.1,60.7,58.1,58.0,56.2,38.8$, 38.7, 35.2, 32.1, 24.4, 20.8, 20.7, 17.0, 16.5, 16.5; IR (neat) 3433, 2930, 1599, 1495, 1452, 1386, $1102,1032,734,704 \mathrm{~cm}^{-1}$.
tert-Butyl ((2R,3R)-3-(2-((2R,3R)-3-(4-methoxy-5,5-diphenylpentyl)-3-methyloxiran-2-yl)et-hyl)-3-methyloxiran-2-yl)methyl carbonate (1.53)


The diepoxy alcohol A25 (74 mg, 0.174 mmol$)$ was dissolved in dry toluene ( 1.7 mL ) and cooled to $0{ }^{\circ} \mathrm{C}$. $N$-Methylimidazole ( $18 \mu \mathrm{~L}, 0.226 \mathrm{mmol}$ ) and $\mathrm{Boc}_{2} \mathrm{O}(152 \mathrm{mg}, 0.696 \mathrm{mmol})$ were added sequentially. The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. The reaction was quenched with water ( 10 mL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by column chromatography ( $20 \%-25 \%$ EtOAc in hexanes containing $0.5 \%$
$\mathrm{Et}_{3} \mathrm{~N}$ ) to give the desired product $1.53(72 \mathrm{mg}, 78 \%$, dr $\sim 1: 1$ regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.41-7.18(\mathrm{~m}, 10 \mathrm{H}), 4.30-4.14(\mathrm{~m}, 2 \mathrm{H}), 4.02(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.98-3.85$ (m, 1H), 3.19/3.18 (s, 3H), $3.05(\mathrm{t}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.66-2.63(\mathrm{~m}, 1 \mathrm{H}), 1.81-1.39(\mathrm{~m}, 8 \mathrm{H}), 1.52$ $(\mathrm{s}, 9 \mathrm{H}), 1.35(\mathrm{~s}, 3 \mathrm{H}), 1.20 / 1.19(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 153.5,142.9,142.5,142.4$, $129.0,128.9,128.7,128.6,128.4,126.6,126.4,83.7,83.6,82.7,65.6,62.9,62.8,61.0,60.3$, $59.3,58.1,58.0,56.3,38.8,38.8,34.8,32.2,27.9,24.4,20.9,20.8,17.1,16.5,16.5$; IR (neat) 3026, 2934, 1743, 1495, 1453, 1370, 1279, 1256, 1163, 1098, 859, 747, $705 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{32} \mathrm{H}_{44} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 547.3036$, found $547.3002 ;[\alpha]_{\mathrm{D}}=+11.0\left(\mathrm{CHCl}_{3}, c 2.01\right)$.

## ((2S,3S)-3-((E)-8-Methoxy-4-methyl-9,9-diphenylnon-3-enyl)-3-methyloxiran-2-yl)methanol

 (A26)

A suspension of the activated $4 \AA$ molecular sieves powder ( 115.0 mg ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3.3 \mathrm{~mL})$ was treated with $(+)$-diisopropyl tartrate $(9.6 \mu \mathrm{~L}, 45.8 \mu \mathrm{~mol})$ and the mixture was cooled to -35 to $-30{ }^{\circ} \mathrm{C}$. The mixture was stirred for 10 min and $\mathrm{Ti}(\mathrm{O}-i-\mathrm{Pr})_{4}(11.4 \mu \mathrm{~L}, 38.2 \mu \mathrm{~mol})$ was added and the mixture was stirred for 15 min more. After that time, $t$-butyl hydroperoxide (5.0-6.0 M in decane, $0.23 \mathrm{~mL}, \sim 1.15 \mathrm{mmol}$ ) was added dropwise and the mixture was stirred for 30 min . Dienol $\mathbf{1 . 6 5}$ ( $150.0 \mathrm{mg}, 0.382 \mathrm{mmol}$, dissolved in 1.0 mL of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) was added dropwise and the flask formerly containing the allylic alcohol was rinsed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 0.5 \mathrm{~mL})$. The reaction mixture was stirred at -35 to $-30{ }^{\circ} \mathrm{C}$ for 1.5 h and then water $(0.6 \mathrm{~mL})$ was added. The mixture was allowed to warm up to $0^{\circ} \mathrm{C}$ and stirred for 1 h . A solution of $30 \%$ of NaOH saturated with $\mathrm{NaCl}(0.2 \mathrm{~mL})$ was added and the mixture was warmed up to room temperature and stirred for
1.5 h . The suspension was filtered through a pad of Celite and the filtrate was dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography (30\%-40\% EtOAc in hexanes) to give the monoepoxy alcohol A26 (156.0 mg, 97.3\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 7.44-7.23(\mathrm{~m}, 10 \mathrm{H}), 5.09(\mathrm{t}, J=6.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.06(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.99-$ $3.93(\mathrm{~m}, 1 \mathrm{H}), 3.86-3.83(\mathrm{~m}, 1 \mathrm{H}), 3.70(\mathrm{dd}, J=12.0,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.22(\mathrm{~s}, 3 \mathrm{H}), 3.02-2.99(\mathrm{~m}$, 1H), 2.46 (br s, 1H), $2.10(\mathrm{q}, ~ J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 1.96-1.91(\mathrm{~m}, 2 \mathrm{H}), 1.72-1.43(\mathrm{~m}, 6 \mathrm{H}), 1.59(\mathrm{~s}$, $3 \mathrm{H}), 1.33(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 143.0,142.5,135.7,128.9,128.6,128.3,126.4$, $126.3,123.5,83.6,63.2,61.5,61.2,57.9,56.2,39.6,38.6,31.6,23.6,23.2,16.9,15.9$; IR (neat) 3425, 3027, 2934, 1599, 1495, 1452, 1385, 1103, 1032, 910, 733, $703 \mathrm{~cm}^{-1}$; HRMS (ESI): m/z calcd for $\mathrm{C}_{27} \mathrm{H}_{36} \mathrm{O}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 431.2562$, found 431.2556; $[\alpha]_{\mathrm{D}}=-3.8\left(\mathrm{CHCl}_{3}, c\right.$ 1.17 $)$.
tert-Butyl ((2S,3S)-3-(2-((2R,3R)-3-(4-methoxy-5,5-diphenylpentyl)-3-methyloxiran-2-yl)et-hyl)-3-methyloxiran-2-yl)methyl carbonate (1.54)


A solution of monoepoxy alcohol A26 (145.0 $\mathrm{mg}, 0.3549 \mathrm{mmol}$ ) in $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{DMM}(5.3 \mathrm{~mL}$, 1:2, $\mathrm{v} / \mathrm{v}$ ) was treated a 0.05 M solution of $\mathrm{Na}_{2} \mathrm{~B}_{4} \mathrm{O}_{7}$ in $4 \times 10^{-4} \mathrm{M} \mathrm{Na} 2$ (EDTA) ( 3.5 mL ), $\mathrm{Bu}_{4} \mathrm{NHSO}_{4}(4.8 \mathrm{mg}, 14.2 \mu \mathrm{~mol})$ and Shi ketone $(45.8 \mathrm{mg}, 0.177 \mathrm{mmol})$ sequentially. The mixture was cooled to $0^{\circ} \mathrm{C}$, and the Oxone ( $301 \mathrm{mg}, 0.490 \mathrm{mmol}$ ), dissolved in $4 \times 10^{-4} \mathrm{M}$ $\mathrm{Na}_{2}$ (EDTA) $(2.3 \mathrm{~mL})$, and $\mathrm{K}_{2} \mathrm{CO}_{3}(284 \mathrm{mg}, 2.06 \mathrm{mmol})$, dissolved in water ( 2.3 mL ), were added simultaneously via a syringe pump over 2.0 h . After the addition was completed, the blue reaction mixture was stirred further for 15 min at $0^{\circ} \mathrm{C}$, then diluted with water ( 5 mL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( $3 \times 25 \mathrm{~mL}$ ). The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was dissolved in dry toluene ( 3.5 mL ) and cooled to $0{ }^{\circ} \mathrm{C} . \mathrm{N}$ -

Methylimidazole ( $36.8 \mu \mathrm{~L}, 0.4614 \mathrm{mmol}$ ) and $\mathrm{Boc}_{2} \mathrm{O}(232 \mathrm{mg}, 1.06 \mathrm{mmol})$ were added sequentially. The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. The reaction was quenched with water ( 5 mL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was azeotroped with hexanes ( $3 \times 20 \mathrm{~mL}$ ) to removed $t-\mathrm{BuOH}$ and the residue was purified by column chromatography ( $12 \%-24 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the desired product $1.54(153.5 \mathrm{mg}, 82.4 \%, \mathrm{dr} \sim 1: 1$ regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.43-7.20(\mathrm{~m}, 10 \mathrm{H}), 4.29(\mathrm{dd}, J=11.9,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.17(\mathrm{dd}, J=11.9,6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.04(\mathrm{~d}, J$ $=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.99-3.90(\mathrm{~m}, 1 \mathrm{H}), 3.20 / 3.19(\mathrm{~s}, 3 \mathrm{H}), 3.07(\mathrm{t}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.67-2.62(\mathrm{~m}, 1 \mathrm{H})$, $1.77-1.42(\mathrm{~m}, 10 \mathrm{H}), 1.54(\mathrm{~s}, 9 \mathrm{H}), 1.35(\mathrm{~s}, 3 \mathrm{H}), 1.22(\mathrm{br} \mathrm{s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $153.4,142.8,142.4,142.4,128.9,128.8,128.6,128.5,128.3,126.5,126.4,83.5,83.5,82.6,65.6$, $63.0,62.9,60.8,60.8,60.3,59.7,58.0,57.9,56.2,38.7,38.7,35.1,32.1,27.8,24.4,20.8,20.6$, $16.8,16.5,16.4$; IR (neat) $2979,2935,1743,1495,1453,1370,1279,1163,1098,912,859,733$, $704 \mathrm{~cm}^{-1} ;$ HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{32} \mathrm{H}_{44} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$547.3036, found 547.3031; $[\alpha]_{\mathrm{D}}=$ $+0.5\left(\mathrm{CHCl}_{3}, c 1.21\right)$.

## 6,6-Diphenylhexane-1,5-diol (A27)

 In a round-bottomed flask, the $\delta$-valerolactone ( $1.82 \mathrm{~mL}, 20.0 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20.0 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ was treated dropwise with DIBAL-H (1 M in hexanes, $22.0 \mathrm{~mL}, 22.0 \mathrm{mmol}$ ). The mixture was stirred at $-78^{\circ} \mathrm{C}$ for 1 h , then quenched with saturated sodium tartrate $(80 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 100 \mathrm{~mL})$. The extracts were
dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The crude lactol was used in the next step without further purification.

In another two-necked round-bottomed flask, diphenylmethane ( $13.4 \mathrm{~mL}, 80.0 \mathrm{mmol}$ ) in THF $(50.0 \mathrm{~mL})$ was treated dropwise with $n$ - BuLi ( 1.6 M in hexanes, $50.0 \mathrm{~mL}, 80.0 \mathrm{mmol}$ ). The resulting deep orange solution was refluxed for 1 h and then cooled to $0^{\circ} \mathrm{C}$. The as-prepared crude lactol (dissolved in 5 mL THF) was added dropwise and the flask formerly containing the lactol was rinsed with THF ( $2 \times 1.5 \mathrm{~mL}$ ). The deep-orange mixture was stirred at $0^{\circ} \mathrm{C}$ for 0.5 h , then quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}(80 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \mathrm{x} 100 \mathrm{~mL})$. The combined extracts were dried over $\mathrm{MgSO}_{4}$ and evaporated. The resulting residue was purified by column chromatography ( $40 \%-80 \%$ EtOAc in hexanes) to give the diol A27 (4.42 g, 81.8\%) as a white solid: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.57-7.18(\mathrm{~m}, 10 \mathrm{H}), 4.38(\mathrm{app} \mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H})$, $3.90(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.59(\mathrm{t}, J=5.7 \mathrm{~Hz}, 2 \mathrm{H}), 1.88(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 1.81(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 1.67-1.42(\mathrm{~m}$, $6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.6,141.7,129.0,129.0,128.8,128.4,127.0,126.7,73.9$, $62.7,59.0,34.6,32.6,22.0$; IR (neat) $3358,3023,2948,1596,1493,1450,1345,1039,696 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{O}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$293.1517, found 293.1537.

## 6-(1-Phenyl-1H-tetrazol-5-ylthio)-1,1-diphenylhexan-2-ol (A28)


$(1.164 \mathrm{~g}, 4.44 \mathrm{mmol})$ in THF $(30.0 \mathrm{~mL})$ was added dropwise diisopropyl azodicarboxylate $(0.82$ $\mathrm{mL}, 4.07 \mathrm{mmol}$ ). The mixture was warmed to room temperature and stirred for 20 min . The reaction was quenched with water ( 60 mL ) and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 70 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$ and evaporated. The resulting residue was purified by column
chromatography ( $25 \%-40 \%$ EtOAc in hexanes) to give the crude product which was further purified by column chromatography ( $3.5 \% \mathrm{Et}_{2} \mathrm{O}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give $\mathbf{A 2 8}(1.542 \mathrm{~g}, 96.8 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.60-7.50(\mathrm{~m}, 5 \mathrm{H}), 7.40-7.15(\mathrm{~m}, 10 \mathrm{H}), 4.39-4.33$ $(\mathrm{m}, 1 \mathrm{H}), 3.86(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.35(\mathrm{t}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 1.86-1.38(\mathrm{~m}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 154.6,142.4,141.5,133.8,130.2,129.9,129.0,128.9,128.8,128.3,127.0$, $126.8,124.0,73.6,59.1,34.4,33.4,29.1,25.0$; IR (neat) 3439, 3026, 2942, 1597, 1499, 1451, 1387, 1243, 1088, 1015, 910, 760, 733, $704 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{25} \mathrm{H}_{26} \mathrm{~N}_{4} \mathrm{OSNa}$ $[\mathrm{M}+\mathrm{Na}]^{+} 453.1725$, found 453.1705 .

## 5-(5-Methoxy-6,6-diphenylhexylthio)-1-phenyl-1H-tetrazole (A29)

 in mineral oil, $0.335 \mathrm{~g}, 8.38 \mathrm{mmol}$ ) and the yellow suspension was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min . MeI ( $0.84 \mathrm{~mL}, 13.42 \mathrm{mmol}$ ) was added dropwise and the reaction mixture was stirred for 4 h at room temperature. The reaction was quenched with water ( 60 mL ) and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \mathrm{x}$ 50 mL ). The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $13 \%-20 \%$ EtOAc in hexanes) to give the desired product A29 (1.268 g, 85.0\%) as a pale yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.59-7.56(\mathrm{~m}, 5 \mathrm{H})$, 7.38-7.17 (m, 10H), $3.98(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.93-3.88(\mathrm{~m}, 1 \mathrm{H}), 3.34(\mathrm{t}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.15(\mathrm{~s}$, $3 \mathrm{H}), 1.86-1.70(\mathrm{~m}, 2 \mathrm{H}), 1.65-1.40(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 154.6,142.8,142.4$, $133.9,130.2,129.9,128.9,128.7,128.6,128.4,126.6,126.4,124.0,83.5,58.1,56.3,33.4,31.7$, 29.3, 24.2; IR (neat) $3026,2938,1597,1498,1451,1386,1242,1102,759,697 \mathrm{~cm}^{-1} ;$ HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{28} \mathrm{~N}_{4} \mathrm{OSNa}[\mathrm{M}+\mathrm{Na}]^{+} 467.1882$, found 467.1876 .

## 5-(5-Methoxy-6,6-diphenylhexylsulfonyl)-1-phenyl-1H-tetrazole (1.66)

 and then $m \mathrm{CPBA}$ (pure, $435 \mathrm{mg}, 2.52 \mathrm{mmol}$ ) was added in small portions. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 15 min , then at room temperature overnight. The reaction was quenched with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ solution ( 20 mL ), stirred for 30 min and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 25 \mathrm{~mL})$. The extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $16 \%-24 \%$ EtOAc in hexanes) to give the desired sulfone 1.66 ( 336 mg , 95.0\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.70-7.60(\mathrm{~m}, 5 \mathrm{H}), 7.38-7.17(\mathrm{~m}, 10 \mathrm{H})$, $3.99(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.94-3.88(\mathrm{~m}, 1 \mathrm{H}), 3.72-3.64(\mathrm{~m}, 2 \mathrm{H}), 3.15(\mathrm{~s}, 3 \mathrm{H}), 1.96-1.84(\mathrm{~m}, 2 \mathrm{H})$, 1.68-1.53 (m, 3H), 1.49-1.43 (m, 1H); ${ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 153.7,142.6,142.3,133.2$, $131.6,129.9,128.9,128.8,128.6,128.5,126.7,126.6,125.3,83.4,58.2,56.4,56.1,31.7,23.9$, 22.3; IR (neat) $3027,2934,2828,1597,1496,1452,1342,1153,1103,762,705 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{28} \mathrm{~N}_{4} \mathrm{O}_{3} \mathrm{SNa}[\mathrm{M}+\mathrm{Na}]^{+}$499.1780, found 499.1787.
((2E,6E)-11-Methoxy-3-methyl-12,12-diphenyldodeca-2,6-dienyloxy)(tert-butyl)dimethylsilane (A30)

A solution of sulfone 1.66 (300 $\mathrm{mg}, 0.630 \mathrm{mmol}$,
azeotropically dried with benzene $)$ in anhydrous $1,2-$ dimethoxyethane ( 3.8 mL ) at $-78{ }^{\circ} \mathrm{C}$ was treated dropwise with KHMDS ( 0.5 M in $1,2-$ dimethoxyethane, $1.51 \mathrm{~mL}, 0.755 \mathrm{mmol}$ ) and the resulting yellow mixture was stirred at this temperature for 1 h . After that time, aldehyde $\mathbf{1 . 6 0}(183 \mathrm{mg}, 0.755 \mathrm{mmol}$, dissolved in 0.5 mL of

1,2-dimethoxyethane) was added dropwise. The reaction mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for 1.5 h , then at room temperature overnight. The reaction was quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution (5 $\mathrm{mL})$, poured onto water $(10 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{~mL})$. The extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography ( $2 \%$ $3 \%$ EtOAc in hexanes) to give the desired diene $\mathbf{A 3 0}$ ( $196.4 \mathrm{mg}, 63.3 \%$ ) as a pale yellow oil: ${ }^{1} \mathrm{H}$ NMR (500 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 7.38-7.17(\mathrm{~m}, 10 \mathrm{H}), 5.35-5.33(\mathrm{~m}, 2 \mathrm{H}), 5.30(\mathrm{qt}, J=6.3,1.0 \mathrm{~Hz}, 1 \mathrm{H})$, $4.20(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.00(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.91-3.88(\mathrm{~m}, 1 \mathrm{H}), 3.17(\mathrm{~s}, 3 \mathrm{H}), 2.09-2.05(\mathrm{~m}$, $2 \mathrm{H}), 2.20-1.99(\mathrm{~m}, 2 \mathrm{H}), 1.92-1.91(\mathrm{~m}, 2 \mathrm{H}), 1.61(\mathrm{~s}, 3 \mathrm{H}), 1.52-1.48(\mathrm{~m}, 2 \mathrm{H}), 1.45-1.40(\mathrm{~m}, 2 \mathrm{H})$, 0.92 ( $\mathrm{s}, 9 \mathrm{H}$ ), $0.08(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 143.2,142.6,136.9,130.4,130.3$, $129.1,128.7,128.6,128.4,126.5,126.4,124.7,83.8,60.5,58.0,56.3,39.8,32.7,31.7,31.1$, 26.2, 25.2, 18.6, 16.6, -4.8; IR (neat) 3027, 2928, 2855, 1599, 1495, 1451, 1381, 1254, 1105, 1062, 836, 775, $701 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{28} \mathrm{H}_{39} \mathrm{OSi}\left(\mathrm{M}_{-\mathrm{C}}^{4} \mathrm{H}_{9}\right)^{+\cdot} 435.2719$, found 435.2706 .

## (2E,6E)-11-Methoxy-3-methyl-12,12-diphenyldodeca-2,6-dien-1-ol (A31)

 monohydrate ( $125 \mathrm{mg}, 0.478 \mathrm{mmol}$ ). The yellow solution was stirred for 1.5 h and the concentrated. The resulting residue was purified by flash chromatography ( $20 \%-30 \%$ EtOAc in hexanes) to give the allylic alcohol A31 ( $143.1 \mathrm{mg}, 94.8 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.49-7.18(\mathrm{~m}, 10 \mathrm{H}), 5.44-5.36(\mathrm{~m}, 3 \mathrm{H}), 4.15(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.04(\mathrm{~d}, J=8.3$ Hz, 1H), 3.96-3.91 (m, 1H), 3.19 (s, 3H), 2.14-2.02 (m, 4H), 2.00-1.90 (m, 2H), 1.75 (br s, 1H), $1.68(\mathrm{~s}, 3 \mathrm{H}), 1.60-1.39(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 143.0,142.5,139.1,130.4,130.0$,
$129.0,128.6,128.3,126.4,126.3,123.8,83.7,59.3,57.9,56.2,39.6,32.6,31.6,30.9,25.0,16.4 ;$ IR (neat) 3390, 3026, 2930, 1599, 1495, 1451, 1101, 1003, 969, 745, $703 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 401.2457$, found 401.2464.

## ((2R,3R)-3-(2-((2R,3R)-3-(4-Methoxy-5,5-diphenylpentyl)oxiran-2-yl)ethyl)-3-methyloxiran-

## 2-yl)methanol (A32)



To a solution of dienol A31 (100.0 mg, 0.264 mmol$)$ in $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{DMM}(7.9 \mathrm{~mL}, 1: 2, \mathrm{v} / \mathrm{v})$ were added a 0.05 M solution of $\mathrm{Na}_{2} \mathrm{~B}_{4} \mathrm{O}_{7}$ in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}($ EDTA $)(5.3 \mathrm{~mL}), \mathrm{Bu}_{4} \mathrm{NHSO}_{4}(7.2 \mathrm{mg}, 21.1 \mu \mathrm{~mol})$ and Shi ketone (102 mg, 0.396 mmol ) sequentially. The mixture was cooled to $-5^{\circ} \mathrm{C}$, and the Oxone ( $672 \mathrm{mg}, 1.09 \mathrm{mmol}$ ), dissolved in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}($ EDTA $)(3.4 \mathrm{~mL})$, and $\mathrm{K}_{2} \mathrm{CO}_{3}(635 \mathrm{mg}, 4.59$ $\mathrm{mmol})$, dissolved in water ( 3.4 mL ), were added simultaneously via a syringe pump over 2.0 h . After the addition was completed, the slightly blue reaction mixture was stirred further for 15 $\min$ at $0{ }^{\circ} \mathrm{C}$, then diluted with water $(10 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \times 20 \mathrm{~mL})$. The organic extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was purified by flash chromatography (50\% - 70\% EtOAc in hexanes) to give the diepoxy alcohol A32 (101.9 $\mathrm{mg}, 94.0 \%, \mathrm{dr} \sim 1: 1$ regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.39-7.15(\mathrm{~m}, 10 \mathrm{H}), 4.01$ $(\mathrm{d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.94-3.89(\mathrm{~m}, 1 \mathrm{H}), 3.79(\mathrm{dd}, J=12.1,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.69(\mathrm{dd}, J=12.0,6.3 \mathrm{~Hz}$, $1 \mathrm{H}), 3.17 / 3.16(\mathrm{~s}, 3 \mathrm{H}), 2.98-2.95(\mathrm{~m}, 1 \mathrm{H}), 2.66-2.60(\mathrm{~m}, 2 \mathrm{H}), 1.86-1.46(\mathrm{~m}, 10 \mathrm{H}), 1.3(\mathrm{~s}, 3 \mathrm{H}) ;$ ${ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.8,142.4,128.9,128.7,128.6,128.4,126.5,126.4,83.6,62.6$, $61.4,60.7,58.9,58.8,58.3,58.2,58.1,58.0,56.2,34.5,32.1,32.0,27.6,21.6,17.1$; IR (neat)

3426, 2934, 1495, 1452, 1097, 1032, 732, $704 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{O}_{4}\left(\mathrm{M}^{+\bullet}\right)$ 410.2457, found 410.2447; $[\alpha]_{\mathrm{D}}=+14.3\left(\mathrm{CHCl}_{3}, c 1.20\right)$.
tert-Butyl ((2R,3R)-3-(2-((2R,3R)-3-(4-methoxy-5,5-diphenylpentyl)oxiran-2-yl)ethyl)-3-me-thyloxiran-2-yl)methyl carbonate (1.55)


A solution of diepoxy alcohol A32 (99.2 mg, $0.242 \mathrm{mmol})$ in toluene $(2.4 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was added 1-Methylimidazole ( $19 \mu \mathrm{~L}, 0.24 \mathrm{mmol}$ ) and $\mathrm{Boc}_{2} \mathrm{O}(106 \mathrm{mg}, 0.484 \mathrm{mmol})$ sequentially. The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. After that time, the reaction was quenched with water $(10 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 25 \mathrm{~mL})$. The extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated. The residue was azeotroped with hexanes ( $2 \times 20 \mathrm{~mL}$ ) to remove $t-\mathrm{BuOH}$ and the residue was purified by column chromatography $\left(16 \%-20 \% \mathrm{EtOAc}\right.$ in hexanes containing $\left.0.5 \% \mathrm{Et}_{3} \mathrm{~N}\right)$ to give the desired product $1.55(102.0 \mathrm{mg}, 82.6 \%, \mathrm{dr} \sim 1: 1$ regarding the stereochemical outcomes of the homobenzylic center and the two epoxide groups) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.39-7.17(\mathrm{~m}, 10 \mathrm{H}), 4.22(\mathrm{dd}, J=11.8,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 4.14(\mathrm{dd}, J=11.9,6.0 \mathrm{~Hz}, 1 \mathrm{H})$, $4.00(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.93-3.88(\mathrm{~m}, 1 \mathrm{H}), 3.17 / 3.16(\mathrm{~s}, 3 \mathrm{H}), 3.02(\mathrm{t}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.64-2.60$ $(\mathrm{m}, 2 \mathrm{H}), 1.75-1.44(\mathrm{~m}, 10 \mathrm{H}), 1.50(\mathrm{~s}, 9 \mathrm{H}), 1.32(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 153.5$, $142.9,142.4,129.0,128.7,128.6,128.4,126.6,126.4,83.7,82.8,65.6,60.2,59.3,58.8,58.8$, 58.2, 58.1, 56.3, 34.2, 32.2, 27.9, 27.6, 21.7, 17.1; IR (neat) 2978, 2934, 1743, 1495, 1453, 1370, 1279, 1256, 1163, 1097, 859, 746, $704 \mathrm{~cm}^{-1}$; HRMS (ESI): m/z calcd for $\mathrm{C}_{31} \mathrm{H}_{42} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 533.2879, found 533.2866; $[\alpha]_{\mathrm{D}}=+19.1\left(\mathrm{CHCl}_{3}, c 1.08\right)$.

## (S)-4-((R)-Tetrahydro-5-methoxyfuran-2-yl)-1,3-dioxolan-2-one (1.68)



To tert-butyl carbonate $\mathbf{1 . 4 7}(92.0 \mathrm{mg}, \quad 0.223 \mathrm{mmol})$ in dichloroethane/toluene ( $8.6 \mathrm{~mL}, 5: 1 \mathrm{v} / \mathrm{v}$ ) in borosilicate flask at room temperature were added the activated $4 \AA$ molecular sieves ( 184 mg ), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ ( 184 mg ), $\mathrm{NaOAc}(184 \mathrm{mg})$ and $N$-methylquinolinium hexafluorophosphate ( $6.4 \mathrm{mg}, 22.3 \mu \mathrm{~mol}$ ). The mixture was photoirradiated with gentle air bubbling for 3 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with $\mathrm{Et}_{2} \mathrm{O}(40 \mathrm{~mL})$. The filtrate was concentrated and the resulting residue was purified by flash chromatography ( $45 \%-55 \%$ EtOAc in hexanes) to give the product $1.68(24.8 \mathrm{mg}, 59.0 \%)$ in a 1.9:1 ratio as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.04(\mathrm{dd}, J=4.5,1.8 \mathrm{~Hz}, 66 \%$ of $1 \mathrm{H}), 5.01-4.99(\mathrm{~m}, 34 \%$ of 1 H$), 4.67-4.46(\mathrm{~m}, 2.4 \mathrm{H}), 4.39-4.20(\mathrm{~m}, 1.6 \mathrm{H}), 3.33 / 3.32(\mathrm{~s}, 3 \mathrm{H})$, 2.26-1.93 (m, 3H), 1.74-1.66 (m, 1H); ${ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 155.1$ (minor), 154.9 (major), 105.7 (minor), 105.6 (major), 79.4 (minor), 77.8 (minor), 77.1 (major), 66.8 (major), 66.4 (minor), 55.1 (major), 32.7 (minor), 31.7 (major), 25.9 (minor), 25.5 (major); IR (neat) 2920, 1807, 1464, 1376, 1170, 1088, 1031, $955 \mathrm{~cm}^{-1}$; HRMS (EI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{7} \mathrm{H}_{9} \mathrm{O}_{4}(\mathrm{M}-$ $\left.\mathrm{CH}_{3} \mathrm{O}\right)^{+\bullet} 157.0501$, found 157.0499 .

## (S)-4-((R)-Tetrahydro-5-oxofuran-2-yl)-1,3-dioxolan-2-one (1.69)



At $0{ }^{\circ} \mathrm{C}$, the acetal $1.68(6.4 \mathrm{mg}, 34.0 \mu \mathrm{~mol})$ in acetone $(1.0 \mathrm{~mL})$ was treated with Jones reagent $(2.67 \mathrm{M}, 60 \mu \mathrm{~L}, 0.160 \mathrm{mmol})$. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h , and Jones reagent $(2.67 \mathrm{M}, 60 \mu \mathrm{~L}, 0.160 \mathrm{mmol})$ was added. The mixture was stirred at $0^{\circ} \mathrm{C}$ fro 30 min , then at room temperature for 2 h . After that time, the mixture was purified by column chromatography $\left(10 \%-20 \% \mathrm{EtOAc}^{2} \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give the lactone $\mathbf{1 . 6 9}$ (4.3
$\mathrm{mg}, 74.1 \%$ ) as a pale yellow solid: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.74(\mathrm{ddd}, J=8.0,6.7,5.8 \mathrm{~Hz}$, $1 \mathrm{H}), 4.64(\mathrm{t}, J=8.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.65-4.58(\mathrm{~m}, 1 \mathrm{H}), 4.40(\mathrm{dd}, J=8.9,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.68-2.62(\mathrm{~m}$, 2H), 2.60-2.50(m, 1H), 2.20-2.10(m, 1H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 175.2,154.0,78.1$, 76.2, 66.7, 27.5, 24.0; IR (neat) 2919, 1778, 1462, 1401, 1328, 1173, 1087, $1048 \mathrm{~cm}^{-1} ;$ HRMS (EI): $m / z$ calcd for $\mathrm{C}_{7} \mathrm{H}_{8} \mathrm{O}_{5}[\mathrm{M}+\mathrm{H}]^{+}$173.0450, found 173.0455.

To monoepoxide $1.48(102.0 \mathrm{mg}, 0.239 \mathrm{mmol})$ in dichloroethane/toluene ( $9.2 \mathrm{~mL}, 5: 1, \mathrm{v} / \mathrm{v}$ ) in borosilicate flask at room temperature were added the activated $4 \AA$ molecular sieves ( 204 mg ), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(204 \mathrm{mg}), \mathrm{NaOAc}(204 \mathrm{mg})$ and N -methylquinolinium hexafluorophosphate ( $6.9 \mathrm{mg}, 23.9 \mu \mathrm{~mol}$ ). The mixture was photoirradiated with gentle air bubbling for 3 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{~mL})$. The filtrate was concentrated and the resulting yellowish-green residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2.0 \mathrm{~mL})$. To this solution were added $\mathrm{Et}_{3} \mathrm{~N}$ ( $0.22 \mathrm{~mL}, 1.6 \mathrm{mmol}$ ), $\mathrm{Ac}_{2} \mathrm{O}(57 \mu \mathrm{~L}, 0.6 \mathrm{mmol})$ and DMAP $(2.4 \mathrm{mg}, 20 \mu \mathrm{~mol})$ sequentially. The mixture was stirred at room temperature for 3 h , then concentrated and purified by column chromatography ( $20 \%-50 \%$ EtOAc in hexanes) to provide the cyclization products, which were further purified by column chromatography $\left(4 \%-10 \% \mathrm{EtOAc}\right.$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give exo-product $1.71(14.6 \mathrm{mg}, 30 \%, \mathrm{dr}=2: 1)$ and endo-product $1.72(10.6 \mathrm{mg}, 22 \%, \mathrm{dr}=3.4: 1)$ as colorless oils.

## (S)-4-((R)-Tetrahydro-6-methoxy-2H-pyran-2-yl)-1,3-dioxolan-2-one (1.71)

${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.74$ (app d, $J=2.5 \mathrm{~Hz}, 67 \%$ of 1 H ),
 4.62-4.45 (m, 3H), 4.37 (dd, $J=9.3,2.2 \mathrm{~Hz}, 33 \%$ of 1 H ), 3.96 (ddd, $J$ $=11.8,4.4,1.9 \mathrm{~Hz}, 67 \%$ of 1 H$), 3.66(\mathrm{ddd}, J=11.2,5.6,2.2 \mathrm{~Hz}, 33 \%$ of 1 H$), 3.46(\mathrm{~s}, 33 \%$ of
$3 \mathrm{H}), 3.36\left(\mathrm{~s}, 67 \%\right.$ of 3 H ), 1.98-1.16 (m, 6H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 155.1$ (major), 155.0 (minor), 103.4 (minor), 98.4 (major), 78.0 (major), 77.4 (minor), 75.3 (minor), 68.2 (major), 66.4 (minor), 66.1 (major), 56.4 (minor), 55.0 (major), 30.9 (minor), 29.5 (major), 26.6 (minor), 26.5 (major), 21.2 (minor), 17.2 (major); IR (neat) 2952, 2851, 1799, 1389, 1174, 1078, $1031 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{8} \mathrm{H}_{11} \mathrm{O}_{4}\left(\mathrm{M}^{+\bullet}\right)$ 171.0657, found 171.0650 .

## (4aR,9aS)-Hexahydro-6-methoxy-4H-[1,3]dioxino[5,4-b]oxepin-2-one (1.72)

 $4.29(\mathrm{~m}, 77 \%$ of 1 H$), 4.22-4.06(\mathrm{~m}, 2.8 \mathrm{H}), 3.79(\mathrm{dt}, J=9.7,5.8 \mathrm{~Hz}, 23 \%$ of 1 H ), $3.42(\mathrm{~s}, 23 \%$ of 3 H$), 3.36(\mathrm{~s}, 77 \%$ of 3 H$), 2.37-2.14(\mathrm{~m}, 3 \mathrm{H}), 1.96-1.93(\mathrm{~m}, 23 \%$ of 1 H$)$, 1.77-1.42 (m, 5H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 148.5$ (minor), 148.1 (major), 104.9 (minor), 102.8 (major), 81.2 (minor), 80.8 (major), 69.3 (major), 69.2 (minor), 68.2 (minor), 61.5 (major), 56.4 (minor), 55.8 (major), 35.5 (major), 34.2 (major), 33.5 (minor), 27.9 (minor), 17.7 (major), 16.6 (minor); IR (neat) $2943,1760,1403,1382,1224,1140,1057 \mathrm{~cm}^{-1} ;$ HRMS (EI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{8} \mathrm{H}_{11} \mathrm{O}_{4}\left(\mathrm{M}^{+\bullet}\right)$ 171.0657, found 171.0651 ; an analytical sample of the major diastereomer was obtained through purifying the above mixture by column chromatography ( $35 \%-45 \%$ EtOAc in hexanes): ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 4.02$ (dd, $J=8.9,5.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.60 (dd, $J=$ $10.1,5.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.42(\mathrm{t}, J=10.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.34-3.27(\mathrm{~m}, 2 \mathrm{H}), 2.85(\mathrm{~s}, 3 \mathrm{H}), 1.76-1.67(\mathrm{~m}, 1 \mathrm{H})$, $1.57-1.47(\mathrm{~m}, 1 \mathrm{H}), 1.10($ dddd, $J=15.3,11.6,8.9,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 0.95-0.85(\mathrm{~m}, 2 \mathrm{H}), 0.82-0.71$ (m, 1H).

## (R)-Tetrahydro-6-((S)-2-oxo-1,3-dioxolan-4-yl)pyran-2-one (1.73)



A solution of acetal $1.71(6.0 \mathrm{mg}, 29.7 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.6 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ was treated with $m \mathrm{CPBA}$ (pure, $6.7 \mathrm{mg}, 38.6 \mu \mathrm{~mol}$ ) and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(4.5$ $\mu \mathrm{L}, 35.6 \mu \mathrm{~mol})$ sequentially. After stirred at $0^{\circ} \mathrm{C}$ for 10 min and then at room temperature for 1.5 h , the mixture was cooled to $0{ }^{\circ} \mathrm{C}$ and $\mathrm{Et}_{3} \mathrm{~N}(20.7 \mu \mathrm{~L}, 148 \mu \mathrm{~mol})$ was added dropwise. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h , then concentrated, and the resulting residue was purified by column chromatography $\left(15 \%-25 \% \mathrm{EtOAc}\right.$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give the desired lactone $\mathbf{1 . 7 3}(4.6 \mathrm{mg}$, $83.6 \%$ ) as a colorless liquid: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.69-4.59(\mathrm{~m}, 2 \mathrm{H}), 4.56-4.41(\mathrm{~m}$, 2H), 2.69 (dddd, $J=18.0,6.8,4.8,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.54$ (ddd, $J=17.9,9.3,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.24-2.16$ $(\mathrm{m}, 1 \mathrm{H}), 2.08-1.90(\mathrm{~m}, 2 \mathrm{H}), 1.64(\mathrm{dtd}, J=13.8,11.0,5.2 \mathrm{~Hz}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $169.3,154.2,79.0,76.3,66.9,29.8,24.6,18.2$; IR (neat) $2919,1790,1732,1376,1239,1166$, $1056 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{8} \mathrm{H}_{10} \mathrm{O}_{5}\left(\mathrm{M}^{+\bullet}\right) 186.0528$, found 186.0536.

## 4-(Tetrahydro-5-methoxy-2-methylfuran-2-yl)-1,3-dioxolan-2-one (1.76), Hexahydro-6-methoxy-8a-methylpyrano[3,2-d][1,3]dioxin-2-one (1.77) and (4aR,6S,8aR)-Hexahydro-6-methoxy-8a-methylpyrano[3,2-d][1,3]dioxin-2-one (1.78)




carbonate $1.49(125.2 \mathrm{mg}, 0.294 \mathrm{mmol})$ in dichloroethane/toluene ( $11.3 \mathrm{~mL}, 5: 1, \mathrm{v} / \mathrm{v}$ ) in borosilicate flask at room temperature were added the activated $4 \AA$ molecular sieves ( 250 mg ), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(250 \mathrm{mg}), \mathrm{NaOAc}(250 \mathrm{mg})$ and N -methylquinolinium hexafluorophosphate $(8.5 \mathrm{mg}, 29.4 \mu \mathrm{~mol})$. The mixture was photoirradiated with gentle air bubbling for 2.5 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel
and the residue was washed with $\operatorname{EtOAc}(30 \mathrm{~mL})$. The filtrate was concentrated and the resulting residue was purified by flash chromatography ( $30 \%-45 \%$ EtOAc in hexanes) to provide a mixture of $\mathbf{1 . 7 6}$ and $\mathbf{1 . 7 7}(19.7 \mathrm{mg}, 33.2 \%$, a pale yellow oil) with a molar ratio of 4.8:1 and cisfused endo-product $1.78(3.8 \mathrm{mg}, 8.0 \%)$ as a white solid. For the mixture of $\mathbf{1 . 7 6}$ and 1.77: IR (neat) $2928,2835,1791,1755,1463,1375,1170,1084,1034,951 \mathrm{~cm}^{-1}$. For cis-fused endoproduct 1.78: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.78(\operatorname{app~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.66(\mathrm{dd}, J=12.1,2.7$ $\mathrm{Hz}, 1 \mathrm{H}), 4.34(\mathrm{dd}, J=12.1,0.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.86(\operatorname{app} \mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.41(\mathrm{~s}, 3 \mathrm{H}), 2.12-2.00(\mathrm{~m}$, $1 \mathrm{H}), 1.94(\mathrm{dd}, J=12.8,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.87-1.81(\mathrm{~m}, 1 \mathrm{H}), 1.67-1.61(\mathrm{~m}, 1 \mathrm{H}), 1.45(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 148.5,98.3,78.6,69.3,63.2,55.4,29.4,25.3,25.1$; IR (neat) 2932, 1748, 1212, 1178, 1130, 1060, $1024 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{9} \mathrm{H}_{15} \mathrm{O}_{5}[\mathrm{M}+\mathrm{H}]^{+}$203.0919, found 203.0929.

## (S)-4-((R)-Tetrahydro-2-methyl-5-oxofuran-2-yl)-1,3-dioxolan-2-one (1.79)



To a solution of the mixture of $\mathbf{1 . 7 6}$ and $1.77(18.9 \mathrm{mg}, 93.5 \mu \mathrm{~mol})$ in acetone $(3.0 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was added Jones reagent $(0.3 \mathrm{~mL})$. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 15 min and then at room temperature for 3 h . After that time, the reaction was quenched with isopropyl alcohol (1 drop), concentrated and purified by column chromatography $\left(2 \%-20 \% \mathrm{EtOAc}\right.$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give the unreacted acetal 1.77 ( $2.9 \mathrm{mg}, 15.3 \%$ ) as a white solid and the title lactone 1.79 ( $11.0 \mathrm{mg}, \sim 74.8 \%$ based on unreacted acetal): ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.72(\mathrm{dd}, J=8.4,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.57(\mathrm{t}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.36(\mathrm{dd}, J=9.2$, $6.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.70(\mathrm{t} . J=8.8 \mathrm{~Hz}, 2 \mathrm{H}), 2.34-2.24(\mathrm{~m}, 1 \mathrm{H}), 2.19-2.09(\mathrm{~m}, 1 \mathrm{H}), 1.47(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 175.0,154.1,83.9,78.7,65.4,30.0,28.2,20.8$; IR (neat) 2920, 1789,

1463, 1267, 1167, $1082 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{8} \mathrm{H}_{11} \mathrm{O}_{5}[\mathrm{M}+\mathrm{H}]^{+}$187.0606, found 187.0612.
(4aR,6S,8aS)-Hexahydro-6-methoxy-8a-methylpyrano[3,2-d][1,3]dioxin-2-one (1.77)

$3.38(\mathrm{~s}, 3 \mathrm{H}), 2.13-2.04(\mathrm{~m}, 1 \mathrm{H}), 1.95-1.76(\mathrm{~m}, 3 \mathrm{H}), 1.50(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 148.1, $98.3,77.5,67.0,62.7,55.3,31.1,27.8,17.5$; IR (neat) 2917, 1757, 1464, 1196, 1111, $1068 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{9} \mathrm{H}_{15} \mathrm{O}_{5}[\mathrm{M}+\mathrm{H}]^{+}$203.0919, found 203.0930.
(4aR,9aS)-Hexahydro-6-methoxy-9a-methyl-4H-[1,3]dioxino[5,4-b]oxepin-2-one (1.81)


To tert-butyl carbonate $\mathbf{1 . 5 0} \quad(65.8 \quad \mathrm{mg}, \quad 149 \quad \mu \mathrm{~mol})$ in dichloroethane/toluene ( $5.7 \mathrm{~mL}, 5: 1 \mathrm{~h} / \mathrm{v}$ ) in borosilicate flask at room temperature were added the activated $4 \AA$ molecular sieves ( 132 mg ), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(132 \mathrm{mg}), \mathrm{NaOAc}(132 \mathrm{mg})$ and N -methylquinolinium hexafluorophosphate ( $4.3 \mathrm{mg}, 14.9 \mu \mathrm{~mol}$ ). The mixture was photoirradiated with gentle air bubbling for 2 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with $\mathrm{EtOAc}(20 \mathrm{~mL})$. The filtrate was concentrated and the resulting residue was purified by flash chromatography ( $5 \%-15 \% \mathrm{EtOAc}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to provide the desired compound $1.81(23.7 \mathrm{mg}, 73.4 \%)$ as a mixture of two diastereomers in a 1.2:1 ratio: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.76$ (t, $J=3.8 \mathrm{~Hz}, 46 \%$ of 1 H ), 4.66 (dd, $J=8.8,5.8 \mathrm{~Hz}, 54 \%$ of $1 \mathrm{H}), 4.34(\mathrm{dd}, J=10.8,6.4 \mathrm{~Hz}, 46 \%$ of 1 H$), 4.29-4.18(\mathrm{~m}, 54 \%$ of 2 H$), 4.19(\mathrm{t}, J=10.8 \mathrm{~Hz}$, $46 \%$ of 1 H$), 3.88(\mathrm{dd}, J=10.6,6.4 \mathrm{~Hz}, 46 \%$ of 1 H$), 3.42(\mathrm{~s}, 46 \%$ of 3 H$), 3.35(\mathrm{~s}, 54 \%$ of 3 H$)$,
$2.23-2.01(\mathrm{~m}, \sim 1.5 \mathrm{H}), 1.96-1.92(\mathrm{~m}, 46 \%$ of 1 H$), 1.75-1.58(\mathrm{~m}, \sim 3.5 \mathrm{H}), 1.51(\mathrm{~s}, 46 \%$ of 3 H$)$, $1.48(\mathrm{~s}, 54 \%$ of 3 H$), 1.45-1.35(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 148.2,148.0,104.3$, $102.8,84.2,83.0,68.1,66.7,66.3,62.7,56.3,55.7,43.2,41.6,34.5,34.4,19.5,19.3,18.3,16.7 ;$ IR (neat) 2941, 1755, 1464, 1384, 1252, 1199, 1128, 1091, 1050, $969 \mathrm{~cm}^{-1} ;$ HRMS (EI): m/z calcd for $\mathrm{C}_{9} \mathrm{H}_{13} \mathrm{O}_{4} \mathrm{Na}\left(\mathrm{M}-\mathrm{CH}_{3} \mathrm{O}\right)^{+\bullet}$ 185.0814, found 185.0811. An analytical sample of the slightly major diastereomer was obtained through purifying the above mixture by column chromatography ( $35 \%-40 \%$ EtOAc in hexanes): ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 4.00(\mathrm{dd}, J=8.8$, $5.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.60(\mathrm{~d}, J=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.58(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.44(\mathrm{dd}, J=10.4,6.8 \mathrm{~Hz}, 1 \mathrm{H})$, $2.85(\mathrm{~s}, 3 \mathrm{H}), 1.56-1.48(\mathrm{~m}, 2 \mathrm{H}), 1.19-1.10(\mathrm{~m}, 2 \mathrm{H}), 0.98-0.88(\mathrm{~m}, 1 \mathrm{H}), 0.94(\mathrm{~s}, 3 \mathrm{H}), 0.75-0.66(\mathrm{~m}$, $1 \mathrm{H})$.

## (S)-4-((2R,5S)-Tetrahydro-5-((R)-tetrahydro-5-methoxy-2-methylfuran-2-yl)-2-

methylfuran-2-yl)-1,3-dioxolan-2-one (1.85) and (4aR,5aS,9aR,11aS)-8-Methoxy-5a,11a-dimethyldecahydro-1,3,5,9-tetraoxadibenzo[a,d]cyclohepten-2-one (1.86)

$\mathrm{mL}, 5: 1, \mathrm{v} / \mathrm{v}$ ) in borosilicate flask at room temperature were added the activated $4 \AA$ molecular sieves ( 258 mg ), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(258 \mathrm{mg}), \mathrm{NaOAc}(258 \mathrm{mg})$ and N -methylquinolinium hexafluorophosphate $(7.3 \mathrm{mg}, 25.3 \mu \mathrm{~mol})$. The mixture was photoirradiated with gentle air bubbling for 4.5 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc ( 40 mL ). The filtrate was concentrated and the resulting residue was purified by flash chromatography ( $35 \%-50 \% \mathrm{EtOAc}$
in hexanes) to provide a mixture of the above two products ( $28.7 \mathrm{mg}, 39.6 \%$ ) as a colorless oil: IR (neat) 2926, 1796, 1754, 1460, 1374, 1166, 1085, 1036, 1006, $952 \mathrm{~cm}^{-1}$.
(S)-4-((2R,5S)-Tetrahydro-5-((R)-tetrahydro-2-methyl-5-oxofuran-2-yl)-2-methylfuran-2-yl)-1,3-dioxolan-2-one (A33)


A mixture of acetals 1.85 and $1.86(20.8 \mathrm{mg}, 72.6 \mu \mathrm{~mol})$ in acetone $(2.1 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated dropwise with Jones reagent $(0.2 \mathrm{~mL})$. The mixture was stirred at $0^{\circ} \mathrm{C}$ for 10 min , then at room temperature for 1.5 h and purified without workup by column chromatography (50\%-90\% EtOAc in hexanes) to give the unreacted acetal 1.86 ( $3.2 \mathrm{mg}, 15.4 \%$, nearly pure) and lactone A33 (13.8 mg, ~80\%). For lactone A33: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.60(\mathrm{dd}, J=8.4,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.50(\mathrm{t}, J=8.6 \mathrm{~Hz}$, $1 \mathrm{H}), 4.38(\mathrm{dd}, J=8.7,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{dd}, J=8.6,6.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.64-2.58(\mathrm{~m}, 2 \mathrm{H}), 2.31-2.19$ $(\mathrm{m}, 1 \mathrm{H}), 2.10-2.04(\mathrm{~m}, 2 \mathrm{H}), 1.94-1.73(\mathrm{~m}, 3 \mathrm{H}), 1.39(\mathrm{~s}, 3 \mathrm{H}), 1.27(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 176.7,155.1,86.9,83.5,83.2,79.5,66.0,34.1,29.3,29.2,26.7,23.8,21.0$; IR (neat) 2958, 2924, 2853, 1790, 1770, 1456, 1382, 1248, 1166, 1085, 1020, 944, 770, $728 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{6}\left(\mathrm{M}^{+\bullet}\right) 270.1103$, found 270.1094; $[\alpha]_{\mathrm{D}}=+4.5\left(\mathrm{CHCl}_{3}, c 0.24\right)$.

## (4aR,5aS,9aR,11aS)-5a,11a-Dimethyloctahydro-1,3,5,9-tetraoxadibenzo[a,d]cycloheptene-

 2,8-dione (A34)

A solution of acetal $1.86(2.9 \mathrm{mg}, 11.2 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ at 0
${ }^{\circ} \mathrm{C}$ was treated with $m \mathrm{CPBA}$ (pure, $2.5 \mathrm{mg}, 14.6 \mu \mathrm{~mol}$ ) and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ $(1.9 \mu \mathrm{~L}, 13.4 \mu \mathrm{~mol})$ sequentially. After stirred at $0{ }^{\circ} \mathrm{C}$ for 10 min and then at room temperature for 30 min , the mixture was cooled to $0{ }^{\circ} \mathrm{C}$ and $\mathrm{Et}_{3} \mathrm{~N}(7.8 \mu \mathrm{~L}, 56.0$
$\mu \mathrm{mol}$ ) was added dropwise. The mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min , and purified by column chromatography ( $10 \%-20 \% \mathrm{EtOAc}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the desired lactone $\mathbf{A 3 4}$ (1.8 mg, 66.7\%) as colorless needles: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.28(\mathrm{dd}, J=8.6,5.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.21-4.14(\mathrm{~m}$, $1 \mathrm{H}), 4.09$ (dd, $J=10.1,8.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.06(\mathrm{dd}, J=11.0,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.80(\mathrm{ddd}, J=18.3,9.4,5.5$ $\mathrm{Hz}, 1 \mathrm{H}), 2.64(\mathrm{ddd}, J=18.3,8.7,7.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.35-2.28(\mathrm{~m}, 1 \mathrm{H}), 2.17-1.88(\mathrm{~m}, 5 \mathrm{H}), 1.50(\mathrm{~s}$, 3H), 1.38 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 170.4,148.2,83.1,82.7,77.4,75.7,66.5,65.2$, $37.2,34.3,27.5,24.8,22.4,16.0$; IR (neat) 2923, 1747, 1463, 1408, 1229, 1124, $1068 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (300 MHz, C ${ }_{6} \mathrm{D}_{6}$ ) $\delta 3.44(\mathrm{~d}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.43(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.18(\mathrm{dd}, J=9.6,7.7$ $\mathrm{Hz}, 1 \mathrm{H}), 2.93$ (dd, $J=10.9,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.02-1.97(\mathrm{~m}, 2 \mathrm{H}), 1.65(\mathrm{td}, J=15.3,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.43-$ $1.26(\mathrm{~m}, 3 \mathrm{H}), 1.17-1.04(\mathrm{~m}, 2 \mathrm{H}), 0.81(\mathrm{~s}, 3 \mathrm{H}), 0.48(\mathrm{~s}, 3 \mathrm{H})$; HRMS (ESI): m/z calcd for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$293.1001, found 293.1020; $[\alpha]_{\mathrm{D}}=+101\left(\mathrm{CHCl}_{3}, c 0.15\right)$.

## (R)-4-((2S,5S)-tetrahydro-5-((R)-tetrahydro-5-methoxy-2-methylfuran-2-yl)-2-

methylfuran-2-yl)-1,3-dioxolan-2-one (1.87) and (4aS,5aS,9aR,11aR)-8-Methoxy-5a,11a-dimethyldecahydro-1,3,5,9-tetraoxadibenzo[a,d]cyclohepten-2-one (1.88)
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To diepoxide 1.52 ( 150 mg , 0.294 mmol in dichloroethane/toluene (11.3
$\mathrm{mL}, 5: 1, \mathrm{v} / \mathrm{v}$ ) in borosilicate flask at room temperature were added the activated $4 \AA$ molecular sieves (300 mg), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(300 \mathrm{mg}), \mathrm{NaOAc}(300 \mathrm{mg})$ and $N$-methylquinolinium hexafluorophosphate $(8.5 \mathrm{mg}, 29.4 \mu \mathrm{~mol})$. The mixture was photoirradiated with gentle air bubbling for 4.5 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc $(40 \mathrm{~mL})$. The filtrate was
concentrated and the resulting residue was purified by flash chromatography ( $35 \%-50 \% \mathrm{EtOAc}$ in hexanes) to provide a mixture of the above two products ( $51.2 \mathrm{mg}, 60.9 \%$ ): IR (neat) 2925 , $1797,1750,1462,1384,1259,1167,1120 \mathrm{~cm}^{-1}$.

## (R)-4-((2S,5S)-tetrahydro-5-((R)-tetrahydro-2-methyl-5-oxofuran-2-yl)-2-methylfuran-2-

## yl)-1,3-dioxolan-2-one (A35)



A mixture of acetals 1.87 and $1.88(34.9 \mathrm{mg}, 122 \mu \mathrm{~mol})$ in acetone $(1.8 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated dropwise with Jones reagent $(0.3 \mathrm{~mL})$. The mixture was stirred at $0^{\circ} \mathrm{C}$ for 10 min , then at room temperature for 1.5 h and purified without workup by column chromatography (50\%-90\% EtOAc in hexanes) to give the unreacted acetal 1.88 (4.9 mg, nearly pure) and lactone A35 (22.1 mg, ~81\%). For lactone A35: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.58(\mathrm{dd}, J=8.3,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.48(\mathrm{t}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.32(\mathrm{dd}, J$ $=8.8,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{dd}, J=8.8,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.65-2.59(\mathrm{~m}, 2 \mathrm{H}), 2.24(\mathrm{ddd}, J=12.9,9.6,6.9$ $\mathrm{Hz}, 1 \mathrm{H}), 2.07-1.81(\mathrm{~m}, 5 \mathrm{H}), 1.38(\mathrm{~s}, 3 \mathrm{H}), 1.27(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 176.7, $155.0,86.9,85.1,83.1,80.3,66.2,34.5,29.4,29.1,27.0,23.4,21.3$; IR (neat) 2979, 2880, 1790, 1767, 1454, 1382, 1170, 1111, 1085, $944 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{6}\left(\mathrm{M}^{+\cdot}\right)$ 270.1103, found 270.1095; $[\alpha]_{\mathrm{D}}=-11.3\left(\mathrm{CHCl}_{3}, c 1.03\right)$.

## (4aS,5aS,9aR,11aR)-5a,11a-Dimethyloctahydro-1,3,5,9-tetraoxadibenzo[a,d]cycloheptene-

## 2,8-dione (A36)



A solution of acetal $1.88(4.5 \mathrm{mg}, 15.7 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ at 0 ${ }^{\circ} \mathrm{C}$ was treated with $m \mathrm{CPBA}$ (pure, $3.5 \mathrm{mg}, 20.4 \mu \mathrm{~mol}$ ) and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ $(2.7 \mu \mathrm{~L}, 18.8 \mu \mathrm{~mol})$ sequentially. After stirred at $0^{\circ} \mathrm{C}$ for 10 min and
at room temperature for 20 min , the mixture was cooled to $0^{\circ} \mathrm{C}$ and $\mathrm{Et}_{3} \mathrm{~N}(10.9 \mu \mathrm{~L}, 78.5 \mu \mathrm{~mol})$ was added dropwise. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min and purified by column chromatography ( $15 \%-25 \% \mathrm{EtOAc}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the desired lactone $\mathbf{A 3 6}$ (3.0 mg, 71.4\%) as colorless needles: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.40(\mathrm{app} \mathrm{dd}, J=10.0,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.23(\mathrm{dd}$, $J=10.5,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.09(\mathrm{t}, J=10.2, \mathrm{~Hz}, 1 \mathrm{H}), 4.00(\mathrm{dd}, J=10.1,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.88(\mathrm{ddd}, J=$ $18.3,11.2,4.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.72$ (ddd, $J=18.3,9.6,5.5 \mathrm{~Hz}, 1 \mathrm{H}) ; 2.21-1.76(\mathrm{~m}, 6 \mathrm{H}), 1.49(\mathrm{~s}, 3 \mathrm{H})$, $1.24(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 170.7,147.4,83.8,81.4,77.4,66.6,64.9,39.1,30.8$, 28.1, 24.4, 20.5, 19.5; IR (neat) 2924, 1748, 1463, 1408, 1229, 1124, $1068 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$293.1001, found 293.0988; $[\alpha]_{\mathrm{D}}=+48.7\left(\mathrm{CHCl}_{3}, c 0.23\right)$.

## (4aR,5aS,10aR,12aS)-9-Methoxy-5a,12a-dimethyldecahydro-1,3,5,10-tetraoxabenzo[b]-

 heptalen-2-one (1.91)

To diepoxide 1.53 ( $145 \mathrm{mg}, 276 \mu \mathrm{~mol}$ ) in dichloroethane/toluene $(10.6 \mathrm{~mL}, 5: 1, \mathrm{v} / \mathrm{v})$ in borosilicate flask at room temperature were added the activated $4 \AA$ molecular sieves ( 290 mg ), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(290 \mathrm{mg}), \mathrm{NaOAc}(290 \mathrm{mg})$ and N -methylquinolinium hexafluorophosphate $(8.0 \mathrm{mg}$, $27.6 \mu \mathrm{~mol}$ ). The mixture was photoirradiated with gentle air bubbling for 2 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc $(40 \mathrm{~mL})$. The filtrate was concentrated and the resulting residue was purified by flash chromatography ( $25 \%-35 \%$ EtOAc in hexanes) to provide the product $1.91\left(44.8 \mathrm{mg}, 54.0 \%\right.$, pale yellow liquid) as two diastereomers in about $1: 1$ ratio: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.54-4.48(\mathrm{~m}, 1 \mathrm{H}), 4.25-3.98(\mathrm{~m}, 3 \mathrm{H}), 3.92-3.85(\mathrm{~m}, 0.5 \mathrm{H}), 3.56(\mathrm{dd}, J=11.2$, $2.4 \mathrm{~Hz}, 0.5 \mathrm{H}), 3.40 / 3.37(\mathrm{~s}, 3 \mathrm{H}), 3.24(\mathrm{dd}, J=10.8,3.8 \mathrm{~Hz}, 0.5 \mathrm{H}), 2.26-1.94$ (m, 2.5H), 1.91-
$1.72(\mathrm{~m}, 3.5 \mathrm{H}), 1.65-1.52(\mathrm{~m}, 3.5 \mathrm{H}), 1.47 / 1.44(\mathrm{~s}, 3 \mathrm{H}), 1.33 / 1.29(\mathrm{~s}, 3 \mathrm{H}), 1.21-1.18(\mathrm{~m}, 0.5 \mathrm{H}) ;$
${ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 149.0,148.6,105.2,102.6,83.3,83.2,81.4,80.3,79.7,75.6,67.0$, $67.0,65.2,63.8,56.1,55.9,44.6,43.3,37.5,37.0,35.3,33.6,27.6,26.2,22.3,21.6,19.2,17.7$, 17.0, 16.7; IR (neat) 2940, 1759, 1454, 1384, 1209, 1111, 1053, 1008, $921 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 323.1471$, found 323.1500; $[\alpha]_{\mathrm{D}}=+31.5\left(\mathrm{CHCl}_{3}, c 1.45\right)$.
(4aR,5aS,10aR,12aS)-5a,12a-Dimethyldecahydro-1,3,5,10-tetraoxabenzo[b]heptalene-2,9dione (1.92)


A solution of acetal $1.91(15.6 \mathrm{mg}, 51.9 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ at
$0{ }^{\circ} \mathrm{C}$ was treated with $m \mathrm{CPBA}$ (pure, $11.6 \mathrm{mg}, 67.5 \mu \mathrm{~mol}$ ) and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(7.2 \mu \mathrm{~L}, 57.1 \mu \mathrm{~mol})$ sequentially. After stirred at $0{ }^{\circ} \mathrm{C}$ for 10 min , then at room temperature for 1 h , the mixture was cooled to $0{ }^{\circ} \mathrm{C}$ and $\mathrm{Et}_{3} \mathrm{~N}(36.2 \mu \mathrm{~L}, 256$ $\mu \mathrm{mol})$ was added dropwise. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1.5 h , the quenched with a mixture of saturated $\mathrm{NaHCO}_{3} /$ saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(4 \mathrm{~mL}, 1: 1, \mathrm{v} / \mathrm{v})$. The mixture was poured onto water ( 5 mL ) and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 25 \mathrm{~mL})$. The extracts were dried over $\mathrm{MgSO}_{4}$, filtered and concentrated, and the resulting residue was purified by column chromatography ( $10 \%-20 \%$ EtOAc in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the desired lactone 1.92 ( $9.9 \mathrm{mg}, 66.9 \%$ ) as a white crystalline solid: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.26-4.20(\mathrm{~m}, 2 \mathrm{H}), 4.14-4.06(\mathrm{~m}, 2 \mathrm{H}), 2.70-2.55(\mathrm{~m}, 2 \mathrm{H}), 2.35-$ $2.23(\mathrm{~m}, 2 \mathrm{H}), 1.99-1.81(\mathrm{~m}, 4 \mathrm{H}), 1.77-1.68(\mathrm{~m}, 2 \mathrm{H}), 1.46(\mathrm{~s}, 3 \mathrm{H}), 1.31(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 173.3,148.6,84.3,82.2,78.9,66.8,64.3,43.2,36.2,33.6,26.6,22.0,20.0,15.8 ;$ IR (neat) 2989, 2941, 2871, 1748, 1727, 1501, 1454, 1365, 1328, 1272, 1212, 1098, $1040 \mathrm{~cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$307.1158, found 307.1158. $[\alpha]_{\mathrm{D}}=+50.4$ $\left(\mathrm{CHCl}_{3}, c 0.42\right)$.

## (4aS,5aS,10aR,12aR)-9-Methoxy-5a,12a-dimethyldecahydro-1,3,5,10-tetraoxabenzo[b]-

 heptalen-2-one (1.93)

To diepoxide 1.54 ( $48.2 \mathrm{mg}, 91.9 \mu \mathrm{~mol}$ ) in dichloroethane/toluene $(3.5 \mathrm{~mL}, 5: 1, \mathrm{v} / \mathrm{v})$ in borosilicate flask at room temperature were added the activated $4 \AA$ molecular sieves ( 96 mg ), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(96 \mathrm{mg}), \mathrm{NaOAc}(96 \mathrm{mg})$ and $N$-methylquinolinium hexafluorophosphate ( $2.6 \mathrm{mg}, 9.2$ $\mu \mathrm{mol})$. The mixture was photoirradiated with gentle air bubbling for 3 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc ( 30 mL ). The filtrate was concentrated and the resulting residue was purified by flash chromatography ( $25 \%-35 \%$ EtOAc in hexanes) to provide the product $\mathbf{1 . 9 3}$ (21.7 $\mathrm{mg}, 78.6 \%$, pale yellow solid) as two diastereomers in about 1:1 ratio: ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 4.69(\mathrm{dd}, J=3.8,2.2 \mathrm{~Hz}, 0.5 \mathrm{H}), 4.54(\mathrm{dd}, J=8.9,5.7 \mathrm{~Hz}, 0.5 \mathrm{H}), 4.17(\mathrm{dd}, J=10.7,6.6$ $\mathrm{Hz}, 1 \mathrm{H}), 4.02(\mathrm{t}, J=10.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.90(\mathrm{dd}, J=10.7,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.90-3.85(\mathrm{~m}, 0.5 \mathrm{H}), 3.52(\mathrm{dd}$, $J=10.1,0.8 \mathrm{~Hz}, 0.5 \mathrm{H}), 3.40 / 3.37(\mathrm{~s}, 3 \mathrm{H}), 2.08-2.00(\mathrm{~m}, 2 \mathrm{H}), 1.89-1.53(\mathrm{~m}, 8 \mathrm{H}), 1.44(\mathrm{~s}, 3 \mathrm{H})$, $1.21 / 1.17(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) $\delta$ 151.6, 148.0, 102.7, 102.4, 83.6, 83.6, 81.3, $80.6,78.8,74.1,67.0(2 \mathrm{C}), 64.0,63.9,56.0,55.8,40.5,40.2,39.8,39.5,33.7,33.4,27.3$ (2C), 20.8, 20.3, 19.4, 19.3, 19.3, 17.5; IR (neat) 2940, 1755, 1461, 1382, 1246, 1223, 1116, 1051, 913 $\mathrm{cm}^{-1}$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{6} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$323.1471, found 323.1462; $[\alpha]_{\mathrm{D}}=+26.6$ $\left(\mathrm{CHCl}_{3}, c 0.55\right)$.

## (4aS,5aS,10aR,12aR)-5a,12a-Dimethyldecahydro-1,3,5,10-tetraoxabenzo[b]heptalene-2,9-

## dione (1.94)



A solution of acetal $1.93(19.0 \mathrm{mg}, 63.2 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2.0 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with $m \mathrm{CPBA}$ (pure, $14.2 \mathrm{mg}, 82.2 \mu \mathrm{~mol}$ ) and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(9.5 \mu \mathrm{~L}, 75.8 \mu \mathrm{~mol})$ sequentially. The mixture was stirred at $0^{\circ} \mathrm{C}$ for 10 min , and then at room temperature for 1 h . After that time, the mixture was cooled to $0{ }^{\circ} \mathrm{C}$ and $\mathrm{Et}_{3} \mathrm{~N}(44.0 \mu \mathrm{~L}, 316 \mu \mathrm{~mol})$ was added dropwise. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min, then concentrated, and the resulting residue was purified by column chromatography ( $15 \%$ $25 \% \mathrm{EtOAc}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the desired lactone 1.94 (14.4 mg, $80.0 \%$ ) as a white solid: ${ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.47(\mathrm{dd}, J=10.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.20(\mathrm{dd}, J=10.7,6.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.05(\mathrm{t}, J$ $=10.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.98(\mathrm{dd}, J=10.5,6.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.70(\mathrm{dt}, J=14.1,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.64(\mathrm{ddd}, J=$ $14.1,5.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.12(\mathrm{ddd}, J=13.6,5.9,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.07-1.98(\mathrm{~m}, 3 \mathrm{H}), 1.90(\mathrm{dddd}, J=$ $14.7,5.8,2.6,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.84(\mathrm{app} \mathrm{dt}, J=13.6,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.78-1.70(\mathrm{~m}, 2 \mathrm{H}), 1.48(\mathrm{~s}, 3 \mathrm{H})$, 1.14 (s, 3H); ${ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 173.4,147.8,83.6,82.7,79.0,66.8,64.6,39.3$, 38.3, 33.4, 26.4, 20.4, 19.14, 19.10; IR (neat) 2984, 2941, 1747, 1732, 1444, 1388, 1274, 1252, 1200, 1116, 1100, 1070, $1049 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{O}_{6}\left(\mathrm{M}^{+\bullet}\right) 284.1260$, found 284.1254; $[\alpha]_{\mathrm{D}}=+17.2\left(\mathrm{CHCl}_{3}, c 0.52\right)$.

Key NOESY enhancements observed in lactone 1.94:

(S)-4-((2R,5S)-tetrahydro-5-((R)-tetrahydro-6-methoxy-2H-pyran-2-yl)-2-methylfuran-2-yl)-1,3-dioxolan-2-one (1.97) and (4aR,5aS,10aR,12aS)-9-Methoxy-12a-methyldecahydro-1,3,5,10-tetraoxabenzo[b]heptalen-2-one (1.98)


To diepoxide 1.55 (52.8 mg, $103 \mu \mathrm{~mol}$ in dichloroethane/toluene (4.0 $\mathrm{mL}, 5: 1, \mathrm{v} / \mathrm{v}$ ) in borosilicate flask at room temperature were added the activated $4 \AA$ molecular sieves (106 mg), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(106 \mathrm{mg}$, $), \mathrm{NaOAc}(106 \mathrm{mg})$ and N -methylquinolinium hexafluorophosphate $(3.0 \mathrm{mg}, 10.3 \mu \mathrm{~mol})$. The mixture was photoirradiated with gentle air bubbling for 4 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc ( 20 mL ). The filtrate was concentrated and the resulting residue was purified by flash chromatography (5\%-20\% EtOAc in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to provide the exo, exo-product $1.97(7.3 \mathrm{mg}, 24.7 \%)$ as a white solid and endo, endoproduct 1.98 ( $8.8 \mathrm{mg}, 29.7 \%$ ) as a colorless oil. For exo, exo-product $1.97(\mathrm{dr}=2: 1):{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.71(\mathrm{br} \mathrm{s}, 67 \%$ of 1 H$), 4.61-4.56(\mathrm{~m}, 1 \mathrm{H}), 4.54-4.50(\mathrm{~m}, 1 \mathrm{H}), 4.44-4.41(\mathrm{~m}$, $1 \mathrm{H}), 4.32(\mathrm{dd}, J=9.5,2.0 \mathrm{~Hz}, 33 \%$ of 1 H$), 4.02(\mathrm{dd}, J=7.1,4.9 \mathrm{~Hz}, 33 \%$ of 1 H$), 3.98(\mathrm{dd}, J=$ $7.4,4.4 \mathrm{~Hz}, 67 \%$ of 1 H ), 3.72 (ddd, $J=11.6,4.2,2.0 \mathrm{~Hz}, 67 \%$ of 1 H ), $3.48(\mathrm{~s}, 33 \%$ of 3 H$), 3.40$ (ddd, $J=11.3,4.7,1.9 \mathrm{~Hz}, 33 \%$ of 1 H$), 3.33(\mathrm{~s}, 67 \%$ of 3 H$), 2.05-1.95(\mathrm{~m}, 4 \mathrm{H}), 1.90-1.78(\mathrm{~m}$, $3 \mathrm{H}), 1.73-1.65(\mathrm{~m}, 1 \mathrm{H}), 1.60(\mathrm{~s}, 3 \mathrm{H}), 1.55-1.48(\mathrm{~m}, 1 \mathrm{H}), 1.31-1.22(\mathrm{~m}, 1 \mathrm{H}), 1.28(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 155.3$ (major), 151.1 (minor), 103.7 (minor), 98.7 (major), 82.2 (major, 2C), 81.9 (minor), 79.6 (minor), 79.2 (major), 69.7 (major), 66.0 (major), 56.3 (minor), 54.7 (major), 35.1 (major), 34.7 (minor), 31.3 (minor), 29.9 (major), 27.6 (minor), 27.3 (major), 26.7
(minor), 26.3 (major), 22.0 (minor), 20.9 (minor), 20.5 (major), 17.8 (major); IR (neat) 2943, 1798, 1455, 1374, 1166, 1033, $949 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{O}_{6}\left(\mathrm{M}^{+\cdot}\right)$ 286.1416, found 286.1419; $[\alpha]_{\mathrm{D}}=-24.1\left(\mathrm{CHCl}_{3}, c\right.$ 0.71). For endo, endo-product $1.98(\mathrm{dr}=2.3: 1):{ }^{1} \mathrm{H}$ NMR ( $600 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 4.56(\mathrm{dd}, J=8.8,5.8 \mathrm{~Hz}, 70 \%$ of 1 H ), 4.49-4.46 (m, $30 \%$ of 1 H ), 4.39-4.34 (m, 1H), $4.11(\mathrm{t}, J=10.6 \mathrm{~Hz}, 70 \%$ of 1 H$), 4.10(\mathrm{t}, J=10.6 \mathrm{~Hz}, 30 \%$ of 1 H$), 3.94(\mathrm{dd}$, $J=11.3,6.5 \mathrm{~Hz}, 70 \%$ of 1 H$), 3.84(\mathrm{dd}, J=11.0,6.3 \mathrm{~Hz}, 30 \%$ of 1 H$), 3.68(\mathrm{dt}, J=8.5,4.5 \mathrm{~Hz}$, $70 \%$ of 1 H ), $3.62-3.59(\mathrm{~m}, 30 \%$ of 1 H$), 3.49-3.47(\mathrm{~m}, 30 \%$ of 1 H$), 3.42(\mathrm{~s}, 30 \%$ of 3 H$), 3.38(\mathrm{~s}$, $70 \%$ of 3 H$), 3.37-3.33(70 \%$ of 1 H$), 2.22-1.97(\mathrm{~m}, 4 \mathrm{H}), 1.92-1.78(\mathrm{~m}, 2 \mathrm{H}), 1.65-1.59(\mathrm{~m}, 2 \mathrm{H})$, $1.46(\mathrm{~s}, 30 \%$ of 3 H$), 1.43(\mathrm{~s}, 70 \%$ of 3 H$), 1.38-1.33(\mathrm{~m}, 1 \mathrm{H}), 1.28-1.25(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (151 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 149.0$ (minor), 148.9 (major), 106.9 (minor), 102.6 (major), 86.4 (major), 83.3 (minor), 82.6 (minor), 82.4 (major), 79.5 (minor), 75.2 (minor), 74.0 (major), 73.2 (minor), 73.1 (major), 66.5 (major), 56.1 (minor), 55.9 (major), 39.5 (minor), 36.7 (major), 35.9 (minor), 35.6 (major), 34.9 (minor), 33.5 (major), 29.7 (major), 28.6 (minor), 28.0 (minor), 21.0 (major), 18.9 (major), 17.9 (minor); IR (neat) 2939, 1755, 1455, 1384, 1255, 1205, 1109, 1042, $999 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{22} \mathrm{O}_{6}\left(\mathrm{M}^{+\bullet}\right)$ 286.1416, found 286.1414; $[\alpha]_{\mathrm{D}}=+11.8\left(\mathrm{CHCl}_{3}, c\right.$ 0.85).

## (4aR,5aS,10aR,12aS)-12a-Methyldecahydro-1,3,5,10-tetraoxa-benzo[b]heptalene-2,9-dione

 (1.99)

To a solution of acetal $1.98(8.0 \mathrm{mg}, 27.9 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ were added $m \mathrm{CPBA}$ (pure, $6.2 \mathrm{mg}, 36.3 \mu \mathrm{~mol}$ ) and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ (4.2 $\mu \mathrm{L}, 33.5 \mu \mathrm{~mol})$ sequentially. After stirred at room temperature for 30 min , the mixture was cooled to $0{ }^{\circ} \mathrm{C}$ and $\mathrm{Et}_{3} \mathrm{~N}(19.4 \mu \mathrm{~L}, 140 \mu \mathrm{~mol})$ was added dropwise. The
mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min , then concentrated, and the resulting residue was purified by column chromatography ( $10 \%-20 \% \mathrm{EtOAc}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give lactone $1.99(5.3 \mathrm{mg}, 70.2 \%)$ as a white crystalline solid: ${ }^{1} \mathrm{H}$ NMR $\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.43-4.39(\mathrm{~m}, 1 \mathrm{H}), 4.40(\mathrm{dd}, J=10.4$, $6.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.13(\mathrm{dd}, J=11.2,10.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.86(\mathrm{dd}, J=11.3,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.53(\mathrm{ddd}, J=10.6$, 8.0, $3.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.70-2.61(\mathrm{~m}, 2 \mathrm{H}), 2.22-2.17(\mathrm{~m}, 3 \mathrm{H}), 2.07-2.01(\mathrm{~m}, 2 \mathrm{H}), 1.92(\mathrm{ddd}, J=15.4$, 9.6, $2.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.77-1.73(\mathrm{~m}, 2 \mathrm{H}), 1.48(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 173.9,148.2$, $85.5,81.9,81.2,78.4,66.5,35.7,34.5,33.6,27.5,21.0,19.2 ;{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ ) $\delta 3.58$ $(\mathrm{dd}, J=10.0,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.46(\mathrm{dd}, J=11.2,10.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.38-3.34(\mathrm{~m}, 1 \mathrm{H}), 2.66(\mathrm{dd}, J=$ $11.2,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.58(\mathrm{ddd}, J=11.2,7.8,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.22-2.18(\mathrm{~m}, 1 \mathrm{H}), 1.73-1.65(\mathrm{~m}, 2 \mathrm{H})$, 1.52 (dddd, $J=15.8,8.8,3.8,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.42-1.38(\mathrm{~m}, 1 \mathrm{H}), 1.30(\mathrm{ddd}, J=14.4,8.8,1.4 \mathrm{~Hz}$, $1 \mathrm{H}), 1.22(\mathrm{dddd}, J=17.1,11.6,5.2,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.16-1.11(\mathrm{~m}, 2 \mathrm{H}), 0.92-0.84(\mathrm{~m}, 1 \mathrm{H}), 0.80(\mathrm{~s}$, 3H); IR (neat) 2922, 2850, 1747, 1453, 1387, 1273, 1204, 1106, 1058, $1015 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{6}\left(\mathrm{M}^{+\bullet}\right) 270.1103$, found 270.1111; $[\alpha]_{\mathrm{D}}=+12.7\left(\mathrm{CHCl}_{3}, c 0.26\right)$.
(R)-Tetrahydro-6-((2S,5R)-tetrahydro-5-methyl-5-((S)-2-oxo-1,3-dioxolan-4-yl)furan-2-yl)-pyran-2-one (A37)

To a solution of acetal $1.97(6.8 \mathrm{mg}, 23.7 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5$
 mL ) at $0^{\circ} \mathrm{C}$ were added $m$ CPBA acid (pure, $5.3 \mathrm{mg}, 30.8 \mu \mathrm{~mol}$ ) and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(4.0 \mu \mathrm{~L}, 28.4 \mu \mathrm{~mol})$ sequentially. After stirred at room temperature for 30 min , the mixture was cooled to $0{ }^{\circ} \mathrm{C}$ and $\mathrm{Et}_{3} \mathrm{~N}(16.5 \mu \mathrm{~L}, 118 \mu \mathrm{~mol})$ was added dropwise. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 30 min , then concentrated, and the resulting residue was purified by column chromatography ( $15 \%-25 \% \mathrm{EtOAc}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give the desired lactone $\mathbf{A 3 7}$ ( $5.2 \mathrm{mg}, 81.2 \%$ ) as a white solid: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.64(\mathrm{dd}, J=8.4,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.52(\mathrm{t}, J=8.8$
$\mathrm{Hz}, 1 \mathrm{H}), 4.45(\mathrm{dd}, J=8.8,6.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.30(\mathrm{dd}, J=11.4,4.6,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.10(\mathrm{dt}, J=7.2,4.6$ $\mathrm{Hz}, 1 \mathrm{H}), 2.62$ (dddd, $J=17.8,6.6,4.8,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.46(\mathrm{ddd}, J=17.8,9.3,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.19-$ $2.12(\mathrm{~m}, 1 \mathrm{H}), 2.07-2.02(\mathrm{~m}, 1 \mathrm{H}), 2.01-1.93(\mathrm{~m}, 3 \mathrm{H}), 1.90-1.85(\mathrm{~m}, 1 \mathrm{H}), 1.84-1.79(\mathrm{~m}, 1 \mathrm{H}), 1.50-$ $1.44(\mathrm{~m}, 1 \mathrm{H}), 1.28(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (125 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 171.1,155.2,83.0,81.1,80.7,79.0$, 66.1, 34.5, 29.9, 26.4, 24.8, 20.7, 18.5; IR (neat) 2957, 2929, 1789, 1731, 1242, 1173, 1084, 1049, 1018, $771 \mathrm{~cm}^{-1}$; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{6}\left(\mathrm{M}^{+\bullet}\right) 270.1103$, found 270.1104; $[\alpha]_{\mathrm{D}}$ $=-42.8\left(\mathrm{CHCl}_{3}, c 0.50\right)$.

















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| T0 | 65536 |
| SOLVENT | CDC13 |
| NS | 16 |
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| cx | 20.00 cm |
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| PC | 1.00 |
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| cX | 20.00 cm |
| F1P | 10.000 ppm |
| F1 | 6008.30 Hz |
| F2P | -0.500 ppm |
| F2 | $-300.42 \mathrm{~Hz}$ |
| PPMCM | $0.52500 \mathrm{ppm} / \mathrm{cm}$ |
| HZCM | $315.43576 \mathrm{~Hz} / \mathrm{cm}$ |



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## APPENDIX B

## EFFORTS TOWARDS THE TOTAL SYNTHESIS OF (+)LACTODEHYDROTHYRSIFEROL AND ITS ANALOGS (SUPPORTING INFORMATION)

General Experimental Proton ( ${ }^{1} \mathrm{H}$ NMR) and carbon $\left({ }^{13} \mathrm{C}\right.$ NMR) nuclear magnetic resonance spectra were recorded at ambient temperature on Bruker Avance 300 spectrometer at 300 MHz and 75 MHz or Bruker Avance 500 spectrometer at 500 MHz and 125 MHz if specified. The chemical shifts are given in parts per million (ppm) on the delta ( $\delta$ ) scale. The solvent peak was used as a reference value, for ${ }^{1} \mathrm{H}$ NMR: $\mathrm{CDCl}_{3}=7.27 \mathrm{ppm}$, for ${ }^{13} \mathrm{C}$ NMR: $\mathrm{CDCl}_{3}=77.23$. Data are reported as follows: $(\mathrm{s}=$ singlet; $\mathrm{d}=$ doublet; $\mathrm{t}=$ triplet; $\mathrm{q}=$ quartet; $\mathrm{dd}=$ doublet of doublets; ddd $=$ doublet of doublet of doublets; $\mathrm{dt}=$ doublet of triplets; $\mathrm{td}=$ triplet of doublets; $\mathrm{br}=$ broad $).$ High resolution mass spectra were recorded on a MICROMASS AUTOSPEC (for EI) or WATERS Q-TOF API-US (for ESI) spectrometer. Infrared (IR) spectra were collected on a Mattson Cygnus 100 spectrometer. Samples for IR were prepared as a thin film on a NaCl plate by dissolving the compound in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and then evaporating the $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. Tetrahydrofuran was distilled from sodium and benzophenone. Activated $4 \AA$ molecular sieves were obtained through drying in oven at $150{ }^{\circ} \mathrm{C}$ overnight. $\mathrm{Boc}_{2} \mathrm{O}$ and N -methylimidole were purchased from Acros and
used without further purification. Anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ was purchased from Aldrich and used as received. Toluene and 1,2-dichloroethane were purchased from Fisher Scientific and dried with 4 $\AA$ molecular sieves overnight prior to use. Anhydrous DMF and MeI were purchased from Acros. $\mathrm{NiCl}_{2}, \mathrm{CrCl}_{3}, \mathrm{Mn}$, and anhydrous LiCl were purchased from Aldrich and used without further purification. TMSCl (purchased from Aldrich) was distilled from anhydrous $\mathrm{K}_{2} \mathrm{CO}_{3}$ before use. Analytical TLC was performed on E. Merck pre-coated ( 25 mm ) silica gel 60F-254 plates. Visualization was done under UV (254 nm). Flash chromatography was done using ICN SiliTech 32-63 $60 \AA$ silica gel. Reagent grade ethyl acetate, diethyl ether and hexanes (commercial mixture) were purchased from EM Science and used as is for chromatography. All reactions were performed in flame-dried glassware under nitrogen with magnetic stirring unless otherwise noted.

## 1-Hydroxy-5-methoxy-6,6-diphenylhexan-2-one (2.11)



To a solution of terminal olefin $2.10(407 \mathrm{mg}, 1.53 \mathrm{mmol})$ in acetone $/ \mathrm{H}_{2} \mathrm{O}(15 \mathrm{~mL}, 4.5: 1, \mathrm{v} / \mathrm{v})$ and $\mathrm{AcOH}(0.58 \mathrm{~mL})$ was added dropwise a solution of $\mathrm{KMnO}_{4}(435 \mathrm{mg}, 2.75 \mathrm{mmol})$ in acetone $/ \mathrm{H}_{2} \mathrm{O}$ $(6.9 \mathrm{~mL})$. The mixture was stirred for 5 min and $\mathrm{EtOH}(0.8 \mathrm{~mL})$ was added. After 20 min , the mixture was filtered through Celite and the residue was washed with acetone ( 50 mL ). The filtrate was concentrated and the residue was azeotroped with acetone ( 3 x ) and purified by column chromatography ( $35 \%-40 \%$ EtOAc in hexanes) to give hydroxyl ketone 2.11 ( 258 mg , $57 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.38-7.18(\mathrm{~m}, 10 \mathrm{H}), 4.29-4.14(\mathrm{~m}, 2 \mathrm{H})$, 4.02-3.95 (m, 2H), 3.10 (s, 3H), 3.08-3.06 (m, 1H), 2.58-2.38 (m, 2H), 2.07-1.95 (m, 1H), 1.73$1.64(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 209.7,142.2,128.9,128.8,128.6,126.9,126.7,82.6$,
68.2, 58.1, 56.4, 33.9, 26.3; IR (neat) 3435, 2931, 1718, 1494, 1451, 1102, 748; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{O}_{3} \mathrm{~K}[\mathrm{M}+\mathrm{K}]^{+} 337.1206$, found 337.1176.

## 5-Methoxy-2-methylene-6,6-diphenylhexan-1-ol (2.12)



To a solution of hydroxyl ketone $2.11(235 \mathrm{mg}, 0.788 \mathrm{mmol})$ in THF $(7 \mathrm{~mL})$ were added $\mathrm{PPh}_{3}(227 \mathrm{mg}, 0.867 \mathrm{mmol}),\left(\mathrm{Ph}_{3} \mathrm{P}\right)_{3} \mathrm{RhCl}(18.2$ $\mathrm{mg}, 19.7 \mu \mathrm{~mol})$ and ${ }^{i} \operatorname{PrOH}(0.6 \mathrm{~mL}, 7.88 \mathrm{mmol})$ sequentially. After $5 \mathrm{~min}, \mathrm{TMSCHN}_{2}(0.63 \mathrm{~mL}$, 1.26 mmol ) was added. The yellow mixture was stirred for $18 \mathrm{~h}, \mathrm{PPh}_{3}(200 \mathrm{mg}, 0.788 \mathrm{mmol})$ was added followed by $\mathrm{TMSCHN}_{2}(0.63 \mathrm{~mL}, 1.26 \mathrm{mmol})$. The reaction was stirred for another 18 h and TBAF ( 1 M in THF, $3.0 \mathrm{~mL}, 3.0 \mathrm{mmol}$ ) was added. The mixture was stirred for 30 min , then treated with saturated $\mathrm{NH}_{4} \mathrm{Cl}(15 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(4 \times 20 \mathrm{~mL})$. The extract was dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and concentrated. The residue was purified column chromatography ( $20 \%-45 \%$ EtOAc in hexanes) to give allylic alcohol 2.12 ( $168 \mathrm{mg}, 72 \%$ ) and unreacted starting material 2.11 ( 23 mg , 10\%). For allylic alcohol 2.12: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.40-7.18$ $(\mathrm{m}, 10 \mathrm{H}), 5.00(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.82(\mathrm{~d}, J=0.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.04(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.00(\mathrm{br} \mathrm{s}, 2 \mathrm{H})$, 3.97-3.90(m, 1H), 3.16(s, 3H), 2.27-2.09 (m, 2H), 1.78-1.52 (m, 3H); ${ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 149.0,142.8,142.4,129.0,128.8,128.7,128.5,126.7,126.5,109.8,83.4,66.1,58.1$, 56.4, 30.6, 28.7; IR (neat) 3397, 2928, 1599, 1494, 1451, 1103, 1030, 898, 745; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{O}_{2} \mathrm{~K}[\mathrm{M}+\mathrm{K}]^{+} 335.1413$, found 335.1404.

## (2-(3-Methoxy-4,4-diphenylbutyl)oxiran-2-yl)methanol (2.13)



To a solution of allylic alcohol $2.12(157 \mathrm{mg}, 0.530 \mathrm{mmol})$ in benzene $(3.0 \mathrm{~mL})$ was added $\mathrm{VO}(\mathrm{acac})_{2}(2.8 \mathrm{mg}, 10.6 \mu \mathrm{~mol})$ followed by
dropwise addition of ${ }^{t} \mathrm{BuOOH}(5-6 \mathrm{M}$ in decane, $0.12 \mathrm{~mL}, \sim 0.65 \mathrm{mmol}$ ). The mixture was heated to $80^{\circ} \mathrm{C}$ for 20 min , then cooled to room temperature and saturated $\mathrm{Na}_{2} \mathrm{SO}_{3}(2 \mathrm{~mL})$ was added. The mixture was stirred for 10 min , the treated with saturated $\mathrm{NaHCO}_{3}(10 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(4 \times 20 \mathrm{~mL})$. The extract was dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and concentrated. The residue was purified column chromatography $\left(30 \%-60 \% \mathrm{EtOAc}\right.$ in hexanes containing $\left.0.5 \% \mathrm{Et}_{3} \mathrm{~N}\right)$ to give epoxy alcohol 2.13 ( $148 \mathrm{mg}, 90 \%$, $\mathrm{dr} \sim 1: 1$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.39-7.19(\mathrm{~m}, 10 \mathrm{H}), 3.98(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.96-3.90(\mathrm{~m}, 1 \mathrm{H}), 3.65(\mathrm{ddd}, J=12.3,6.6,4.4$ $\mathrm{Hz}, 1 \mathrm{H}), 3.53(\mathrm{dd}, J=12.2,8.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.16 / 3.15(\mathrm{~s}, 3 \mathrm{H}), 2.81(\mathrm{dd}, J=4.6,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.56$ $(\mathrm{dd}, J=4.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.98-1.56(\mathrm{~m}, 4 \mathrm{H}), 1.56-1.38(\mathrm{~m}, 1 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $142.6,142.3,128.9,128.8,128.6,128.5,126.8,126.6,83.4,83.3,63.2,62.9,59.8,59.7,58.2$, 58.1, 56.3, 56.3, 50.2, 49.7, 27.5, 27.2, 27.0, 26.8; IR (neat) 3432, 2929, 1642, 1600, 1494, 1452, 1101, 747; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{20} \mathrm{H}_{24} \mathrm{O}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 335.1623$, found 335.1607.

## tert-Butyl 2-(3-methoxy-4,4-diphenylbutyl)oxiranylmethyl carbonate (2.14)



To a solution of epoxy alcohol $2.13(135 \mathrm{mg}, 0.432 \mathrm{mmol})$ in dry toluene ( 4.3 mL ) at $0{ }^{\circ} \mathrm{C}$ were added 1-methylimidazole ( $34 \mu \mathrm{~L}$, $0.432 \mathrm{mmol})$ and di-tert-butyl dicarbonate ( $188 \mathrm{mg}, 0.864 \mathrm{mmol}$ ).

The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. After that time, water ( 2 drops) was added. After 10 min , the mixture was concentrated and the residue was azeotroped with acetone ( 5 mL ), then hexanes ( $2 \times 5 \mathrm{~mL}$ ) and purified by flash chromatography ( $10 \%-20 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the tert-butyl carbonate $2.14(162 \mathrm{mg}, 91 \%, \mathrm{dr} \sim 1: 1)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta$ 7.41-7.19 (m, 10H), 4.12-3.94(m, 4H), 3.18/3.17(s, 3 H$), 2.72(\mathrm{t}, J=4.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.58$
$(\mathrm{dd}, J=4.5,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.97-1.58(\mathrm{~m}, 4 \mathrm{H}), 1.51(\mathrm{~s}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 153.4$, $153.4,142.6,142.3,142.3,129.0,128.9,128.8,128.8,128.6,128.5,126.7,126.7,126.5,83.2$, $82.6,68.5,68.1,58.0,58.0,57.3,57.2,56.2,56.2,50.8,50.3,27.9,27.4,26.9,26.8,26.6$; IR (neat) 2979, 2933, 1742, 1494, 1453, 1279, 1161, 1102, 743; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{25} \mathrm{H}_{32} \mathrm{O}_{5} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 435.2147$, found 435.2150.

## (5S)-8-Methoxy-1,3,7-trioxaspiro[4.5]decan-2-one (2.17)



To epoxide 2.14 ( $70 \mathrm{mg}, 0.17 \mathrm{mmol}$ ) in dichloroethane/toluene ( 5.6 mL , $5: 1, \mathrm{v} / \mathrm{v}$ ) in borosilicate flask at room temperature were added the activated $4 \AA$ molecular sieves ( 140 mg ), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}(140 \mathrm{mg})$,
$\mathrm{NaOAc}(140 \mathrm{mg})$ and N -methylquinolinium hexafluorophosphate ( $24.6 \mathrm{mg}, 85 \mu \mathrm{~mol}$ ). The mixture was photoirradiated with gentle air bubbling for 8 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with $\mathrm{Et}_{2} \mathrm{O}(30 \mathrm{~mL})$. The filtrate was concentrated and the resulting residue was purified by flash chromatography ( $30 \%-60 \%$ EtOAc in hexanes) to produce the bicycle 2.17 ( $25.2 \mathrm{mg}, 79 \%$, dr 1.1:1) as a pale yellow oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.70(\mathrm{t}, J=2.3 \mathrm{~Hz}, 0.5 \mathrm{H}), 4.60(\mathrm{app} \mathrm{dd}$, $J=3.0,2.3 \mathrm{~Hz}, 0.5 \mathrm{H}), 4.50(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 0.5 \mathrm{H}), 4.20(\mathrm{dd}, J=8.6,1.4 \mathrm{~Hz}, 0.5 \mathrm{H}), 4.10(\mathrm{~d}, J=$ $8.8 \mathrm{~Hz}, 0.5 \mathrm{H}), 4.06(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 0.5 \mathrm{H}), 3.88(\mathrm{dd}, J=11.0,1.0 \mathrm{~Hz}, 0.5 \mathrm{H}), 3.72(\mathrm{~d}, J=12.4 \mathrm{~Hz}$, $0.5 \mathrm{H}), 3.66(\mathrm{dd}, J=12.4,2.2 \mathrm{~Hz}, 0.5 \mathrm{H}), 3.52(\mathrm{dd}, J=11.1,2.4 \mathrm{~Hz}, 0.5 \mathrm{H}), 3.39(\mathrm{~s}, 1.5 \mathrm{H}), 3.38(\mathrm{~s}$, $1.5 \mathrm{H}), 2.33(\mathrm{dt}, J=12.8,4.5 \mathrm{~Hz}, 0.5 \mathrm{H}), 2.15-2.00(\mathrm{~m}, 1 \mathrm{H}), 1.99-1.90(\mathrm{~m}, 1 \mathrm{H}), 1.84(\mathrm{dtd}, J=$ 12.4, 4.1, $2.4 \mathrm{~Hz}, 0.5 \mathrm{H}), 1.76-1.57(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 154.2,154.2,97.6$, $96.9,79.4,78.6,72.8,71.7,64.5,63.3,55.5,55.3,28.2,28.1,27.4,26.1$; IR (neat) 2921, 1804,

1389, 1183, 1125, 1050, 772; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{8} \mathrm{H}_{11} \mathrm{O}_{5}\left(\mathrm{M}^{+\bullet}\right)$ 187.0606, found 187.0606.

## 6-(tert-Butyldiphenylsilanyloxy)-2-(tert-butyldiphenylsilanyloxymethyl)hex-1-en-3-ol (2.25)



To a mixture of (2-bromoallyloxy)(tert-butyl)diphenylsilane 2.23 $(2.764 \mathrm{~g}, 7.36 \mathrm{mmol})$, aldehyde $2.24(1.200 \mathrm{~g}, 3.68 \mathrm{mmol}), \mathrm{CrCl}_{3}(116$ $\mathrm{mg}, 0.736 \mathrm{mmol}), \mathrm{NiCl}_{2}(95 \mathrm{mg}, 0.736 \mathrm{mmol})$ and $\mathrm{Mn}(1.011 \mathrm{~g}, 18.4$ mmol ) in anhydrous DMF ( 6 mL , degassed with argon prior to use) was added TMSCl ( 1.12 mL , $8.83 \mathrm{mmol})$. After stirring overnight, the reaction was quenched with water ( 15 mL ) and transferred to a beaker. $\mathrm{HCl}(1 \mathrm{~N})$ was added until all the manganese metal was completely consumed. The mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(4 \mathrm{x} 40 \mathrm{~mL})$, and the extract was dried (MgSO4) and concentrated. The residue was purified by column chromatography ( $2 \%-10 \%$ EtOAc in hexanes) to give allylic alcohol $2.25(1.948 \mathrm{~g}, 85 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.71-7.64(\mathrm{~m}, 8 \mathrm{H}), 7.46-7.34(\mathrm{~m}, 12 \mathrm{H}), 5.20(\mathrm{q}, J=1.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.12(\mathrm{~s}, 1 \mathrm{H})$, $4.30(\mathrm{~d}, J=13.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.23-4.19(\mathrm{~m}, 2 \mathrm{H}), 3.65(\operatorname{app~t}, J=6.0 \mathrm{~Hz}, 2 \mathrm{H}), 2.63(\mathrm{~d}, J=4.9 \mathrm{~Hz}$, $1 \mathrm{H}), 1.70-1.51(\mathrm{~m}, 4 \mathrm{H}), 1.07(\mathrm{~s}, 9 \mathrm{H}), 1.04(\mathrm{~s}, 9 \mathrm{H})$.

## Ethyl (Z)-8-(tert-butyldiphenylsilanyloxy)-4-(tert-butyldiphenylsilanyloxymethyl)oct-4-eno-

 ate (2.26) A mixture of allylic alcohol 2.25 ( $1.868 \mathrm{~g}, 3.00$ ), triethyl orthoacetate ( $2.2 \mathrm{~mL}, 12.0 \mathrm{mmol}$ ) and propionic acid (11 $\mu \mathrm{L}, 0.15 \mathrm{mmol}$ ) was heated to $100{ }^{\circ} \mathrm{C}$ for 4 h and the unreacted triethyl orthoacetate was distilled out under reduced pressure. The residue was purified
by column chromatography ( $1 \%-3 \%$ EtOAc in hexanes) to give the ethyl ester $2.26(1.724 \mathrm{~g}$, $83 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.69-7.60(\mathrm{~m}, 8 \mathrm{H}), 7.43-7.32(\mathrm{~m}, 12 \mathrm{H})$, $5.19(\mathrm{t}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 4.20(\mathrm{~s}, 2 \mathrm{H}), 4.12(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.53(\mathrm{t}, J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.51-2.41$ $(\mathrm{m}, 4 \mathrm{H}), 1.86(\mathrm{q}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 1.52-1.43(\mathrm{~m}, 2 \mathrm{H}), 1.23(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.04(\mathrm{~s}, 9 \mathrm{H}), 1.00$ ( $\mathrm{s}, 9 \mathrm{H}$ ).

## (Z)-10-(tert-Butyldiphenylsilanyloxy)-6-(tert-butyldiphenylsilanyloxymethyl)-2-methyldeca-

## 1,6-dien-3-ol (2.27)



A solution of ethyl ester $2.26(1.700 \mathrm{~g}, 2.45 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ was treated with DIBAL-H (1 M in hexanes, $2.6 \mathrm{~mL}, 2.6 \mathrm{mmol}$ ) over 15 min . After 30 min , DIBAL-H ( $0.36 \mathrm{~mL}, 0.36 \mathrm{mmol}$ ) was added over 2 min . The mixture was stirred for 15 min and isopropenylmagnesium bromide ( 0.5 M in THF, $9.8 \mathrm{~mL}, 4.9 \mathrm{mmol}$ ) was added dropwise. The reaction was stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h , then warmed to room temperature and quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution $(10 \mathrm{~mL})$ and saturated sodium tartrate solution $(25 \mathrm{~mL})$. The mixture was stirred vigorously for 1 h and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{~mL})$. The organic extract was dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $3 \%-$ $12 \%$ EtOAc in hexanes) to give secondary alcohol 2.27 ( $1.333 \mathrm{~g}, 79 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.70-7.63(\mathrm{~m}, 8 \mathrm{H}), 7.45-7.34(\mathrm{~m}, 12 \mathrm{H}), 5.23(\mathrm{t}, J=7.3 \mathrm{~Hz}, 1 \mathrm{H})$, 4.97-4.96 (m, 1H), 4.87-4.86 (m, 1H), $4.22(\mathrm{~d}, J=2.5 \mathrm{~Hz}, 2 \mathrm{H}), 4.08-4.05(\mathrm{~m}, 1 \mathrm{H}), 3.56(\mathrm{t}, J=$ $6.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.28-2.21(\mathrm{~m}, 2 \mathrm{H}), 1.91(\mathrm{q}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 1.78-1.67(\mathrm{~m}, 5 \mathrm{H}), 1.56-1.49(\mathrm{~m}, 2 \mathrm{H})$, $1.06(\mathrm{~s}, 9 \mathrm{H}), 1.02(\mathrm{~s}, 9 \mathrm{H})$.

Ethyl (4E,8Z)-12-(tert-butyldiphenylsilanyloxy)-8-(tert-butyldiphenylsilanyloxymethyl)-4-methyldodeca-4,8-dienoate (2.28)
 A mixture of allylic alcohol 2.27 ( $1.333 \mathrm{~g}, 1.93$ mmol ), triethyl orthoacetate ( $1.4 \mathrm{~mL}, 7.72 \mathrm{mmol}$ ) and propionic acid $(7 \mu \mathrm{~L}, 96 \mu \mathrm{~mol})$ was heated to
$115{ }^{\circ} \mathrm{C}$ for 2 h and the unreacted triethyl orthoacetate was distilled out under reduced pressure. The residue was purified by column chromatography ( $2 \%-4 \%$ EtOAc in hexanes) to give the ethyl ester $2.28(1.216 \mathrm{~g}, 83 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 7.69-7.60 (m, $8 \mathrm{H}), 7.41-7.32(\mathrm{~m}, 12 \mathrm{H}), 5.16(\mathrm{t}, J=6.8 \mathrm{~Hz}, 2 \mathrm{H}), 4.18(\mathrm{~s}, 2 \mathrm{H}), 4.12(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 3.55(\mathrm{t}$, $J=6.4 \mathrm{~Hz}, 2 \mathrm{H}), 2.42-2.37(\mathrm{~m}, 2 \mathrm{H}), 2.32-2.28(\mathrm{~m}, 2 \mathrm{H}), 2.19-2.06(\mathrm{~m}, 4 \mathrm{H}), 1.89(\mathrm{q}, J=7.4 \mathrm{~Hz}$, 2H), $1.60(\mathrm{br} \mathrm{s}, 3 \mathrm{H}), 1.54-1.47(\mathrm{~m}, 2 \mathrm{H}), 1.25(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}), 1.04(\mathrm{~s}, 9 \mathrm{H}), 1.00(\mathrm{~s}, 9 \mathrm{H})$.
(5E,9Z)-13-(tert-Butyldiphenylsilanyloxy)-9-(tert-butyldiphenylsilanyloxymethyl)-5-methyl-1-phenyltrideca-5,9-dien-2-ol (2.29)
 A solution of ethyl ester 2.28 ( $600 \mathrm{mg}, 0.788$ mmol) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ was treated with DIBAL-H ( 1 M in hexanes, $0.83 \mathrm{~mL}, 0.83$
$\mathrm{mmol})$ over 10 min . After 30 min , DIBAL-H ( $0.12 \mathrm{~mL}, 0.12 \mathrm{mmol}$ ) was added and the mixture was stirred for 30 min .

In a separate round-bottom flask, a mixture of $\mathrm{CuCN}(282 \mathrm{mg}, 3.15 \mathrm{mmol})$ and $\mathrm{LiCl}(294 \mathrm{mg}$, $6.93 \mathrm{mmol})$ in THF $(10 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ was treated with $\mathrm{BnMgCl}(2 \mathrm{M}$ in THF, $1.6 \mathrm{mmol}, 3.2$ $\mathrm{mmol})$ dropwise. After $1 \mathrm{~h}, \mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(0.28 \mathrm{~mL}, 2.26 \mathrm{mmol})$ was added and the mixture was stirred for 5 min . The reaction mixture from the first reaction was cannulated into the second
flask followed by rinse ( $2 \times 0.5 \mathrm{~mL}$ THF). The mixture was stirred at $-78^{\circ} \mathrm{C}$ for 1 h and then at room temperature for 1 h . After that time, the reaction was quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution $(10 \mathrm{~mL})$ /saturated sodium tartrate solution $(5 \mathrm{~mL})$ and stirred vigorously for 1 h . The mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(4 \times 20 \mathrm{~mL})$ and the organic extract was dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was purified by column chromatography (3\% - $12 \%$ EtOAc in hexanes) to give secondary alcohol 2.29 ( $496 \mathrm{mg}, 78 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.70-7.62(\mathrm{~m}, 8 \mathrm{H}), 7.44-7.30(\mathrm{~m}, 12 \mathrm{H}), 7.30-7.20(\mathrm{~m}, 5 \mathrm{H}), 5.21-5.15(\mathrm{~m}, 2 \mathrm{H}), 4.19(\mathrm{~s}$, $2 \mathrm{H}), 3.86-3.76(\mathrm{~m}, 1 \mathrm{H}), 3.55(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 2.84(\mathrm{dd}, J=13.6,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.67(\mathrm{dd}, J=$ $13.5,8.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.23-1.98(\mathrm{~m}, 2 \mathrm{H}), 1.60(\mathrm{~s}, 3 \mathrm{H}), 1.66-1.47(\mathrm{~m}, 6 \mathrm{H}), 1.05(\mathrm{~s}, 9 \mathrm{H}), 1.01(\mathrm{~s}, 9 \mathrm{H})$.
(Z)-6-((E)-7-Methoxy-4-methyl-8-phenyloct-3-enyl)-2,2-13,13-tetramethyl-3,3,12,12-tetraphenyl-4,11-dioxa-3,12-disilatetradec-6-ene (B1)


A solution of alcohol 2.29 ( $486 \mathrm{mg}, 0.600 \mathrm{mmol}$ ) in DMF ( 5 mL ) at $0{ }^{\circ} \mathrm{C}$ was treated with NaH ( $60 \%$ weight in mineral oil, $60 \mathrm{mg}, 1.50 \mathrm{mmol}$ ).

After 20 min , $\mathrm{MeI}(0.15 \mathrm{~mL}, 2.4 \mathrm{mmol})$ was added and the mixture was stirred overnight at room temperature. After that time, the reaction was quenched with water ( 20 mL ) cautiously and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. The organic extract was dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $2 \%-5 \%$ EtOAc in hexanes) to give methyl ether B1 (431 mg, 87\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.69-7.61(\mathrm{~m}, 8 \mathrm{H})$, 7.43-7.30 (m, 12H), 7.29-7.26 (m, 2H), 7.22-7.18 (m, 3H), 5.20-5.10 (m, 2H), 4.19 (s, 2H), 3.55 $(\mathrm{t}, J=6.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.36-3.30(\mathrm{~m}, 1 \mathrm{H}), 3.31(\mathrm{~s}, 3 \mathrm{H}), 2.84(\mathrm{dd}, J=13.7,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.72(\mathrm{dd}, J=$
$13.8,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.22-1.96(\mathrm{~m}, 6 \mathrm{H}), 1.89(\mathrm{q}, J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 1.58-1.48(\mathrm{~m}, 7 \mathrm{H}), 1.04(\mathrm{~s}, 9 \mathrm{H})$, 1.00 ( $\mathrm{s}, 9 \mathrm{H}$ ).

## (2Z)-2-((E)-7-Methoxy-4-methyl-8-phenyloct-3-enyl)hex-2-ene-1,6-diol (2.30)



To a solution of silyl ether B1 ( $409 \mathrm{mg}, 0.497 \mathrm{mmol})$ in THF ( 5 mL ) was added $\mathrm{TBAF} \cdot \mathrm{H}_{2} \mathrm{O}(312 \mathrm{mg}, 1.19$ $\mathrm{mmol})$. The reaction was stirred for 4 h , then concentrated and the residue was purified by column chromatography (50\%-60\% EtOAc in hexanes) to give the diol 2.30 ( $144 \mathrm{mg}, 84 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 7.32-7.19 (m, 5H), 5.28 (t, $J=7.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.16-4.99(\mathrm{~m}, 1 \mathrm{H}), 4.17(\mathrm{~s}, 2 \mathrm{H}), 3.63(\mathrm{t}, J=5.9 \mathrm{~Hz}$, $2 \mathrm{H}), 3.39-3.35(\mathrm{~m}, 1 \mathrm{H}), 3.32(\mathrm{~s}, 3 \mathrm{H}), 2.86(\mathrm{dd}, J=13.7,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.71(\mathrm{dd}, J=13.7,6.3 \mathrm{~Hz}$, $1 \mathrm{H}), 2.24(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.15-1.97(\mathrm{~m}, 8 \mathrm{H}), 1.68-1.60(\mathrm{~m}, 2 \mathrm{H}), 1.58-1.49(\mathrm{~m}, 5 \mathrm{H})$.

## 3-((2S,3R)-3-(Hydroxymethyl)-3-((E)-7-methoxy-4-methyl-8-phenyloct-3-enyl)oxiran-2-

 yl)propan-1-ol (B2)

To a mixture of diol $2.30(138 \mathrm{mg}, 0.398 \mathrm{mmol})$ and activated $4 \AA$ molecular sieves $(120 \mathrm{mg})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4$ $\mathrm{mL})$ at $-20^{\circ} \mathrm{C}$ was added D --diisopropyl tartrate $(8 \mu \mathrm{~L}$, $48 \mu \mathrm{~mol})$. After $10 \mathrm{~min}, \mathrm{Ti}\left(\mathrm{O}^{i} \mathrm{Pr}\right)_{4}(12 \mu \mathrm{~L}, 40 \mu \mathrm{~mol})$ was introduced. The mixture was stirred for 30 min , and ${ }^{t} \mathrm{BuOOH}(5-6 \mathrm{M}$ in decane, $0.22 \mathrm{~mL}, \sim 1.2 \mathrm{mmol}$ ) was added dropwise. The reaction was stirred for 5 h at $-20^{\circ} \mathrm{C}$, and then stored in at $-20^{\circ} \mathrm{C}$ overnight. Water $(0.5 \mathrm{~mL})$ was added the mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h . After that time, $30 \% \mathrm{NaOH}$ saturated with $\mathrm{NaCl}(0.5 \mathrm{~mL})$ was added and the mixture was stirred at room temperature for 2.5 h . The mixture was filtered
through a 1:1 mixture of $\mathrm{MgSO}_{4} /$ Celite and the residue was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$. The combined filtrates were concentrated and the residue was purified by column chromatography $\left(60 \%-80 \% \mathrm{EtOAc}\right.$ in hexanes containing $\left.1 \% \mathrm{Et}_{3} \mathrm{~N}\right)$ to give monoepoxy diol $\mathbf{B} 2(47 \mathrm{mg}, 33 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.32-7.19(\mathrm{~m}, 5 \mathrm{H}), 5.11(\mathrm{t}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.78-$ $3.65(\mathrm{~m}, 4 \mathrm{H}), 3.39-3.34(\mathrm{~m}, 1 \mathrm{H}), 3.32(\mathrm{~s}, 3 \mathrm{H}), 2.90-2.83(\mathrm{~m}, 2 \mathrm{H}), 2.70(\mathrm{ddd}, J=13.8,6.3,2.5 \mathrm{~Hz}$, 1H), 2.38 (br s, 1H), 2.16-1.95 (m, 6H), 1.86-1.49 (m, 10H).

## 3-((2S,3R)-3-(Hydroxymethyl)-3-(2-((2R,3R)-3-(3-methoxy-4-phenylbutyl)-3-methyloxiran-

 2-yl)ethyl)oxiran-2-yl)propan-1-ol (B3)

A solution of monoepoxide $\mathbf{B} 2(45 \mathrm{mg}, 0.124 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{DMM}(1.8 \mathrm{~mL}, 1: 2, \mathrm{v} / \mathrm{v})$ was treated with 0.05 M solution of $\mathrm{Na}_{2} \mathrm{~B}_{4} \mathrm{O}_{7}$ in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}$ (EDTA) (1.2 $\mathrm{mL}), \mathrm{Bu}_{4} \mathrm{NHSO}_{4}(1.7 \mathrm{mg}, 4.5 \mu \mathrm{~mol})$ and Shi ketone $(16.0 \mathrm{mg}, 62 \mu \mathrm{~mol})$ sequentially. The mixture was cooled to $-5^{\circ} \mathrm{C}$. Oxone ( $122 \mathrm{mg}, 0.198 \mathrm{mmol}$ ), dissolved in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}$ (EDTA) $(0.8 \mathrm{~mL})$, and $\mathrm{K}_{2} \mathrm{CO}_{3}(115 \mathrm{mg}, 0.831 \mathrm{mmol})$, dissolved in water $(0.8 \mathrm{~mL})$, were added simultaneously via a syringe pump over 2.0 h . After the addition was completed, the blue mixture was stirred further for 10 min , and anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ was added in portions until all the water disappeared. The mixture was filtered and the residues was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The combined filtrates were concentrated and the residue was purified by flash chromatography $\left(80 \%-100 \% \mathrm{EtOAc}\right.$ in hexanes containing $\left.1 \% \mathrm{Et}_{3} \mathrm{~N}\right)$ to give diepoxide $\mathbf{B 3}(42 \mathrm{mg}, 89 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.30-7.18(\mathrm{~m}, 5 \mathrm{H}), 3.90-3.49(\mathrm{~m}, 4 \mathrm{H}), 3.38-3.34(\mathrm{~m}$, $1 \mathrm{H}), 3.32(\mathrm{~s}, 3 \mathrm{H}), 2.91-2.82(\mathrm{~m}, 2 \mathrm{H}), 2.76-2.64(\mathrm{~m}, 2 \mathrm{H}), 2.20-1.41(\mathrm{~m}, 14 \mathrm{H}), 1.22 / 1.22 / 1.21(\mathrm{~s}$, $3 \mathrm{H})$.

## tert-Butyl 3-((2S,3R)-3-tert-butoxycarbonyloxymethyl-3-(2-[3-(3-methoxy-4-phenylbutyl)-(2R,3R)-3-methyloxiranyl]ethyl)oxiranyl)propyl carbonate (ent-2.21)



A solution of the diepoxide $\mathbf{B} 3(39 \mathrm{mg}, 0.103 \mathrm{mmol})$
in dry toluene $(1.0 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with N methylimidazole ( $16 \mu \mathrm{~L}, 0.206 \mathrm{mmol}$ ) followed by di-tert-butyl dicarbonate ( $90 \mathrm{mg}, 0.412 \mathrm{mmol}$ ). The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. After that time, the reaction was quenched with water $(5 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 15 \mathrm{~mL})$. The organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and concentrated. The residue was purified by flash chromatography $\left(15 \%-25 \%\right.$ EtOAc in hexanes containing $1 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the tert-butyl carbonate ent-2.21 (31 $\mathrm{mg}, 52 \%$, $\mathrm{dr} \sim 2: 1$ regarding the stereochemical outcomes of the two epoxide groups) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.32-7.18(\mathrm{~m}, 5 \mathrm{H}), 4.20-4.05(\mathrm{~m}, 4 \mathrm{H}), 3.36-3.34(\mathrm{~m}$, $1 \mathrm{H}), 3.31(\mathrm{~s}, 3 \mathrm{H}), 2.90-2.81(\mathrm{~m}, 2 \mathrm{H}), 2.72-2.63(\mathrm{~m}, 2 \mathrm{H}), 1.90-1.70(\mathrm{~m}, 6 \mathrm{H}), 1.66-1.56(\mathrm{~m}, 5 \mathrm{H})$, 1.54-1.43 (m, 1H), $1.49(\mathrm{~s}, 9 \mathrm{H}), 1.20 / 1.20 / 1.19(\mathrm{~s}, 3 \mathrm{H})$.
tert-Butyl 3-[8-((R)-2-methyl-5-oxotetrahydrofuran-2-yl)-(5S,6S,8S)-2-oxo-1,3,7-trioxaspi-ro[4.5]dec-6-yl]propyl carbonate (2.32)


Diepoxide ent-2.21 (30 mg, $51.8 \quad \mu \mathrm{~mol})$ in dichloroethane/toluene ( $1.7 \mathrm{~mL}, 5: 1, \mathrm{v} / \mathrm{v}$ ) in borosilicate flask at room temperature was treated with the activated $4 \AA$ molecular sieves ( 60 mg ), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}\left(\begin{array}{ll}60 \mathrm{mg}\end{array}\right)$, $\mathrm{NaOAc}(60 \mathrm{mg})$ and N methylquinolinium hexafluorophosphate ( $7.5 \mathrm{mg}, 26 \mu \mathrm{~mol}$ ). The mixture was photoirradiated
with gentle air bubbling for 3.5 h while stirring at room temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with $\mathrm{Et}_{2} \mathrm{O}(25 \mathrm{~mL})$. The filtrate was concentrated and the resulting residue was purified by column chromatography ( $25 \%$ - $40 \%$ EtOAc in hexanes) to provide a pale yellow oil 2.31 ( $9.5 \mathrm{mg}, 43 \%$, containing small amounts of unknown materials) and another pale yellow oil ( $3.0 \mathrm{mg}, 13 \%$ ). For the major product 2.31: IR (neat) 2976, 1809, 1739, 1280, 1163, 1067.

The acetal $2.31(3.2 \mathrm{mg})$ was dissolved in acetone $(0.3 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ and Jones reagent ( $16 \mu \mathrm{~L}$ ) was added. The mixture was stirred for 1 h , then purified by column chromatography ( $50 \%$ $80 \%$ EtOAc in hexanes) to give title lactone 2.33 ( $2.6 \mathrm{mg}, 84 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( 500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 4.54(\mathrm{~d}, J=8.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.41(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.08(\mathrm{t}, J=5.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.50$ $(\mathrm{dd}, J=10.6,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.46(\mathrm{dd}, J=11.6,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.62-2.58(\mathrm{~m}, 2 \mathrm{H}), 2.24(\mathrm{ddd}, J=13.1$, $9.6,6.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.18(\mathrm{td}, J=12.8,3.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.10(\mathrm{dt}, J=13.4,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 1,96-1.91(\mathrm{~m}$, $2 H), 1.90-1.86(\mathrm{~m}, 1 \mathrm{H}), 1.74-1.70(\mathrm{~m}, 2 \mathrm{H}), 1.53-1.46(\mathrm{~m}, 1 \mathrm{H}), 1.49(\mathrm{~s}, 9 \mathrm{H}), 1.39-1.33(\mathrm{~m}, 1 \mathrm{H})$, $1.35(\mathrm{~s}, 3 \mathrm{H})$.
(5S,6S,8S)-6-(3-Hydroxypropyl)-8-((R)-2-methyl-5-oxotetrahydrofuran-2-yl)-1,3,7-
trioxaspiro[4.5]decan-2-one (2.35)


A solution of lactone $2.33(7.2 \mathrm{mg}, 17.4 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.8$ $\mathrm{mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with 2,6 -lutidine $(7.1 \mu \mathrm{~L}, 61 \mu \mathrm{~mol})$ followed by TMSOTf ( $10 \mu \mathrm{~L}, 52 \mu \mathrm{~mol}$ ). The reaction was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h , and then quenched with saturated $\mathrm{NaHCO}_{3}(0.5 \mathrm{~mL})$. Anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ was added and the mixture was filtered. The residue was washed with EtOAc $(40 \mathrm{~mL})$ and the filtrate was concentrated. The resulting residue was purified by column chromatography ( $70 \%-100 \%$

EtOAc in hexanes) to give the alcohol $2.35(5.4 \mathrm{mg}, 98 \%)$ as a white solid: ${ }^{1} \mathrm{H}$ NMR ( 500 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 4.54(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.11(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.67(\mathrm{br} \mathrm{s}, 2 \mathrm{H}), 3.54(\mathrm{dd}, J=11.9$, $1.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.50(\mathrm{dd}, J=11.8,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.68-2.54(\mathrm{~m}, 2 \mathrm{H}), 2.24(\mathrm{ddd}, J=13.1,10.2,5.5 \mathrm{~Hz}$, $1 \mathrm{H}), 2.19$ (ddd, $J=3.0,4.2,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.11(\mathrm{dt}, J=13.4,4.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.94-1.88(\mathrm{~m}, 2 \mathrm{H})$, $1.78-1.69(\mathrm{~m}, 2 H), 1.66-1.60(\mathrm{~m}, 2 \mathrm{H}), 1.66-1.60(\mathrm{~m}, 1 \mathrm{H}), 1.53-1.48(\mathrm{~m}, 1 \mathrm{H}), 1.40-1.32(\mathrm{~m}, 1 \mathrm{H})$, 1.37 (s, 3H).

3-(5S,6S,8S)-[8-((R)-2-Methyl-5-oxotetrahydrofuran-2-yl)-2-oxo-1,3,7-trioxaspiro[4.5]dec-6-yl]propionaldehyde (ent-2.19)


A solution of alcohol $2.35(5.2 \mathrm{mg}, 16 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ was treated with $\mathrm{NaHCO}_{3}(5.5 \mathrm{mg}, 66 \mu \mathrm{~mol})$ and Dess-Martin periodinane ( $10.5 \mathrm{mg}, 25 \mu \mathrm{~mol}$ ) sequentially. The mixture was stirred for 30 min , then loaded onto column and purified ( $60 \%$ $80 \%$ EtOAc in hexanes) to give the aldehyde ent-2.19 (3.2 mg, 61\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $\left.500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.77(\mathrm{~s}, 1 \mathrm{H}), 4.57(\mathrm{~d}, J=8.9 \mathrm{~Hz}, 1 \mathrm{H}), 4.13(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.52(\mathrm{dd}, J$ $=11.1,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.47(\mathrm{dd}, J=11.8,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.67-2.54(\mathrm{~m}, 4 \mathrm{H}), 2.23-2.17(\mathrm{~m}, 2 \mathrm{H}), 2.10$ $(\mathrm{dt}, J=8.6,4.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.99-1.86(\mathrm{~m}, 3 \mathrm{H}), 1.84-1.76(\mathrm{~m}, 1 \mathrm{H}), 1.39-1.32(\mathrm{~m}, 2 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H})$.

## 2-(tert-Butyldiphenylsilanyloxymethyl)hepta-1,6-dien-3-ol (2.38)

 A solution of methyl ester $2.37(2.050 \mathrm{~g}, 12.0 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(24 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with $\mathrm{LiAlH}_{4}\left(1 \mathrm{M}\right.$ in $\left.\mathrm{Et}_{2} \mathrm{O}, 12.0 \mathrm{~mL}\right)$ dropwise over 20 min . The reaction was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h and then quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ $(20 \mathrm{~mL})$ cautiously. Saturated sodium tartrate solution $(40 \mathrm{~mL})$ was added and the mixture was
stirred vigorously for 2 h . The mixture was extracted with EtOAc ( $3 \times 60 \mathrm{~mL}$ ) and the organic extract was dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and concentrated. The crude product was purified by column chromatography $(40 \%-60 \%$ EtOAc in hexanes) to give a colorless oil $(1.033 \mathrm{~g})$. This oil was separated into two parts ( 325 mg and 708 mg ). The first part ( 325 mg ) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(10 \mathrm{~mL})$ and treated with $\operatorname{TBDPSCl}(0.47 \mathrm{~mL}, 1.83 \mathrm{mmol})$ and imidazole $(155 \mathrm{mg}, 2.28 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 15 min , then quenched with water $(20 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 25 \mathrm{~mL})$. The organic extract was dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The crude product was purified by column chromatography ( $5 \%-10 \% \mathrm{EtOAc}$ in hexanes) to give desired silyl ether. The second part ( 708 mg ) was treated in a similar manner with TBDPSCl ( $1.15 \mathrm{~mL}, 4.48 \mathrm{mmol}$ ) and imidazole ( $373 \mathrm{mg}, 5.48 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$. The combined silyl ether 2.38 ( $1.180 \mathrm{~g}, 26 \%$, two steps) was obtained as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.71-7.69(\mathrm{~m}, 4 \mathrm{H}), 7.48-7.38(\mathrm{~m}, 6 \mathrm{H}), 5.89-5.75(\mathrm{~m}, 1 \mathrm{H}), 5.18(\mathrm{q}, J=1.4 \mathrm{~Hz}$, $1 \mathrm{H}), 5.12(\mathrm{~s}, 1 \mathrm{H}), 5.06-4.96(\mathrm{~m}, 2 \mathrm{H}), 4.32(\mathrm{~d}, J=13.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.24-4.18(\mathrm{~m}, 2 \mathrm{H}), 2.25(\mathrm{~d}, J=$ $4.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.20-2.01(\mathrm{~m}, 2 \mathrm{H}), 1.71-1.63(\mathrm{~m}, 2 \mathrm{H}), 1.08(\mathrm{~s}, 9 \mathrm{H})$.

## Ethyl (Z)-4-(tert-butyldiphenylsilanyloxymethyl)nona-4,8-dienoate (B4)



A mixture of allylic alcohol $2.38(1.180 \mathrm{~g}, 3.10 \mathrm{mmol})$, triethyl orthoacetate $(2.3 \mathrm{~mL}, 12.4 \mathrm{mmol})$ and propionic acid $(11 \mu \mathrm{~L}, 0.16$ mmol) was heated to $100{ }^{\circ} \mathrm{C}$ for 6 h and the unreacted triethyl orthoacetate was removed by distillation under reduced pressure. The residue was purified by column chromatography ( $1 \%-4 \%$ EtOAc in hexanes) to give ethyl ester B4 $(0.903 \mathrm{~g}, 65 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.71-7.68(\mathrm{~m}, 4 \mathrm{H}), 7.47-7.36(\mathrm{~m}, 6 \mathrm{H}), 5.76-5.63(\mathrm{~m}$,
$1 \mathrm{H}), 5.24(\mathrm{t}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 4.97-4.89(\mathrm{~m}, 2 \mathrm{H}), 4.21(\mathrm{~s}, 2 \mathrm{H}), 4.14(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 2.56-2.45$ $(\mathrm{m}, 4 \mathrm{H}), 2.01-1.85(\mathrm{~m}, 4 \mathrm{H}), 1.26(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}), 1.05(\mathrm{~s}, 9 \mathrm{H})$.

## (Z)-6-(tert-Butyldiphenylsilanyloxymethyl)-2-methylundeca-1,6,10-trien-3-ol (2.39)



A solution of the ethyl ester B4 $(897 \mathrm{mg}, 1.99 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(6$ $\mathrm{mL})$ at $-78{ }^{\circ} \mathrm{C}$ was treated with DIBAL-H $(1 \mathrm{M}$ in hexanes, 2.10 $\mathrm{mL}, 2.1 \mathrm{mmol}$ ) over 30 min . After 40 min , DIBAL-H ( 0.30 mL , 0.30 mmol ) was added over 5 min . The mixture was stirred for 15 min and isopropenylmagnesium bromide ( 0.5 M in THF, $6.0 \mathrm{~mL}, 3.0 \mathrm{mmol}$ ) was added dropwise. The reaction was stirred at $-78{ }^{\circ} \mathrm{C}$ for 15 min , then warmed to room temperature and quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution $(10 \mathrm{~mL})$ and saturated sodium tartrate solution $(8 \mathrm{~mL})$. The mixture was stirred vigorously for 1 h and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{~mL})$. The organic extract was dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $3 \%-$ $10 \%$ EtOAc in hexanes) to give secondary alcohol 2.39 ( $678 \mathrm{mg}, 76 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.71-7.68(\mathrm{~m}, 4 \mathrm{H}), 7.47-7.36(\mathrm{~m}, 6 \mathrm{H}), 5.76-5.65(\mathrm{~m}, 1 \mathrm{H}), 5.26(\mathrm{t}, J=$ $7.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.99-4.90(\mathrm{~m}, 3 \mathrm{H}), 4.87-4.85(\mathrm{~m}, 1 \mathrm{H}), 4.24-4.16(\mathrm{~m}, 2 \mathrm{H}), 4.11-4.06(\mathrm{~m}, 1 \mathrm{H}), 2.29-$ $2.22(\mathrm{~m}, 2 \mathrm{H}), 2.04-1.89(\mathrm{~m}, 4 \mathrm{H}), 1.79-1.66(\mathrm{~m}, 2 \mathrm{H}), 1.74(\mathrm{~s}, 3 \mathrm{H}), 1.62(\mathrm{~d}, J=3.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.06$ (s, 9H).

Ethyl (4E,8Z)-8-(tert-butyldiphenylsilanyloxymethyl)-4-methyltrideca-4,8,12-trienoate (B5)


A mixture of allylic alcohol 2.39 ( $670 \mathrm{mg}, 1.49 \mathrm{mmol}$ ), triethyl orthoacetate $(1.1 \mathrm{~mL}, 6.0 \mathrm{mmol})$ and propionic $\operatorname{acid}(5.5 \mu \mathrm{~L}, 74 \mu \mathrm{~mol})$ was heated to $135^{\circ} \mathrm{C}$ for 2 h and
then purified by column chromatography ( $1 \%-3 \%$ EtOAc in hexanes) to give ethyl ester B5 ( $594 \mathrm{mg}, 77 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.71-7.68(\mathrm{~m}, 4 \mathrm{H}), 7.46-7.36$ $(\mathrm{m}, 6 \mathrm{H}), 5.76-5.65(\mathrm{~m}, 1 \mathrm{H}), 5.23-5.15(\mathrm{~m}, 2 \mathrm{H}), 5.01-4.91(\mathrm{~m}, 2 \mathrm{H}), 4.19(\mathrm{~s}, 2 \mathrm{H}), 4.13(\mathrm{q}, J=7.2$ $\mathrm{Hz}, 2 \mathrm{H}), 2.43-2.37(\mathrm{~m}, 2 \mathrm{H}), 2.35-2.28(\mathrm{~m}, 2 \mathrm{H}), 2.22-2.10(\mathrm{~m}, 4 \mathrm{H}), 2.03-1.88(\mathrm{~m}, 4 \mathrm{H}), 1.61(\mathrm{~s}$, $3 \mathrm{H}), 1.26(\mathrm{t}, J=7.1 \mathrm{~Hz}, 3 \mathrm{H}), 1.05(\mathrm{~s}, 9 \mathrm{H})$.
(5E,9Z)-9-(tert-Butyldiphenylsilanyloxymethyl)-5-methyl-1-phenyltetradeca-5,9,13-trien-2ol (B6)


A solution of the ethyl ester B5 (588 mg, 1.13 mmol$)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5.6 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$ was added DIBAL-H (1 M in hexanes, $1.2 \mathrm{~mL}, 1.2 \mathrm{mmol}$ ) over 15 min . After 30 min ,

DIBAL-H ( $0.20 \mathrm{~mL}, 0.20 \mathrm{mmol}$ ) was added over 5 min . The mixture was stirred for 20 min .
In a separate round-bottom flask, to a mixture of $\mathrm{CuCN}(405 \mathrm{mg}, 4.52 \mathrm{mmol})$ and $\mathrm{LiCl}(422 \mathrm{mg}$, $9.94 \mathrm{mmol})$ in THF $(14 \mathrm{~mL})$ at $-78^{\circ} \mathrm{C}$ was added $\mathrm{BnMgCl}(2 \mathrm{M}$ in THF, $2.3 \mathrm{mmol}, 4.6 \mathrm{mmol})$ dropwise over 15 min . After $1 \mathrm{~h}, \mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(0.28 \mathrm{~mL}, 2.26 \mathrm{mmol})$ was added and the mixture was stirred for 5 min . The reaction mixture from the first reaction was cannulated to the second flask followed by rinse ( $2 \times 1 \mathrm{~mL}$ THF). The mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h and then at room temperature for 2 h . After that time, the reaction was quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution $(10 \mathrm{~mL}) /$ saturated sodium tartrate solution $(8 \mathrm{~mL})$ and stirred vigorously for 1 h . The mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{~mL})$ and the organic extract was dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was purified by column chromatography (3\% - $12 \%$ EtOAc in hexanes) to give secondary alcohol $\mathbf{B 6}(520 \mathrm{mg}, 81 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.73-7.70(\mathrm{~m}, 4 \mathrm{H}), 7.48-7.37(\mathrm{~m}, 6 \mathrm{H}), 7.35-7.22(\mathrm{~m}, 5 \mathrm{H}), 5.77-5.66(\mathrm{~m}, 1 \mathrm{H}), 5.24-5.19$
$(\mathrm{m}, 2 \mathrm{H}), 4.99-4.91(\mathrm{~m}, 2 \mathrm{H}), 4.20(\mathrm{~s}, 2 \mathrm{H}), 3.87-3.78(\mathrm{~m}, 1 \mathrm{H}), 2.84(\mathrm{dd}, J=13.5,4.4 \mathrm{~Hz}, 1 \mathrm{H})$, $2.68(\mathrm{dd}, J=13.5,8.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.24-2.08(\mathrm{~m}, 6 \mathrm{H}), 2.06-1.88(\mathrm{~m}, 4 \mathrm{H}), 1.62(\mathrm{~s}, 3 \mathrm{H}), 1.66-1.58$ $(\mathrm{m}, 2 \mathrm{H}), 1.07(\mathrm{~s}, 9 \mathrm{H})$.
((2Z,5E)-9-Methoxy-6-methyl-2-(pent-4-enylidene)-10-phenyldec-5-enyloxy)(tertbutyl)diphenylsilane (B7)

A solution of the secondary alcohol $\mathbf{B 6}(510 \mathrm{mg}, 0.900$
 $\mathrm{mmol})$ in DMF $(8 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with $\mathrm{NaH}(60 \%$ weight in mineral oil, $90 \mathrm{mg}, 2.25 \mathrm{mmol}$ ). After 30 min , MeI ( $0.22 \mathrm{~mL}, 3.60 \mathrm{mmol}$ ) was added and the mixture was stirred overnight at room temperature. After that time, the reaction was quenched with water ( 20 mL ) cautiously and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 30 \mathrm{~mL})$ and the organic extract was dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $1 \%-4 \%$ EtOAc in hexanes) to give the methyl ether B7 (474 mg, 91\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 7.72-7.66 (m, $4 \mathrm{H}), 7.47-7.36(\mathrm{~m}, 6 \mathrm{H}), 7.32-7.25(\mathrm{~m}, 2 \mathrm{H}), 7.23-7.18(\mathrm{~m}, 3 \mathrm{H}), 5.79-5.66(\mathrm{~m}, 1 \mathrm{H}), 5.21(\mathrm{app} \mathrm{t}, J$ $=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{appq} \mathrm{q}, ~ J=6.8,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.99-4.91(\mathrm{~m}, 2 \mathrm{H}), 4.20(\mathrm{~s}, 2 \mathrm{H}), 3.39-3.32(\mathrm{~m}$, $1 \mathrm{H}), 3.32(\mathrm{~s}, 3 \mathrm{H}), 2.86(\mathrm{dd}, J=13.7,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.72(\mathrm{dd}, J=13.7,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.25-2.07(\mathrm{~m}$, $5 \mathrm{H}), 2.04-1.88(\mathrm{~m}, 5 \mathrm{H}), 1.59-1.51(\mathrm{~m}, 2 \mathrm{H}), 1.54(\mathrm{br} \mathrm{s}, 3 \mathrm{H}), 1.06(\mathrm{~s}, 9 \mathrm{H})$.

## (2Z,5E)-9-Methoxy-6-methyl-2-(pent-4-enylidene)-10-phenyldec-5-en-1-ol (2.40)



A solution of the silyl ether B7 ( $464 \mathrm{mg}, 0.799 \mathrm{mmol})$ in THF ( 5 mL ) was treated with TBAF ( 1 M in THF, 1.9 mL , 1.9 mmol ). The reaction was stirred for 4 h , then
concentrated and the residue was purified by column chromatography ( $10 \%-22 \% \mathrm{EtOAc}$ in hexanes) to give the trienol $2.40(230 \mathrm{mg}, 84 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 7.32-7.28 (m, 2H), 7.23-7.18 (m, 3H), $5.82(\mathrm{tdd}, J=16.8,10.2,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.31(\mathrm{t}, J=6.8 \mathrm{~Hz}$, $1 \mathrm{H}), 5.14-5.10(\mathrm{~m}, 1 \mathrm{H}), 5.06-4.98(\mathrm{~m}, 2 \mathrm{H}), 4.12(\mathrm{~d}, J=5.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.38-3.30(\mathrm{~m}, 1 \mathrm{H}), 3.32(\mathrm{~s}$, $3 \mathrm{H}), 2.86(\mathrm{dd}, J=13.7,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.72(\mathrm{dd}, J=13.7,6.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.23-1.96(\mathrm{~m}, 10 \mathrm{H}), 1.57-$ $1.50(\mathrm{~m}, 2 \mathrm{H}), 1.55(\mathrm{~s}, 3 \mathrm{H}), 1.15(\mathrm{t}, J=5.7 \mathrm{~Hz}, 1 \mathrm{H})$.
((2S,3R)-3-(But-3-enyl)-2-((E)-7-methoxy-4-methyl-8-phenyloct-3-enyl)oxiran-2yl)methanol (B8)


A mixture of trienol $2.40(223 \mathrm{mg}, 0.651 \mathrm{mmol})$ and activated $4 \AA$ molecular sieves $(195 \mathrm{mg})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(6 \mathrm{~mL})$ at $-20^{\circ} \mathrm{C}$ was treated with L-diisopropyl tartrate $(16 \mu \mathrm{~L}$, $78 \mu \mathrm{~mol})$. After $15 \mathrm{~min}, \mathrm{Ti}\left(\mathrm{O}^{i} \operatorname{Pr}\right)_{4}(20 \mu \mathrm{~L}, 65 \mu \mathrm{~mol})$ was introduced. The mixture was stirred for 30 min , and ${ }^{t} \mathrm{BuOOH}(5-6 \mathrm{M}$ in decane, $0.36 \mathrm{~mL}, \sim 1.9 \mathrm{mmol}$ ) was added dropwise. The reaction was stirred for 40 min at $-20^{\circ} \mathrm{C}$, and then water $(0.3 \mathrm{~mL})$ was added. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h . After that time, $30 \% \mathrm{NaOH}$ saturated with $\mathrm{NaCl}(0.3 \mathrm{~mL})$ was added and the mixture was stirred at room temperature for 4 h . The mixture was filtered through a $1: 1$ mixture of $\mathrm{MgSO}_{4} /$ Celite and the residue was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The combined filtrates were concentrated and the residue was purified by column chromatography ( $20 \%-25 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give a colorless oil ( 150 mg , containing $7 \mathrm{~mol} \%$ (+)-DIPT) which was treated with a mixture of $30 \% \mathrm{NaOH}$ saturated with $\mathrm{NaCl}(0.3 \mathrm{~mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3$ mL ) for 5 h . The mixture was concentrated and the residue was purified by column chromatography $\left(20 \%-25 \% \mathrm{EtOAc}\right.$ in hexanes containing $\left.0.5 \% \mathrm{Et}_{3} \mathrm{~N}\right)$ to give the monoepoxide

B8 (134 mg, 58\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.31-7.20(\mathrm{~m}, 5 \mathrm{H}), 5.84$ (tdd, $J=16.9,10.2,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.10-5.02(\mathrm{~m}, 3 \mathrm{H}), 3.76(\mathrm{dd}, J=11.8,6.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.67(\mathrm{dd}, J=11.8$, $5.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.36-3.34(\mathrm{~m}, 1 \mathrm{H}), 3.32(\mathrm{~s}, 3 \mathrm{H}), 2.90-2.84(\mathrm{~m}, 2 \mathrm{H}), 2.71(\mathrm{dd}, J=13.7,6.2 \mathrm{~Hz}, 1 \mathrm{H})$, 2.28-1.99 (m, 6H), 1.94-1.90 (m, 1H), 1.78-1.45 (m, 6H), $1.55(\mathrm{~s}, 3 \mathrm{H})$.

## ((2S,3R)-3-(But-3-enyl)-2-(2-((2S,3S)-3-(3-methoxy-4-phenylbutyl)-3-methyloxiran-2-

 yl)ethyl)oxiran-2-yl)methanol (B9)

A solution of $\mathbf{B 8}(125 \mathrm{mg}, 0.349 \mathrm{mmol})$ in $\mathrm{CH}_{3} \mathrm{CN} / \mathrm{DMM}$ $(5.2 \mathrm{~mL}, 1: 2, \mathrm{v} / \mathrm{v})$ was treated with a 0.05 M solution of $\mathrm{Na}_{2} \mathrm{~B}_{4} \mathrm{O}_{7}$ in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}$ (EDTA) ( 3.5 mL ), $\mathrm{Bu}_{4} \mathrm{NHSO}_{4}$
( $4.7 \mathrm{mg}, 14.0 \mu \mathrm{~mol}$ ) and ent-Shi ketone ( $27.0 \mathrm{mg}, 0.105 \mathrm{mmol}$ ) sequentially. The mixture was cooled to $-5^{\circ} \mathrm{C}$. Oxone ( $268 \mathrm{mg}, 0.436 \mathrm{mmol}$ ), dissolved in $4 \times 10^{-4} \mathrm{M} \mathrm{Na}_{2}$ (EDTA) ( 2.0 mL ), and $\mathrm{K}_{2} \mathrm{CO}_{3}(253 \mathrm{mg}, 1.83 \mathrm{mmol})$, dissolved in water $(2.0 \mathrm{~mL})$, were added simultaneously via a syringe pump over 2.0 h . After the addition was completed, the slightly blue reaction mixture was stirred further for 15 min , then diluted with water $(10 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{x}$ $15 \mathrm{~mL})$. The organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and concentrated. The residue was purified by flash chromatography ( $25 \%-38 \%$ EtOAc in hexanes) to give unreacted starting material B8 ( $37.4 \mathrm{mg}, 30 \%$ ) and the diepoxide B9 ( $79.7 \mathrm{mg}, 61 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta$ 7.32-7.19 (m, 5H), 5.91-5.78 (m, 1H), 5.11-5.01 (m, 2H), 3.73-3.71 (m, $2 \mathrm{H}), 3.38-3.34(\mathrm{~m}, 1 \mathrm{H}), 3.32(\mathrm{~s}, 3 \mathrm{H}), 2.92-2.83(\mathrm{~m}, 2 \mathrm{H}), 2.71-2.65(\mathrm{~m}, 1 \mathrm{H}), 2.32-2.18(\mathrm{~m}, 2 \mathrm{H})$, $2.03-1.91(\mathrm{~m}, 1 \mathrm{H}), 1.87(\mathrm{dt}, J=5.9,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.81-1.41(\mathrm{~m}, 8 \mathrm{H}), 1.22(\mathrm{~s}, 3 \mathrm{H})$.
tert-Butyl ((2S,3R)-3-(but-3-enyl)-2-(2-((2S,3S)-3-(3-methoxy-4-phenylbutyl)-3-methyloxira-n-2-yl)ethyl)oxiran-2-yl)methyl carbonate (2.36)


A solution of the diepoxy alcohol B9 (78 mg, 0.208 $\mathrm{mmol})$ in dry toluene $(2.0 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with di-tert-butyl dicarbonate ( $91 \mathrm{mg}, 0.416 \mathrm{mmol}$ ) and 1methylimidazole ( $16 \mu \mathrm{~L}, 0.208 \mathrm{mmol}$ ). The reaction mixture was stirred overnight, allowing the temperature to warm to room temperature slowly. After that time, the reaction was quenched with water $(10 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 15 \mathrm{~mL})$. The organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and concentrated. The residue was purified by flash chromatography ( $15 \%$ $20 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the tert-butyl carbonate $2.36(90 \mathrm{mg}$, $91 \%, \mathrm{dr} \sim 4.6: 1$ regarding the stereochemical outcomes of the two epoxide groups) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.31-7.28(\mathrm{~m}, 2 \mathrm{H}), 7.23-7.19(\mathrm{~m}, 3 \mathrm{H}), 5.85-5.80(\mathrm{~m}, 1 \mathrm{H})$, $5.08(\mathrm{td}, J=17.1,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.02(\mathrm{dd}, J=10.4,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 4.21-4.12(\mathrm{~m}, 2 \mathrm{H}), 3.38-3.33(\mathrm{~m}$, $1 \mathrm{H}), 3.31(\mathrm{~s}, 3 \mathrm{H}), 2.89-2.83(\mathrm{~m}, 2 \mathrm{H}), 2.70-2.65(\mathrm{~m}, 2 \mathrm{H}), 2.29-2.21(\mathrm{~m}, 2 \mathrm{H}), 1.85-1.81(\mathrm{~m}, 1 \mathrm{H})$, $1.78-1.60(\mathrm{~m}, 7 \mathrm{H}), 1.54-1.42(\mathrm{~m}, 2 \mathrm{H}), 1.45(\mathrm{~s}, 9 \mathrm{H}), 1.21 / 1.20(\mathrm{~s}, 3 \mathrm{H})$.
(5R,6R,8R)-6-But-3-enyl-8-((S)-2-methyl-5-oxo-tetrahydrofuran-2-yl)-1,3,7-trioxaspiro[4.5]decan-2-one (B10)


Diepoxide $2.36(44 \mathrm{mg}, 92.7 \mu \mathrm{~mol})$ in dichloroethane/toluene ( 3.1 $\mathrm{mL}, 5: 1, \mathrm{v} / \mathrm{v}$ ) in borosilicate flask at room temperature was treated with activated $4 \AA$ molecular sieves ( 88 mg ), anhydrous $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ $(88 \mathrm{mg}), \mathrm{NaOAc}(88 \mathrm{mg})$ and $N$-methylquinolinium hexafluorophosphate ( $13.4 \mathrm{mg}, 46.4 \mu \mathrm{~mol}$ ). The mixture was photoirradiated with gentle air bubbling for 6.5 h while stirring at room
temperature. The reaction mixture was filtered through a small plug of silica gel and the residue was washed with EtOAc $(20 \mathrm{~mL})$. The filtrate was concentrated and the resulting residue was purified by flash chromatography ( $25 \%-35 \%$ EtOAc in hexanes) to provide a pale yellow oil $(10.5 \mathrm{mg})$, which was dissolved in acetone $(1.0 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. Jones reagent $(2.67 \mathrm{M}, 6$ drops) was added dropwise ( 1 drop $/ 5 \mathrm{~min}$ ). After completion of addition, the mixture was stirred for 1 h and then concentrated. The residues was purified by column chromatography (50\%-70\% EtOAc in hexanes) to give the lactone $\mathbf{B 1 0}$ ( $4.6 \mathrm{mg}, 17 \%$, two steps) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 5.83-5.69(\mathrm{~m}, 1 \mathrm{H}), 5.11-5.01(\mathrm{~m}, 2 \mathrm{H}), 4.54(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.12(\mathrm{~d}, J=9.0 \mathrm{~Hz}$, $1 \mathrm{H}), 3.50(\mathrm{dd}, J=10.8,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.46(\mathrm{dd}, J=11.7,2.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.64-2.57(\mathrm{~m}, 2 \mathrm{H}), 2.32-$ $2.05(\mathrm{~m}, 5 \mathrm{H}), 1.97-1.87(\mathrm{~m}, 2 \mathrm{H}), 1.69(\mathrm{dtd}, J=13.6,8.1,2.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.49(\mathrm{dtd}, J=13.6,5.7$, $2.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.40-1.32(\mathrm{~m}, 1 \mathrm{H}), 1.36(\mathrm{~s}, 3 \mathrm{H})$.

## 3-((5R,6R,8R)-[8-((S)-2-Methyl-5-oxotetrahydrofuran-2-yl)-2-oxo-1,3,7-trioxaspiro[4.5]dec-6-yl])propionaldehyde (2.19)



A solution of the lactone $\mathbf{B 1 0}(3.3 \mathrm{mg}, 10.6 \mu \mathrm{~mol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.8$ $\mathrm{mL})$ at $-78{ }^{\circ} \mathrm{C}$ was treated dropwise with a saturated solution of $\mathrm{O}_{3}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $-78^{\circ} \mathrm{C}$ until all the starting material disappeared. $\mathrm{PPh}_{3}$ (a crystal) was added and the cold bath was removed. The reaction was stirred for 10 h , then concentrated and the residues was purified by column chromatography ( $60 \%-80 \% \mathrm{EtOAc}$ in hexanes) to give the aldehyde 2.19 ( 3.6 mg (containing $\sim 17 \% \mathrm{Ph}_{3} \mathrm{P}=\mathrm{O}, 83 \%$ ) as a colorless oil, which has identical spectral data to ent-2.19.

## (S)-1-((2R,5R)-Tetrahydro-5-(2-hydroxypropan-2-yl)-2-methylfuran-2-yl)ethane-1,2-diol

## (2.42)



A mixture of AD-mix- $\beta(12.35 \mathrm{~g})$ in ${ }^{t} \mathrm{BuOH} / \mathrm{H}_{2} \mathrm{O}(88 \mathrm{~mL}, 1: 1, \mathrm{v} / \mathrm{v})$ at 0 ${ }^{\circ} \mathrm{C}$ was treated with $\mathrm{CH}_{3} \mathrm{SO}_{2} \mathrm{NH}_{2}(0.839 \mathrm{~g}, 8.82 \mathrm{mmol})$. After 15 min , the epoxy alcohol $(1.50 \mathrm{~g}, 8.82 \mathrm{mmol}$, prepared from geraniol through Sharpless asymmetric epoxidation) was added dropwise and the flask formerly containing the epoxy alcohol was rinsed with ${ }^{t} \mathrm{BuOH} / \mathrm{H}_{2} \mathrm{O}(2 \times 2 \mathrm{~mL}, 1: 1, \mathrm{v} / \mathrm{v})$. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 2 days, then concentrated to $\sim 50 \mathrm{~mL}$ and extracted with EtOAc ( $25 \times 30 \mathrm{~mL}$ ). The combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated. The residue was purified by column chromatography $(90 \%-100 \%$ EtOAc in hexanes followed by $5 \%-15 \% \mathrm{MeOH}$ in EtOAc) to give a mixture of 2.22 and $2.42(1.720 \mathrm{~g})$. This mixture was dissolved in $\mathrm{PhMe}(200 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$ and CSA•pyridine ( $262 \mathrm{mg}, 0.842 \mathrm{mmol}$ ) was added. The reaction was stirred for 1.5 h , and $\mathrm{Et}_{3} \mathrm{~N}$ ( 1 mL ) was added. The solvent was removed under reduced pressure and the residue was purified by column chromatography ( $90 \%-100 \%$ EtOAc in hexanes followed by $5 \%-15 \%$ MeOH in EtOAc) to give triol $2.42\left(1.627 \mathrm{~g}, 90 \%\right.$, two steps) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.79(\mathrm{t}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.75-3.70(\mathrm{~m}, 2 \mathrm{H}), 3.58-3.52(\mathrm{~m}, 1 \mathrm{H}), 2.88(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, $2.53-2.51(\mathrm{~m}, 1 \mathrm{H}), 2.18(\mathrm{~s}, 1 \mathrm{H}), 2.10(\mathrm{td}, J=11.3,10.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.90-1.81(\mathrm{~m}, 2 \mathrm{H}), 1.65-1.57$ $(\mathrm{m}, 1 \mathrm{H}), 1.22(\mathrm{~s}, 3 \mathrm{H}), 1.18(\mathrm{~s}, 3 \mathrm{H}), 1.14(\mathrm{~s}, 3 \mathrm{H})$; HRMS (ESI): m/z calcd for $\mathrm{C}_{10} \mathrm{H}_{20} \mathrm{O}_{4} \mathrm{Na}(\mathrm{M}+$ $\mathrm{Na}^{+}$) 227.1259, found 227.1246.

## (R)-2-Hydroxy-2-[(5R)-5-(1-hydroxy-1-methylethyl)-(2R)-2-methyltetrahydrofuran-2-

## yl]ethyl toluene-4-sulfonate (B11)



A solution of triol $2.42(1.627 \mathrm{~g}, 7.96 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ was treated with pyridine ( $1.3 \mathrm{~mL}, 15.9 \mathrm{mmol}), \mathrm{TsCl}(1.670 \mathrm{~g}, 8.76 \mathrm{mmol})$ and DMAP ( $49 \mathrm{mg}, 0.40 \mathrm{mmol}$ ) sequentially. The reaction was stirred overnight, then concentrated and the residue was purified by column chromatography ( $30 \%-60 \% \mathrm{EtOAc}$ in hexanes) to give the tosylate $\mathbf{B 1 1}(1.284 \mathrm{~g}, 45 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.82(\mathrm{td}, J=8.4,1.9 \mathrm{~Hz}, 2 \mathrm{H}), 7.38-7.35(\mathrm{~m}, 2 \mathrm{H}), 4.26(\mathrm{dd}, J=10.4,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.02(\mathrm{dd}, J=$ $10.5,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.78(\mathrm{td}, J=7.6,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.72(\mathrm{dd}, J=8.9,6.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.52(\mathrm{~d}, J=3.2$ $\mathrm{Hz}, 1 \mathrm{H}), 2.46(\mathrm{~s}, 3 \mathrm{H}), 2.06(\mathrm{td}, J=11.9,9.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.96(\mathrm{~s}, 1 \mathrm{H}), 1.89-1.81(\mathrm{~m}, 2 \mathrm{H}), 1.66$ (ddd, $J=11.9,6.2,3.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.18(\mathrm{~s}, 3 \mathrm{H}), 1.13(\mathrm{~s}, 3 \mathrm{H}), 1.11(\mathrm{~s}, 3 \mathrm{H})$; HRMS (ESI): m/z calcd for $\mathrm{C}_{17} \mathrm{H}_{26} \mathrm{O}_{6} \mathrm{NaS}[\mathrm{M}+\mathrm{Na}]^{+}$381.1348, found 381.1340.

## 2-((2R,5R)-Tetrahydro-5-methyl-5-((S)-oxiran-2-yl)furan-2-yl)propan-2-ol (B12)



A solution of the tosylate $\mathbf{B 1 1}(270 \mathrm{mg}, 0.753 \mathrm{mmol})$ in dry $\mathrm{MeOH}(21$ mL ) was treated with anhydrous $\mathrm{K}_{2} \mathrm{CO}_{3}(104 \mathrm{mg}, 0.753 \mathrm{mmol})$. The mixture was stirred for 1.5 h , then concentrated and the residue was purified by column chromatography ( $50 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the epoxide $\mathbf{B 1 2}$ (134 $\mathrm{mg}, 96 \%)$ as a colorless oil: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.81-3.76(\mathrm{~m}, 1 \mathrm{H}), 3.04(\mathrm{dd}, J=4.1$, $2.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.74(\mathrm{dd}, J=5.0,4.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.58(\mathrm{dd}, J=5.0,2.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.11(\mathrm{~s}, 1 \mathrm{H}), 1.90-1.78$ $(\mathrm{m}, 2 \mathrm{H}), 1.68-1.58(\mathrm{~m}, 1 \mathrm{H}), 1.28(\mathrm{~s}, 3 \mathrm{H}), 1.23(\mathrm{~s}, 3 \mathrm{H}), 1.13(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 86.8,81.4,70.7,57.2,43.9,32.8,27.5,26.3,24.3,24.3$.

## (2-((2R,5R)-Tetrahydro-5-methyl-5-((S)-oxiran-2-yl)furan-2-yl)propan-2-yloxy)triethyl-

 silane (2.43)

A solution of the tertiary alcohol B12 (178 mg, 0.956 mmol$)$ in DMF $(5.8 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ were treated with imidazole $(130 \mathrm{mg}, 1.91 \mathrm{mmol})$, $\operatorname{TESCl}(0.24 \mathrm{~mL}, 1.43 \mathrm{mmol})$ and DMAP sequentially. The mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 2 h and then at room temperature overnight. After that time, the reaction was quenched with water $(15 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. The organic extract was dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and concentrated. The residue was purified by column chromatography $\left(2 \%-8 \% \mathrm{Et}_{2} \mathrm{O}\right.$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the silyl ether $2.43(261 \mathrm{mg}, 91 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 3.72(\mathrm{t}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.99(\mathrm{dd}, J=4.1,2.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.73(\mathrm{dd}, J=$ $4.9,4.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.58(\mathrm{dd}, J=5.1,2.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.98-1.73(\mathrm{~m}, 3 \mathrm{H}), 1.58(\mathrm{ddd}, J=13.7,8.2,5.2$ $\mathrm{Hz}, 1 \mathrm{H}), 1.24(\mathrm{~s}, 3 \mathrm{H}), 1.20(\mathrm{~s}, 3 \mathrm{H}), 1.19(\mathrm{~s}, 3 \mathrm{H}), 0.95(\mathrm{t}, J=8.1 \mathrm{~Hz}, 9 \mathrm{H}), 0.58(\mathrm{q}, J=7.8 \mathrm{~Hz}$, $6 \mathrm{H}){ }^{13}{ }^{13} \mathrm{CNMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 87.7,81.6,74.3,57.4,44.2,33.1,28.1,26.6,25.5,24.0,7.3$, 6.9; IR (neat) 2958, 2876, 1240, 1174, 1044, 726. HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{16} \mathrm{H}_{32} \mathrm{O}_{3} \mathrm{NaSi}$ $\left[^{M}+\mathrm{Na}\right]^{+} 323.2018$, found 323.1993.
(R)-2-Methyl-(5R)-5-(1-methyl-1-triethylsilanyloxyethyl)-2-((S)-1-triethylsilanyloxypent-4ynyl)tetrahydrofuran (B13)


A mixture of $n-\mathrm{BuLi}(1.6 \mathrm{M}$ in hexanes, $23.2 \mathrm{~mL}, 37.2 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O} \quad(8 \quad \mathrm{~mL}) \quad$ at $-78 \quad{ }^{\circ} \mathrm{C}$ was treated with tetramethylethylenediamine $(1.4 \mathrm{~mL}, 9.3 \mathrm{mmol})$ followed by dropwise addition of propargyl bromide ( $80 \%$ weight in $\mathrm{PhMe}, 2.1 \mathrm{~mL}, 18.6 \mathrm{mmol}$ ). The resulting yellow suspension was stirred at $-78{ }^{\circ} \mathrm{C}$ for 20 min , then transferred to a solution of epoxide $2.43(180 \mathrm{mg}, 0.6 \mathrm{mmol})$ in
anhydrous $\mathrm{Et}_{2} \mathrm{O}(1.0 \mathrm{~mL})$ at $-78^{\circ} \mathrm{C}$ quickly via syringe. The mixture was stirred at $-78{ }^{\circ} \mathrm{C}$ for 1 h, then quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{~mL}) /$ water $(10 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20$ $\mathrm{mL})$. The ether solution was dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and concentrated. The residue was purified by column chromatography ( $4 \%-12 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give starting material $2.43(65 \mathrm{mg}$, $36 \%$ ) and the terminal alkyne B13 (128 mg, 63\%) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.71-3.61(\mathrm{~m}, 2 \mathrm{H}), 2.49-2.40(\mathrm{~m}, 2 \mathrm{H}), 2.33(\mathrm{ddd}, J=16.8,8.0,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.04(\mathrm{td}, J=11.2$, $8.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.97(\mathrm{t}, J=2.7 \mathrm{~Hz}, 1 \mathrm{H}), 1.96-1.83(\mathrm{~m}, 2 \mathrm{H}), 1.74-1.65(\mathrm{~m}, 1 \mathrm{H}), 1.56-1.44(\mathrm{~m}, 2 \mathrm{H})$, $1.22(\mathrm{~s}, 3 \mathrm{H}), 1.18(\mathrm{~s}, 3 \mathrm{H}), 1.13(\mathrm{~s}, 3 \mathrm{H}), 0.96(\mathrm{t}, J=7.7 \mathrm{~Hz}, 9 \mathrm{H}), 0.59(\mathrm{q}, J=8.0 \mathrm{~Hz}, 6 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 88.9,85.9,84.7,75.6,74.1,68.6,31.1,30.9,27.7,26.9,26.1,24.0$, 16.1, 7.3, 7.0.

## (S)-1-[(2R)-2-Methyl-(5R)-5-(1-methyl-1-triethylsilanyloxy-ethyl)-tetrahydro-furan-2-yl]-

 pent-4-yn-1-ol (2.44)

A solution of the secondary alcohol B13 ( $120 \mathrm{mg}, 0.352 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3.5 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with imidazole $(27 \mathrm{mg}$, $0.396 \mathrm{mmol})$, $\operatorname{TESCl}(65 \mu \mathrm{~L}, 0.387 \mathrm{mmol})$ and $\operatorname{DMAP}(2.1 \mathrm{mg}, 17 \mu \mathrm{~mol})$ sequentially, and the cold bath was removed. After 1 h , the reaction mixture was concentrated and the residue was purified by column chromatography ( $1 \%-3 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give the bis-silyl ether $\mathbf{2 . 4 4}$ $(145 \mathrm{mg}, 91 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.67-3.63(\mathrm{~m}, 2 \mathrm{H}), 2.38-2.29(\mathrm{~m}$, $1 \mathrm{H}), 2.23$ (ddd, $J=16.9,7.9,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 1.99-1.74(\mathrm{~m}, 5 \mathrm{H}), 1.59-1.47(\mathrm{~m}, 2 \mathrm{H}), 1.19(\mathrm{~s}, 3 \mathrm{H})$, $1.17(\mathrm{~s}, 3 \mathrm{H}), 1.09(\mathrm{~s}, 3 \mathrm{H}), 1.00-0.93(\mathrm{~m}, 18 \mathrm{H}), 0.68-0.55(\mathrm{~m}, 12 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $87.4,85.8,85.2,76.3,74.3,68.4,34.6,32.8,28.1,26.5,25.7,22.7,15.9,7.3,7.3,7.0,5.7$.

## (2R)-2-Methyl-(5R)-5-(1-methyl-1-triethylsilanyloxyethyl)-2-((1S)-4-tributylstannanyl-1-

## triethylsilanyloxypent-4-enyl)tetrahydrofuran (B14)



A solution of ${ }^{i} \mathrm{Pr}_{2} \mathrm{NH}(0.57 \mathrm{~mL}, 4.05 \mathrm{mmol})$ in THF (3.5 $\mathrm{mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with $n-\mathrm{BuLi}(1.6 \mathrm{M}$ in hexanes, 2.3 $\mathrm{mL}, 3.6 \mathrm{mmol})$. After 30 min , the flask was cooled to $-30^{\circ} \mathrm{C}$ and a solution of $n-\mathrm{Bu} \mathrm{u}_{3} \mathrm{SnH}(0.97$ $\mathrm{mL}, 3.6 \mathrm{mmol})$ in THF ( 2.1 mL ) was added dropwise. The pale yellow solution was stirred for 1 h and $\mathrm{Et}_{2} \mathrm{AlCl}(1 \mathrm{M}$ in heptane, $3.0 \mathrm{~mL}, 3.0 \mathrm{mmol})$ was added. The mixture was stirred at $-30^{\circ} \mathrm{C}$ for 5 h to form an $n-\mathrm{Bu}_{3} \mathrm{SnAlEt}_{2}$ solution ( $\left.\sim 0.244 \mathrm{M}\right)$.

A solution of terminal alkyne $\mathbf{2 . 4 4}(140 \mathrm{mg}, 0.308 \mathrm{mmol})$ in THF $(11 \mathrm{~mL})$ at $-30^{\circ} \mathrm{C}$ was treated with $n-\mathrm{Bu}_{3} \mathrm{SnAlEt}_{2}$ solution ( $6.2 \mathrm{~mL}, \sim 1.5 \mathrm{mmol}$ ) followed by $\mathrm{CuCN}(8.3 \mathrm{mg}, 92.4 \mu \mathrm{~mol})$. After 1 h , the $n-\mathrm{Bu}_{3} \mathrm{SnAlEt}_{2}$ solution ( $2.0 \mathrm{~mL}, \sim 0.5 \mathrm{mmol}$ ) and $\mathrm{CuCN}(8.0 \mathrm{mg}, 89 \mu \mathrm{~mol})$ were added. The mixture was stirred at $-30^{\circ} \mathrm{C}$ for 5 h , and then stored at $-20^{\circ} \mathrm{C}$ overnight. The reaction was quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{~mL})$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. The ether solution was dried $\left(\mathrm{MgSO}_{4}\right)$, filtered and concentrated. The residue was purified by column chromatography ( $2 \%-2.5 \% \mathrm{Et}_{2} \mathrm{O}$ in hexanes) to give the desired vinyl stannane $\mathbf{B 1 4}(67.8 \mathrm{mg}$, 29\%) and unreacted starting material 2.44 ( $71.9 \mathrm{mg}, 51 \%$ ). For vinyl stannane: ${ }^{1} \mathrm{H}$ NMR (500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 5.70(J=84.6,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(J=38.7,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.64(\mathrm{dd}, J=5.5,3.7 \mathrm{~Hz}$, $1 \mathrm{H}), 3.52(\mathrm{dd}, J=5.0,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.42(\mathrm{dt}, J=8.2,2.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.22(\mathrm{dt}, J=7.5,2.8 \mathrm{~Hz}, 1 \mathrm{H})$, $1.97(\mathrm{dt}, J=6.5,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.92-1.84(\mathrm{~m}, 1 \mathrm{H}), 1.82-1.76(\mathrm{~m}, 1 \mathrm{H}), 1.66(\mathrm{tt}, J=8.0,2.0 \mathrm{~Hz}, 1 \mathrm{H})$, $1.57-1.29(\mathrm{~m}, 14 \mathrm{H}), 1.19(\mathrm{~s}, 3 \mathrm{H}), 1.17(\mathrm{~s}, 3 \mathrm{H}), 1.08(\mathrm{~s}, 3 \mathrm{H}), 0.99-0.88(\mathrm{~m}, 33 \mathrm{H}), 0.66-0.56(\mathrm{~m}$, 12H).

2-((1S)-4-Iodo-1-triethylsilanyloxypent-4-enyl)-(2R)-2-methyl-(5R)-5-(1-methyl-1-triethylsilanyloxy-ethyl)tetrahydrofuran (2.45)


A solution of the vinyl stannane B14 ( $62 \mathrm{mg}, 83 \mu \mathrm{~mol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1.6 \mathrm{~mL})$ was treated with $\mathrm{I}_{2}(23 \mathrm{mg}, 91 \mu \mathrm{~mol})$. The slightly purple solution was stirred for 10 min , then quenched with saturated $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ solution (5 $\mathrm{mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 15 \mathrm{~mL})$. The organic extract was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, filtered and concentrated. The residue was purified by column chromatography $\left(2 \% \mathrm{Et}_{2} \mathrm{O}\right.$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the vinyl iodide $2.45(41.7 \mathrm{mg}, 86 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 6.04(\mathrm{q}, J=1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.69(\mathrm{~d}, J=0.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.66-3.61(\mathrm{~m}, 1 \mathrm{H}), 3.54$ (dd, $J=7.5,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.60-2.41(\mathrm{~m}, 2 \mathrm{H}), 1.99-1.73(\mathrm{~m}, 4 \mathrm{H}), 1.60-1.50(\mathrm{~m}, 2 \mathrm{H}), 1.19(\mathrm{~s}, 3 \mathrm{H})$, $1.17(\mathrm{~s}, 3 \mathrm{H}), 1.10(\mathrm{~s}, 3 \mathrm{H}), 1.00-0.93(\mathrm{~m}, 18 \mathrm{H}), 0.67-0.54(\mathrm{~m}, 12 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 125.2,113.1,87.4,85.9,76.8,74.2,42.9,34.9,33.9,28.0,26.5,25.8,22.6,7.4,7.0,5.7$; IR (neat) 2956, 2876, 1459, 1238, 1173, 1099, 1068. HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{25} \mathrm{H}_{51} \mathrm{IO}_{3} \mathrm{NaSi}$ $[\mathrm{M}+\mathrm{Na}]^{+} 605.2319$, found 605.2341.
















































## APPENDIX C

## MULTICOMPONENT APPROACH TO THE SYNTHESIS OF OXIDIZED AMIDES THROUGH NITRILE HYDROZIRCONATION (SUPPORTING INFORMATION)

General Experimental Proton ( ${ }^{1} \mathrm{H}$ NMR) and carbon $\left({ }^{13} \mathrm{C}\right.$ NMR) nuclear magnetic resonance spectra were recorded at ambient temperature on Bruker Avance 300 spectrometer at 300 MHz and 75 MHz or Bruker Avance 500 spectrometer at 500 MHz and 125 MHz if specified. The chemical shifts are given in parts per million (ppm) on the delta ( $\delta$ ) scale. The solvent peak was used as a reference value, for ${ }^{1} \mathrm{H}$ NMR: $\mathrm{CDCl}_{3}=7.27 \mathrm{ppm}, \mathrm{CD}_{3} \mathrm{OD}=3.31$, for ${ }^{13} \mathrm{C}$ NMR: $\mathrm{CDCl}_{3}$ $=77.23, \mathrm{CD}_{3} \mathrm{OD}=49.00$. Data are reported as follows: $(\mathrm{s}=$ singlet; $\mathrm{d}=$ doublet; $\mathrm{t}=$ triplet; $\mathrm{q}=$ quartet; sept $=$ septet; sext $=$ sextet; dd $=$ doublet of doublets; ddd $=$ doublet of doublet of doublets; $\mathrm{dt}=$ doublet of triplets; $\mathrm{td}=$ triplet of doublets; $\mathrm{dtd}=$ doublet of triplet of doublets; $\mathrm{br}=$ broad; app $=$ apparently). High resolution mass spectra were recorded on a MICROMASS AUTOSPEC (for EI) or WATERS Q-TOF API-US (for ESI) spectrometer. Infrared (IR) spectra were collected on a Mattson Cygnus 100 spectrometer. Samples for IR were prepared as a thin film on a NaCl plate by dissolving the compound in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and then evaporating the $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. Optical rotations were measured on a Perkin-Elmer 241 polarimeter at ambient temperature. Tetrahydrofuran and diethyl ether were distilled from sodium and benzophenone. Methylene
chloride and benzene was distilled under $\mathrm{N}_{2}$ from $\mathrm{CaH}_{2}$. Anhydrous $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was obtained through distillation from $\mathrm{CaH}_{2}$. $\mathrm{PhOCOCl},{ }^{i} \mathrm{PrCOCl}, \mathrm{MeOCH}_{2} \mathrm{COCl}, \mathrm{CbzCl}, \mathrm{PhSH}, \mathrm{Et}_{3} \mathrm{~N}$, and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ were distilled prior to use. MeOH and ${ }^{t} \mathrm{BuOH}$ were distilled from Mg and stored over $4 \AA$ molecular sieves prior to use. PhOH was azeotroped with toluene and dried under high vacuum before use. Methanesulfonic anhydride was purchased from Aldrich and used without further purification. Anhydrous $\mathrm{Mg}\left(\mathrm{ClO}_{4}\right)_{2}$ and $\mathrm{Zn}(\mathrm{OTf})_{2}$ were purchased from Aldrich and Fluka, respectively, stored in dessicator, and used as received. Analytical TLC was performed on E. Merck pre-coated ( 25 mm ) silica gel 60F-254 plates. Visualization was done under UV (254 $\mathrm{nm})$. Flash chromatography was done using ICN SiliTech 32-63 $60 \AA$ silica gel. Reagent grade ethyl acetate, diethyl ether and hexanes (commercial mixture) were purchased from EM Science and used as is for chromatography. All reactions were performed in oven or flame-dried glassware under Ar with magnetic stirring unless otherwise noted. Schwartz' reagent, though commercially available, was prepared according to the literature. ${ }^{112}$ All the compounds in this work were prepared in their racemic form unless otherwise noted.

Phenyl (1R,2R)-1,2-dimethoxyoctylcarbamate (3.2) and phenyl (1S,2R)-1,2-dimethoxyoctylcarbamate (3.3)



A solution of 2-methoxyoctanenitrile 3.1 ( $70.0 \mathrm{mg}, 0.451 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $\mathrm{mL})$ was treated with $\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(140 \mathrm{mg}, 0.541 \mathrm{mmol})$. The reaction was stirred for 15 min , then cooled to $0{ }^{\circ} \mathrm{C}$ and phenyl chloroformate ( $79 \mu \mathrm{~L}, 0.631 \mathrm{mmol}$ ) was added dropwise. The cold bath was removed and the mixture was stirred for 10 min . After that time, the flask was cooled to $0{ }^{\circ} \mathrm{C}$ and phenyl chloroformate ( $56 \mu \mathrm{~L}, 0.451 \mathrm{mmol}$ ) was added. The mixture was
stirred at room temperature for 15 min and then cooled to $0^{\circ} \mathrm{C}$. A solution of $\mathrm{MeOH}(0.36 \mathrm{ml}$, $9.02 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.6 \mathrm{~mL})$ was added dropwise. The reaction was stirred for 15 min at $0{ }^{\circ} \mathrm{C}$ and then quenched with saturated $\mathrm{NaHCO}_{3}(25 \mathrm{~mL})$. The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4$ x 20 mL ) and the combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $6 \%-120 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the desired product ( $77.2 \mathrm{mg}, 55.3 \%$ ) as a colorless oil in a 2.4:1.0 diastereomeric ratio. Further purification by column chromatography (8\%-14\% EtOAc in hexanes containing $0.5 \%$ $\mathrm{Et}_{3} \mathrm{~N}$ ) yielded analytically pure samples. For faster eluting anti-product 3.2 : ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) 7.38(\operatorname{app} \mathrm{t}, J=7.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.24-7.12(\mathrm{~m}, 3 \mathrm{H}), 5.90(\mathrm{~d}, J=9.8 \mathrm{~Hz}, 1 \mathrm{H}), 4.88(\mathrm{~d}, J=$ $10.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.59-3.49(\mathrm{~m}, 1 \mathrm{H}), 3.52(\mathrm{~s}, 3 \mathrm{H}), 3.44(\mathrm{~s}, 3 \mathrm{H}), 1.55-1.27(\mathrm{~m}, 10 \mathrm{H}), 0.90(\mathrm{t}, J=6.9$ $\mathrm{Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) 155.2, 151.0, 129.5, 125.6, 121.7, 85.6, 82.4, 59.7, 56.0, 31.9, 31.4, 29.5, 25.6, 22.8, 14.3; IR (neat) 3322, 2930, 2857, 1747, 1515, 1487, 1334, 1206, 1103, 1025, 952, 738; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{NO}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$332.1838, found 332.1830. For slower eluting syn-product 3.3: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.40-7.34 (m, 2H), 7.29-7.22 (m, 3H), $5.82(\mathrm{~d}, J=9.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.00(\mathrm{dd}, J=10.0,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.41(\mathrm{~s}, 3 \mathrm{H}), 3.40(\mathrm{~s}$, $3 \mathrm{H}), 3.18(\mathrm{dt}, J=6.8,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.62-1.55(\mathrm{~m}, 2 \mathrm{H}), 1.40-1.24(\mathrm{~m}, 8 \mathrm{H}), 0.90(\mathrm{t}, J=6.8 \mathrm{~Hz}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 154.9, 151.0, 129.5, 125.7, 121.7, 82.9, 82.6, 58.4, 56.5, 31.9, 29.7, 29.0, 25.6, 22.8, 14.3; IR (neat) 3324, 2928, 2857, 1747, 1523, 1488, 1356, 1209, 1086, 954; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{27} \mathrm{NO}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 332.1838$, found 332.1841.

## (E)-Ethyl non-2-enoate (3.7)


1.50-1.41 (m, 2H), 1.36-1.27 (m, 9H), $0.89(\mathrm{t}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $167.0,149.7,121.4,60.3,32.4,31.8,29.0,28.2,22.8,14.5,14.3$.

## (2S,3R)-Ethyl 2,3-dihydroxynonanoate (C1)

 A mixture of $\mathrm{AD}-\mathrm{mix}-\beta$ in ${ }^{t} \mathrm{BuOH} / \mathrm{H}_{2} \mathrm{O}(20 \mathrm{~mL}, 1: 1, \mathrm{v} / \mathrm{v})$ at $0{ }^{\circ} \mathrm{C}$ was treated with $\mathrm{CH}_{3} \mathrm{SO}_{2} \mathrm{NH}_{2}(0.190 \mathrm{~g}, 2.00 \mathrm{mmol})$ followed by a solution of enoate $3.7(0.368 \mathrm{~g}, 2.00 \mathrm{mmol})$ in ${ }^{t} \mathrm{BuOH}(0.5 \mathrm{~mL})$. The mixture was stirred at $0^{\circ} \mathrm{C}$ for 6 h and then at room temperature for 10 h . The reaction was cooled to $0^{\circ} \mathrm{C}$ and quenched with $\mathrm{Na}_{2} \mathrm{SO}_{3}$ solution $(10 \%, 30 \mathrm{~mL})$. After stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h , the mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{x}$ $30 \mathrm{~mL})$. The organic extracts were dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $30 \%-40 \%$ EtOAc in hexanes) to give the diol $\mathbf{C 1}(0.402 \mathrm{~g}, 92.0 \%)$ as a white solid: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 4.29(\mathrm{q}, J=7.1 \mathrm{~Hz}, 2 \mathrm{H}), 4.08(\mathrm{dd}, J=5.3,2.0 \mathrm{~Hz}$, $1 \mathrm{H}), 3.88(\mathrm{dtd}, J=8.9,6.9,2.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.12(\mathrm{~d}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.98(\mathrm{~d}, J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.64-$ $1.58(\mathrm{~m}, 2 \mathrm{H}), 1.52-1.44(\mathrm{~m}, 1 \mathrm{H}), 1.39-1.25(\mathrm{~m}, 10 \mathrm{H}), 0.89(\mathrm{t}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $(75$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $173.9,73.2,72.7,62.3,34.0,32.0,29.4,25.9,22.8,14.4,14.3$; IR (neat) 3377, 2925, 2854, 1737, 1462, 1294, 1136, 1099, 1072; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{11} \mathrm{H}_{22} \mathrm{O}_{4} \mathrm{Na}$ $[\mathrm{M}+\mathrm{Na}]^{+} 241.1416$, found 241.1420; $[\alpha]_{\mathrm{D}}=+12.6\left(\mathrm{CHCl}_{3}, c 0.98\right)$.

## (2S,3R)-Ethyl 2,3-dimethoxynonanoate (C2)



A solution of the diol $\mathbf{C 1}(170.0 \mathrm{mg}, 0.779 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4.0 \mathrm{~mL})$ were treated with $\mathrm{Ag}_{2} \mathrm{O}(271 \mathrm{mg}, 1.17 \mathrm{mmol})$ and $\mathrm{MeI}(0.22 \mathrm{~mL}, 3.50 \mathrm{mmol})$. The reaction was refluxed for 10 h , and $\mathrm{Ag}_{2} \mathrm{O}(271 \mathrm{mg}, 1.17 \mathrm{mmol})$ and $\mathrm{MeI}(0.22 \mathrm{~mL}, 3.50$ $\mathrm{mmol})$ were added sequentially. After $12 \mathrm{~h}, \mathrm{Ag}_{2} \mathrm{O}(271 \mathrm{mg}, 1.17 \mathrm{mmol})$ and $\mathrm{MeI}(0.22 \mathrm{~mL}, 3.50$
mmol ) were added. The mixture was refluxed for another 6 h , then filter through Celite and the residue was washed with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The combined filtrate was concentrated and the resulting residue was purified by column chromatography (5\%-15\% EtOAc in hexanes) to give the desired product $\mathbf{C} 2(59.4 \mathrm{mg}, 31.0 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 4.32$4.21(\mathrm{~m}, 2 \mathrm{H}), 3.78(\mathrm{~d}, J=4.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.51(\mathrm{dt}, J=6.5,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.44(\mathrm{~s}, 3 \mathrm{H}), 3.39(\mathrm{~s}, 3 \mathrm{H})$, $1.61-1.54(\mathrm{~m}, 2 \mathrm{H}), 1.34-1.26(\mathrm{~m}, 11 \mathrm{H}), 0.89(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $171.4,82.7,82.0,61.1,59.2,58.6,32.0,30.1,29.6,25.8,22.8,14.5,14.3$; IR (neat) 2927, 1747, 1464, 1261, 1190, 1143, 1105, 1031; HRMS (ESI): m/z calcd for $\mathrm{C}_{13} \mathrm{H}_{26} \mathrm{O}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 269.1729, found 269.1713; [ $\alpha]_{\mathrm{D}}=-29.7\left(\mathrm{CHCl}_{3}, c 0.63\right)$.

## (2S,3R)-2,3-Dimethoxynonanoic acid (3.8)



A solution of the ethyl ester C2 ( $40.0 \mathrm{mg}, 0.162 \mathrm{mmol}$ ) in $1,2-$ dimethoxyethane $/ \mathrm{H}_{2} \mathrm{O}(2.8 \mathrm{~mL}, 4: 1, \mathrm{v} / \mathrm{v})$ was treated with $\mathrm{LiOH} \cdot \mathrm{H}_{2} \mathrm{O}(13.6$ $\mathrm{mg}, 0.324 \mathrm{mmol})$. After 3 and $4 \mathrm{~h}, \mathrm{LiOH} \cdot \mathrm{H}_{2} \mathrm{O}(6.8 \mathrm{mg}, 0.162 \mathrm{mmol})$ was added, respectively. The reaction was stirred for another 3 h , then quenched with $\mathrm{HCl}(0.5 \mathrm{~N}$, $\sim 1.0 \mathrm{~mL})$ to $\mathrm{pH} \sim 1.5$ and extracted with $\mathrm{Et}_{2} \mathrm{O}(5 \mathrm{x} 10 \mathrm{~mL})$. The combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $30 \%$ EtOAc in hexanes followed by $50 \% \mathrm{MeOH}$ in EtOAc ) to give the unreacted ester ( 7.6 mg , $19.0 \%$ ) and carboxylic acid 3.8 ( $28.1 \mathrm{mg}, 79.4 \%$ ) as a white sticky solid: ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CD}_{3} \mathrm{OD}\right) 3.66(\mathrm{~d}, J=3.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.54(\mathrm{dt}, J=6.7,3.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.42(\mathrm{~s}, 3 \mathrm{H}), 3.41(\mathrm{~s}, 3 \mathrm{H}), 1.69-$ $1.57(\mathrm{~m}, 2 \mathrm{H}), 1.46-1.29(\mathrm{~m}, 8 \mathrm{H}), 0.91(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}){ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) 178.2, 84.9, 83.9, 59.4, 58.9, 32.9, 31.2, 30.5, 26.8, 23.7, 14.4; IR (neat) 3401, 2926, 2856, 1618, 1418,

1194, 1091; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{11} \mathrm{H}_{22} \mathrm{O}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$241.1416, found 241.1407; $[\alpha]_{\mathrm{D}}=$ -26.0 ( $\left.\mathrm{CH}_{3} \mathrm{OH}, c 0.77\right)$.

Phenyl (1S,2R)-1,2-dimethoxyoctylcarbamate ((-)-3.3)
( diphenyl phosphoryl azide ( $61 \mu \mathrm{~L}, 0.282 \mathrm{mmol}$ ). After 2 h , diphenylphosphoryl azide ( $30 \mu \mathrm{~L}$, 0.140 mmol ) was added. The reaction was stirred for 2 h , then quenched with water ( 10 mL ) and extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$. The organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $5 \%-15 \%$ EtOAc in hexanes) to give the carbamate (-)-3.3 (11.4 mg, 52.3\%) as a colorless oil: $[\alpha]_{\mathrm{D}}=-3.8\left(\mathrm{CHCl}_{3}, c \times 0.52\right)$. No other diastereomer was observed.

## 2-Ethoxyoctanenitrile (3.11)

$\mathrm{NC} \mathrm{H}_{4}$ A mixture of heptanal ( $4.00 \mathrm{~g}, 35.0 \mathrm{mmol}$ ), absolute EtOH ( 80 ml ), (EtO) ${ }_{3} \mathrm{CH}$ $(5.8 \mathrm{~mL}, 35.0 \mathrm{mmol})$ and the activated $4 \AA$ molecular sieves $(4.00 \mathrm{~g})$ at $0{ }^{\circ} \mathrm{C}$ was treated dropwise with concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}(2.0 \mathrm{ml})$ and the mixture was stirred at room temperature overnight. After that time, the reaction mixture was concentrated to $\sim 30 \mathrm{~mL}$ and slowly poured onto a cold saturated $\mathrm{NaHCO}_{3}$ solution $(80 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$. The resulting mixture was filtered through Celite. The filtrate was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 80 \mathrm{~mL})$ and the extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The resulting residue was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(70 \mathrm{~mL})$, and $\mathrm{BiBr}_{3}(1.57 \mathrm{~g}, 3.50 \mathrm{mmol})$ and $\mathrm{TMSCN}(5.60 \mathrm{ml}, 42.0 \mathrm{mmol})$ were added sequentially. The reaction was stirred overnight, then quenched with saturated $\mathrm{NaHCO}_{3}$ solution $(50 \mathrm{~mL}) /$ water
$(20 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 100 \mathrm{~mL})$. The combined extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $2 \%-5 \% \mathrm{EtOAc}$ in hexanes) to give the ethoxy nitrile 3.11 ( $4.51 \mathrm{~g}, 76.0 \%$ ) as a colorless liquid. ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 4.11(\mathrm{t}, J=6.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.82(\mathrm{qd}, J=8.8,6.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.51(\mathrm{qd}, J=8.9,7.0 \mathrm{~Hz}$, $1 \mathrm{H}), 1.87-1.80(\mathrm{~m}, 2 \mathrm{H}), 1.52-1.44(\mathrm{~m}, 2 \mathrm{H}), 1.38-1.30(\mathrm{~m}, 6 \mathrm{H}), 1.26(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.89(\mathrm{t}, J$ $=6.7 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 118.9,69.0,66.4,33.8,31.7,28.9,24.9,22.7,15.0$, 14.2; IR (neat) 2957, 2930, 2860, 1468, 1335, 1126, 1108, 735; HRMS (EI): m/z calcd for $\mathrm{C}_{10} \mathrm{H}_{19} \mathrm{NO}\left(\mathrm{M}^{+\bullet}\right)$ 169.1467, found 169.1474

Representative procedure for the preparation of acyl aminals:
$N$-((1R,2R)-2-Ethoxy-1-methoxyoctyl)isobutyramide (3.12) and $N$-((1S,2R)-2-ethoxy-1methoxyoctyl)isobutyramide (3.13)


A solution of ethoxynitrile 3.11 (100.0
$\mathrm{mg}, 0.591 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4.5 \mathrm{~mL})$ was treated with $\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(229 \mathrm{mg}$,
0.886 mmol ). The reaction was stirred for 15 min , then cooled to $0^{\circ} \mathrm{C}$ and isobutyryl chloride $(94 \mu \mathrm{~L}, 0.886 \mathrm{mmol})$ was added dropwise. The mixture was stirred for 15 min at $0{ }^{\circ} \mathrm{C}$ and $\mathrm{MeOH}(1.0 \mathrm{~mL}, 23.6 \mathrm{mmol})$ was added dropwise. The reaction was stirred for 15 min at $0^{\circ} \mathrm{C}$ and quenched with $\mathrm{AcOH}(2.0 \mathrm{~mL}) /$ water $(20 \mathrm{~mL})$. The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{x}$ $25 \mathrm{~mL})$ and the combined organic extracts were washed with saturated $\mathrm{NaHCO}_{3}(15 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $15 \%-30 \%$ EtOAc in hexanes) to give the desired product ( $121.3 \mathrm{mg}, 75.1 \%$ ) as a white solid in a $2.3: 1.0$ diastereomeric ratio. Further purification ( $15 \%-30 \%$ EtOAc in hexanes) yielded analytically
pure samples. For faster eluting anti-product 3.12: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 6.20(\mathrm{~d}, J=9.5$ $\mathrm{Hz}, 1 \mathrm{H}), 5.01(\mathrm{dd}, J=9.7,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{qd}, J=9.4,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.54(\mathrm{qd}, 9.4,7.1 \mathrm{~Hz}, 1 \mathrm{H})$, 3.47-3.43 (m, 1H), $3.29(\mathrm{~s}, 3 \mathrm{H}), 2.40(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.37-1.23(\mathrm{~m}, 10 \mathrm{H}), 1.18-1.14(\mathrm{~m}$, 9H), $0.84(\operatorname{app~t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 177.9,82.6,80.6,67.3,55.8,36.1$, $31.9,31.8,29.4,25.6,22.7,19.8,19.7,15.8,14.2$; IR (neat) $3271,2965,2920,1653,1540,1467$, 1233, 1113, 1101; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{28} \mathrm{NO}_{2}\left(\mathrm{M}-\mathrm{CH}_{3} \mathrm{O}\right)^{+\bullet} 242.2120$, found 242.2123 . For slower eluting syn-product 3.13: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 6.20(\mathrm{~d}, J=9.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.17$ (dd, $J=9.8,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.66(\mathrm{qd}, J=9.4,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.46(\mathrm{qd}, J=9.3,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.36(\mathrm{~s}$, $3 \mathrm{H}), 3.24(\mathrm{dt}, J=6.8,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.42(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.63-1.55(\mathrm{~m}, 2 \mathrm{H}), 1.40-1.25(\mathrm{~m}$, $10 \mathrm{H}), 1.23-1.17(\mathrm{~m}, 9 \mathrm{H}), 0.88(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 177.5, 81.0, 80.0, 66.1, 56.4, 36.2, 31.9, 29.8, 29.7, 25.6, 22.8, 19.9, 19.7, 15.8, 14.2; IR (neat) 3273, 2971, 2921, 1651, 1538, 1467, 1154, 1103, 1072; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{28} \mathrm{NO}_{2}\left(\mathrm{M}-\mathrm{CH}_{3} \mathrm{O}\right)^{+\bullet}$ 242.2120, found 242.2119 .
$N$-((1R,2R)-2-Ethoxy-1-methoxyoctyl)-2-methoxyacetamide (3.14) and $N$-((1S,2R)-2-ethoxy-1-methoxyoctyl)-2-methoxyacetamide (3.15)

representative procedure with the following amounts of reagents: ethoxynitrile $\mathbf{3 . 1 1}(100.0 \mathrm{mg}$, $0.591 \mathrm{mmol}), \mathrm{CH}_{2} \mathrm{Cl}_{2}(4.5 \mathrm{~mL}), \mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(168 \mathrm{mg}, 0.650 \mathrm{mmol})$, methoxyacetyl chloride ( 65 $\mu \mathrm{L}, 0.709 \mathrm{mmol}) \mathrm{MeOH}(1.0 \mathrm{ml}, 23.6 \mathrm{mmol})$. The reaction was quenched with $1 \mathrm{~N} \mathrm{HCl}(2.0$ $\mathrm{mL}) /$ water $(15 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 25 \mathrm{~mL})$. The combined organic extracts were washed with saturated $\mathrm{NaHCO}_{3}(15 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was
purified by column chromatography ( $20 \%-40 \%$ EtOAc in hexanes) to give the desired product $(111.8 \mathrm{mg}, 68.7 \%)$ as a colorless oil in a 1.7:1.0 diastereomeric ratio. Further purification ( $20 \%$ $40 \%$ EtOAc in hexanes) yielded analytically pure samples. For the faster eluting anti-product 3.14: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.18(\mathrm{~d}, J=9.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.06(\mathrm{dd}, J=10.0,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.98$ $(\mathrm{d}, J=15.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.90(\mathrm{~d}, J=15.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.76(\mathrm{qd}, J=9.4,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.56(\mathrm{qd}, J=9.3$, $7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.49-3.45(\mathrm{~m}, 1 \mathrm{H}), 3.43(\mathrm{~s}, 3 \mathrm{H}), 3.33(\mathrm{~s}, 3 \mathrm{H}), 1.45-1.25(\mathrm{~m}, 10 \mathrm{H}), 1.18(\mathrm{t}, J=7.0$ $\mathrm{Hz}, 3 \mathrm{H}), 0.86(\mathrm{t}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 170.8, 82.4, 80.5, 72.0, 67.3, 59.4, $56.1,31.9,31.8,29.4,25.7,22.8,15.8,14.2$; IR (neat) 3413, 2930, 2858, 1695, 1506, 1113; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{26} \mathrm{NO}_{3}\left(\mathrm{M}-\mathrm{CH}_{3} \mathrm{O}\right)^{+\bullet} 244.1913$, found 244.1925. For the slower eluting syn-product 3.15: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $7.22(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.18(\mathrm{dd}, J=$ $10.1,3.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.98(\mathrm{~d}, J=15.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.92(\mathrm{~d}, J=15.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.66(\mathrm{qd}, J=9.2,7.0 \mathrm{~Hz}$, $1 \mathrm{H}), 3.50(\mathrm{qd}, J=9.2,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.43(\mathrm{~s}, 3 \mathrm{H}), 3.36(\mathrm{~s}, 3 \mathrm{H}), 3.26(\mathrm{dt}, J=6.8,3.0 \mathrm{~Hz}, 1 \mathrm{H})$, $1.63-1.55(\mathrm{~m}, 2 \mathrm{H}), 1.41-1.29(\mathrm{~m}, 8 \mathrm{H}), 1.21(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.88(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 170.4, 81.0, 79.8, 72.0, 66.5, 59.4, 56.5, 31.9, 29.9, 29.6, 25.7, 22.8, 15.7, 14.3; IR (neat) 3417, 2928, 2858, 1686, 1510, 1112, 1078; HRMS (EI): m/z calcd for $\mathrm{C}_{13} \mathrm{H}_{26} \mathrm{NO}_{3}\left(\mathrm{M}-\mathrm{CH}_{3} \mathrm{O}\right)^{+\bullet}$ 244.1913, found 244.1917.

## Benzyl (1R,2R)-2-ethoxy-1-methoxyoctylcarbamate (3.16) and benzyl (1S,2R)-2-ethoxy-1methoxyoctylcarbamate (3.17)




The title compounds were prepared by following the representative procedure with the following amounts of reagents: ethoxynitrile $3.11(60.0 \mathrm{mg}, 0.354 \mathrm{mmol})$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3.5 \mathrm{~mL}), \mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(110.0 \mathrm{mg}, 0.425 \mathrm{mmol})$. After completion of hydrozirconation,
the reaction mixture was cooled to $0{ }^{\circ} \mathrm{C}$ and benzyl chloroformate ( $71 \mu \mathrm{~L}, 0.500 \mathrm{mmol}$ ) was added dropwise. The cold bath was removed and the mixture was stirred for 10 min . After that time, the flask was cooled to $0{ }^{\circ} \mathrm{C}$ and benzyl chloroformate ( $50 \mu \mathrm{~L}, 0.354 \mathrm{mmol}$ ) was added. The mixture was stirred at room temperature for 30 min and then cooled to $0^{\circ} \mathrm{C}$. A solution of $\mathrm{MeOH}(0.28 \mathrm{ml}, 7.08 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ was added dropwise. The reaction was stirred for 10 min at $0{ }^{\circ} \mathrm{C}$ and then quenched with saturated $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$. The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$ and the combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography (5\%-20\% EtOAc in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the desired product ( $76.3 \mathrm{mg}, 63.8 \%$ ) as a colorless oil in a 1.5:1.0 diastereomeric ratio. Further purification ( $10 \%-13 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) yielded analytically pure materials. For faster eluting anti-product 3.16: ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) 7.39-7.30(\mathrm{~m}, 5 \mathrm{H}), 5.66(\mathrm{~d}, J=9.8 \mathrm{~Hz}, 1 \mathrm{H}), 5.14(\mathrm{~s}, 2 \mathrm{H}), 4.82(\mathrm{dd}, J=9.9,1.0 \mathrm{~Hz}, 1 \mathrm{H})$, $3.74(\mathrm{qd}, J=9.3,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.56(\mathrm{qd}, J=9.2,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.48-3.44(\mathrm{~m}, 1 \mathrm{H}), 3.37(\mathrm{~s}, 3 \mathrm{H})$, 1.46-1.28(m, 10H), $1.17(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.89(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}(75 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 156.8,136.6,128.7,128.4,128.2,85.6,80.6,67.4,67.1,55.7,31.9,29.5,25.7,22.8$, 15.9, 14.3; IR (neat) 3337, 2929, 2858, 1731, 1497, 1456, 1326, 1216, 1107, 966, 735; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{31} \mathrm{NO}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 360.2151$, found 360.2148 . For slower eluting synproduct 3.17: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.40-7.31(\mathrm{~m}, 5 \mathrm{H}), 5.54(\mathrm{~d}, J=10.0 \mathrm{~Hz}, 1 \mathrm{H})$, $5.15 / 5.14$ (two s, 2 H ), $4.94(\mathrm{dd}, J=10.1,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.64(\mathrm{qd}, J=9.2,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.48(\mathrm{qd}, J$ $=9.2,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.38(\mathrm{~s}, 3 \mathrm{H}), 3.28(\mathrm{dt}, J=6.8,2.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.63-1.52(\mathrm{~m}, 2 \mathrm{H}), 1.41-1.26(\mathrm{~m}$, $8 \mathrm{H}), 1.20(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.89(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 156.6, 136.5, $128.8,128.5,128.4,82.9,81.1,77.4,67.2,66.2,56.3,31.9,29.7,29.6,25.7,22.8,15.8,14.3$; IR
(neat) 3334, 2928, 2858, 1729, 1501, 1455, 1232, 1097, 737; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{31} \mathrm{NO}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 360.2151$, found 360.2149 .

## $N$-(2-Ethoxy-1-methoxyoctyl)methanesulfonamide (3.18 and 3.19)

 the following amounts of reagents: ethoxynitrile 3.11 ( $100.0 \mathrm{mg}, 0.591 \mathrm{mmol}$ ), $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4.5 \mathrm{~mL})$, $\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(228 \mathrm{mg}, 0.886 \mathrm{mmol})$. After addition of methanesulfonic anhydride ( $144 \mathrm{mg}, 0.827$ $\mathrm{mmol})$, The mixture was stirred for 2 min at $0^{\circ} \mathrm{C}$ and $\mathrm{MeOH}(1.0 \mathrm{~mL}, 23.6 \mathrm{mmol})$ was added dropwise. The reaction was stirred for 10 min at $0{ }^{\circ} \mathrm{C}$ and quenched with saturated $\mathrm{NaHCO}_{3}(15$ $\mathrm{mL})$. The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{x} 20 \mathrm{~mL})$ and the combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography $(20 \%-30 \%$ EtOAc in hexanes) to give the desired product ( $40.8 \mathrm{mg}, 24.5 \%$ ) as a colorless oil in a 2.4:1.0 diastereomeric ratio: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 5.42(\mathrm{~d}, J=9.4 \mathrm{~Hz}, 71 \%$ of 1 H$), 5.22$ (d, $J=9.5 \mathrm{~Hz}, 29 \%$ of 1 H$), 4.64(\mathrm{dd}, J=9.5,3.1 \mathrm{~Hz}, 29 \%$ of 1 H$), 4.48(\mathrm{dd}, J=9.4,2.4 \mathrm{~Hz}, 71 \%$ of 1 H$), 3.72-3.52(\mathrm{~m}, 2 \mathrm{H}), 3.45(\mathrm{~s}, 29 \%$ of 3 H$), 3.46-3.42(\mathrm{~m}, 71 \%$ of 1 H$), 3.40(\mathrm{~s}, 71 \%$ of 3 H$)$, 3.32 (ddd, $J=7.2,5.6,3.1 \mathrm{~Hz}, 29 \%$ of 1 H ), $3.06(\mathrm{~s}, 29 \%$ of 3 H$), 3.05(\mathrm{~s}, 71 \%$ of 3 H$), 1.56-1.25$ $(\mathrm{m}, 10 \mathrm{H}), 1.21(\mathrm{t}, J=6.9 \mathrm{~Hz}, 29 \%$ of 3 H$), 1.18(\mathrm{t}, J=7.0 \mathrm{~Hz}, 71 \%$ of 3 H$), 0.88(\operatorname{app} \mathrm{t}, J=6.8$ $\mathrm{Hz}, 3 \mathrm{H}$ ) ; ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 88.1 (major), 86.0 (minor), 81.2 (minor), 79.5 (major), 67.2 (major), 66.4 (minor), 56.5, (minor), 55.7 (major), 43.3 (major), 43.2 (minor), 31.9, 31.6 (major), 29.6 (minor), 29.5 (major), 29.3 (minor), 25.8 (minor), 25.4 (major), 22.8, 15.8 (major), 15.7 (minor), 14.2; IR (neat) 3286, 2926, 2858, 1458, 1328, 1161, 1110, 978, 766; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{11} \mathrm{H}_{24} \mathrm{NO}_{3} \mathrm{~S}\left(\mathrm{M}-\mathrm{CH}_{3} \mathrm{O}\right)^{+\bullet} 250.1477$, found 250.1466 .
$N$-((1R,2R)-2-Ethoxy-1-tert-butoxyoctyl)isobutyramide (3.21) and $N$-((1S,2R)-2-ethoxy-1-tert-butoxyoctyl)isobutyramide (3.22)
 procedure with the following amounts of reagents: ethoxynitrile $3.11(60.0 \mathrm{mg}, 0.354 \mathrm{mmol})$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3.5 \mathrm{~mL}), \mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(110.0 \mathrm{mg}, 0.425 \mathrm{mmol})$. After addition of isobutyryl chloride ( 52 $\mu \mathrm{L}, 0.500 \mathrm{mmol}$ ), the cold bath was removed and the mixture was stirred for 10 min . After that time, the flask was cooled to $0^{\circ} \mathrm{C}$ and a solution of ${ }^{t} \mathrm{BuOH}(0.67 \mathrm{ml}, 7.08 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5$ mL ) was added dropwise to the reaction mixture over 3 min . The reaction was stirred for 10 min at $0{ }^{\circ} \mathrm{C}$, then diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ and quenched with saturated $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$. The organic layer was separated and the aqueous layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 15 \mathrm{~mL})$. The combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $10 \%-20 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the desired product ( $79.2 \mathrm{mg}, 70.8 \%$ ) as a white solid in a 1.0:2.0 diastereomeric ratio. Further purification ( $12 \%-18 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) yielded analytically pure samples. For faster eluting anti-product 3.21: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 6.10(\mathrm{~d}, J=9.2 \mathrm{~Hz}$, $1 \mathrm{H}), 5.32(\mathrm{dd}, J=9.4,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.84(\mathrm{qd}, J=9.6,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.58(\mathrm{qd}, J=9.6,7.0 \mathrm{~Hz}, 1 \mathrm{H})$, 3.30-3.26 (m, 1H), $2.33(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.38-1.27(\mathrm{~m}, 10 \mathrm{H}), 1.22(\mathrm{~s}, 9 \mathrm{H}), 1.19-1.12(\mathrm{~m}$, 9H), $0.87(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}),{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 175.8,83.0,75.5,74.6,67.6,36.1$, 32.0, 31.6, 29.5, 28.6, 25.9, 22.8, 19.7, 19.4, 15.9, 14.3; IR (neat) 3246, 2969, 2922, 2858, 1648, 1552, 1466, 1109, 1069; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{37} \mathrm{NO}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$338.2671, found 338.2663. For slower eluting syn-product 3.22: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 6.00(\mathrm{~d}, J=9.2 \mathrm{~Hz}$,
$1 \mathrm{H}), 5.39(\mathrm{dd}, J=9.4,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.64-3.54(\mathrm{~m}, 2 \mathrm{H}), 3.14-3.09(\mathrm{~m}, 1 \mathrm{H}), 2.30(\mathrm{sept}, J=6.9 \mathrm{~Hz}$, $1 \mathrm{H}), 1.51-1.38(\mathrm{~m}, 2 \mathrm{H}), 1.35-1.24(\mathrm{~m}, 8 \mathrm{H}), 1.20(\mathrm{~s}, 9 \mathrm{H}), 1.19-1.11(\mathrm{~m}, 9 \mathrm{H}), 0.87(\mathrm{t}, J=6.9 \mathrm{~Hz}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 175.3, 82.1, 74.9, 74.3, 66.8, 36.1, 32.0, 30.2, 29.6, 28.5, 26.0, $22.8,19.5,19.4,15.8,14.3$; IR (neat) $3254,2960,2920,2856,1646,1544,1459,1365,1193$, 1109, 1072, 731; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{37} \mathrm{NO}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 338.2671$, found 338.2666 .
$N$-((1R,2R)-2-Ethoxy-1-phenoxyoctyl)isobutyramide (3.23) and $N$-((1S,2R)-2-ethoxy-1phenoxyoctyl)isobutyramide (3.24)


The title compounds were prepared by following the representative procedure with the following amounts of reagents: ethoxynitrile $3.11(60.0 \mathrm{mg}, 0.354 \mathrm{mmol}), \mathrm{CH}_{2} \mathrm{Cl}_{2}(3.5 \mathrm{~mL}), \mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}$ ( $110.0 \mathrm{mg}, 0.425 \mathrm{mmol}$ ). After addition of isobutyryl chloride ( $52 \mu \mathrm{~L}, 0.500 \mathrm{mmol}$ ), the cold bath was removed and the mixture was stirred for 10 min . The mixture was cooled to ${ }^{\circ} \mathrm{C}$ and a solution of $\mathrm{PhOH}(333 \mathrm{mg}, 3.54 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ was added dropwise. The reaction was stirred at ${ }^{\circ} \mathrm{C}$ for 40 min , then quenched with saturated $\mathrm{NaHCO}_{3}$ solution ( 20 mL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \times 15 \mathrm{~mL})$. The organic extracts were washed with saturated $\mathrm{Na}_{2} \mathrm{CO}_{3}$ solution ( 20 mL ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography $\left(7 \%-10 \% \mathrm{EtOAc}\right.$ in hexanes containing $\left.0.5 \% \mathrm{Et}_{3} \mathrm{~N}\right)$ to give the desired product ( $81.7 \mathrm{mg}, 68.7 \%$ ) as a white solid in a 5.6:1.0 diastereomeric ratio. Further purification ( $7 \%$ $10 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) yielded analytically pure samples. For faster eluting anti-product 3.23: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 7.30-7.23(\mathrm{~m}, 2 \mathrm{H}), 7.03(\operatorname{app~td}, J=7.8$, $1.0 \mathrm{~Hz}, 2 \mathrm{H}), 6.96(\mathrm{app} \mathrm{tt}, J=7.3,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.36(\mathrm{~d}, J=9.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.92(\mathrm{dd}, J=9.9,1.4 \mathrm{~Hz}$,
$1 \mathrm{H}), 3.96(\mathrm{qd}, J=9.4,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.72(\mathrm{qd}, J=9.5,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.68-3.64(\mathrm{~m}, 1 \mathrm{H}), 2.37$ (sept, $J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.47-1.37(\mathrm{~m}, 4 \mathrm{H}), 1.33-1.24(\mathrm{~m}, 9 \mathrm{H}), 1.15(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 1.08(\mathrm{~d}, J=6.9$ $\mathrm{Hz}, 3 \mathrm{H}), 0.88(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 177.1,156.4,129.7,121.8,116.5$, 80.7, 80.0, 68.1, 36.0, 31.9, 29.4, 25.7, 22.8, 19.6, 19.5, 16.0, 14.2; IR (neat) 3290, 2964, 2929, 2859, 1657, 1595, 1534, 1495, 1222, 1107, 753; HRMS (ESI): m/z calcd for $\mathrm{C}_{20} \mathrm{H}_{33} \mathrm{NO}_{3} \mathrm{Na}$ $[\mathrm{M}+\mathrm{Na}]^{+} 358.2358$, found 358.2359. For slower eluting syn-product 3.24 (containing trace amount of unknown impurity): ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.30-7.24 (m, 2H), 7.07-7.04 (m, 2H), 6.97 (app tt, $J=7.3,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.24(\mathrm{~d}, J=9.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.02(\mathrm{dd}, J=9.8,3.4 \mathrm{~Hz}, 1 \mathrm{H})$, $3.72(\mathrm{qd}, J=9.4,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.62(\mathrm{qd}, J=9.4,7.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.44(\mathrm{dt}, J=7.0,3.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.36$ (sept, $J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.71-1.62(\mathrm{~m}, 2 \mathrm{H}), 1.47-1.21(\mathrm{~m}, 11 \mathrm{H}), 1.15(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 1.10(\mathrm{~d}$, $J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 0.87(\operatorname{app} \mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 176.5,156.8,129.7$, $121.9,116.2,81.2,77.8,66.8,36.0,31.9,30.1,29.6,25.7,22.8,19.6,19.5,15.9,14.3$; IR (neat) 3288, 2963, 2926, 2857, 1653, 1535, 1495, 1220, 1109, 1042, 752; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{20} \mathrm{H}_{33} \mathrm{NO}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 358.2358$, found 358.2328 .
$N-((1 R, 2 R)-2-E t h o x y-1-(p h e n y l t h i o) o c t y l) i s o b u t y r a m i d e ~(3.25) ~ a n d ~ N-((1 S, 2 R)-2-e t h o x y-1-~$ (phenylthio)octyl)isobutyramide (3.26)



The title compounds were prepared by following the representative procedure with the following amounts of reagents: ethoxynitrile $3.11(60.0 \mathrm{mg}, 0.354 \mathrm{mmol})$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3.5 \mathrm{~mL}), \mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(110.0 \mathrm{mg}, 0.425 \mathrm{mmol})$. After addition of isobutyryl chloride ( 52 $\mu \mathrm{L}, 0.500 \mathrm{mmol}$ ), the cold bath was removed and the mixture was stirred for 10 min . The mixture was cooled to ${ }^{\circ} \mathrm{C}$ and a solution of $\mathrm{PhSH}(117 \mathrm{mg}, 1.06 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.3 \mathrm{~mL})$ was
added dropwise. The reaction was stirred at ${ }^{\circ} \mathrm{C}$ for 10 min , then quenched with saturated $\mathrm{NaHCO}_{3}$ solution $(20 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \times 15 \mathrm{~mL})$. The organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography (7\% $13 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the desired product $(89.4 \mathrm{mg}, 71.7 \%)$ as a white solid in a 1.0:7.1 diastereomeric ratio. Further purification ( $10 \%-16 \%$ EtOAc in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) yielded analytically pure samples. For faster eluting anti-product $3.25:{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.49-7.45 (m, 2H), 7.31-7.20 (m, 3H), $6.00(\mathrm{~d}, J=9.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.58$ $(\mathrm{dd}, J=10.0,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.76-3.63(\mathrm{~m}, 2 \mathrm{H}), 3.56(\mathrm{dt}, J=6.5,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.29(\mathrm{sept}, J=6.9$ $\mathrm{Hz}, 1 \mathrm{H}), 1.60-1.48(\mathrm{~m}, 1 \mathrm{H}), 1.40-1.14(\mathrm{~m}, 12 \mathrm{H}), 1.08(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 1.00(\mathrm{~d}, J=6.9 \mathrm{~Hz}$, $3 \mathrm{H}), 0.87(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 176.3, 133.9, 132.3, 129.1, 127.5, 81.7, 67.3, 60.0, 35.9, 32.5, 31.9, 29.4, 25.7, 22.8, 19.7, 19.6, 15.9, 14.3; IR (neat) 3302, 2962, 2928, 2859, 1652, 1497, 1440, 1379, 1223, 1098, 739; HRMS (ESI): m/z calcd for $\mathrm{C}_{20} \mathrm{H}_{33} \mathrm{NO}_{2} \mathrm{SNa}[\mathrm{M}+\mathrm{Na}]^{+}$374.2130, found 374.2130. For slower eluting syn-product 3.26: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.47-7.44 (m, 2H), 7.30-7.18 (m, 3 H$), 6.00(\mathrm{~d}, J=9.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.63$ (dd, $J=9.7,3.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.69-3.49(\mathrm{~m}, 3 \mathrm{H}), 2.25(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.73-1.66(\mathrm{~m}, 2 \mathrm{H})$, $1.42-1.28(\mathrm{~m}, 8 \mathrm{H}), 1.21(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 1.06(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 0.98(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H})$, $0.88(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ 176.0, 133.6, 132.0, 129.1, 127.3, 82.0, $66.0,59.8,35.8,31.9,31.6,29.5,25.8,22.7,19.6,19.5,15.7,14.2$; IR (neat) $3293,2962,2927$, 2858, 1650, 1526, 1223,, 1100, 736; HRMS (ESI): m/z calcd for $\mathrm{C}_{20} \mathrm{H}_{33} \mathrm{NO}_{2} \mathrm{SNa}[\mathrm{M}+\mathrm{Na}]^{+}$ 374.2130, found 374.2115 .

## 1-Cyanoheptyl benzoate (3.27)



A solution of 2-hydroxyoctanenitrile $(0.600 \mathrm{~g}, 4.25 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(14 \mathrm{~mL})$ was treated with $\mathrm{Et}_{3} \mathrm{~N}(1.2 \mathrm{~mL}, 8.50 \mathrm{mmol})$, DMAP ( $\left.5.2 \mathrm{mg}, 42.5 \mu \mathrm{~mol}\right)$ and benzoyl chloride ( $0.60 \mathrm{~mL}, 5.10 \mathrm{mmol}$ ). The reaction was stirred for 1 h , then quenched with water $(30 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$. The organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography $\left(5 \%-10 \% \mathrm{Et}_{2} \mathrm{O}\right.$ in hexanes) to give the benzoate 3.27 ( $1.042 \mathrm{~g}, 94.1 \%$ ) as a colorless oil. For spectral data, see ref. 119.

## 1-(Isobutyramido)-1-methoxyoctan-2-yl benzoate (3.28)

 $\mathrm{mg}, 0.408 \mathrm{mmol}), \mathrm{CH}_{2} \mathrm{Cl}_{2}(4.0 \mathrm{~mL}), \mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(158 \mathrm{mg}, 0.612 \mathrm{mmol})$, isobutyryl chloride ( 52 $\mu \mathrm{L}, 0.490 \mathrm{mmol}), \mathrm{MeOH}(0.7 \mathrm{~mL}, 17.3 \mathrm{mmol})$. After the reaction was complete, it was quenched with $1 \mathrm{~N} \mathrm{HCl}(1.5 \mathrm{~mL})$ and water $(15 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 25 \mathrm{~mL})$. The organic extracts were washed with saturated $\mathrm{NaHCO}_{3}(15 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $15 \%-30 \%$ EtOAc in hexanes) to give the product $3.28(90.4 \mathrm{mg}, 63.5 \%$, containing $4 \% \mathrm{BnOH})$ as a white solid in a 1.4:1.0 diastereomeric ratio. Further purification ( $15 \%-30 \%$ EtOAc in hexanes) yielded analytically pure materials. For faster eluting product: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.03-8.01 (m, 2H), 7.60-7.56 (m, 1H), 7.48-7.44 (m, 2H), $5.97(\mathrm{~d}, J=9.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.24(\mathrm{dd}, J=9.6,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.19(\mathrm{ddd}, J=8.6$, $6.6,3.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.38(\mathrm{~s}, 3 \mathrm{H}), 2.32(\mathrm{sept}, J=7.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.86-1.80(\mathrm{~m}, 1 \mathrm{H}), 1.78-1.70(\mathrm{~m}, 1 \mathrm{H})$, $1.44-1.23(\mathrm{~m}, 7 \mathrm{H}), 1.21-1.15(\mathrm{~m}, 1 \mathrm{H}), 1.10(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 1.00(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 0.86(\mathrm{t}$,
$J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 178.0,167.0,133.5,130.0,129.9,128.7,81.9$, $74.5,56.3,36.1,31.8,31.3,29.3,25.3,22.8,19.6,19.5,14.2$; IR (neat) $3295,2959,2929,2858$, 1721, 1663, 1529, 1452, 1273, 1113, 712 ; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{20} \mathrm{H}_{31} \mathrm{NO}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$ 372.2151, found 372.2123. For slower eluting product: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.08 (app d, $J=7.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.59(\operatorname{appt} \mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.47(\operatorname{appt}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.01(\mathrm{~d}, J=9.6 \mathrm{~Hz}$, $1 \mathrm{H}), 5.33$ (dd, $J=9.8,4.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.12(\mathrm{td}, J=8.6,4.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.38(\mathrm{~s}, 3 \mathrm{H}), 2.40(\mathrm{sept}, J=6.9$ $\mathrm{Hz}, 1 \mathrm{H}), 1.82-1.73(\mathrm{~m}, 2 \mathrm{H}), 1.44-1.26(\mathrm{~m}, 8 \mathrm{H}), 1.18(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 1.17(\mathrm{~d}, J=7.0 \mathrm{~Hz}$, 3H), $0.87(\mathrm{t}, J=6.7 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 177.6,166.7,133.4,130.1,130.0$, $128.7,81.0,75.8,56.7,36.1,31.8,30.6,29.3,25.4,22.7,19.8,19.7,14.2$; IR (neat) 3299,2929 , 2858, 1722, 1661, 1527, 1453, 1273, 1113, 712; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{20} \mathrm{H}_{31} \mathrm{NO}_{4} \mathrm{Na}$ $[\mathrm{M}+\mathrm{Na}]^{+} 372.2151$, found 372.2133 .

## 1-(Isobutyramido)-1-hydroxyoctan-2-yl benzoate (3.29)



The title compound was prepared by following the representative procedure with the following amounts of reagents: benzoate 3.27 (100.0 $\mathrm{mg}, 0.408 \mathrm{mmol}), \mathrm{CH}_{2} \mathrm{Cl}_{2}(4.0 \mathrm{~mL}), \mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(158 \mathrm{mg}, 0.612 \mathrm{mmol})$, isobutyryl chloride ( $52 \mu \mathrm{~L}, 0.490 \mathrm{mmol}$ ). The reaction was quenched with water ( 15 mL ) and extraction of the mixture with EtOAc ( $3 \times 25 \mathrm{~mL}$ ). After evaporation of the solvent, the crude product was purified by column chromatography ( $20 \%-60 \%$ EtOAc in hexanes containing $0.5 \%$ $\left.\mathrm{Et}_{3} \mathrm{~N}\right)$ gave the product $3.29(71.6 \mathrm{mg}, 52.4 \%$, containing trace amount of impurity) as a colorless oil in a 3.0:1.0 diastereomeric ratio. ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 8.10-8.07 (m, 1.5H), 8.03-7.99 $(\mathrm{m}, 0.5 \mathrm{H}), 7.61-7.56(\mathrm{~m}, 1 \mathrm{H}), 7.48-7.42(\mathrm{~m}, 2 \mathrm{H}), 7.11(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 0.75 \mathrm{H}), 6.74(\mathrm{~d}, J=8.3 \mathrm{~Hz}$, $0.25 \mathrm{H}), 5.53-5.36(\mathrm{~m}, 1 \mathrm{H}), 5.22-5.17(\mathrm{~m}, 0.25 \mathrm{H}), 5.15-5.10(\mathrm{~m}, 0.75 \mathrm{H}), 4.74(\mathrm{br} \mathrm{s}, 0.25 \mathrm{H}), 4.52$
(br s, 0.75 H ), 2.47-2.22 (m, 1H), 1.92-1.78 (m, 2H), 1.44-1.20 (m, 8H), $1.13(\mathrm{~d}, J=6.8 \mathrm{~Hz}$, $2.25 \mathrm{H}), 1.11(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 2.25 \mathrm{H}), 1.04(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 0.75 \mathrm{H}), 0.99(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 0.75 \mathrm{H}), 0.85$ $(\mathrm{t}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (for major diastereomer) 178.4, 167.7, 133.6, $130.1,128.6,76.2,75.0,35.6,31.8,30.7,29.2,25.5,22.7,19.5,19.3,14.2$; IR (neat) 3338, 2959, 2928, 2858, 1720, 1657, 1530, 1451, 1274, 1119, 1070, 711; HRMS (EI): m/z calcd for $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{NO}_{3}(\mathrm{M}-\mathrm{OH})^{+\bullet} 318.2069$, found 318.2064.

## $N$-(1-Methoxynonyl)isobutyramide (3.31)



The title compound was prepared by following the representative procedure with the following amounts of reagents: octyl cyanide $\mathbf{3 . 3 0}$ $(84.0 \mathrm{mg}, 0.603 \mathrm{mmol})$, THF ( 6.0 mL ), $\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(194 \mathrm{mg}, 0.754 \mathrm{mmol})$. The hydrozirconation reaction was stirred for 30 min , then cooled to $0^{\circ} \mathrm{C}$ and isobutyryl chloride ( 95 $\mu \mathrm{L}, 0.904 \mathrm{mmol}$ ) was added dropwise. The reaction was stirred for 10 min at $0^{\circ} \mathrm{C}$ and MeOH $(0.73 \mathrm{~mL}, 18.1 \mathrm{mmol})$ was added dropwise. The reaction was stirred at $0^{\circ} \mathrm{C}$ for 15 min , then quenched with a solution of $\mathrm{Et}_{3} \mathrm{~N}(0.25 \mathrm{~mL})$ in water $(15 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \times 20$ $\mathrm{mL})$. The combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo. The residue was purified by column chromatography (15\%-25\% EtOAc in hexanes containing 0.5\% $\mathrm{Et}_{3} \mathrm{~N}$ ) to gave the title product ( $91.7 \mathrm{mg}, 62.3 \%$ ) as a white solid: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $5.66(\mathrm{~d}, J=9.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.10(\mathrm{td}, J=9.8,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.31(\mathrm{~s}, 3 \mathrm{H}), 2.37(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H})$, $1.66-1.59(\mathrm{~m}, 1 \mathrm{H}), 1.52-1.43(\mathrm{~m}, 1 \mathrm{H}), 1.38-1.24(\mathrm{~m}, 12 \mathrm{H}), 1.18(\operatorname{app} \mathrm{~d}, J=6.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.16$ (app d, $J=6.4 \mathrm{~Hz}, 3 \mathrm{H}), 0.86(\mathrm{t}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 177.4, 81.1, 55.9, $36.1,35.8,32.0,29.6,29.5,29.4,25.0,22.8,19.9,19.7,14.2$; IR (neat) $3281,2920,2853,1651$,

1538, 1466, 1377, 1236, 1081, 929, 720; HRMS (EI): m/z calcd for $\mathrm{C}_{13} \mathrm{H}_{26} \mathrm{NO}\left(\mathrm{M}-\mathrm{CH}_{3} \mathrm{O}\right)^{+\bullet}$ 212.2014 found 212.2010 .

## $N$-(1-Hydroxynonyl)isobutyramide (3.32)



The title compound was prepared by following the representative procedure with the following amounts of reagents: octyl cyanide $\mathbf{3 . 3 0}$ $(84.0 \mathrm{mg}, 0.603 \mathrm{mmol}), \mathrm{CH}_{2} \mathrm{Cl}_{2}(4.5 \mathrm{~mL}), \mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(171 \mathrm{mg}, 0.663 \mathrm{mmol})$. After hydrozirconation was complete, a solution of isobutyryl chloride ( $76 \mu \mathrm{~L}, 0.724 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}$ ( $0.25 \mathrm{~mL}, 1.81 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ was added dropwise at $0^{\circ} \mathrm{C}$. The reaction was stirred for 15 min at $0{ }^{\circ} \mathrm{C}$ and quenched with water $(20 \mathrm{~mL})$. The mixture was acidified by adding 1 N HCl to $\mathrm{pH} \sim 1.0$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 25 \mathrm{~mL})$. The combined organic extracts were washed with saturated $\mathrm{NaHCO}_{3}(20 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo. The residue was purified by column chromatography ( $10 \%$ - $70 \%$ EtOAc in hexanes) to give acyl hemiaminal 3.32 ( $74.7 \mathrm{mg}, 54.0 \%$ ) as white solids: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 6.17(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, $5.30(\mathrm{q}, J=6.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.27(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 2.35(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.73-1.61(\mathrm{~m}, 1 \mathrm{H}), 1.59-$ $1.48(\mathrm{~m}, 1 \mathrm{H}), 1.40-1.26(\mathrm{~m}, 12 \mathrm{H}), 1.15(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 1.14(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 0.87(\mathrm{t}, J=$ 6.9 Hz, 3H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 178.4, 74.5, 35.7, 35.3, 32.0, 29.6, 29.5, 29.4, 25.1, 22.8, 19.6, 19.4, 14.3; IR (neat) 3298, 2934, 2854, 1653, 1540, 1462, 1231, 1095; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{26} \mathrm{NO}(\mathrm{M}-\mathrm{OH})^{+\bullet} 212.2014$ found 212.2015.

## N -(Methoxy(phenyl)methyl)isobutyramide (3.34)



By following the representative procedure, reaction of benzonitrile $\mathbf{3 . 3 3}$ $(60.0 \mathrm{mg}, 0.582 \mathrm{mmol})$ with $\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(240 \mathrm{mg}, 0.931 \mathrm{mmol})$ in THF ( 5.8
mL ) for 2.5 h followed by acylation with isobutyryl chloride ( $92 \mu \mathrm{~L}, 0.873 \mathrm{mmol}$ ) and addition of $\mathrm{MeOH}(0.71 \mathrm{~mL}, 17.5 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ gave the title product $3.34(87.9 \mathrm{mg}$, $72.9 \%$ ) as a white solid: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $7.42-7.30(\mathrm{~m}, 5 \mathrm{H}), 6.14(\mathrm{~d}, J=9.4 \mathrm{~Hz}$, $1 \mathrm{H}), 6.02(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.45(\mathrm{~s}, 3 \mathrm{H}), 2.40(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.21(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H})$, $1.18(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) 177.4, 139.6, 128.8, 128.6, 126.0, 81.3, 56.1, 36.0, 19.8, 19.6; IR (neat) 3286, 2967, 1653, 1535, 1451, 1230, 1099, 1046, 951, 746; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{11} \mathrm{H}_{14} \mathrm{NO}_{2}\left(\mathrm{M}-\mathrm{CH}_{3} \mathrm{O}\right)^{+\bullet} 192.1024$, found 192.1031.

## $N$-((E)-Non-1-enyl)isobutyramide (3.36)



A solution of octyl cyanide 3.30 ( $84.0 \mathrm{mg}, 0.603 \mathrm{mmol}$ ) in THF ( 6.0 mL ) was treated with $\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(171 \mathrm{mg}, 0.663 \mathrm{mmol})$. The reaction was stirred for 20 min , then cooled to $0^{\circ} \mathrm{C}$ and a solution of isobutyryl chloride ( $60 \mu \mathrm{~L}, 0.573 \mathrm{mmol}$ ) and $\mathrm{Et}_{3} \mathrm{~N}(0.25 \mathrm{~mL}, 1.81 \mathrm{mmol})$ in THF $(4.0 \mathrm{~mL})$ was added dropwise. The flask formerly containing the isobutyryl chloride and $\mathrm{Et}_{3} \mathrm{~N}$ was rinsed with THF ( $2 \times 1 \mathrm{~mL}$ ). The reaction was stirred for 10 min at $0{ }^{\circ} \mathrm{C}$ and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(98 \mu \mathrm{~L}, 0.784 \mathrm{mmol})$ was added dropwise. The cold bath was removed and the mixture was stirred overnight. After that time, the reaction was quenched with water ( 30 mL ) and extracted with EtOAc ( $4 \times 30 \mathrm{~mL}$ ). The combined organic extracts were washed with water $(30 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo. The residue was purified by column chromatography ( $10 \%-20 \%$ EtOAc in hexanes) to gave the title product 3.36 (73.1 $\mathrm{mg}, 57.3 \%$ ) as a white solid: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $7.24(\mathrm{~d}, J=9.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.74$ (app dd, $J=14.2,10.5 \mathrm{~Hz}, 1 \mathrm{H}), 5.15(\mathrm{td}, J=14.2,7.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.37(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.00(\mathrm{q}, J=$ $6.6 \mathrm{~Hz}, 2 \mathrm{H}), 1.36-1.22(\mathrm{~m}, 10 \mathrm{H}), 1.17(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 6 \mathrm{H}), 0.87(\mathrm{t}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 75 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 174.2,122.7,113.3,35.7,32.0,30.1,29.9,29.3,29.2,22.8,19.6,14.3$; IR (neat)

3283, 2967, 2921, 2851, 1680, 1647, 1526, 1467, 1238, 950, 723; HRMS (EI): m/z calcd for $\mathrm{C}_{13} \mathrm{H}_{25} \mathrm{NO}\left(\mathrm{M}^{+\bullet}\right) 211.1936$, found 211.1938.

## $N$-(Isobutyryl)-N-((E)-non-1-enyl)isobutyramide (3.38)

${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 6.22(\mathrm{td}, J=13.9,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.47(\mathrm{td}, J=$ $14.2,7.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.21(\mathrm{sept}, J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.16(\mathrm{dq}, J=7.3,1.4 \mathrm{~Hz}$, 2H), 1.44 (pent, 2H), 1.34-1.26 (m, 8H), 1.17 (d, $J=6.8 \mathrm{~Hz}, 12 \mathrm{H}), 0.89$ $(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (125 MHz, $\left.\mathrm{CDCl}_{3}\right) 181.0,132.6,125.8,35.0,32.0,30.1,29.4$, 29.3, 29.1, 22.8, 19.7, 14.3; IR (neat) 2962, 2928, 1706, 1466, 1383, 1187, 1162, 1092; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{31} \mathrm{NO}_{2} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+} 304.2252$, found 304.2246.

## (E)-2-Methyltridec-2-en-1-ol (3.39)

A solution of 1-dodecene (freshly distilled, $0.842 \mathrm{~g}, 5.00 \mathrm{mmol}$ ) and methacrolein $(90 \%, 3.89 \mathrm{~g}, 50.0 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$ was treated with Grubbs' $2^{\text {nd }}$ generation catalyst ( $64.0 \mathrm{mg}, 75 \mu \mathrm{~mol}$ ). The reaction was refluxed for 1.5 h and then concentrated. The residue was purified by column chromatography ( $1 \%-5 \% \mathrm{EtOAc}$ in hexanes) to give the desired product ( 1.222 g , contaminated with unknown impurities). This product was dissolved in $\mathrm{MeOH}(29 \mathrm{~mL})$ and cooled to $0{ }^{\circ} \mathrm{C} . \mathrm{NaBH}_{4}(219 \mathrm{mg}, 5.80 \mathrm{mmol})$ was added. The reaction was stirred for 30 min and quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution $(1 \mathrm{~mL})$. The mixture was stirred for 10 min while warming to room temperature and then concentrated. The residue was purified by column chromatography ( $10 \%-20 \%$ EtOAc in hexanes) to give the allylic alcohol $3.39(0.621 \mathrm{~g}, 58.5 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $5.41(\mathrm{app} \mathrm{t}, J=7.1$ Hz, 1H), $4.00(\mathrm{~s}, 2 \mathrm{H}), 2.06-1.99(\mathrm{~m}, 2 \mathrm{H}), 1.66(\mathrm{~s}, 3 \mathrm{H}), 1.45-1.27(\mathrm{~m}, 16 \mathrm{H}), 0.88(\mathrm{t}, J=7.0 \mathrm{~Hz}$,

3H); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) 134.7, 126.9, 69.3, 32.1, 29.9, 29.8, 29.7, 29.6, 27.8, 22.9, 14.3, 13.8; IR (neat) 3335, 2924, 2854, 1464, 1378, 1012; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{28} \mathrm{O}$ $\left(\mathrm{M}^{+\bullet}\right) 212.2140$, found 212.2150 .

## (E)-3-Methyltetradec-3-enenitrile (3.40)

A solution of allylic alcohol $3.39(505 \mathrm{mg}, 2.38 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(0.66 \mathrm{~mL}$,

$4.78 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(12 \mathrm{~mL}) /$ THF $(6 \mathrm{~mL})$ was cooled to $-42{ }^{\circ} \mathrm{C}$ and methanesulfonyl chloride $(0.24 \mathrm{~mL}, 3.09 \mathrm{mmol})$ was added dropwise. The mixture was stirred for 30 min and anhydrous $\mathrm{LiBr}(620 \mathrm{mg}, 7.14 \mathrm{mmol})$ was added followed by THF ( 18 mL ). The mixture was warmed to $0^{\circ} \mathrm{C}$ and stirred for 1.5 h . After that time, the reaction was diluted with hexanes $(150 \mathrm{~mL})$, washed with water $(80 \mathrm{~mL})$ and brine $(50 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The crude allylic bromide was dissolved in DMF ( 4.5 mL ) and CuCN ( 213 mg , 2.38 mmol ) was added in one portion. The reaction was stirred overnight, then quenched with water $(30 \mathrm{~mL})$ and extracted with EtOAc $(3 \times 25 \mathrm{~mL})$. The organic extract was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The resulting residue was purified by column chromatography ( $3 \%-5 \%$ EtOAc in hexanes) to give allylic nitrile 3.40 ( $295 \mathrm{mg}, 56.0 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 5.49(\mathrm{sext}$ of $\mathrm{t}, J=7.2,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 3.03(\mathrm{~s}, 2 \mathrm{H}), 2.04(\mathrm{q}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.73(\mathrm{~s}$, $3 \mathrm{H}), 1.38-1.27(\mathrm{~m}, 16 \mathrm{H}), 0.89(\mathrm{t}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 130.2, 124.0, $118.0,32.0,29.8,29.6,29.5,29.4,29.3,28.2,27.4,22.8,16.1,14.2$; IR (neat) 2925, 2854, 2249, 1464, 1412, 1114, 721; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{27} \mathrm{~N}\left(\mathrm{M}^{+\bullet}\right)$ 221.2144, found 221.2152.

## $N$-((1E,3E)-3-Methyltetradeca-1,3-dienyl)isobutyramide (3.41)



A solution of allylic nitrile $3.40(90.0 \mathrm{mg}, 0.406 \mathrm{mmol})$ in THF ( 5.0 mL ) was treated with $\mathrm{Cp}_{2} \mathrm{Zr}(\mathrm{H}) \mathrm{Cl}(157 \mathrm{mg}, 0.609 \mathrm{mmol})$, The reaction was stirred for 30 min , then cooled to $0^{\circ} \mathrm{C}$ and a solution of isobutyryl chloride ( $51 \mu \mathrm{~L}$, $0.487 \mathrm{mmol})$ and $\mathrm{Et}_{3} \mathrm{~N}(0.18 \mathrm{~mL}, 1.26 \mathrm{mmol})$ in THF ( 2.0 mL ) was added dropwise. The flask formerly containing the isobutyryl chloride and $\mathrm{Et}_{3} \mathrm{~N}$ was rinsed with THF ( 0.5 mL ). The reaction was stirred for 2 min at $0{ }^{\circ} \mathrm{C}$ and $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}(76 \mu \mathrm{~L}, 0.609 \mathrm{mmol})$ was added dropwise. The cold bath was removed and the mixture was stirred for 2 h . After that time, the reaction was diluted with $\mathrm{Et}_{2} \mathrm{O}(3 \mathrm{~mL})$ and filtered through a small plug of silica gel. The residue was washed with $\mathrm{Et}_{2} \mathrm{O}(30 \mathrm{~mL})$ and the combined filtrate was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo. The residue was purified by column chromatography (5\%-17\% EtOAc in hexanes containing $0.5 \%$ $\mathrm{Et}_{3} \mathrm{~N}$ ) to gave the title product $3.41(74.1 \mathrm{mg}, 62.1 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) 7.10(\mathrm{~d}, J=10.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.91(\mathrm{dd}, J=14.2,10.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.83(\mathrm{~d}, J=14.3 \mathrm{~Hz}, 1 \mathrm{H})$, $5.34(\mathrm{t}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.40(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.10(\mathrm{q}, J=7.0 \mathrm{~Hz}, 2 \mathrm{H}), 1.75(\mathrm{~s}, 3 \mathrm{H}), 1.40-$ $1.27(\mathrm{~m}, 16 \mathrm{H}), 1.20(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 6 \mathrm{H}), 0.88(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $174.3,131.8,130.8,120.4,118.5,35.9,32.1,30.0,29.9,29.8,29.6,28.4,22.9,19.7,14.3,12.7 ;$ IR (neat) 3276, 2924, 2854, 1644, 1531, 1467, 1253, 950; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{35} \mathrm{NO}$ $\left(\mathrm{M}^{+\bullet}\right)$ 293.2719, found 293.2717.

## (2S,4R,6R)-Tetrahydro-4-methoxy-5,5,6-trimethyl-2H-pyran-2-carbonitrile (3.43)

 $4.5,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 1.84(\mathrm{ddd}, J=13.4,11.8,6.1 \mathrm{~Hz}, 1 \mathrm{H}), 1.12(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 3 \mathrm{H}), 0.96(\mathrm{~s}, 3 \mathrm{H})$,
0.83 (s, 3H); ${ }^{13} \mathrm{C}$ NMR (75 MHz, $\mathrm{CDCl}_{3}$ ) 118.0, 81.4, 78.0, 64.2, 57.8, 39.4, 29.3, 22.6, 14.6, 12.2; IR (neat) 2980, 2941, 2874, 1470, 1450, 1391, 1164, 1104, 954, 867, 718; HRMS (EI): $m / z$ calcd for $\mathrm{C}_{10} \mathrm{H}_{17} \mathrm{NO}_{2}\left(\mathrm{M}^{+\bullet}\right)$ 183.1259, found 183.1255.
$N-((S)-((2 S, 4 R, 6 R)-T e t r a h y d r o-4-m e t h o x y-5,5,6-t r i m e t h y l-2 H-p y r a n-2-$
yl)(methoxy)methyl)isobutyramide (3.44), $N-((R)-((2 S, 4 R, 6 R)-T e t r a h y d r o-4-m e t h o x y-5,5,6-$ trimethyl-2H-pyran-2-yl)(methoxy)methyl)isobutyramide (3.45) and $N$-(((2S,4R,6R)-Tetrahydro-4-methoxy-5,5,6-trimethyl-2H-pyran-2-yl)methyl)isobutyramide (3.46)

tetrahydropyranyl cyanide $3.43(50.0 \mathrm{mg}, 0.273 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2.7 \mathrm{~mL})$ was treated with Schwartz reagent ( $84.5 \mathrm{mg}, 0.328 \mathrm{mmol}$ ). The mixture was stirred for 15 min , then cooled to 0 ${ }^{\circ} \mathrm{C}$ and isobutyryl chloride ( $40 \mu \mathrm{~L}, 0.382 \mathrm{mmol}$ ) was added dropwise. The cold bath was removed and the mixture was stirred for 10 min . After that time, the flask was cooled to $-78{ }^{\circ} \mathrm{C}$ and $\mathrm{Mg}\left(\mathrm{ClO}_{4}\right)_{2}(61 \mathrm{mg}, 0.273 \mathrm{mmol})$ was added in one portion. After 30 min , a pre-cooled solution $\left(-78{ }^{\circ} \mathrm{C}\right)$ of $\mathrm{MeOH}(0.22 \mathrm{ml}, 5.46 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ was cannulated dropwise to the reaction mixture over 5 min . After completion of addition, the reaction was stirred at -78 ${ }^{\circ} \mathrm{C}$ for 15 min , then quenched with saturated $\mathrm{NaHCO}_{3}$ solution $(15 \mathrm{~mL})$ and warmed to room temperature. The biphasic mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$ and the combined organic extracts were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography ( $20 \%-70 \% \mathrm{EtOAc}$ in hexanes containing $0.5 \% \mathrm{Et}_{3} \mathrm{~N}$ ) to give the desired products 3.44 and $3.45(60.2 \mathrm{mg}, 76.8 \%)$ in a 2.3:1.0 diastereomeric ratio as a colorless oil and
the over-reduction product $\mathbf{3 . 4 6}(6.8 \mathrm{mg}, 9.7 \%)$ as a colorless oil. For 3.46: ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{CDCl}_{3}\right) 5.77(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 4.05-3.97(\mathrm{~m}, 1 \mathrm{H}), 3.48-3.39(\mathrm{~m}, 3 \mathrm{H}), 3.32(\mathrm{~s}, 3 \mathrm{H}), 3.03(\mathrm{t}, J=6.4 \mathrm{~Hz}$, $1 \mathrm{H}), 2.38(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.74-1.69(\mathrm{~m}, 2 \mathrm{H}), 1.18-1.14(\mathrm{~m}, 9 \mathrm{H}), 0.96(\mathrm{~s}, 3 \mathrm{H}), 0.88(\mathrm{~s}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 177.3, 82.0, 74.5, 69.0, 57.7, 41.0, 38.6, 35.9, 27.6, 24.5, 19.9, $19.8,15.6,15.5$; IR (neat) $3305,2970,2933,2874,1651,1548,1468,1386,1243,1103$; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{27} \mathrm{NO}_{3} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$280.1889, found 280.1899. Further purification $\left(20 \%-40 \% \mathrm{EtOAc}\right.$ in hexanes containing $\left.0.5 \% \mathrm{Et}_{3} \mathrm{~N}\right)$ of the mixture of 3.44 and 3.45 yielded analytically pure diastereomers. For the faster eluting product 3.44 (major, white solid): ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $6.01(\mathrm{~d}, J=9.0 \mathrm{~Hz}, 1 \mathrm{H}), 5.26(\mathrm{dd}, J=9.5,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.84-3.80(\mathrm{~m}, 1 \mathrm{H})$, $3.39(\mathrm{~s}, 3 \mathrm{H}), 3.36(\mathrm{q}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.33(\mathrm{~s}, 3 \mathrm{H}), 3.04(\mathrm{dd}, J=8.6,4.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.44(\mathrm{sept}, J=$ $6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.92(\mathrm{td}, J=13.7,4.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.66(\mathrm{ddd}, J=13.8,8.6,5.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.20(\mathrm{~d}, J=6.9$ $\mathrm{Hz}, 3 \mathrm{H}), 1.19(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}), 1.10(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 0.94(\mathrm{~s}, 3 \mathrm{H}), 0.86(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (125 MHz, $\mathrm{CDCl}_{3}$ ) 178.2, 82.0, 80.0, 76.2, 70.9, 57.7, 56.4, 38.2, 36.2, 26.0, 24.5, 19.9, 19.8, 16.1, 15.3; IR (neat) 3300, 2972, 2938, 1659, 1536, 1468, 1387, 1103; HRMS (ESI): $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{15} \mathrm{H}_{29} \mathrm{NO}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$310.1994, found 310.1985. For the slower eluting product 3.45 (minor, colorless oil): ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 6.26(\mathrm{~d}, J=9.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.16(\mathrm{dd}, J=9.6$, $3.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.83-3.78(\mathrm{~m}, 1 \mathrm{H}), 3.67(\mathrm{q}, J=6.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.38(\mathrm{~s}, 3 \mathrm{H}), 3.31(\mathrm{~s}, 3 \mathrm{H}), 3.18(\mathrm{dd}, J=$ $7.0,3.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.42(\mathrm{sept}, J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.96(\mathrm{ddd}, J=13.7,6.6,3.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.71-1.63(\mathrm{~m}$, $1 \mathrm{H}), 1.22-1.17(\mathrm{~m}, 9 \mathrm{H}), 1.00(\mathrm{~s}, 3 \mathrm{H}), 0.88(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 177.5, 82.4, 81.7, $70.3,57.8,56.7,38.0,36.1,26.3,25.4,19.9,19.7,17.6,15.6$; IR (neat) 3293, 2970, 2934, 1658, 1531, 1468, 1387, 1170, 1102; HRMS (ESI): $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{29} \mathrm{NO}_{4} \mathrm{Na}[\mathrm{M}+\mathrm{Na}]^{+}$310.1994, found 310.2002.



















sulfonamide








































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