# STRAIN-ENABLED PHOSPHINATION AND FLUORINATION REACTIONS 

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Submitted to the Graduate Faculty of
The Kenneth P. Dietrich School of Arts and Science in partial fulfillment of the requirements for the degree of Doctor of Philosophy

## UNIVERSITY OF PITTSBURGH

## THE KENNETH P. DIETRICH SCHOOL OF ARTS AND SCIENCES

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# STRAIN-ENABLED PHOSPHINATION AND FLUORINATION REACTIONS 

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Chapter one of this thesis discusses the hydrophosphination of carbodiimides with secondary phosphine boranes. These reactions provide access to phosphaguanidine boranes, which have previously not been studied. A cyclic carbodiimide was also prepared, and hydrophosphination of this substrate provided access to structurally unique phosphaguanidines that bear resemblance to cyclophanes.

Chapter two delineates the extension of this hydrophosphination method to functionalize strained carbocycles, namely bicyclo[1.1.0]butanes. The scope and stereoselectivity of this transformation was investigated, and further transformations of the cyclobutyl phosphine products were conducted. Studies on the addition of aromatic and heteroaromatic thiols to bicyclo[1.1.0]butanes are also described in this chapter.

Chapter three describes a strain-relieving deoxyfluorination of bicyclo[1.1.0]butyl alcohols. The products of these transformations are fluorinated methylenecyclobutanes, which can be oxidatively transformed into cyclobutanones. Bicyclo[1.1.0]butyl amides were also subjected to fluorination with a high degree of diastereoselectivity.

Chapter four details the development of bicyclo[1.1.0]butyl amides as potential androgen receptor antagonists for the treatment of castration-resistant prostate cancer. In the course of these medicinal chemistry studies, an acid-mediated isomerization of bicyclo[1.1.0]butanes to cyclobutenes was discovered.

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## LIST OF ABBREVIATIONS

| A.................. angstrom ( $10^{-10}$ meter) |
| :---: |
| Ac................ acetyl |
| AR ................ androgen receptor |
| ATR.............. attenuated total reflectance |
| aq.................. aqueous |
| BHT.............. butylated hydroxytoluene |
| BINAP........... (2,2'-bis(diphenylphosphino)-1,1'-binaphthyl |
| BINOL ........... 1,1'-bi-2-naphthol |
| BIPI............... Boehringer Ingelheim Phosphine-Imidazole |
| BPE ............... 1,2-bis((2,5)-diphenylphospholano)ethane |
| Boc............... $t$-butyloxycarbonyl |
| Bn................. benzyl |
| Bz................. benzoyl |
| calcd .............. calculated |
| CCDC............ Cambridge Crystallographic Data Centre |
| centrimide ...... cetyltrimethyl ammonium bromide |
| COD .............. 1,5-cyclooctadiene |
| CRPC ............ castration-resistant prostate cancer |
| CSA.............. 10-camphorsulfonic acid |
| Cy................. cyclohexyl |
| d................... days |
| DAST ............. diethylaminosulfur trifluoride |
| DABCO.......... 1,4-diazabicyclo[2.2.2]octane |

Deoxo-Fluor ${ }^{\circledR}$. bis(2-methoxyethyl)aminosulfur trifluoride
DIBAL ........... diisobutylaluminum hydride
DBU ............... 1,8-diazabicyclo[5.4.0]undec-7-ene
DCC $\qquad$ dicyclohexylcarbodiimide

DCE 1,2-dichloroethane

DCE-d 4 ........... deuterated 1,2-dichloroethane
DEPT $\qquad$ distortionless enhancement by polarization transfer (NMR method)

DHP. 3,4-dihydro-2H-pyran

DMAc $\qquad$ dimethylacetamide

DMAP ............ 4-dimethylaminopyridine
DMF $\qquad$ dimethylformamide

DMP $\qquad$ Dess-Martin periodinane

DMPU $\qquad$ 1,3-dimethyl-3,4,5,6-tetrahydro-2(1H)-pyrimidinone

DMSO $\qquad$ dimethylsulfoxide
dppe $\qquad$ 1,2-bis(diphenylphosphino)ethane
ee $\qquad$ enantiomeric excess
eq. $\qquad$ equivalents
$\mathrm{EC}_{50}$ $\qquad$ effective concentration for $50 \%$ inhibition

EtOAc $\qquad$ ethyl acetate

ESI. $\qquad$ electrospray ionization

FDA. $\qquad$ United States Food and Drug Administration

Fluolead.......... 4-tert-butyl-2,6-dimethylphenylsulfur trifluoride
HFIP. $\qquad$ hexafluoroisopropanol

HMBC $\qquad$ heteronuclear multiple bond correlation (NMR method)

HPLC $\qquad$ high performance liquid chromatography

| HRMS ........... high resolution mass spectrometry |
| :---: |
| HSQC ............ heteronuclear single quantum coherence spectroscopy (NMR method) |
| Hz .................. hertz |
| IR.................. infrared |
| $\mathrm{k}_{\text {rel }}$................. relative rate |
| LAH .............. lithium aluminum hydride |
| M ................... molar (mol/L, concentration unit) |
| $m$-CPBA ........ meta-chloroperbenzoic acid |
| Me ................ methyl |
| min ................ minutes |
| mp ................. melting point |
| Ms ................. mesyl; methanesulfonyl |
| MS................ mass spectrometry |
| NBS............... $N$-bromosuccinimide |
| NfF ................ nonafluorobutyl sulfonyl fluoride |
| NFSI.............. $N$-fluorobenzenesulfonimide |
| NMO ............. $N$-methylmorpholine- $N$-oxide |
| NMP .............. $N$-methyl-2-pyrrolidone |
| NMR ............. nuclear magnetic resonance |
| NOESY .......... nuclear Overhauser effect spectroscopy (NMR method) |
| NTTL ............N-1,2-naphthaloyl-(S)-tert-leucine |
| PCC............... pyridinium chlorochromate |
| PPTS ............. pyridinium $p$-toluenesulfonate |
| Ph ................. phenyl |
| PyFluor.......... 2-pyridinesulfonylfluoride |

$\mathrm{R}_{\mathrm{f} . . . . . . . . . . . . . . . . . . . . ~ r e t a r d a t i o n ~ f a c t o r ~}^{\text {r }}$
rt $\qquad$
RSM $\qquad$ recovered starting material
$s$-Bu $\qquad$ secondary butyl

SFC $\qquad$ supercritical fluid chromatography

Skewphos ....... 2,4-bis(diphenylphosphino)pentane
$t_{1 / 2}$ half life

T3P $\qquad$ 2,4,6-tripropyl-1,3,5,2,4,6-trixatriphosphorinane-2,4,6-trioxide
$t$-Bu $\qquad$ tertiary butyl

THF $\qquad$ tetrahydrofuran

TLC. $\qquad$ thin layer chromatography

Ts. $\qquad$ p-toluenesulfonyl

UV $\qquad$ ultraviolet

## ACKNOWLEDGEMENTS

First and foremost, I thank Dr. Peter Wipf for providing me with the opportunity, resources, and guidance to learn organic chemistry at a very high level (dating back to my days as a visiting summer undergraduate). I also acknowledge Dr. Kay Brummond and Dr. Craig Wilcox, both of whom have continued to be supportive members of my thesis committee despite taking positions of leadership at the University during my tenure. Dr. W. Seth Horne is acknowledged for his helpful mentorship in the development of my departmental proposal document.

I acknowledge our collaborators in the Chemical Development department BoehringerIngelheim (BI), particularly Dr. Carl Busacca for both serving on my thesis committee and graciously welcoming me into his lab for the summer of 2015. Both he and Dr. Chris Senanayake have been engaging collaborators who have helped to broaden my knowledge of chemistry. I also acknowledge my fellow BI co-interns from the summer of 2015 for the good camaraderie that we shared.

In the Wipf group, the opportunity to work with many peers from all over the world has been a highly enriching experience. I am especially grateful for the friendship and scientific guidance that I received from Mike Frasso, Dr. Joe Salamoun, Dr. Gilmar Brito, Dr. Raffaele Colombo, Dr. Lalith Samankumara, Mike Kerner, and Leila Terrab. The administrative support provided by Desirae Crocker, Mary Liang, Taber Maskrey, and numerous other staff mebers is greatfully acknowledged. Mike Frasso, Mike Kerner, Leila Terrab, and Evan Carder are also acknowledged for reading sections of my thesis draft.

I had the pleasure of working collaboratively during the early stages of the phosphaguanidine research described in Chapter 1 (with Dr. Eakkaphon Rattanangkool) and in the
development of the androgen receptor antagonists described in Chapter 4 (with Dr. James Johnson, Dr. Keita Takubo, Serene Tai, Taber Maskrey, and numerous other collaborators from outside the group). I also acknowledge the NMR, MS, and X-ray crystallography facility directors (Dr. Damodaran Achary, Dr. Bhaskar Godugu, and Dr. Steve Geib, respectively) for maintaining these essential resources for the research described herein.

Finally, I am very grateful for the many individuals who have been extraordinarily supportive throughout my graduate studies, including my friends (both within and beyond the department), parents, parents-in-law, and extended family. I especially grateful for my wife Michelle, who has been unwavering source of support and encouragement.

### 1.0 HYDROPHOSPHINATION OF CARBODIIMIDES WITH PHOSPHINE BORANES

### 1.1 INTRODUCTION

### 1.1.1 Hydrophosphination of Carbodiimides

The hydrophosphination of carbon-carbon and carbon-heteroatom multiple bonds is a highly atom economical means to prepare substituted phosphines. The earliest hydrophosphinations, which were reported in the 1950's, involved reacting alkenes with primary phosphines under harsh conditions. ${ }^{1}$ An array of catalytic methods for alkene and alkyne hydrophosphination have since been developed, including methods that are mediated by bases ${ }^{2}$, organocatalysts, ${ }^{3}$ radical initiators, ${ }^{4}$ or Lewis acids. ${ }^{5}$ Hydrophosphinations have also been catalyzed by late transition metals ${ }^{6}$ or lanthanides. ${ }^{7}$ More recently, overarching efforts to develop more sustainable catalytic methods have produced hydrophosphination catalysts derived from coinage ${ }^{8}$ and alkali earth ${ }^{9}$ metals. The development of catalytic, asymmetric hydrophosphinations of hydrocarbons is also an area of ongoing research. ${ }^{10}$

Although carbodiimides are most prominently known as peptide coupling agents, ${ }^{11}$ they too can been subjected to hydrophosphination. Early protocols for these transformations involved harsh conditions such as heating a neat mixture of a secondary phosphine with a carbodiimide. ${ }^{12}$ In modern practice, various methods using base, transition metal, or lanthanide catalysts are available for the hydrophosphination of carbodiimides. ${ }^{13}$ Deprotonated secondary phosphines can be used to formally hydrophosphinate carbodiimides through attack of the phosphide anion onto
the carbodiimide followed by aqueous workup. ${ }^{14}$ The hydrophosphination of other electrophiles such as isocyanates, isothiocyanates, and aldehydes has also been investigated (Figure 1). ${ }^{15}$


Figure 1. General overview of hydrophosphination chemistry.

### 1.1.2 Hydrophosphination with Phosphine Boranes

Phosphine boranes are Lewis acid/base adducts that have been known for over a century, but interest in the chemistry of these molecules has increased in recent decades. ${ }^{16}$ Aside from serving as air-stable precursors of functionalized phosphines, these adducts have been used as hydrogen atom donors in Barton-McCombie deoxygenations ${ }^{17}$ and as polyphosphinoborane monomers. ${ }^{18} \mathrm{~A}$ marked advantage of phosphine boranes is their tendency to be air-stable crystalline solids, whereas the parent phosphines (particularly primary and secondary alkyl phosphines) are often air sensitive, pyrophoric, and/or malodorous. These adducts are generally synthesized from the corresponding phosphines with $\mathrm{BH}_{3} \cdot \mathrm{THF}$ or $\mathrm{BH}_{3} \cdot \mathrm{SMe}_{2}$ as the borane donor, although alternative methods for their preparation have been described, such as a one-pot reduction/boron complexation of phosphine oxides. ${ }^{19}$

In addition to protecting the phosphine against oxidation, the borane decreases the magnitude of the pKa of hydrogens adjacent to phosphorus (as with tertiary phosphine boranes) or directly attached to phosphorus (in the case of secondary phosphine boranes). ${ }^{20}$ This property enables phosphine boranes to be convenient, air-stable reagents that can be readily elaborated into more complex phosphines. For example, dimethylphenyl tertiary phosphine boranes can be subjected to an enantioselective deprotonation using $s$ - BuLi and (-)-sparteine. Subsequent addition of a copper (II) salt promotes oxidative anion dimerization to provide access to $P$-chiral dppe derivatives. ${ }^{21}$ The $\mathrm{P}-\mathrm{H}$ bond in secondary phosphine boranes (1-6) can be deprotonated with $n$-BuLi, NaH , or $t$-BuOK, and the resulting anion can be subjected to a hydrophosphination or alkylation reaction.

Imamoto and co-workers first reported in 1990 that secondary phosphine boranes (1-6) can hydrophosphinate Michael acceptors in the presence of substoichiometric potassium hydroxide. ${ }^{22}$ Since Imamoto's pioneering work, Gaumont and co-workers have found that hydrophosphinations of alkenes and alkynes can be conducted with secondary phosphine boranes by simply heating a neat mixture of the two reactants (Figure 2). ${ }^{23}$ For example, terminal alkynes were hydrophosphinated to afford vinyl phosphines (1-8) with high regioselectivity and $E / Z$ selectivity. ${ }^{24}$ Similar selectivity was observed in the thermal hydrophosphination of vinyl ethers ${ }^{25}$ (1-9) and 1,1-disubstituted alkenes. ${ }^{26} \mathrm{~A}$ thermal hydrophosphination of $\beta$-pinene afforded the limonene-derived product 1-10 through a ring opening, which suggests that these thermal transformations proceed through a radical mechanism. ${ }^{26}$


Figure 2. Thermal hydrophosphinations of secondary phosphine boranes.
In 2009, Busacca and co-workers serendipitously discovered a base-mediated hydrophosphination of alkynes using secondary phosphine boranes. This unprecedented reaction was discovered while attempting to prepare ligand 1-14 through nucleophilic aromatic substitution of imidazoline 1-11 (Scheme 1). Further investigation revealed that the hydrophosphination is general for a range of substituted alkynes. Vinyl phosphine boranes such as $\mathbf{1 - 1 3}$ with are obtained in good yields and $E: Z$ ratios ranging from 3:1 to $>20: 1 .{ }^{27}$ In cases where differentially substituted alkynes were used, complete regioselectivity was attained based on the electronic parameters of the alkyne substituents. This method offers advantages over Gaumont's thermal hydrophosphination in that it is effective on internal alkynes, proceeds at ambient temperature, and does not require a large excess of the alkyne.


Scheme 1. Discovery of a base-mediated hydrophosphination of internal alkynes.

This operationally simple hydrophosphination method enables access to phosphines that are potentially useful for catalytic applications. The vinyl phosphines derived from alkyne hydrophosphination (1-15) were enantioselectively hydrogenated with cationic rhodium and ( $R$, R)-Skewphos ${ }^{28}$ to afford chiral monodentate phosphine boranes with $>99 \%$ ee. ${ }^{27}$ Alkyne derivatives such as propargyl amines, and propargyl alcohols, allenyl phosphine oxides also effectively undergo this hydrophosphination (1-16 and 1-17, Figure 3). ${ }^{29}$ These more highly functionalized hydrophosphination products can also serve as ligand precursors. For example, the allene hydrophosphination products (1-16) are protected equivalents of bis-phosphine monooxides, which are a useful class of ligands for catalysis. ${ }^{30}$ The products of propargyl amine or alcohol hydrophosphination are also precursors to $\mathrm{P}-\mathrm{N}$ or $\mathrm{P}-\mathrm{O}$ bidentate ligands. The bidentate binding mode of 1-17 $(\mathrm{X}=\mathrm{NHPh})$ was confirmed by preparing an ${ }^{15} \mathrm{~N}$ labeled substrate and evaluating the ${ }^{15} \mathrm{~N}$ and ${ }^{31} \mathrm{P}$ coupling constants of Pt or Rh complexes. ${ }^{29}$


1-17




1-16

Figure 3. Base-mediated hydrophosphinations with phosphine boranes.
Given the simplicity of this base-mediated hydrophosphination protocol, it became of interest to extend this method to the hydrophosphination of heterocumulenes such as carbodiimides. In an initial screening effort, Busacca and co-workers found that secondary
phosphine boranes do indeed hydrophosphinate carbodiimides to provide phosphaguanidine boranes 1-19-1-24 (Table 1). Carbodiimides with $N, N$ '-dialkyl, $N, N$ '-diaryl and $N$-alkyl, $N$ '-aryl substituents undergo this transformation equally well (entry 1 vs. entry 3 ). Di- $t$-butyl and dicyclohexyl phosphine boranes were also similarly effective in the reaction. The products of the reaction were typically crystalline solids that were easily purified by chromatography on $\mathrm{SiO}_{2}$. Xray crystal structures of two representative phosphaguanidine boranes (1-21 and 1-23) were obtained (Figure 4). ${ }^{31}$

Table 1. Preliminary results from Busacca et al. on the hydrophosphination of carbodiimides.


| entry | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | $\mathrm{R}^{3}$ | product | yield (\%) |
| :--- | :--- | :--- | :---: | :---: | :---: |
| 1 | Cy | Cy | Cy | $\mathbf{1 - 1 9}$ | 71 |
| 2 | Cy | Cy | $\mathrm{t}-\mathrm{Bu}$ | $\mathbf{1 - 2 0}$ | 58 |
| 3 | Ph | Ph | Cy | $\mathbf{1 - 2 1}$ | 69 |
| 4 | Ph | Ph | $\mathrm{t}-\mathrm{Bu}$ | $\mathbf{1 - 2 2}$ | 77 |
| 5 | Cy | Ph | Cy | $\mathbf{1 - 2 3}$ | 63 |
| 6 | Cy | Ph | $\mathrm{t}-\mathrm{Bu}$ | $\mathbf{1 - 2 4}$ | 58 |



1-21


1-23

Figure 4. Phosphaguanidine boranes crystallized by Busacca et al.

### 1.1.3 Structure and Utility of Phosphaguanidines

Guanidines have long been of interest to the chemistry community due to their structure, reactivity, and occurrence in biologically active natural products. ${ }^{32}$ The guanidine structure is wellrepresented among common nucleophilic organocatalysts, ${ }^{33}$ and a significant number of chiral guanidines have been designed to accomplish various asymmetric catalytic transformations. ${ }^{34}$ Phosphaguanidines, wherein one nitrogen from the guanidine unit is substituted for a phosphine, have received significantly less attention. These molecules are typically prepared by the hydrophosphination of carbodiimides with secondary phosphines. ${ }^{13 \mathrm{a}}$

An extensive NMR, computational, and crystallographic study of phosphaguanidines was undertaken by Coles and co-workers. ${ }^{35}$ Their research demonstrated that phosphaguanidines have distinct conformational preferences that are governed by the nature of the $P$ and $N$ substituents. A nomenclature system was developed to define the eight possible conformations of a given phosphaguanidine that arise from three binary factors: $E$ vs $Z$ stereochemistry about the $\mathrm{C}=\mathrm{N}$ bond, syn vs anti orientation of the $N-\mathrm{R}$ substituents and $\alpha$ vs $\beta$ alignment of the guanidine $N-H$ with the $P-R$ substituents (Figure 5). It was found that these orientations are dependent on the nature of the $P-R$ substituent: $P$-diphenyl phosphines appear only in the $E_{\text {syn }} \alpha$ conformation, whereas $P$ cyclohexylphosphines show a mixture of $E_{\text {syn }} \alpha$ and $Z_{\text {anti }} \alpha / \beta$ isomers. When the phosphines were transformed to the corresponding phosphine sulfides or selenides, a mixture of $E_{\text {syn }} \beta$ and $Z_{\text {syn }}$ $\beta$ conformers were observed, with the former being preferred in $P$-diphenyl phosphines and the latter in $P$-dicyclohexyl phosphines.


Figure 5. Eight possible conformations of phosphaguanidines annotated with the observations of Coles et al. ${ }^{32}$
The $Z_{\text {syn }} \beta$ conformation allows for an interaction between the guanidine $\mathrm{N}-\mathrm{H}$ and the chalcogen atom in the phosphine sulfides and selenides. In the case of the phosphine selenide $\mathbf{1 -}$ 25, the interaction was evident through a $J_{\mathrm{SeH}}$ NMR coupling of 6 Hz . A similar alignment between the $\mathrm{N}-\mathrm{H}$ and the borane was observed in the crystal structures of phosphaguanidine boranes $\mathbf{1 - 2 1}$ and 1-23, as illustrated in Figure 6. The distances between the phosphaguanidine $\mathrm{N}-\mathrm{H}$ and boron/selenium atoms are comparable between the two structures $(2.68 \AA \mathrm{H}-$ Se distance vs. 2.57 $\AA$ H-B distance).


Figure 6. Comparison between the conformations of phosphine borane 1-21 and phosphine selenide 1-25.

As part of their study, Coles and co-workers prepared the first example of a phosphaguanidine borane (1-27) by treating $\mathbf{1 - 2 6}$ with $\mathrm{BH}_{3} \bullet$ THF (Scheme 2 ). ${ }^{35}$ While this synthesis was successful in providing crystals of 1-27 suitable for X-ray diffraction, the yield of the transformation was low, and the product was reported to be unstable in solution. Interestingly, 1-26 adopted an $E_{\text {syn }}$ (pseudo) $\alpha$ conformation in the solid state, which stands in contrast to the $Z_{\text {syn }} \beta$ conformation observed in the phosphaguanidine boranes 1-21 and 1-23 prepared by Busacca and co-workers (Figure 4).




Scheme 2: Phosphaguanidine borane 1-27 prepared by Coles et al. ${ }^{35}$
Despite these studies on the preparation and structure of phosphaguanidines, no examples of their use in catalysis have been reported to date. This is somewhat surprising considering that phosphaguanidines have been shown to exhibit either mono- or bidentate binding modes to metals such as molybdenum. ${ }^{14 \mathrm{~b}}$ Upon deprotonation of the phosphaguanidine $N-H$, the resulting anion has exhibited a wide array of binding modes to lithium, ${ }^{14 \mathrm{a}}$ aluminum, ${ }^{36}$ titanium, ${ }^{36}$ zirconium, ${ }^{36}$ thallium, ${ }^{37}$ and rhodium. ${ }^{38}$ Similarly, there have been no reports on the application of phosphaguanidines or their borane-protected analogs as organocatalysts, where the nitrogen atoms could plausibly participate in nucleophilic ${ }^{33}$ or hydrogen bonding ${ }^{39}$ catalysis. The opportunity to identify new types of catalysts provided further motivation to develop a convenient method for the hydrophosphination of carbodiimides to produce phosphaguanidine boranes.

### 1.2 RESULTS AND DISCUSSION

### 1.2.1 Hydrophosphination of Acyclic Carbodiimides

Given the previous success with using deprotonated secondary phosphine boranes to conduct hydrophosphination reactions, ${ }^{27,29}$ we sought to extend the scope of this method to include the hydrophosphination of carbodiimides. Preliminary studies by collaborators at Boehringer Ingelheim suggested that under the previously developed conditions ( NaH in DMAc), hydrophosphination of alkyl and aryl carbodiimides could be conducted with aliphatic phosphine boranes (Table 1). However, it remained to be explored as to whether aromatic phosphine boranes or other carbodiimides would be tolerated in the transformation.

Thus, a series of carbodiimides containing both electron-rich and electron-poor aryl substituents was prepared by desulfurization of the corresponding disubstituted thioureas. ${ }^{40}$ In cases where the necessary thioureas were not commercially available or obtained from collaborators, they were prepared by reacting the appropriate primary amine with an isothiocyanate. This approach provided convenient access to carbodiimides 1-28-1-31 (Table 2).

Table 2. Preparation of carbodiimides through desulfurization of the corresponding thioureas.

|  |  | $\xrightarrow[\substack{\mathrm{MsCl} \\ \text { DMAP, } \mathrm{Et}_{3} \mathrm{~N}}]{\mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{rt}}$ | $R^{1}{ }^{N}$ | $N^{-R^{2}}$ |
| :---: | :---: | :---: | :---: | :---: |
| entry | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | product | yield (\%) |
| 1 | $\left(3,5-\mathrm{CF}_{3}\right)_{2} \mathrm{C}_{6} \mathrm{H}_{3}$ | Cy | 1-28 | 83 |
| 2 | $(2-\mathrm{OMe}) \mathrm{C}_{6} \mathrm{H}_{4}$ | Cy | 1-29 | 72 |
| 3 | $(4-\mathrm{OMe}) \mathrm{C}_{6} \mathrm{H}_{4}$ | Ph | 1-30 | 87 |
| 4 | allyl | allyl | 1-31 | 58 |

These carbodiimides, along with commercially available dicyclohexylcarbodiimide (DCC), were tested in the hydrophosphination reaction (Table 3). ${ }^{31}$ Both electron-rich and
electron-poor aryl carbodiimides were effective substrates (entries $1-3$ ), although a diallyl carbodiimide gave a somewhat lower yield (entry 4). Aryl secondary phosphine boranes showed poor reactivity, regardless of the nature of the carbodiimide (entry 5). Commercially available $N$, $N$ '-bistrimethylsilyl carbodiimide could not be hydrophosphinated with either cyclohexyl or $t$ butyl phosphine boranes, perhaps due to the high steric encumbrance of this substrate (entry 6).

Table 3. Hydrophosphination of carbodiimides.

|  | $R^{R^{1}}{ }_{N}-R^{2}$ |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| entry | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | $\mathrm{R}^{3}$ | product | yield (\%) |
| 1 | Ph | (4-OMe) $\mathrm{C}_{6} \mathrm{H}_{4}$ | $t$-Bu | 1-32 | 78 |
| 2 | Cy | $\left(3,5-\mathrm{CF}_{3}\right)_{2} \mathrm{C}_{6} \mathrm{H}_{3}$ | Cy | 1-33 | 55 |
| 3 | Cy | (2-OMe) $\mathrm{C}_{6} \mathrm{H}_{4}$ | Cy | 1-34 | 58 |
| 4 | allyl | allyl | Cy | 1-35 | 37 |
| 5 | Cy | Cy | (4-Me) $\mathrm{C}_{6} \mathrm{H}_{4}$ | (-) | <10 |
| 6 | $\mathrm{SiMe}_{3}$ | $\mathrm{SiMe}_{3}$ | Cy or $t$-Bu | (-) | 0 |

While the borane motif on products 1-32-1-35 prevents oxidation of the phosphaguanidine, it must ultimately be removed to enable ligation of the phosphine to a metal center. Heating a toluene solution of a phosphine borane with DABCO is an effective approach for the deboronation of aromatic phosphine boranes. ${ }^{31}$ Alternatively, removal of the borane can be conducted with $\mathrm{HBF}_{4}$ to provide a tralkylphosphonium tetrafluoroborate salt. These phosphonium salts can be free-based in situ, and thus serve as convenient air-stable surrogates of alkyl phosphines for use in catalysis. ${ }^{41}$

Phosphaguanidine 1-19 was subjected to this acid-mediated deboronation. These conditions, however, did not cleave the alkyl phosphaguanidine boranes, but rather protonated the guanidine functionality to afford phosphaguanidinium borane tetrafluoroborate salt 1-36 (Scheme 4). In contrast to generic trialkylphosphines, the protonation of the guanidine apparently deactivates the phosphine borane and consequently prevents deboronation. The structure of 1-36,
which was confirmed by X-ray crystallography, features an unmodified borane and a phosphaguanidinium core with identical $\mathrm{C}-\mathrm{N}$ bond lengths of $1.316 \AA$. The guanidinium substituents of $\mathbf{1 - 3 6}$ crystallized in an $\mathrm{E}_{\text {syn }} / Z_{\text {anti }}$ conformation, much like a similar phosphaguanidinium salt that was previously crystallized by Mansfield et al. ${ }^{42}$


Scheme 3. Attempted acidic deboronation with $\mathrm{HBF}_{4}$ results in tetrafluoroborate salt 1-36.
The oxidation of the free phosphaguanidines to the phosphaguanidine oxides was also explored. Interestingly, Hitchcock and co-workers reported that phosphaguanidines undergo $\mathrm{C}-\mathrm{P}$ bond cleavage when treated with $t-\mathrm{BuOOH}$ to afford an amidinium phosphinate salt. ${ }^{42}$ In agreement with this result, treatment of phosphine $\mathbf{1 - 3 7}$ with $t$ - BuOOH using Hitchcock's procedure resulted in a complex mixture that likely contains an analogous species (Table 4). A cleaner oxidation occurred when $m$-CPBA was used as the oxidant, and the desired phosphine oxide 1-38 was isolated in $33 \%$ yield after chromatography on $\mathrm{SiO}_{2}$. Dilute aqueous hydrogen peroxide, however, provided phosphine oxide 1-38 in nearly quantitative yield. This series of experiments suggests that despite their relative paucity in the literature, ${ }^{43}$ phosphaguanidine oxides are stable entities that can be prepared through the oxidation of phosphaguanidines with the appropriate oxidizing agent.

Table 4. Oxidation of phosphaguanidine 1-37.


### 1.2.2 Preparation and Hydrophosphination of a Cyclic Carbodiimide

While this base-mediated hydrophosphination of acyclic carbodiimides is operationally convenient, the method has limitations. Aryl phosphine boranes showed poor reactivity, which significantly limits the scope of the phosphaguanidines that can be obtained. In addition, the products have a congested steric environment around the phosphorous center that disfavors the binding of metals and reduces the ability of the guanidine motif to act as an organocatalyst. Attempts to use the deprotected phosphaguanidines in Cu-catalyzed conjugate additions, Rh-catalyzed hydrogenation, and organocatalytic transformations were unsuccessful. These results may suggest that the phosphine center of these molecules is poorly accessible for metal binding. Further complicating this structural issue is the fact that some phosphaguanidine products are a mixture of $E / Z$ isomers in solution (for example, tetracyclohexyl product $\mathbf{1 - 1 9}$ is a $2: 1$ mixture of $Z_{\text {syn }} / E_{\text {syn }}$ isomers as judged by ${ }^{13} \mathrm{C}$ NMR). ${ }^{31}$ This isomerism suboptimal for the design of phosphaguanidine ligands for use in stereoselective catalysis.

To produce phosphaguanidines with a greater potential for catalytic applications, analogs with a well-defined rigid structure and a more accessible phosphorus center were pursued. The
most intriguing approach toward this end is to constrain the nitrogen substituents in a $\mathrm{Z}_{\text {syn }}$ conformation by embedding them in a medium-sized ring. Efficient access to such a scaffold could be achieved through the hydrophosphination of a cyclic carbodiimide. Although seven-membered or smaller cyclic carbodiimides are unprecedented, eight-membered carbodiimides have been prepared from cyclic thioureas or amidooximes. ${ }^{44}$ These cyclic molecules are useful tools for experimental studies of the nitrogen inversion barrier in carbodiimides. ${ }^{45}$

Molina and co-workers have reported an efficient preparation of cyclic carbodiimide 1-41 from commercially available $2,2^{\prime}$-ethylenedianiline 1-39 (Scheme 4). ${ }^{46}$ A diazotization/azidation sequence, formation of a bis-iminophosphorane, and double Staudinger-aza-Wittig ring closure with $\mathrm{CO}_{2}$ provided an efficient access to the cyclic carbodiimide $\mathbf{1 - 4 1}$. Although the originally reported preparation used stoichiometric DMAP and $\mathrm{Boc}_{2} \mathrm{O}$ to form a $\mathrm{CO}_{2}$ equivalent in situ, ${ }^{47}$ bubbling $\mathrm{CO}_{2}$ gas through the solution with a balloon was found to be a more convenient and equally effective means to conduct this ring closure. The three-step sequence is operationally straightforward, uses inexpensive reagents, and requires only one chromatographic purification.


Scheme 4. Preparation of cyclic carbodiimide 1-41.
The nine-membered cyclic carbodiimide 1-41 is a strain-activated molecule. Although this ring strain has not been experimentally quantified, the strained nature of 1-41 and related derivatives has been verified spectroscopically. ${ }^{46,} 48$ For example, the abnormally high carbodiimide IR stretch of $\mathbf{1 - 4 1}\left(2186 \mathrm{~cm}^{-1}\right.$, as compared to the typical range of $\left.2120-2145 \mathrm{~cm}^{-1}\right)$ is suggestive of a strain-induced weakening of the carbodiimide $\mathrm{C}=\mathrm{N}$ bonds. This strain activation may render 1-41 more reactive than DCC and other acyclic carbodiimides in hydrophosphination
chemistry. Nonetheless, carbodiimide $\mathbf{1 - 4 1}$ is a bench-stable solid that showed no detectable decomposition after several months of storage at room temperature under air.

The hydrophosphination of carbodiimide 1-41 was conducted with a variety of phosphine boranes (Table 5). In contrast to acyclic carbodiimides, consistent yields were obtained for both aliphatic (entries 1,2 ) and aromatic (entries 3,4 ) secondary phosphine boranes using the standard hydrophosphination conditions. A mixed phenylcyclohexyl phosphine borane also performed well (entry 5). This improved reaction scope may be attributed to the enhanced reactivity of carbodiimide 1-41, which benefits from strain-activated $\mathrm{C}=\mathrm{N}$ bonds and high accessibility of the central $s p$-hybridized carbon. The products resulting from hydrophosphination of 1-41 (1-42-146) constitute the first examples of a phosphaguanidines that are embedded in a medium ring system.

Table 5. Hydrophosphinations on cyclic carbodiimide 1-41.


The cyclic phosphaguanidines were typically colorless, crystalline solids that could be purified by chromatography on $\mathrm{SiO}_{2}$ or by recrystallization. Single crystal X-ray diffraction was conducted on suitable crystals of phosphaguanidines 1-42 and 1-44. Interestingly, in both cases, the nine-membered ring adopts a compact boat-like conformation (Figure 7). This produces a tight dihedral angle $\left(17-24^{\circ}\right)$ for the $\mathrm{H}-\mathrm{C}-\mathrm{C}-\mathrm{H}$ bonds of the ethylene bridge. Consequently, the aromatic
rings of the guanidine sit nearly aligned, with an average inter-ring distance of about $4 \AA$. Although the shortest inter-ring distances $(2.8 \AA)$ are comparable to those of [2,2]-p-cyclophanes (2.7-3.2 $\AA$ ) and graphite ( $3.4 \AA$ ), the rings angle outward to give much longer maximal inter-ring distances $(5.2-5.3 \AA)($ Table 6$) .{ }^{49}$


1-42





Figure 7. Crystal structures for compounds 1-42 and 1-44.
Table 6. Pertinent data from the crystal structures of 1-42 and 1-44.


| $R$ | ethylene bridge <br> H-C-C-H <br> dihedral angle $\left(^{\circ}\right)$ | $\mathrm{B}-\mathrm{P}-\mathrm{C}-\mathrm{NH}$ <br> dihedral angle $\left(^{\circ}\right)$ | $\mathrm{C}_{1}-\mathrm{C}_{1}{ }^{\prime}$ <br> distance $(\AA)$ | $\mathrm{C}_{4}-\mathrm{C}_{4}{ }^{\prime}$ <br> distance $(\AA)$ | $\mathrm{C}_{n}-\mathrm{C}_{n}{ }^{\prime}$ average <br> distance $(\AA)$ |
| :--- | :---: | :---: | :---: | :---: | :---: |
| $t-\mathrm{Bu}$ | 22 | 4 | 2.80 | 5.15 | 3.98 |
| $4-\mathrm{MeC}_{6} \mathrm{H}_{4}$ | 24 | 21 | 2.83 | 5.33 | 4.07 |

In addition, 1-42 and 1-44 were found to have an $\alpha$ (eclipsed) orientation between the $\mathrm{N}-$ H and $\mathrm{BH}_{3}$ in the solid state. This phenomenon was also observed in the acyclic phosphaguanidines 1-21 and 1-23 that were crystallized by Busacca and co-workers (Figure 6). Modeling studies of $t$-butyl phosphine borane $\mathbf{1 - 4 2}$ at the $\mathrm{HF} / 6-311+\mathrm{G}^{* *}$ level of theory revealed that the
$\alpha(\mathrm{N}-\mathrm{H}$ eclipsed) orientation is preferred over the $\beta$ orientation by $10 \mathrm{kcal} / \mathrm{mol}$, although the exact origin of this preference is not known.

Deprotection of the aromatic cyclic phosphaguanidine boranes using DABCO in toluene proceeded smoothly to provide phosphines 1-47 and 1-48 (Table 7). The mixed aromatic/aliphatic phosphaguanidine borane 1-46 also successfully underwent deprotection to afford racemic phosphine 1-49. The free phosphines 1-47 and 1-48 were oxidized using dilute hydrogen peroxide to provide phosphine oxides $\mathbf{1 - 5 0}$ and $\mathbf{1 - 5 1}$ in high yield with no evidence of phosphaguanidine cleavage (Table 8).

Table 7. Deprotection of cyclic phosphaguanidine boranes.


Table 8. Oxidation of cyclic phosphaguanidines.


### 1.2.3 Hydrophosphination with Phospholane Boranes and Attempted Catalytic

## Applications

The use of strained cyclic carbodiimide 1-41 as a hydrophosphination substrate enabled the preparation of both aromatic and aliphatic phosphaguanidine boranes in good yield (Table 5). As evidenced by crystallography studies on cyclic phosphaguanidine boranes 1-42 and 1-44 (Figure 7), these products adopt predictable, well-defined geometries that can enable the rational design of ligands and organocatalysts. Moreover, the ability to readily remove the borane and expose the free phosphines 1-47-1-49 (which can be manipulated on the benchtop using standard precautions) also bodes well for the application of the molecules as phosphine ligands in catalytic applications.

Ideally, chirality could be introduced into cyclic phosphaguanidine scaffold to provide access to molecules that can catalyze enantioselective transformations. Attempts were made to produce a chiral derivative of cyclic carbodiimide 1-41 by installing a cyclic acetal on the ethylene backbone, but these efforts were unsuccessful. Alternatively, chirality can be introduced by conducting the hydrophosphinations with a chiral secondary phosphine borane. If the chiral secondary phosphine borane was also cyclic in nature, this approach could provide a phosphaguanidine core with a highly accessible phosphine center.

Five-membered phospholane boranes have the greatest precedent among cyclic secondary phosphine boranes, and were the primary targets of this study. Phospholane borane 1-54 was prepared from 1,4-dibromobutane $\mathbf{1 - 5 2}$ through double Grignard formation, trapping with a dichlorophosphine, formation of a tertiary phosphine borane, and reduction with lithium metal (Scheme 5). ${ }^{50}$ This preparation was low yielding, but the inexpensive reagents allowed the reactions to be sufficiently scaled to provide $>200 \mathrm{mg}$ of $\mathbf{1 - 5 4}$.


Scheme 5. Preparation of phospholane borane 1-54.
Chiral $C 2$ symmetric phospholanes have been prepared by introducing phenyl substituents at the 2- and 5-positions of the phospholane ring. These chiral phospholanes are the basis of the BPE ligand class, which are useful for asymmetric hydrogenation and other transformations. ${ }^{51}$ Several methods to prepare these phospholanes have been reported; however, the method previously used by Busacca proved to be the most effective. ${ }^{27}$ The phospholane scaffold was accessed from diene 1-55 by using a McCormack [4+1] cyclization. ${ }^{52}$ Subsequent hydrogenation and epimerization/hydrolysis afforded phosphinic acid 1-57, which was subjected to a classical resolution with quinine. Reduction with phenylsilane and borane complexation then afforded each enantiomer of phospholane borane 1-58 (Scheme 6).


Scheme 6. Preparation of chiral phospholane borane 1-58.

The phospholanes 1-54 and 1-58 were used in hydrophosphination reactions with cyclic carbodiimide 1-41 (Table 9). While unsubstituted phospholane borane 1-54 was less effective in the transformation, each of the 1-58 enantiomers underwent hydrophosphination in moderate to good yield to afford chiral phospholane borane 1-60. It was found that the sign of the specific rotation for 1-60 is opposite of that of the corresponding secondary phosphine borane precursor (1-58).

Table 9. Hydrophosphination of cyclic carbodiimide 1-41 with phospholane boranes.



| R | enantiomer product | yield (\%) |  |
| :---: | :---: | :---: | :---: |
| H | $(-)$ | $\mathbf{1 - 5 9}$ | 35 |
| Ph | $(R, R)$ | $(-)-1-60$ | 79 |
| Ph | $(S, S)$ | $(+)-1-60$ | 61 |

Like the other phosphaguanidines, 1-60 can be deprotected by heating in the presence of excess DABCO (Scheme 7). The resulting phosphine 1-61 is sufficiently air-stable to be briefly manipulated in air, but the material was placed under an inert atmosphere for storage. The potential of this phosphine to bind to transition metals was evaluated by forming a complex with platinum (II) chloride. When a solution of $\mathbf{1 - 6 1}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was treated with 0.5 equivalents of $\mathrm{PtCl}_{2}$ and left undisturbed, the trans $-\mathrm{PtCl}_{2}$ (phosphine) $)_{2}$ complex 1-62 formed as light yellow prisms. The structure of this complex was confirmed by X-ray crystallography (Figure 8). The observed Pt-P bond length $(2.305 \AA)$ and $J_{\text {PtP }}$ coupling constant $(2352 \mathrm{~Hz})$ of 1-62 is consistent with similar $\mathrm{PtCl}_{2}$ complexes with sterically bulky phosphines. ${ }^{53}$

( $R, R-$ ) 1-60

( $R, R-$ ) 1-61


1-62

Scheme 7. Deprotection and platinum complex formation with chiral phosphine borane 1-61.


Figure 8. X-ray crystal structure of platinum complex 1-62. Hydrogens are omitted for clarity.
Despite the success in forming complex 1-62, the application of 1-61 in asymmetric catalysis did not yield successful results. For example, enantioselective borylation ${ }^{54}$ or Heck reactions ${ }^{55}$ were 1-61 was used as a ligand either showed poor conversion or gave extensive decomposition of the substrate. Rhodium-catalyzed hydrogenation was successfully carried out using 1-61; however, the hydrogenated product had no detectable enantiomeric excess.

Attempts to use the either the free phosphaguanidines or their borane complexes as guanidine-type hydrogen bonding catalysts were similarly unsuccessful. The addition of diphenylphosphite to nitrostyrene (1-63), for example, is reported to proceed in excellent yield and enantioselectivity using a chiral, bicyclic guanidine catalyst (Table 10). ${ }^{56}$ However, $20 \mathrm{~mol} \%$ of
achiral phosphine borane 1-59 was required to get similarly high yields in this transformation. Disappointingly, when the same catalyst loading of the chiral derivative 1-60 was used, a much lower yield was obtained. This is possibly due to the steric encumbrance of the phosphaguanidine motif imparted by the phenyl rings on the phospholane. In general, the poor catalytic activity of these molecules suggests that alternative phosphaguanidine architectures will need to be considered for the preparation of phosphaguanidines with high utility in catalysis.

Table 10. Attempted hydrogen bonding catalysis with phosphaguanidine boranes.


### 1.3 CONCLUSIONS

A method for the hydrophosphination of carbodiimides using air-stable secondary phosphine boranes was developed. Various aromatic and aliphatic substituents were tolerated on the carbodiimide substrate in the transformation. However, aryl phosphine boranes were less effective than alkyl phosphine boranes. An acidic deboronation protocol was attempted on the alkyl phosphaguanidine products, but protonation of the guanidine occurred and phosphaguanidinium borane salts were isolated. The free phosphines could, however, be oxidized to phosphine oxides in good yield without cleavage of the $\mathrm{C}-\mathrm{P}$ bond.

A cyclic carbodiimide was prepared, which underwent a facile hydrophosphination with a wider scope of phosphine boranes. Thus, a series of nine-membered cyclic phosphaguanidines were prepared, which were found to have interesting boat-like structures that were elucidated by X-ray crystallography. Chiral derivatives of the cyclic phosphaguanidines were prepared through hydrophosphination with chiral, cyclic phospholane boranes. The chiral phospholane borane was deprotected, and a crystalline platinum (II) complex was prepared with this material. Unfortunately, attempts to use this chiral phosphaguanidine either as a ligand in asymmetric catalysis or as an organocatalyst were not fruitful.

### 2.0 NUCLEOPHILIC ADDITIONS TO BICYCLO[1.1.0]BUTYL NITRILES

### 2.1 INTRODUCTION

### 2.1.1 Bicyclo[1.1.0]butanes: Structure and Reactivity

The concept of molecular strain has a storied history that dates back to the proclamations of Adolf von Baeyer in the late ninetieth century. ${ }^{57}$ Indeed, strain activation underlies many aspects of modern synthetic chemistry, such as strategic design in total synthesis, ${ }^{58}$ strain-enabled reaction methods, ${ }^{59}$ and bioorthogonal tranformations. ${ }^{60}$ Numerous strained molecules containing threeand four-membered rings have been designed and applied in target-oriented synthesis. ${ }^{61}$

Among these strained rings, bicyclo[1.1.0]butanes are distinguished by their high strain energy ( $64 \mathrm{kcal} / \mathrm{mol}$ ) and unique structural properties. ${ }^{57}$ Though they are quite strained, many derivatives are readily prepared and are bench-stable. The first synthesis of an authentic bicyclo[1.1.0]butane derivative was published by Wiberg in 1959, who prepared methyl bicyclo[1.1.0]butanoate by ring contraction of 3-bromocyclobutane methyl ester. ${ }^{62}$ The parent bicyclo[1.1.0]butane hydrocarbon, which is a gas at room temperature (boiling point $8^{\circ} \mathrm{C}$ ), can be synthesized in a similar fashion through intramolecular Wurtz coupling of 1,3dihalocyclobutanes. ${ }^{63}$ More versatile synthetic methods to access functionalized bicyclo[1.1.0]butanes have since been developed, including those based on Simmons-Smith chemistry,,${ }^{64}$ intramolecular carbenoid cyclizations, ${ }^{65}$ or ring closures of appropriately substituted cyclopropanes. ${ }^{66}$

Studies of the structure and bonding in bicyclo[1.1.0]butanes have elucidated several unique features of these molecules. For example, bicyclo[1.1.0]butanes adopt a "butterfly" like geometry, with an interflap angle of about $123^{\circ}$ (Figure 9). The $\mathrm{C}-\mathrm{C}$ bond distances of both the lateral and central bonds are $1.507 \AA$, which is slightly shorter than those of typical cyclopropanes $(1.512 \AA) .{ }^{67}$ The $\mathrm{C}-\mathrm{H}$ bonds range in length from $1.142 \AA$ (methine) to $1.194 \AA$ (exo methylene).


Figure 9. Key structural parameters of bicyclo[1.1.0]butanes. ${ }^{67 \mathrm{a}}$
Although the $\mathrm{C}-\mathrm{C}$ bonds of bicyclo[1.1.0]butanes are equal in length, the central bond is distinguished from the lateral bonds by its substantial $\pi$ character. ${ }^{68}$ Experimental verification of this character was reported by Moore and co-workers, who found that bicyclo[1.1.0]butane dimer 2-1 has a substantial UV absorption at 195 nm that is characteristic of a conjugated $\pi$ system (Figure 10). ${ }^{69}$ The nature of this central bond imparts a high $s$ character to the methine $\mathrm{C}-\mathrm{H}$ bonds, which is manifested by an unusually large ${ }^{13} \mathrm{C}-\mathrm{H}$ coupling constant $(202 \mathrm{~Hz})$ that is intermediate between ethylene $(156 \mathrm{~Hz})$ and acetylene $(249 \mathrm{~Hz}) .{ }^{62 b}$

Much like cyclopropanes, ${ }^{70}$ the central bond of bicyclo[1.1.0]butanes can impact the reactivity of adjacent carbinols through hyperconjugative stabilization. In fact, the hydrolysis of bicyclo[1.1.0]butane ester 2-3 was found to be 3 orders of magnitude faster than the corresponding cyclopropane ester 2-2 (Figure 10). ${ }^{62 \mathrm{~b}}$ In contrast to cyclopropane 2-2, which formed the expected cyclopropylmethanol upon hydrolysis, bicyclobutane $\mathbf{2 - 3}$ reacted preferentially at the central bond to produce cyclobutanols.


Figure 10. Experimental verifications of the unusual $\pi$ character of the central bond of bicyclo[1.1.0]butanes.
Despite the very interesting structure and reactivity of bicyclo[1.1.0]butanes, their application in preparative organic synthesis has been limited. Most synthetic studies of these strained hydrocarbons have focused on their solvolysis and transition-metal mediated isomerizations to dienes and related structures, which was a popular research focus in the 1970's and early 1980 's. ${ }^{71}$ Various metals were studied in this context, including palladium, nickel, rhodium, silver, and platinum. ${ }^{72}$

More recently, Walczak and Wipf have demonstrated the value of bicyclo[1.1.0]butanes as precursors of structurally novel heterocycles (Scheme 8). ${ }^{73}$ For example, an intramolecular Rhmediated isomerization of alkene-tethered bicyclo[1.1.0]butanes (2-4) can selectively provide pyrrolidine 2-5 or azepine 2-6 in good yield, depending on the ligand and rhodium source employed. ${ }^{74}$ When phenyl-substituted bicyclo[1.1.0]butyl phosphinamides are alkylated with allyl and propargyl electrophiles using phase-transfer catalysis, the intermediate alkylated products spontaneously rearrange to spirocyclic cyclobutenes such as 2-7 through an Alder-ene reaction. In contrast, the use of cinnamyl bromide derivatives enables the synthesis of pyrrrolidines such as 28 through a formal [2+2] reaction pathway. ${ }^{75}$ Studies of these thermal transformations using electron spin resonance spectroscopy suggest that they occur through biradical intermediates. ${ }^{76}$ The unique heterocycles that can be obtained from bicyclo[1.1.0]butanes have been applied in medicinal chemistry studies ${ }^{77}$ and in a synthetic approach toward the daphniglaucin family of polycyclic alkaloids. ${ }^{78}$


Scheme 8. Intramolecular transformations of bicyclo[1.1.0]butanes developed by Walczak and Wipf. ${ }^{\text {74-75 }}$ ${ }^{\text {a }}$ The thermal rearrangements were conducted through phase-transfer allylations of the bicyclobutyl phosphinamides with the appropriate allyl bromide. The intermediate allylated materials (i.e., 2-4) underwent rearrangement in situ at rt.

### 2.1.2 Nucleophilic Additions to Bicyclo[1.1.0]butanes

Both the highly-strained nature of the bicyclo[1.1.0]butane scaffold and the substantial $\pi$ character of its central bond enable ring-opening transformations to provide cyclobutanes. In fact, the ability of bicyclo[1.1.0]butanes to undergo facile hydrolysis to cyclobutanol derivatives was recognized soon after the initial discovery of these carbocycles. ${ }^{62 b,}{ }^{79}$ Bicyclo[1.1.0]butanes have been converted into functionalized cyclobutanes through acidic solovolysis, ${ }^{80}$ electrophilic halogenation, ${ }^{81}$ and photochemical irradiation. ${ }^{82}$ These reactions provide an intriguing alternative to $[2+2]$ cyclizations for the selective synthesis of 1,3-difunctionalized cyclobutane derivatives.

The importance of these studies notwithstanding, the addition of carbon-based nucleophiles through an anionic pathway provides a more versatile and synthetically useful method for the synthesis of cyclobutanes from bicyclo[1.1.0]butanes. The high $\pi$ character of the strained central bond enables bicyclo[1.1.0]butanes to be functional equivalents of Michael acceptors, especially when substituted with an electron withdrawing group at the ring junction. Gaoni and co-workers were the first to study this mode of reactivity by investigating the addition of organocuprates to sulfones such as 2-9 (Scheme 9). ${ }^{83}$ These additions were found to be quite facile, but the cyclobutane products such as $\mathbf{2 - 1 0}$ were typically formed as inseparable mixtures of cis/trans-diastereomers. ${ }^{84}$ However, the researchers later designed a series of constrained polycyclic bicyclo[1.1.0]butane derivatives that underwent diastereoselective organocuprate additions. ${ }^{83 b}, 83 \mathrm{c}$ This synthetic method was also applied to the total synthesis of citrilol acetate, a cyclobutane-containing monoterpene natural product. ${ }^{83 b}$


Scheme 9. Nucleophilic addition to bicyclo[1.1.0]butane sulfones developed by Gaoni et al.
Fox and co-workers have developed a tandem approach to functionalized cyclobutanes through an enantioselective bicyclo[1.1.0]butane formation and nucleophilic addition sequence. ${ }^{65 \mathrm{a}}$ Upon treatment of diazo ester 2-11 with a chiral rhodium carboxylate complex, bicyclo[1.1.0]butane ester 2-12 is formed with good enantioselectivity (typically $>90 \%$ ee, Scheme 10). The ester can be either isolated or subjected to an in situ nucleophilic attack by an organocuprate. Unfortunately, the nucleophilic addition proceeds with negligible diastereoselectivity. An epimerization method was therefore developed whereby 2-13 can be
deprotonated and treated with a bulky proton source (BHT) to provide the desired cyclobutanes with $>20: 1$ diastereoselectivity.


Scheme 10. Enantioselective bicyclo[1.1.0]butane synthesis/nucleophilic addition sequence.
Until recently, the nucleophilic addition of heteroatomic nucleophiles to bicyclo[1.1.0]butanes was typically limited to solvolysis-type reactions. ${ }^{80}$ However, Baran and coworkers realized the potential of bicyclo[1.1.0]butyl sulfones (along with related strained molecules such as aza-bicyclo[1.1.0]butane and [1.1.1]propellane) to enable the late-stage derivation of biologically relevant amines, which is a synthetic paradigm they dubbed "strainrelease amination" (Scheme 11). ${ }^{85}$ Upon heating a secondary amine with sulfone 2-14, for example, an excellent yield of aminocyclobutane 2-15 was obtained. Various sulfones were screened in this transformation, and it was found that electron-deficient sulfones such as the difluorophenyl derivative 2-14 are necessary to promote efficient nucleophilic attack of the amine. This addition lacked diastereoselectivity. However, this was inconsequential as the sulfonyl group was typically removed under reductive conditions to provide the cyclobutyl amine (2-16). This method was successfully used for many amine substrates, as well as for the selective cyclobutylation of cysteine residues of peptides.


Scheme 11. "Strain-release amination" protocol developed by Baran et al.

### 2.1.3 Cyclobutyl Phosphines: Synthesis and Utility

Phosphines have played an indispensable role in the development of organometallic chemistry and asymmetric catalysis. Many of the most frequently employed phosphines are bidentate derivatives that are supported by a chiral backbone, such as BINAP. ${ }^{86}$ The ability of chiral, monodentate phosphines to effectively moderate catalytic transformations has more recently come to be appreciated. ${ }^{87}$ Perhaps the most widely employed of the monodentate phosphines are the Buchwald biaryl ligands. Extensive structural analyses of this ligand class have revealed that steric bulk at phosphorus, often provided through cyclohexyl phosphine substituents, is essential for their effective performance in Suzuki-Miyaura cross-couplings. ${ }^{88}$ Therefore, the design of novel phosphines with bulky, cyclic substituents may give rise to ligands that are useful for transitionmetal mediated transformations.

A cyclobutane is an intriguing substituent to consider for the development of phosphine ligands. Aside from providing some degree of steric bulk, the cyclobutane architecture allows for the introduction of substituents adjacent to the phosphine center with a well-defined spatial relationship. One of the few phosphine derivatives that is based on this architecture is tetraphosphine 2-17, which was prepared through photochemical [2+2] dimerization of an alkenyl diphosphine. ${ }^{89}$ Treatment of 2-17 with (cylcooctadiene)palladium (II) chloride affords a mixture of complexes 2-18 and 2-19, both of which were studied by X-ray crystallography (Scheme 12). Although the catalytic applications of these tetraphosphines have been limited to CO-ethylene copolymerization reactions, these studies illustrate the potential of cyclobutanes to be effective scaffolds on which to design phosphines with specific structural parameters. Modulating the bite angle of bidentate ligands is known to have a major effect on their performance in catalysis, ${ }^{90}$ thus,
bidentate ligands derived from this underexplored cyclobutane scaffold may find unique catalytic applications in organic synthesis.


Scheme 12. Cyclobutyl tetraphosphines as ligands in palladium complexes.

Aside from this example, few other cyclobutyl phosphines have been prepared to date, and many of them are highly unusual structures that lack functional handles for further diversification. For example, phosphine oxide $\mathbf{2 - 2 0}{ }^{91}$ and phosphine $\mathbf{2 - 2 1}{ }^{92}$ have been prepared from pre-formed phosphinane or carbohydrate scaffolds, respectively (Figure 11). Functionalization of cubane and [1.1.1]-propellane has been exploited to produce phosphines $\mathbf{2 - 2 2},{ }^{93} \mathbf{2 - 2 3},{ }^{94}$ and $\mathbf{2 - 2 4} .{ }^{95}$ Intramolecular cyclization of phosphine borane $\mathbf{2 - 2 5}$ has recently been used to prepare cyclobutylphosphine borane 2-26, but this approach is not amenable to the synthesis of functionalized cyclobutyl phosphines (Scheme 13). ${ }^{96}$ To the best of our knowledge, none of these phosphines based on the cyclobutane architecture have been applied in catalysis.


2-20


2-21


2-22


2-23, $\mathrm{R}=\mathrm{H}$
2-24, $\mathrm{R}=\mathrm{PPh}_{2}$

Figure 11. Examples of cyclobutylphosphines.


Scheme 13. Preparation of cyclobutyl phosphine borane 2-26 through intramolecular cyclization.
As these examples illustrate, a general method to produce functionalized cyclobutylphosphines is lacking, as are studies on their utility in catalytic transformations. This prompted the development of a hydrophosphination reaction of bicyclo[1.1.0]butanes with secondary phosphine boranes to produce cyclobutylphosphines. Unlike existing methods, the preparation of cyclobutylphosphines through a hydrophosphination approach would allow for a modular synthesis and would provide products with functional handles for further diversification. Although hydrophosphination with secondary phosphine boranes has been successfully conducted on carbodiimides ${ }^{31}$ and alkynes, ${ }^{27}$ extension of the method to include strained carbocycles would provide access to novel phosphines that are rich in $\mathrm{sp}^{3}$ carbons. Moreover, while the addition of carbon and nitrogen nucleophiles to bicyclo[1.1.0]butanes have been demonstrated (Section 2.1.2), the development of a method to add phosphorus nucleophiles would further expand the synthetic utility of these strained rings.

### 2.2 RESULTS AND DISCUSSION

### 2.2.1 Hydrophosphination of Bicyclo[1.1.0]butyl Nitriles

Studies on the hydrophosphination of strained rings commenced with a screening of potential reaction substrates. It was initially anticipated that the relief of ring strain would be a driving force to enable the synthesis of a wide variety of phosphine derivatives through hydrophosphination.

Unfortunately, this was not the case. As shown in Figure 12, bicyclo[1.1.0]butyl alcohols (2-27, $\mathbf{2 - 2 8}$ ), methylenecyclopropanes (2-29), cyclopropenes (2-30), bicyclic ethers (2-31), and activated cyclobutanes (2-32) were all inert to the standard hydrophosphination conditions.




Figure 12. Initial screening studies for the hydrophosphination of strained rings.
In contrast, bicyclo[1.1.0]butanes substituted with an electron withdrawing group at the ring junction ( $\mathbf{2} \mathbf{- 3 3}, \mathbf{2 - 3 4}$ ) reacted to afford a complex mixture of products. This result suggests that the "hydrophosphination" reaction is, to some degree, a Michael addition of a deprotonated phosphine borane nucleophile onto the electron deficient central bond of the bicyclo[1.1.0]butane. As previously noted (Section 2.1.1), the central bond of this strained ring has substantial $\pi$ character that presumably facilitates such an addition. This finding is also consistent with previous studies of nucleophilic addition to the central bond of bicyclo[1.1.0]butanes, where sulfone ${ }^{83,85}$ or ester ${ }^{65 a}$ substituents were found to be necessary for effective cyclobutane formation (Section 2.1.2).

Based on this hypothesis, a nitrile was selected as the optimal activating group to promote the hydrophosphination. Unlike aldehydes (2-33) or primary amides (2-34), nitriles are not affected by deprotonated secondary phosphine boranes. The bicyclo[1.1.0]butyl nitrile of choice
for these studies was 2-37, which can be efficiently accessed in two steps from commercially available methylenecyclobutane 2-35 (Scheme 14). ${ }^{80}$ Using this protocol, 2-35 was subjected to HBr addition to afford 2-36 as a mixture of diastereomers. Treatment of 2-36 with NaH in THF at reflux affords bicyclobutane 2-37 as a liquid that can be purified by bulb-to-bulb distillation. In this manner, nitrile 2-37 was prepared on decagram scale. Differential scanning calorimetry studies of 2-37 showed that this strained molecule is a moderate explosion hazard upon heating, with a decomposition energy of $1600 \mathrm{~J} / \mathrm{g}$ at an onset temperature of $\sim 100^{\circ} \mathrm{C}$. Nitrile $\mathbf{2 - 3 7}$ was therefore stored in a $-20^{\circ} \mathrm{C}$ freezer and handled with appropriate personal protective equipment.


Scheme 14. Synthesis of bicyclo[1.1.0]butyl nitrile 2-37.
When 2-37 was subjected to the previously developed hydrophosphination conditions with dicyclohexylphosphine borane, $\mathrm{a} \sim 2: 1$ mixture of cyclobutane diastereomers cis-2-38 and trans-2-38 was obtained upon workup (Scheme 15). ${ }^{97}$ Notably, and in contrast to previous bicyclo[1.1.0]butane nucleophilic addition studies, these two phosphine borane diastereomers had significantly distinct $\mathrm{R}_{\mathrm{f}}$ values on $\mathrm{SiO}_{2}$ and could be readily separated by chromatography.

2-37

cis-2-38 49\%
NC,

trans-2-38, 15\%

Scheme 15. Hydrophosphination of nitrile 2-37 with dicyclohexylphosphine borane.
The identity of the major cis-diastereomer was assigned by NOESY (Figure 13). The characteristic correlation between the methyl doublet at $1.52 \mathrm{ppm}\left({ }^{3} J_{\mathrm{HP}}=9.2 \mathrm{~Hz}\right)$ and the methine at 3.39 ppm , along with the correlation of both peaks to a single methylene at 2.05 ppm , provides
strong spectroscopic evidence for this assignment of relative stereochemistry. This assignment was later confirmed by obtaining an X-ray crystal structure of a cis-cyclobutane derivative (Section 2.2.3).


Figure 13. NOESY spectrum of cis-2-38.
An isomerization study was conducted to investigate the nature of the diastereoselectivity in the transformation. When either pure diastereomer of $\mathbf{2 - 3 8}$ was subjected to the reaction conditions (1.2 eq NaH in DMAc), HPLC analysis of a quenched aliquot showed formation of the original 2:1 mixture of isomers (Scheme 16). This suggests that the diastereoselectivity is thermodynamic in origin. Therefore, a higher yield of the cis-isomer (which is potentially more valuable for the preparation of bidentate phosphine ligands) can be achieved by iterative "recycling" of the minor trans-isomer through isomerization and purification by chromatography on $\mathrm{SiO}_{2}$.


Scheme 16. Isomerization studies of 2-38.
The hydrophosphination of nitrile 2-37 was conducted with various phosphine boranes (Table 11). A variety of aliphatic and aromatic phosphine boranes were effective in the transformation, each giving a $\sim 2: 1$ mixture of readily separable diastereomers. A generally consistent yield was obtained for all the substrates except for furyl derivative 2-41. In each case, the identity of the major cis-isomer was confirmed by NOESY and by the similarity of the ${ }^{1} \mathrm{H}$ NMR shifts to other analogs.

Table 11. Hydrophosphinations of nitrile 2-37.


| $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | product | cis yield (\%) | trans yield (\%) |
| :---: | :---: | :---: | :---: | :---: |
| Cy | Cy | $\mathbf{2 - 3 8}$ | 49 | 15 |
| $t-\mathrm{Bu}$ | $t$ - Bu | $\mathbf{2 - 3 9}$ | 57 | 26 |
| Cy | Ph | $\mathbf{2 - 4 0}$ | 47 | 23 |
| $n-\mathrm{Bu}$ | 2 -furyl | $\mathbf{2 - 4 1}$ | 28 | 13 |
| $4-(\mathrm{Me}) \mathrm{C}_{6} \mathrm{H}_{4}$ | $4-(\mathrm{Me}) \mathrm{C}_{6} \mathrm{H}_{4}$ | $\mathbf{2 - 4 2}$ | 48 | 28 |
| $3,5-(\mathrm{Me}) \mathrm{C}_{6} \mathrm{H}_{3}$ | $3,5-(\mathrm{Me}) \mathrm{C}_{6} \mathrm{H}_{3}$ | $\mathbf{2 - 4 3}$ | 48 | 24 |
| $4-(\mathrm{MeO}) \mathrm{C}_{6} \mathrm{H}_{4}$ | $4-(\mathrm{MeO}) \mathrm{C}_{6} \mathrm{H}_{4}$ | $\mathbf{2 - 4 4}$ | 43 | 16 |
| $4-(\mathrm{Ph}) \mathrm{C}_{6} \mathrm{H}_{4}$ | $4-(\mathrm{Ph}) \mathrm{C}_{6} \mathrm{H}_{4}$ | $\mathbf{2 - 4 5}$ | 42 | 22 |

A chiral cyclobutylphosphine borane was also prepared using this method. Secondary phosphine borane 2-46 (which was prepared by collaborators at Boehringer Ingelheim over several steps from naproxen) was used to prepare cyclobutane 2-47 in the typical $\sim 2: 1$ ratio of readily
separable diastereomers (Scheme 17). The hydrophosphination was also attempted with chiral phospholane borane 1-58. However, unlike in the previous carbodiimide studies, the phospholane underwent epimerization under the reaction conditions to afford a mixture of 4 products (cis-and trans-cyclobutane isomers, each with a pair of C2-symmetric or meso phospholanes).




Scheme 17. Preparation of chiral phosphine borane 2-47.

### 2.2.2 Preparation and Hydrophosphination of other Bicyclo[1.1.0]butane Nitriles

Nitrile 2-37 was suitably reactive in hydrophosphination reactions, convenient to prepare, and stable upon storage. Despite these advantages, the exclusive use of 2-37 as the hydrophosphination substrate only provides access to phosphines with a methyl substituent on the cyclobutane. Access to bicyclo[1.1.0]butyl nitriles with other substituents at the ring junction would enable the preparation of a wider range of cyclobutyl phosphines. However, published methods for the synthesis of bicyclo[1.1.0]butyl nitriles require multiple steps and lack generality. ${ }^{98}$

A more versatile approach to bicyclo[1.1.0]butane nitriles was envisioned to proceed through the direct installation of a nitrile onto a bicyclo[1.1.0]butane scaffold rather than a ring contraction of a cyclobutylnitrile. In practice, this could be accomplished by trapping a bicyclo[1.1.0]butyllithium species with an electrophilic " $\mathrm{CN}^{+}$" source rather than an imine or
aldehyde (the method commonly used for the synthesis of bicyclo[1.1.0]butyl amines and alcohols by Wipf and co-workers). ${ }^{73 a}$ Coincidentally, our collaborators at Boehringer Ingelheim had recently developed a convenient method that could enable such a transformation. As part of a research effort to develop new methods for the process-scale preparation of aryl nitriles, it was found that dimethylmalononitrile is a safe and inexpensive reagent for the electrophilic cyanation of aryl lithiums and Grignard reagents. ${ }^{99}$ The cyanation reaction is proposed to proceed through the addition of an organometallic reagent to this dinitrile reagent followed by a retro-Thorpe collapse of the anionic intermediate to afford the desired aryl nitrile.

When dibromocyclopropane 2-48 was subjected to the previously reported dilithiation sequence, ${ }^{66 a, ~ 73 b}$ the resulting anion 2-49 underwent this "transnitrilation" to afford nitrile 2-50 (Scheme 18). The analogous bicyclo[1.1.0] butyllithium species $\mathbf{2 - 5 2}$ also underwent this transformation to provide nitrile $\mathbf{2 - 5 3}$, albeit in a low yield of $18 \%$. This poor yield may be attributed to the lack of an aryl substituent to provide conjugative stabilization of the central bicyclo[1.1.0]butane bond. Nonetheless, this approach provided efficient access to these alternative bicyclo[1.1.0]butane nitriles for hydrophosphination studies.



Scheme 18. Preparation of bicyclobutane nitriles 2-50 and 2-53.
The hydrophosphination of nitriles 2-50 and 2-53 was conducted with dicyclohexyl and di-p-tolyl phosphine boranes using the established protocol. ${ }^{97}$ Hydrophosphinations of 2-50 afforded
a nearly 1:1 mixture of diastereomers that were readily separable on $\mathrm{SiO}_{2}$ (Table 12). In contrast, hydrophosphinations of unsubstituted nitrile 2-53 afforded only the cis-cyclobutane products, albeit in modest yield (Table 13). These lower yields may be a consequence of the somewhat unstable nature of 2-53, which showed significant decomposition and polymerization after several days at $-20^{\circ} \mathrm{C}$.

Table 12. Hydrophosphination of bicyclobutane nitrile 2-50.


Table 13. Hydrophosphination of bicyclobutane nitrile 2-53.


These studies suggest that the ratio of cis/trans-isomers obtained through this hydrophosphination is heavily influenced by the substituent at the bicyclo[1.1.0]butane ring junction. This can be rationalized in light of the extensive studies on the stereochemical course of nucleophilic additions to bicyclo[1.1.0]butanes conducted by Hoz and co-workers. ${ }^{100}$ Both theoretical and experimental evidence suggests that upon equatorial attack by a nucleophile, the incipient nitrile anion will adopt the thermodynamically less favored tetrahedral conformation wherein the nitrile is trans to the nucleophile. ${ }^{100 \mathrm{~b}}$ The anion will equilibrate between two
diastereomeric species prior to workup, presumably through the intermediacy of a planar keteneimine tautomer. ${ }^{100 a}$ The isomerization study shown in Scheme 16 demonstrates that the thermodynamically favored cis- isomer, wherein both substituents sit in equatorial positions, is preferentially formed under thermodynamic equilibration (Figure 14). As the $A$-value of the $\mathrm{R}^{2}$ substituent increases from 1.7 (methyl) to 3 (phenyl), the thermodynamic differentiation between the two anionic forms is less pronounced. Thus, a mixture of diastereomers is obtained upon aqueous workup of the reaction.


Figure 14. Explanation for the diastereoselectivity observed in the hydrophosphination of bicyclo[1.1.0]butyl nitriles.

### 2.2.3 Utility of Cyclobutylphosphine Products

An advantage of using nitriles as hydrophosphination substrates is the potential to diversify the cyclobutyl phosphine products. This strategy can be leveraged to enable the synthesis of an array of diversified phosphines. Attempts were made to enact this strategy by converting the cyclobutylnitrile to an imidazoline through a Pinner reaction. This would provide access to cyclobutyl analogs of the bidentate phosphine/imidazoline "BIPI" ligands, ${ }^{101}$ but unfortunately, this effort was unsuccessful. However, the cyclobutyl nitriles could be readily be reduced. A two-
step reduction of 2-42 and 2-43 using DIBAL and $\mathrm{NaBH}_{4}$ afforded the corresponding primary alcohols 2-60 and 2-61 (Scheme 19).


Scheme 19. Reduction of cyclobutyl nitriles.
The alcohols 2-60 and 2-61 are versatile precursors of bidentate ligands. For example, the primary alcohol could be treated with a $(\mathrm{RO})_{2} \mathrm{PCl}$ reagent to produce a phosphine/phosphite bidentate system. This class of ligands has been used in asymmetric catalysis, most frequently in asymmetric hydrogenations and hydroformylations. ${ }^{102}$ To prepare this type of ligand on the ciscyclobutane scaffold, phosphine borane 2-60 was first deprotected using the standard DABCO method. The free phosphine was then subjected to the chlorophosphite derived from $\mathrm{PCl}_{3}$ and $S$ BINOL to afford ligand 2-63 (Scheme 20).


Scheme 20. Synthesis of phosphine/phosphite 2-63.
When ligand 2-63 was treated with $\left[\mathrm{Rh}(\mathrm{COD}) \mathrm{Cl}_{2}\right.$ in $\mathrm{DCE}-\mathrm{d}_{4}$, evidence of bidentate Rh binding was observed by ${ }^{31} \mathrm{P}$ NMR: the phosphite shifted upfield from 144.9 to 134.7 ppm and a $J_{\mathrm{Rh}-\mathrm{P}}$ of 262 Hz was observed, while the phosphine shifted downfield from 11.2 to 45.7 ppm and a $J_{\text {Rh-P }}$ of 151 Hz was observed. Encouraged by this data, 2-63 was applied in several asymmetric catalytic transformations. The asymmetric hydrogenation of dehydrophenylalanine derivative 2-

64 with ligand 2-63 and a cationic rhodium source provided the unnatural enantiomer of N phenylalanine methyl ester (2-65) in 91\% ee (Scheme 21). The absolute configuration of 2-65 was confirmed by optical rotation, and the ee was determined using chiral SFC. Unfortunately, ligand 2-63 did not provide high reactivity or selectivity in the asymmetric hydrogenation of more challenging alkene substrates or in the asymmetric hydrocyanation of alkenes.


Scheme 21. Application of 2-63 in enantioselective hydrogenation.
The preparation of diphosphines from alcohol 2-61 was also explored through formation of a sulfonate ester and treatment with a phosphine borane nucleophile. While the mesylation of alcohol 2-61 was effective, the resulting mesylate was an insufficiently reactive electrophile. The more reactive $p$-bromosulfonate of $\mathbf{2 - 6 1}$ was thus prepared (Scheme 22). Treatment of the sulfonate with a diaryl phosphine borane and base afforded diphosphine borane 2-66. The structure of this diphosphine borane was secured by X-ray crystallography, which confirms the assignment of relative stereochemistry on the cyclobutane (which was previously based on NOESY).





Scheme 22. Synthesis and X-ray crystal structure of diphosphine borane 2-66.
The application of 2-66 and monophosphines such as 2-42 and 2-43 (as $\mathrm{HBF}_{4}$ salts) in catalysis proved difficult. For example, application of these ligands in challenging $\mathrm{C}-\mathrm{H}$ activations resulted in low conversion and/or extensive substrate decomposition.

While these cyclobutyl phosphines were effective some Suzuki cross-couplings, the yields of the reactions were no better than those employing other more accessible ligands. ${ }^{104}$ Future studies should be conducted to identify transformations where the structural rigidity of this cyclobutylphosphine scaffold would be particularly advantageous in facilitating catalysis.

### 2.2.4 Addition of Thiols to Bicyclo[1.1.0]butanes

The success of phosphine borane additions to bicyclo[1.1.0]butyl nitriles raised the question of whether thiols, which are similarly good nucleophiles, could add to these strained rings to afford 1,3-difunctionalized cyclobutyl thioethers. Other approaches to cyclobutyl thioethers and their oxidized analogs (sulfoxides and sulfones) include [2+2] cycloaddtions, ${ }^{105}$ additions of thiol nucleophiles to cyclobutyl bromides, ${ }^{106}$ or additions to preformed sulfonyl bicyclo[1.1.0]butanes. ${ }^{107}$ In contrast, the nucleophilic addition of thiols to readily obtained bicyclo[1.1.0]butyl nitriles offers a straightforward access to versatile 1,3-difunctionalized cyclobutyl thioethers. Moreover, this operationally simple approach is a viable alternative to photochemical irradiation, which has previously been used to add thiols across bicyclo[1.1.0]butanes. ${ }^{82}$

Interestingly, early reports on the reactivity of nitrile 2-37 noted that hydrolysis and aminolysis reactions of the central bond occur with complete trans-diastereoselectivity. ${ }^{108}$ This selectivity was also observed when thiophenol was added to $\mathbf{2 - 3 7}$ in methanol. Experimental and theoretical studies of this addition concluded that the trans diastereomer is kinetically favored, and fast protonation of the anionic intermediate in protic solvents selectively provides the trans product. ${ }^{100 b}$

Despite these insightful mechanistic studies, the addition of thiols to bicyclobutanes appears to have gained limited traction from a preparative standpoint. No systematic study has investigated the scope, yield, and selectivity for the addition of a range of thiophenol nucleophiles, nor have additions of heterocyclic thiols been studied. We were therefore encouraged to develop this addition as a method to access stereodefined cyclobutylthioethers as potentially useful synthetic building blocks. There is also an increasing interest in the development of small molecules that are capable of selective covalent modification of cysteine residues in enzyme active sites. ${ }^{109}$ Bicyclo[1.1.0] butanes are bench-stable, strain activated electrophiles that are well suited for this purpose. In fact, the bioconjugation of peptides with a sulfonyl bicyclo[1.1.0]butane bearing an ${ }^{131}$ I radiolabel has quite recently been described. ${ }^{110}$ Further investigation of the scope and parameters of thiol additions to bicyclo[1.1.0]butanes will enable a better understanding of the potential of these strained rings to be used as tools for medicinal chemistry and chemical biology applications.

In contrast to the studies with phosphine boranes, treatment of a mixture of thiophenol and 2-37 with NaH in DMAc afforded only a modest yield of adduct 2-67 with low diastereoselectivity. However, in agreement with previous studies, ${ }^{100 b}$ treatment of the mixture with $\mathrm{K}_{2} \mathrm{CO}_{3}$ at room temperature cleanly afforded the desired cyclobutane with high trans-diastereoselectivity. A series of thiophenol derivatives were then employed in the reaction, all of which reacted with similarly high yield and diastereoselectivity (Table 14). Like the cyclobutylphosphine boranes, the mixture of diastereomers obtained upon workup were quite separable by chromatography on $\mathrm{SiO}_{2}$.

Table 14. Addition of thiophenol derivatives to bicyclobutane 2-37.


Various heterocyclic thiols were tested in the addition reaction (Table 15). Except for an electron-rich indole-based thiol (entry 5), these substrates required heating and longer reaction times (usually 12-24 h) to reach completion. The yields of these transformations were slightly lower than those with thiophenols, however, the trans-diastereoselectivity remained high. An exception to this trend was the poorly nucleophilic tetrazole-based thiol (entry 4). In this case, a non-nucleophilic alcohol solvent $\left(\mathrm{CF}_{3} \mathrm{CH}_{2} \mathrm{OH}\right)$ was necessary to prevent degradation of the tetrazole product 2-75.

The fact that bicyclo[1.1.0]butyl nitriles only react with heterocyclic thiols at elevated temperature suggests that they may be stable and selective bioconjugation agents. These studies, along with the related bioconjugation studies by Baran and co-workers, ${ }^{85}$ bodes well for the potential of bicyclo[1.1.0]butanes as components of targeted covalent inhibitors that are not prone to attack by glutathione and other off-target nucleophiles.

Table 15. Addition of heterocyclic thiols to bicyclobutane 2-37.


### 2.3 CONCLUSIONS

The ability of bicyclo[1.1.0]butyl nitriles to undergo addition by phosphorous and sulfur nucleophiles has been demonstrated. Most these studies were conducted with methyl-substituted bicyclo[1.1.0]butyl nitrile 2-37. To expand the reaction scope, the transnitrilation method developed by Reeves and co-workers ${ }^{99}$ was applied to access bicyclo[1.1.0]butyl nitriles 2-50 and 2-53, and the previously developed hydrophosphination conditions were used to hydrophosphinate these substrates. Modifications of the resulting cyclobutyl nitrile products were conducted, which provided access to phosphine/phosphite 2-63 and diphosphine borane 2-66. The addition of
thiophenols and heterocyclic thiols to 2-37 was also investigated. These transformations were found to proceed with high trans-diastereoselectivity.

### 3.0 FLUORINATION OF BICYCLO[1.1.0]BUTANES

### 3.1 INTRODUCTION

### 3.1.1 Overview of Selective Monofluorination in Organic Chemistry

With a bond dissociation energy of $105 \mathrm{kcal} / \mathrm{mol}$, the carbon-fluorine bond is the strongest of those commonly encountered in organic chemistry. Fluorine is also the most electronegative of the elements, therefore, $\mathrm{C}-\mathrm{F}$ bonds can induce a substantial dipole moment in fluorinated molecules. However, the poor polarizability of the fluorine atom renders these strong bonds inert in many transformations such as oxidative addition and nucleophilic substitution. This feature, along with the similarity of the $\mathrm{C}-\mathrm{O}$ and $\mathrm{C}-\mathrm{F}$ bond lengths ( $1.43 \AA$ and $1.35 \AA$, respectively) and atomic radii ( $1.52 \AA$ and $1.47 \AA$, respectively), make the C-F bond an effective isostere of ethers and other polar functional groups. ${ }^{111}$ Strategic incorporation of C-F bonds in organic molecules has become a popular way to modulate the conformation, solubility, reactivity, and intermolecular interactions of molecules. Consequently, fluorinated molecules are well represented in the pharmaceutical, agrochemical, and material industries. ${ }^{112}$

Due to this demand, the development of methods to selectively fluorinate organic molecules has been a subject of intensive research efforts. ${ }^{113}$ These methods can be broadly divided into three categories: electrophilic fluorination, where a carbon nucleophile attacks an $\mathrm{F}^{+}$ equivalent; radical fluorination, where carbon and fluorine radicals join; and nucleophilic fluorination, where an $\mathrm{F}^{-}$source attacks an organic molecule, often with the aid of $\mathrm{H}^{+}$or another activator (Figure 15).

Electrophilic fluorination

Radical fluorination



Nucleophilic fluorination


Figure 15. General overview of fluorination methods.
Electrophilic fluorination is very commonly employed. ${ }^{114}$ Historically, $\mathrm{F}^{+}$sources such as $\mathrm{F}_{2}$ gas, $\mathrm{CF}_{3} \mathrm{OF}$, and $\mathrm{XeF}_{2}$ have been used in these transformations, although the drawbacks associated with their handling $\left(\mathrm{F}_{2}, \mathrm{CF}_{3} \mathrm{OF}\right)$ and expense $\left(\mathrm{XeF}_{2}\right)$ have limited their widespread adoption. Most electrophilic fluorinations are now conducted with N-F reagents such as Selectfluor ${ }^{\circledR}$, NFSI, and fluoropyridinium salts. A wide range of substrates have been employed in electrophilic fluorination, including alkenes, arenes, heterocycles, glycals, and enol ethers. ${ }^{114}$ The wide scope of electrophilic fluorination with N-F reagents enables access to fluorinated organic molecules for a variety of applications.

Radical fluorination is a less developed methodology. Recent studies by Boger ${ }^{115}$ and Hiroya ${ }^{116}$ have demonstrated that N-F reagents can be used to conduct radical fluorinations of alkenes in the presence of an iron or cobalt catalyst. Radial fluorination has also been investigated in the context of the burgeoning field of photoredox catalysis. For example, MacMillan and coworkers have developed a photocatalytic method for the conversion of carboxylic acids to the corresponding alkyl fluorides. ${ }^{117}$ The activation of $\mathrm{SF}_{6}$ gas through photoredox catalysis has also been recently used as a method for alcohol deoxyfluorination. ${ }^{118}$

Nucleophilic fluorination is a well-studied field that has several variations. Direct $\mathrm{S}_{\mathrm{N}} 2$ reactions with inexpensive and easily handled fluoride salts is an ideal approach for fluorine introduction. However, the poor solubility and low nucleophilicty of $\mathrm{F}^{-}(\sim 1600$-fold less reactive toward $\mathrm{CH}_{3} \mathrm{I}$ than $\mathrm{Br}^{-}$in $\mathrm{MeOH}^{119}$ ) make this approach less effective. In certain cases, anhydrous
tetraalkylammonium fluorides can affect the nucleophilic fluorination of good electrophiles, but the generality of this transformation is limited. ${ }^{120}$

A more effective approach to nucleophilic fluorination is the use of HF to activate a Lewis basic substrate, which is then attacked by $\mathrm{F}^{-}$to provide a fluorinated product. Anhydrous HF has been used for this purpose, but its low boiling point $\left(19.6^{\circ} \mathrm{C}\right)$ and extreme toxicity significantly deter its use. In the 1970's, Olah and co-workers discovered that HF can be dissolved in pyridine to provide poly(hydrogen fluoride, pyridine $\cdot 9 \mathrm{HF}$ ), a much more easily handled source of HF that is now widely known as Olah's reagent. ${ }^{121}$ Pyridine • 9 HF can effectively fluorinate alkenes, alcohols, and isocyanates, among other functional groups. Alternative reagents have been developed where HF is dissolved in solvents such as DPMU. ${ }^{122}$ Quite recently, it has been reported that HF can be condensed with solid $\mathrm{KHSO}_{4}$ to yield a stable liquid HF reagent. ${ }^{123}$ These alternative HF reagents are reported to be superior to Olah's reagent in some nucleophilic fluorinations.

The concept of using a single reagent to both activate a substrate and conduct a nucleophilic attack with fluoride is also the basis of the deoxyfluorination of alcohols, which is a widely popular fluorination method. The most commonly employed reagents for deoxyfluorination are sulfur trifluoride reagents such as diethylaminosulfur trifluoride (DAST) and the closely related Deoxofluor ${ }^{\circledR} .{ }^{124}$ Despite their utility, the toxicity and explosive potential of these reagents have prompted the development of innocuous and bench stable deoxyfluorination reagents. Examples of deoxyfluorination reagents that have recently gained popularity include the pyridine sulfonyl fluoride (PyFluor) developed by Doyle and co-workers, ${ }^{125}$ the fluoroimidiazolium salts (PhenoFluor ${ }^{\text {TM } / A l k y l F l u o r ~}{ }^{\text {TM }}$ ) developed by Ritter and co-workers, ${ }^{126}$ and the bench-stable sulfur trifluoride (Fluolead ${ }^{\text {TM }}$ ) developed by Umemoto and co-workers (Figure 16). ${ }^{127}$


PyFluor


PhenoFluor ${ }^{\text {TM }}$


Fluolead $^{\text {TM }}$

Figure 16. Recently developed deoxyfluorination reagents.
The fluorination of organic molecules through activation by transition metal catalysis is also a growing area of research. ${ }^{128}$ The fluorination of alkenes and alkynes can be enabled by late transition metals such as palladium ${ }^{129}$ or gold. ${ }^{130}$ Palladium catalysis has been used to form allylic fluorides by attack of $\mathrm{F}^{-}$onto $\pi$-allyl complexes. ${ }^{131}$ A few examples of Pd -catalyzed $\mathrm{C}-\mathrm{H}$ fluorination have also been reported, ${ }^{132}$ however, these transformations require directing groups and/or are limited in scope. The development of selective, widely applicable $\mathrm{C}-\mathrm{H}$ fluorination methods (and $\mathrm{C}-\mathrm{H}$ activation methods in general) is a major unsolved challenge in organic synthesis.

### 3.1.2 Fluorinations Involving the Opening of Strained Rings

Although many classes of molecules have been subjected to nucleophilic and electrophilic fluorination reactions, the fluorination of molecules containing strained rings is somewhat underdeveloped. By leveraging the inherent strain energy of small carbocyclic rings and heterocycles, efficient and selective fluorinations can lead to structurally novel fluorinated chemotypes.

Illustrative examples of unique fluorinated products that can be accessed through the fluorination of strained rings of can be found in the work of Roberts and co-workers (Scheme 23). Upon treatment with $\mathrm{Et}_{3} \mathrm{~N} \cdot 3 \mathrm{HF}$, ketone 3-1 undergoes diastereoselective fluorination to provide cyclobutanone 3-2 in good yield. ${ }^{133}$ A similar approach was also used to prepare a fluorinated
analog of 3'-azidothymidine (AZT), the antiretroviral molecule that is one of the great triumphs of $20^{\text {th }}$ century medicinal chemistry. This AZT analog was prepared from ketone 3-3 through a diastereoselective fluorination with $\mathrm{Et}_{3} \mathrm{~N} \cdot 3 \mathrm{HF}$ to afford 3-4. Through a series of steps, 3-4 was converted into the carbocyclic AZT analog 3-5. Unfortunately, 3-5 had much weaker antiretroviral activity than the parent compound (AZT). ${ }^{134}$



Scheme 23. Nucleophilic fluorination of strained cyclopropanes 3-1 and 3-3.
Not all cyclopropanes undergo such facile nucleophilic fluorination. Despite being activated by diester substituents, cyclopropyl lactone 3-6 was inert to fluorination under a variety of standard conditions (Scheme 24). Only upon prolonged heating of 3-6 with excess $\mathrm{Et}_{3} \mathrm{~N} \cdot 3 \mathrm{HF}$ and $\mathrm{FeF}_{3}$ did sufficient conversion to primary fluoride 3-7 occur. This transformation was investigated while developing a process-scale route to the antidiabetic agent carmegliptin, but the harsh nature of this fluorination prompted the researchers to develop an alternative route involving the deoxyfluorination of a primary alcohol. ${ }^{135}$


Scheme 24. Nucleophilic fluorination of lactone 3-6.

Alternatively, cyclopropanes can be subjected to nucleophilic difluorination through an oxidative ring-opening (Scheme 25). ${ }^{136}$ This method was recently developed by Jacobsen and coworkers as an extension of similar oxidative fluorinations of alkenes ${ }^{137}$ and alkenyl lactones. ${ }^{138}$ The transformation is presumed to occur through the catalytic formation of a hypervalent iodonium fluoride, which facilitates ring-opening of the cyclopropane. Variations of the reaction were developed to provide access to other 1,3-difunctionalized products from 3-8.


Scheme 25. Oxidative difluorination of cyclopropanes developed by Jacobsen and co-workers.
Electrophilic N-F reagents have been used to conduct fluorinations of cyclopropyl or cyclobutyl carbinols with concomitant ring expansion (Scheme 26). In the presence of catalytic silver nitrate, Selectfluor ${ }^{\circledR}$ promotes a radical-based ring opening fluorination of cyclopropane 310 to afford ketone 3-11. ${ }^{139}$ Photocatalysis has also been used to conduct similar transformations. ${ }^{140}$ Alexakis and co-workers have used chiral phosphoric acids to prepare enantioenriched fluorinated ketones through this approach. A representative transformation of this type is the conversion of cyclobutane carbinol 3-12 to spirocyclic ketone 3-13. ${ }^{141}$



Scheme 26. Ring expansion of carbinols using electrophilic fluorination.
The ring-opening fluorination of three- and four-membered heterocycles has been explored, and many of these transformations do not require the use of HF complexes (Scheme 27). ${ }^{142}$ For example, the symmetrical epoxide 3-14 undergoes diastereoselective fluorination when treated with excess $\mathrm{BF}_{3} \cdot \mathrm{OEt}_{2}$ at room temperature for an extended time. ${ }^{143}$ Aziridines can be fluorinated under similar conditions, or alternatively, through nucleophilic catalysis by using an acyl fluoride as a fluorine source. This method was used for the conversion of aziridine 3-16 to 317. ${ }^{144}$ Although oxetanes are less strained than their epoxide counterparts, fluorination of oxetane 3-18 was accomplished with $\mathrm{SiF}_{4}$ and $\mathrm{Me}_{2} \mathrm{~S} .{ }^{145}$ In contrast, fluorination of the highly strained azabicyclo[1.1.0]butane heterocycle 3-20 is reported to occur in excellent yield upon treatment with Olah's reagent. ${ }^{146}$ However, no method for the fluorination of the carbocyclic analogs (bicyclo[1.1.0]butanes) has been reported.


Scheme 27. Fluorination of 3- and 4-membered heterocycles.

### 3.1.3 Synthesis and Utility of Fluorinated Cyclobutanes

The introduction of fluorine substituents can modulate the reactivity and conformational preferences of cyclobutanes. ${ }^{147}$ For example, perfluorocyclobutanes and related carbocycles exhibit unique isomerization behavior, and studies of this reactivity have broadened modern understanding of fluorine substitution effects in organic molecules. ${ }^{188}$ In the context of medicinal chemistry, fluorinated carbocycles have been leveraged to improve the potency and pharmacokinetic properties of lead compounds. ${ }^{149}$ The structural rigidity of the cyclobutane scaffold makes fluorinated cyclobutanes particularly intriguing for the exploitation of carbon-fluorine bond interactions with specific residues in enzyme active sites. ${ }^{147}$

Three examples of monofluorinated cyclobutanes that have been developed by pharmaceutical companies for various indications are shown in Scheme 28. The cyclobutyl nucleoside analog 3-24 was developed by Bristol-Myers Squibb as a treatment for herpes infections. ${ }^{150}$ This fluorinated cyclobutane was prepared through a diastereoselective [2+2] cycloaddition between fumarate 3-22 and a fluorinated ketene acetal. However, cycloaddition product 3-23 was obtained in modest yield and with low diastereoselectivity.





Scheme 28. Synthetic approaches to monofluorinated cyclobutanes of medicinal interest.
Deoxyfluorination of cyclobutyl alcohols has more commonly been used to access monofluorinated cyclobutanes. For example, deoxyfluorination of secondary alcohol 3-25 with DAST afforded 3-26 as a single diastereomer in 79\% yield (Scheme 29). This intermediate was carried forward through 5 steps to triazole 3-27, which is a hydroxysteroid dehydrogenase inhibitor for the treatment of cardiovascular ailments. ${ }^{151}$ Deoxyfluorination was also used in the last step of a process-scale route to cyclobutane $\mathbf{3 - 3 0}$, an $\mathrm{H}_{3}$ receptor antagonist developed by Pfizer for the treatment of neurological disorders. ${ }^{152}$

The preparation of fluorinated cyclobutanes through fluorinations that involve the opening of strained rings is a mechanistically intriguing alternative to [2+2] or deoxyfluorination approaches. An early example of such a transformation was reported as a side reaction by Adcock and co-workers. As part of a studies on the isomerization behavior of bicyclo[1.1.1]pentanes, it was found that [1.1.1] propellane (3-31) undergoes rearrangement to 3-
fluoromethylenecyclobutane 3-32 in the presence of $\mathrm{XeF}_{2}$ (Scheme 29). ${ }^{153}$ The transformation is presumably mediated by trace amounts of HF generated under the reaction conditions.


Scheme 29. Rearrangement of [1.1.1]-propellane.
Substituted 3-fluoromethylenecyclobutanes can be prepared through the ring expansion of methylenecyclopropyl alcohols. When treated with nonafluorobutanesulfonyl fluoride, alcohol 333 underwent ring expansion to provide a mixture of fluorinated product 3-34 and ether 3-35 (Scheme 30). ${ }^{154}$ Similarly, alcohol 3-36 underwent DAST-mediated ring expansion to provide fluorocyclobutane 3-37 in good yield. ${ }^{155}$ These transformations are proposed to proceed through formation of a non-classical bicyclobutonium carbocation, ${ }^{156}$ which is then trapped with a fluoride nucleophile to afford the cyclobutane product. The conversion of 3-33 to a cyclobutyl amide has also been accomplished through a Ritter reaction on this putative non-classical carbocation intermediate. ${ }^{157}$


Scheme 30. Expansions of methylenecyclopropanes to fluorinated methylenecyclobutanes.
A series of conceptually similar fluorinations involving cyclopropane ring expansion were developed by Yoshioka and co-workers (Scheme 31). ${ }^{158}$ These transformations are conducted with Olah's reagent that is modified with $\mathrm{KHF}_{2}$ and diisopropyl amine to provide a more nucleophilic
and monomeric source of fluoride. When alcohol 3-38 $(R=H)$ is treated with this reagent, 3-39 is obtained in good yield as a single alkene isomer. ${ }^{159}$ This method provides superior access to the homoallylic fluoride motif, as the direct treatment of homoallylic alcohols with deoxyfluorination reagents is often complicated by side reactions. When derivatives of $\mathbf{3 - 3 8}$ where R is an alkyl or aryl substituent were employed, a ring expansion occurred to provide tertiary fluorocyclobutane 3-40 in good yield with high diastereoselectivity. ${ }^{160}$ Electrophilic ring opening of an epoxide can also be used to trigger the ring expansion and fluorination. This is exemplified by the conversion of 3-41 to primary alcohol 3-42. ${ }^{160}$ An extension of this method was developed that provides access to fluorinated cyclopentenones. ${ }^{161}$ The method also was used as a key step in the total synthesis of a fluorinated analog of grandisol, a cyclobutane-containing monoterpene. ${ }^{162}$


Scheme 31. Fluorinations involving cyclopropane ring expansion developed by Yoshioka et al. ${ }^{158}$

### 3.2 RESULTS AND DISCUSSION

### 3.2.1 Deoxyfluorination of Bicyclo[1.1.0]butyl Alcohols

The successful development of methods to add phosphorus and sulfur nucleophiles to bicyclo[1.1.0]butanes prompted consideration as to whether poor nucleophiles could also be used in these strain-releasing transformations. The addition of fluoride, a prototypical poor nucleophile, would provide an intriguing access to monofluorinated cyclobutane products. However, despite the substantial strain energy of bicyclo[1.1.0]butanes, preliminary attempts to use fluoride salts to fluorinate bicyclobutyl amides, sulfones, and nitriles were unsuccessful. In the representative transformation shown in Scheme 32, potassium fluoride fails to fluorinate amide 3-43 even with extensive heating in the presence of 18 -crown- 6 . When heated above $110{ }^{\circ} \mathrm{C}, \mathbf{3 - 4 3}$ began to decompose.


Scheme 32. Attempted direct fluorination of bicyclo[1.1.0]butyl amide 3-43 with KF/18-crown-6.
Since the direct addition of fluoride salts appeared unfeasible, a fluorination method involving the activation of bicyclo[1.1.0]butyl alcohols was developed. The work of Yoshioka and co-workers on the deoxyfluorination of cyclopropanes ${ }^{158}$ (Scheme 31) was an inspiration toward this end, and the modified reagent that they developed $\left(\mathrm{KHF}_{2} / i-\mathrm{Pr}_{2} \mathrm{NH} /\right.$ pyridine $\left.\cdot 9 \mathrm{HF}\right)$ was expected to fluorinate the structurally related bicyclo[1.1.0]butyl alcohols. Some accounts from the literature also suggested that bicyclo[1.1.0]butyl alcohols could be converted to methylenecyclobutanes through deoxygenation and nucleophilic attack. Wiberg and co-workers
noted in the original full paper on bicyclo[1.1.0]butanes that treatment of bicyclo[1.1.0]butyl methanol with HBr resulted in a rearrangement to form 3-bromomethylenecyclobutane. ${ }^{62 \mathrm{~b}}$ Additionally, Szeimies and co-workers prepared a 3-chloromethylenecyclobutane from the corresponding bicyclo[1.1.0]butyl methanol using $N$-chlorosuccinimide and dimethyl sulfide. ${ }^{163}$

Alcohol 3-45 was successfully converted to methylenecyclobutane 3-46 in moderate yield when subjected to the HF reagent developed by Yoshioka (Table 16). A substantial increase in yield was realized by adding the substrate to the pre-mixed reagent, as opposed to adding reagents to a solution of the substrate (entry 1 vs entry 2 ). The yield was also found to be correlated to the stoichiometry of pyridine $\cdot 9 \mathrm{HF}$ : the yield dropped when less than 50 equivalents of HF was used (entry 1 vs entry 3 ), but no advantage was observed with $>70$ equivalents. The use of a large excess of pyridine $\cdot 9 \mathrm{HF}(100 \mathrm{eq} \mathrm{HF})$ is also required for the difluorination chemistry recently developed by Jacobsen and co-workers. ${ }^{136-138}$ Despite this high stoichiometric excess, only 1.8 mL of pyridine - $9 \mathrm{HF}(1 \mathrm{~mL}$ reagent $=38 \mathrm{mmol} \mathrm{HF})$ is required to fluorinate 1 mmol of a bicyclobutyl alcohol using this newly developed method. While the hazardous and corrosive nature of pyridine $\cdot 9 \mathrm{HF}$ necessitates the use of a plastic reaction vessel (typically a polypropylene syringe sealed on one end), the reagent can be safely quenched upon workup with aqueous NaOH . Unfortunately, less reactive and more easily handled sources of HF such as $\mathrm{Et}_{3} \mathrm{~N} \cdot 3 \mathrm{HF}$ or $\mathrm{KHF}_{2}$ alone did not enable this fluorination (entries 4 and 5).

Standard deoxyfluorination reagents such as DAST and Deoxofluor ${ }^{\circledR}$ also successfully converted alcohol 3-45 to the fluorinated methylenecyclobutane 3-46 (Table 16, entries 6 and 7). However, the yields of these transformations tended to be lower and their purification was often complicated by side products. Sulfonyl fluoride-based deoxyfluorination reagents such as nonfluorobutanesulfonyl fluoride ( NfF ) and PyFluor ${ }^{125}$ provided only modest amounts of 3-46
(entries 8-11). The use of alkali fluoride salts in combination with various Lewis or Bronsted acids was also ineffective (entry 12).

Table 16. Screening of deoxyfluorination conditions.

|  |  |  | Ph |  |
| :---: | :---: | :---: | :---: | :---: |
| entry | conditions | solvent temp | mperature ( ${ }^{\circ} \mathrm{C}$ ) | yield (\%) |
| 1 | $\mathrm{KHF}_{2}(1 \mathrm{eq}), i-\mathrm{Pr}_{2} \mathrm{NH}(4 \mathrm{eq})$, pyridine $\cdot 9 \mathrm{HF}(70 \mathrm{eq} \mathrm{HF})^{\text {a }}$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | -78 | 57 |
| 2 | $\mathrm{KHF}_{2}(1 \mathrm{eq}), i-\mathrm{Pr}_{2} \mathrm{NH}(4 \mathrm{eq})$, pyridine $\cdot 9 \mathrm{HF}$ ( 70 eq HF ) | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | -78 | 38 |
| 3 | $\mathrm{KHF}_{2}(1 \mathrm{eq}), \mathrm{i}-\mathrm{Pr}_{2} \mathrm{NH}(4 \mathrm{eq})$, pyridine $\cdot 9 \mathrm{HF}$ (10.5 eq HF) | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | -78 | 33 |
| 4 | $\mathrm{KHF}_{2}(1 \mathrm{eq}), i-\mathrm{Pr}_{2} \mathrm{NH}(4 \mathrm{eq}), \mathrm{Et}_{3} \mathrm{~N} \cdot 3 \mathrm{HF}(3 \mathrm{eq})$ | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 25 | 0 |
| 5 | $\mathrm{KHF}_{2}$ (1 eq) | DCE | 80 | 0 |
| 6 | DAST (1 eq) | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 0 | $51^{\text {b }}$ |
| 7 | Deoxo-fluor (1 eq) | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 0 | 32 |
| 8 | Fluolead (1 eq) | $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ | 25 | 0 |
| 9 | NfF (1.5 eq), DBU (3 eq) | toluene | 25 | 23 |
| 10 | same as entry 9, w/ CsF (3 eq) | toluene | 25 | 26 |
| 11 | PyFluor (1.1 eq), DBU (2.0 eq) | toluene | 25 | 14 |
| 12 | NaF (3 eq) $\quad \mathrm{CH}_{2} \mathrm{C}$ | $\mathrm{Cl}_{2} / \mathrm{HFIP}$ (10:1) | 1) 25 | 0 |

The conversion of alcohol 3-45 to methylenecyclobutane 3-46 likely proceeds through an acid-mediated dehydration to form a cationic intermediate (Scheme 33). This putative cation is highly stabilized, as it has both benzylic and non-classical bicyclobutonium character. ${ }^{156}$ Evidence for the non-classical character of the bicyclo[1.1.0]butyl-1-carbinyl cation was found through spectroscopic and computational studies by Wiberg and co-workers. ${ }^{164}$ Attack of this stable cationic intermediate by the fluoride nucleophile at the ring junction results in an opening of the bicyclic system to relieve $\sim 17 \mathrm{kcal} / \mathrm{mol}$ of ring strain and provide product 3-46.


Scheme 33. Proposed mechanism for the deoxyfluorination of 3-45.
A mechanistic alternative is that the benzylic cation rearranges to a tertiary 3methylenecyclobutyl cation prior to fluoride attack. However, the computational studies conducted by Wiberg and co-workers at the MP4/6-31G* level of theory suggest that the 3methylenecyclobutyl cation is $16.6 \mathrm{kcal} / \mathrm{mol}$ less stable than the non-classical bicyclo[1.1.0]butyl-1-carbinyl cation. ${ }^{164}$ This bicyclo[1.1.0]butane/methylenecyclobutane isomerization is a rare instance in which one isomeric form (the bicyclobutane) is strongly favored as the cationic intermediate, while the other (the methylenecyclobutane) is preferred as the product.

To evaluate the scope of this deoxyfluorination, a series of bicyclo[1.1.0]butyl alcohols were prepared (Table 17). This was accomplished using the method discovered by Brinker ${ }^{66 a}$ and adapted by Walczak and Wipf, ${ }^{73 b}$ where a trihalocyclopropane is dilithiated and trapped with an aldehyde or ketone electrophile. The $\mathrm{R}^{2}$ substituent on the alcohol products was easily varied by trapping with various benzaldehyde derivatives (entries 1-6). Acetophenone (entry 7) and 3pentanone (entries 16 and 17) were also successfully trapped to provide tertiary alcohols. The $\mathrm{R}^{1}$ substituent was varied by using several trihalocyclopropane precursors, which were in turn prepared through dibromocyclopropanation of the appropriate allylic halides (entries 8-17).

Table 17. Preparation of bicyclo[1.1.0]butyl alcohol substrates.

|  |  | $R^{1}$ <br> X | $\xrightarrow[\mathrm{Et}_{2} \mathrm{O},-78^{\circ} \mathrm{C}]{\substack{\text { 1. } \mathrm{MeLi} \\ \text { 2. } t \text { - } \mathrm{BuLi} \\ \text { 3. } \mathrm{R}^{2} \mathrm{R}^{3} \mathrm{CO}}}$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| entry | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | $\mathrm{R}^{3}$ | X | product | yield (\%) |
| 1 | $\mathrm{CH}_{3}$ | H | Ph | Cl | 3-45 | 80 |
| 2 | $\mathrm{CH}_{3}$ | H | $4-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | Cl | 3-47 | 64 |
| 3 | $\mathrm{CH}_{3}$ | H | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | Cl | 3-48 | 56 |
| 4 | $\mathrm{CH}_{3}$ | H | $4-\mathrm{OMeC}_{6} \mathrm{H}_{4}$ | Cl | 3-49 | 46 |
| 5 | $\mathrm{CH}_{3}$ | H | 2-(allyloxy) $\mathrm{C}_{6} \mathrm{H}_{4}$ | Cl | 3-50 | 36 |
| 6 | $\mathrm{CH}_{3}$ | H | 2-pyridyl | Cl | 3-51 | 49 |
| 7 | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | Ph | Cl | 3-52 | 77 |
| 8 | $n-\mathrm{Pr}$ | H | Ph | Br | 3-53 | 34 |
| 9 | $i-\mathrm{Pr}$ | H | Ph | Br | 3-54 | 40 |
| 10 | $\mathrm{Ph}\left(\mathrm{CH}_{2}\right)_{2}$ | H | Ph | Br | 3-55 | 47 |
| 11 | Ph | H | $n-\mathrm{Bu}$ | Br | 3-56 | 81 |
| 12 | Ph | H | $t$-Bu | Br | 3-57 | 38 |
| 13 | 4- $\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | H | $t$-Bu | Br | 3-58 | 27 |
| 14 | $4-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | H | Ph | Br | 3-59 | 54 |
| 15 | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | H | Ph | Br | 3-60 | 67 |
| 16 | 4- $\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | Et | Et | Br | 3-61 | 63 |
| 17 | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | Et | Et | Br | 3-62 | 71 |

The deoxyfluorination of these bicyclo[1.1.0]butyl alcohols was evaluated using the pyridine $\cdot 9 \mathrm{HF} / \mathrm{KHF}_{2} / i-\mathrm{Pr}_{2} \mathrm{NH}$ reagent (Table 18). The yields for these transformations generally ranged from $40-65 \%$. Tertiary (entries 7,14 , and 15) and secondary alcohols performed equally well in the deoxyfluorination. Although the presence of a benzylic alcohol presumably facilitates the transformation by providing a more stabilized cationic intermediate, benzylic activation is not required as evidenced by the successful preparation of 3-71, albeit with a reduced yield (entry 11). Appended functional groups appeared detrimental to the reaction, as evidenced by the lower yield of allyloxy-substituted derivative 3-66. The only heterocyclic alcohol attempted, pyridine 3-51, afforded a complex mixture under the reaction conditions.

Table 18. Deoxyfluorination of bicyclo[1.1.0]butyl alcohols.

|  |  | $\xrightarrow[\substack{\mathrm{CH}_{2} \mathrm{Cl}_{2} \\-78{ }^{\circ} \mathrm{C}, 10 \mathrm{~min}}]{\substack{\mathrm{KHF}_{2}(1 \mathrm{eq}) \\ i-\mathrm{Pr}_{2} \mathrm{NH}(4 \mathrm{eq})}} \mathrm{HF} \mathrm{\cdot Pyr(50-70eq} \mathrm{HF)}$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
| entry | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | $\mathrm{R}^{3}$ | product | yield (\%) |
| 1 | $\mathrm{CH}_{3}$ | H | Ph | 3-46 | 57 |
| 2 | $\mathrm{CH}_{3}$ | H | $4-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | 3-63 | 46 |
| 3 | $\mathrm{CH}_{3}$ | H | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | 3-64 | 58 |
| 4 | $\mathrm{CH}_{3}$ | H | $4-\mathrm{OMeC}_{6} \mathrm{H}_{4}$ | 3-65 | 46 |
| 5 | $\mathrm{CH}_{3}$ | H | 2-(allyloxy) $\mathrm{C}_{6} \mathrm{H}_{4}$ | 3-66 | 34 |
| 6 | $\mathrm{CH}_{3}$ | H | 2-pyridyl | - | 0 |
| 7 | $\mathrm{CH}_{3}$ | $\mathrm{CH}_{3}$ | Ph | 3-67 | 63 |
| 8 | $n-\mathrm{Pr}$ | H | Ph | 3-68 | 60 |
| 9 | $i-\mathrm{Pr}$ | H | Ph | 3-69 | 47 |
| 10 | $\mathrm{Ph}\left(\mathrm{CH}_{2}\right)_{2}$ | H | Ph | 3-70 | 58 |
| 11 | Ph | H | $n-\mathrm{Bu}$ | 3-71 | 31 |
| 12 | 4-CF3 $\mathrm{C}_{6} \mathrm{H}_{4}$ | H | Ph | 3-72 | 62 |
| 13 | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | H | Ph | 3-73 | 41 |
| 14 | $4-\mathrm{CF}_{3} \mathrm{C}_{6} \mathrm{H}_{4}$ | Et | Et | 3-74 | 51 |
| 15 | $4-\mathrm{ClC}_{6} \mathrm{H}_{4}$ | Et | Et | 3-75 | 48 |

No side products were isolated upon purification of these reactions. However, the fact that some yellow/brown material is typically adsorbed to the base of the $\mathrm{SiO}_{2}$ column suggests that some degree of polymerization or decomposition occurs. Despite moderate yields obtained in these transformations, this method is the first example of a bicyclo[1.1.0]butane fluorination and is one of the only means available for the synthesis of substituted 3-fluoromethylenecyclobutanes. ${ }^{154}$

### 3.2.2 Anomalous Deoxyfluorination Substrates

Not all the bicyclo[1.1.0]butyl alcohols that were tested in the deoxyfluorination reaction provided the expected methylenecyclobutane product. These limitations revealed some interesting aspects of the deoxyfluorination mechanism and the stability of the methylenecyclobutane products. For example, when $t$-butyl-substituted alcohols 3-57 and 3-58 were subjected to the deoxyfluorination conditions, allylic alcohols 3-76 and 3-77 (respectively) were obtained in low yield (Scheme 34).

The structure of these alcohols was assigned using a combination of one- and two-dimensional NMR experiments. It is hypothesized that 3-76 and 3-77 originate from the expected 3methylenecyclobutane, but the unfavorable stearic clash between the $t$-butyl group and the cyclobutane render the products unstable. Upon aqueous workup, hydrolytic alkene transposition would relieve both this interaction and the cyclobutane ring strain to afford the observed allylic alcohol product.


Scheme 34. Deoxyfluorination of 3-57 and 3-58 affords rearranged allylic alcohol products.
The deoxyfluorination was also impacted by substitution at the bicyclo[1.1.0]butane ring junction. As Table 18 shows, a variety of alkyl and aryl substituents are tolerated in the transformation. In contrast, the attempted deoxyfluorination of bicyclobutane 3-78 (which does not bear a substituent at the ring junction) afforded aldehyde 3-79 in low yield (Scheme 35). This aldehyde may arise from protonation of the bicyclo[1.1.0]butane at the less substituted ring junction, which would trigger a semipinicol rearrangement (1,2-arene shift) of the resulting tertiary carbocation to afford 3-79. Although products resulting from alternative reaction pathways are presumably formed, 3-79 was the only tractable product isolated from the crude reaction mixture.


Scheme 35. Attempted deoxyfluorination of bicyclobutanes without a ring junction substituent.

Sulfonyl-substituted bicyclo[1.1.0]butane 3-80 was completely inert to the pyridine $\bullet 9 \mathrm{HF}$ reagent (Scheme 36). When treated with Deoxofluor ${ }^{\circledR}$, however, a deoxfluorination occurred to provide bicyclo[1.1.0]butyl fluoride 3-81 with no formation of a methylenecyclobutane or other rearranged product. The failure of $\mathbf{3 - 8 0}$ to undergo rearrangement supports the hypothesis that the fluorination proceeds through a delocalized carbocation intermediate. The formation of such an intermediate would be strongly disfavored by the strong electron withdrawing capacity of the sulfone at the ring junction.


Scheme 36. Deoxyfluorination of sulfonyl bicyclobutane 3-80.
It was hypothesized that cyclopropenyl carbinols, inasmuch as they are structurally and functionally similar to the bicyclo[1.1.0]butyl alcohols previously studied, would form fluorinated methylenecyclopropanes when subjected to the deoxyfluorination conditions. Encouragingly, cyclopropene 3-82 has been converted to metheylenecyclopropanes through organocuprate addition or [2,3]-sigmatropic rearrangement. ${ }^{165}$ However, when 3-82 was subjected to the deoxyfluorination conditions, allene 3-83 was the only product obtained (Scheme 37). This allene product is presumably formed though activation of the alcohol and attack of the fluoride nucleophile on the cyclopropyl methylene. While the rearrangement of similar cyclopropene carbinols to allenes has been accomplished using perchloric acid ${ }^{166}$ or silica gel, ${ }^{167}$ this is the first example of a rearrangement of this type involving fluorination. A potentially fruitful line of future research would identify divergent conditions that provide selective access to either fluorinated allenes or methylenecyclopropanes with various substitution patterns.


Scheme 37. Deoxyfluorination of cyclopropene alcohol 3-82 provides allene 3-83.

### 3.2.3 Transformations of Fluorinated Methylenecyclobutanes

With a method to access 3-fluoromethylenecyclobutanes established, the utility of these products was explored. A prime objective along these lines was to convert the methylenecyclobutanes into 3-fluorocyclobutanones through oxidative scission of the exocyclic alkene. These cyclobutanone products were envisioned to be versatile precursors to more complex fluorinated carbocycles. Moreover, there are only a few scattered reports of 3-fluorocyclobutanones in the patent literature, and their utility remains unexplored.

A problematic aspect of this oxidative scission was the loss of material due to the generation of volatile, low molecular weight ketones in small quantities. Thus, the larger phenethylamine substituted methylenecyclobutane 3-70 was the primary substrate used in these studies. When subjected to a two-step sequence of dihydroxylation and oxidative diol cleavage, cyclobutanone 3-84 was obtained (Scheme 38). Ketone 3-84 was converted into oxime 3-85, which is a potential precursor of fluorinated cyclobutyl amines ${ }^{168}$ or fluorinated spirocycles through [3+2] dipolar cycloaddition. ${ }^{169}$ Reduction of ketone 3-84 occurred with moderate diastereoselectivity to afford alcohol 3-86 in 66\% yield and a 6.3:1 diastereomeric ratio, and the relative stereochemistry of the major alcohol isomer was tentatively assigned by NOESY.


Scheme 38. Transformations of ketone 3-84.
Unfortunately, ketone 3-84 and analogs thereof appear to be unstable to acid-mediated transformations. When an oxidative alkene cleavage was conducted on n-propyl substituted methylenecyclobutane 3-68, the cyclobutanone product 3-87 was carried directly forward to a reductive amination due to concerns of volatility (Scheme 39). No secondary amine was isolated from this reaction, but rather, the unexpected vinylogous amide product 3-88 was obtained in 34\% yield. Similarly, when ketone 3-84 was subjected to an Ugi multicomponent reaction, vinylogous amide 3-89 was obtained in 47\% yield, with no evidence of Ugi product formation. The structure of these vinylogous amides was assigned by ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, DEPT 135, and HMBC. The connectivity of the molecule was confirmed by analysis of HMBC correlations (Figure 17). The alkene geometry was assigned on the basis of the high $\mathrm{N}-\mathrm{H}{ }^{1} \mathrm{H}$ NMR shift (11.17 ppm) that is indicative of intramolecular hydrogen bonding in these systems. ${ }^{170}$


Scheme 39. Attempted acid-mediated transformations of cyclobutanones result in vinylogous amides.


Figure 17. Key HMBC correlations of 3-88.
The formation of these vinylogous amides likely occurs by enol formation, $4 \pi$ electrocyclic ring opening, and addition/elimination of the amine nucleophile into the $\beta$-fluoroenone (Scheme 40). Alternatively, an initial elimination of fluoride from 3-84 would provide a reactive cyclobuteneone that undergoes nucleophilic attack by the amine, $4 \pi$ electrocyclic ring opening, and tautomerization to afford 3-89.


Scheme 40. Possible mechanism for the conversion of 3-84 to 3-89.
Other transformations were investigated that retain the aryl substituent of the deoxyfluorination products. Bromohydrin formation of 3-63 proceeded with regio- and diastereoselectivity to provide 3-90 as a single regioisomer and as a separable 5.8:1 mixture of diastereomers ( $69 \%$ and $12 \%$ isolated yields, Scheme 41). The structure and relative stereochemistry of 3-90 was assigned using ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, NOESY, and HSQC. The secondary alcohol of the bromohydrin was converted to the corresponding ketone 3-91 in $87 \%$ yield using Dess-Martin periodinane. Attempted Bayer-Villager oxidation of 3-91 was unsuccessful, as this ketone was resilient to $m$-chloroperbenzoic acid and other oxidation conditions.


Scheme 41. Regio- and diastereoselective bromohydrin formation from 3-63.

### 3.2.4 Diastereoselective Fluorination of Bicyclo[1.1.0]butyl amides

With the demonstrated ability of the pyridine - $9 \mathrm{HF} / \mathrm{KHF}_{2} / i-\mathrm{Pr}_{2} \mathrm{NH}$ reagent to conduct deoxyfluorination reactions, the fluorination of other bicyclo[1.1.0]butane derivatives was considered. Specifically, a bicyclo[1.1.0]butane substituted with a Lewis basic functional group such as an amide was expected to undergo a ring-opening fluorination when treated with this HF reagent. This transformation would provide access to fluorinated cyclobutyl amides, which are useful structures in medicinal chemistry and other applications (Section 3.1.3).

To test this hypothesis, a series of bicyclo[1.1.0]butyl amides were prepared through dilithiation of the appropriate trihalocyclopropane precursor and trapping with an isocyanate (Table 19). The yields of isocyanate trapping tend to be lower than those of aldehyde or imine trapping. In the case where $\mathrm{R}^{1}$ was an $n$-propyl substituent, the yield was poor (entry 4). Nonetheless, amides 3-92-3-96 were obtained in sufficient quantity to test their feasibility in fluorination reactions.

Table 19. Preparation of bicyclo[1.1.0] butyl amides.


Treatment of the bicyclo[1.1.0]butyl amides with the modified pyridine • 9 HF reagent afforded 3-fluoro-3-alkylcyclobutyl amides (Table 20). The reaction tended to be slower than the alcohol deoxyfluorination, thus, an additional time period at $0{ }^{\circ} \mathrm{C}$ was added to ensure full conversion. The products were obtained as single regioisomers, and the diastereoselectivity of the transformation was also high. In cases where $\mathrm{R}^{1}=\mathrm{CH}_{3}$ (entries $1-3$ ) only one diastereomer (with the methyl and amide in a cis relationship) was isolated after chromatography. Larger substituents at the ring junction, however, resulted in attenuated diastereoselectivity. This was most pronounced in the fluorination of 3-96 $\left(\mathrm{R}^{1}=\mathrm{Ph}\right)$, which yielded $\mathbf{3 - 1 0 1}$ as separable diastereomers in isolated yields of 39 and $25 \%$. The poor diastereoselectivity in this case may be the consequence of the more planar stabilized benzylic cation that is a putative intermediate in the conversion of 396 to 3-101.

Table 20. Diastereoselective fluorination of bicyclo[1.1.0]butyl amides.


The relative stereochemistry of each of the major amide isomers was assigned by NOESY (in 3-97, for example, the methyl and methine protons each correlate to methylene protons on opposite faces of the cyclobutane). To confirm these assignments, a single crystal of amide 3-98 was grown and analyzed by X-ray crystallography. The crystal structure obtained for 3-98 confirms the regio- and stereochemical assignments of the cyclobutane products (Figure 18).



Figure 18. X-ray crystal structure of amide 3-98.
As observed in bicyclo[1.1.0]butyl alcohol deoxyfluorinations, attempted fluorination of a bicyclobutane derivative without a ring junction substituent resulted in an unexpected product. When amide 3-43 was treated under the typical reaction conditions, attack of the fluoride nucleophile at the bicyclobutane methylene afforded racemic cyclopropane 3-102 as the major product in 32\% yield (Scheme 42). An inseparable mixture of byproducts was also obtained upon purification, a component of which appears to be the expected 3-fluorocyclobutyl amide. While the acidic methanolysis of bicyclo[1.1.0]butanes is known to produce a mixture of 3-
methoxycyclobutane and trans-cyclopropane products, ${ }^{79-80}$ the origin of the preference for the cyclopropane product in this case is not obvious.


Scheme 42. Fluorination of amide 3-43 affords cyclopropane 3-102.
This amide fluorination reaction is interesting from a methodological and mechanistic perspective, but unlike the alcohol deoxyfluorination, viable alternatives exist to access the fluorinated products. Commercially available 3-oxocyclobutanecarboxylic acid can be used as a synthetic building block to access fluorinated amides such as 3-88-3-93. For example, a three-step sequence of amide coupling, treatment with a Grignard reagent, and deoxyfluorination with Deoxofluor ${ }^{\circledR}$ was used to access the fluorinated amide 3-30 on process scale (Scheme 28). ${ }^{152}$

However, the de novo preparation of the cyclobutane ring via a bicyclo[1.1.0]butane provides opportunities for unique modifications such as isotopic labelling. As a demonstration of this potential, deuterium-labeled bicyclo[1.1.0]butyl amide 3-107 was prepared and fluorinated to afford the labeled and fluorinated cyclobutane 3-108 (Scheme 43). Substrate 3-107 was accessed through copper-catalyzed addition of phenyl magnesium bromide to propargyl alcohol (3-103), which upon quenching with $\mathrm{D}_{2} \mathrm{O}$ provides allylic alcohol 3-104 as a single alkene isomer. ${ }^{171}$ This alcohol was then protected, subjected to dibromocyclopropanation, and deprotected to afford racemic alcohol 3-105. Conversion to bromide 3-106 and subjection of the resulting tribromide to the dilithiation/isocyanate trapping procedure afforded racemic bicyclo[1.1.0]butane 3-107 in good yield. Like the non-labeled material 3-96, fluorination of 3-107 proceeded with very little diastereoselectivity. However, the 3-108 diastereomers were readily separable on $\mathrm{SiO}_{2}$, and each
has a trans-relationship between the deuterium and amide substituents. One could envision using a similar strategy to produce fluorinated cyclobutanes with other substitution patterns or isotopic labels (such as ${ }^{13} \mathrm{C}$ ).


Scheme 43. Preparation and fluorination of a deuterium labeled bicyclo[1.1.0]butane.

### 3.3 CONCLUSIONS

A method to convert bicyclo[1.1.0]butyl alcohols to fluorinated methylenecyclobutanes has been developed. Screening of a variety of deoxyfluorination conditions revealed that a modified version of Olah's reagent (pyridine $\cdot 9 \mathrm{HF}$ with $\mathrm{KHF}_{2}$ and $i$ - $\mathrm{Pr}_{2} \mathrm{NH}$ additives) was most effective in this reaction. A series of bicyclo[1.1.0]butyl alcohols with various substitution patterns was prepared and subjected to the optimized deoxyfluorination conditions. Many of these alcohols reacted smoothly to form the expected methylenecyclobutane products, but derivatives with sterically demanding appendages or alternative ring junction substituents gave unexpected products. The utility of the fluorinated methylenecyclobutane products was explored through oxidative cleavage of the exocyclic alkene or diastereoselective bromohydrin formation. Preliminary studies on the
fluorination of bicyclo[1.1.0]butyl amides were also undertaken, and a deuterium-labelled cyclobutyl amide was prepared using this method.

### 4.0 SYNTHESIS OF ANDROGEN RECEPTOR ANTAGONISTS CONTAINING STRAINED CARBOCYCLES

### 4.1 INTRODUCTION

### 4.1.1 Androgen Receptor Antagonists as Prostate Cancer Therapeutics

Prostate cancer is the third most common type of cancer diagnosed in the United States. Approximately $12 \%$ of men will be diagnosed with prostate cancer in their lifetime, and an estimated 161,000 new cases were diagnosed in the United States in 2017 alone. Fortunately, current treatment technologies limit the fatality rate from prostate cancer to 20.1 per 100,000 men in the U.S. ${ }^{172}$ A common method to treat high-risk prostate cancer is radiation therapy (with or without prostatectomy) followed by androgen deprivation therapy, including chemical castration. While this approach is an effective short-term treatment, $10-20 \%$ of prostate cancer treated by castration progresses to castration-resistant prostate cancer (CRPC) within 5 years. The estimated period of survival for individuals living with CRPC is 14 months. ${ }^{173}$

A key player in the development of CRPC is the androgen receptor (AR), a nuclear transcription factor that regulates the growth of healthy prostate cells. Upon binding to endogenous $5 \alpha$-dihydrotestosterone, the androgen receptor translocates into the prostate cell nucleus. ${ }^{174}$ It has been demonstrated that excessive AR localized in the cell nucleus enables the progression of CRPC. ${ }^{175}$ Thus, treatment with a small-molecule AR antagonist is a viable strategy to slow CRPC progression. Enzalutamide (4-1, Figure 19) is an FDA-approved AR antagonist for the treatment of CRPC, but it is not completely effective and enables an extension
of patient lifespan by just 3-5 months. ${ }^{176}$ Therefore, there is a need for new AR antagonists that show an improved ability to slow the progression of CRPC, perhaps through a novel mechanism of action.


Figure 19. Enzalutamide, an FDA-approved treatment for castration-resistant prostate cancer.

### 4.1.2 Previous Medicinal Chemistry Efforts in the Wipf group

In an effort to identify new small molecule AR antagonists for the potential treatment of CRPC, a library of 219,055 compounds was screened for suppression of AR activity in the prostate cell nucleus. ${ }^{177}$ The compounds were tested in a castration-resistant prostate cancer cell line (C4-2) transfected with an AR expression vector containing a green fluorescent protein tag (2GFP-AR). A promising hit from this screening was isoxazole thioether 4-2 (Figure 19). In a similar luciferase reporter assay using C4-2 CRPC cells transfected with luciferase-tagged AR vector (C4-2-PSArl ), 4-2 showed promising inhibition of AR activity with an $\mathrm{EC}_{50}$ of $7.3 \mu \mathrm{M}$.

An extensive structure-activity relationship study of 4-2 was conducted in the Wipf group with the aim of developing more potent analogs. ${ }^{178}$ Specific features of 4-2 that were evaluated include the isoxazole appendage, thioether and amide linkages, piperazine heterocycle, and arene appendage (Figure 20). These studies revealed the importance of the $o$-methyl aniline and piperazine features. The isoxazole and thioether, however, were found to be replaceable. More potent analogs were attained by substituting the flexible thioether with a more rigid cyclopropane. The most successful compound in this series was 4-3 (known as JJ-450). Aside from having a $>2$
fold improvement in activity over 4-2, cyclopropane 4-3 has a higher drug-likeness score and eliminates the metabolically liable thioether motif. Extensive evaluations of the capability of 4-3 to treat CRPC in vivo is ongoing.


Figure 20. Optimization of high-throughput screening hit 4-2 yielded lead 4-3.

### 4.2 RESULTS AND DISCUSSION

### 4.2.1 Synthesis of Androgen Receptor Antagonists Containing Bicyclo[1.1.0]butanes

Despite the high potency $\left(\mathrm{EC}_{50}=1.7 \mu \mathrm{M}\right)$ of $(-) \mathbf{- 4 - 3}$ in the luciferase reporter assay, the metabolic stability of this compound is suboptimal. When tested in human, mouse, and dog liver microsomes, (-)-4-3 displayed poor stability with half-lives of $31.7,5.3$, and 5.6 min , respectively. The aryl cyclopropane, piperazine heterocycle, and o-methyl arene substituent are each features of 4-3 that may contribute to this fast metabolic degradation by cytochrome P450 enzymes. ${ }^{179}$ Mass spectrometry analyses of the degradation products from these microsomal stability assays affirmed that oxidative cleavage of the cyclopropane is a major metabolic pathway.

To mitigate these metabolic shortcomings and discover compounds with even greater potency, modifications and replacements of the cyclopropane core of 4-3 were developed.

Alternative carbocycles such as bicyclo[1.1.0]butanes were envisioned to be structurally novel cyclopropane replacements. Others have found that unusual carbocycles such as bicyclo[1.1.1]pentanes and cubanes can be used to enhance the properties of lead compounds. ${ }^{180}$ Despite the structural rigidity and novelty of bicyclo[1.1.0]butanes, to our knowledge, they have not been exploited for this purpose. While the potential for undesired reactivity of bicyclobutanes substituted with an electron withdrawing group (such as an amide) may raise concerns of off-target reactivity, previous thiol addition studies (Section 2.2.4) suggest that endogenous nucleophiles such as glutathione would only add to these systems at elevated temperature or in the presence of base.

In order to prepare bicyclo[1.1.0]butane-derived analogs of 4-3, it was necessary to prepare the appropriate tribromide precursors with electron deficient arenes (Scheme 44). The $p$ fluorophenyl tribromide 4-6 was prepared through allylic bromination of commercially available methyl styrene 4-4 and dibromocyclopropanation of the resulting allylic bromide product (4-5). The same approach has been used to prepare the parent phenyl tribromide, ${ }^{73 b}$ however, the presence of the fluoride substituent decreased the yield for this transformation. Bromination of more highly deactivated trifluoromethyl styrene was completely ineffective. An alternative procedure involving the Cu -mediated addition of $p$-trifluoromethyl magnesium bromide to propargyl alcohol was therefore used to access allylic alcohol 4-8. ${ }^{181}$ The alcohol 4-8 was then subjected to a protection/dibromocyclopropanation/deprotection sequence to provide alcohol 4-9. Bromination of 4-9 afforded the tribromide 4-10.


Scheme 44. Preparation of tribromides 4-6 and 4-10.
With the necessary tribromides in hand, a dilithiation and trapping sequence was proposed to access the amide targets. As outlined in Section 3.2.4, bicyclo[1.1.0]butyl amides can be accessed by trapping the bicyclo[1.1.0]butyl lithium species with isocyanates, but this transformation cannot provide direct access to the tertiary amides needed for these studies. Attempts were made to trap electrophiles (such as 1,1-carbonyldiimidazole (CDI) or chloroformates) that would provide a bicyclobutyl amide precursor at the correct oxidation state. Ultimately, the most successful approach was trapping of the bicyclobutane anion with $\mathrm{CO}_{2}$ to form a carboxylic acid (Table 21). By conducting a pH -controlled workup of this reaction, bicyclo[1.1.0]butyl acids 4-11 and 4-12 were obtained in acceptable purity without purification by chromatography. Although some reports describe the preparation of certain bicyclo[1.1.0]butyl carboxylic acids through anionic trapping of $\mathrm{CO}_{2},{ }^{83 \mathrm{c}, 182}$ this work is the first example of a general approach to aryl-substituted bicyclo[1.1.0]butyl carboxylic acids.

Table 21. Preparation of carboxylic acids 4-11 and 4-12.


The bicyclo[1.1.0]butane carboxylic acids 4-11 and 4-12 were coupled to the appropriate aryl piperizines using the $\mathrm{T} 3 \mathrm{P}^{\circledR}$ reagent (Table 22). ${ }^{183}$ These piperazine coupling partners were prepared by collaborators through Buchwald-Hartwig coupling of N -Boc piperazine with an aryl bromide. ${ }^{178}$ The carbamate protecting group was then removed with HCl in dioxane to afford the piperazine HCl salts, which were typically free-based before amide coupling. In this manner, three bicyclobutyl amide analogs 4-13, 4-14, and 4-15 were prepared for testing of their AR antagonist activity.

Table 22. Amide coupling of carboxylic acids 4-11 and 4-12.


### 4.2.2 Rearrangement of Bicyclo[1.1.0]butyl Amides to Cyclobutenes

Initially, the amide couplings described in Table 22 were conducted by pre-mixing a solution of bicyclo[1.1.0]butane carboxylic acid with piperazine HCl salt, treating with excess $\mathrm{Et}_{3} \mathrm{~N}$, and then adding the $\mathrm{T} 3 \mathrm{P}^{\circledR}$ coupling reagent. This approach gave perplexingly inconsistent results. While the expected bicyclo[1.1.0]butyl amides were isolated in some cases, in many others an unexpected cyclobutene product was obtained in low yield (as in the conversion of 4-12 to 4-16, Scheme 45). Through a systematic evaluation of reaction parameters, it was determined that this cyclobutene formation was triggered by excess HCl in the piperazine HCl salt, possibly resulting from
incomplete washing of the solid during large-scale preparations. When the piperazine HCl salt was partitioned between $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and 1 M NaOH before use in the coupling, bicyclobutyl amides such as $\mathbf{4 - 1 5}$ could be reliably prepared.

With this knowledge, several acids were screened for their effectiveness in promoting the isomerization of bicyclobutane 4-15 to cyclobutene 4-16. The optimally clean and high yielding protocol was found to be treatment of a $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution of $\mathbf{4 - 1 5}$ with a catalytic ( $15 \mathrm{~mol} \%$ ) amount of camphorsulfonic acid followed by heating at $60^{\circ} \mathrm{C}$. Camphorsulfonic acid is an advantageous acid source in that it is an easily weighed solid that does not exist as a hydrate. Acids with appreciable water content tended to promote the competitive formation of a cyclobutanol hydration product in addition to cyclobutene formation.

To confirm the structure of cyclobutene 4-16, it was subjected to hydrogenation to afford cyclobutane 4-17 as a single diastereomer (Scheme 45). Bicyclo[1.1.0]butyl amide 4-15 was also subjected to hydrogenation across the strained central bond at elevated pressure to afford the same cyclobutane product (4-17), which affirms the structural assignment of $\mathbf{4 - 1 6}$. The relative stereochemistry of cyclobutane 4-17 was confirmed by NOESY, wherein a correlation is observed from both methines to one of the two methylenes.


Scheme 45. Preparation of cyclobutane 4-16 and confirmation of structure through conversion to 4-17.

The isomerization of bicyclo[1.1.0]butanes to cyclobutenes has been conducted with cobalt (II) porphyrin complexes, ${ }^{184}$ photochemical irradiation, ${ }^{185}$ or reduction over lithium metal. ${ }^{186}$ However, no general method for the isomerization of functionalized bicyclobutanes under mild conditions has been described. The scope of this catalytic acid-mediated isomerization was therefore investigated (Table 23). Both secondary (NHPh) and tertiary (piperazine) amides were well tolerated in the isomerization (entries 1-5). The electronic nature of the arene ring was unimportant, as evidenced by the similar yields obtained in entries $4\left(\mathrm{R}^{1}=H\right)$ and $5\left(\mathrm{R}^{1}=\mathrm{CF}_{3}\right)$. A ketone substrate (obtained through DMP oxidation of the corresponding bicyclo[1.1.0]butyl alcohol) afforded the expected cyclobutane in modest yield (entry 6). The lower yield may be attributed to the poor stability of bicyclo[1.1.0]butyl ketones, which are more susceptible to decomposition than other bicyclobutane derivatives.

Table 23. Conversion of bicyclo[1.1.0]butyl amides to cyclobutenes.


| entry | $\mathrm{R}^{1}$ | $\mathrm{R}^{2}$ | product |  |
| :--- | :---: | :---: | :---: | :---: |
| yield (\%) |  |  |  |  |
| 1 | $\mathrm{CF}_{3}$ | $\mathbf{A}$ | $\mathbf{4 - 1 6}$ | 71 |
| 2 | H | NHPh | $\mathbf{4 - 1 8}$ | 66 |
| 3 | Cl | NHPh | $\mathbf{4 - 1 9}$ | 84 |
| 4 | H | $\mathbf{B}$ | $\mathbf{4 - 2 0}$ | 75 |
| 5 | $\mathrm{CF}_{3}$ | $\mathbf{B}$ | $\mathbf{4 - 2 1}$ | 76 |
| 6 | $\mathrm{CF}_{3}$ | Ph | $\mathbf{4 - 2 2}$ | 30 |




An arene substituent at the ring junction is necessary for this isomerization. When methylsubstituted amide 3-92 was subjected to the optimized conditions, no conversion occurred. More vigorous heating at $120^{\circ} \mathrm{C}$ eventually triggered formation of methylenecyclobutane 4-23 in 17\% yield (Scheme 46). This result is consistent with an isomerization mechanism whereby amide protonation results in a cyclobutyl cation that is stabilized by the aryl substituent. Elimination then
affords the racemic cyclobutene product. In the case of 4-23, thermodynamic formation of the less strained methylenecyclobutane is favored over cyclobutene formation.


Scheme 46. Conversion of amide 3-92 to methylenecyclobutane 4-23.

### 4.2.3 Biological Activity of Amide Analogs

Bicyclo[1.1.0]amides 4-13, 4-14, and 4-15, as well as cyclobutane 4-17 (all of which are achiral) were tested for their ability to inhibit AR activity in the prostate cell nucleus using the AR luciferase dual-reporter assay. ${ }^{177}$ Select analogs were subjected to additional microsome stability tests and cell proliferation assays in both AR positive (22Rv1) and AR negative (PC3) CRPC cell lines (Table 24). The respective enantiomers of cyclobutane $\mathbf{4 - 1 6}$ (which is chiral due to the desymmetrization from the cyclic olefin) were separated by preparative chiral SFC and evaluated separately in the assays. Although many other analogs were prepared by collaborators and tested in these assays, only one (oxetane 4-24) contained a four-membered ring. The biological data for both enantiomers of 4-24 are shown in Table 24 to enable comparison to cyclobutane 4-17. Assay data for both enantiomers of the lead compound 4-3 (JJ-450) are also shown for comparison.

Table 24. Biological testing of androgen receptor antagonist analogs.




4-16


4-17: $\mathrm{X}=\mathrm{CH}_{2}$
4-24: $\mathrm{X}=\mathrm{O}$ (prepared by a collaborator)

| entry | compound | luciferase <br> assay <br> $\mathrm{EC}_{50}(\mu \mathrm{M})$ | human liver <br> microsome <br> $\mathrm{t}_{1 / 2}(\mathrm{~min})$ | mouse liver <br> microsome <br> $\mathrm{t}_{1 / 2}(\mathrm{~min})$ | 22Rv1 cell assay <br> $(\%$ inibition at 30 <br> $\mu \mathrm{M})$ | PC3 cell assay (\% <br> inibition at $30 \mu \mathrm{M})$ |
| :---: | :--- | :---: | :---: | :---: | :---: | :---: |
| 1 | $(-)-4-3$ | 1.7 | 31.7 | 5.3 | 88 | 44 |
| 2 | $(+)-4-3$ | 14.8 | 16.3 | 7.3 | 78 | 33 |
| 3 | $\mathbf{4 - 1 3}$ | $>25$ | not tested | not tested | not tested | not tested |
| 4 | $\mathbf{4 - 1 4}$ | 2.8 | 36.5 | 43.1 | 70 | 30 |
| 5 | $4-15$ | 12.3 | 26.9 | 31.2 | 83 | 49 |
| 6 | $(-)-4-16$ | 8.2 | not tested | not tested | not tested | not tested |
| 7 | $(+)-4-16$ | $>25$ | not tested | not tested | not tested | not tested |
| 8 | $\mathbf{4 - 1 7}$ | $>25$ | $>1000$ | 178.7 | 24 | 10 |
| 9 | $(-)-4-24$ | 15.1 | 21.3 | 16.0 | 78 | 46 |
| 10 | $(+)-4-24$ | 12.7 | 129.0 | 88.4 | 82 | 35 |

While the $p$-fluoro bicyclo[1.1.0]butyl amide $\mathbf{4 - 1 3}$ showed $>25 \mu \mathrm{M}$ activity in the luciferase dual-reporter assay and was not considered for cell-based evaluation (entry 3), ptrifluoromethyl substrates 4-14 and 4-15 were comparable in activity to lead structure (-)-4-3 (entries 4 and 5 vs 1 and 2). Like (-)-4-3, these bicyclobutanes showed strong and selective growth inhibition of the human 22Rv1 (androgen receptor positive) CRPC cell line over the PC3 (androgen receptor negative) CRPC cell line. The stability of these bicyclo[1.1.0]butane substrates in human liver microsomes was comparable to lead compound (-)-4-3. In the case of mice liver microsomes, however, the half-life was nearly an order of magnitude greater. These results suggest
that despite their high strain energy, bicyclo[1.1.0]butanes can serve as cyclopropane, cyclobutane, or cis-alkene substitutes that are stable to metabolic degradation in vivo.

Cyclobutane 4-17 was significantly less potent than many other analogs (including oxetane 4-24) in both the luciferase reporter and cell growth assays (entry 8). However, 4-17 exhibited dramatically higher microsomal stability than any of the comparison substrates. Others have observed that cyclobutane derivatives of lead structures have an enhanced metabolic stability, ${ }^{180 a}$ and molecules containing cyclobutanes have shown effective pharmacokinetic performance in human clinical trials. ${ }^{187}$ Taken together, these results encourage continued studies of these and other small carbocyclic ring systems in medicinal chemistry.

### 4.3 CONCLUSIONS

Several bicyclo[1.1.0]butyl amides were prepared as part of a medicinal chemistry to identify androgen receptor antagonists for the treatment of castration-resistant prostate cancer. A method for the synthesis of aryl bicyclo[1.1.0]butane carboxylic acids was developed to attain the necessary synthetic targets. These new analogs had comparable androgen receptor antagonist activity and greater microsomal stability than the previous lead compound 4-3 (JJ-450). During these studies, an acid-mediated isomerization of bicyclo[1.1.0]butanes to cyclobutenes was discovered. The scope of this transformation was investigated.

### 5.0 EXPERIMENTAL

### 5.1 GENERAL EXPERIMENTAL

All glassware was oven dried and cooled in a desiccator prior to use. All moisture sensitive reactions were performed under an atmosphere of dry $\mathrm{N}_{2}$ or Ar. Reactions carried out below $0{ }^{\circ} \mathrm{C}$ employed an acetone/dry ice bath. Reagents obtained from commercial sources were used as received unless otherwise specified. THF and $\mathrm{Et}_{2} \mathrm{O}$ were distilled from sodium/benzophenone ketyl and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and toluene were distilled over $\mathrm{CaH}_{2}$. Concentrating under reduced pressure refers to the use of a rotary evaporator connected to a PIAB Lab Vac H40, followed by a $1-$ 1.5 torr high vacuum pump to remove residual solvent.

Melting points were determined using a Laboratory Devices Mel-Temp II in open capillary tubes and are uncorrected. Infrared spectra were determined as neat solids or oils on a Smiths Detection IdentifyIR FT-IR spectrometer. ${ }^{1} \mathrm{H}$ NMR, ${ }^{13} \mathrm{C}$ NMR, ${ }^{11} \mathrm{~B}$ NMR, ${ }^{19} \mathrm{~F}$ NMR and ${ }^{31} \mathrm{P}$ NMR spectra were recorded on a Bruker Advance $300 \mathrm{MHz}, 400 \mathrm{MHz}$, or 500 MHz instruments. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR chemical shifts ( $\delta$ ) were reported in parts per million with the residual solvent peak used as an internal standard $\delta$ and are tabulated as follows: chemical shift, multiplicity $(\mathrm{s}=$ singlet, $\mathrm{d}=$ doublet, $\mathrm{t}=$ triplet, $\mathrm{q}=$ quartet, $\mathrm{m}=$ multiplet $)$, number of protons, and coupling constant(s). ${ }^{13} \mathrm{C}$ NMR spectra were obtained using a proton-decoupled pulse sequence and are tabulated by observed peak. ${ }^{11} \mathrm{~B}$ NMR, ${ }^{19} \mathrm{~F}$ NMR, and ${ }^{31} \mathrm{P}$ NMR spectra were taken without an internal standard. Thin-layer chromatography was performed using precoated silica gel 60 F254 plates (EMD, $250 \mu \mathrm{~m}$ thickness) and visualization was accomplished with a 254 nm UV light and by staining with a $\mathrm{KMnO}_{4}$ solution $\left(1.5 \mathrm{~g}\right.$ of $\mathrm{KMnO}_{4}$ and 1.5 g of $\mathrm{K}_{2} \mathrm{CO}_{3}$ in 100 mL of a $0.1 \%$

NaOH solution). Flash chromatography on $\mathrm{SiO}_{2}$ (Silicycle, Silia-P Flash Silica Gel or SiliaFlash ${ }^{\circledR}$ P60, 40-63 $\mu \mathrm{m}$ ) was used to purify crude reaction mixtures.

Unless otherwise stated, starting materials were purchased from commercial sources and used without further purification. Thiourea starting materials were either obtained from commercial sources, from collaborators, or were prepared from the corresponding primary amine and isocyanate. Secondary phosphine boranes were supplied from Boehringer-Ingelheim (Ridgefield, CT ) or were prepared according to the referenced procedure.

### 5.2 CHAPTER 1 EXPERIMENTAL PART



General Procedure A (Carbodiimide formation from the corresponding thiourea): N -(3, 5Bistrifluoromethyl)phenyl $N^{\prime}$-cyclohexylcarbodiimide (1-28). A 25 mL round bottom flask was charged with the corresponding thiourea ( $0.263 \mathrm{~g}, 0.710 \mathrm{mmol}$ ), triethylamine $(0.30 \mathrm{~mL}, 1.4 \mathrm{mmol}$, 2.0 eq), DMAP ( $0.004 \mathrm{~g}, 0.03 \mathrm{mmol}, 5 \mathrm{~mol} \%)$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$. The reaction mixture was cooled to $0^{\circ} \mathrm{C}$ and mesyl chloride ( $0.10 \mathrm{~mL}, 1.4 \mathrm{mmol}, 2.0 \mathrm{eq}$ ) was added via syringe over 5 min . The solution was stirred at room temperature for 1 h then was concentrated under reduced pressure. The crude material was slurried in hexane and filtered through a plug of Celite ${ }^{\circledR}$. The filtrate was concentrated under reduced pressure and purified by chromatography on $\mathrm{SiO}_{2}$ (hexanes) under $\mathrm{N}_{2}$ pressure to afford $\mathbf{1 - 2 8}(0.197 \mathrm{~g}, 0.585 \mathrm{mmol}, 83 \%)$ as a colorless oil: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 7.56(\mathrm{~s}, 1 \mathrm{H}), 7.46(\mathrm{~s}, 2 \mathrm{H}), 3.63-3.58(\mathrm{~m}, 1 \mathrm{H}), 2.03-2.00(\mathrm{~m}, 2 \mathrm{H}), 1.77(\mathrm{dd}, 2 \mathrm{H}, J=9.6,4.0$ $\mathrm{Hz}), 1.60-1.49(\mathrm{~m}, 3 \mathrm{H}), 1.43-1.40(\mathrm{~m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 143.8,133.4\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}\right.$
$=33 \mathrm{~Hz}), 123.2\left({ }^{3} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right), 123.0\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=271 \mathrm{~Hz}\right), 117.6\left({ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 56.8,34.7,25.2,24.1$; MS (ESI) ${ }^{+} m / z 337$ (100); HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}_{2} \mathrm{~F}_{6}[\mathrm{M}+\mathrm{H}]$ 337.1139, found 337.1141 .

$N$-2-Methoxyphenyl $N$ '-cyclohexylcarbodiimide (1-29). Prepared according to General Procedure A from the corresponding thiourea $(0.327 \mathrm{~g}, 1.23 \mathrm{mmol})$, DMAP $(0.008 \mathrm{~g}, 0.06 \mathrm{mmol})$, triethylamine ( $0.52 \mathrm{~mL}, 3.7 \mathrm{mmol}$ ) and mesyl chloride $(0.19 \mathrm{~mL}, 2.5 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(15 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ afforded $\mathbf{1 - 2 9}(0.204 \mathrm{~g}, 0.885 \mathrm{mmol}, 72 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.12-6.97(\mathrm{~m}, 2 \mathrm{H}), 6.91-6.81(\mathrm{~m}, 2 \mathrm{H}), 3.88(\mathrm{~s}, 3 \mathrm{H}), 3.49-3.36$ (m, 1 H$), 2.08-1.93(\mathrm{~m}, 2 \mathrm{H}), 1.84-1.68(\mathrm{~m}, 2 \mathrm{H}), 1.62-1.08(\mathrm{~m}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 154.0,136.4,129.3,125.2,124.6,121.0,111.2,56.5,55.7,34.6,25.5,24.5 ;$ IR (ATR) 3027, 2134, 1593, 1485, 1232, 1192, 827, 769, $692 \mathrm{~cm}^{-1} ; \mathrm{MS}(\mathrm{ESI})^{+} m / z 249\left(100, \mathrm{M}+\mathrm{H}_{2} \mathrm{O}\right), 231$ (10), 149 (7); $\mathrm{HRMS}(\mathrm{ESI})^{+} m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{~N}_{2} \mathrm{O}_{2}\left[\mathrm{M}+\mathrm{H}+\mathrm{H}_{2} \mathrm{O}\right]$ 249.1603, found 249.1655.

$\boldsymbol{N}$-4-Methoxyphenyl $\boldsymbol{N}$ '-phenylcarbodiimide (1-30). ${ }^{188}$ Prepared according to General Procedure A from the corresponding thiourea $(1.63 \mathrm{~g}, 6.31 \mathrm{mmol})$, triethylamine $(2.66 \mathrm{~mL}, 18.9$ mmol, 3.00 eq ), mesyl chloride ( $0.97 \mathrm{~mL}, 13 \mathrm{mmol}, 2.0 \mathrm{eq}$ ) and DMAP ( $0.039 \mathrm{~g}, 0.32 \mathrm{mmol}, 5$ $\mathrm{mol} \%$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(60 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc} /$ hexanes $)$ afforded $\mathbf{1 - 3 0}(1.23 \mathrm{~g}, 5.49 \mathrm{mmol}, 87 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.35-7.30(\mathrm{~m}, 2$ H), 7.19-7.13 (m, 5 H$), 6.85(\mathrm{dd}, 2 \mathrm{H}, J=6.9 \mathrm{~Hz}, 2.4 \mathrm{~Hz}), 3.80(\mathrm{~s}, 3 \mathrm{H})$.


1-31
$\boldsymbol{N}, \boldsymbol{N}^{\prime}$-Diallylcarbodiimide (1-31). Prepared according to General Procedure A from the corresponding thiourea ( $1.00 \mathrm{~g}, 6.40 \mathrm{mmol}$ ), DMAP $(0.039 \mathrm{~g}, 0.32 \mathrm{mmol})$, triethylamine ( 2.69 $\mathrm{mL}, 19.2 \mathrm{mmol}$ ), and mesyl chloride ( $1.00 \mathrm{~mL}, 12.8 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. Purification by filtration through Celite afforded carbodiimide $\mathbf{1 - 3 1}(0.456 \mathrm{~g}, 3.73 \mathrm{mmol}, 58 \%)$ as a clear oil: ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 5.90,5.27,5.18\left(\mathrm{ABX}, 2 \times 3 \mathrm{H}, J_{\mathrm{AX}}=17.2 \mathrm{~Hz}, J_{\mathrm{BX}}=10.0 \mathrm{~Hz}, J_{\mathrm{AB}}=\right.$ $1.6 \mathrm{~Hz}), 3.82(\mathrm{~d}, 2 \times 2 \mathrm{H}, J=4.8 \mathrm{~Hz}) .{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 141.0,134.6,116.4,48.8$.


General Procedure B (Hydrophosphination of Carbodiimides): ( $N$-Cyclohexyl, $N^{\prime}$-(3, 5trifluoromethyl) phenylcarbamimidoyl)-dicyclohexylphosphine borane (1-32). A 5 mL vial was flushed with Ar and charged with di-tert-butyl phosphine borane $(0.064 \mathrm{~g}, 0.40 \mathrm{mmol})$ and carbodiimide 1-30 ( $0.075 \mathrm{~g}, 0.33 \mathrm{mmol})$. These materials were dissolved in dimethylacetamide ( 2 mL ) and sparged with Ar for 10 min . The flask was then cooled in an ice bath. Sodium hydride $(0.016 \mathrm{~g}, 0.40 \mathrm{mmol}, 60 \%$ dispersion $)$ was then added at once, causing immediate gas evolution, and foaming to a white suspension. After 15 min , the ice bath was removed and then the reaction mixture allowed to warm to rt . After 15 min at rt , the reaction was stirred vigorously while open to the air for 30 minutes (the solution had a distinct green color). The reaction mixture was treated with sat aq $\mathrm{NH}_{4} \mathrm{Cl}(2 \mathrm{~mL})$, and extracted with diethyl ether ( $2 \times 30 \mathrm{~mL}$ ). The combined organic phases were then washed with water $(30 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated to give a crude sticky oil. Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc} /$ hexane $)$ afforded $\mathbf{1 - 3 2}(0.100 \mathrm{~g}$,
$0.260 \mathrm{mmol}, 78 \%)$ as a clear oil: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.96(\mathrm{~s}, 0.5 \mathrm{H}), 7.86(\mathrm{~s}, 0.5 \mathrm{H})$, 6.96-6.86 (m, 2 H$), 6.82(\mathrm{t}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}), 6.69(\mathrm{t}, 0.5 \mathrm{H}, J=7.5 \mathrm{~Hz}), 6.61-6.46(\mathrm{~m}, 4 \mathrm{H}), 6.44$ $(\mathrm{dd}, 1 \mathrm{H}, J=7.0 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 3.65(\mathrm{~s}, 3 \mathrm{H}), 1.53(\mathrm{~s}, 9 \mathrm{H}), 1.51(\mathrm{~s}, 9 \mathrm{H}), 1.1-0.3(\mathrm{br}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3},{ }^{1} J_{\mathrm{CP}}$ not resolved) $\delta 128.0,127.9,124.1,123.7,122.1,121.8,121.3,120.4$, $113.4\left({ }^{3} J_{\mathrm{CP}}=14 \mathrm{~Hz}\right), 55.5,33.8,28.4,27.3 ;{ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 52.1 ;{ }^{11} \mathrm{~B}$ NMR ( 128 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-43$; HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{22} \mathrm{H}_{32} \mathrm{~N}_{2} \mathrm{OP}\left[\mathrm{M}+\mathrm{H}-\mathrm{BH}_{3}\right]$ 371.2252, found 371.2265. This compound exhibits rotomers and/or $E-Z$ isomers by NMR.


## (3,5-Bis(trifluoromethyl)phenyl)- N -cyclohexylcarbamimidoyldicyclohexyl

phosphine borane (1-33). ${ }^{31}$ Prepared according to General Procedure $B$ from dicyclohexylphosphine borane $(0.080 \mathrm{~g}, 0.38 \mathrm{mmol})$, carbodiimide $\mathbf{1 - 2 8}(0.106 \mathrm{~g}, 0.315 \mathrm{mmol})$ and sodium hydride ( $0.015 \mathrm{~g}, 0.38 \mathrm{mmol}, 60 \%$ ) in DMAc ( 1.2 mL ). Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc} /$ hexane $)$ afforded phosphine borane $\mathbf{1 - 3 3}(0.095 \mathrm{~g}, 0.17$ mmol, $55 \%$ ) as a clear oil that solidified to a colorless solid on standing: Mp 97.5-98.9 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 6.43(\mathrm{t}, 1 \mathrm{H}, J=9.2 \mathrm{~Hz}), 6.30(\mathrm{~d}, 2 \mathrm{H}, J=6.8 \mathrm{~Hz}), 5.84(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, 3.07 (br s, 1 H ), 2.10-2.00 (m, 2 H ), 1.90-0.75 (m, 30 H ), 1.0-0.0 (br, 3 H ); ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}, \mathrm{CF}_{3}$ not resolved) $\delta 164.5,164.3,162.0,161.9,153.2,147.9\left(\mathrm{~d},{ }^{1} J_{\mathrm{CP}}=68 \mathrm{~Hz}\right), 103.5(\mathrm{~d}$, $\left.{ }^{3} J_{\mathrm{CP}}=25 \mathrm{~Hz}\right), 97.2\left(\mathrm{t},{ }^{3} J_{\mathrm{CP}}=27 \mathrm{~Hz}\right), 52.0,33.1,32.3,31.9,26.9,26.69,26.66,25.9,25.2,24.2 ;$ ${ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 41.1 ;{ }^{11} \mathrm{~B} \mathrm{NMR}\left(128 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-44 ; \mathrm{HRMS}(\mathrm{ESI}){ }^{+} \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{27} \mathrm{H}_{38} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{P}\left[\mathrm{M}+\mathrm{H}-\mathrm{BH}_{3}\right] 535.2677$, found 535.2661.

( $N$-Phenyl, $N^{\prime}$-2-methoxylphenylylcarbamimidoyl)-dicyclohexylphosphine borane (1-34). Prepared according to general Procedure B from dicyclohexylphosphine borane ( $0.536 \mathrm{~g}, 2.53$ $\mathrm{mmol})$, carbodiimide $\mathbf{1 - 2 9}(0.485 \mathrm{~g}, 2.10 \mathrm{mmol})$ and sodium hydride $(0.101 \mathrm{~g}, 1.06 \mathrm{mmol}, 60 \%)$ in DMAc ( 10 mL ). Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc} /$ hexane $)$ afforded phosphine borane $\mathbf{1 - 3 4}(0.540 \mathrm{~g}, 1.22 \mathrm{mmol}, 58 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 6.93(\mathrm{t}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 6.86-6.71(\mathrm{~m}, 3 \mathrm{H}), 5.67(\mathrm{t}, 3 \mathrm{H}, J=3.2 \mathrm{~Hz}), 3.75(\mathrm{~s}, 3 \mathrm{H})$, 2.95-2.79 (m, 1 H$), 2.23-2.07(\mathrm{~m}, 2 \mathrm{H}), 1.95-1.48(\mathrm{~m}, 18 \mathrm{H}), 1.45-0.70(\mathrm{~m}, 13 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 150.2,147.0\left({ }^{1} J_{\mathrm{CP}}=77 \mathrm{~Hz}\right), 139.6\left({ }^{3} J_{\mathrm{CP}}=13 \mathrm{~Hz}\right), 122.8,121.1,120.5$, $110.6,55.1,51.5\left({ }^{3} J_{\mathrm{CP}}=6 \mathrm{~Hz}\right), 34.7,33.7,32.1,31.8,31.6,26.8,26.82,26.77,26.69,26.65$, 26.63, 26.61, 25.29, 25.26, 24.4, 14.1; ${ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 39.1 ;{ }^{11} \mathrm{~B}$ NMR ( 128 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta-45$; $\mathrm{HRMS}(\mathrm{ESI})^{+} m / z$ calcd for $\mathrm{C}_{26} \mathrm{H}_{45} \mathrm{~N}_{2} \mathrm{OPB}[\mathrm{M}+\mathrm{H}] 443.3362$, found 433.3353.

( $N, N^{\prime}$-Diallylcarbamimidoyl)-dicyclohexylphosphine borane (1-35). Prepared according to General Procedure B from carbodiimide 1-31 ( $0.150 \mathrm{~g}, 1.23 \mathrm{mmol}$ ), dicyclohexylphosphine borane ( $0.312 \mathrm{~g}, 1.47 \mathrm{mmol}$ ), and NaH dispersion ( $0.059 \mathrm{~g}, 1.47 \mathrm{mmol}$ ) in DMAc ( 6 mL ). Purification by chromatography on $\mathrm{SiO}_{2}$ ( $5 \% \mathrm{EtOAc}$ in hexane) to afforded $\mathbf{1 - 3 5}$ ( $0.153 \mathrm{~g}, 0.458 \mathrm{mmol}, 37 \%$ ) as a yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 6.08-5.70(\mathrm{~m}, 2 \mathrm{H}), 5.35-5.04(\mathrm{~m}, 4 \mathrm{H}), 4.14(\mathrm{~s}, 2 \mathrm{H})$, 3.95 (s, 2 H ), 2.40-1.05 (m, 22 H ), 0.90-0.05 (br, 3 H ); IR (ATR): 3368, 2934, 2376, 1629, 1450,

1077, $982,917 \mathrm{~cm}^{-1} ;{ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 38.4 ;{ }^{11} \mathrm{~B}$ NMR ( $128 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-44$; HRMS m/z calcd for $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{~N}_{2} \mathrm{P}\left[\mathrm{M}+\mathrm{H}-\mathrm{BH}_{3}\right]$ 321.2454, found 321.2432. Significant broadening was observed in the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra.


## ( $N$, $N$ '-Dicyclohexylcarbamimidoylium)- dicyclohexylphosphine borane tetrafluoroborate

 (1-36). ${ }^{31}$ A solution of phosphine borane $\mathbf{1 - 1 9}(0.077 \mathrm{~g}, 0.18 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1.5 \mathrm{~mL})$ was cooled to $0{ }^{\circ} \mathrm{C}$ under Ar and treated with $\mathrm{HBF}_{4}$ diethyl ether complex ( $0.37 \mathrm{~mL}, 2.8 \mathrm{mmol}, 15 \mathrm{eq}$ ). After 30 min at $0^{\circ} \mathrm{C}$, the reaction was warmed to rt over 30 min . The solution was opened to air, treated with aq $\mathrm{HBF}_{4}(1.5 \mathrm{~mL})$, and stirred for 30 min more. The solution was then diluted with water (5 $\mathrm{mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 15 \mathrm{~mL})$. The combined extracts were washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and were concentrated to afford $\mathbf{1 - 3 6}(0.075 \mathrm{~g}, 0.15 \mathrm{mmol}, 81 \%)$ as a colorless crystalline solid: Mp 259-261 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.46-7.34$ (br s, 1 H ), 4.75-4.65 (m, 1 H ), 3.62-3.38 (m, 2 H ), 2.40-2.25 (m, 2 H ), 1.95-1.10(m, 41 H$) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$, ) $\delta$ $158.1\left({ }^{1} J_{\mathrm{CP}}=30 \mathrm{~Hz}\right), 156.2\left({ }^{1} J_{\mathrm{CP}}=27 \mathrm{~Hz}\right), 58.6,53.3\left({ }^{3} J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 34.9,34.7,32.8,32.6,32.1$, $31.8,31.2,28.3,27.8,27.7,27.2,26.42,26.38,26.3,26.0,25.9,25.8,25.7,25.42,25.40,25.1$, 25.0, 24.4, 24.2, 23.2; ${ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 60.0,56.9 ;{ }^{11} \mathrm{~B}$ NMR ( $128 \mathrm{MHz}, \mathrm{CDCl}_{3}$, measured without lock) $\delta-0.9,-45.4$. A single crystal suitable for X-ray analysis was grown from hexanes/acetone (1:1). The X-ray crystal structure data for the compound has been deposited with CCDC \#997593.

## $N$-Cyclohexyl, $N$ '-phenylcarbamimidoyl)-di-(3, 5-dimethyl-4-methoxyphenyl)

phosphine oxide (1-38). ${ }^{31}$ A solution of phosphine $\mathbf{1 - 3 7}(0.052 \mathrm{~g}, 0.10 \mathrm{mmol}$, obtained from collaborators) in acetone ( 1 mL ) was treated with hydrogen peroxide $(0.12 \mathrm{~mL}, 0.11 \mathrm{mmol}, 3 \%$ aq. solution). The reaction mixture was stirred for 1 h at rt ; TLC analysis ( $10 \% \mathrm{EtOAc} /$ hexanes) showed a clean and quantitative oxidation. The solution was diluted with water ( 5 mL ) and extracted with $\mathrm{Et}_{2} \mathrm{O}(2 \times 10 \mathrm{~mL})$. The combined organic extracts were washed with water ( 10 mL ), brine ( 10 mL ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated to afford phosphine oxide $\mathbf{1 - 3 8}(0.053 \mathrm{~g}, 0.10$ mmol, $99 \%$ ) as a sticky colorless gum: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (significant broadening) $\delta$ 7.60-7.41 (m, 4 H), 7.22-6.60 (m, 5H), 6.30-6.20 (br s, 1 H), 3.74 (s, 6 H), 2.28 (s, 12 H), 1.78$0.95(\mathrm{~m}, 11 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 160.4$ (s), 132.9, 132.8, 131.3, 131.2, 128.2, 126.3, $125.3,122.0,120.8,59.6,33.3,25.4,24.7,16.2 ;{ }^{31} \mathrm{P}$ NMR (162 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 20.0 ;$ HRMS $(\mathrm{ESI})^{+} m / z$ calcd for $\mathrm{C}_{31} \mathrm{H}_{40} \mathrm{~N}_{2} \mathrm{O}_{3} \mathrm{P}[\mathrm{M}+\mathrm{H}]$ 519.2777, found 519.2764.


5, 6-Didehydro-12, 13-dihydrodibenzodiazonine (1-41). ${ }^{46} \mathrm{~A}$ solution of dianiline $1-39$ ( 3.50 g , $16.4 \mathrm{mmol})$ in $2 \mathrm{M} \mathrm{HCl}(60 \mathrm{~mL})$ was cooled to $0^{\circ} \mathrm{C}$ and treated dropwise with a solution of sodium nitrite ( $3.41 \mathrm{~g}, 49.5 \mathrm{mmol}, 3.00 \mathrm{eq}$ ) in $\mathrm{H}_{2} \mathrm{O}(30 \mathrm{~mL})$. The yellow solution was stirred for 1 h at 0 ${ }^{\circ} \mathrm{C}$, then was treated dropwise with a solution of sodium azide $(4.29 \mathrm{~g}, 65.0 \mathrm{mmol}, 4.00 \mathrm{eq})$ in water ( 35 mL ), which caused considerable foaming. The mixture warmed to rt and stirred for 24
$h$, which resulted in formation of a tan precipitate. After addition of $\mathrm{CH}_{2} \mathrm{Cl}_{2}(75 \mathrm{~mL})$ and water $(10 \mathrm{~mL})$, the layers were separated and the organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated in vacuo. The crude material was filtered through a short plug of $\mathrm{SiO}_{2}$ with $30 \% \mathrm{EtOAc}$ in hexanes. Concentration of the eluent afforded diazide $\mathbf{1 - 4 0}(3.61 \mathrm{~g}, 13.7 \mathrm{mmol}, 83 \%)$ as a $\tan$ solid: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.26(\mathrm{td}, 2 \mathrm{H}, J=9.0 \mathrm{~Hz}, 1.8 \mathrm{~Hz}), 7.12(\mathrm{t}, 4 \mathrm{H}, J=6.6 \mathrm{~Hz}), 7.04$ (app $\mathrm{t}, 2 \mathrm{H}, J=7.2 \mathrm{~Hz}), 2.82(\mathrm{~s}, 4 \mathrm{H}) ; \operatorname{IR}(\mathrm{ATR}) 3085,2993,2115,1485,1286,1163,751 \mathrm{~cm}^{-1}$.

A solution of diazide $\mathbf{1 - 4 0}(3.61 \mathrm{~g}, 13.7 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(100 \mathrm{~mL})$ under $\mathrm{N}_{2}$ at rt was treated with triphenylphosphine $(7.17 \mathrm{~g}, 27.3 \mathrm{mmol})$. The opaque brown solution was stirred at rt for 20 h and developed a brown precipitate. The solvent was removed in vacuo to afford the iminophosphorane $(10.1 \mathrm{~g})$ as a buff solid that was used without further purification: ${ }^{1} \mathrm{H}$ NMR (400 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 7.85-7.23(\mathrm{~m}, 32 \mathrm{H}), 7.12-7.01(\mathrm{~d}, 3 \mathrm{H}, J=6.6 \mathrm{~Hz}), 6.80-6.45(\mathrm{~m}, 6 \mathrm{H}), 3.23-$ 3.05 (s, 4 H); IR (ATR) 3026, 1594, 1479, 1450, 1340, 1103, 726, $691 \mathrm{~cm}^{-1}$.

A suspension of the iminophosphorane ( $2.40 \mathrm{~g}, 3.27 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(200 \mathrm{~mL})$ was sparged with a balloon $\mathrm{CO}_{2}$ at rt for 4 h , over which period the reaction mixture turned from opaque to transparent yellow/brown. The reaction mixture was concentrated, and the concentrate was purified by chromatography on $\mathrm{SiO}_{2}(10 \% \mathrm{EtOAc} /$ hexane $)$ to afford carbodiimide $1-41(0.68 \mathrm{~g}$, $3.1 \mathrm{mmol}, 94 \%$ ) as a colorless solid: $\mathrm{Mp} 141.9-142.8^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 7.37-7.20 (m, 4 H), 7.16-7.01 (m, 4 H), 3.41-3.27 (m, 2 H), 3.12-2.97 (m, 2 H); IR (ATR) 2974, 2104, 2091, 1737, $1595,1450,1140,734 \mathrm{~cm}^{-1}$. This material contained ca. $10 \%$ of dimeric bis-carbodiimide that was readily removed in subsequent transformations.


6-(Di-t-butylphosphanyl)-12, 13-dihydrodibenzodiazonine borane (1-42). ${ }^{31}$ Prepared according to General Procedure B from carbodiimide $\mathbf{1 - 4 1}(0.065 \mathrm{~g}, 0.30 \mathrm{mmol})$, di- $t$-butyl phosphine borane $(0.057 \mathrm{~g}, 0.35 \mathrm{mmol})$ and $\mathrm{NaH}(0.014 \mathrm{~g}, 0.35 \mathrm{mmol}, 60 \%$ dispersion $)$ in DMAc $(1.2 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc} /$ hexanes $)$ afforded $\mathbf{1 - 4 2}(0.075 \mathrm{~g}$, $0.20 \mathrm{mmol}, 67 \%$ ) as a colorless crystalline solid: $\mathrm{Mp} 174-177{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.61(\mathrm{~s}, 1 \mathrm{H}), 6.92-6.73(\mathrm{~m}, 5 \mathrm{H}), 6.69(\mathrm{~d}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.63(\mathrm{t}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.49(\mathrm{~d}, 1 \mathrm{H}$, $J=7.6 \mathrm{~Hz}), 3.45-3.26(\mathrm{~m}, 2 \mathrm{H}), 2.97-2.83(\mathrm{~m}, 2 \mathrm{H}), 1.58(\mathrm{~s}, 9 \mathrm{H}), 1.54(\mathrm{~s}, 9 \mathrm{H}), 1.2-0.1(\mathrm{br}, 3 \mathrm{H}) ;$ ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 147.9\left({ }^{3} J_{\mathrm{CP}}=11 \mathrm{~Hz}\right), 146.3\left({ }^{1} J_{\mathrm{CP}}=75 \mathrm{~Hz}\right), 137.7,137.5\left({ }^{3} J_{\mathrm{CP}}=8\right.$ $\mathrm{Hz}), 130.9,130.2,128.4,127.0,126.7,126.6,126.5,122.3,120.3,34.3\left({ }^{1} J_{\mathrm{CP}}=27 \mathrm{~Hz}\right), 33.6\left({ }^{1} J_{\mathrm{CP}}\right.$ $=26 \mathrm{~Hz}), 30.4\left({ }^{2} J_{\mathrm{CP}}=7 \mathrm{~Hz}\right), 28.8,28.2 ;{ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 52.6 ;{ }^{11} \mathrm{~B} \mathrm{NMR}(128 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta-42$; $\mathrm{HRMS}(\mathrm{ESI})^{+} \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{23} \mathrm{H}_{34} \mathrm{~N}_{2} \mathrm{~PB}[\mathrm{M}+\mathrm{H}]$ 381.2631, found 381.2622. The X-ray crystal structure data for this compound has been deposited (CCDC \#1008171).


6-(Dicyclohexylphosphanyl)-12, 13-dihydrodibenzodiazonine borane (1-43). Prepared according to General Procedure $B$ from carbodiimide $\mathbf{1 - 4 1}(0.15 \mathrm{~g}, 0.68 \mathrm{mmol})$, dicyclohexylphosphine borane ( $0.173 \mathrm{~g}, 0.817 \mathrm{mmol}$ ), and $\mathrm{NaH}(0.032 \mathrm{~g}, 0.82 \mathrm{mmol}, 60 \%$ dispersion) in DMAc (3 mL). Purification by chromatography on $\mathrm{SiO}_{2}(10 \% \mathrm{EtOAc} /$ hexane $)$
afforded phosphine borane $\mathbf{1 - 4 3}(0.212 \mathrm{~g}, 0.049 \mathrm{mmol}, 72 \%)$, with a trace amount of inseparable phosphine borane starting material, as a clear oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.25-7.20(\mathrm{~s}, 1$ H), 6.91-6.73 (m, 5H), $6.67(\mathrm{~d}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.60(\mathrm{t}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.42(\mathrm{~d}, 1 \mathrm{H}, J=7.6$ Hz), 3.36-3.23 (m, 2 H), 3.02-2.80 (m, 2 H), 2.47-2.20 (m, 2 H ), 2.15-1.18 (m, 20 H ), 0.9-0.1 (br, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 148.0\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=12 \mathrm{~Hz}\right), 145.9\left(\mathrm{~d},{ }^{1} J_{\mathrm{CP}}=78 \mathrm{~Hz}\right), 137.4(\mathrm{~d}$, $\left.{ }^{3} J_{\mathrm{CP}}=8 \mathrm{~Hz}\right), 130.8,130.4,128.5,126.9,126.7,126.6,126.1,122.5,120.2,32.0,31.63,31.57$, $31.2,30.7,30.0,29.4,29.3,29.1,28.7,27.7,27.0,26.83,26.76,26.72,26.67,26.65,26.6,26.5$, 26.4, 26.3, 26.1, 26.1, 25.8; ${ }^{31} \mathrm{P} \operatorname{NMR}\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 38.3 ;{ }^{11} \mathrm{~B} \mathrm{NMR}\left(128 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ -43.1; IR (ATR) 3334, 2933, 2862, 2378, 1629, 1480, 1325, 1215, 1077, 767, $734 \mathrm{~cm}^{-1}$; HRMS $(\mathrm{ESI})^{+} \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{27} \mathrm{H}_{39} \mathrm{~N}_{2} \mathrm{~PB}[\mathrm{M}+\mathrm{H}] 433.2944$, found 433.2930 .


6-(Ditolylphosphanyl)-12, 13-dihydrodibenzodiazonine borane (1-44). ${ }^{31}$ Prepared according to General Procedure B from carbodiimide 1-41 ( $0.200 \mathrm{~g}, 0.907 \mathrm{mmol}$ ), di-tolyl phosphine borane $(0.249 \mathrm{~g}, 1.09 \mathrm{mmol})$ and $\mathrm{NaH}(0.022 \mathrm{~g}, 0.55 \mathrm{mmol}, 60 \%$ dispersion) in DMAc ( 5 mL ). Purification by chromatography on $\mathrm{SiO}_{2}(10 \% \mathrm{EtOAc} /$ hexanes $)$ afforded phosphine borane 1-44 $(0.319 \mathrm{~g}, 0.712 \mathrm{mmol}, 78 \%)$ as a colorless crystalline solid: Mp 189.7-191.5 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.01(\mathrm{~d}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.98(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.85(\mathrm{~d}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.82$ $(\mathrm{d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.42(\mathrm{~s}, 1 \mathrm{H}), 7.31(\mathrm{td}, 4 \mathrm{H}, J=6.4 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 6.86-6.80(\mathrm{~m}, 3 \mathrm{H}), 6.79-6.72$ $(\mathrm{m}, 2 \mathrm{H}), 6.63-6.54(\mathrm{~m}, 2 \mathrm{H}), 6.42(\mathrm{~d}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.23-3.15(\mathrm{~m}, 1 \mathrm{H}), 2.97-2.86(\mathrm{~m}, 1 \mathrm{H})$, 2.78-2.66(m, 2 H$), 2.43(\mathrm{~s}, 3 \mathrm{H}), 2.41(\mathrm{~s}, 3 \mathrm{H}), 1.5-0.8(\mathrm{br}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$
$148.7\left({ }^{1} J_{\mathrm{CP}}=91 \mathrm{~Hz}\right), 148.0\left({ }^{3} J_{\mathrm{CP}}=13 \mathrm{~Hz}\right), 142.7\left({ }^{4} J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 142.1\left({ }^{4} J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 137.3\left({ }^{3} J_{\mathrm{CP}}=\right.$ $10 \mathrm{~Hz}), 136.9,133.6\left({ }^{2} J_{\mathrm{CP}}=10 \mathrm{~Hz}\right), 133.2\left({ }^{2} J_{\mathrm{CP}}=10 \mathrm{~Hz}\right), 130.7,130.4,129.6\left({ }^{3} J_{\mathrm{CP}}=11 \mathrm{~Hz}\right)$, $128.4,126.6,126.5,125.4,124.2\left({ }^{1} J_{\mathrm{CP}}=64 \mathrm{~Hz}\right), 123.3\left({ }^{1} J_{\mathrm{CP}}=61 \mathrm{~Hz}\right), 122.6,119.6,31.1,29.2$, 21.7, 21.6; ${ }^{31} \mathrm{P}$ NMR (162 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 19.4 ;{ }^{11} \mathrm{~B} \mathrm{NMR} \mathrm{(128} \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-40.4$; IR (ATR) $3340,3026,2378,1637,1486,1356,1128,1045,908,802,734 \mathrm{~cm}^{-1} ;$ HRMS (ESI) ${ }^{+} \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{29} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{P}\left[\mathrm{M}+\mathrm{H}-\mathrm{BH}_{3}\right]$ 435.1990, found 435.1975. An X-ray crystal structure was obtained for this molecule.


## 6-(Di-(3, 5-dimethyl-4-methyoxy)phenylphosphanyl)-12, 13-dihydrodibenzodiazonine

 borane (1-45). ${ }^{31}$ Prepared according to General Procedure B from carbodiimide 1-41 (0.084 g, $0.38 \mathrm{mmol})$, di-(3,5-dimethyl-4-methoxy)phosphine borane ( $0.133 \mathrm{~g}, 0.545 \mathrm{mmol}$ ) and NaH ( $0.018 \mathrm{~g}, 0.46 \mathrm{mmol}, 60 \%$ dispersion) in DMAc ( 3 mL ). Purification by chromatography on $\mathrm{SiO}_{2}$ (10\% EtOAc/hexanes) afforded phosphine borane $\mathbf{1 - 4 5}(0.171 \mathrm{~g}, 0.319 \mathrm{mmol}, 83 \%)$ as a colorless foam: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.79(\mathrm{~d}, 2 \mathrm{H}, J=10.8 \mathrm{~Hz}), 7.54(\mathrm{~d}, 2 \mathrm{H}, J=11.2 \mathrm{~Hz}), 7.43$ $(\mathrm{s}, 1 \mathrm{H}), 6.89-6.83(\mathrm{~m}, 3 \mathrm{H}), 6.80-6.74(\mathrm{~m}, 2 \mathrm{H}), 6.63-6.55(\mathrm{~m}, 2 \mathrm{H}), 6.44(\mathrm{~d}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.79$ (s, 3 H ), $3.76(\mathrm{~s}, 3 \mathrm{H}), 3.27-3.18(\mathrm{~m}, 1 \mathrm{H}), 2.92-2.81(\mathrm{~m}, 1 \mathrm{H}), 2.79-2.68(\mathrm{~m}, 2 \mathrm{H}), 2.36(\mathrm{~s}, 6 \mathrm{H})$, $2.33(\mathrm{~s}, 6 \mathrm{H}), 1.5-0.2(\mathrm{br}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 160.5\left({ }^{4} J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 159.9\left({ }^{4} J_{\mathrm{CP}}=\right.$ $2 \mathrm{~Hz}), 149.2\left({ }^{1} J_{\mathrm{CP}}=91 \mathrm{~Hz}\right), 148.0\left({ }^{3} J_{\mathrm{CP}}=13 \mathrm{~Hz}\right), 137.3\left({ }^{3} J_{\mathrm{CP}}=10 \mathrm{~Hz}\right), 136.6,134.3\left({ }^{2} J_{\mathrm{CP}}=11\right.$ $\mathrm{Hz}), 133.8\left({ }^{2} J_{\mathrm{CP}}=10 \mathrm{~Hz}\right), 131.8\left({ }^{3} J_{\mathrm{CP}}=24 \mathrm{~Hz}\right), 131.8,130.6\left({ }^{3} J_{\mathrm{CP}}=17 \mathrm{~Hz}\right), 128.4,126.7,126.5$, $126.4,125.1,122.6,122.0\left({ }^{1} J_{\mathrm{CP}}=65 \mathrm{~Hz}\right), 120.9\left({ }^{1} J_{\mathrm{CP}}=60 \mathrm{~Hz}\right), 119.4,59.8,59.7,31.2,28.8,16.4$,16.3; ${ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 19.8 ;{ }^{11} \mathrm{~B}$ NMR ( $128 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-38.3$; IR (ATR) 3345 , 2934, 2378, 1637, 1480, 1278, 1114, 917, $734 \mathrm{~cm}^{-1}$; HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{33} \mathrm{H}_{36} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}$ $\left[\mathrm{M}+\mathrm{H}-\mathrm{BH}_{3}\right]$ 523.2514, found 523.2494.


6-(Cyclohexylphenylphosphanyl)-12, 13-dihydrodibenzodiazonine borane (1-46). ${ }^{31}$ Prepared according to general Procedure $B$ from carbodiimide $\mathbf{1 - 4 1}(0.100 \mathrm{~g}, 0.454 \mathrm{mmol})$, phenylcyclohexyl phosphine borane $(0.112 \mathrm{~g}, 0.545 \mathrm{mmol})$ and $\mathrm{NaH}(0.022 \mathrm{~g}, 0.55 \mathrm{mmol}, 60 \%$ dispersion) in DMAc ( 2 mL ). Purification by chromatography on $\mathrm{SiO}_{2}$ (hexanes) to afforded phosphine borane $1-46(0.134 \mathrm{~g}, 0.314 \mathrm{mmol}, 69 \%)$ as a colorless solid that is an equilibrating mixture of rotomers/isomers in solution: Mp 164-166 ${ }^{\circ} \mathrm{C}$; IR (ATR) 3353, 2934, 2385, 1644, 1474, 1435, 1260, 726, $692 \mathrm{~cm}^{-1},{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 8.18-8.09 (m, 2 H ), 7.63-7.48 (m, 3 H ), $7.28(\mathrm{~s}, 0.7 \mathrm{H}), 7.13(\mathrm{~s}, 0.3 \mathrm{H}), 6.93-6.74(\mathrm{~m}, 4 \mathrm{H}), 6.69(\mathrm{~d}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 6.65-6.54(\mathrm{~m}, 2 \mathrm{H})$, $6.49(\mathrm{~d}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.20-3.09(\mathrm{~m}, 0.5 \mathrm{H}), 3.10(\mathrm{dd}, 1 \mathrm{H}, J=9.6 \mathrm{~Hz}, 2.4 \mathrm{~Hz}), 3.04-2.71(\mathrm{~m}$, $1.5 \mathrm{H}), 2.69-2.60(\mathrm{~m}, 1 \mathrm{H}), 2.58-2.43(\mathrm{~m}, 1.5 \mathrm{~Hz}), 2.30-2.15(\mathrm{~m}, 1 \mathrm{H}), 1.97-1.86(\mathrm{~m}, 1 \mathrm{H}), 1.80-$ $1.65(\mathrm{~m}, 3 \mathrm{H}), 1.48-1.17(\mathrm{~m}, 6 \mathrm{H}), 1.1-0.4(\mathrm{br}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 148.1, 148.0, $147.4\left({ }^{1} J_{\mathrm{CP}}=57 \mathrm{~Hz}\right), 137.2,137.1,133.0,132.9,132.1,130.7,130.5,130.4,128.8,128.7,128.7$, $128.6,128.5,127.1,126.8,126.5,126.4,126.3,125.5,124.9,122.6,119.9,119.8,33.5,33.2,33.1$, 31.1, 30.7, 30.2, 28.4, 26.7, 26.5, 26.4, 25.9, 25.2; ${ }^{31} \mathrm{P}$ NMR (162 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 27.1,25.3 ;{ }^{11} \mathrm{~B}$ NMR (128 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta-44.3$; HRMS $(\mathrm{ESI})^{+} m / z$ calcd for $\mathrm{C}_{2} \mathrm{H}_{33} \mathrm{~N}_{2} \mathrm{~PB}[\mathrm{M}+\mathrm{H}] 427.2474$, found 427.2466.


General Procedure C (Phosphine borane deprotection): 6-(Di-tolylphosphanyl)-12, 13dihydrodibenzodiazonine (1-47). ${ }^{31}$ A solution of phosphine borane $\mathbf{1 - 4 4}(0.071 \mathrm{~g}, 0.16 \mathrm{mmol})$ in toluene ( 2 mL ) was treated under an Ar atmosphere with DABCO ( $0.025 \mathrm{~g}, 0.22 \mathrm{mmol})$ and stirred at $60^{\circ} \mathrm{C}$ for 90 min . After cooling to rt , the reaction mixture was purified by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc} / \mathrm{hexanes})$ to afford phosphine 1-47 ( $\left.0.056 \mathrm{~g}, 0.13 \mathrm{mmol}, 81 \%\right)$ as clear sticky oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 7.72-7.48 (m, 4 H$), 7.25-7.19(\mathrm{~m}, 3 \mathrm{H}), 6.88-6.76(\mathrm{~m}, 3 \mathrm{H}), 6.76-$ $6.59(\mathrm{~m}, 4 \mathrm{H}), 6.54(\mathrm{~d}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.38(\mathrm{~d}, 2 \mathrm{H}, J=8.8 \mathrm{~Hz}), 2.83(\mathrm{~d}, 2 \mathrm{H}, J=9.2 \mathrm{~Hz}), 2.38$ (s, 6 H$) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 156.5,139.9,134.3\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=19 \mathrm{~Hz}\right), 131.3\left(\mathrm{~d},{ }^{3} J_{\mathrm{CP}}=11\right.$ Hz ), 130.4, 129.9, 129.8, 126.3, 30.5, 21.4; ${ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-3.7$; HRMS (ESI) ${ }^{+}$ $m / z$ calcd for $\mathrm{C}_{29} \mathrm{H}_{28} \mathrm{~N}_{2} \mathrm{P}[\mathrm{M}+\mathrm{H}] 435.1990$, found 435.1991.


6-(Di-(3, 5-dimethyl-4-methoxy)phenylphosphanyl)-12, 13-dihydrodibenzo]diazonine (148). ${ }^{31}$ Prepared according to General Procedure C from phosphine borane $\mathbf{1 - 4 5}(0.176 \mathrm{~g}, 0.328$ $\mathrm{mmol})$ and $\mathrm{DABCO}(0.147 \mathrm{~g}, 1.31 \mathrm{mmol})$ in toluene $(2 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc} /$ hexanes $)$ afforded phosphine $\mathbf{1 - 4 8}(0.112 \mathrm{~g}, 0.214 \mathrm{mmol}, 65 \%)$ as a sticky oil:
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.60-7.29$ (br m, 4 H ), 6.89-6.50 (br m, 8 H ), 6.25-6.10 (br s, 1 H ), 3.80 (s, 6 H), 3.65-3.15 (br m, 2 H ), 3.00-2.68 (br m, 2 H ), 2.35 ( $\mathrm{s}, 12 \mathrm{H}$ ), ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 158.5,156.7,156.6,135.2,135.0,131.7,130.5,129.5,129.4,126.3,59.7,31.4,29.5$, 16.3; ${ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-3.2$; $\mathrm{HRMS}(\mathrm{ESI})^{+} \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{33} \mathrm{H}_{36} \mathrm{~N}_{2} \mathrm{O}_{2} \mathrm{P}[\mathrm{M}+\mathrm{H}]$ 523.2514, found 523.2494.


6-(Cyclohexylphenylphosphanyl)-12, 13-dihydrodibenzodiazonine (1-49). ${ }^{31}$ Prepared according to General Procedure C from phosphine borane 1-46 ( $0.099 \mathrm{~g}, 0.23 \mathrm{mmol})$ and DABCO $(0.104 \mathrm{~g}, 0.929 \mathrm{mmol})$ in toluene $(2 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \%$ EtOAc/hexanes) afforded phosphine $\mathbf{1 - 4 9}(0.090 \mathrm{~g}, 0.22 \mathrm{mmol}, 94 \%)$ as a colorless oil that is an equilibrating mixture of diastereomers in solution: ${ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.98-7.75(\mathrm{~m}, 2$ H), 7.51-7.38 (m, 3 H), 6.94-6.73 (m, 4 H), 6.74-6.63 (m, 2 H), 6.63-6.51 (m, 2 H), 6.51-6.43 (m, $2 \mathrm{H}), 6.03-6.08(\mathrm{~s}, 0.3 \mathrm{H}), 3.52-3.23(\mathrm{~m}, 1.5 \mathrm{H}), 3.04-2.57(\mathrm{~m}, 2.5 \mathrm{H}), 2.49(\mathrm{~d}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz})$, 2.39-2.24 (m, 1 H$), 1.95-1.28(\mathrm{~m}, 9 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 155.1,149.5,138.6,136.4$, $135.1,135.0,134.7,134.6,133.9,133.8,130.6,130.4,130.1,128.8,128.4,126.6,126.4,126.1$, $125.5,125.1,124.1,121.7,120.3,119.6,35.8,34.8,31.6,31.1,30.7,30.2,29.9,29.8,28.6,28.2$, 27.7, 27.6, 26.8, 26.5, 26.3, 26.1, 25.9; ${ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.6,2.5$; $\mathrm{HRMS}(\mathrm{ESI})^{+} \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{27} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{P}[\mathrm{M}+\mathrm{H}] 413.2147$, found 413.2144.


6-(Di-tolylphosphanyloxide)-12, 13-dihydrodibenzodiazonine (1-50). A solution of phosphine $\mathbf{1 - 4 7}(0.019 \mathrm{~g}, 0.044 \mathrm{mmol})$ in acetone $(1 \mathrm{~mL})$ was treated with hydrogen peroxide $(0.05 \mathrm{~mL}, 0.04$ $\mathrm{mmol}, 3 \% \mathrm{aq} . \mathrm{soln}$ ) at rt . After stirring for 15 min , the reaction mixture was partitioned between water and EtOAc ( 10 mL each). The phases were separated, and the aqueous phase was reextracted with EtOAc ( 10 mL ). The combined organic phases were washed with water ( 15 mL ), brine ( 15 mL ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated to afford phosphine oxide $\mathbf{1 - 5 0}(0.017 \mathrm{~g}, 0.038$ $\mathrm{mmol}, 86 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.10-7.91(\mathrm{~m}, 5 \mathrm{H}), 7.33(\mathrm{dd}, 4 \mathrm{H}, J$ $=7.6 \mathrm{~Hz}, 2.4 \mathrm{~Hz}), 6.91-6.72(\mathrm{~m}, 5 \mathrm{H}), 6.67-6.54(\mathrm{~m}, 2 \mathrm{H}), 6.42(\mathrm{~d}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.16-3.04(\mathrm{~m}$, $1 \mathrm{H}), 3.02-2.90(\mathrm{~m}, 1 \mathrm{H}), 2.81-2.65(\mathrm{~m}, 2 \mathrm{H}), 2.43(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 150.6$, $149.1,147.7,147.5,142.9,142.9,142.8,142.8,137.0,137.0,136.9,132.3,132.2,132.1,132.0$, $130.7,130.3,129.3,129.3,129.2,129.2,128.6,128.3,128.1,127.2,127.0,126.6,126.6,126.5$, $125.8,122.5,120.0,30.9,29.4,21.7 ;{ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 18.2$.


6-(Di-(3, 5-dimethyl-4-methoxy)phenylphosphanyloxide)-12, 13-dihydrodibenzo-
diazonine (1-51). ${ }^{31}$ A solution of phosphine $\mathbf{1 - 4 8}(0.094 \mathrm{~g}, 0.16 \mathrm{mmol})$ in acetone $(1 \mathrm{~mL})$ was treated with hydrogen peroxide ( $0.2 \mathrm{~mL}, 0.2 \mathrm{mmol}, 3 \% \mathrm{aq} . \operatorname{soln}$ ) at rt . After stirring for 15 min , the reaction mixture was partitioned between water and EtOAc ( 20 mL each). The phases were separated, and the aqueous phase was extracted with EtOAc ( 20 mL ). The combined organic phases were washed with water $(25 \mathrm{~mL})$ and brine $(25 \mathrm{~mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated to afford phosphine oxide $\mathbf{1 - 5 1}(0.088 \mathrm{~g}, 0.16 \mathrm{mmol}, 91 \%)$ as a colorless crystalline solid: Mp 236-238 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.90(\mathrm{~d}, 1 \mathrm{H}, J=2.8 \mathrm{~Hz}), 7.81(\mathrm{~d}, 2 \mathrm{H}, J=$ $12.0 \mathrm{~Hz}), 7.76(\mathrm{~d}, 2 \mathrm{H}, J=12.0 \mathrm{~Hz}), 6.89-6.73(\mathrm{~m}, 5 \mathrm{H}), 6.66-6.56(\mathrm{~m}, 2 \mathrm{H}), 6.43(\mathrm{~d}, 1 \mathrm{H}, J=7.6$ Hz ), 3.79 (s, 3 H ), 3.78 ( $\mathrm{s}, 3 \mathrm{H}$ ), 3.21-3.09 (m, 1 H ), 2.98-2.87 (m, 1 H ), 2.81-2.66 (m, 2 H ), 2.34 (s, 12 H ) ${ }^{13}{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 160.7,160.6,160.5,160.5,150.6,149.1,147.8,147.6$, $137.1,137.0,136.7,133.1,133.0,132.8,132.6,131.6,131.5,131.5,131.4,130.7,130.4,128.7$, 126.7, 126.5, 126.5, 126.1, 125.9, 125.6, 125.0, 124.9, 122.6, 119.9, 59.7, 59.7, 31.0, 29.0, 16.3, 16.2; ${ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 18.5$; IR (ATR) 3202, 2934, 1627, 1480, 1286, 1215, 1163, 1115, 994, $758 \mathrm{~cm}^{-1}$; HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{33} \mathrm{H}_{36} \mathrm{~N}_{2} \mathrm{PO}_{3}[\mathrm{M}+\mathrm{H}] 539.2458$, found 539.2443.


6-(Phospholan-1-yl)-12,13-dihydro-5H-dibenzo[1,3]diazonine borane (1-59). Prepared according to General Procedure B from phospholane borane $\mathbf{1 - 5 4}(0.080 \mathrm{~g}, 0.78 \mathrm{mmol})$, carbodiimide $1-41(0.207 \mathrm{~g}, 0.942 \mathrm{mmol})$ and sodium hydride $(0.037 \mathrm{~g}, 0.94 \mathrm{mmol}, 60 \%$ dispersion) in DMAc ( 3 mL ). Purification by was chromatography on $\mathrm{SiO}_{2}$ (hexanes to $15 \%$ EtOAc/hexanes) afforded $\mathbf{1 - 5 9}(0.153 \mathrm{~g}, 0.475 \mathrm{mmol}, 35 \%)$ as a colorless solid: $\mathrm{Mp} 118-121^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.19(\mathrm{~s}, 1 \mathrm{H}), 6.92-6.75(\mathrm{~m}, 5 \mathrm{H}), 6.64(\mathrm{~d}, 1 \mathrm{H}, J=6.4 \mathrm{~Hz}), 6.60(\mathrm{t}$,
$1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.42(\mathrm{~d}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.26(\mathrm{td}, 1 \mathrm{H}, J=10.0 \mathrm{~Hz}, 2.8 \mathrm{~Hz}), 3.20-3.10(\mathrm{~m}, 1 \mathrm{H})$, $2.95(\mathrm{t}, 1 \mathrm{H}, J=10.4 \mathrm{~Hz}, 2.8 \mathrm{~Hz}), 2.83(\mathrm{ddd}, 1 \mathrm{H}, J=12.8 \mathrm{~Hz}, 10.0 \mathrm{~Hz}, 6.4 \mathrm{~Hz}), 2.57(\mathrm{ddd}, 1 \mathrm{H}, J$ $=14.4 \mathrm{~Hz}, 6.8 \mathrm{~Hz}, 6.8 \mathrm{~Hz}), 2.51-2.40(\mathrm{~m}, 1 \mathrm{H}), 2.21-1.97(\mathrm{~m}, 4 \mathrm{H}), 0.83\left(\mathrm{q}, 3 \mathrm{H},{ }^{1} J_{\mathrm{BH}}=82 \mathrm{~Hz}\right)$; ${ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 149.5\left({ }^{1} J_{\mathrm{CP}}=78 \mathrm{~Hz}\right), 147.7\left({ }^{3} J_{\mathrm{CP}}=12 \mathrm{~Hz}\right), 137.2,137.1,136.6$, $130.7,130.4,128.1,126.8,126.7,126.6,125.5,122.6,119.9,30.8,29.8,27.6,27.5\left({ }^{3} J_{\mathrm{CP}}=8 \mathrm{~Hz}\right)$, $29.9\left({ }^{1} J_{\mathrm{CP}}=36 \mathrm{~Hz}\right), 26.1\left({ }^{1} J_{\mathrm{CP}}=36 \mathrm{~Hz}\right), 24.2\left({ }^{1} J_{\mathrm{CP}}=38 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR $\left(161 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 40.1$; ${ }^{11} \mathrm{~B}_{\mathrm{NMR}}\left(128 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-39$; IR (ATR) $3358,2951,2365,1629,1474,1053,751 \mathrm{~cm}^{-1} ; \mathrm{MS}$ $(\mathrm{ESI})^{+} m / z 321,309,221$; HRMS $(\mathrm{ESI})^{+} m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{P}\left[\mathrm{M}+\mathrm{H}-\mathrm{BH}_{3}\right]$ 309.1521, found 309.1514.


6-(2,5-Diphenylphospholan-1-yl)-12,13-dihydro-5H-dibenzo[1,3]diazonine borane (1-60). Prepared according to General Procedure B from phosphine borane $(S, S)-\mathbf{1 - 5 8}^{27}(0.119 \mathrm{~g}, 0.468$ $\mathrm{mmol})$, carbodiimide $1-41(0.086 \mathrm{~g}, 0.39 \mathrm{mmol})$, and sodium hydride $(0.019 \mathrm{~g}, 0.47 \mathrm{mmol}, 60 \%)$ in DMAc ( 2 mL ). Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc} /$ hexane $)$ afforded 1-60 $(0.113 \mathrm{~g}, 0.238 \mathrm{mmol}, 61 \%)$ as a colorless solid: $[\alpha]_{\mathrm{D}}+22\left(\mathrm{c}=0.5, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ; \mathrm{Mp} 184-188{ }^{\circ} \mathrm{C}$; IR (ATR) $3340,3026,3058,2951,2378,1629,1485,907,727 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.67-7.53(\mathrm{~m}, 2 \mathrm{H}), 7.52-7.21(\mathrm{~m}, 8 \mathrm{H}), 6.89-6.73(\mathrm{~m}, 2 \mathrm{H}), 6.71-6.48(\mathrm{~m}, 4 \mathrm{H}), 6.34(\mathrm{~d}, 0.5 \mathrm{H}, J$ $=7.6 \mathrm{~Hz}), 5.56(\mathrm{~d}, 0.5 \mathrm{H}, J=7.2 \mathrm{~Hz}), 4.89-4.73(\mathrm{~m}, 1 \mathrm{H}), 3.96-3.75(\mathrm{~m}, 1 \mathrm{H}), 3.19(\operatorname{app} \mathrm{~d}, 1 \mathrm{H}, J$ $=8.4 \mathrm{~Hz}), 2.91-2.73(\mathrm{~m}, 1 \mathrm{H}), 2.74-2.15(\mathrm{~m}, 6 \mathrm{H}), 1.35-1.15(\mathrm{~m}, 2 \mathrm{H}), 1.1-0.1(\mathrm{br} \mathrm{q}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 147.9\left({ }^{2} J_{\mathrm{CP}}=12 \mathrm{~Hz}\right), 147.5\left({ }^{2} J_{\mathrm{CP}}=12 \mathrm{~Hz}\right), 146.8\left({ }^{1} J_{\mathrm{CP}}=72 \mathrm{~Hz}\right), 146.3\left({ }^{1} J_{\mathrm{CP}}\right.$
$=74 \mathrm{~Hz}), 138.0,137.1,137.0,136.9,136.7,136.5,136.4,135.8\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 136.7\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right)$, $130.4\left({ }^{1} J_{\mathrm{CP}}=73 \mathrm{~Hz}\right), 130.5,130.4,130.0,129.0,128.8,128.7,128.61,128.59,128.54,128.49$, $128.3\left(J_{\mathrm{CP}}=4 \mathrm{~Hz}\right), 128.0\left(J_{\mathrm{CP}}=4 \mathrm{~Hz}\right), 127.6,127.5,127.3,127.1,126.7,126.6,126.5,126.4$, 126.3, 125.4, 122.8, 122.5, 120.5, 120.3, 120.3, $48.0\left({ }^{1} J_{\mathrm{CP}}=29 \mathrm{~Hz}\right), 47.5\left({ }^{1} J_{\mathrm{CP}}=29 \mathrm{~Hz}\right), 45.6\left({ }^{1} J_{\mathrm{CP}}\right.$ $=32 \mathrm{~Hz}), 43.4\left({ }^{1} J_{\mathrm{CP}}=32 \mathrm{~Hz}\right), 38.8,32.8\left({ }^{2} J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 32.5\left({ }^{2} J_{\mathrm{CP}}=6 \mathrm{~Hz}\right), 30.9,30.8,30.5\left({ }^{2} J_{\mathrm{CP}}=\right.$ 6 Hz ), 29.9, 29.0; ${ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 51.7$; HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{31} \mathrm{H}_{33} \mathrm{~N}_{2} \mathrm{~PB}$ $[\mathrm{M}+\mathrm{H}] 475.2469$, found 475.2473.


6-(2,5-Diphenylphospholan-1-yl)-12,13-dihydro-5H-dibenzo[1,3]diazonine (1-61). Prepared according to General Procedure C from phosphine borane 1-60 ( $0.050 \mathrm{~g}, 0.11 \mathrm{mmol}$ ) and DABCO $(0.047 \mathrm{~g}, 0.42 \mathrm{mmol})$ in toluene $(1.5 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \%$ EtOAc/hexane) afforded phosphine 1-61 ( $0.037 \mathrm{~g}, 0.080 \mathrm{mmol}, 76 \%)$ as a colorless oil: ${ }^{31} \mathrm{P}$ NMR (202 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 28.0$.


Bis[6-(2,5-Diphenylphospholan-1-yl)-12,13-dihydro-5H-dibenzo[1,3]diazonine] platinum chloride complex (1-62). A solution of phosphine 1-61 ( $0.169 \mathrm{~g}, 0.38 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$ was treated with $\mathrm{PtCl}_{2}(0.102 \mathrm{~g}, 0.380 \mathrm{mmol})$. The vial was sealed under Ar and stirred at rt for 1
$h$, at which point NMR indicated formation of a complex. The heterogeneous reaction mixture was filtered through a syringe filter, and the eluent was placed in a vapor diffusion chamber with methyl $t$-butyl ether ( 30 mL ). After standing for 9 days, yellow rectangular crystals Pt complex 1-62 formed: ${ }^{31} \mathrm{P}$ NMR ( $\left.202 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 52.9(\mathrm{JPtP}=2532 \mathrm{~Hz})$. An X-ray crystal structure was obtained for this complex.

### 5.3 CHAPTER 2 EXPERIMENTAL PART



3-Methylbicyclo[1.1.0]butane-1-carbonitrile (2-37). ${ }^{97}$ A mixture of 3methylenecyclobutanecarbonitrile 2-35 (10.0 g, 0.107 mol$)$ and $48 \% \mathrm{HBr}(21 \mathrm{~mL}, 0.376 \mathrm{~mol}, 3.5$ eq) was vigorously stirred at room temperature for 24 h . The reaction mixture was partitioned between water and ether ( 100 mL ), the phases were separated, and the aqueous layer was extracted with ether $(2 \times 75 \mathrm{~mL})$. The combined ethereal solution was washed with water $(100 \mathrm{~mL})$ then sat. aq. $\mathrm{NaHCO}_{3}(100 \mathrm{~mL})$. The organic layer was dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and concentrated to afford the bromocyclobutane $\mathbf{2 - 3 6}(13.44 \mathrm{~g}, 77.20 \mathrm{mmol}, 72 \%)$ as a colorless liquid with a $d r$ of 1.7:1. Characteristic signal for the major isomer: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.95(\mathrm{~s}, 3 \mathrm{H})$. Characteristic signal for the minor isomer: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.89(\mathrm{~s}, 3 \mathrm{H})$.

A solution of bromide $\mathbf{2 - 3 6}(13.43 \mathrm{~g}, 77.17 \mathrm{mmol})$ in THF $(70 \mathrm{~mL})$ in a dry 3-necked 250mL round bottom flask fitted with a stopper, septum, and $\mathrm{N}_{2}$ inlet was treated with $\mathrm{NaH}(3.06 \mathrm{~g}$, 77.2 mmol ) in portions over 5 min . The flask was stirred at $35^{\circ} \mathrm{C}$ for 12 h , then at reflux for 5 h . After cooling, the grey/brown reaction mixture was quenched with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(25 \mathrm{~mL})$. The
reaction mixture was then partitioned between diethyl ether $(100 \mathrm{~mL})$ and water $(100 \mathrm{~mL})$, the layers were separated, and the aqueous layer was re-extracted with ether ( 75 mL ). The combined organic layers were washed with water $(100 \mathrm{~mL})$ and brine $(100 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated to afford a yellow-tinted cloudy liquid. The crude material was purified by bulb-tobulb distillation (bath temp $\left.75^{\circ} \mathrm{C}, 2 \mathrm{mmHg}\right)$ ) to afford bicyclobutane 2-37(3.54 g, 38.0 mmol , $49 \%$ ) as a colorless liquid: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.97(\mathrm{~s}, 2 \mathrm{H}), 1.71(\mathrm{~s}, 3 \mathrm{H}), 1.28(\mathrm{~s}, 2$ H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 120.2, 40.5, 37.0, 23.1, 12.7, -3.4 . SAFETY NOTE: The distillation should be conducted using the strongest accessible vacuum source and the minimum possible temperature. Although we encountered no difficulty conducting this distillation, differential scanning calorimetry analysis suggests that it is unsafe to heat this material above $\sim 100$ ${ }^{\circ} \mathrm{C}$.


General Procedure D (Bicyclo[1.1.0]butane hydrophosphination): 3-(Dicyclohexylphosphanyl)-3-methylcyclobutane-1-carbonitrile borane (2-38). ${ }^{97}$ A solution of bicyclobutane 2-37 ( $0.472 \mathrm{~g}, 5.07 \mathrm{mmol}, 1.2 \mathrm{eq}$ ) and dicyclohexyl phosphine borane ( $0.896 \mathrm{~g}, 4.22$ $\mathrm{mmol})$ in DMAc ( 20 mL ) was sparged with Ar for 10 min , cooled to $0^{\circ} \mathrm{C}$, and treated with NaH $(0.203 \mathrm{~g}, 5.07 \mathrm{mmol}, 60 \%$ in oil $)$ in one portion. The reaction mixture was stirred for 10 min at 0 ${ }^{\circ} \mathrm{C}$, then the bath was removed and the solution was warmed to rt over 2 h . The orange/brown reaction mixture was quenched with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(5 \mathrm{~mL})$, and the material was partitioned between water ( 30 mL ) and diethyl ether ( 30 mL ). The aqueous layer was extracted with ether (30 $\mathrm{mL})$. The combined extracts were washed with water $(40 \mathrm{~mL})$ and brine $(40 \mathrm{~mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The crude residue was purified by chromatography on $\mathrm{SiO}_{2}(15 \%$

EtOAc/hexane) to afford cis-2-38 ( $0.634 \mathrm{~g}, 2.08 \mathrm{mmol}, 49 \%)$ and trans-2-38 $(0.197 \mathrm{~g}, 0.645 \mathrm{~g}$, $15 \%$ ), both as colorless solids. cis-2-38: Mp 193-195 ${ }^{\circ} \mathrm{C}$; IR (ATR) 2933, 2386, 1450, 1070, 840 $\mathrm{cm}^{-1}$; MS $\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 306,245,215,172 ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.37$ (pent, $1 \mathrm{H}, J=9.6$ Hz ), 3.22-3.09 (m, 2 H ), 2.08-2.01 (m, 2 H ), 1.91-1.77 (m, 8 H ), 1.77-1.67 (br s, 2 H ), 1.66-1.58 $(\mathrm{m}, 2 \mathrm{H}), 1.55\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} \mathrm{~J}_{\mathrm{PH}}=9.2 \mathrm{~Hz}\right), 1.48-1.31(\mathrm{~m}, 4 \mathrm{H}), 1.30-1.15(\mathrm{~m}, 6 \mathrm{H}), 0.9-0.1(\mathrm{br} \mathrm{q}, 3 \mathrm{H})$; ${ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 35.8 ;{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 120.5\left({ }^{4} J_{\mathrm{CP}}=2.5 \mathrm{~Hz}\right), 36.3$, $32.7\left({ }^{1} J_{\mathrm{CP}}=26 \mathrm{~Hz}\right), 31.5\left({ }^{1} J_{\mathrm{CP}}=29 \mathrm{~Hz}\right), 27.69,27.67,27.66,27.65,27.2,27.11,27.08,27.0$, 25.95, 25.94, 22.1, 22.1, $19.7\left(J_{\mathrm{CP}}=17 \mathrm{~Hz}\right)$; HRMS $\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{18} \mathrm{H}_{34} \mathrm{BNP}[\mathrm{M}+\mathrm{H}]$ 306.2516, found 306.2506. trans-2-38: Mp 165-168 ${ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.32-3.23$ (m, 1 H$), 3.11(\mathrm{dd}, 2 \mathrm{H}, J=10.8 \mathrm{~Hz}, 2.4 \mathrm{~Hz}), 2.16-2.07(\mathrm{~m}, 2 \mathrm{H}), 1.88-1.75(\mathrm{~m}, 8 \mathrm{H}), 1.75-1.61$ $(\mathrm{m}, 2 \mathrm{H}), 1.67\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=9.6 \mathrm{~Hz}\right), 1.47-1.32(\mathrm{~m}, 4 \mathrm{H}), 1.30-1.14(\mathrm{~m}, 8 \mathrm{H}), 0.7-0.0(\mathrm{br} \mathrm{q}, 3 \mathrm{H})$; ${ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 34.3 ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 122.9\left({ }^{4} \mathrm{~J}_{\mathrm{CP}}=1.0 \mathrm{~Hz}\right), 36.4$, $33.0\left({ }^{1} J_{\mathrm{CP}}=26 \mathrm{~Hz}\right), 31.3\left({ }^{1} J_{\mathrm{CP}}=29 \mathrm{~Hz}\right), 27.9,27.1,27.03,26.98,26.9,25.8,25.1\left(J_{\mathrm{CP}}=3 \mathrm{~Hz}\right)$, $17.8\left(J_{\mathrm{CP}}=9 \mathrm{~Hz}\right)$; IR (ATR) 2933, 2390, 1442, 1072, $734 \mathrm{~cm}^{-1} ; \mathrm{HRMS}^{\left(\mathrm{ESI}^{+}\right) \mathrm{m} / z \text { calcd for }}$ $\mathrm{C}_{18} \mathrm{H}_{34} \mathrm{BNP}[\mathrm{M}+\mathrm{H}] 306.2516$, found 306.2502 .


3-(Di-t-butylphosphanyl)-3-methylcyclobutane-1-carbonitrile borane (2-39). ${ }^{97}$ Prepared according to General Procedure D from di-t-butyl phosphine borane ( $0.363 \mathrm{~g}, 2.27 \mathrm{mmol}$ ), bicyclobutane 2-37 ( $0.264 \mathrm{~g}, 2.84 \mathrm{mmol}$ ), and $\mathrm{NaH}(0.100 \mathrm{~g}, 2.49 \mathrm{mmol}, 60 \%$ dispersion in oil) in DMAc ( 8 mL ). Purification by chromatography on $\mathrm{SiO}_{2}$ (10-30\% EtOAc in hexanes) afforded cis-2-39 ( $0.327 \mathrm{~g}, 1.29 \mathrm{mmol}, 57 \%$ ) and trans-2-39 $(0.147 \mathrm{~g}, 0.58 \mathrm{mmol}, 26 \%)$, both as colorless
solids. cis-2-39: $\mathrm{Mp} 125-127{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.47-3.31(\mathrm{~m}, 3 \mathrm{H}), 2.14-2.02(\mathrm{~m}$, $2 \mathrm{H}), 1.83\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=8.8 \mathrm{~Hz}\right), 1.33(\mathrm{~s}, 9 \mathrm{H}), 1.30(\mathrm{~s}, 9 \mathrm{H}), 0.61\left(\mathrm{brq}, 3 \mathrm{H},{ }^{1} J_{\mathrm{BH}}=98 \mathrm{~Hz}\right),{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 120.6\left({ }^{4} J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 38.1,35.8\left({ }^{1} J_{\mathrm{CP}}=20 \mathrm{~Hz}\right), 35.1\left({ }^{1} J_{\mathrm{CP}}=23 \mathrm{~Hz}\right)$, 29.6, $23.7\left(J_{\mathrm{CP}}=4 \mathrm{~Hz}\right), 20.1\left(J_{\mathrm{CP}}=18 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 54.7$; IR (ATR) 2974, 2378, 2235, 1480, $1072 \mathrm{~cm}^{-1} ; \mathrm{MS}^{\left(\mathrm{ESI}^{+}\right) ~} \mathrm{~m} / \mathrm{z} 240,223,208,196,162,142,114 ; \mathrm{HRMS}^{2}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{14} \mathrm{H}_{30} \mathrm{BNP}[\mathrm{M}+\mathrm{H}]$ 254.2203, found 254.2197. trans-2-39: Mp 53-56 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.48(\operatorname{appq}, 2 \mathrm{H}, J=12.4 \mathrm{~Hz}), 3.27-3.17(\mathrm{~m}, 1 \mathrm{H}), 2.09\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=9.2\right), 2.13-$ $2.05(\mathrm{~m}, 2 \mathrm{H}), 1.35(\mathrm{~s}, 9 \mathrm{H}), 1.32(\mathrm{~s}, 9 \mathrm{H}), 0.53\left(\mathrm{brq}, 3 \mathrm{H},{ }^{1} J_{\mathrm{BH}}=98 \mathrm{~Hz}\right),{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 123.5,37.5,36.9\left({ }^{1} J_{\mathrm{CP}}=20 \mathrm{~Hz}\right), 35.1\left({ }^{1} J_{\mathrm{CP}}=23 \mathrm{~Hz}\right), 29.6,26.5\left(J_{\mathrm{CP}}=4 \mathrm{~Hz}\right), 19.3\left(J_{\mathrm{CP}}\right.$ $=15 \mathrm{~Hz}) ;{ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 53.6$.


3-(Cyclohexyl(phenyl)phosphanyl)-3-methylcyclobutane-1-carbonitrile borane (2-40). ${ }^{97}$ Prepared according to General Procedure D from racemic phenyl cyclohexyl phosphine borane ( $0.275 \mathrm{~g}, 1.33 \mathrm{mmol}$ ), bicyclobutane $\mathbf{2 - 3 7}(0.136 \mathrm{~g}, 1.47 \mathrm{mmol}, 1.20 \mathrm{eq})$, and $\mathrm{NaH}(0.059 \mathrm{~g}, 1.5$ mmol, 1.2 eq$)$ in DMAc ( 8 mL ). Purification by chromatography on $\mathrm{SiO}_{2}$ ( $15-40 \% \mathrm{EtOAc} /$ hexane $)$ afforded cis-2-40 ( $0.189 \mathrm{~g}, 0.632 \mathrm{mmol}, 47 \%$ ) as a colorless solid and trans-2-40 (0.092 g, 0.31 mmol, $23 \%$ ) as a colorless oil. cis-2-40: Mp $137-139^{\circ} \mathrm{C}$; IR (ATR) 2934, 2397, 1448, 1072, 751 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.68(\mathrm{td}, 2 \mathrm{H}, J=8.4,1.6 \mathrm{~Hz}), 7.55-7.42(\mathrm{~m}, 3 \mathrm{H}), 3.30$ (pent, $1 \mathrm{H}, J=9.2 \mathrm{~Hz}$ ), 3.13 (ddd, $1 \mathrm{H}, J=12.8 \mathrm{~Hz}, 10.0 \mathrm{~Hz}, 7.6 \mathrm{~Hz}$ ), $2.90(\mathrm{ddd}, 1 \mathrm{H}, J=13.6 \mathrm{~Hz}, 11.6$ $\mathrm{Hz}, 9.6 \mathrm{~Hz}), 2.41-2.28(\mathrm{~m}, 1 \mathrm{H}), 2.10(\mathrm{ddd}, 1 \mathrm{H}, J=16.0 \mathrm{~Hz}, 4.4 \mathrm{~Hz}, 3.2 \mathrm{~Hz}), 1.93-1.83(\mathrm{~m}, 1 \mathrm{H})$, $1.79-1.55(\mathrm{~m}, 5 \mathrm{H}), 1.51-1.18(\mathrm{~m}, 5 \mathrm{H}), 1.42\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=11.2 \mathrm{~Hz}\right), 1.1-0.3(\mathrm{br} \mathrm{m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$

NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 132.9\left(J_{\mathrm{CP}}=7 \mathrm{~Hz}\right), 131.5\left(J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 128.8,128.7$, 125.0, 124.5, 120.7, 34.8, 34.2, $32.2\left({ }^{1} J_{\mathrm{CP}}=30 \mathrm{~Hz}\right), 30.3\left({ }^{1} J_{\mathrm{CP}}=33 \mathrm{~Hz}\right), 26.6,26.5,26.4,26.3,25.7,20.9,18.3$ $\left({ }^{2} J_{\mathrm{CP}}=18 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 32.0 ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 300,286,231,209,173,150 ;$ HRMS ( $\mathrm{ESI}^{+}$) $m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{BNP}[\mathrm{M}+\mathrm{H}], 300.2047$, found 300.2060. trans-2-40: IR (ATR) 2939, 2858, 2384, 2239, 1437, 1109, 1066, $742 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.70(\mathrm{td}, 2 \mathrm{H}$, $J=8.0 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 7.54-7.41(\mathrm{~m}, 3 \mathrm{H}), 3.32$ (dddd, $1 \mathrm{H}, J=9.6 \mathrm{~Hz}, 9.6 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 7.2 \mathrm{~Hz}), 3.15$ (app q, $1 \mathrm{H}, J=12.0 \mathrm{~Hz}$ ), $2.71(\mathrm{td}, 1 \mathrm{H}, J=13.2 \mathrm{~Hz}, 11.2 \mathrm{~Hz}), 2.42-2.29(\mathrm{~m}, 1 \mathrm{H}), 2.23$ (ddd, 1 H , $J=12.4 \mathrm{~Hz}, 6.8 \mathrm{~Hz}, 6.4 \mathrm{~Hz}), 1.91-1.78(\mathrm{~m}, 2 \mathrm{H}), 1.77-1.62(\mathrm{~m}, 3 \mathrm{H}), 1.53\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=10.4\right.$ $\mathrm{Hz}), 1.49-1.38(\mathrm{~m}, 1 \mathrm{H}), 1.38-1.15(\mathrm{~m}, 5 \mathrm{H}), 1.2-0.2(\mathrm{br} \mathrm{m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $133.1\left(J_{\mathrm{CP}}=8 \mathrm{~Hz}\right), 131.7\left(J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 128.8,128.7,124.7,124.2,122.6,35.7\left(J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 34.9$, $32.3\left({ }^{1} J_{\mathrm{CP}}=28 \mathrm{~Hz}\right), 30.6\left({ }^{1} J_{\mathrm{CP}}=33 \mathrm{~Hz}\right), 26.6\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 26.5,26.4,26.3,26.2,25.6,24.0\left(J_{\mathrm{CP}}\right.$ $=3 \mathrm{~Hz}), 17.0\left(J_{\mathrm{CP}}=7 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P} \operatorname{NMR}\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 30.2 ; \operatorname{IR}(\mathrm{ATR}) 2939,2858,2384,2239$, 1437, 1109, 1066, $742 \mathrm{~cm}^{-1}$


3-(n-Butyl(furan-2-yl)phosphanyl)-3-methylcyclobutane-1-carbonitrile borane (2-41). ${ }^{97}$ Prepared according to General Procedure D from racemic furyl butyl phosphine borane ( 0.261 g , $1.54 \mathrm{mmol})$, bicyclobutane $\mathbf{2 - 3 7}(0.172 \mathrm{~g}, 1.8 \mathrm{mmol}, 1.20 \mathrm{eq})$, and $\mathrm{NaH}(0.074 \mathrm{~g}, 1.8 \mathrm{mmol}, 1.2$ eq, $60 \%$ in oil) in DMAc ( 8 mL ). Purification by chromatography on $\mathrm{SiO}_{2}$ ( $15-40 \%$ EtOAc/hexane) afforded cis-2-41 (0.115 g, 0.437 mmol, 28\%) and trans-2-41 (0.052 g, 0.20 mmol , 13\%), both as colorless oils. cis-2-41: IR (ATR) 2956, 2933, 2386, 2240, 1456, 1128, 1072, 757 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.74(\mathrm{~s}, 1 \mathrm{H}), 7.18-7.15(\mathrm{~m}, 1 \mathrm{H}), 6.53-6.50(\mathrm{~m}, 1 \mathrm{H}), 3.28$
(pent, $1 \mathrm{H}, J=9.2 \mathrm{~Hz}$ ), $2.88(\mathrm{td}, 1 \mathrm{H}, 13.2 \mathrm{~Hz}, 9.2 \mathrm{~Hz}), 2.69(\mathrm{td}, 1 \mathrm{H}, 13.6 \mathrm{~Hz}, 9.2 \mathrm{~Hz}), 2.19-1.98$ $(\mathrm{m}, 3 \mathrm{H}), 1.77-1.66(\mathrm{~m}, 1 \mathrm{H}), 1.60-1.47(\mathrm{~m}, 1 \mathrm{H}), 1.44\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=14.4 \mathrm{~Hz}\right), 1.35$ (sextet, 2 H , $J=7.2 \mathrm{~Hz}), 1.28-1.13(\mathrm{~m}, 1 \mathrm{H}), 0.87(\mathrm{t}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}), 1.1-0.0(\mathrm{br} \mathrm{q}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 149.1\left({ }^{3} J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 143.2\left({ }^{1} J_{\mathrm{CP}}=61 \mathrm{~Hz}\right), 125.8\left({ }^{2} J_{\mathrm{CP}}=18 \mathrm{~Hz}\right), 121.1\left({ }^{4} J_{\mathrm{CP}}=2 \mathrm{~Hz}\right)$, $111.3\left({ }^{3} J_{\mathrm{CP}}=8 \mathrm{~Hz}\right), 34.1,33.9,32.7\left({ }^{1} J_{\mathrm{CP}}=35 \mathrm{~Hz}\right), 25.0,24.1\left({ }^{2} J_{\mathrm{CP}}=14 \mathrm{~Hz}\right), 22.1,\left({ }^{3} J_{\mathrm{CP}}=3 \mathrm{~Hz}\right)$, $19.0\left({ }^{1} J_{\mathrm{CP}}=36 \mathrm{~Hz}\right), 17.1\left({ }^{2} J_{\mathrm{CP}}=16 \mathrm{~Hz}\right), 13.6 ;{ }^{31} \mathrm{P} \mathrm{NMR}\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 18.6 ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ 264, 250, 228, 200, 184, 154, 110; HRMS (ESI ${ }^{+}$) $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{24}$ BNOP [M+H] 264.1683, found 264.1685. trans-2-41: IR (ATR) 2958, 2380, 2238, 1555, 1458, 1212, 1128, 1071, 1049, 1007, $909,732 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.71(\mathrm{~s}, 1 \mathrm{H}), 7.16(\mathrm{~d}, 1 \mathrm{H}, J=1.2 \mathrm{~Hz}), 6.51$ $(\mathrm{t}, 1 \mathrm{H}, J=1.6 \mathrm{~Hz}), 2.98$ (pent, $1 \mathrm{H}, J=7.6 \mathrm{~Hz}$ ), 2.89-2.77 (m, 2 H), 2.32-2.12 (m, 2 H$), 2.08$ (dd, $1 \mathrm{H}, J=14.8,4.0 \mathrm{~Hz}), 1.76-1.63(\mathrm{~m}, 1 \mathrm{H}), 1.58-1.44(\mathrm{~m}, 1 \mathrm{H}), 1.53\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=12.0 \mathrm{~Hz}\right), 1.35$ (pent, $2 \mathrm{H}, J=7.2 \mathrm{~Hz}$ ), 1.29-1.18 (m, 1 H$), 0.86(\mathrm{t}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}), 1.1-0.1(\mathrm{br} \mathrm{q}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 148.9\left({ }^{3} J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 143.2\left({ }^{1} J_{\mathrm{CP}}=61 \mathrm{~Hz}\right), 125.5\left({ }^{2} J_{\mathrm{CP}}=18 \mathrm{~Hz}\right), 122.1,111.4$ $\left({ }^{3} J_{\mathrm{CP}}=8 \mathrm{~Hz}\right), 35.0,34.9\left({ }^{3} J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 33.2\left({ }^{1} J_{\mathrm{CP}}=33 \mathrm{~Hz}\right), 25.0,24.1\left({ }^{2} J_{\mathrm{CP}}=13 \mathrm{~Hz}\right), 18.8\left({ }^{1} J_{\mathrm{CP}}=\right.$ $36 \mathrm{~Hz}), 16.2\left({ }^{3} J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 13.6 ;{ }^{31} \mathrm{P}$ NMR $\left(202 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 17.6.


3-(Di-p-tolylphosphanyl)-3-methylcyclobutane-1-carbonitrile borane (2-42). ${ }^{97}$ Prepared according to General Procedure D from di-p-tolylphosphine borane ( $0.500 \mathrm{~g}, 2.19 \mathrm{mmol}$ ), bicyclobutane 2-37 ( $0.245 \mathrm{~g}, 2.63 \mathrm{mmol}$ ), and $\mathrm{NaH}(0.105 \mathrm{~g}, 2.63 \mathrm{mmol})$ in DMAc ( 10 mL ). Purification by chromatography on $\mathrm{SiO}_{2}(20-40 \% \mathrm{EtOAc} /$ hexane $)$ afforded cis-2-42 ( $0.341 \mathrm{~g}, 1.06$ $\mathrm{mmol}, 48 \%)$ as a colorless solid and trans-2-42 $(0.199 \mathrm{~g}, 0.620 \mathrm{mmol}, 28 \%)$ as a colorless foam.
cis-2-42: Mp 162-163 ${ }^{\circ} \mathrm{C}$; IR (ATR) 2392, 1429, 1066, $807 \mathrm{~cm}^{-1},{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $7.53(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.51(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.29(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.28(\mathrm{~d}, 2 \mathrm{H}, J=8.0$ $\mathrm{Hz}), 3.31$ (pent, $1 \mathrm{H}, J=9.2 \mathrm{~Hz}), 2.88-2.78(\mathrm{~m}, 2 \mathrm{H}), 2.40(\mathrm{~s}, 6 \mathrm{H}), 2.34-2.26(\mathrm{~m}, 2 \mathrm{H}), 1.54(\mathrm{~d}, 3$ $\left.\mathrm{H},{ }^{3} J_{\mathrm{PH}}=14.0 \mathrm{~Hz}\right), 1.2-0.5(\mathrm{br} \mathrm{q}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.2\left(J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 133.4$ $\left(J_{\mathrm{CP}}=7 \mathrm{~Hz}\right), 129.8\left(J_{\mathrm{CP}}=8 \mathrm{~Hz}\right), 123.3,122.8,120.5\left({ }^{4} J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 34.7,33.4\left({ }^{1} J_{\mathrm{CP}}=34 \mathrm{~Hz}\right), 22.1$ $\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 21.5,18.2\left({ }^{2} J_{\mathrm{CP}}=17 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P} \operatorname{NMR}\left(202 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 27.7 ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 240$, 219, 167, 139, 122; HRMS $\left(\mathrm{ESI}^{+}\right) m / z$ calcd for $\mathrm{C}_{20} \mathrm{H}_{26} \mathrm{BNP}[\mathrm{M}+\mathrm{H}] 322.1890$, found 322.1879 . trans-2-42: IR (ATR) 2956, 2386, 1498, 1066, $814 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.51$ (d, 2 $\mathrm{H}, J=8.4 \mathrm{~Hz}), 7.49(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.28(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.27(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 3.07$ (ddd, $2 \mathrm{H}, J=13.6 \mathrm{~Hz}, 13.6 \mathrm{~Hz}, 10.0 \mathrm{~Hz}), 2.80(\mathrm{ddd}, 1 \mathrm{H}, J=17.0 \mathrm{~Hz}, 9.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}), 2.46-2.36$ (m, 2 H ), $2.39(\mathrm{~s}, 6 \mathrm{H}), 1.60\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} \mathrm{~J}_{\mathrm{PH}}=12.8 \mathrm{~Hz}\right), 1.4-0.5(\mathrm{br} \mathrm{q}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 125 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 142.10,142.08,133.2,133.1,129.8,129.7,123.6,123.2,122.1\left({ }^{4} J_{\mathrm{CP}}=0.5 \mathrm{~Hz}\right), 35.7$, $33.2\left({ }^{1} J_{\mathrm{CP}}=31 \mathrm{~Hz}\right), 25.7\left(J_{\mathrm{CP}}=1.5 \mathrm{~Hz}\right), 21.5\left(J_{\mathrm{CP}}=1.0 \mathrm{~Hz}\right), 16.9\left({ }^{2} J_{\mathrm{CP}}=6 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR $(202$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 28.2.


3-(Bis(3,5-dimethylphenyl)phosphanyl)-3-methylcyclobutane-1-carbonitrile borane (2-43).
Prepared according to General Procedure D from di-m-xylyl phosphine borane ( $5.00 \mathrm{~g}, 19.5$ mmol ), bicyclobutane 2-37 ( $3.1 \mathrm{~mL}, 29 \mathrm{mmol}$ ), and sodium hydride $(0.780 \mathrm{~g}, 19.5 \mathrm{mmol}, 60 \%$ dispersion) in DMAc (100 mL). Purification by chromatography on $\mathrm{SiO}_{2}$ (10-15-20-25\% EtOAc in hexane) afforded cis-2-43 (3.24 g, 9.28 mmol, 48\%) and trans-2-43 (1.66 g, $4.76 \mathrm{mmol}, 24 \%)$, both as colorless solids: cis-2-43: Mp 162-165 ${ }^{\circ} \mathrm{C}$; IR (ATR) 2932, 2390, 1448, 1128, 1066, 692
$\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.23(\mathrm{~s}, 2 \mathrm{H}), 7.21(\mathrm{~s}, 2 \mathrm{H}), 7.16(\mathrm{~s}, 2 \mathrm{H}), 3.31$ (pent, $1 \mathrm{H}, J=$ $9.5 \mathrm{~Hz}), 2.85(\mathrm{td}, 2 \mathrm{H}, J=10.0 \mathrm{~Hz}, 2.5 \mathrm{~Hz}), 2.35(\mathrm{~s}, 12 \mathrm{H}), 2.24-2.37(\mathrm{~m}, 2 \mathrm{H}), 1.53\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}\right.$ $=14.5 \mathrm{~Hz}), 1.2-0.5(\mathrm{br} \mathrm{q}, 3 \mathrm{H}) ;{ }^{31} \mathrm{P} \operatorname{NMR}\left(202 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 29.0 ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 138.1\left(J_{\mathrm{CP}}=10 \mathrm{~Hz}\right), 137.8,133.4\left(J_{\mathrm{CP}}=8 \mathrm{~Hz}\right), 131.6\left(J_{\mathrm{CP}}=8 \mathrm{~Hz}\right), 129.0\left(J_{\mathrm{CP}}=4 \mathrm{~Hz}\right)$, 128.7, 127.6, 127.4, $125.4\left({ }^{1} J_{\mathrm{CP}}=53 \mathrm{~Hz}\right), 120.2,43.8\left({ }^{3} J_{\mathrm{CP}}=25 \mathrm{~Hz}\right), 36.9,35.8,21.4$, $19.2\left({ }^{3} J_{\mathrm{CP}}=19\right.$ $\mathrm{Hz}), 13.0 ;{ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 30.6 ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 350,259,212,155 ; \mathrm{HRMS}^{\left(\mathrm{ESI}^{+}\right)}$ $m / z$ calcd for $\mathrm{C}_{22} \mathrm{H}_{30} \mathrm{BNP}[\mathrm{M}+\mathrm{H}]$ 350.2203, found 350.2190. trans-2-43: Mp 130-134 ${ }^{\circ} \mathrm{C}$; IR (ATR) 2937, 2386, 1450, 1128, 852, $692 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.19$ ( $\mathrm{s}, 2 \mathrm{H}$ ), 7.17 (s, 2 H), $7.14(\mathrm{~s}, 2 \mathrm{H}), 3.08(\mathrm{td}, 2 \mathrm{H}, J=13.5 \mathrm{~Hz}, 10.0 \mathrm{~Hz}), 2.83-2.75(\mathrm{~m}, 1 \mathrm{H}), 2.46-2.37(\mathrm{~m}, 2$ H), $2.33(\mathrm{~s}, 12 \mathrm{H}), 1.62\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} \mathrm{~J}_{\mathrm{PH}}=13.0 \mathrm{~Hz}\right), 1.4-0.5(\mathrm{br} \mathrm{q}, 3 \mathrm{H}) ;{ }^{31} \mathrm{P}$ NMR ( $202 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 26.8 ;{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 138.6,138.5,133.4,133.4,130.8,130.7,126.7,126.3$, $122.2\left({ }^{4} J_{\mathrm{CP}}=1 \mathrm{~Hz}\right), 35.8,33.1\left({ }^{1} J_{\mathrm{CP}}=31 \mathrm{~Hz}\right), 25.8\left(J_{\mathrm{CP}}=1 \mathrm{~Hz}\right), 21.4,21.4,16.9\left(J_{\mathrm{CP}}=6 \mathrm{~Hz}\right)$.


## (3-(Bis(4-methoxyphenyl)phosphanyl)-3-methylcyclobutane-1-carbonitrile borane (2-44). ${ }^{97}$

 Prepared according to General Procedure D from di-p-methoxy phosphine borane ( $0.222 \mathrm{~g}, 0.854$ mmol ), and bicyclobutane $\mathbf{2 - 3 7}(0.254 \mathrm{~g}, 2.73 \mathrm{mmol})$ and $\mathrm{NaH}(0.038 \mathrm{~g}, 0.94 \mathrm{mmol})$ in DMAc ( 6 $\mathrm{mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(10-30 \% \mathrm{EtOAc} /$ hexane $)$ afforded cis-2-44 (0.130 $\mathrm{g}, 0.368 \mathrm{mmol}, 43 \%)$ as a colorless foam and trans-2-44 ( $0.048 \mathrm{~g}, 0.14 \mathrm{mmol}, 16 \%)$ as a colorless oil. $\underline{\text { cis-2-44: }}$ IR (ATR) 2394, 2385, 1600, 1499, 1245, 1109, 1021, $798 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.57(\mathrm{t}, 4 \mathrm{H}, J=9.2 \mathrm{~Hz}), 6.99(\mathrm{dd}, 4 \mathrm{H}, J=8.8 \mathrm{~Hz}, 1.6 \mathrm{~Hz}), 3.84(\mathrm{~s}, 6 \mathrm{H}), 3.31$ (pent, 1 $\mathrm{H}, J=9.2 \mathrm{~Hz}), 2.79(\mathrm{qd}, 2 \mathrm{H}, J=10.0 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 2.34-2.24(\mathrm{~m}, 2 \mathrm{H}), 1.50\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=14.4\right.$Hz ), 1.2-0.3 (br m, 3 H ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 162.23,162.21,135.1,135.0,120.6\left({ }^{4} J_{\mathrm{CP}}\right.$ $=2 \mathrm{~Hz}), 117.6,117.0,114.7,114.6,55.4,34.6,33.6\left({ }^{1} J_{\mathrm{CP}}=34 \mathrm{~Hz}\right), 22.1\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 18.2\left(J_{\mathrm{CP}}=\right.$ $17 \mathrm{~Hz}) ;{ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 26.2$; MS ( $\mathrm{ESI}^{+}$) $\mathrm{m} / \mathrm{z} 318,316,259,247$; HRMS (ESI ${ }^{+}$) $m / z$ calcd for $\mathrm{C}_{20} \mathrm{H}_{26} \mathrm{BNO}_{2} \mathrm{P}[\mathrm{M}+\mathrm{H}]$ 354.1789, found 354.1793. trans-2-44: ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.53(\mathrm{tt}, 4 \mathrm{H}, J=9.2 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 6.96(\mathrm{dd}, 4 \mathrm{H}, J=8.8 \mathrm{~Hz}, 1.6 \mathrm{~Hz}), 3.84(\mathrm{~s}, 6 \mathrm{H}), 3.04$ ( td, $2 \mathrm{H}, 13.6 \mathrm{~Hz}, 10.0 \mathrm{~Hz}$ ), 2.74 (ddd, $1 \mathrm{H}, J=17.6 \mathrm{~Hz}, 9.6 \mathrm{~Hz}, 7.6 \mathrm{~Hz}), 2.39(\mathrm{td}, 2 \mathrm{H}, J=13.2$ $\mathrm{Hz}, 7.6 \mathrm{~Hz}), 1.59\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=12.8 \mathrm{~Hz}\right), 1.4-0.5(\mathrm{br} \mathrm{m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $162.09,162.07,134.8,134.7,122.1,117.8,117.2,114.6,114.5,55.3,35.6,33.4\left({ }^{1} J_{\mathrm{CP}}=32 \mathrm{~Hz}\right)$, $25.7\left(J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 16.8\left(J_{\mathrm{CP}}=6 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P} \operatorname{NMR}\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 26.3$; IR (ATR) 2940, 2379, $2237,1902,1594,1501,1289,1253,1180,1061,1024,801 \mathrm{~cm}^{-1}$


## 3-(Di([1,1'-biphenyl]-4-yl)phosphanyl)-3-methylcyclobutane-1-carbonitrile borane

 (2-45). ${ }^{97}$ Prepared according to General Procedure D from biphenylphosphine borane ( 0.300 g , 0.850 mmol ), bicyclobutane 2-37(0.095 g, 1.0 mmol ), $\mathrm{NaH}(0.036 \mathrm{~g}, 0.89 \mathrm{mmol})$ in DMAc (4 $\mathrm{mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(15-30 \% \mathrm{EtOAc} /$ hexane $)$ afforded cis-2-45 (0.158 $\mathrm{g}, 0.35 \mathrm{mmol}, 42 \%)$ and trans-2-45 ( $0.083 \mathrm{~g}, 0.186 \mathrm{mmol}, 22 \%$ ), both as sticky waxes. cis-2-45: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.79-7.70(\mathrm{~m}, 8 \mathrm{H}), 7.64-7.59(\mathrm{~m}, 4 \mathrm{H}), 7.48(\mathrm{t}, 4 \mathrm{H}, J=7.5 \mathrm{~Hz})$, 7.40 (t, $2 \mathrm{H}, J=7.2 \mathrm{~Hz}$ ), 3.38 (pent, $1 \mathrm{H}, J=9.2 \mathrm{~Hz}$ ), $2.94(\mathrm{dd}, 2 \mathrm{H}, J=12.8 \mathrm{~Hz}, 9.6 \mathrm{~Hz}), 2.42-$ $2.35(\mathrm{~m}, 2 \mathrm{H}), 1.61\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} \mathrm{~J}_{\mathrm{PH}}=14.4 \mathrm{~Hz}\right), 1.4-0.7(\mathrm{brq}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $144.6\left(J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 139.7,134.0,129.1,128.3,127.7,127.3,125.3,124.7,120.5,34.8,33.5\left({ }^{1} J_{\mathrm{CP}}\right.$$=33 \mathrm{~Hz}), 22.3\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 18.3\left({ }^{2} J_{\mathrm{CP}}=17 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR $\left(202 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 29.0$; IR (ATR) 3027, 2384, 1077, 911, $726 \mathrm{~cm}^{-1}$; MS (ESI ${ }^{+}$) $\mathrm{m} / \mathrm{z} 423,324,292,224,124 ; \mathrm{HRMS}^{\left(\mathrm{ESI}^{+}\right)} \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{30} \mathrm{H}_{30} \mathrm{BNP}[\mathrm{M}+\mathrm{H}]$ 446.2203, found 446.2211. trans-2-45: IR (ATR) 3029, 2980, 2374, 2236, 1597, 1483, 1389, 1109, 1061, 1006, 909, 834, 736, $695 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.76-$ $7.68(\mathrm{~m}, 8 \mathrm{H}), 7.61(\mathrm{~d}, 4 \mathrm{H}, J=7.5 \mathrm{~Hz}), 7.49(\mathrm{t}, 4 \mathrm{H}, J=7.5 \mathrm{~Hz}), 7.42(\mathrm{t}, 2 \mathrm{H}, J=7.5 \mathrm{~Hz}), 3.20$ (dd, $2 \mathrm{H}, J=13.6 \mathrm{~Hz}, 10.0 \mathrm{~Hz}), 3.03-2.90(\mathrm{~m}, 1 \mathrm{H}), 2.49(\mathrm{td}, 2 \mathrm{H}, J=14.0 \mathrm{~Hz}, 7.0 \mathrm{~Hz}), 1.71(\mathrm{~d}, 3$ $\left.\mathrm{H},{ }^{3} J_{\mathrm{PH}}=12.8 \mathrm{~Hz}\right), 1.4-0.7(\mathrm{br} \mathrm{m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 144.5\left(J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 139.6$, $133.8,129.1,128.4,127.6,127.2,125.5,125.0,122.1,35.7,33.4\left({ }^{1} J_{\mathrm{CP}}=31 \mathrm{~Hz}\right), 25.8,17.1\left({ }^{2} J_{\mathrm{CP}}\right.$ $=6 \mathrm{~Hz}) ;{ }^{31} \mathrm{P}$ NMR $\left(202 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 29.1$.

cis-2-47, 38\%


## 3-(Bis((S)-2-(6-methoxynaphthalen-2-yl)propyl)phosphanyl)-3-methylcyclobutane-1-

carbonitrile (2-47). ${ }^{97}$ Prepared according to General Procedure D from di-(S)-naproxen phosphine borane ( $3.77 \mathrm{~g}, 8.49 \mathrm{mmol}$ ), bicyclobutane nitrile $\mathbf{2 - 3 7}(0.790 \mathrm{~g}, 8.49 \mathrm{mmol})$ and $\mathrm{NaH}(0.340 \mathrm{~g}$, $8.49 \mathrm{mmol}, 60 \%$ dispersion $)$ in DMAc ( 40 mL ). Purification by automated chromatography ( 120 $\mathrm{g} \mathrm{SiO}_{2}$ cartridge, 5-10-15-50\% EtOAc in hexane) afforded cis-2-47 (1.73 g, $3.22 \mathrm{mmol}, 38 \%$ ) and trans-2-47 (0.773 g, $1.43 \mathrm{mmol}, 17 \%)$, both as colorless solids. cis-2-47: Mp $162-163{ }^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}-$ $24.1\left(c 0.220, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.73(\mathrm{appt}, 3 \mathrm{H}, J=9.2 \mathrm{~Hz}), 7.52(\mathrm{~d}, 2 \mathrm{H}$, $J=8.4 \mathrm{~Hz}), 7.36(\mathrm{dd}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 7.22-7.16(\mathrm{~m}, 2 \mathrm{H}), 7.14(\mathrm{~d}, 1 \mathrm{H}, J=2.4 \mathrm{~Hz}), 7.09$ $(\mathrm{dd}, 1 \mathrm{H}, J=8.8 \mathrm{~Hz}, 2.4 \mathrm{~Hz}), 7.02(\mathrm{~d}, 1 \mathrm{H}, J=2.4 \mathrm{~Hz}), 6.92(\mathrm{dd}, 1 \mathrm{H}, J=8.4,1.6 \mathrm{~Hz}), 3.95(\mathrm{~s}, 3$ H), $3.89(\mathrm{~s}, 3 \mathrm{H}), 3.51-3.40(\mathrm{~m}, 1 \mathrm{H}), 3.10(\mathrm{pent}, 1 \mathrm{H}, J=9.2 \mathrm{~Hz}), 2.97-2.86(\mathrm{~m}, 1 \mathrm{H}), 2.79$ (app
$\mathrm{q}, 1 \mathrm{H}, J=12.0 \mathrm{~Hz}), 2.66(\mathrm{td}, 1 \mathrm{H}, J=13.2 \mathrm{~Hz}, 10.0 \mathrm{~Hz}), 1.98-1.87(\mathrm{~m}, 2 \mathrm{H}), 1.70-1.59(\mathrm{~m}, 2 \mathrm{H})$, $1.46(\mathrm{~d}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}), 1.21\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=11.0 \mathrm{~Hz}\right), 1.26-1.14(\mathrm{~m}, 1 \mathrm{H}), 1.08(\mathrm{ddd}, 1 \mathrm{H}, J=$ 15.2 Hz, 9.2 Hz, 6.4 Hz ), $0.92(\mathrm{~d}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}), 1.0-0.3(\mathrm{br} \mathrm{m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 157.7,157.5,141.5,141.4,141.04,141.00,133.6,133.3,129.2,129.03,128.98,128.8$, $127.5,127.2,126.0,125.6,125.5,124.9,120.6\left({ }^{4} J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 105.67,105.57,55.4,55.3,35.2$, 35.0, 33.6, $31.9\left({ }^{1} J_{\mathrm{CP}}=31 \mathrm{~Hz}\right), 30.0\left({ }^{1} J_{\mathrm{CP}}=31 \mathrm{~Hz}\right), 29.3\left({ }^{1} J_{\mathrm{CP}}=27 \mathrm{~Hz}\right), 25.6,25.0,20.5,17.8$ $\left({ }^{2} J_{\mathrm{CP}}=18 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR (162 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 28.7$; IR (ATR) 2933, 2386, 1603, 1450, 1215, 837 $\mathrm{cm}^{-1} ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) m / z 538,500,447,338 ; \operatorname{HRMS}\left(\mathrm{ESI}^{+}\right) m / z$ calcd for $\mathrm{C}_{34} \mathrm{H}_{42} \mathrm{BNO}_{2} \mathrm{P}[\mathrm{M}+\mathrm{H}]$ 538.3041, found 538.3025. trans-2-47: $[\alpha]_{\mathrm{D}}+27.1\left(c=0.240, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$; IR (ATR) 2933, 2395, $1605,1028,857 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.78-7.72(\mathrm{~m}, 3 \mathrm{H}), 7.53(\mathrm{dd}, 2 \mathrm{H}, J=8.4$ $\mathrm{Hz}, 6.0 \mathrm{~Hz}), 7.38(\mathrm{dd}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}, 1.6 \mathrm{~Hz}), 7.24-7.15(\mathrm{~m}, 3 \mathrm{H}), 7.11(\mathrm{dd}, 1 \mathrm{H}, J=8.8 \mathrm{~Hz}, 2.4$ $\mathrm{Hz}), 7.05(\mathrm{~d}, 1 \mathrm{H}, J=2.4 \mathrm{~Hz}), 6.89(\mathrm{dd}, 1 \mathrm{H}, J=8.4,2.4 \mathrm{~Hz}), 3.96(\mathrm{~s}, 3 \mathrm{H}), 3.90(\mathrm{~s}, 3 \mathrm{H}), 3.53-$ 3.43 (m, 1 H), 3.21-3.13 (m, 1 H), 2.94-2.80 (m, 2 H), 2.49 (dd, $1 \mathrm{H}, J=13.2 \mathrm{~Hz}, 10.4 \mathrm{~Hz}$ ), 2.11 $(\mathrm{td}, 1 \mathrm{H}, J=12.4 \mathrm{~Hz}, 7.2 \mathrm{~Hz}), 1.94(\mathrm{td}, 1 \mathrm{H}, J=14.0 \mathrm{~Hz}, 9.6 \mathrm{~Hz}), 1.70-1.55(\mathrm{~m}, 3 \mathrm{H}), 1.47(\mathrm{~d}, 3$ $\mathrm{H}, J=6.4 \mathrm{~Hz}), 1.35\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=10.4 \mathrm{~Hz}\right), 1.13-1.04(\mathrm{~m}, 1 \mathrm{H}), 0.93(\mathrm{~d}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}), 1.0-0.3$ (br m, 3 H ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 157.7,157.5,141.6,141.5,140.9,140.9,133.7$, 133.3, $129.2,129.0,128.8,127.6,127.2,126.1,125.6,125.3,124.8,122.5,119.3,105.7,105.6,55.4$, $55.3,35.2,35.0,34.7,34.3,34.2,34.2,32.1\left({ }^{1} J_{\mathrm{CP}}=31 \mathrm{~Hz}\right), 30.4\left({ }^{1} J_{\mathrm{CP}}=32 \mathrm{~Hz}\right), 29.6\left({ }^{1} J_{\mathrm{CP}}=27\right.$ $\mathrm{Hz}), 25.7\left(J_{\mathrm{CP}}=11 \mathrm{~Hz}\right), 24.9\left(J_{\mathrm{CP}}=7 \mathrm{~Hz}\right), 23.6\left(J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 16.5\left({ }^{2} J_{\mathrm{CP}}=7 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR $(162$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 27.5.


2-50

General Procedure E (Bicyclobutane nitrile synthesis through transnitrilation): 3-Phenylbicyclo[1.1.0]butane-1-carbonitrile (2-50). ${ }^{189}$ A solution of dibromocyclopropane (8.28 $\mathrm{g}, 25.5 \mathrm{mmol})$ in diethyl ether ( 60 mL ) was cooled to $-78{ }^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$ in a $250-\mathrm{mL}$ round bottom flask. Methyl lithium ( $15.6 \mathrm{~mL}, 25.5 \mathrm{mmol}, 1.63 \mathrm{M}$ in $\mathrm{Et}_{2} \mathrm{O}$ ) was added dropwise via syringe over 15 min . After stirring for 1 h at $-78^{\circ} \mathrm{C}$, $t$-butyl lithium ( $15.0 \mathrm{~mL}, 25.5 \mathrm{mmol}, 1.70 \mathrm{M}$ in pentane) was added dropwise via syringe over 15 min . After stirring for an additional 1 h at $-78^{\circ} \mathrm{C}$, dimethyl malononitrile ( $2.40 \mathrm{~g}, 25.5 \mathrm{mmol}$ ) was added as a solution in THF ( 10 mL ), and the reaction mixture turned from a clear orange into an opaque yellow solution. The cooling bath was removed and the solution was warmed to rt over 1.5 h , affording an orange/brown suspension. The mixture was quenched with sat. aq. $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{~mL})$, and partitioned between water ( 30 mL ) and diethyl ether $(50 \mathrm{~mL})$. The aqueous layer was extracted with ether $(50 \mathrm{~mL})$. The combined organic extracts were washed with water $(60 \mathrm{~mL})$ and brine $(60 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Chromatography of the crude black residue on $\mathrm{SiO}_{2}(5-20 \% \mathrm{EtOAc} /$ hexane $)$ afforded bicyclobutane 2-50 (1.71 g, $11.0 \mathrm{mmol}, 43 \%$ ) as a yellow oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.45-$ $7.29(\mathrm{~m}, 5 \mathrm{H}), 2.71(\mathrm{~s}, 2 \mathrm{H}), 1.68(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 131.8,128.9,128.5$, $128.0,126.5,118.8,37.9,29.5,5.8$.


2-53
2,2-Dimethylbicyclo[1.1.0]butane-1-carbonitrile (2-53). ${ }^{98}$ Prepared according to General

Procedure E from dibromocyclopropane ( $10.0 \mathrm{~g}, 18.1 \mathrm{mmol}$ ), methyl lithium ( $22.2 \mathrm{~mL}, 36.2$ $\mathrm{mmol}, 1.63 \mathrm{M}$ in $\mathrm{Et}_{2} \mathrm{O}$ ), $t$-butyl lithium ( $21.3 \mathrm{~mL}, 36.2 \mathrm{mmol}, 1.70 \mathrm{M}$ in pentane), and dimethylmalononitrile ( $4.43 \mathrm{~g}, 47.0 \mathrm{mmol}, 1.30 \mathrm{eq}$ ) in $\mathrm{Et}_{2} \mathrm{O}(80 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ (5-20\% EtOAc in hexanes) afforded 2-53 (0.694 g, $6.48 \mathrm{mmol}, 18 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 2.30(\mathrm{dd}, 1 \mathrm{H}, J=3.0 \mathrm{~Hz}, 3.0 \mathrm{~Hz}), 2.21(\mathrm{t}, 1 \mathrm{H}, J=$ $3.0 \mathrm{~Hz}), 1.86(\mathrm{t}, 1 \mathrm{H}, J=3.0 \mathrm{~Hz}), 1.35(\mathrm{~s}, 3 \mathrm{H}), 0.96(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 120.6$, 53.8, 35.3, 24.9, 22.0, 13.8, 2.6.


3-(Dicyclohexylphosphanyl)-3-phenylcyclobutane-1-carbonitrile (2-54). ${ }^{97}$ Prepared according to General Procedure D from nitrile 2-50 ( $0.083 \mathrm{~g}, 0.54 \mathrm{mmol}, 1.3 \mathrm{eq})$, dicyclohexyl phosphine borane ( $0.087 \mathrm{~g}, 0.41 \mathrm{mmol}$ ) and $\mathrm{NaH}(0.016 \mathrm{~g}, 0.41 \mathrm{mmol}, 60 \%$ in oil) in DMAc ( 2 mL ). Purification by chromatography on $\mathrm{SiO}_{2}(15 \% \mathrm{EtOAc} /$ hexane $)$ afforded less polar cis-2-54 (0.043 $\mathrm{g}, 0.117 \mathrm{mmol}, 28 \%)$ and the more polar trans-2-54 ( $0.058 \mathrm{~g}, 0.158 \mathrm{mmol}, 38 \%$ ), both as colorless solids. cis-2-54: Mp 198-201 ${ }^{\circ} \mathrm{C}$; IR (ATR) 2934, 2844, 2391, 1448, $1072 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.45(\mathrm{t}, 2 \mathrm{H}, J=6.9 \mathrm{~Hz}), 7.35(\mathrm{t}, 3 \mathrm{H}, J=7.8 \mathrm{~Hz}), 3.64-3.48(\mathrm{~m}, 2 \mathrm{H}), 3.07$ (pent, $1 \mathrm{H}, J=9.2 \mathrm{~Hz}), 2.76(\mathrm{tt}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}, 3.0 \mathrm{~Hz}), 1.95-1.53(\mathrm{~m}, 11 \mathrm{H}), 1.32-1.05(\mathrm{~m}, 11 \mathrm{H}), 1.1-$ $0.1(\mathrm{br} \mathrm{m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 139.1\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 128.9,128.5,127.8\left(J_{\mathrm{CP}}=2\right.$ $\mathrm{Hz}), 127.6,127.6,124.5,120.2\left({ }^{4} J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 42.1\left({ }^{1} J_{\mathrm{CP}}=21 \mathrm{~Hz}\right), 36.9,36.6,31.5\left({ }^{1} J_{\mathrm{CP}}=28 \mathrm{~Hz}\right)$, 27.4, 27.2, 27.1, 27.0, 26.9, 26.8, 25.9, $20.6\left({ }^{2} J_{\mathrm{CP}}=18 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR ( $\left.162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 39.6$; MS (ESI $\left.{ }^{+}\right) m / z 385\left(\mathrm{M}+\mathrm{NH}_{4}{ }^{+}\right), 368,273,215 ; \mathrm{HRMS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{23} \mathrm{H}_{36} \mathrm{BNP}[\mathrm{M}+\mathrm{H}]$ 368.2673, found 368.2696. trans-2-54: Mp 205-207 ${ }^{\circ} \mathrm{C}$; IR (ATR) 2930, 2856, 2377, 2239, 1446,

1064, $760 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.41(\mathrm{t}, 2 \mathrm{H}, J=7.2 \mathrm{~Hz}$ ), $7.33(\mathrm{app} \mathrm{t}, 1 \mathrm{H}, J=7.6$ $\mathrm{Hz}), 7.13(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 3.67(\mathrm{pent}, 1 \mathrm{H}, J=9.2 \mathrm{~Hz}), 3.38(\mathrm{td}, 2 \mathrm{H}, J=13.2 \mathrm{~Hz}, 10.0 \mathrm{~Hz})$, 2.98-2.86 (m, 2 H), 1.95-1.49 (m, 10 H$), 1.47-1.32(\mathrm{~m}, 2 \mathrm{H}), 1.32-1.02(\mathrm{~m}, 10 \mathrm{H}), 0.9-0.0(\mathrm{br} \mathrm{m}$, $3 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 143.0\left({ }^{2} J_{\mathrm{CP}}=6 \mathrm{~Hz}\right), 128.8,127.4,126.3\left({ }^{3} J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 121.8$, $41.4\left({ }^{1} J_{\mathrm{CP}}=21 \mathrm{~Hz}\right), 37.3,31.3\left({ }^{1} J_{\mathrm{CP}}=29 \mathrm{~Hz}\right), 28.6,27.6,27.2,27.1,26.7,26.6,25.8,17.6\left({ }^{2} J_{\mathrm{CP}}=\right.$ $3 \mathrm{~Hz}) ;{ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 35.9$.

cis-2-55

trans-2-55

3-(Di-p-tolylphosphanyl)-3-phenylcyclobutane-1-carbonitrile (2-55). ${ }^{97}$ Prepared according to General Procedure D from nitrile 2-50 ( $0.621 \mathrm{~g}, 4.01 \mathrm{mmol}$ ), di-p-tolyl phosphine borane ( 0.684 $\mathrm{g}, 3.00 \mathrm{mmol}$ ) and $\mathrm{NaH}(0.120 \mathrm{~g}, 3.0 \mathrm{mmol})$ in DMAc ( 15 mL ). Purification by automated chromatography ( $80 \mathrm{~g} \mathrm{SiO}_{2}$ cartridge, 5-10-50\% EtOAc in hexane) afforded cis-2-55 (0.365 g, $0.95 \mathrm{mmol}, 32 \%)$ as a light yellow solid and trans-2-55 ( $0.457 \mathrm{~g}, 1.19 \mathrm{mmol}, 40 \%$ ) as a white crystalline solid. cis-2-55: Mp $68-72{ }^{\circ} \mathrm{C}$; IR (ATR) 2368, 1600, 1497, 1446, 1190, 1063, 806, 655 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.48(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.45(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.24-7.13$ (m, 7 H ), 7.09-7.02 (m, 2 H ), 3.30 (ddd, $2 \mathrm{H}, J=13.2 \mathrm{~Hz}, 10.0 \mathrm{~Hz}, 2.4 \mathrm{~Hz}$ ), 3.03 (pent, $1 \mathrm{H}, J=$ $9.2 \mathrm{~Hz}), 2.91-2.84(\mathrm{~m}, 2 \mathrm{H}), 2.38(\mathrm{~s}, 6 \mathrm{H}), 1.5-0.5(\mathrm{br} \mathrm{m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $142.2\left(J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 137.6,133.9\left(J_{\mathrm{CP}}=6 \mathrm{~Hz}\right), 129.4\left(J_{\mathrm{CP}}=11 \mathrm{~Hz}\right), 128.9\left(J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 127.8$, 127.8, 127.7, 127.5, 127.4, 122.6, 122.1, $120.2\left({ }^{4} J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 43.7\left({ }^{1} J_{\mathrm{CP}}=27 \mathrm{~Hz}\right), 35.8,21.5,19.1$ $\left(J_{\mathrm{CP}}=19 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR (202 MHz, $\left.\mathrm{CDCl}_{3}\right)$ § 29.0. trans-2-55: Mp 172-174 ${ }^{\circ} \mathrm{C}$; IR (ATR) 2933, 2391, 1604, 1450, $1064 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.44(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.42(\mathrm{~d}, 2$
$\mathrm{H}, J=8.0 \mathrm{~Hz}), 7.25-7.13(\mathrm{~m}, 7 \mathrm{H}), 6.66(\mathrm{dd}, 2 \mathrm{H}, J=6.4 \mathrm{~Hz}, 1.6 \mathrm{~Hz}), 3.55($ pent, $1 \mathrm{H}, J=9.2 \mathrm{~Hz}$ ), 3.45-3.34 (m, 2 H$), 2.96-2.87(\mathrm{~m}, 2 \mathrm{H}), 2.38(\mathrm{~s}, 6 \mathrm{H}), 1.3-0.7$ (br m, 3 H$),{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 142.3\left(J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 142.2\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 134.2\left(J_{\mathrm{CP}}=98 \mathrm{~Hz}\right), 129.2$, 129.1, $128.0\left(J_{\mathrm{CP}}=\right.$ $2 \mathrm{~Hz}), 127.6\left(J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 127.1,122.1,121.7,121.5,42.9\left({ }^{1} J_{\mathrm{CP}}=24 \mathrm{~Hz}\right), 36.7,21.5,17.7 ;{ }^{31} \mathrm{P}$ NMR (162 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 25.7$; IR (ATR) 2933, 2383, 1437, 1065, 757, $744 \mathrm{~cm}^{-1} ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ 384, 370, 345, 155; HRMS (ESI $\left.{ }^{+}\right) m / z$ calcd for $\mathrm{C}_{25} \mathrm{H}_{28} \mathrm{BNP}[\mathrm{M}+\mathrm{H}]$ 384.2047, found 384.2038.


2-56
3-(Dicyclohexylphosphanyl)-2,2-dimethylcyclobutane-1-carbonitrile (2-56). ${ }^{97}$ Prepared according to General Procedure D from bicyclobutane 2-53 ( $0.148 \mathrm{~g}, 1.37 \mathrm{mmol}$ ), dicyclohexyl phosphine borane ( $0.225 \mathrm{~g}, 1.06 \mathrm{mmol}), \mathrm{NaH}(0.055 \mathrm{~g}, 1.37 \mathrm{mmol}, 60 \%$ in mineral oil $)$ in DMAc ( 7.5 mL ). Purification by chromatography on $\mathrm{SiO}_{2}(10-30 \% \mathrm{EtOAc} /$ hexane $)$ afforded 2-56 (0.144 $\mathrm{g}, 0.339 \mathrm{mmol}, 34 \%)$ as a colorless solid: Mp 135-139 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 2.90$ (app t, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}$ ), 2.68 (pent, $1 \mathrm{H}, J=10.8 \mathrm{~Hz}$ ), $2.50(\mathrm{td}, 1 \mathrm{H}, J=10.8 \mathrm{~Hz}, 8.0 \mathrm{~Hz}$ ), 2.31 (dtd, $1 \mathrm{H}, J=10.4 \mathrm{~Hz}, 8.0 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 2.05-1.96(\mathrm{~m}, 1 \mathrm{H}), 1.94-1.55(\mathrm{~m}, 11 \mathrm{H}), 1.51(\mathrm{~s}, 3 \mathrm{H})$, $1.28(\mathrm{~s}, 3 \mathrm{H}), 1.37-1.14(\mathrm{~m}, 10 \mathrm{H}), 1.1-0.0(\mathrm{br} \mathrm{m}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 118.4\left({ }^{4} J_{\mathrm{CP}}\right.$ $=6 \mathrm{~Hz}), 45.3\left({ }^{3} J_{\mathrm{CP}}=1 \mathrm{~Hz}\right), 36.3\left({ }^{1} J_{\mathrm{CP}}=25 \mathrm{~Hz}\right), 34.3\left({ }^{1} J_{\mathrm{CP}}=23 \mathrm{~Hz}\right), 33.6\left(J_{\mathrm{CP}}=10 \mathrm{~Hz}\right), 33.3\left(J_{\mathrm{CP}}\right.$ $=9 \mathrm{~Hz}), 31.2,27.8,27.53,27.46,27.4,27.3,27.14,27.05,27.00,26.95,26.9,26.0,25.9\left(J_{\mathrm{CP}}=19\right.$ $\mathrm{Hz})$, 25.4, $21.9\left(J_{\mathrm{CP}}=4 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 26.7$; IR (ATR) 2924, 2365, 1448, $1066 \mathrm{~cm}^{-1}$; MS (ESI $\left.{ }^{+}\right) m / z 320,306,268,241,215,273,152 ; \operatorname{HRMS}^{2}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{19} \mathrm{H}_{36} \mathrm{BNP}[\mathrm{M}+\mathrm{H}] 320.2673$, found 320.2684.

(3-(Di-p-tolylphosphanyl)-2,2-dimethylcyclobutane-1-carbonitrile borane (2-57). ${ }^{97}$ Prepared according to General Procedure D from bicyclobutane 2-53 ( $0.049 \mathrm{~g}, 0.46 \mathrm{mmol}$ ), di- $p$ tolylphosphine borane $(0.081 \mathrm{~g}, 0.36 \mathrm{mmol})$, and $\mathrm{NaH}(0.018 \mathrm{~g}, 0.46 \mathrm{mmol}, 60 \%$ in mineral oil) in DMAc ( 2.5 mL ). Purification by chromatography on $\mathrm{SiO}_{2}$ (5-25\% EtOAc/hexane) afforded $0.032 \mathrm{~g}(0.095 \mathrm{mmol}, 27 \%)$ of $\mathbf{2 - 5 7}$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.59(\mathrm{~d}, 1 \mathrm{H}$, $J=8.0 \mathrm{~Hz}), 7.56(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.45(\mathrm{~d}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.42(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.27(\mathrm{dd}$, $2 \mathrm{H}, J=8.0 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 7.22(\mathrm{dd}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}, 1.6 \mathrm{~Hz}), 3.04(\mathrm{ddd}, 1 \mathrm{H}, J=11.2 \mathrm{~Hz}, 8.4 \mathrm{~Hz}$, $8.4 \mathrm{~Hz}), 2.93(\mathrm{td}, 1 \mathrm{H}, J=10.0 \mathrm{~Hz}, 0.8 \mathrm{~Hz}), 2.73$ (pent, $1 \mathrm{H}, J=11.2 \mathrm{~Hz}$ ), $2.48(\mathrm{ddd}, 1 \mathrm{H}, J=19.2$ Hz, $8.4 \mathrm{~Hz}, 2.4 \mathrm{~Hz}$ ), $2.39(\mathrm{~s}, 3 \mathrm{H}), 2.37(\mathrm{~s}, 3 \mathrm{H}), 1.27(\mathrm{~s}, 3 \mathrm{H}), 1.16(\mathrm{~s}, 3 \mathrm{H}), 1.2-0.6(\mathrm{br} \mathrm{m}, 3 \mathrm{H}) ;$ ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 141.9\left({ }^{1} J_{\mathrm{CP}}=35 \mathrm{~Hz}\right), 141.9\left({ }^{1} J_{\mathrm{CP}}=35 \mathrm{~Hz}\right), 132.5\left({ }^{2} J_{\mathrm{CP}}=15 \mathrm{~Hz}\right)$, $132.4\left({ }^{2} J_{\mathrm{CP}}=14 \mathrm{~Hz}\right), 129.7,129.7,129.6,129.6,126.3,125.8,125.3,124.7,118.5\left({ }^{4} J_{\mathrm{CP}}=6 \mathrm{~Hz}\right)$, $45.3,39.9\left({ }^{1} J_{\mathrm{CP}}=28 \mathrm{~Hz}\right), 33.8\left(J_{\mathrm{CP}}=25 \mathrm{~Hz}\right), 30.9,24.9,21.5,21.5,20.9\left({ }^{3} J_{\mathrm{CP}}=5 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 13.7$; IR (ATR) 2974, 2391, 2235, 1450, 1103, 1059, $734 \mathrm{~cm}^{-1}$; MS (ESI ${ }^{+}$) $m / z$ 336, 322, 282, 249, 228, 209; HRMS (ESI $\left.{ }^{+}\right) m / z$ calcd for $\mathrm{C}_{21} \mathrm{H}_{28} \mathrm{BNP}[\mathrm{M}+\mathrm{H}]$ 336.2047, found 336.2061 .


2-58

3-(Ditolylphosphanyl)-3-methylcyclobutane-1-carbaldehyde borane (2-58). ${ }^{97}$ A solution of
nitrile cis-2-42 $(0.755 \mathrm{~g}, 2.35 \mathrm{mmol})$ in THF $(12 \mathrm{~mL})$ was cooled to $0^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$. DIBAL ( 3.5 $\mathrm{mL}, 3.5 \mathrm{mmol}, 1.5 \mathrm{eq}, 1 \mathrm{M}$ in hexanes) was added via syringe over 2 min . The ice bath was removed, and the resulting solution was stirred at rt for 16 h , then was quenched with $2 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$ $(10 \mathrm{~mL})$. The solution was stirred for 1 h and gradually became an opaque white suspension. The suspension was partitioned between water $(40 \mathrm{~mL})$ and $\mathrm{Et}_{2} \mathrm{O}(40 \mathrm{~mL})$, and the aqueous layer was extracted with $\mathrm{Et}_{2} 0(40 \mathrm{~mL})$. The combined extracts were washed with water ( 40 mL ), brine ( 40 $\mathrm{mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The crude oil was purified by chromatography on $\mathrm{SiO}_{2}$ ( $20 \%$ EtOAc in hexanes) to afford aldehyde 2-58 ( $0.484 \mathrm{~g}, 1.49 \mathrm{mmol}, 64 \%$ ) as a colorless foam: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.24(\mathrm{~d}, 1 \mathrm{H}, J=2.0 \mathrm{~Hz}), 7.54(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.52(\mathrm{~d}, 2 \mathrm{H}, J$ $=8.4 \mathrm{~Hz}), 7.26(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.25(\mathrm{~d}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.26($ pent d, $1 \mathrm{H}, J=9.2 \mathrm{~Hz}, 2.4$ $\mathrm{Hz}), 2.88-2.74(\mathrm{~m}, 2 \mathrm{H}), 2.41(\mathrm{~s}, 6 \mathrm{H}), 2.13-2.04(\mathrm{~m}, 2 \mathrm{H}), 1.61\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} \mathrm{~J}_{\mathrm{PH}}=14.8 \mathrm{~Hz}\right), 1.2-0.4$ (br q, 3 H ); ${ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 201.1,141.8,133.5,133.4,129.6,129.5,124.0,123.5$, $40.3\left(J_{\mathrm{CP}}=12 \mathrm{~Hz}\right), 31.9\left({ }^{1} J_{\mathrm{CP}}=32 \mathrm{~Hz}\right), 30.6,23.6\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 21.5 ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) m / z 324,311$, 295, 213; ${ }^{31} \mathrm{P}$ NMR ( $162 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 27.1; $\mathrm{HRMS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{20} \mathrm{H}_{27} \mathrm{BOP}[\mathrm{M}+\mathrm{H}]$ 325.1887, found 325.1892.


3-(Di-m-xylylphosphanyl)-3-methylcyclobutane-1-carbaldehyde borane (2-59). A solution of cis-2-43 (1.00 g, 2.86 mmol$)$ in THF $(25 \mathrm{~mL})$ was cooled to $0^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$ in a 100 mL flask. The solution was treated with DIBAL ( $4.29 \mathrm{~mL}, 4.29 \mathrm{mmol}, 1 \mathrm{M}$ in hexane, 1.5 eq ) via syringe over 20 s . The cooling bath was removed, and the resulting colorless solution was stirred at rt for 24 h .

The reaction was quenched with $2 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}(15 \mathrm{~mL})$. The resulting solution was stirred for 30 min, was extracted with ether ( $2 \times 50 \mathrm{~mL}$ ). The combined extracts were washed with water (30 $\mathrm{mL})$, brine $(30 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated. The crude material was purified by chromatography on $\mathrm{SiO}_{2}$ (15-25\% EtOAc/hexane) to afford aldehyde 2-59 ( $0.556 \mathrm{~g}, 1.58 \mathrm{mmol}$, $55 \%$ ) as a colorless solid: $\mathrm{Mp} 133-134{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 9.24(\mathrm{~d}, 1 \mathrm{H}, J=2.0$ Hz), 7.25 (s, 2 H$), 7.22(\mathrm{~s}, 2 \mathrm{H}), 7.13(\mathrm{~s}, 2 \mathrm{H}), 3.24$ (pent, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}), 2.87-2.75(\mathrm{~m}, 2 \mathrm{H})$, $2.33(\mathrm{~s}, 12 \mathrm{H}), 2.14-2.05(\mathrm{~m}, 2 \mathrm{H}), 1.60\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=14.0 \mathrm{~Hz}\right), 1.2-0.3(\mathrm{br} \mathrm{q}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 201.2,138.4,138.3,133.18,133.16,131.1,131.0,127.1,126.6,40.3\left(J_{\mathrm{CP}}=\right.$ $11 \mathrm{~Hz}), 31.7\left({ }^{1} J_{\mathrm{CP}}=31 \mathrm{~Hz}\right), 30.8,23.8\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 21.6,13.8 ;{ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 28.4 IR (ATR) 2933, 2373, 1713, 1128, 1069, 839, $692 \mathrm{~cm}^{-1} ; \mathrm{MS}^{\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 367,353,341,241, ~}$ 212; $\mathrm{HRMS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{22} \mathrm{H}_{31} \mathrm{BOP}[\mathrm{M}+\mathrm{H}] 353.2200$, found 353.2180 .


3-(Ditolyl-3-methylcyclobutyl)methanol borane (2-60). ${ }^{97}$ A solution of aldehyde 2-58 (0.608 g, $1.88 \mathrm{mmol}, 1.00 \mathrm{eq})$ in $\mathrm{MeOH}(10 \mathrm{~mL})$ was cooled to $0^{\circ} \mathrm{C}$. Sodium borohydride $(0.074 \mathrm{~g}, 2.0$ mmol, 1.1 eq ) was added to the colorless solution in one portion. Bubbling ensued, and then subsided after ca. 10 min . The reaction mixture was stirred at rt for 2 h , and quenched with 1 M $\mathrm{HCl}(5 \mathrm{~mL})$. The contents of the flask were then partitioned between ether $(30 \mathrm{~mL})$ and water ( 30 $\mathrm{mL})$, and the aqueous phase was extracted with ether $(30 \mathrm{~mL})$. The combined organic phases were washed with water $(30 \mathrm{~mL})$ and brine $(30 \mathrm{~mL})$, were $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by chromatography on $\mathrm{SiO}_{2}(25 \% \mathrm{EtOAc} /$ hexane $)$ to afford alcohol 2-60 ( $0.503 \mathrm{~g}, 1.54$
$\mathrm{mmol}, 82 \%)$ as a colorless solid: $\mathrm{Mp} 108-110{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.54(\mathrm{~d}, 2 \mathrm{H}, J$ $=8.0 \mathrm{~Hz}), 7.52(\mathrm{~d}, 2 \mathrm{H}, J=8.8 \mathrm{~Hz}), 7.23(\mathrm{~d}, 4 \mathrm{H}, J=6.8 \mathrm{~Hz}), 3.19(\mathrm{~d}, 2 \mathrm{H}, J=6.4 \mathrm{~Hz}), 2.74$ (pent, $1 \mathrm{H}, J=8.0 \mathrm{~Hz}), 2.37(\mathrm{~s}, 6 \mathrm{H}), 2.38-2.23(\mathrm{~m}, 2 \mathrm{H}), 2.01-1.89(\mathrm{~m}, 2 \mathrm{H}), 1.54\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=14.8\right.$ Hz ), 1.2-0.3 (br q, 3 H ); ${ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 141.49$, 141.47, 133.54, 133.45, 129.4, 129.3, 124.7, 124.2, 66.2, 33.1, $32.2\left({ }^{2} J_{\mathrm{CP}}=13 \mathrm{~Hz}\right), 31.0\left({ }^{1} J_{\mathrm{CP}}=32 \mathrm{~Hz}\right), 23.5\left(J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 21.5$; ${ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 27.3.


3-((Di-m-xylylphosphanyl)-3-methylcyclobutyl)methanol borane (2-61). A solution of aldehyde 2-59 $(0.414 \mathrm{~g}, 1.18 \mathrm{mmol})$ in $\mathrm{MeOH}(8 \mathrm{~mL})$ and THF $(3 \mathrm{~mL})$ was treated with sodium borohydride $(0.049 \mathrm{~g}, 1.3 \mathrm{mmol})$ at $0^{\circ} \mathrm{C}$. The ice bath was removed, and the colorless reaction mixture was stirred at rt for 1.25 h . The reaction was quenched with $1 \mathrm{M} \mathrm{HCl}(5 \mathrm{~mL})$, then was partitioned between water $(40 \mathrm{~mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(40 \mathrm{~mL})$, and the aqueous phase was re-extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(40 \mathrm{~mL})$. The combined extracts were washed with water $(30 \mathrm{~mL}), 1: 1$ water/brine (30 mL), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and were concentrated. Chromatography on $\mathrm{SiO}_{2}$ (15-30\% EtOAc/hexane) afforded alcohol 2-61 ( $0.318 \mathrm{~g}, 0.897 \mathrm{mmol}, 76 \%)$ as a colorless solid: Mp 142$144{ }^{\circ} \mathrm{C}$; IR (ATR) 3360 (br), 2934, 2384, 1450, 1127, 911, $726 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.27(\mathrm{~s}, 2 \mathrm{H}), 7.25(\mathrm{~s}, 2 \mathrm{H}), 7.13(\mathrm{~s}, 2 \mathrm{H}), 3.25(\mathrm{~d}, 2 \mathrm{H}, J=6.4 \mathrm{~Hz}), 2.75($ pent, $1 \mathrm{H}, J=7.6 \mathrm{~Hz})$, 2.42-2.24 (m, 2 H$), 2.35(\mathrm{~s}, 12 \mathrm{H}), 2.03-1.92(\mathrm{~m}, 2 \mathrm{H}), 1.58\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=14.4 \mathrm{~Hz}\right), 1.45(\mathrm{br} \mathrm{s}, 1$ H), 1.3-0.5 (br q, 3 H ); ${ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 138.2$, 138.1, 132.89, 132.87, 131.13, $131.05,127.8,127.3,66.2,33.3,32.2,\left(J_{\mathrm{CP}}=12 \mathrm{~Hz}\right), 31.0\left({ }^{1} J_{\mathrm{CP}}=30 \mathrm{~Hz}\right), 23.6\left({ }^{3} J_{\mathrm{CP}}=3 \mathrm{~Hz}\right), 21.4$; ${ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 28.4; HRMS $\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{22} \mathrm{H}_{31} \mathrm{BOP}[\mathrm{M}-\mathrm{H}] 353.2200$,
found 353.2188 .


3-(Ditolyl-3-methylcyclobutyl)methanol (2-62). Prepared according to General Procedure C from phosphine borane $\mathbf{2 - 6 0}(0.160 \mathrm{~g}, 0.512 \mathrm{mmol})$ and $\mathrm{DABCO}(0.230 \mathrm{~g}, 2.05 \mathrm{mmol})$ in toluene ( 2 mL ) to afford phosphine 2-62 ( $0.119 \mathrm{~g}, 0.381 \mathrm{mmol}, 78 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.28(\mathrm{~d}, 4 \mathrm{H}, J=7.2 \mathrm{~Hz}), 7.15(\mathrm{~d}, 4 \mathrm{H}, J=9.6 \mathrm{~Hz}), 3.21(\mathrm{~d}, 2 \mathrm{H}, J=6.4$ $\mathrm{Hz}), 2.68$ (pent, $1 \mathrm{H}, J=8.0 \mathrm{~Hz}), 2.35(\mathrm{~s}, 6 \mathrm{H}), 2.07(\mathrm{~d}, 1 \mathrm{H}, J=12.4 \mathrm{~Hz}), 2.02(\mathrm{~d}, 1 \mathrm{H}, J=11.2$ $\mathrm{Hz}), 2.00-1.91(\mathrm{~m}, 2 \mathrm{H}), 1.48\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} \mathrm{~J}_{\mathrm{PH}}=12.8 \mathrm{~Hz}\right) ;{ }^{31} \mathrm{P} \operatorname{NMR}\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.7$.

(11S)-4-((3-(Di-p-tolylphosphanyl)-3-methylcyclobutyl)methoxy)dinaphtho[2,1-d:1',2'-
$f][\mathbf{1 , 3 , 2}]$ dioxaphosphepine (2-63). ${ }^{97} S$-BINOL $(0.223 \mathrm{~g}, 0.778 \mathrm{mmol})$ was placed under $\mathrm{N}_{2}$ into a $50-\mathrm{mL}$ flask and treated with $\mathrm{PCl}_{3}(1.1 \mathrm{~mL}, 12 \mathrm{mmol}, 17 \mathrm{eq})$ and $N$-methylpyrrolidone ( 1 drop). The mixture was stirred at $60^{\circ} \mathrm{C}$ for 15 min , cooled, and slowly concentrated under high vacuum to give a semi-solid. The residue was azeotroped with toluene ( 5 mL ) and dissolved in THF (5 $m L)$. An NMR of an aliquot showed the clean chlorophosphite ( ${ }^{31} \mathrm{P}$ NMR $\delta 174$ ). In a separate $50-$ mL flask, a solution of phosphine 2-62 ( $0.221 \mathrm{~g}, 0.707 \mathrm{mmol})$ in THF $(3 \mathrm{~mL})$ was cooled to $0^{\circ} \mathrm{C}$ under $\mathrm{N}_{2} . \mathrm{NaH}(0.031 \mathrm{~g}, 0.79 \mathrm{mmol}, 1.1 \mathrm{eq}, 60 \% \mathrm{in}$ oil $)$ was added in one portion. The resulting alkoxide was stirred for 10 min at $0^{\circ} \mathrm{C}$. The chlorophosphine solution was then transferred to the
alkoxide solution via syringe at $0^{\circ} \mathrm{C}$. The cooling bath was removed, and the reaction mixture was stirred for 19 h at rt , concentrated, and purified by chromatography on $\mathrm{SiO}_{2}$ (5-40\% EtOAc/hexane) to afford 2-63 ( $0.121 \mathrm{~g}, 0.193 \mathrm{mmol}, 27 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.96(\mathrm{~d}, 1 \mathrm{H}, J=8.8 \mathrm{~Hz}), 7.95-7.86(\mathrm{~m}, 3 \mathrm{H}), 7.47(\mathrm{~d}, 1 \mathrm{H}, J=8.8 \mathrm{~Hz}), 7.42(\mathrm{td}, 2 \mathrm{H}, J$ $=6.8 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 7.37-7.22(\mathrm{~m}, 9 \mathrm{H}), 7.18-7.12(\mathrm{~m}, 4 \mathrm{H}), 3.55\left(\operatorname{app} \mathrm{dd}, 1 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=17.2 \mathrm{~Hz}, J=\right.$ $7.6 \mathrm{~Hz}), 3.37\left(\mathrm{app} \mathrm{dd}, 1 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=17.6 \mathrm{~Hz}, J=7.6 \mathrm{~Hz}\right), 2.72($ pent, $1 \mathrm{H}, J=8.0 \mathrm{~Hz}), 2.34(\mathrm{~s}, 6 \mathrm{H})$, $2.05(\mathrm{dd}, 2 \mathrm{H}, J=20 \mathrm{~Hz}, 12 \mathrm{~Hz}), 1.96-1.88(\mathrm{~m}, 2 \mathrm{H}), 1.45\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{PH}}=12.8 \mathrm{~Hz}\right) ;{ }^{13} \mathrm{C}$ NMR ( 100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 148.5\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 147.5,138.5\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 134.3\left(J_{\mathrm{CP}}=6 \mathrm{~Hz}\right), 134.1\left(J_{\mathrm{CP}}=6\right.$ $\mathrm{Hz}), 133.1,133.0,132.8,132.6,131.5,131.0,130.3,129.9,129.1\left(J_{\mathrm{CP}}=7 \mathrm{~Hz}\right), 128.3\left(J_{\mathrm{CP}}=6 \mathrm{~Hz}\right)$, $127.0,126.2\left(J_{\mathrm{CP}}=9 \mathrm{~Hz}\right), 125.0,124.9,124.1\left(J_{\mathrm{CP}}=5 \mathrm{~Hz}\right), 122.7\left(J_{\mathrm{CP}}=2 \mathrm{~Hz}\right), 121.9,121.7,68.5$ $\left({ }^{2} J_{\mathrm{CP}}=7 \mathrm{~Hz}\right), 34.73,34.69,34.63,34.55,31.8\left({ }^{1} J_{\mathrm{CP}}=18 \mathrm{~Hz}\right), 31.2,\left(J_{\mathrm{CP}}=10 \mathrm{~Hz}\right), 26.9,25.4,25.3$ $\left(J_{\mathrm{CP}}=4 \mathrm{~Hz}\right), 21.4,20.7,14.2 ;{ }^{31} \mathrm{P}$ NMR $\left(162 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.8,7.2 ; \mathrm{HRMS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{40} \mathrm{H}_{37} \mathrm{O}_{3} \mathrm{P}_{2}[\mathrm{M}+\mathrm{H}]$ 627.2212, found 627.2206.


## 2-65

Methyl acetyl-D-phenylalaninate (2-65). ${ }^{97}$ A solution of 2-64 (0.116 g, 0.532 mmol$)$ in MeOH $(1 \mathrm{~mL})$ in a stainless steel pressure vessel was charged with a solution of phosphine-phosphite 2$63(0.010 \mathrm{~g}, 0.016 \mathrm{mmol}, 3 \mathrm{~mol} \%)$ in $\mathrm{MeOH}(0.3 \mathrm{~mL})$ and a solution of $\mathrm{Rh}(\mathrm{NBD}) \mathrm{BF}_{4}(0.002 \mathrm{~g}$, $0.005 \mathrm{mmol}, 1 \mathrm{~mol} \%)$ in $\mathrm{MeOH}(0.3 \mathrm{~mL})$ under an Ar blanket provided by an inverted funnel. The resulting yellow solution was sealed in the Parr autoclave and placed under 13.99 bar of $\mathrm{H}_{2}$. After 3.5 h , the remaining 13.96 bar was vented. The solution was concentrated, re-dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(1 \mathrm{~mL})$, and filtered through neutral $\mathrm{Al}_{2} \mathrm{O}_{3}$ to afford $\mathbf{2 - 6 5}(0.086 \mathrm{~g}, 0.39 \mathrm{mmol}, 73 \%)$ as a colorless solid: $[\alpha]_{\mathrm{D}}-0.855\left(\mathrm{c}=1.0, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.38-7.21(\mathrm{~m} 3 \mathrm{H}), 7.16-7.07$
(m, 2 H), $6.09(\mathrm{~d}, 1 \mathrm{H}, J=6.6 \mathrm{~Hz}), 4.90(\mathrm{ddd}, 1 \mathrm{H}, J=7.8 \mathrm{~Hz}, 6.0 \mathrm{~Hz}, 5.7 \mathrm{~Hz}), 3.74(\mathrm{~s}, 3 \mathrm{H}), 3.21-$ 3.04 (m, 2 H), 1.99 (s, 3 H); Chiral SFC (Chiralpak IA, 5-15\% i-PrOH modifier gradient) $\mathrm{T}_{\mathrm{r}} 8.48$ $\min$ (major, $D$-isomer), $11.42 \min$ (minor, $L$-isomer).


## ((3-(Bis(3,5-dimethylphenyl)phosphanyl)-3-methylcyclobutyl)methyl)bis(3,5-

dimethylphenyl)phosphane diborane (2-66). A solution of alcohol 2-61 (0.119 g, 0.336 mmol ) and $\mathrm{Et}_{3} \mathrm{~N}(0.05 \mathrm{~mL}, 0.03 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$ was cooled to $0{ }^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$. Brosyl chloride $(0.086 \mathrm{~g}, 0.34 \mathrm{mmol})$ was added in one portion. The solution was warmed to rt over 2.5 h , then was directly placed on an $\mathrm{SiO}_{2}$ column and purified by chromatography ( $10 \% \mathrm{EtOAc} /$ hexane) to afford the corresponding $p$-bromosulfonate $(0.068 \mathrm{~g}, 0.119 \mathrm{mmol}, 35 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 7.76-7.66(\mathrm{~m}, 4 \mathrm{H}), 7.20(\mathrm{~s}, 2 \mathrm{H}), 7.16(\mathrm{~s}, 2 \mathrm{H}), 7.11(\mathrm{~s}, 2 \mathrm{H}), 3.64(\mathrm{~d}$, $2 \mathrm{H}, J=7.2 \mathrm{~Hz}), 2.85(\mathrm{pent}, 1 \mathrm{H}, J=8.1 \mathrm{~Hz}), 2.32(\mathrm{~s}, 12 \mathrm{H}), 2.35-2.22(\mathrm{~m}, 2 \mathrm{H}), 2.02-1.91(\mathrm{~m}, 2$ H), $1.52\left(\mathrm{~d}, \quad 3 \mathrm{H}, \quad{ }^{3} J_{\mathrm{PH}}=14.1 \mathrm{~Hz}\right)$, ${ }_{1.3-0.1}(\mathrm{br} \quad \mathrm{q}, \quad 3 \quad \mathrm{H})$.

A solution of this brosylate $(0.068 \mathrm{~g}, 0.119 \mathrm{mmol})$ and di- $m$-xylyl phosphine borane $(0.046 \mathrm{~g}, 0.178 \mathrm{mmol})$ in DMAc $(1 \mathrm{~mL})$ under $\mathrm{N}_{2}$ was treated with sodium hydride $(0.007 \mathrm{~g}, 0.2$ $\mathrm{mmol}, 60 \%$ in oil) in one portion. The reaction was stirred for 19 h at rt , then was quenched with sat aq $\mathrm{NH}_{4} \mathrm{Cl}(10 \mathrm{~mL})$. The material was partitioned between ether and water ( 30 mL each), the layers were separated, and the aqueous layer was extracted with ether ( 30 mL ). The combined extracts were washed with water ( 30 mL ), brine $(30 \mathrm{~mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Two iterations of chromatography on $\mathrm{SiO}_{2}(10 \% \mathrm{EtOAc} /$ hexane $)$
afforded diphosphine borane 2-66 ( $0.034 \mathrm{~g}, 0.057 \mathrm{mmol}, 48 \%$ ) as a colorless solid: $\mathrm{Mp} 225-227$ ${ }^{\circ} \mathrm{C} ; \operatorname{IR}(\mathrm{ATR}) 2933,2378,1737,1461,1128,1066,915,734 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $7.18(\mathrm{~s}, 2 \mathrm{H}), 7.16(\mathrm{~s}, 4 \mathrm{H}), 7.14(\mathrm{~s}, 2 \mathrm{H}), 7.09(\mathrm{~s}, 4 \mathrm{H}), 2.85-2.74(\mathrm{~m}, 1 \mathrm{H}), 2.35-2.26(\mathrm{~m}, 2 \mathrm{H})$, 2.33 (s, 12 H ), 2.31 (s, 12 H ), 1.97 (dd, $2 \mathrm{H}, J=11.0 \mathrm{~Hz}, 7.0 \mathrm{~Hz}$ ), 1.94-1.87 (m, 2 H ), 1.42 (d, 3 $\mathrm{H},{ }^{3} \mathrm{~J}_{\mathrm{PH}}=14.0 \mathrm{~Hz}$ ), 1.1-0.4 (br q, 6 H ); ${ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 138.3, 138.2, 138.1, $138.0,132.9,132.8,131.1,131.0,129.7,129.6,129.1,127.7,127.2,38.0\left(J_{\mathrm{CP}}=9 \mathrm{~Hz}\right), 31.5\left(J_{\mathrm{CP}}\right.$ $=5 \mathrm{~Hz}), 31.2\left({ }^{1} J_{\mathrm{CP}}=31 \mathrm{~Hz}\right), 25.4\left(J_{\mathrm{CP}}=14 \mathrm{~Hz}\right), 23.3,21.4,21.3 ;{ }^{31} \mathrm{P} \operatorname{NMR}\left(202 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 29.5, 11.8. An X-ray crystal structure was obtained for this compound.

trans 2-67

cis 2-67

## General Procedure $F$ (Thiol additions to bicyclo[1.1.0]butyl nitriles): 3-Methyl-3-

 (phenylthio)cyclobutane-1-carbonitrile (2-67). A 50 mL flask under $\mathrm{N}_{2}$ was charged with a solution of thiophenol $(0.19 \mathrm{~mL}, 1.8 \mathrm{mmol})$ in $\mathrm{MeOH}(10 \mathrm{~mL})$. This solution was treated with anhydrous $\mathrm{K}_{2} \mathrm{CO}_{3}(0.251 \mathrm{~g}, 1.82 \mathrm{mmol}, 1.00 \mathrm{eq})$ and stirred 10 min at rt. Nitrile 2-37(0.203 g, $2.18 \mathrm{mmol}, 1.20 \mathrm{eq})$ was then added neat by syringe over 1 min . The solution was stirred at rt for 21 h , at which point it was concentrated. The crude, cloudy liquid was purified by chromatography on $\mathrm{SiO}_{2}(10-20 \% \mathrm{EtOAc} /$ hexanes $)$ to afford the less polar trans-2-67 ( $0.260 \mathrm{~g}, 1.27 \mathrm{mmol}, 70 \%$ ) and the more polar cis-2-67 ( $0.042 \mathrm{~g}, 0.21 \mathrm{mmol}, 11 \%$ ), both as colorless oils: trans-2-67: ${ }^{1} \mathrm{HNMR}$ ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.50-7.44(\mathrm{~m}, 2 \mathrm{H}), 7.39-7.32(\mathrm{~m}, 3 \mathrm{H}), 3.23$ (pent, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}$ ), 2.60 (ddd, $2 \mathrm{H}, J=10.8 \mathrm{~Hz}, 8.4 \mathrm{~Hz}, 2.4 \mathrm{~Hz}$ ), 2.41 (ddd, $2 \mathrm{H}, J=10.4 \mathrm{~Hz}, 8.4 \mathrm{~Hz}, 2.4 \mathrm{~Hz}$ ), 1.55 (s, 3 $\mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 135.5,132.3,129.0,128.9,122.2,48.2,39.6,29.7$, 16.7. cis-2-67: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.49-7.45(\mathrm{~m}, 2 \mathrm{H}), 7.40-7.34(\mathrm{~m}, 3 \mathrm{H}), 3.05$ (pent, $1 \mathrm{H}, J=$8.5 Hz ), 2.64 (app dt, $2 \mathrm{H}, J=9.5 \mathrm{~Hz}, 2.5 \mathrm{~Hz}$ ), 2.34 (ddd, $2 \mathrm{H}, J=10.5 \mathrm{~Hz}, 8.0 \mathrm{~Hz}, 2.5 \mathrm{~Hz}$ ), 1.49 (s, 3 H ) ${ }^{13}{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 135.7,131.9,129.1,129.0,121.4,46.3,39.8,39.6,26.9$, 16.5; IR (ATR) 2956, 2235, 1720, 1474, 917, 745, $695 \mathrm{~cm}^{-1} ; \mathrm{MS}^{\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 204,159,118 ; \text { HRMS }}$ $\left(\mathrm{ESI}^{+}\right) m / z$ calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{NS}[\mathrm{M}+\mathrm{H}]$ 204.0842, found 204.0842.

trans 2-68

cis 2-68

3-Methyl-3-(naphthalen-2-ylthio)cyclobutane-1-carbonitrile (2-68). Prepared according to General Procedure F from 2-naphthlenethiol $(0.122 \mathrm{~g}, 0.761 \mathrm{mmol}), \mathrm{K}_{2} \mathrm{CO}_{3}(0.105 \mathrm{~g}, 0.761 \mathrm{mmol}$, 1.00 eq ), and nitrile $\mathbf{2 - 3 7}$ ( $0.203 \mathrm{~g}, 2.18 \mathrm{mmol}, 1.20 \mathrm{eq}$ ) in $\mathrm{MeOH}(5 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ (10-20\% EtOAc/hexanes) afforded trans-2-68 ( $0.119 \mathrm{~g}, 0.470 \mathrm{mmol}$, 62\%) as a colorless solid and cis-2-68 ( $0.032 \mathrm{~g}, 0.13 \mathrm{mmol}, 17 \%$ ) as a colorless oil: trans-2-68: Mp 61-64 ${ }^{\circ} \mathrm{C}$; IR (ATR) 2970, 2236, 1583, 1375, 1196, 1114, 863, 827, $749 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.99(\mathrm{~s}, 1 \mathrm{H}), 7.88-7.77(\mathrm{~m}, 3 \mathrm{H}), 7.57-7.48(\mathrm{~m}, 3 \mathrm{H}), 3.27($ pent, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}$ ), 2.67 (ddd, $2 \mathrm{H}, J=11.2 \mathrm{~Hz}, 8.8 \mathrm{~Hz}, 2.4 \mathrm{~Hz}$ ), 2.43 (ddd, $2 \mathrm{H}, J=10.8 \mathrm{~Hz}, 8.0 \mathrm{~Hz}, 2.4 \mathrm{~Hz}$ ), 1.60 (s, $3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 135.0,133.5,133.0,132.1,129.8,128.6,127.82,127.76$, $127.0,126.8,122.2,48.6,39.7,29.8,16.8 . \underline{c i s-2-68}:{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.99(\mathrm{~d}, 1 \mathrm{H}$, $J=1.2 \mathrm{~Hz}), 7.88-7.75(\mathrm{~m}, 3 \mathrm{H}), 7.56-7.46(\mathrm{~m}, 3 \mathrm{H}), 3.08(\mathrm{pent}, 1 \mathrm{H}, J=8.8 \mathrm{~Hz}), 2.73(\mathrm{dd}, 2 \mathrm{H}, J$ $=8.8 \mathrm{~Hz}, 2.4 \mathrm{~Hz}), 2.43(\mathrm{ddd}, 2 \mathrm{H}, J=10.4 \mathrm{~Hz}, 8.0 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 1.52(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 135.2,133.6,133.1,132.1,129.3,128.7,127.82,127.76,126.9,126.7,121.5,46.6,40.0$, 27.0, 16.7; HRMS (ASAP) $m / z$ calcd for $\mathrm{C}_{16} \mathrm{H}_{16} \mathrm{NS}[\mathrm{M}+\mathrm{H}]$ 254.1003, found 254.1035.

trans 2-69

cis 2-69

3-Methyl-3-(o-tolylthio)cyclobutane-1-carbonitrile (2-69). Prepared according to General Procedure F from nitrile 2-37 ( $0.050 \mathrm{~g}, 0.54 \mathrm{mmol}$ ), o-thiocresol ( $0.06 \mathrm{~mL}, 0.49 \mathrm{mmol}$ ), and $\mathrm{K}_{2} \mathrm{CO}_{3}$ $(0.074 \mathrm{~g}, 0.54 \mathrm{mmol})$ in $\mathrm{MeOH}(3 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(10-20 \%$ EtOAc/hexanes) to afford the less polar trans-2-69 ( $0.084 \mathrm{~g}, 0.39 \mathrm{mmol}, 79 \%)$ and the more polar cis-2-69 ( $0.008 \mathrm{~g}, 0.04 \mathrm{mmol}, 8 \%$ ), both as colorless oils. trans-2-69: IR (ATR): 2940, 2238, 1732, 1589, 1487, 1263, 1114, 1063, $748 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.36(\mathrm{dd}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}$, $0.8 \mathrm{~Hz}), 7.29-7.20(\mathrm{~m}, 2 \mathrm{H}), 7.17(\mathrm{td}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 3.25($ pent, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}), 2.63$ (ddd, $2 \mathrm{H}, J=11.2 \mathrm{~Hz}, 9.2 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 2.48-2.39(\mathrm{~m}, 2 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}), 1.60(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 141.7,134.7,132.2,130.8,128.5,126.4,122.3,48.7,39.8,29.5,21.4,17.0$; HRMS (ESI) ${ }^{+}$calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{NS}[\mathrm{M}+\mathrm{H}]$ 218.1003, found 218.0991. cis-2-69: ${ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.39(\mathrm{~d}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 7.32-7.23(\mathrm{~m}, 2 \mathrm{H}), 7.19(\mathrm{td}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}, 2.4 \mathrm{~Hz})$, 3.05 (pent, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}$ ), $2.63(\mathrm{td}, 2 \mathrm{H}, J=9.6 \mathrm{~Hz}, 2.4 \mathrm{~Hz}), 2.48-2.35(\mathrm{~m}, 2 \mathrm{H}), 2.44(\mathrm{~s}, 3 \mathrm{H})$, 1.49 (s, 3 H ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 142.3,136.1,131.6,130.8,129.0,126.5,45.8,40.1$, $26.5,21.5,16.8$.

trans-2-70

cis-2-70

3-((2-Bromophenyl)thio)-3-methylcyclobutane-1-carbonitrile (2-70). Prepared according to General Procedure F from bicyclobutane 2-37 ( $0.210 \mathrm{~g}, 2.25 \mathrm{mmol}, 1.20 \mathrm{eq}$ ), 2-bromothiophenol ( $0.22 \mathrm{~mL}, 1.87 \mathrm{mmol}$ ) and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.286 \mathrm{~g}, 2.07 \mathrm{mmol}, 1.10 \mathrm{eq})$ in $\mathrm{MeOH}(5 \mathrm{~mL})$. Purification by
chromatography on $\mathrm{SiO}_{2}(20-30 \% \mathrm{EtOAc} /$ hexanes $)$ afforded trans-2-70 $(0.367 \mathrm{~g}, 1.30 \mathrm{mmol}$, 69\%) and cis-2-70 ( $0.067 \mathrm{~g}, 0.24 \mathrm{mmol}, 13 \%$ ), both as colorless oils. trans-2-70 : IR (ATR) 2940, $2238,1731,1445,1426,1103,1019,712 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.65(\mathrm{dd}, 1 \mathrm{H}, J=$ $8.0 \mathrm{~Hz}, 1.6 \mathrm{~Hz}), 7.39(\mathrm{dd}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}, 1.6 \mathrm{~Hz}), 7.29(\mathrm{td}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 7.16(\mathrm{td}, 1$ $\mathrm{H}, J=7.6 \mathrm{~Hz}, 1.6 \mathrm{~Hz}$ ), 3.37 (pent, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}$ ), 2.70 (ddd, $2 \mathrm{H}, J=10.8 \mathrm{~Hz}, 9.2 \mathrm{~Hz}, 2.4 \mathrm{~Hz}$ ), 2.48 (ddd, $2 \mathrm{H}, J=10.8 \mathrm{~Hz}, 8.4 \mathrm{~Hz}, 2.4 \mathrm{~Hz}), 1.66(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 134.5, $134.4,133.8,129.3,128.7,122.0,49.1,39.8,29.4,17.2 ; \mathrm{MS}_{(\mathrm{ESI})^{+} \mathrm{m} / \mathrm{z} 281,261,233,212,199,}$ 144, 127, 114; HRMS (ESI) ${ }^{+}$calcd for $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{NSBr}[\mathrm{M}+\mathrm{H}]$ 281.9947, found 281.9938. cis-2-70 : ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.68(\mathrm{dd}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}, 1.6 \mathrm{~Hz}), 7.47(\mathrm{dd}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}, 1.6$ Hz ), $7.31(\mathrm{td}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 7.20(\mathrm{td}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}, 1.6 \mathrm{~Hz}), 3.08$ (pent, $1 \mathrm{H}, J=8.8$ $\mathrm{Hz}), 2.72(\mathrm{td}, 2 \mathrm{H}, J=9.6 \mathrm{~Hz}, 2.4 \mathrm{~Hz}), 2.54-2.43(\mathrm{~m}, 2 \mathrm{H}), 1.53(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 136.3,133.9,133.8,130.0,127.9,121.3,47.1,40.3,26.3,17.0$.

trans-2-71

cis-2-71

3-((3-Fluorophenyl)thio)-3-methylcyclobutane-1-carbonitrile (2-71). Prepared according to General Procedure F from 3-fluorothiophenol ( $0.06 \mathrm{~mL}, 0.49 \mathrm{mmol}$ ), nitrile 2-37 ( $0.067 \mathrm{~g}, 0.72$ mmol, 1.1 eq ) and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.099 \mathrm{~g}, 0.72 \mathrm{mmol}$, 1.1 eq ) in $\mathrm{MeOH}(3 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(10-20 \% \mathrm{EtOAc} /$ hexanes $)$ afforded the less polar trans-2-71 ( 0.074 g , $0.33 \mathrm{mmol}, 51 \%)$ and the more polar cis-2-71 ( $0.023 \mathrm{~g}, 0.10 \mathrm{mmol}, 16 \%$ ), both as colorless oils. trans-2-71: IR (ATR) 2942, 2262, 1597, 1578, 1472, 1261, 1215, 1112, $863 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.31(\mathrm{td}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}, 6.0 \mathrm{~Hz}), 7.20(\mathrm{~d}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.14(\mathrm{dt}, 1 \mathrm{H}, J=8.0$ $\mathrm{Hz}, 2.0 \mathrm{~Hz}$ ), 7.05 (td, $1 \mathrm{H}, J=8.0 \mathrm{~Hz}, 2.0 \mathrm{~Hz}$ ), 3.28 (pent, $1 \mathrm{H}, J=8.4 \mathrm{~Hz}$ ), 2.60 (ddd, $2 \mathrm{H}, J=$ $10.8 \mathrm{~Hz}, 9.2 \mathrm{~Hz}, 2.0 \mathrm{~Hz}$ ), 2.44 (ddd, $2 \mathrm{H}, J=10.8 \mathrm{~Hz}, 8.4 \mathrm{~Hz}, 2.4 \mathrm{~Hz}$ ), $1.60(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR
$\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 162.5\left({ }^{1} J_{\mathrm{CF}}=248 \mathrm{~Hz}\right), 134.6\left({ }^{3} J_{\mathrm{CF}}=7 \mathrm{~Hz}\right), 130.4\left({ }^{4} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right), 130.3\left({ }^{3} J_{\mathrm{CF}}\right.$ $=9 \mathrm{~Hz}), 122.0,121.3\left({ }^{2} J_{\mathrm{CF}}=21 \mathrm{~Hz}\right), 115.8\left({ }^{2} J_{\mathrm{CF}}=21 \mathrm{~Hz}\right), 48.4,39.6,29.6,16.8 ;{ }^{19} \mathrm{~F}$ NMR $(376$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-111.7$; HRMS (ESI) ${ }^{+}$calcd for $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{NSF}[\mathrm{M}+\mathrm{H}]$ 222.0723, found 222.0753. cis-2-71: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.34(\mathrm{td}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}, 6.0 \mathrm{~Hz}), 7.22(\mathrm{dt}, 1 \mathrm{H}, J=7.6$ $\mathrm{Hz}, 1.2 \mathrm{~Hz}$ ), $7.15(\mathrm{dt}, 1 \mathrm{H}, J=8.8 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 7.08(\mathrm{tdd}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}, 2.8 \mathrm{~Hz}, 0.8 \mathrm{~Hz}), 3.09$ (pent, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}$ ), 2.67 (ddd, $2 \mathrm{H}, J=10.4 \mathrm{~Hz}, 8.8 \mathrm{~Hz}, 2.0 \mathrm{~Hz}$ ), 2.43 (ddd, $2 \mathrm{H}, J=10.8 \mathrm{~Hz}$, $8.8 \mathrm{~Hz}, 2.0 \mathrm{~Hz}), 1.50(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 162.5\left({ }^{1} J_{\mathrm{CF}}=248 \mathrm{~Hz}\right), 134.2\left({ }^{3} J_{\mathrm{CF}}\right.$ $=8 \mathrm{~Hz}), 130.8\left({ }^{4} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right), 130.4\left({ }^{3} J_{\mathrm{CF}}=8 \mathrm{~Hz}\right), 121.8\left({ }^{2} J_{\mathrm{CF}}=22 \mathrm{~Hz}\right), 121.3,116.0\left({ }^{2} J_{\mathrm{CF}}=21\right.$ $\mathrm{Hz}), 46.4,40.0,26.9,16.7 ;{ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$-111.7.

trans 2-72

cis 2-72

3-Methyl-3-(pyridin-2-ylthio)cyclobutane-1-carbonitrile (2-72). Prepared according to General Procedure F from 2-mercaptopyridine ( $0.093 \mathrm{~g}, 0.84 \mathrm{mmol}$ ), nitrile 2-37 ( $0.086 \mathrm{~g}, 0.92 \mathrm{mmol}$ ) and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.116 \mathrm{~g}, 0.839 \mathrm{mmol})$ in $\mathrm{MeOH}(4 \mathrm{~mL})$ with heating at $70{ }^{\circ} \mathrm{C}$. Purification by chromatography on $\mathrm{SiO}_{2}$ ( $25 \% \mathrm{EtOAc} /$ hexanes ) afforded trans-2-72 ( $0.088 \mathrm{~g}, 0.43 \mathrm{mmol}, 51 \%$ ) as a colorless oil that solidified upon standing and cis-2-72 ( $0.024 \mathrm{~g}, 0.12 \mathrm{mmol}, 14 \%)$ as a colorless oil. trans-2-72: Mp 55-57 ${ }^{\circ} \mathrm{C}$; IR (ATR) 2957, 2240, 1578, 1564, 1457, 1416, 1111, $760,723 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.41(\mathrm{dt}, 1 \mathrm{H}, J=4.0 \mathrm{~Hz}, 0.8 \mathrm{~Hz}$ ), $7.48(\mathrm{ddt}, 1 \mathrm{H}, J$ $=7.6 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 0.8 \mathrm{~Hz}), 7.14(\mathrm{dd}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}, 0.8 \mathrm{~Hz}), 7.04-6.98(\mathrm{~m}, 1 \mathrm{H}), 3.38$ (pent, 1 H , $J=8.4 \mathrm{~Hz}), 2.91-2.81(\mathrm{~m}, 2 \mathrm{H}), 2.60-2.51(\mathrm{~m}, 2 \mathrm{H}), 1.79(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $158.2,149.6,136.1,123.9,122.4,120.1,47.9,40.0,29.8,17.4 ; \operatorname{MS}(E S I)^{+} \mathrm{m} / \mathrm{z} 205,171,144,112$ (5); HRMS (ESI) ${ }^{+}$calcd for $\mathrm{C}_{11} \mathrm{H}_{13} \mathrm{~N}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]$ 205.0799, found 205.0781. cis-2-72: ${ }^{1} \mathrm{H}$ NMR (400
$\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.48(\mathrm{~d}, 1 \mathrm{H}, J=4.0 \mathrm{~Hz}), 7.51(\mathrm{td}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 7.16(\mathrm{~d}, 1 \mathrm{H}, J=8.0$ $\mathrm{Hz}), 7.03(\mathrm{t}, 1 \mathrm{H}, J=6.0 \mathrm{~Hz}), 3.19(\mathrm{pent}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}), 3.13-3.04(\mathrm{~m}, 2 \mathrm{H}), 2.58(\mathrm{ddd}, 2 \mathrm{H}, J=$ $10.4 \mathrm{~Hz}, 8.8 \mathrm{~Hz}, 2.0 \mathrm{~Hz}$ ), $1.69(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 158.2,149.6,136.2,123.9$, 122.4, 122.0, 45.8, 40.0, 27.7, 17.3.

trans-2-73

cis-2-73

3-Methyl-3-((5-methyl-1,3,4-thiadiazol-2-yl)thio)cyclobutane-1-carbonitrile (2-73). Prepared according to General Procedure F from 5-methylthiadiazole-2-thiol ( $0.084 \mathrm{~g}, 0.63 \mathrm{mmol}$ ), nitrile $\mathbf{2 - 3 7}(0.065 \mathrm{~g}, 0.70 \mathrm{mmol}, 1.1 \mathrm{eq})$, and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.096 \mathrm{~g}, 0.70 \mathrm{mmol}, 1.1 \mathrm{eq})$ in $\mathrm{MeOH}(3 \mathrm{~mL})$ with heating at $60^{\circ} \mathrm{C}$. Purification by chromatography on $\mathrm{SiO}_{2}$ (10-20\% EtOAc/hexanes) afforded the less polar trans-2-73 ( $0.075 \mathrm{~g}, 0.33 \mathrm{mmol}, 52 \%$ ) and the more polar cis-2-37 $(0.015 \mathrm{~g}, 0.07 \mathrm{mmol}$, 10\%), both as colorless oils. trans-2-73: IR (ATR) 2963, 2239, 1426, 1383, 1190, 1039, 911, 726 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.37$ (pent, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}$ ), 2.87 (ddd, $2 \mathrm{H}, J=12.0 \mathrm{~Hz}, 9.2$ $\mathrm{Hz}, 2.4 \mathrm{~Hz}), 2.72(\mathrm{~s}, 3 \mathrm{H}), 2.56(\mathrm{ddd}, 2 \mathrm{H}, J=11.6 \mathrm{~Hz}, 8.4 \mathrm{~Hz}, 2.8 \mathrm{~Hz}), 1.77(\mathrm{~s}, 3 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR (125 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 166.2,161.8,121.6,50.7,39.7,29.3,16.9,15.5 ; \mathrm{HRMS}^{(\mathrm{ESI})^{+}}$calcd for $\mathrm{C}_{9} \mathrm{H}_{12} \mathrm{~N}_{3} \mathrm{~S}_{2}[\mathrm{M}+\mathrm{H}] 226.0473$, found 226.0479. cis-2-73: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 3.19$ (pent, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}), 2.95-2.85(\mathrm{~m}, 2 \mathrm{H}), 2.75(\mathrm{~s}, 3 \mathrm{H}), 2.66(\mathrm{ddd}, 2 \mathrm{H}, J=11.2 \mathrm{~Hz}, 8.8 \mathrm{~Hz}, 2.4 \mathrm{~Hz})$, 1.75 (s, 3 H ); ${ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 166.3,161.7,121.2,48.4,40.2,27.1,17.2,15.6$.


3-Methyl-3-((1-methyl-1H-imidazol-2-yl)thio)cyclobutane-1-carbonitrile (2-74). Prepared according to General Procedure F from bicyclobutane 2-37 ( $0.070 \mathrm{~g}, 0.75 \mathrm{mmol}, 1.20 \mathrm{eq}), 1-$ methyl-2-mercaptoimidazole $(0.072 \mathrm{~g}, 0.063 \mathrm{mmol})$ and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.095 \mathrm{~g}, 0.69 \mathrm{mmol}, 1.10 \mathrm{eq})$ in $\mathrm{MeOH}(3 \mathrm{~mL})$ with heating at $60^{\circ} \mathrm{C}$. Purification by chromatography on $\mathrm{SiO}_{2}\left(5 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ afforded 2-74 ( $0.097 \mathrm{~g}, 0.467 \mathrm{mmol}, 75 \%)$ as a colorless oil. The material is an inseparable $5: 1$ mixture of trans (major) and cis diastereomers: trans-2-74: IR (ATR) 3378, 2945, 2337, 1637, 1453, 1310, 1214, 1112, $753 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.14(\mathrm{~d}, 1 \mathrm{H}, J=1.2 \mathrm{~Hz}$ ), 7.04 (d, $1 \mathrm{H}, J=1.2 \mathrm{~Hz}), 3.71(\mathrm{~s}, 3 \mathrm{H}), 3.28$ (pent, $1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 2.82(\mathrm{ddd}, 2 \mathrm{H}, J=11.2 \mathrm{~Hz}, 9.2$ $\mathrm{Hz}, 2.8 \mathrm{~Hz}$ ), 2.44 (ddd, $2 \mathrm{H}, J=10.8 \mathrm{~Hz}, 8.0 \mathrm{~Hz}, 2.8 \mathrm{~Hz}$ ), $1.56(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\mathrm{CDCl}_{3}$ ) $\delta 138.8,130.1,123.3,122.1,50.8,39.6,34.0,29.6,16.6 ; \mathrm{HRMS}^{(\mathrm{ESI})^{+} \text {calcd for }}$ $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{~N}_{3} \mathrm{~S}[\mathrm{M}+\mathrm{H}]$ 208.0908, found 208.0902.

trans 2-75
trans-3-Methyl-3-((1-phenyl-1H-tetrazol-5-yl)thio)cyclobutane-1-carbonitrile
Prepared according to General Procedure F from bicyclobutane 2-37 ( $0.064 \mathrm{~g}, 0.69 \mathrm{mmol}, 1.1 \mathrm{eq}$ ), $N$-phenyl 2-mercaptotetrazole $(0.112 \mathrm{~g}, 0.063 \mathrm{mmol})$ and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.087 \mathrm{~g}, 0.63 \mathrm{mmol}, 1.0 \mathrm{eq})$ in $\mathrm{CF}_{3} \mathrm{CH}_{2} \mathrm{OH}(3 \mathrm{~mL})$ with heating at $60{ }^{\circ} \mathrm{C}$. After stirring at $60^{\circ} \mathrm{C}$ for 13 h , the reaction was concentrated and purified by chromatography on $\mathrm{SiO}_{2}(30 \% \mathrm{EtOAc} /$ hexanes $)$ to afford tetrazole 2-75 ( $0.032 \mathrm{~g}, 0.118 \mathrm{mmol}, 19 \%$ ) as a colorless solid: Mp 119-120 ${ }^{\circ} \mathrm{C}$; IR (ATR) $\mathrm{cm}^{-1} 3064,2962$,
$2240,1596,1499,1459,1387,1239,1015,762,695 ;{ }^{1} \mathrm{H}$ NMR (400 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 7.61-7.50(\mathrm{~m}$, 5 H ), 3.42 (pent, $1 \mathrm{H}, J=8.8 \mathrm{~Hz}$ ), 2.99 (ddd, $2 \mathrm{H}, J=12.0 \mathrm{~Hz}, 9.6 \mathrm{~Hz}, 2.4 \mathrm{~Hz}$ ), 2.68 (ddd, $2 \mathrm{H}, J$ $=11.6 \mathrm{~Hz}, 8.4 \mathrm{~Hz}, 2.8 \mathrm{~Hz}), 1.89(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 152.2,133.4$, 130.4, 129.9, 124.2, 121.6, 50.7, 39.9, 29.4, 17.2; MS (ESI) ${ }^{+} m / \mathrm{z} 272,262,179,151,83 ;$ HRMS (ESI) $^{+}$ calcd for $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{~N}_{5} \mathrm{~S}[\mathrm{M}+\mathrm{H}]$ 272.0964, found 272.0954.


## trans 2-76

3-((1H-Indol-3-yl)thio)-3-methylcyclobutane-1-carbonitrile (2-76). Prepared according to General Procedure F from indole-3-thiol ( $0.327 \mathrm{~g}, 2.20 \mathrm{mmol}$ ) nitrile 2-37 ( $0.225 \mathrm{~g}, 2.42 \mathrm{mmol}$, $1.10 \mathrm{eq})$, and $\mathrm{K}_{2} \mathrm{CO}_{3}(0.304 \mathrm{~g}, 2.19 \mathrm{mmol}, 1.00 \mathrm{eq})$ in $\mathrm{MeOH}(10 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ (10-20\% EtOAc/hexanes) afforded trans-2-76 ( $0.218 \mathrm{~g}, 0.900 \mathrm{mmol}$, $41 \%$ ) as a colorless solid in $\sim 80-90 \%$ purity. The cis isomer co-eluted with several impurities and was not characterized. trans-2-76: IR (ATR) 3321, 2956, 2245, 1492, 1452, 1416, 1266, 1110, 1009, $739 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 8.71(\mathrm{~s}, 1 \mathrm{H}), 7.82(\mathrm{~d}, 1 \mathrm{H}, J=6.6 \mathrm{~Hz}), 7.51-7.46$ $(\mathrm{m}, 1 \mathrm{H}), 7.37(\mathrm{~d}, 1 \mathrm{H}, J=2.7 \mathrm{~Hz}), 7.34-7.23(\mathrm{~m}, 2 \mathrm{H}), 3.32-3.16(\mathrm{~m}, 1 \mathrm{H}), 2.61(\mathrm{ddd}, 2 \mathrm{H}, J=$ 11.4 Hz, 9.6 Hz, 1.8 Hz), 2.35 (ddd, $2 \mathrm{H}, J=11.1 \mathrm{~Hz}, 8.1 \mathrm{~Hz}, 2.1 \mathrm{~Hz}$ ), 1.57 (s, 3 H ) ${ }^{13}{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 136.2,131.7,130.5,122.9,122.7,121.0,119.4,111.8,103.2,49.0,39.3$, 29.7, 16.6; HRMS $(\mathrm{ESI})^{+}$calcd for $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{~N}_{2} \mathrm{~S}[\mathrm{M}+\mathrm{H}]$ 243.0951, found 243.0940.

### 5.4 CHAPTER 3 EXPERIMENTAL PART



3-45

General Procedure G (Preparation of Bicyclo[1.1.0butyl alcohols/amides): (3-Methylbicyclo[1.1.0]butan-1-yl)(phenyl)methanol (3-45). ${ }^{73 \mathrm{~b}}$ A solution of 1-chloromethyl-1-methyl-2,2-dibromocyclopropane ( $10 \mathrm{~g}, 38 \mathrm{mmol}$ ) in $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{~mL})$ was cooled to $-78{ }^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$. A solution of $\mathrm{MeLi}\left(25 \mathrm{~mL}, 38 \mathrm{mmol}, \mathrm{c}=1.3 \mathrm{M} \mathrm{in}_{\mathrm{Et}}^{2} \mathrm{O}\right)$ was added via syringe over 15 min . After 1 h at $-78^{\circ} \mathrm{C}$, a solution of $t-\mathrm{BuLi}(25 \mathrm{~mL}, 38 \mathrm{mmol}, \mathrm{c}=1.5 \mathrm{M}$ in pentane) was added via syringe over 15 min . The reaction mixture was stirred for an additional 1 h at $-78^{\circ} \mathrm{C}$, then neat benzaldehyde ( $1.5 \mathrm{~mL}, 15 \mathrm{mmol}$ ) was added in one portion. The reaction mixture was quenched at $-78{ }^{\circ} \mathrm{C}$ with sat aq $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{~mL})$, warmed up to rt , and extracted ( $3 \times 50 \mathrm{~mL}$ ) with $\mathrm{Et}_{2} \mathrm{O}$. The combined organic layers were washed with water $(50 \mathrm{~mL})$, brine $(50 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO} 4\right)$, and concentrated. The crude product was purified by Kugelrohr distillation ( 1 mmHg , fractions collected at $105-115{ }^{\circ} \mathrm{C}$ oven temperature) to afford alcohol 3-45 (2.13 g, $12.2 \mathrm{mmol}, 80 \%$ ) as colorless liquid: IR (ATR) 3383, 1493, 1450, $1088 \mathrm{~cm} ;{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.42-7.26$ (m, 5 H), $5.05(\mathrm{~d}, J=4.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.96(\mathrm{~d}, 1 \mathrm{H}, J=4.2 \mathrm{~Hz}),, 1.48(\mathrm{~d}, 1 \mathrm{H}, J=6.6 \mathrm{~Hz}),, 1.46(\mathrm{~s}, 3$ H), $1.13(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 1 \mathrm{H}), 0.73(\mathrm{~s}, 1 \mathrm{H}), 0.64(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 143.1$, $128.2,127.5,126.5,73.3,32.8,32.6,16.1,11.1,10.5 ; \mathrm{MS}$ (EI) $m / z 174$ (71), 159 (66), 141 (53), 115 (76), 105 (100), 91 (77), 77 (90); HRMS (EI) calcd for $\mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}\left[\mathrm{M}^{+}\right]$174.1045, found 174.1042.

(3-Methylbicyclo[1.1.0]butan-1-yl)(4-(trifluoromethyl)phenyl)methanol (3-47). Prepared according to General Procedure G using 1-chloromethyl-1-methyl-2,2-dibromocyclopropane $(2.00 \mathrm{~g}, 7.62 \mathrm{mmol}, 2.50 \mathrm{eq}), \operatorname{MeLi}(4.7 \mathrm{~mL}, 7.6 \mathrm{mmol}, 2.5 \mathrm{eq}), t-\operatorname{BuLi}(4.5 \mathrm{~mL}, 7.6 \mathrm{mmol}, 2.5$ eq), and $p$-trifluoromethylbenzaldehyde ( $0.42 \mathrm{~mL}, 3.0 \mathrm{mmol}, 1.0 \mathrm{eq})$. Purification by chromatography on $\mathrm{SiO}_{2}$ ( $15 \% \mathrm{EtOAc} /$ hexanes ) afforded $\mathbf{3 - 4 7}(0.472 \mathrm{~g}, 1.95 \mathrm{mmol}, 64 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.60(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.50(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz})$, $5.11(\mathrm{~d}, 1 \mathrm{H}, J=4.0 \mathrm{~Hz}), 2.05(\mathrm{~d}, 1 \mathrm{H}, J=3.2 \mathrm{~Hz}), 1.47(\mathrm{~s}, 3 \mathrm{H}), 1.10(\mathrm{~d}, 1 \mathrm{H}, J=6.4 \mathrm{~Hz}), 0.71$ $(\mathrm{s}, 1 \mathrm{H}), 0.64(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 147.0,129.8\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 126.8,125.5$, $125.3,122.8,72.8,32.9,32.6,16.4,11.3,11.2$.

(4-Chlorophenyl)(3-methylbicyclo[1.1.0]butan-1-yl)methanol (3-48). Prepared according to General Procedure G using 1-chloromethyl-1- methyl-2,2-dibromocyclopropane (1.20 g, 4.57 $\mathrm{mmol}, 2.50 \mathrm{eq}), \mathrm{MeLi}(2.9 \mathrm{~mL}, 4.6 \mathrm{mmol}, 2.5 \mathrm{eq}), t-\operatorname{BuLi}(2.7 \mathrm{~mL}, 4.6 \mathrm{mmol}, 2.5 \mathrm{eq})$ and $p-$ chlorobenzaldehyde $(0.257 \mathrm{~g}, 1.83 \mathrm{mmol})$ in diethyl ether $(8 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ afforded 3-48 ( $0.215 \mathrm{~g}, 1.03 \mathrm{mmol}, 56 \%$ ) as a colorless oil: IR (ATR) 3358, 2923, 1489, 1091, 1013, $967,815 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.32(\mathrm{~s}, 4 \mathrm{H}), 5.02(\mathrm{~d}, 1 \mathrm{H}, J=4.0 \mathrm{~Hz})$, $1.97(\mathrm{~d}, 1 \mathrm{H}, J=3.6 \mathrm{~Hz}), 1.44(\mathrm{~d}, 1 \mathrm{H}, J=6.0 \mathrm{~Hz}) 1.44(\mathrm{~s}, 3 \mathrm{H}), 1.09(\mathrm{~d}, 1 \mathrm{H}, J=6.4 \mathrm{~Hz}), 0.70(\mathrm{~s}$, $1 \mathrm{H}), 0.62(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 141.6,133.3,128.5,127.9,72.9,32.8,32.6$,
16.3, 11.2, 11.0; MS (ESI ${ }^{+}$) 191 (70), 177 (100), 146 (30), 133 (90); HRMS (ESI ${ }^{+}$) calcd for $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{Cl}\left[\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}\right]$ 191.0628, found 191.0622.

(4-Methoxyphenyl)(3-methylbicyclo[1.1.0]butan-1-yl)methanol (3-49). Prepared according to General Procedure G using 1-chloromethyl-1-methyl-2,2-dibromocyclopropane ( $2.00 \mathrm{~g}, 7.62$ $\mathrm{mmol}, 2.50 \mathrm{eq}), \mathrm{MeLi}(4.8 \mathrm{~mL}, 7.6 \mathrm{mmol}, 2.5 \mathrm{eq}), t-\operatorname{BuLi}(4.5 \mathrm{~mL}, 7.6 \mathrm{mmol}, 2.5 \mathrm{eq})$, and $p$ anisaldehyde ( $0.37 \mathrm{~mL}, 3.0 \mathrm{mmol}, 1.0 \mathrm{eq}$ ). Purification by chromatography on $\mathrm{SiO}_{2}$ afforded 3-49 ( $0.284 \mathrm{~g}, 1.30 \mathrm{mmol}, 46 \%$ ) as a colorless oil: IR (ATR) 3393, 2921, 1612, 1511, 1246, 1172, 1034, $832 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.31(\mathrm{~d}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}), 6.88(\mathrm{~d}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}), 4.38$ $(\mathrm{d}, 1 \mathrm{H}, J=3.9 \mathrm{~Hz}), 3.81(\mathrm{~s}, 3 \mathrm{H}), 1.86(\mathrm{~d}, 1 \mathrm{H}, J=3.9 \mathrm{~Hz}), 1.47(\mathrm{~d}, 1 \mathrm{H}, J=6.6 \mathrm{~Hz}), 1.43(\mathrm{~s}, 3$ H), $1.12(\mathrm{~d}, 1 \mathrm{H}, J=6.6 \mathrm{~Hz}), 0.71(\mathrm{~s}, 1 \mathrm{H}), 0.62(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 159.1$,
 139 (10), 121 (5); HRMS [ESI $\left.{ }^{+}\right] m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{15} \mathrm{O}\left[\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}\right]$ 187.1117, found 187.1117.

(2-(Allyloxy)phenyl)(3-methylbicyclo[1.1.0]butan-1-yl)methanol (3-50). Prepared according to General Procedure G using 1-chloromethyl-1-methyl-2,2-dibromocyclopropane (1.21 g, 4.62 $\mathrm{mmol}, 2.50 \mathrm{eq}), \operatorname{MeLi}(2.9 \mathrm{~mL}, 4.6 \mathrm{mmol}, 2.5 \mathrm{eq}), t-\mathrm{BuLi}(2.7 \mathrm{~mL}, 4.6 \mathrm{mmol}, 2.5 \mathrm{eq})$, and a solution of $O$-allylsalicyaldehyde ( $0.300 \mathrm{~g}, 1.85 \mathrm{mmol}, 1.00 \mathrm{eq}$ ) in THF ( 1 mL ). Purification by chromatography on $\mathrm{SiO}_{2}$ afforded $\mathbf{3 - 5 0}(0.154 \mathrm{~g}, 0.669 \mathrm{mmol}, 36 \%)$ as a colorless oil: IR (ATR)

3418, 2920, 1601, 1489, 1454, 1237, 998, $753 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (300 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 7.38$ (dd, 1 H , $J=7.5 \mathrm{~Hz}, 1.8 \mathrm{~Hz}), 7.22(\mathrm{td}, 1 \mathrm{H}, J=7.5 \mathrm{~Hz}, 1.5 \mathrm{~Hz}), 6.96(\mathrm{app} \mathrm{t}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.87(\mathrm{~d}, 1 \mathrm{H}$, $J=8.1 \mathrm{~Hz}), 6.06,5.42,5.30\left(\mathrm{ABX}, 3 \mathrm{H}, J_{\mathrm{AB}}=1.5 \mathrm{~Hz}, J_{\mathrm{AX}}=15.9 \mathrm{~Hz}, J_{\mathrm{BX}}=10.5 \mathrm{~Hz}\right), 5.22(\mathrm{~d}, 1$ $\mathrm{H}, J=6.6 \mathrm{~Hz}), 4.58(\mathrm{dt}, 2 \mathrm{H}, J=5.1 \mathrm{~Hz}, 1.5 \mathrm{~Hz}), 2.90(\mathrm{~d}, 1 \mathrm{H}, J=6.6 \mathrm{~Hz}), 1.51(\mathrm{~d}, 1 \mathrm{H}, J=6.3$ $\mathrm{Hz}), 1.42(\mathrm{~s}, 3 \mathrm{H}), 1.12(\mathrm{~d}, 1 \mathrm{H}, J=6.6 \mathrm{~Hz}), 0.73(\mathrm{~s}, 1 \mathrm{H}), 0.64(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 155.6,133.0,131.6,128.5,128.2,120.9,117.7,111.8,70.0,68.9,33.4,32.8,15.3,11.1$, 10.7; MS $\left[\mathrm{ESI}^{+}\right] m / z 213$ (100), 172 (35), 131 (15), 118 (20); $\mathrm{HRMS}\left[\mathrm{ESI}^{+}\right]$calcd for $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{O}$ $\left[\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}\right]$ 213.1274, found 213.1273.


3-51
(3-Methylbicyclo[1.1.0]butan-1-yl)(pyridin-2-yl)methanol (3-51). Prepared according to General Procedure G from 1-chloromethyl-1-methyl-2,2-dibromocyclopropane (1.50 g, 5.72 $\mathrm{mmol})$, $\mathrm{MeLi}\left(3.6 \mathrm{~mL}, 5.7 \mathrm{mmol}, \mathrm{c}=1.6 \mathrm{M} \mathrm{in}_{\mathrm{Et}}^{2} \mathrm{O}\right), t-\mathrm{BuLi}(3.4 \mathrm{~mL}, 5.7 \mathrm{mmol}, \mathrm{c}=1.7 \mathrm{M}$ in pentane), and 2-pyridinecarboxaldehyde ( $0.22 \mathrm{~mL}, 2.3 \mathrm{mmol}$ ) in $\mathrm{Et}_{2} \mathrm{O}(12 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ ( $15 \% \mathrm{EtOAc} /$ hexanes) afforded alcohol 3-51 ( $0.198 \mathrm{~g}, 1.13 \mathrm{mmol}, 49 \%$ ) as a yellow oil: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.54(\mathrm{~d}, 1 \mathrm{H}, J=4.4 \mathrm{~Hz}), 7.70(\mathrm{td}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}$, $1.6 \mathrm{~Hz}), 7.35(\mathrm{~d}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.21(\mathrm{dd}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}, 5.2 \mathrm{~Hz}), 5.01(\mathrm{~d}, 1 \mathrm{H}, J=4.0 \mathrm{~Hz})$, $4.45(\mathrm{~d}, 1 \mathrm{H}, J=4.0 \mathrm{~Hz}), 1.58(\mathrm{~s}, 3 \mathrm{H}), 1.47(\mathrm{~d}, 1 \mathrm{H}, J=6.4 \mathrm{~Hz}), 1.35(\mathrm{~d}, 1 \mathrm{H}, J=6.4 \mathrm{~Hz}), 0.71$ $(\mathrm{s}, 1 \mathrm{H}), 0.53(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 161.3,148.0,136.6,122.4,121.0,71.1$, 34.1, 31.3, 16.1, 10.5.


1-(3-Methylbicyclo[1.1.0]butan-1-yl)-1-phenylethan-1-ol (3-52). Prepared according to General Procedure G from 1-chloromethyl-1-methyl-2,2-dibromocyclopropane (2.00 g, 7.62 $\mathrm{mmol}, 2.50 \mathrm{eq}), \mathrm{MeLi}\left(4.8 \mathrm{~mL}, 7.6 \mathrm{mmol}, \mathrm{c}=1.6 \mathrm{M} \mathrm{in}_{\mathrm{Et}}^{2} \mathrm{O}\right), t-\mathrm{BuLi}(4.5 \mathrm{~mL}, 7.6 \mathrm{mmol}, \mathrm{c}=1.7$ M in pentane) and acetophenone ( $0.36 \mathrm{~mL}, 3.0 \mathrm{mmol}$ ). Purification by chromatography on $\mathrm{SiO}_{2}$ ( $15 \% \mathrm{EtOAc} /$ hexanes ) to afforded 3-52 ( $0.440 \mathrm{~g}, 2.34 \mathrm{mmol}, 77 \%$ ) as a colorless oil: IR (ATR) 3432, 2921, 1446, 1372, 1088, 961, 760, $698 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.54(\mathrm{~d}, 2 \mathrm{H}, J$ $=7.6 \mathrm{~Hz}), 7.34(\mathrm{t}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.25(\mathrm{t}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 1.89(\mathrm{~s}, 1 \mathrm{H}), 1.61(\mathrm{~s}, 3 \mathrm{H}), 1.60(\mathrm{~s}$, $3 \mathrm{H}), 1.34(\mathrm{~d}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 1.26(\mathrm{~d}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 0.58(\mathrm{~s}, 1 \mathrm{H}), 0.47(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 148.0,128.0,126.8,125.2,73.9,32.2,30.3,28.6,20.7,12.6,10.8 ; \mathrm{MS}^{2}\left[\mathrm{ESI}^{+}\right]$ $m / z 171$ (60), 154 (30), 143 (100), 129 (50); HRMS [ESI $\left.{ }^{+}\right] m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{15}\left[\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}\right]$ 171.1174, found 171.1169.


Phenyl(3-propylbicyclo[1.1.0]butan-1-yl)methanol (3-53). Prepared according to General Procedure G using the corresponding tribromide ( $2.03 \mathrm{~g}, 6.06 \mathrm{mmol}, 2.54 \mathrm{eq}$ ), $\mathrm{MeLi}(3.8 \mathrm{~mL}, 6.1$ $\mathrm{mmol}, \mathrm{c}=1.6 \mathrm{M}$ in $\mathrm{Et}_{2} \mathrm{O}$ ), $t$ - $\mathrm{BuLi}(3.6 \mathrm{~mL}, 6.1 \mathrm{mmol}, \mathrm{c}=1.7 \mathrm{M}$ in pentane), and benzaldehyde $(0.25 \mathrm{~mL}, 2.4 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(15 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(15-25 \%$ EtOAc/hexanes) afforded alcohol 3-53 ( $0.165 \mathrm{~g}, 0.816 \mathrm{mmol}, 34 \%$ over 3 steps) as a colorless oil: IR (ATR) $3353,2926,1453,1085,1002,964,754,698 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.44-$ $7.27(\mathrm{~m}, 5 \mathrm{H}), 5.05(\mathrm{~d}, 1 \mathrm{H}, J=3.9 \mathrm{~Hz}), 1.93(\mathrm{~d}, 1 \mathrm{H}, J=4.2 \mathrm{~Hz}), 1.80-1.59(\mathrm{~m}, 2 \mathrm{H}), 1.52-1.38$
$(\mathrm{m}, 2 \mathrm{H}), 1.48(\mathrm{~d}, 1 \mathrm{H}, J=6.6 \mathrm{~Hz}), 1.14(\mathrm{~d}, 1 \mathrm{H}, J=6.6 \mathrm{~Hz}), 0.95(\mathrm{t}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}), 0.68(\mathrm{~s}, 1$ H), $0.61(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 143.2,128.4,128.3,127.6,127.1,126.6,73.4$, 31.9, 31.5, 28.0, 22.7, 16.7, 15.0, 14.0; $\mathrm{MS}(\mathrm{ESI})^{+} m / z 185(75), 143(90), 129(100) ; \mathrm{HRMS}^{(\mathrm{ESI})^{+}}$ $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{17}\left[\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}\right]$ 185.1325, found 185.1325.

(3-Isopropylbicyclo[1.1.0]butan-1-yl)(phenyl)methanol (3-54). Prepared according to General Procedure G using from the corresponding tribromide ( $3.30 \mathrm{~g}, 9.86 \mathrm{mmol}, 2.17 \mathrm{eq}$ ), MeLi $(7.1$ $\mathrm{mL}, 11.4 \mathrm{mmol}, \mathrm{c}=1.6 \mathrm{M}$ in $\left.\mathrm{Et}_{2} \mathrm{O}\right), t-\operatorname{BuLi}(6.7 \mathrm{~mL}, 11.4 \mathrm{mmol}, \mathrm{c}=1.7 \mathrm{M}$ in pentane $)$, and benzaldehyde ( $0.46 \mathrm{~mL}, 4.5 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(20 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ (15$25 \% \mathrm{EtOAc} /$ hexanes $)$ afforded alcohol 3-54 ( $0.372 \mathrm{~g}, 1.84 \mathrm{mmol}, 40 \%$ ) as a colorless oil: IR (ATR) 3379, 2959, 1453, 1083, 1002, 963, 891, 751, $698 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.44-7.27(\mathrm{~m}, 5 \mathrm{H}), 5.12(\mathrm{~d}, 1 \mathrm{H}, J=4.0 \mathrm{~Hz}), 2.04($ pent, $1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 1.89(\mathrm{~d}, 1 \mathrm{H}, J=4.4$ $\mathrm{Hz}), 1.51(\mathrm{~d}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 1.18(\mathrm{~d}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 1.01(\mathrm{~d}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}), 0.95(\mathrm{~d}, 3 \mathrm{H}, J$ $=6.8 \mathrm{~Hz}), 0.53(\mathrm{~s}, 1 \mathrm{H}), 0.48(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 143.4,128.4,127.7,126.6$, 72.7, 29.3, 28.3, 24.6, 21.8, 21.2, 18.8; MS (ESI) ${ }^{+} m / z 203,201$ (20), 185 (100), 157 (3); HRMS $(\mathrm{ESI})^{+} m / z$ cald for $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{O}[\mathrm{M}+\mathrm{H}]$ 203.1430, found 203.1433.

(3-Phenethylbicyclo[1.1.0]butan-1-yl)(phenyl)methanol (3-55). Prepared using General Procedure G from the corresponding tribromide $(0.968 \mathrm{~g}, 2.44 \mathrm{mmol}, 2.00 \mathrm{eq}), \mathrm{MeLi}(1.5 \mathrm{~mL}, 2.4$ $\mathrm{mmol}, \mathrm{c}=1.6 \mathrm{M}$ in $\left.\mathrm{Et}_{2} \mathrm{O}\right), t-\mathrm{BuLi}(1.4 \mathrm{~mL}, 2.4 \mathrm{mmol}, \mathrm{c}=1.7 \mathrm{M}$ in pentane $)$, and benzaldehyde $(0.12 \mathrm{~mL}, 1.2 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(8 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(15-25 \%$

EtOAc/hexanes) afforded alcohol 3-55 ( $0.153 \mathrm{~g}, 0.579 \mathrm{mmol}, 47 \%$ ) as a yellow tinted oil: IR (ATR) $3376,3026,2922,1603,1495,1453,1013,699 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.41-$ $7.31(\mathrm{~m}, 4 \mathrm{H}), 7.30-7.26(\mathrm{~m}, 3 \mathrm{H}), 7.21-7.15(\mathrm{~m}, 3 \mathrm{H}), 5.02(\mathrm{~d}, 1 \mathrm{H}, J=4.4 \mathrm{~Hz}), 2.82-2.66(\mathrm{~m}, 2$ H), 2.18-2.08 (m, 1 H$), 2.07-1.97(\mathrm{~m}, 1 \mathrm{H}), 1.86(\mathrm{~d}, 1 \mathrm{H}, J=4.0 \mathrm{~Hz}), 1.48(\mathrm{~d}, 1 \mathrm{H}, J=6.4 \mathrm{~Hz})$, $1.18(\mathrm{~d}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 0.67(\mathrm{~s}, 1 \mathrm{H}), 0.60(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 143.2,142.1$, $128.5,128.4,128.3,127.7,126.5,125.8,73.2,35.7,31.8,31.4,28.2,17.5,15.1$.


1-(3-Phenylbicyclo[1.1.0]butan-1-yl)pentan-1-ol (3-56). Prepared according to General Procedure G from the corresponding tribromide $(0.957 \mathrm{~g}, 2.59 \mathrm{mmol}, 2.50 \mathrm{eq}), \mathrm{MeLi}(1.6 \mathrm{~mL}, 2.6$ $\mathrm{mmol}, \mathrm{c}=1.62 \mathrm{M}$ in $\left.\mathrm{Et}_{2} \mathrm{O}\right), t-\mathrm{BuLi}(1.5 \mathrm{~mL}, 2.6 \mathrm{mmol}, \mathrm{c}=1.7 \mathrm{M}$ in pentane $)$, and valeraldehyde $(0.11 \mathrm{~mL}, 1.0 \mathrm{mmol}, 1.0 \mathrm{eq})$ in $\mathrm{Et}_{2} \mathrm{O}(6 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(10 \%$ EtOAc/hexanes) afforded alcohol 3-56 ( $0.181 \mathrm{~g}, 0.837 \mathrm{mmol}, 81 \%$ ) as a yellow-tinted oil: IR (ATR) $3396,2929,1602,1446,1104,1026,761,695 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.31-$ 7.21 (m, 4 H ), 7.14 (tt, $1 \mathrm{H}, J=6.8 \mathrm{~Hz}, 1.6 \mathrm{~Hz}$ ), 3.81 (ddd, $1 \mathrm{H}, J=4.0 \mathrm{~Hz}, 4.0 \mathrm{~Hz}, 4.0 \mathrm{~Hz}$ ), 2.35 $(\mathrm{d}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 2.19(\mathrm{dd}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}, 0.8 \mathrm{~Hz}), 1.67-1.21(\mathrm{~m}, 6 \mathrm{H}), 1.15(\mathrm{~s}, 1 \mathrm{H}), 1.13(\mathrm{~d}$, $1 \mathrm{H}, J=4.0 \mathrm{~Hz}), 1.09(\mathrm{~s}, 1 \mathrm{H}), 0.85(\mathrm{t}, 3 \mathrm{H}, J=6.8 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 137.0$, 128.6, 125.3, 125.2, 125.1, 70.2, 36.3, 12.2, 30.2, 27.6, 22.5, 17.4, 14.0; MS (ESI) $m / z 217$ (10), 199 (100), 177 (15), 143 (40), 117 (10); HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{O}[\mathrm{M}+\mathrm{H}]$ 217.1587, found 217.1586.


2,2-Dimethyl-1-(3-phenylbicyclo[1.1.0]butan-1-yl)propan-1-ol (3-57). Prepared according to General Procedure G from the corresponding tribromide ( $2.24 \mathrm{~g}, 6.05 \mathrm{mmol}, 1.75 \mathrm{eq}$ ), MeLi ( 4.31 $\left.\mathrm{mL}, 6.91 \mathrm{mmol}, \mathrm{c}=1.6 \mathrm{M} \mathrm{in}_{\mathrm{Et}}^{2} \mathrm{O}\right), t-\mathrm{BuLi}(4.07 \mathrm{~mL}, 6.91 \mathrm{mmol}, \mathrm{c}=1.7 \mathrm{M}$ in pentane $)$, and neat trimethylacetaldehyde $(0.40 \mathrm{~mL}, 3.5 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(12 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(25 \% \mathrm{EtOAc} /$ hexanes $)$ followed by a second chromatography on neutral $\mathrm{Al}_{2} \mathrm{O}_{3}(10 \%$ EtOAc/hexanes) to afford alcohol 3-57 ( $0.282 \mathrm{~g}, 1.30 \mathrm{mmol}, 38 \%$ ) as a light yellow oil: IR (ATR) 3562, 2951, 2866, 1602, 1479, 1362, 1107, 1007, 950, 773, $695 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 7.37-7.23(\mathrm{~m}, 4 \mathrm{H}), 7.21-7.13(\mathrm{~m}, 1 \mathrm{H}), 3.41(\mathrm{~d}, 1 \mathrm{H}, J=3.3 \mathrm{~Hz}), 2.51(\mathrm{~d}, 1 \mathrm{H}, J=6.3 \mathrm{~Hz}), 2.20$ $(\mathrm{d}, 1 \mathrm{H}, J=6.3 \mathrm{~Hz}), 1.28(\mathrm{~s}, 1 \mathrm{H}), 1.13(\mathrm{~s}, 1 \mathrm{H}), 1.01(\mathrm{~s}, 9 \mathrm{H}), 0.94(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 137.4,128.7,125.2,125.1,77.7,36.6,35.2,29.6,25.9,25.7,14.1 ; \mathrm{MS}(\mathrm{ESI})^{+} \mathrm{m} / \mathrm{z} 217$, 215, 213, 247 (20), 231 (30), 215 (30), 199 (100), 143 (20); HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{21} \mathrm{O}$ [M+H] 217.1587, found 217.1586.


## 2,2-Dimethyl-1-(3-(4-(trifluoromethyl)phenyl)bicyclo[1.1.0]butan-1-yl)propan-1-ol

Prepared according to General Procedure G from the corresponding tribromide ( $0.822 \mathrm{~g}, 1.88$ $\mathrm{mmol}, 2.00 \mathrm{eq}), \mathrm{MeLi}\left(1.2 \mathrm{~mL}, 1.9 \mathrm{mmol}, \mathrm{c}=1.62 \mathrm{M} \mathrm{in}_{\mathrm{Et} 2} \mathrm{O}\right), t-\mathrm{BuLi}(1.1 \mathrm{~mL}, 1.9 \mathrm{mmol}, \mathrm{c}=1.7$ M in pentane $)$, and trimethylacetaldehyde $(0.11 \mathrm{~mL}, 0.94 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(5 \mathrm{~mL})$. Purification by two iterations of chromatography on $\mathrm{SiO}_{2}(5-10 \% \mathrm{EtOAc} /$ hexanes $)$ afforded alcohol 3-58 (0.071 g, $0.25 \mathrm{mmol}, 27 \%$ ) as a colorless solid: $\mathrm{Mp} 80-84^{\circ} \mathrm{C}$; IR (ATR) 2955, 1614, 1324, 1163, 1109,

1062, 1013, 951, 841, $688 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.50(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}$ ), $7.32(\mathrm{~d}$, $2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 3.34(\mathrm{~d}, 1 \mathrm{H}, J=4.4 \mathrm{~Hz}), 2.53(\mathrm{~d}, 1 \mathrm{H}, J=6.4 \mathrm{~Hz}), 2.21(\mathrm{dd}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}, 1.2$ $\mathrm{Hz}), 1.34(\mathrm{~s}, 1 \mathrm{H}), 1.18(\mathrm{~s}, 1 \mathrm{H}), 1.01(\mathrm{~s}, 1 \mathrm{H}), 0.98(\mathrm{~s}, 9 \mathrm{H}), 0.95(\mathrm{~d}, 1 \mathrm{H}, J=4.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.6,129.4,126.9\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 125.4,125.3\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right), 124.4(\mathrm{q}$, $\left.{ }^{1} J_{\mathrm{CF}}=270 \mathrm{~Hz}\right), 80.8,36.8,35.5,30.2,28.1,25.8,14.5 ;{ }^{19} \mathrm{~F} \mathrm{NMR}\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-62.1$; MS $(\mathrm{ESI})^{+} m / z 267$ (100), 189 (40), 137 (50); $\mathrm{HRMS}(\mathrm{ESI})^{+} m / z$ calcd for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~F}_{3}\left[\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}\right]$ 267.1361, found 267.1354.


Phenyl(3-(4-(trifluoromethyl)phenyl)bicyclo[1.1.0]butan-1-yl)methanol (3-59). Prepared according to General Procedure G the corresponding tribromide ( $1.49 \mathrm{~g}, 3.41 \mathrm{mmol}, 2.00 \mathrm{eq}$ ), $\operatorname{MeLi}\left(2.1 \mathrm{~mL}, 3.4 \mathrm{mmol}, \mathrm{c}=1.6 \mathrm{M}\right.$ in $\left.\mathrm{Et}_{2} \mathrm{O}\right), t-\mathrm{BuLi}(2.0 \mathrm{~mL}, 341 \mathrm{mmol}, \mathrm{c}=1.7 \mathrm{M}$ in pentane $)$, and benzaldehyde ( $0.17 \mathrm{~mL}, 1.7 \mathrm{mmol}, 1.0 \mathrm{eq}$ ) in $\mathrm{Et}_{2} \mathrm{O}(8 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(15 \% \mathrm{EtOAc} /$ hexanes $)$ afforded alcohol 3-59 ( $0.281 \mathrm{~g}, 54 \%$ ) as a yellow tinted solid. ${ }^{1} \mathrm{H}$ NMR revealed that the material was contaminated with minor impurities, so the material was recrystallized from hexanes $(\sim 20 \mathrm{~mL})$ to afford $0.132 \mathrm{~g}(25 \%)$ of the product as light yellow crystals: Mp 99-102 ${ }^{\circ} \mathrm{C}$; IR (ATR) $3327,2922,1617,1324,1106,1065,1006,835 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.48(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.29-7.23(\mathrm{~m}, 3 \mathrm{H}), 7.19(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.15-$ $7.09(\mathrm{~m}, 2 \mathrm{H}), 4.85(\mathrm{~s}, 1 \mathrm{H}), 2.49(\mathrm{~d}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 2.10(\mathrm{dd}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}, 0.8 \mathrm{~Hz}), 1.81(\mathrm{~d}$, $1 \mathrm{H}, J=2.0 \mathrm{~Hz}), 1.31(\mathrm{~s}, 1 \mathrm{H}), 1.25(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 141.7,141.3,128.4$, 128.0, 127.2 $\left({ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 126.5,125.7,125.1\left({ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 124.5\left({ }^{1} J_{\mathrm{CF}}=270\right), 123.1$, 72.6,
32.6, 31.7, 29.8, 18.4; MS (ESI) ${ }^{+} m / z 287$ (100), 238 (10), 200 (10), 159 (10); HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{~F}_{3}\left[\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}\right]$ 287.1048, found 287.1038.

(3-(4-Fhlorophenyl)bicyclo[1.1.0]butan-1-yl)(phenyl)methanol (3-60). Prepared according to General Procedure G from the corresponding tribromide ( $0.860 \mathrm{~g}, 2.13 \mathrm{mmol}, 2.00 \mathrm{eq}$ ), MeLi (1.3 $\mathrm{mL}, 2.1 \mathrm{mmol}, \mathrm{c}=1.6 \mathrm{M}$ in $\left.\mathrm{Et}_{2} \mathrm{O}\right), t-\mathrm{BuLi}(1.3 \mathrm{~mL}, 2.1 \mathrm{mmol}, \mathrm{c}=1.7 \mathrm{M}$ in pentane $)$, and benzaldehyde ( $0.11 \mathrm{~mL}, 1.1 \mathrm{mmol}, 1.0 \mathrm{eq}$ ) in $\mathrm{Et}_{2} \mathrm{O}(5 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(10-25 \% \mathrm{EtOAc} / \mathrm{hexanes})$ afforded alcohol 3-60 ( $0.193 \mathrm{~g}, 0.713 \mathrm{mmol}, 67 \%$ ) as a red-tinted oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 7.30-7.24 (m, 3 H ), 7.22, 7.04 (AA'BB', 4 H ), 7.17-7.12 (m, 2 H), $4.85(\mathrm{~s}, 1 \mathrm{H}), 2.40(\mathrm{~d}, 1 \mathrm{H}, J=6.4 \mathrm{~Hz}), 2.01(\mathrm{~d}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 1.76(\mathrm{~s}, 1 \mathrm{H}), 1.24(\mathrm{~s}, 1 \mathrm{H})$, $1.18(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 141.8,135.1,130.9,128.4,128.3,127.9,126.9$, $126.5,72.7,32.4,31.3,28.3,18.0,14.2$.


3-(3-(4-(Trifluoromethyl)phenyl)bicyclo[1.1.0]butan-1-yl)pentan-3-ol (3-61). Prepared according to General Procedure G from the corresponding tribromide $(0.911 \mathrm{~g}, 2.09 \mathrm{mmol}, 2.50$ $\mathrm{eq}), \mathrm{MeLi}\left(1.3 \mathrm{~mL}, 2.1 \mathrm{mmol}, \mathrm{c}=1.6 \mathrm{M} \mathrm{in}^{\mathrm{Et}} \mathrm{t}_{2} \mathrm{O}\right), t-\mathrm{BuLi}(1.2 \mathrm{~mL}, 2.1 \mathrm{mmol}, \mathrm{c}=1.7 \mathrm{M}$ in pentane $)$, and 3-pentanone ( $0.09 \mathrm{~mL}, 0.83 \mathrm{mmol}, 1.0 \mathrm{eq}$ ) in $\mathrm{Et}_{2} \mathrm{O}(5 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ (15\% EtOAc/hexanes) afforded alcohol 3-61 ( $0.149 \mathrm{~g}, 0.524 \mathrm{mmol}, 63 \%$ ) (OH not visible) as a colorless oil: IR (ATR) 3469, 2971, 1616, 1321, 1162, 1106, $841 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz ,
$\left.\mathrm{CDCl}_{3}\right) \delta 7.51(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.37(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 2.31(\mathrm{~s}, 2 \mathrm{H}), 1.65,1.40,0.85\left(\mathrm{ABX}_{3}\right.$, $\left.2+2+6 \mathrm{H}, J_{\mathrm{AB}}=7.6 \mathrm{~Hz}, \partial_{\mathrm{AB}}=0.15 \mathrm{ppm}\right), 1.02(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.2$, $126.9\left({ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 126.0,125.3\left({ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 124.5\left({ }^{1} J_{\mathrm{CF}}=269\right), 74.5,34.1,30.6,30.5,19.1$, 7.9; MS (ESI) ${ }^{+} m / z 267(100), 225(10), 173(5) ; \mathrm{HRMS}(\mathrm{ESI})^{+} m / z$ calcd for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~F}_{3}\left[\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}\right]$ 267.1361, found 267.1352.


3-(3-(4-Chlorophenyl)bicyclo[1.1.0]butan-1-yl)pentan-3-ol (3-62). Prepared according to General Procedure G from the corresponding tribromide (1.14 g, $2.83 \mathrm{mmol}, 2.50 \mathrm{eq})$, MeLi ( 1.8 $\mathrm{mL}, 2.8 \mathrm{mmol}, \mathrm{c}=1.6 \mathrm{M}$ in $\left.\mathrm{Et}_{2} \mathrm{O}\right), t-\mathrm{BuLi}(1.7 \mathrm{~mL}, 2.8 \mathrm{mmol}, \mathrm{c}=1.7 \mathrm{M}$ in pentane $)$, and 3pentanone ( $0.12 \mathrm{~mL}, 1.1 \mathrm{mmol}, 1.0 \mathrm{eq}$ ) in $\mathrm{Et}_{2} \mathrm{O}(6 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ ( $10 \% \mathrm{EtOAc} /$ hexanes ) afforded alcohol 3-62 ( $0.202 \mathrm{~g}, 0.806 \mathrm{mmol}, 71 \%$ ) as a colorless oil: IR (ATR) $3469,2971,1616,1321,1162,1106,841 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (OH not visible) $\delta 7.25-7.17(\mathrm{~m}, 4 \mathrm{H}), 2.23(\mathrm{~s}, 2 \mathrm{H}), 1.63,1.41,0.84\left(\mathrm{ABX}_{3}, 2+2+6 \mathrm{H}, J_{\mathrm{AB}}=7.6 \mathrm{~Hz}, \partial_{\mathrm{AB}}=\right.$ $0.24 \mathrm{ppm}), 0.96(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 135.9,130.6,128.6,127.2,74.5,32.3$, 30.6, 30.2, 18.7, 8.0; MS (ESI) ${ }^{+} m / z 233$ (100), 191 (20), 167 (20), 1491 (10), 135 (40); HRMS $(\mathrm{ESI})^{+} \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{Cl}\left[\mathrm{M}+\mathrm{H}-\mathrm{H}_{2} \mathrm{O}\right]$ 233.1097, found 233.1090.


General Procedure $H \quad$ (Alcohol Deoxyfluorination): ((3-Fluoro-3methylcyclobutylidene)methyl)benzene (3-46). To a suspension of $\mathrm{KHF}_{2}(0.070 \mathrm{~g}, 0.90 \mathrm{mmol})$ in dichloromethane ( 1 mL ) was added diisopropylamine ( $0.51 \mathrm{~mL}, 3.6 \mathrm{mmol}, 4.0 \mathrm{eq}$ ) in a polypropylene reaction vessel. The suspension was cooled under $\mathrm{N}_{2}$ to $-78{ }^{\circ} \mathrm{C}$, then pyridine $\cdot 9$ HF ( $1.2 \mathrm{~mL}, 46 \mathrm{mmol} \mathrm{HF}, 51 \mathrm{eq} \mathrm{HF}$ ) was added. The suspension was stirred for 5 min at $-78^{\circ} \mathrm{C}$, then a solution of alcohol $\mathbf{3 - 4 5}(0.157 \mathrm{~g}, 0.90 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ was added over $\sim 30 \mathrm{~s}$. After stirring at $-78^{\circ} \mathrm{C}$ for 10 min , the mixture was quenched at $-78^{\circ} \mathrm{C}$ with sat aq. $\mathrm{KF}(5 \mathrm{~mL})$, then the reaction was warmed to rt and diluted with water $(25 \mathrm{~mL})$. The material was extracted with ether $(2 \times 20 \mathrm{~mL})$. The organic layers were washed with saturated aq. $\mathrm{NaHCO}_{3}(30 \mathrm{~mL})$ and brine ( 30 mL ), then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Chromatography on $\mathrm{SiO}_{2}(2.5 \%$ EtOAc/hexanes) afforded methylenecyclobutane 3-46 ( $0.090 \mathrm{~g}, 0.51 \mathrm{mmol}, 57 \%$ ) as a colorless oil: IR (ATR) 2794, 1602, 1490, 1448, 1381, 1241, 1145, 914, $738 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.32(\mathrm{dd}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}, 7.2 \mathrm{~Hz}), 7.20(\mathrm{~d}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.31(\mathrm{t}, 1 \mathrm{H}, J=2.4 \mathrm{~Hz})$, $3.39\left(\mathrm{tdd}, 1 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=18.0 \mathrm{~Hz}, J=3.2 \mathrm{~Hz}, 3.2 \mathrm{~Hz}\right), 3.26\left(\mathrm{tdd}, 1 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=18.4 \mathrm{~Hz}, J=3.2 \mathrm{~Hz}, 3.2\right.$ $\mathrm{Hz}), 3.11-3.00(\mathrm{~m}, 1 \mathrm{H}), 2.95-2.84(\mathrm{~m}, 1 \mathrm{H}), 1.56\left(\mathrm{~d}, 3 \mathrm{H},{ }^{4} J_{\mathrm{CF}}=21.6 \mathrm{~Hz}\right) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 137.4,131.9\left({ }^{3} \mathrm{~J}_{\mathrm{CF}}=16 \mathrm{~Hz}\right), 128.5,127.2,126.5,123.8,123.7,92.8\left({ }^{1} J_{\mathrm{CF}}=206 \mathrm{~Hz}\right)$, $46.1\left({ }^{3} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 45.7\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 24.2\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-$ 131.2; HRMS [ES $\left.{ }^{+}\right]$calcd for $\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{~F}\left[\mathrm{M}^{+}\right]$176.1001, found 175.0056.

Notes on pyridine•9 HF: For simplicity, the pyridine•9 HF stoichiometry was initially calculated based on a molecular formula of $\mathrm{C}_{5} \mathrm{H}_{6} \mathrm{FN}$ (molecular weight $=99.11$ ), which is
typically shown on data sheets for this reagent. However, it is more appropriate to consider stoichiometry in terms of equivalents of HF. The equivalents of HF can be calculated in the following manner: $X m L$ reagent * $1.1 \mathrm{~g} / \mathrm{mL}$ (density) * 0.7 (percentage HF by weight) / $20 \mathrm{~g} / \mathrm{mol}$ $($ HF molecular weight $) * 1000=$ mmol HF. For reference, 1 mmol of pyridine $\bullet 9$ HF (as calculated by a molecular weight of 99.11) $=3.5 \mathrm{mmol}$ of HF .

This reagent should be handled with great care using appropriate personal protective equipment. If skin contact with this reagent occurs, apply calcium gluconate gel immediately and liberally. After quenching the fluorinations with aqueous NaOH , the basicity of the aqueous layer should be confirmed by pH paper during the extraction step.


1-((3-Fluoro-3-methylcyclobutylidene)methyl)-4-(trifluoromethyl)benzene (3-63). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.051 \mathrm{~g}, 0.65 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine $(0.37 \mathrm{~mL}, 2.6 \mathrm{mmol}, 4.0 \mathrm{eq})$, pyridine $\cdot 9 \mathrm{HF}(1.2 \mathrm{~mL}, 46 \mathrm{mmol} \mathrm{HF}, 71 \mathrm{eq} \mathrm{HF})$, and alcohol 3-47 ( $0.158 \mathrm{~g}, 0.652 \mathrm{mmol}, 1.00 \mathrm{eq}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \%$ EtOAc in hexanes) afforded methylenecyclobutane 3-63 ( $0.073 \mathrm{~g}, 0.30 \mathrm{mmol}, 46 \%$ ) as a colorless oil: IR (ATR) 2977, 2919, 1617, 1415, 1382, 1323, 1243, 1164, 1121, $1067 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (500 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.56(\mathrm{~d}, 2 \mathrm{H}, J=8.5 \mathrm{~Hz}), 7.28(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 6.34(\mathrm{t}, 1 \mathrm{H}, J=2.0 \mathrm{~Hz}), 3.41$ $\left(\mathrm{ddt}, 1 \mathrm{H},{ }^{3} J_{\mathrm{FH}}=19.5 \mathrm{~Hz}, J=17.0 \mathrm{~Hz}, 3.0 \mathrm{~Hz}\right), 3.41\left(\mathrm{ddt}, 1 \mathrm{H},{ }^{3} J_{\mathrm{FH}}=19.0 \mathrm{~Hz}, J=17.0 \mathrm{~Hz}, 3.0\right.$ $\mathrm{Hz}), 3.10-3.01(\mathrm{~m}, 1 \mathrm{H}), 2.97-2.87(\mathrm{~m}, 1 \mathrm{H}), 1.56\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} \mathrm{~J}_{\mathrm{FH}}=21.5 \mathrm{~Hz}\right) ;{ }^{13} \mathrm{C}$ NMR ( 125 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 140.8,135.4\left({ }^{3} J_{\mathrm{CF}}=15 \mathrm{~Hz}\right), 128.2\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 127.2,125.6,125.4\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=4\right.$ $\mathrm{Hz}), 122.9,122.8,92.6\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=207 \mathrm{~Hz}\right), 46.2\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 45.8\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 24.2(\mathrm{~d}$,
$\left.{ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\delta\left(471 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-131.4,-62.4 ;$ HRMS [ES] calcd for $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{~F}_{4}[\mathrm{M}]$ 244.0875, found 244.0854.


1-Chloro-4-((3-fluoro-3-methylcyclobutylidene)methyl)benzene (3-64). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.035 \mathrm{~g}, 0.45 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.25 \mathrm{~mL}, 1.8$ mmol, 4.0 eq ), pyridine • $9 \mathrm{HF}(0.81 \mathrm{~mL}, 31 \mathrm{mmol} \mathrm{HF}, 69 \mathrm{eq} \mathrm{HF})$, and alcohol 3-48 ( $0.094 \mathrm{~g}, 0.45$ mmol, 1.0 eq ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 3 mL ). Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc}$ in hexanes) afforded methylenecyclobutane 3-64 ( $0.055 \mathrm{~g}, 0.26 \mathrm{mmol}, 58 \%$ ) as a colorless oil: IR (ATR) 2975, 2913, 1655, 1492, 1405, 1381, 1242, 1146, 1092, $915,827 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.27,7.11\left(\mathrm{AA}^{\prime} \mathrm{BB}^{\prime}, 4 \mathrm{H}, J_{\mathrm{AB}}=8.4 \mathrm{~Hz}, \partial_{\mathrm{AB}}=0.13 \mathrm{ppm}\right), 6.26(\mathrm{t}, 1 \mathrm{H}, J=2.4 \mathrm{~Hz}), 3.36(\mathrm{ddt}, 1$ $\left.\mathrm{H},{ }^{3} J_{\mathrm{HF}}=19.2 \mathrm{~Hz}, J=16.4 \mathrm{~Hz}, 2.8 \mathrm{~Hz}\right), 3.25\left(\mathrm{ddt}, 1 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=19.2 \mathrm{~Hz}, J=16.8 \mathrm{~Hz}, 2.4 \mathrm{~Hz}\right), 3.07-$ $2.96(\mathrm{~m}, 1 \mathrm{H}), 2.95-2.84(\mathrm{~m}, 1 \mathrm{H}), 1.55\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=21.6 \mathrm{~Hz}\right) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 135.9,132.7\left({ }^{3} J_{\mathrm{CF}}=15 \mathrm{~Hz}\right), 132.0,128.6,128.3,122.7\left({ }^{4} J_{\mathrm{CF}}=7 \mathrm{~Hz}\right), 92.7\left({ }^{1} J_{\mathrm{CF}}=207 \mathrm{~Hz}\right), 46.1$ $\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 45.6\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 24.2\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-131.3 ;$ HRMS (ESI ${ }^{+}$) $m / z$ calcd for $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{ClF}[\mathrm{M}-\mathrm{H}]$ 209.0533, found 209.0803.


1-((3-Fluoro-3-methylcyclobutylidene)methyl)-4-methoxybenzene (3-65). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.050 \mathrm{~g}, 0.64 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.36 \mathrm{~mL}, 2.6$ mmol, 4.0 eq), pyridine • $9 \mathrm{HF}(1.16 \mathrm{~mL}, 44.7 \mathrm{mmol}$, 69.7 eq HF), and alcohol 3-49 ( 0.131 g , $0.641 \mathrm{mmol}, 1.00 \mathrm{eq})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc}$ in
hexanes) afforded methylenecyclobutane 3-65 (0.061 g, $0.30 \mathrm{mmol}, 46 \%)$ as a colorless oil: IR (ATR) $\mathrm{cm}^{-1} 2972,1607,1511,1296,1246,1175,1144,1034,830 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 7.14(\operatorname{app~d}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}), 6.87(\operatorname{app~d}, 2 \mathrm{H}, J=9.0 \mathrm{~Hz}), 6.25(\mathrm{t}, 1 \mathrm{H}, J=2.1 \mathrm{~Hz})$, $3.81(\mathrm{~s}, 3 \mathrm{H}), 3.44-3.16(\mathrm{~m}, 2 \mathrm{H}), 3.09-2.95(\mathrm{~m}, 1 \mathrm{H}), 2.95-2.80(\mathrm{~m}, 1 \mathrm{H}), 1.56\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=21.6\right.$ $\mathrm{Hz}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 158.3,130.5,129.2,129.0,128.3,123.2\left({ }^{3} J_{\mathrm{CF}}=7 \mathrm{~Hz}\right), 114.0$, $92.8\left({ }^{1} J_{\mathrm{CF}}=206 \mathrm{~Hz}\right), 55.3,46.0\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 45.4\left({ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 24.3\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR (376 MHz, $\mathrm{CDCl}_{3}$ ) $\delta$-131.1; MS (ESI') $m / z 207$ (100), 187 (90), 171 (20), 140 (20); HRMS $\left(\mathrm{ESI}^{+}\right) m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{OF}[\mathrm{M}+\mathrm{H}]$ 207.1185, found 207.1179.


1-(Allyloxy)-2-((3-fluoro-3-methylcyclobutylidene)methyl)benzene (3-66). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.019 \mathrm{~g}, 0.24 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine $(0.14 \mathrm{~mL}$, $0.96 \mathrm{mmol}, 4.0 \mathrm{eq})$, pyridine • $9 \mathrm{HF}(0.43 \mathrm{~mL}$, 17 mmol HF , 69 eq HF ), and alcohol 3-50 ( 0.155 $\mathrm{g}, 0.239 \mathrm{mmol}, 1.00 \mathrm{eq})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1.5 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5-20 \%$ EtOAc in hexanes) afforded methylenecyclobutane 3-66 ( $0.019 \mathrm{~g}, 0.082 \mathrm{mmol}, 34 \%)$ as a colorless oil: IR (ATR) 2973, 2912, 1597, 1487, 1451, 1240, 1144, 997, 917, $748 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.24-7.13(\mathrm{~m}, 2 \mathrm{H}), 6.92(\mathrm{t}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.86(\mathrm{~d}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}), 6.69(\mathrm{app} \mathrm{s}, 1 \mathrm{H})$, $6.08,5.42,5.29\left(\mathrm{ABX}, 3 \mathrm{H}, J_{\mathrm{AB}}=1.2, J_{\mathrm{AX}}=17.2, J_{\mathrm{BX}}=10.4 \mathrm{~Hz}\right), 4.56(\mathrm{~d}, 2 \mathrm{H}, J=4.8 \mathrm{~Hz}), 3.44-$ $3.18(\mathrm{~m}, 2 \mathrm{H}), 3.05-2.85(\mathrm{~m}, 2 \mathrm{H}), 1.55\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=21.6 \mathrm{~Hz}\right) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 155.2,133.5,131.4\left({ }^{3} J_{\mathrm{CF}}=16 \mathrm{~Hz}\right), 127.5,127.4,126.7,120.6,117.9\left({ }^{4} J_{\mathrm{CF}}=7 \mathrm{~Hz}\right), 117.4,112.2$, $92.9\left({ }^{1} J_{\mathrm{CF}}=207 \mathrm{~Hz}\right), 69.2,46.2\left({ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 45.7\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 24.2\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR
(376 MHz, $\mathrm{CDCl}_{3}$ ) $\delta$-131.0; MS (ESI ${ }^{+} \mathrm{m} / \mathrm{z} 233$ (15), 213 (100), 172 (30), 140 (30), 128 (30); HRMS ( $\mathrm{ESI}^{+}$) $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{OF}[\mathrm{M}+\mathrm{H}]$ 213.1342, found 213.1335.


3-67
(1-(3-Fluoro-3-methylcyclobutylidene)ethyl)benzene (3-67). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.069 \mathrm{~g}, 0.88 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.50 \mathrm{~mL}, 3.5 \mathrm{mmol}, 4.0$ eq), pyridine • 9 HF ( $1.59 \mathrm{~mL}, 61.2 \mathrm{mmol} \mathrm{HF}, 69.4 \mathrm{eq} \mathrm{HF}$ ), and alcohol 3-52 ( $0.166 \mathrm{~g}, 0.882$ $\mathrm{mmol}, 1.00 \mathrm{eq})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(2.5 \% \mathrm{EtOAc}$ in hexanes) afforded methylenecyclobutane 3-67 ( $0.106 \mathrm{~g}, 0.557 \mathrm{mmol}, 63 \%$ ) as a colorless oil: IR (ATR) $\mathrm{cm}^{-1} 2974,1599,1496,1444,1379,1241,1168,912,759,697 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 7.34(\mathrm{t}, 2 \mathrm{H}, J=7.6), 7.29(\mathrm{t}, 2 \mathrm{H}, J=7.2 \mathrm{~Hz}), 7.22(\mathrm{td}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 3.29(\mathrm{t}$, $1 \mathrm{H}, J=18.8 \mathrm{~Hz}), 3.23(\mathrm{t}, 1 \mathrm{H}, J=19.6 \mathrm{~Hz}), 2.87(\mathrm{app} \mathrm{q}, 2 \mathrm{H}, J=12.4 \mathrm{~Hz}), 1.96(\mathrm{~s}, 3 \mathrm{H}), 1.52(\mathrm{~d}$, $\left.3 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=21.6 \mathrm{~Hz}\right) ;{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 140.8,128.9,128.1,128.0,126.54,126.51$, 126.4, $125.8\left({ }^{3} J_{\mathrm{CF}}=15 \mathrm{~Hz}\right), 92.1\left({ }^{1} J_{\mathrm{CF}}=205 \mathrm{~Hz}\right), 45.4\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 44.5\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 24.4$ $\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 17.5 ;{ }^{19} \mathrm{~F}$ NMR $\left(470 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-132.4 ;$ HRMS $\left(\mathrm{EI}^{+}\right) \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{13} \mathrm{H}_{15} \mathrm{~F}$ $\left[\mathrm{M}^{+}\right]$190.1158, found 190.1192.


3-68
((3-Fluoro-3-propylcyclobutylidene)methyl)benzene (3-68). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.029 \mathrm{~g}, 0.38 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.21 \mathrm{~mL}, 1.5 \mathrm{mmol}, 4.0$ eq), pyridine • $9 \mathrm{HF}(0.68 \mathrm{~mL}, 26.2 \mathrm{mmol} \mathrm{HF}, 68.9 \mathrm{eq} \mathrm{HF})$, and alcohol 3-53 ( $0.076 \mathrm{~g}, 0.38 \mathrm{mmol}$,
$1.0 \mathrm{eq})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc}$ in hexanes) afforded methylenecyclobutane $\mathbf{3 - 6 8}(0.046 \mathrm{~g}, 0.23 \mathrm{mmol}, 60 \%)$ as a colorless oil: IR (ATR) 2959, 1598, 1449, 1254, 1128, 1028, 912, 735, $692 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.22(\mathrm{app} \mathrm{t}, 2$ $\mathrm{H}, J=7.6 \mathrm{~Hz}), 7.14-7.06(\mathrm{~m}, 3 \mathrm{H}), 6.20(\mathrm{~s}, 1 \mathrm{H}), 3.31\left(\mathrm{ddt}, 1 \mathrm{H},{ }^{3} J_{\mathrm{FH}}=20.0 \mathrm{~Hz}, J=16.8 \mathrm{~Hz}, 2.8\right.$ $\mathrm{Hz}), 3.18\left(\mathrm{ddt}, 1 \mathrm{H},{ }^{3} \mathrm{JH}_{\mathrm{FH}}=20.0 \mathrm{~Hz}, J=17.2 \mathrm{~Hz}, 2.8 \mathrm{~Hz}\right), 3.14-3.02(\mathrm{~m}, 1 \mathrm{H}), 2.91-2.86(\mathrm{~m}, 1 \mathrm{H})$, $1.81-1.68(\mathrm{~m}, 2 \mathrm{H}), 1.47$ (pent, $2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 0.95(\mathrm{t}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 137.6,132.7,132.6,128.5,127.2,126.4,123.6,95.1\left({ }^{1} J_{\mathrm{CF}}=208 \mathrm{~Hz}\right), 44.5\left({ }^{2} J_{\mathrm{CF}}=41\right.$ $\mathrm{Hz}), 44.2\left({ }^{2} J_{\mathrm{CF}}=41 \mathrm{~Hz}\right), 39.4\left({ }^{2} J_{\mathrm{CF}}=22 \mathrm{~Hz}\right), 16.8\left({ }^{3} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right), 14.3 ;{ }^{19} \mathrm{~F}$ NMR $(376 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta-140.6$; HRMS (ASAP) $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{18} \mathrm{~F}[\mathrm{M}+\mathrm{H}]$ 205.1393, found 205.1409.

((3-Fluoro-3-isopropylcyclobutylidene)methyl)benzene (3-69). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.060 \mathrm{~g}, 0.77 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.43 \mathrm{~mL}, 3.1 \mathrm{mmol}, 4.0$ eq), pyridine • 9 HF ( $1.4 \mathrm{~mL}, 53.9 \mathrm{mmol} \mathrm{HF}, 70 \mathrm{eq} \mathrm{HF}$ ), and alcohol 3-54 ( $0.155 \mathrm{~g}, 0.77 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc}$ in hexanes) afforded methylenecyclobutane 3-69 ( $0.073 \mathrm{~g}, 0.36 \mathrm{mmol}, 47 \%$ ) as a colorless oil: IR (ATR) 2966, 1599, 1471, 1448, 1240, 1113, 1017, 913, $693 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.38-7.29(\mathrm{~m}, 2 \mathrm{H})$, $7.25(\mathrm{~m}, 3 \mathrm{H}), 6.29(\mathrm{t}, 1 \mathrm{H}, J=2.4 \mathrm{~Hz}), 3.35-2.94(\mathrm{~m}, 4 \mathrm{H}), 1.92\left(\mathrm{~d}\right.$ heptet, $1 \mathrm{H},{ }^{3} J_{\mathrm{FH}}=25 \mathrm{~Hz}, J=$ $6.8 \mathrm{~Hz}), 0.99(\mathrm{~d}, 6 \mathrm{H}, J=6.4 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 137.6,132.88,132.75,128.5$, 127.2, 126.4, $123.4\left({ }^{3} J_{\mathrm{CF}}=6 \mathrm{~Hz}\right), 98.0\left({ }^{1} J_{\mathrm{CF}}=210 \mathrm{~Hz}\right), 42.7\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 42.4\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right)$, $34.9\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=22 \mathrm{~Hz}\right), 15.73\left({ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 15.71\left({ }^{3} J_{\mathrm{CF}}=5 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-$
152.6; GC-MS (ESI) ${ }^{+} m / z 204,161,129,115,91,63,51$; HRMS (ASAP) $m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{17} \mathrm{~F}$ $\left[\mathrm{M}^{+}\right]$204.1314, found 204.1304.

(2-(3-Benzylidene-1-fluorocyclobutyl)ethyl)benzene (3-70). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.058 \mathrm{~g}, 0.75 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.42 \mathrm{~mL}, 3.0 \mathrm{mmol}, 4.0$ eq), pyridine • $9 \mathrm{HF}(1.4 \mathrm{~mL}, 52 \mathrm{mmol} \mathrm{HF}, 69 \mathrm{eq} \mathrm{HF})$, and alcohol 3-55 ( $0.198 \mathrm{~g}, 0.75 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc}$ in hexanes) afforded methylenecyclobutane 3-70 ( $0.160 \mathrm{~g}, 0.601 \mathrm{mmol}, 48 \%$ ) as a colorless oil: IR (ATR) 2915, 1599, 1496, 1453, 1305, 1237, 1059, 913, 738, $692 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.38-7.27(\mathrm{~m}, 4$ H), 7.25-7.17 (m, 6 H$), 6.33(\mathrm{t}, 1 \mathrm{H}, J=2.0 \mathrm{~Hz}), 3.39\left(\mathrm{ddt}, 1 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=19.6 \mathrm{~Hz}, J=16.8 \mathrm{~Hz}, 3.2\right.$ $\mathrm{Hz}), 3.25\left(\mathrm{ddt}, 1 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=20.0 \mathrm{~Hz}, J=16.8 \mathrm{~Hz}, 2.8 \mathrm{~Hz}\right), 3.45-3.32(\mathrm{~m}, 1 \mathrm{H}), 3.02-2.91(\mathrm{~m}, 1 \mathrm{H})$, 2.79, 2.13 (AA'BB', $4 \mathrm{H}, J=5.2 \mathrm{~Hz}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 141.6,137.5,132.2,128.51$, $128.49,128.4,127.2,126.0,124.0,123.9,94.8\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=210 \mathrm{~Hz}\right), 44.5\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=37 \mathrm{~Hz}\right), 44.2(\mathrm{~d}$, $\left.{ }^{2} J_{\mathrm{CF}}=37 \mathrm{~Hz}\right), 39.2\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 29.8\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-141.7 ;$ HRMS (ES ${ }^{+}$) $m / z$ calcd for $\mathrm{C}_{19} \mathrm{H}_{19} \mathrm{~F}\left[\mathrm{M}^{+}\right]$266.1471, found 266.1465.


3-71
(1-Fluoro-3-pentylidenecyclobutyl)benzene (3-71). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.040 \mathrm{~g}, 0.51 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.28 \mathrm{~mL}, 2.0 \mathrm{mmol}, 4.0 \mathrm{eq}$ ), pyridine - $9 \mathrm{HF}(0.46 \mathrm{~mL}, 18 \mathrm{mmol} \mathrm{HF}, 34.7 \mathrm{eq} \mathrm{HF})$, and alcohol 3-56 ( $0.110 \mathrm{~g}, 0.509 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2$ $\mathrm{mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ (5\% EtOAc in hexanes) afforded
methylenecyclobutane 3-71 ( $0.034 \mathrm{~g}, 0.16 \mathrm{mmol}, 31 \%$ ) as a colorless oil: IR (ATR) 2957, 2924, 1449, 1301, 1229, 1065, 755, $697 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.50-7.45(\mathrm{~m}, 2 \mathrm{H}), 7.39$ (app t, $2 \mathrm{H}, J=7.6$ ), 7.35-7.28(m, 1 H$), 5.44-5.36(\mathrm{~m}, 1 \mathrm{H}), 3.41-3.23(\mathrm{~m}, 2 \mathrm{H}), 3.22-3.10(\mathrm{~m}, 2$ H), $1.97(\mathrm{q}, 2 \mathrm{H}, J=7.2 \mathrm{~Hz}), 1.41-1.26(\mathrm{~m}, 4 \mathrm{H}), 0.92(\mathrm{t}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 142.8\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=22 \mathrm{~Hz}\right), 128.4,127.71,127.69,125.6,124.3\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=6 \mathrm{~Hz}\right), 124.0(\mathrm{~d}$, $\left.{ }^{3} J_{\mathrm{CF}}=8 \mathrm{~Hz}\right), 94.2\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=207 \mathrm{~Hz}\right), 45.5\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 44.0\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 31.7,28.5$, 22.3, 14.0; ${ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-138.2$; HRMS (ASAP) $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{~F}[\mathrm{M}-\mathrm{H}]$ 217.1414, found 217.1393.


1-(3-Benzylidene-1-fluorocyclobutyl)-4-(trifluoromethyl)benzene (3-72). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.029 \mathrm{~g}, 0.37 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.21 \mathrm{~mL}, 1.5$ mmol, 4.0 eq ), pyridine • $9 \mathrm{HF}(0.68 \mathrm{~mL}, 26 \mathrm{mmol} \mathrm{HF}, 71 \mathrm{eq} \mathrm{HF})$, and alcohol 3-59 ( 0.114 g , 0.375 mmol , 1.00 eq$)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc}$ in hexanes) afforded methylenecyclobutane 3-72 ( $0.071 \mathrm{~g}, 0.023 \mathrm{mmol}, 62 \%$ ) as a colorless oil: IR (ATR) 2967, 1622, 1324, 1165, 1122, 1069, 841, $693 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.62$ $\left(\mathrm{AB} \mathrm{q}, 4 \mathrm{H}, J_{\mathrm{AB}}=8.8 \mathrm{~Hz}, \partial_{\mathrm{AB}}=0.04 \mathrm{ppm}\right), 7.37(\mathrm{app} \mathrm{t}, 2 \mathrm{H}, J=6.8 \mathrm{~Hz}), 7.30-7.21(\mathrm{~m}, 3 \mathrm{H}), 6.45$ $(\mathrm{t}, 1 \mathrm{H}, J=2.4 \mathrm{~Hz}), 3.79\left(\mathrm{ddt}, 1 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=20.8 \mathrm{~Hz}, J=17.2 \mathrm{~Hz}, 3.2 \mathrm{~Hz}\right), 3.65\left(\mathrm{ddt}, 1 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=\right.$ $21.6 \mathrm{~Hz}, J=17.2 \mathrm{~Hz}, 2.8 \mathrm{~Hz}), 3.58-3.46(\mathrm{~m}, 1 \mathrm{H}), 3.41-3.30(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 146.4,\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 137.1,130.6\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=14 \mathrm{~Hz}\right), 130.1\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 128.6,127.3$, $126.8,125.5\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 124.4\left(\mathrm{~d},{ }^{4} J_{\mathrm{CF}}=6 \mathrm{~Hz}\right), 124.3,124.2\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=9 \mathrm{~Hz}\right), 124.1\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}\right.$ $=271 \mathrm{~Hz}), 94.0\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=211 \mathrm{~Hz}\right), 47.5\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 47.2\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $(376$
$\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-62.6,-140.5 ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 287$ (100), 201 (20), 171 (45), 140 (60); HRMS $\left(\mathrm{ESI}^{+}\right) m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{~F}_{3}[\mathrm{M}-\mathrm{F}]$ 287.1040, found 287.1048.


1-(3-Benzylidene-1-fluorocyclobutyl)-4-chlorobenzene (3-73). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.053 \mathrm{~g}, 0.68 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.38 \mathrm{~mL}, 2.7 \mathrm{mmol}, 4.0$ eq), pyridine • $9 \mathrm{HF}(1.2 \mathrm{~mL}, 46 \mathrm{mmol} \mathrm{HF}, 70 \mathrm{eq} \mathrm{HF})$, and alcohol 3-60 ( $0.183 \mathrm{~g}, 0.676 \mathrm{mmol}$, 1.00 eq). Purification by chromatography on $\mathrm{SiO}_{2}$ ( $2.5 \% \mathrm{EtOAc}$ in hexanes) afforded methylenecyclobutane 3-73 ( $0.076 \mathrm{~g}, 0.28 \mathrm{mmol}, 41 \%$ ) as a yellow-tinted oil: IR (ATR) 2961, 2913, 1599, 1493, 1398, 1301, 1242, 1092, 1010, 826, $692 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.41(\operatorname{app} \mathrm{~d}, 2 \mathrm{H}, J=8.8 \mathrm{~Hz}), 7.38-7.31(\mathrm{~m}, 4 \mathrm{H}), 7.25-7.20(\mathrm{~m}, 3 \mathrm{H}), 6.41(\mathrm{t}, 1 \mathrm{H}, J=2.0 \mathrm{~Hz})$, $3.74\left(\mathrm{ddt}, 1 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=20.8 \mathrm{~Hz}, J=17.6 \mathrm{~Hz}, 3.2 \mathrm{~Hz}\right), 3.67-3.43(\mathrm{~m}, 2 \mathrm{H}), 3.33(\mathrm{dddd}, 1 \mathrm{H}, J=16.8$ $\mathrm{Hz}, 13.2 \mathrm{~Hz}, 5.2 \mathrm{~Hz}, 3.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 140.8\left({ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 137.2,133.8$ $\left({ }^{4} J_{\mathrm{CF}}=1 \mathrm{~Hz}\right), 131.1,131.0,128.61,128.59,127.3,126.7,125.5\left({ }^{3} J_{\mathrm{CF}}=8 \mathrm{~Hz}\right), 124.2\left({ }^{3} J_{\mathrm{CF}}=5 \mathrm{~Hz}\right)$, $94.2\left({ }^{1} J_{\mathrm{CF}}=209 \mathrm{~Hz}\right), 47.3\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 47.0\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-$ 137.8; GC-MS (EI ${ }^{+}$) 272, 752, 217, 202, 116. HRMS (ASAP) $m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{14} \mathrm{FCl}[\mathrm{M}]$ 272.0768, found 272.0361.


1-(1-Fluoro-3-(pentan-3-ylidene)cyclobutyl)-4-(trifluoromethyl)benzene (3-74). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.041 \mathrm{~g}, 0.52 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine
( $0.30 \mathrm{~mL}, 2.1 \mathrm{mmol}, 4.0 \mathrm{eq}$ ), pyridine $\cdot 9 \mathrm{HF}(0.94 \mathrm{~mL}, 36 \mathrm{mmol} \mathrm{HF}, 70 \mathrm{eq} \mathrm{HF})$, and alcohol 3$61(0.149 \mathrm{~g}, 0.524 \mathrm{mmol}, 1.00 \mathrm{eq})$. Purification by chromatography on $\mathrm{SiO}_{2}(5 \% \mathrm{EtOAc}$ in hexanes) afforded methylenecyclobutane 3-74 ( $0.077 \mathrm{~g}, 0.27 \mathrm{mmol}, 51 \%$ ) as a colorless oil: IR (ATR) $\mathrm{cm}^{-1} 2966,1622,1462,1408,1325,1165,1125,1069,843 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.63,7.56\left(\mathrm{AA}^{\prime} \mathrm{BB}^{\prime}, 4 \mathrm{H}, J_{\mathrm{AB}}=8.4 \mathrm{~Hz}, \partial_{\mathrm{AB}}=0.07 \mathrm{ppm}\right), 3.39-3.25(\mathrm{~m}, 2 \mathrm{H}), 3.11(\mathrm{ddd}$, $2 \mathrm{H}, J=14.4 \mathrm{~Hz}, 14.4 \mathrm{~Hz}, 3.6 \mathrm{~Hz}), 2.02(\mathrm{q}, 4 \mathrm{H}, J=7.6 \mathrm{~Hz}), 1.00(\mathrm{t}, 6 \mathrm{H}, J=7.6 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 147.4\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 137.8\left(\mathrm{~d},{ }^{4} J_{\mathrm{CF}}=6 \mathrm{~Hz}\right), 129.8\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=31 \mathrm{~Hz}\right), 125.3$ $\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 124.1\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 124.1\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=270 \mathrm{~Hz}\right), 118.9\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=13 \mathrm{~Hz}\right), 93.2(\mathrm{~d}$, $\left.{ }^{1} J_{\mathrm{CF}}=209 \mathrm{~Hz}\right), 43.5\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 23.5,12.6 ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-62.6,-142.5 ;$ MS (ASAP) $m / z 267$ (100), 237 (75), 225 (75), 211 (70); HRMS (ASAP) $m / z$ calcd for $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{~F}_{3}$ [M-HF] 266.1282, found 266.1298.


1-Chloro-4-(1-fluoro-3-(pentan-3-ylidene)cyclobutyl)benzene (3-75). Prepared according to General Procedure H using $\mathrm{KHF}_{2}(0.063 \mathrm{~g}, 0.80 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.45 \mathrm{~mL}, 3.2$ mmol, 4.0 eq ), pyridine • $9 \mathrm{HF}(1.4 \mathrm{~mL}$, $54 \mathrm{mmol} \mathrm{HF}, 67 \mathrm{eq} \mathrm{HF})$, and alcohol 3-62 ( $0.201 \mathrm{~g}, 0.802$ mmol, 1.00 eq). Purification by chromatography on $\mathrm{SiO}_{2}$ ( $2.5 \% \mathrm{EtOAc}$ in hexanes) afforded methylenecyclobutane 3-75 ( $0.097 \mathrm{~g}, 0.38 \mathrm{mmol}, 48 \%$ ) as a colorless oil: IR (ATR) 2963, 1491, 1300, 1093, 1013, $829 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.40-7.31(\mathrm{~m}, 4 \mathrm{H}), 3.35-3.20(\mathrm{~m}, 2$ H), 3.14-3.02 (m, 2 H$), 2.00(\mathrm{q}, 4 \mathrm{H}, J=7.6 \mathrm{~Hz}), 1.00(\mathrm{t}, 6 \mathrm{H}, J=7.6 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 141.8\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 137.5\left({ }^{4} J_{\mathrm{CF}}=6 \mathrm{~Hz}\right), 133.4,128.5,125.4\left({ }^{3} J_{\mathrm{CF}}=9 \mathrm{~Hz}\right), 119.2\left({ }^{3} J_{\mathrm{CF}}\right.$ $=13 \mathrm{~Hz}), 93.4\left({ }^{1} J_{\mathrm{CF}}=208 \mathrm{~Hz}\right), 44.5\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 23.5,12.6 ;{ }^{19} \mathrm{~F} \mathrm{NMR}\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-$
139.7; MS (ASAP) $m / z 233$ (80), 232 (55), 217 (30), 209 (20); HRMS (ASAP) $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{Cl}[\mathrm{M}-\mathrm{F}]$ 233.1097, found 233.1089.


6-Fluoro-2,2-dimethyl-4-methylene-6-phenylhexan-3-ol (3-76). Prepared according General Procedure H from $\mathrm{KHF}_{2}(0.044 \mathrm{~g}, 0.56 \mathrm{mmol}, 1.2 \mathrm{eq})$, diisopropylamine ( $0.32 \mathrm{~mL}, 2.3 \mathrm{mmol}, 5.0$ eq), pyridine • $9 \mathrm{HF}(0.76 \mathrm{~mL}, 29 \mathrm{mmol} \mathrm{HF}, 65 \mathrm{eq} \mathrm{HF})$, and alcohol $3-58(0.098 \mathrm{~g}, 0.45 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(2.5 \% \mathrm{EtOAc} /$ hexanes $)$ afforded allylic alcohol 3-77 ( $0.018 \mathrm{~g}, 0.076 \mathrm{mmol}, 17 \%$ ) as a colorless oil: IR (ATR) $3598,2955,1480,1455$, 1362, 1059, 1004, $698 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.41-7.31(\mathrm{~m}, 5 \mathrm{H}), 5.57$ (ddd, 1 H , $\left.{ }^{2} J_{\mathrm{HF}}=48.0 \mathrm{~Hz}, J=8.4 \mathrm{~Hz}, 4.0 \mathrm{~Hz}\right), 5.20(\mathrm{~s}, 1 \mathrm{H}), 5.18(\mathrm{~s}, 1 \mathrm{H}), 3.75(\mathrm{~s}, 1 \mathrm{H}), 2.87(\mathrm{td}, 1 \mathrm{H}, J=$ $15.2 \mathrm{~Hz}, 9.2 \mathrm{~Hz}), 2.48\left(\mathrm{ddd}, 1 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=32.4 \mathrm{~Hz}, J=14.8 \mathrm{~Hz}, 4.0 \mathrm{~Hz}\right), 1.64(\mathrm{~s}, 1 \mathrm{H}), 0.92(\mathrm{~s}, 9 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 145.8\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=2 \mathrm{~Hz}\right), 139.9\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=20 \mathrm{~Hz}\right), 125.7\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=11\right.$ $\mathrm{Hz}), 125.6\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=10 \mathrm{~Hz}\right), 116.0,94.8\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=171 \mathrm{~Hz}\right), 82.3,41.2\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 35.5$, 26.3; ${ }^{19}$ F NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$-172.5; HRMS (ESI ${ }^{+}$) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{15} \mathrm{H}_{19}\left[\mathrm{M}-\mathrm{H}_{2} \mathrm{O}-\mathrm{F}\right]$ 199.1487, found 199.1482.


6-Fluoro-2,2-dimethyl-4-methylene-6-(4-(trifluoromethyl)phenyl)hexan-3-ol (3-77). Prepared according to General Procedure H from $\mathrm{KHF}_{2}(0.010 \mathrm{~g}, 0.13 \mathrm{mmol}, 1.0 \mathrm{eq})$, diisopropylamine ( $0.07 \mathrm{~mL}, 0.51 \mathrm{mmol}, 4.0 \mathrm{eq}$ ), pyridine • $9 \mathrm{HF}(0.12 \mathrm{~mL}, 4.6 \mathrm{mmol} \mathrm{HF}, 36 \mathrm{eq} \mathrm{HF})$, and alcohol 3$57(0.039 \mathrm{~g}, 0.13 \mathrm{mmol})$ in dichloromethane $(1 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$
( $2.5 \% \mathrm{EtOAc} /$ hexanes ) afforded 3-77 ( $0.014 \mathrm{~g}, 0.046 \mathrm{mmol}, 34 \%$ ) as a colorless oil: IR (ATR) 3598, 2957, 1623, 1325, 1166, 1127, 1069, 1007, $841 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.64$ $(\mathrm{d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.47(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 5.66\left(\mathrm{ddd}, 1 \mathrm{H},{ }^{2} J_{\mathrm{HF}}=47.6 \mathrm{~Hz}, J=8.8 \mathrm{~Hz}, 4.4 \mathrm{~Hz}\right)$, $5.17(\mathrm{~s}, 1 \mathrm{H}), 5.15(\mathrm{~s}, 1 \mathrm{H}), 3.80(\mathrm{~s}, 1 \mathrm{H}), 2.85(\mathrm{td}, 1 \mathrm{H}, J=15.2 \mathrm{~Hz}, 8.8 \mathrm{~Hz}), 2.48\left(\mathrm{ddd}, 1 \mathrm{H},{ }^{3} J_{\mathrm{HF}}\right.$ $=32.0 \mathrm{~Hz}, J=15.2 \mathrm{~Hz}, 4.4 \mathrm{~Hz}), 1.65(\mathrm{~s}, 1 \mathrm{H}), 0.92(\mathrm{~s}, 9 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 145.0$ $\left(\mathrm{d},{ }^{3} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right), 143.9\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=20 \mathrm{~Hz}\right), 130.9\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 125.9\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=7 \mathrm{~Hz}\right), 125.5(\mathrm{q}$, $\left.{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 123.7\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=270 \mathrm{~Hz}\right), 116.3,93.8\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=173 \mathrm{~Hz}\right), 82.4,41.1\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right)$, 35.6, 26.3; ${ }^{19} \mathrm{~F} \mathrm{NMR}\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-176.1,-62.6 ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 267$ (100), 173 (10), 137 (35); HRMS (ESI $\left.{ }^{+}\right) m / z$ calcd for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~F}_{3}\left[\mathrm{M}-\mathrm{H}_{2} \mathrm{O}-\mathrm{F}\right]$ 267.1361, found 267.1353.


1-Phenylcyclobutane-1-carbaldehyde (3-79). Prepared according to General Procedure H from $\mathrm{KHF}_{2}(0.100 \mathrm{~g}, 1.29 \mathrm{mmol})$, diisopropylamine $(0.73 \mathrm{~mL}, 5.1 \mathrm{mmol}, 4.0 \mathrm{eq})$, alcohol 3-78 ${ }^{73 \mathrm{~b}}(0.206$ $\mathrm{g}, 1.29 \mathrm{mmol})$, and pyridine $\cdot 9 \mathrm{HF}(0.93 \mathrm{~mL}, 35.8 \mathrm{mmol} \mathrm{HF}, 28 \mathrm{eq} \mathrm{HF})$ in dichloromethane ( 6 $\mathrm{mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(2.5 \% \mathrm{EtOAc} /$ hexanes $)$ afforded aldehyde 3-79 $(0.028 \mathrm{~g}, 0.17 \mathrm{mmol}, 14 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 9.46(\mathrm{~s}, 1 \mathrm{H}), 7.27(\mathrm{t}$, $2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.17(\mathrm{t}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.05(\mathrm{dd}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 2.68-2.59(\mathrm{~m}, 2 \mathrm{H})$, 2.31 (ddd, $2 \mathrm{H}, J=19.2 \mathrm{~Hz}, 9.6 \mathrm{~Hz}, 2.4 \mathrm{~Hz}$ ), 1.97-1.76 (m, 2 H ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 199.5,141.0,128.8,128.5,126.4,57.6,28.4,15.9$.


1-(Fluoro(phenyl)methyl)-3-(phenylsulfonyl)bicyclo[1.1.0]butane (3-81). A solution of alcohol 3-80 ${ }^{83 \mathrm{~b}}$ ( $\left.0.049 \mathrm{~g}, 0.16 \mathrm{mmol}\right)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$ was treated with Deoxofluor ${ }^{\circledR}(0.03 \mathrm{~mL}$, 0.16 mmol ) at $0{ }^{\circ} \mathrm{C}$. The reaction was stirred at $0{ }^{\circ} \mathrm{C}$ for 2 h , then it was quenched with sat aq $\mathrm{NaHCO}_{3}(5 \mathrm{~mL})$. The mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$, and the combined extracts were washed with water $(30 \mathrm{~mL})$ and brine $(30 \mathrm{~mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Chromatography on $\mathrm{SiO}_{2}$ ( $25 \% \mathrm{EtOAc} /$ hexanes) afforded bicyclobutane 3-81 ( $0.027 \mathrm{~g}, 0.089$ $\mathrm{mmol}, 55 \%$ ) as a colorless oil: $\mathrm{IR}(\mathrm{ATR}) 3065,2958,1447,1317,1307,1146,1084,945,724 \mathrm{~cm}^{-}$ ${ }^{1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.98-7.91(\mathrm{~m}, 2 \mathrm{H}), 7.63(\mathrm{tt}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 7.60-7.49$ $(\mathrm{m}, 4 \mathrm{H}), 7.48-7.37(\mathrm{~m}, 3 \mathrm{H}), 6.15\left(\mathrm{~d}, 1 \mathrm{H},{ }^{2} J_{\mathrm{HF}}=46.8 \mathrm{~Hz}\right), 2.77(\mathrm{~d}, 1 \mathrm{H}, J=6.4 \mathrm{~Hz}), 2.36(\mathrm{dd}, 1$ $\mathrm{H}, J=6.4 \mathrm{~Hz}, 0.8 \mathrm{~Hz}), 1.50(\mathrm{t}, 1 \mathrm{H}, J=0.8 \mathrm{~Hz}), 1.43(\mathrm{~d}, 1 \mathrm{H}, J=4.8 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 142.0,137.4\left({ }^{2} J_{\mathrm{CF}}=22 \mathrm{~Hz}\right), 133.3,129.3\left({ }^{4} J_{\mathrm{CF}}=2 \mathrm{~Hz}\right), 129.2,128.6,127.3,127.0\left({ }^{3} J_{\mathrm{CF}}\right.$ $=5 \mathrm{~Hz}), 89.7\left({ }^{1} J_{\mathrm{CF}}=174 \mathrm{~Hz}\right), 36.9,36.1,30.7\left({ }^{2} J_{\mathrm{CF}}=39 \mathrm{~Hz}\right), 28.9 ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta-163.5 ;$ HRMS $\left(\mathrm{ES}^{+}\right) m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{SO}_{2} \mathrm{~F}\left[\mathrm{M}^{+}\right]$302.0777, found 302.0785.


3-83
(4-Fluoro-3-methylbuta-1,2-dien-1-yl)benzene (3-83). Prepared according to General Procedure H from $\mathrm{KHF}_{2}(0.060 \mathrm{~g}, 0.77 \mathrm{mmol})$, diisopropylamine $(0.44 \mathrm{~mL}, 3.1 \mathrm{mmol}, 4.0 \mathrm{eq})$, pyridine $\cdot 9$ HF ( $1.0 \mathrm{~mL}, 12 \mathrm{mmol}, 15 \mathrm{eq}$ ), and cyclopropene $\mathbf{3 - 8 2}{ }^{190}(0.124 \mathrm{~g}, 0.77 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ (hexanes) afforded fluorinated allene 3-83 ( $0.062 \mathrm{~g}, 0.38$ $\mathrm{mmol}, 49 \%$ ) as a colorless oil: IR (ATR) 2984, 2885, 1956, 1599, 1497, 1464, 1374, 978, $693 \mathrm{~cm}^{-}$ ${ }^{1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.32-7.07(\mathrm{~m}, 5 \mathrm{H}), 6.21(\mathrm{ddd}, 1 \mathrm{H}, J=4.8 \mathrm{~Hz}, 4.8 \mathrm{~Hz}, 2.8 \mathrm{~Hz})$, $4.84\left(\mathrm{dd}, 2 \mathrm{H},{ }^{2} J_{\mathrm{HF}}=47.6 \mathrm{~Hz}, J=2.0 \mathrm{~Hz}\right), 1.91(\mathrm{~d}, 3 \mathrm{H}, J=2.8 \mathrm{~Hz}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$
$\delta 204.1\left({ }^{3} J_{\mathrm{CF}}=10 \mathrm{~Hz}\right), 134.1,128.7,127.2,127.0,100.3\left({ }^{2} J_{\mathrm{CF}}=19 \mathrm{~Hz}\right), 95.2\left({ }^{4} J_{\mathrm{CF}}=2 \mathrm{~Hz}\right), 84.4$ $\left({ }^{1} J_{\mathrm{CF}}=170 \mathrm{~Hz}\right), 15.3 ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-213.0$.


3-Fluoro-3-phenethylcyclobutan-1-one (3-84). To a solution of 3-70 ( $0.025 \mathrm{~g}, 0.094 \mathrm{mmol}$ ) in $\mathrm{THF} / \mathrm{H}_{2} \mathrm{O}(10: 1,1 \mathrm{~mL}$ total) was added potassium osmate(VI) dihydrate $(0.3 \mathrm{mg}, \sim 1 \mu \mathrm{~mol}, 0.5$ $\mathrm{mol} \%$ ) and $\mathrm{NMO}(0.016 \mathrm{~g}, 0.12 \mathrm{mmol}, 1.3 \mathrm{eq})$. The reaction mixture was stirred at room temperature for 19 h , then it was directly placed on an $\mathrm{SiO}_{2}$ column and purified by chromatography (25-50\% EtOAc/hexanes) to afford the corresponding diol ( $0.024 \mathrm{~g}, 0.080 \mathrm{mmol}$, $85 \%$ ) as a colorless oil. The material is a $\sim 1: 1$ mixture of diastereomers: ${ }^{19} \mathrm{~F}$ NMR ( 376 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta-137.8,-143.9$.

A solution of this diol $(0.024 \mathrm{~g}, 0.080 \mathrm{mmol}, 1.0 \mathrm{eq})$ in THF $(0.5 \mathrm{~mL})$ and water $(0.5 \mathrm{~mL})$ was treated with sodium periodate $(0.019 \mathrm{~g}, 0.088 \mathrm{mmol}, 1.1 \mathrm{eq})$. The reaction was stirred at rt for 3 h . The reaction mixture was then directly placed on an $\mathrm{SiO}_{2}$ column and purified by chromatography ( $10 \%$ EtOAc/hexanes) to afford cyclobutanone 3-84 ( $0.009 \mathrm{~g}, 0.047 \mathrm{mmol}, 59 \%$ ) as a colorless film. The material contained a trace of benzaldehyde byproduct but was used without further purification: IR (ATR) 2918, 1792, $909,730 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 7.34-7.28 (m, 2 H), 7.25-7.19 (m, 3 H), 3.45-3.32 (m, $2 H$ ), 3.15-3.04 (m, 2 H), 2.87-2.81 (m, 2 H), 2.34-2.22 (m, 2 H$) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 202.9\left(\mathrm{~d},{ }^{3} \mathrm{~J}_{\mathrm{CF}}=12 \mathrm{~Hz}\right), 140.6,128.6$, $128.4,128.3,126.3,89.6\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=202 \mathrm{~Hz}\right), 57.9\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 39.2\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 30.4(\mathrm{~d}$, $\left.{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-151.7$.


3-Fluoro-3-phenethylcyclobutan-1-one oxime (3-85). A solution of ketone 3-84 (0.016 g, 0.08 $\mathrm{mmol})$ in $\mathrm{EtOH}(0.25 \mathrm{~mL})$ and water $(0.25 \mathrm{~mL})$ was treated with hydroxylamine hydrochloride $(0.007 \mathrm{~g}, 0.11 \mathrm{mmol})$ and sodium acetate $(0.016 \mathrm{~g}, 0.19 \mathrm{mmol})$. The reaction mixture was sealed in a microwave vial and was heated at $60^{\circ} \mathrm{C}$ for 4.5 h . The reaction mixture was then cooled to rt and purified by chromatography on $\mathrm{SiO}_{2}(30 \% \mathrm{EtOAc} /$ hexanes $)$ to afford oxime $\mathbf{3 - 8 5}(0.010 \mathrm{~g}$, $0.05 \mathrm{mmol}, 58 \%$ ) as a colorless oil: IR (ATR) 3262, 2925, 1466, 1710, 1455, 1238, 929, $700 \mathrm{~cm}^{-}$ ${ }^{1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.30(\mathrm{app} \mathrm{t}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.24-7.18(\mathrm{~m}, 3 \mathrm{H}), 3.28-2.92(\mathrm{~m}$, 4 H ), 2.77 (ddd, $2 \mathrm{H}, J=5.6 \mathrm{~Hz}, 5.6 \mathrm{~Hz}, 5.6 \mathrm{~Hz}$ ), 2.20-2.07 (m, 2 H ), 1.8-1.4 ( $\mathrm{br} \mathrm{s}, 1 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 140.9,128.6,128.4,126.2,92.4\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=206 \mathrm{~Hz}\right), 43.7\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=26 \mathrm{~Hz}\right)$, $42.7\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=26 \mathrm{~Hz}\right), 39.2\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 29.8\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta$-146.1; MS (ESI') $m / z 208$ (60), 188 (100), 170 (40), 129 (65); HRMS (ESI ${ }^{+}$) $m / z$ calcd for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{FNO}[\mathrm{M}+\mathrm{H}]$ 208.1138, found 208.1131.


3-Fluoro-3-phenethylcyclobutan-1-ol (3-86). A solution of cyclobutanone 3-84 (0.012 g, 0.062 mmol ) in THF ( 1 mL ) was cooled to $-78^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$ and treated with LAH (1 drop of 4 M solution in ether). The reaction was slowly warmed to rt over 4 h , then it was quenched with water ( 5 mL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \times 25 \mathrm{~mL})$. The organic layers were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Purification by chromatography on $\mathrm{SiO}_{2}$ afforded alcohol 3-86 ( $0.008 \mathrm{~g}, 0.041 \mathrm{mmol}, 66 \%$ ) as a colorless film. The material is a $\sim 6.5: 1$ mixture of diastereomers: IR (ATR) 3344, 2936, 1603,

1496, 1455, 1238, 1040, $699 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.33-7.24(\mathrm{~m}, 2 \mathrm{H}), 7.22-7.13$ $(\mathrm{m}, 3 \mathrm{H}), 3.64(\mathrm{q}, 1 \mathrm{H}, J=6.8 \mathrm{~Hz}), 2.72,1.94,1.88\left(\mathrm{AA}^{\prime} \mathrm{BB}^{\prime}, 4 \mathrm{H}, J_{\mathrm{AB}}=5.2 \mathrm{~Hz}\right), 2.62(\mathrm{dtd}, 2 \mathrm{H}$, $J=10.4 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 3.2 \mathrm{~Hz}), 2.31\left(\mathrm{dddd}, 2 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=23.2 \mathrm{~Hz}, J=10.0 \mathrm{~Hz}, 7.2 \mathrm{~Hz}, 3.2 \mathrm{~Hz}\right), 1.62$ $(\mathrm{s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 141.3,128.5,128.4,128.3,126.0,89.3\left({ }^{1} J_{\mathrm{CF}}=211 \mathrm{~Hz}\right)$, $58.9,44.5\left({ }^{2} J_{\mathrm{CF}}=20 \mathrm{~Hz}\right), 40.4\left({ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 29.6\left({ }^{3} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ -133.8.

( $\boldsymbol{E}$ )-4-((4-Methoxybenzyl)amino)hept-3-en-2-one (3-88). Ketone 3-87 was prepared by the dissolving the corresponding diol $(0.091 \mathrm{~g}, 0.38 \mathrm{mmol}, 1.0 \mathrm{eq})$ in THF ( 1.5 mL ) and water ( 0.5 $\mathrm{mL})$ and treating with sodium periodate $(0.090 \mathrm{~g}, 0.42 \mathrm{mmol}, 1.1 \mathrm{eq})$. The reaction was stirred at rt for 2 h , then was filtered through a 1 " pipette column of $\mathrm{SiO}_{2}$ that was then flushed with THF (1 mL ). The filtrate was directly used without concentration or further purification. The filtrate was treated with 4-methoxybenzylamine ( $0.13 \mathrm{~mL}, 0.95 \mathrm{mmol}, 2.5$ eq $)$, acetic acid ( $0.02 \mathrm{~mL}, 0.38 \mathrm{mmol}, 1.0 \mathrm{eq}$ ) and sodium triacetoxyborohydride $(0.283 \mathrm{~g}, 1.34 \mathrm{mmol}, 3.50 \mathrm{eq})$. The reaction was stirred at rt for 15 h , then was quenched with $1 \mathrm{M} \mathrm{NaOH}(5 \mathrm{~mL})$ and partitioned between water ( 30 mL ) and EtOAc ( 30 mL ). The combined extracts were washed with water (30 $\mathrm{mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Chromatography on $\mathrm{SiO}_{2}(20-30 \% \mathrm{EtOAc}$ in hexanes) afforded vinylogous amide $\mathbf{3 - 8 8}(0.032 \mathrm{~g}, 0.13 \mathrm{mmol}, 34 \%)$ as a colorless oil: IR (ATR) 2961, 1602, 1568, 1517, 1302, 1245, 1175, 1032, 811, $746 ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 11.17$ (br s, 1 H ), $7.16(\mathrm{~d}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}), 6.85(\mathrm{~d}, 2 \mathrm{H}, J=8.7 \mathrm{~Hz}), 5.02(\mathrm{~s}, 1 \mathrm{H}), 4.38(\mathrm{~d}, 2 \mathrm{H}, J=6.0$ Hz ), 3.78 (s, 3 H ), 2.17 (t, $2 \mathrm{H}, J=7.5 \mathrm{~Hz}$ ), $2.02(\mathrm{~s}, 3 \mathrm{H}), 1.54$ (pent, $2 \mathrm{H}, J=7.5 \mathrm{~Hz}$ ), 0.96 (t, 3
$\mathrm{H}, J=7.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 195.3,166.7,159.0,130.1,128.1,114.2,94.7,55.3$, 45.9, 33.9, 29.1, 21.4, 14.0; MS ( $\mathrm{ESI}^{+}$) $m / z 248$ (2), 121 (100); HRMS ( $\mathrm{ESI}^{+}$) $m / z$ calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{NO}_{2}[\mathrm{M}+\mathrm{H}]$ 248.1642, found 248.1645.

( $\boldsymbol{E}$ )-4-(Benzylamino)-6-phenylhex-3-en-2-one (3-89). A solution of crude ketone 3-84 (0.030 g, $0.023 \mathrm{mmol})$ in $\mathrm{MeOH}(0.5 \mathrm{~mL})$ was treated sequentially with benzylamine ( $0.02 \mathrm{~mL}, 0.17 \mathrm{mmol}$ ), $t$-butyl isonitrile ( $0.02 \mathrm{~mL}, 0.17 \mathrm{mmol}$ ), and propionic acid $(0.01 \mathrm{~mL}, 0.17 \mathrm{mmol})$. The reaction was stirred at rt for 16 h , then was directly placed on an $\mathrm{SiO}_{2}$ column and chromatographed ( $30 \%$ EtOAc/hexanes) to afford vinylogous amide 3-89 ( $0.011 \mathrm{~g}, 49 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 11.29(\mathrm{~s}, 1 \mathrm{H}), 7.48-7.20(\mathrm{~m}, 8 \mathrm{H}), 7.20-7.11(\mathrm{~m}, 2 \mathrm{H}), 5.13(\mathrm{~s}, 1 \mathrm{H}), 4.43(\mathrm{~d}, 2 \mathrm{H}$, $J=6.3 \mathrm{~Hz}), 2.83,2.53\left(\mathrm{AA}^{\prime} \mathrm{BB}^{\prime}, 4 \mathrm{H}, J_{\mathrm{AB}}=7.5 \mathrm{~Hz}, J_{\mathrm{A}^{\prime} \mathrm{B}^{\prime}}=5.7 \mathrm{~Hz}\right), 2.09(\mathrm{~s}, 3 \mathrm{H})$.


1-Bromo-3-fluoro-3-methylcyclobutyl)(4-(trifluoromethyl)phenyl)methanol (3-90). A solution of methylenecyclobutane $\mathbf{3 - 4 6}(0.025 \mathrm{~g}, 0.10 \mathrm{mmol})$ in $\mathrm{MeCN}(0.75 \mathrm{~mL})$ and water $(0.25$ $\mathrm{mL})$ was treated with NBS $(0.037 \mathrm{~g}, 0.20 \mathrm{mmol}, 2.0 \mathrm{eq})$. The reaction was stirred at $45^{\circ} \mathrm{C}$ for 18 h. After cooling, the material was directly chromatographed on $\mathrm{SiO}_{2}$ to afford bromohydrin 3-90 $(0.024 \mathrm{~g}, 0.070 \mathrm{mmol}, 69 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.63,7.55(\mathrm{ABq}, 4$ $\left.\mathrm{H}, J_{\mathrm{AB}}=8.4 \mathrm{~Hz}\right), 4.88(\mathrm{~s}, 1 \mathrm{H}), 3.13(\operatorname{appq}, 2 \mathrm{H}, J=16.8 \mathrm{~Hz}), 2.79(\mathrm{dddd}, 1 \mathrm{H}, J=14.2 \mathrm{~Hz}, 13.6$ $\mathrm{Hz}, 4.4 \mathrm{~Hz}, 1.2 \mathrm{~Hz}$ ), 2.65 (dddd, $1 \mathrm{H}, J=13.6 \mathrm{~Hz}, 13.2 \mathrm{~Hz}, 4.4 \mathrm{~Hz}, 1.2 \mathrm{~Hz}$ ), 2.50 (br s, $1 \mathrm{H}, 1.71$
$\left(\mathrm{d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=22.4 \mathrm{~Hz}\right) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 143.2,130.6\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 127.9$, $125.0\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 124.0\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=270 \mathrm{~Hz}\right), 91.6\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=202 \mathrm{~Hz}\right), 79.2,60.7\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=14\right.$ $\mathrm{Hz}), 50.2\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 48.7\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 26.4\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $(470 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta-62.6,-127.5$.


## 1-Bromo-3-fluoro-3-methylcyclobutyl)(4-(trifluoromethyl)phenyl)methanone (3-91). A

 solution of alcohol 3-90 $(0.015 \mathrm{~g}, 0.044 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ was cooled to $0{ }^{\circ} \mathrm{C}$ under $\mathrm{N}_{2}$. Dess-Martin periodinane $(0.028 \mathrm{~g}, 0.032 \mathrm{mmol}, 1.1 \mathrm{eq})$ was added in one portion. The reaction was stirred at rt for 2 h , then it was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ and sat aq $\mathrm{NaHCO}_{3}(10 \mathrm{~mL})$. The organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Purification by chromatography on $\mathrm{SiO}_{2}$ afforded ketone 3-91 ( $0.013 \mathrm{~g}, 0.048 \mathrm{mmol}, 87 \%$ ) as a colorless semi-solid: ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 8.11(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.74(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 3.45,2.94\left(\mathrm{AA}^{\prime} \mathrm{BB}^{\prime}, 4 \mathrm{H}^{3}{ }^{3} J_{\mathrm{HF}}=19.6\right.$ $\mathrm{Hz}, J=16.0 \mathrm{~Hz}, 12.0 \mathrm{~Hz}), 1.83\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=22.4 \mathrm{~Hz}\right) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 192.0,134.8\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 130.5,125.6\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 123.4\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=271 \mathrm{~Hz}\right)$, $91.1\left(\mathrm{~d},{ }^{1} J_{\mathrm{CF}}=204 \mathrm{~Hz}\right), 49.6\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 49.3\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=14 \mathrm{~Hz}\right), 26.4\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR (470 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta-63.3,-127.3$; HRMS (ASAP) $\mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{13} \mathrm{H}_{12} \mathrm{BrOF}_{4}[\mathrm{M}+\mathrm{H}]$ 339.0008 , found 339.0029 .

3-92

3-Methyl- N -phenylbicyclo[1.1.0]butane-1-carboxamide (3-92). ${ }^{73 \mathrm{~b}}$ Prepared according to General Procedure G from A solution of 1-chloromethyl-2,2-dibromocyclopropane (3.85 g, 14.7 $\mathrm{mmol}), \mathrm{MeLi}\left(9.8 \mathrm{~mL}, 14.7 \mathrm{mmol}, 1.5 \mathrm{M} \mathrm{in}_{\mathrm{Et}}^{2} \mathrm{O}\right), t-\mathrm{BuLi}(9.8 \mathrm{~mL}, 14.7 \mathrm{mmol}, 1.5 \mathrm{M}$ in pentane $)$, and phenyl isocyanate ( $0.70 \mathrm{~mL}, 7.38 \mathrm{mmol}$ ) in $\mathrm{Et}_{2} \mathrm{O}(25 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(15 \% \mathrm{EtOAc} /$ hexane $)$ afforded amide 3-92 $(0.538 \mathrm{~g}, 2.87 \mathrm{mmol}, 49 \%)$ as a colorless solid: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(\mathrm{CDCl}_{3}, 300 \mathrm{MHz}\right) \delta 7.53(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.44(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.30(\mathrm{t}, J=7.9 \mathrm{~Hz}, 2 \mathrm{H})$, $7.08(\mathrm{t}, J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.22(\mathrm{~s}, 2 \mathrm{H}), 1.59(\mathrm{~s}, 3 \mathrm{H}), 1.24(\mathrm{~s}, 2 \mathrm{H})$.


3-93
$N$-(4-Methoxyphenyl)-3-methylbicyclo[1.1.0]butane-1-carboxamide
(3-93). Prepared according to General Procedure G from 1-chloromethyl-2,2-dibromocyclopropane ( $3.50 \mathrm{~g}, 13.3$ $\mathrm{mmol}), \mathrm{MeLi}\left(8.3 \mathrm{~mL}, 13.3 \mathrm{mmol}, 1.6 \mathrm{M} \mathrm{in}^{\mathrm{Et}} \mathrm{O}\right), t-\mathrm{BuLi}(7.8 \mathrm{~mL}, 13.4 \mathrm{mmol}, 1.7 \mathrm{M}$ in pentane $)$, and 4-methoxyphenyl isocyanate ( $0.87 \mathrm{~mL}, 6.7 \mathrm{mmol}$ ) in $\mathrm{Et}_{2} \mathrm{O}(25 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ (30-50\% EtOAc/hexane) to afford amide 3-93 (0.961 g, $\left.4.42 \mathrm{mmol}, 66 \%\right)$ as a colorless solid: ${ }^{1} \mathrm{H}$ NMR $\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.40(\mathrm{~d}, 2 \mathrm{H}, J=9.0 \mathrm{~Hz}), 7.11(\mathrm{~s}, 1 \mathrm{H}), 6.85(\mathrm{~d}$, $2 \mathrm{H}, J=9.0 \mathrm{~Hz}), 3.78(\mathrm{~s}, 3 \mathrm{H}), 2.12(\mathrm{~s}, 2 \mathrm{H}), 1.58(\mathrm{~s}, 3 \mathrm{H}), 1.22(\mathrm{~s}, 2 \mathrm{H})$.


N-Allyl-3-methylbicyclo[1.1.0]butane-1-carboxamide (3-94). Prepared according to General Procedure G from 1-chloromethyl-2,2-dibromocyclopropane (1.00 g, 3.81 mmol ), MeLi ( 2.4 mL , $3.8 \mathrm{mmol}, 1.6 \mathrm{M}$ in $\mathrm{Et}_{2} \mathrm{O}$ ), $t$ - $\mathrm{BuLi}(2.2 \mathrm{~mL}, 13.4 \mathrm{mmol}, 1.7 \mathrm{M}$ in pentane $)$, and allyl isocyanate ( $0.17 \mathrm{~mL}, 1.9 \mathrm{mmol}$ ) in $\mathrm{Et}_{2} \mathrm{O}(8 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ ( $30-50 \%$ EtOAc/hexane) afforded amide 3-94 ( $0.129 \mathrm{~g}, 0.853 \mathrm{mmol}, 45 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR (300 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 6.03(\mathrm{~s}, 1 \mathrm{H}), 5.76,5.08,5.02\left(\mathrm{ABX}, 3 \mathrm{H}, J_{\mathrm{AB}}=17.2, J_{\mathrm{AX}}=10.4 \mathrm{~Hz}\right), 3.82(\mathrm{t}, 2 \mathrm{H}$, $J=5.6 \mathrm{~Hz}), 2.02(\mathrm{~s}, 2 \mathrm{H}), 1.43(\mathrm{~s}, 3 \mathrm{H}), 1.04(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 170.8,134.9$, 115.7, 42.0, 37.0, 23.7, 14.2, 12.2.

$N$-Phenyl-3-propylbicyclo[1.1.0]butane-1-carboxamide (3-95). Prepared according to General Procedure G from the corresponding tribromide ( $2.42 \mathrm{~g}, 7.23 \mathrm{mmol}$ ), MeLi ( $4.5 \mathrm{~mL}, 7.2 \mathrm{mmol}$, 1.6 M in $\mathrm{Et}_{2} \mathrm{O}$ ), $t-\mathrm{BuLi}(4.3 \mathrm{~mL}, 7.2 \mathrm{mmol}, 1.7 \mathrm{M}$ in pentane), and phenyl isocyanate ( 0.39 mL , $1.8 \mathrm{mmol})$. Purification by two iterations chromatography on $\mathrm{SiO}_{2}(30 \% \mathrm{EtOAc} /$ hexane $)$ afforded amide 3-95 ( $0.127 \mathrm{~g} 0.590 \mathrm{mmol}, 16 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.60-7.35$ (m, 3 H ), 7.30-7.15 (m, 2 H ), 7.11-7.02 (m, 1 H$), 2.21(\mathrm{~s}, 2 \mathrm{H}), 1.85(\mathrm{t}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 1.49$ (pent, $2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 1.18(\mathrm{~s}, 2 \mathrm{H}), 0.98(\mathrm{t}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz})$. The material was contaminated with $\sim 12 \%$ (by mass) of inseparable $N$-phenyl acetamide (MeLi addition product).


N, 3-Diphenylbicyclo[1.1.0]butane-1-carboxamide (3-96). Prepared according to General Procedure G from the corresponding tribromide ( $1.31 \mathrm{~g}, 3.55 \mathrm{mmol}$ ), MeLi ( $2.2 \mathrm{~mL}, 3.6 \mathrm{mmol}$, 1.6 M in $\mathrm{Et}_{2} \mathrm{O}$ ), $t$ - $\mathrm{BuLi}(2.1 \mathrm{~mL}, 3.6 \mathrm{mmol}, 1.7 \mathrm{M}$ in pentane), and phenyl isocyanate ( 0.15 mL , $1.4 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(8 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(15-40 \% \mathrm{EtOAc} /$ hexanes $)$ afforded amide $\mathbf{3 - 9 6}(0.176 \mathrm{~g}, 0.71 \mathrm{mmol}, 50 \%)$ as a tan crystalline solid: $\mathrm{Mp} 150-152{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR (300 MHz, $\left.\mathrm{CD}_{3} \mathrm{OD}\right)(\mathrm{N}-\mathrm{H}$ absent) $\delta$ 7.39-7.33 (m, 2 H ), 7.33-7.24 (m, 4 H ), 7.24-7.15 (m, $3 \mathrm{H}), 7.01(\mathrm{tt}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 3.07(\mathrm{~s}, 2 \mathrm{H}), 1.62(\mathrm{t}, 2 \mathrm{H}, J=1.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right) \delta 168.6,138.1,133.8,128.14,128.06,126.4,125.6,123.8,120.7,34.0,31.5,24.7 ;$ MS (ESI $\left.{ }^{+}\right) m / z 250(100), 227$ (5), 185 (5), 129 (5); HRMS ( $\mathrm{ESI}^{+}$) $m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{NO}[\mathrm{M}+\mathrm{H}]$ 250.1226, found 250.1225 .


General Procedure I (Bicyclo[1.1.0]butyl amide fluorination): 3-Fluoro-3-methyl- N -phenylcyclobutane-1-carboxamide (3-97). To a suspension of $\mathrm{KHF}_{2}(0.012 \mathrm{~g}, 0.15 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ in a polypropylene reaction vessel was added diisopropylamine $(0.08 \mathrm{~mL}, 0.60$ $\mathrm{mmol}, 4.0 \mathrm{eq})$. The suspension was cooled under $\mathrm{N}_{2}$ to $-78{ }^{\circ} \mathrm{C}$, then pyridine $\cdot 9 \mathrm{HF}(0.27 \mathrm{~mL}, 10$ mmol HF, 69 eq HF) was added. The mixture was stirred for 5 min at $-78^{\circ} \mathrm{C}$, then a solution of the amide 3-92 ${ }^{73 \mathrm{~b}}$ ( $0.028 \mathrm{~g}, 0.15 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(0.5 \mathrm{~mL})$ was slowly added. After stirring at $78{ }^{\circ} \mathrm{C}$ for 10 min , the mixture was warmed to $0^{\circ} \mathrm{C}$ for 10 min and then quenched with sat aq. KF ( 2 mL ), warmed to rt , and diluted with $2 \mathrm{M} \mathrm{NaOH}(50 \mathrm{~mL})$. The material was extracted with ether
( 2 X 20 mL ). The organic layers were washed with saturated aq. $\mathrm{NaHCO}_{3}(30 \mathrm{~mL})$ and brine (30 $\mathrm{mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Chromatography on $\mathrm{SiO}_{2}(25 \% \mathrm{EtOAc} /$ hexanes $)$ afforded amide $\mathbf{3 - 9 7}(0.018 \mathrm{~g}, 0.09 \mathrm{mmol}, 58 \%)$ as a colorless solid: Mp $159-161{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.51(\mathrm{~d}, 2 \mathrm{H}, J=6.4 \mathrm{~Hz}), 7.32(\mathrm{t}, 2 \mathrm{H}, J=6.4 \mathrm{~Hz}), 7.17(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.11(\mathrm{t}, 1$ $\mathrm{H}, J=7.5 \mathrm{~Hz}), 3.20(\mathrm{ddd}, 1 \mathrm{H}, J=16.0 \mathrm{~Hz}, 9.0 \mathrm{~Hz}, 7.0 \mathrm{~Hz}), 2.65-2.46(\mathrm{~m}, 4 \mathrm{H}), 1.53\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{HF}}\right.$ $=22.5 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 172.5,137.8,129.1,124.4,119.8,95.6\left({ }^{1} J_{\mathrm{CF}}=195\right.$ $\mathrm{Hz}), 37.8\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 32.8\left({ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 25.0\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(470 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-$ 133.2; MS (ESI $) m / z 208$ (100), 195 (10); HRMS ( $\mathrm{ESI}^{+}$) $m / z$ calcd for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{NOF}[\mathrm{M}+\mathrm{H}]$ 208.1138, found 208.1132 .


3-Fluoro- $N$-(4-methoxyphenyl)-3-methylcyclobutane-1-carboxamide
(3-98). Prepared according to General Procedure I from $\mathrm{KHF}_{2}(0.032 \mathrm{~g}, 0.41 \mathrm{mmol})$, diisopropylamine ( 0.23 mL , $1.6 \mathrm{mmol}, 4.0 \mathrm{eq})$, pyridine $\cdot 9 \mathrm{HF}(0.73 \mathrm{~mL}, 28 \mathrm{mmol} \mathrm{HF}, 69 \mathrm{eq} \mathrm{HF})$, and amide 3-93 ( 0.088 g , $0.41 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2.5 \mathrm{~mL})$. Chromatography on $\mathrm{SiO}_{2}(20-50 \% \mathrm{EtOAc} /$ hexanes $)$ afforded amide 3-98 ( $0.048 \mathrm{~g}, 0.20 \mathrm{mmol}, 50 \%$ ) as a colorless solid: $\mathrm{Mp} 104-10 \mathrm{~F}^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H} \mathrm{NMR}(500 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 7.39(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.35-7.20(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 6.83(\mathrm{~d}, 2 \mathrm{H}, J=8.5 \mathrm{~Hz}), 3.78(\mathrm{~s}, 3 \mathrm{H})$, $3.16(\mathrm{ddd}, 1 \mathrm{H}, J=16.0 \mathrm{~Hz}, 9.0 \mathrm{~Hz}, 7.0 \mathrm{~Hz}), 2.64-2.38(\mathrm{~m}, 4 \mathrm{H}), 1.51\left(\mathrm{~d}, 3 \mathrm{H},{ }^{3} J_{\mathrm{HF}}=22.5 \mathrm{~Hz}\right) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 172.5,156.5,130.9,121.8,114.1,95.8\left({ }^{1} J_{\mathrm{CF}}=195 \mathrm{~Hz}\right), 55.5,37.8$ $\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 32.6\left({ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 25.1\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(470 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-133.2 ; \mathrm{MS}$ $\left(\mathrm{ESI}^{+}\right) m / z 238$ (100), $200(5), 195(5), 184$ (5); $\mathrm{HRMS}\left(\mathrm{ESI}^{+}\right) m / z$ calcd for $\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{NO}_{2} \mathrm{~F}[\mathrm{M}+\mathrm{H}]$ 238.1243, found 238.1236. An X-ray crystal structure was obtained for this compound.


N-Allyl-3-fluoro-3-methylcyclobutane-1-carboxamide (3-99). Prepared according to General Procedure I from $\mathrm{KHF}_{2}(0.066 \mathrm{~g}, 0.85 \mathrm{mmol})$, diisopropylamine $(0.48 \mathrm{~mL}, 3.4 \mathrm{mmol}, 4.0 \mathrm{eq})$, pyridine• $9 \mathrm{HF}(1.5 \mathrm{~mL}, 58 \mathrm{mmol} \mathrm{HF}, 68 \mathrm{eq} \mathrm{HF})$, and the corresponding amide ( $0.129 \mathrm{~g}, 0.853$ mmol ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(20-50 \% \mathrm{EtOAc} /$ hexanes $)$ afforded amide 3-99 ( $0.063 \mathrm{~g}, 0.37 \mathrm{mmol}, 43 \%$ ) as a colorless semi-solid: ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 5.86-5.62(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 5.78,5.13,5.09\left(\mathrm{ABX}, 3 \mathrm{H}, J_{\mathrm{AX}}=18.4 \mathrm{~Hz}, J_{\mathrm{BX}}=10.0 \mathrm{~Hz}\right), 3.84$ (t, 2 H, $J=5.6 \mathrm{~Hz}$ ), $3.04(\mathrm{ddd}, 1 \mathrm{H}, J=16.0 \mathrm{~Hz}, 9.6 \mathrm{~Hz}, 6.8 \mathrm{~Hz}), 2.57-2.32(\mathrm{~m}, 4 \mathrm{H}), 1.47(\mathrm{~d}, 3 \mathrm{H}$, $\left.{ }^{3} J_{\mathrm{HF}}=22.4 \mathrm{~Hz}\right) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 174.1,134.2,116.4,95.8\left({ }^{1} J_{\mathrm{CF}}=195 \mathrm{~Hz}\right), 42.0$, $37.8\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 31.6\left({ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 25.0\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right) ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-133.3 ;$ MS (ESI ${ }^{+}$) m/z 172 (100), 152 (5), 115 (2); HRMS (ESI ${ }^{+}$) m/z calcd for $\mathrm{C}_{9} \mathrm{H}_{15} \mathrm{NOF}[\mathrm{M}+\mathrm{H}]$ 172.1138, found 172.1132 .


3-Fluoro- $N$-phenyl-3-propylcyclobutane-1-carboxamide (3-100). Prepared according to General Procedure I from $\mathrm{KHF}_{2}(0.046 \mathrm{~g}, 0.59 \mathrm{mmol})$, diisopropylamine ( $0.33 \mathrm{~mL}, 2.3 \mathrm{mmol}, 4.0$ eq), pyridine • $9 \mathrm{HF}(1.1 \mathrm{~mL}, 42 \mathrm{mmol}, 72 \mathrm{eq} \mathrm{HF})$, and amide $3-95(0.127 \mathrm{~g}, 0.59 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (3 mL). Chromatography on $\mathrm{SiO}_{2}(25 \% \mathrm{EtOAc} /$ hexanes $)$ afforded amide 3-100 ( $0.051 \mathrm{~g}, 0.22$ mmol, $37 \%$ ) as a colorless oil that partially solidified on standing. The minor diastereomer was isolated ( $0.013 \mathrm{~g}, 9 \%$ ) but was contaminated with an elimination/diene product. Major isomer: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.51(\mathrm{~d}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.32(\mathrm{t}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.25-7.14$ (br s, 1
H), $7.11(\mathrm{t}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.24$ (pent, $1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 2.54(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 2.48(\mathrm{~d}, 2 \mathrm{H}$, $J=8.4 \mathrm{~Hz}), 1.74\left(\mathrm{AA}^{\prime} \mathrm{BB}^{\prime}, 2 \mathrm{H}, J_{\mathrm{AB}}=8.0 \mathrm{~Hz}, \partial_{\mathrm{AB}}=0.05 \mathrm{ppm}\right), 1.47-1.35(\mathrm{~m}, 2 \mathrm{H}), 0.93(\mathrm{t}, 3 \mathrm{H}$, $J=7.6 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 172.6,137.8,129.1,124.4,119.8,97.9\left({ }^{1} J_{\mathrm{CF}}=196\right.$ $\mathrm{Hz}), 40.3\left({ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 36.4\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 33.5,16.1\left({ }^{3} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right), 14.1 ;{ }^{19} \mathrm{~F}$ NMR $(376 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta-144.5 ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) m / z 236$ (100), 216 (15); HRMS (ESI $\left.{ }^{+}\right) m / z$ calcd for $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{NOF}$ $[\mathrm{M}+\mathrm{H}] 236.1451$, found 236.1443.


3-Fluoro-N,3-diphenylcyclobutane-1-carboxamide (3-101). Prepared according to General Procedure I from $\mathrm{KHF}_{2}(0.013 \mathrm{~g}, 0.16 \mathrm{mmol})$, diisopropylamine $(0.09 \mathrm{~mL}, 0.64 \mathrm{mmol}, 4.0 \mathrm{eq})$, pyridine $\cdot 9 \mathrm{HF}(0.29 \mathrm{~mL}, 11 \mathrm{mmol} \mathrm{HF}, 70 \mathrm{eq} \mathrm{HF})$, and amide 3-96 ( $0.040 \mathrm{~g}, 0.16 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (1.5 mL). Chromatography on $\mathrm{SiO}_{2}$ (25\% EtOAc/hexanes) afforded cis 3-101 (17 mg, 0.063 mmol , 39\%) as a colorless oil and trans 3-101 ( $11 \mathrm{mg}, 0.041 \mathrm{mmol}, 25 \%$ ) as a colorless semi-solid. Major product cis 3-101: ${ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.53(\mathrm{~d}, 2 \mathrm{H}, J=7.5 \mathrm{~Hz}), 7.50(\mathrm{~d}, 2 \mathrm{H}, J=8.0$ $\mathrm{Hz}), 7.40(\mathrm{t}, 2 \mathrm{H}, J=7.5 \mathrm{~Hz}), 7.36-7.29(\mathrm{~m}, 3 \mathrm{H}), 7.24(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.12(\mathrm{t}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.46$ $\left(\mathrm{AB}_{2} X(\operatorname{app} \mathrm{q}), 1 \mathrm{H}, J=8.4 \mathrm{~Hz}\right), 3.12-2.93\left(\mathrm{~A} B_{2} \mathrm{X}(\mathrm{m}), 2 \mathrm{H}\right), 2.88-2.75\left(A \mathrm{~B}_{2} \mathrm{X}(\mathrm{m}), 2 \mathrm{H}\right) ;{ }^{19} \mathrm{~F}$ NMR (470 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta-146.0$. Minor product trans 3-101: ${ }^{1} \mathrm{H}$ NMR ( $500 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.54(\mathrm{~d}$, $2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.50(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.43(\mathrm{t}, 2 \mathrm{H}, J=7.5 \mathrm{~Hz}), 7.40-7.27(\mathrm{~m}, 4 \mathrm{H}), 7.12(\mathrm{t}, 1$ $\mathrm{H}, J=7.5 \mathrm{~Hz}$ ), 3.12-3.01 (AMX (m), 2 H ), 2.91-2.82 (AMX (m), 2 H ), 2.75 (AMX (app pent), 1 $\mathrm{H}, J=8.5 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.125 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 171.2,140.9\left({ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 137.8,129.1,128.6$, $128.5,124.8\left({ }^{3} J_{\mathrm{CF}}=6 \mathrm{~Hz}\right), 124.5,119.8,92.0\left({ }^{1} J_{\mathrm{CF}}=211 \mathrm{~Hz}\right), 38.7\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 31.1\left({ }^{3} J_{\mathrm{CF}}=\right.$
$13 \mathrm{~Hz}) ;{ }^{19} \mathrm{~F}$ NMR ( $470 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$-119.3. MS (ESI ${ }^{+}$m/z 250 (100), 225 (10), 184 (30), 128 (20); HRMS (ESI') $m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{NO}[\mathrm{M}+\mathrm{H}-\mathrm{HF}] 250.1226$, found 250.1225.


2-(Fluoromethyl)- $N$-phenylcyclopropane-1-carboxamide (3-102). Prepared according to General Procedure I from $\mathrm{KHF}_{2}(0.032 \mathrm{~g}, 0.40 \mathrm{mmol})$, diisopropylamine ( $0.23 \mathrm{~mL}, 1.6 \mathrm{mmol}, 4.0$ eq), pyridine • $9 \mathrm{HF}(0.72 \mathrm{~mL}, 28 \mathrm{mmol} \mathrm{HF}, 69 \mathrm{eq} \mathrm{HF})$, and amide $\mathbf{3 - 4 3}(0.070 \mathrm{~g}, 0.40 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \mathrm{~mL})$. Chromatography on $\mathrm{SiO}_{2}(20-50 \% \mathrm{EtOAc} /$ hexanes $)$ afforded a less polar minor product ( 10 mg ), which appears to be $\mathrm{a} \sim 1: 1$ mixture of a cyclopropane diastereomer and an unidentified product. The major product was identified as primary fluoride $\mathbf{3 - 1 0 2}(25 \mathrm{mg}, 0.13$ mmol, $32 \%$ ), a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.59(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.50(\mathrm{~d}, 2 \mathrm{H}, J=8.0$ $\mathrm{Hz}), 7.35-7.24(\mathrm{~m}, 3 \mathrm{H}), 7.10(\mathrm{t}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 4.54\left(\mathrm{ddd}, 1 \mathrm{H},{ }^{2} J_{\mathrm{HF}}=48.4 \mathrm{~Hz}, J=9.6 \mathrm{~Hz}, 5.6\right.$ $\mathrm{Hz}), 4.17\left(\mathrm{ddd}, 1 \mathrm{H},{ }^{2} J_{\mathrm{HF}}=48.4 \mathrm{~Hz}, J=8.4 \mathrm{~Hz}, 8.4 \mathrm{~Hz}\right), 1.97-1.84(\mathrm{~m}, 1 \mathrm{H}), 1.63-1.54(\mathrm{~m}, 1 \mathrm{H})$, 1.42-1.33(m, 1 H), 0.95-0.86(m, 1 H$) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 137.9,129.1,129.0,124.3$, $119.8,84.9\left({ }^{1} J_{\mathrm{CF}}=167 \mathrm{~Hz}\right), 21.0\left({ }^{2} J_{\mathrm{CF}}=24 \mathrm{~Hz}\right), 11.74,11.67 ;{ }^{19} \mathrm{~F} \mathrm{NMR}\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-$ 214.4.

(Z)-2-Phenylprop-2-en-3- $\boldsymbol{d}$-1-ol (3-104). A 500 mL 3-neck flask affixed with an internal thermometer, septum, and condenser with $\mathrm{N}_{2}$ inlet was charged with magnesium turnings ( 2.25 g , 92.8 mmol , 2.6 eq ) and THF ( 120 mL ). Bromobenzene ( $9.5 \mathrm{~mL}, 89 \mathrm{mmol}, 2.5 \mathrm{eq}$ ) was added in
$\sim 1 \mathrm{~mL}$ portions. The system was heated with a heat gun to initiate Grignard formation. Upon complete formation of the Grignard reagent, the reaction was stirred until it cooled to rt . In a separate 500 mL 3-neck flask affixed with a septum, internal thermometer, and $\mathrm{N}_{2}$ inlet. Under an $\mathrm{N}_{2}$ atmosphere, $\mathrm{CuI}(0.679 \mathrm{~g}, 3.57 \mathrm{mmol}, 0.100 \mathrm{eq})$, THF ( 80 mL ), and propargyl alcohol ( 2.1 mL , $36 \mathrm{mmol}, 1.0 \mathrm{eq})$ were added sequentially. The solution was cooled to $-78^{\circ} \mathrm{C}$. The Grignard solution was cannulated to this alcohol solution over a period of 1.5 h . After complete transfer, the cooling bath was removed and the reaction was warmed to rt and stirred for 17 h . The reaction was then cooled to $-78{ }^{\circ} \mathrm{C}$ and treated with $\mathrm{D}_{2} \mathrm{O}(6.5 \mathrm{~mL}, 0.36 \mathrm{~mol}, 10 \mathrm{eq})$. After warming to $\mathrm{rt}, 1 \mathrm{M}$ $\mathrm{HCl}(300 \mathrm{~mL})$ was added. The material was extracted with ether $(2 \times 100 \mathrm{~mL})$. The combined extracts were washed with water $(70 \mathrm{~mL})$ and brine $(70 \mathrm{~mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Purification by chromatography on $\mathrm{SiO}_{2}(30 \% \mathrm{EtOAc} /$ hexanes $)$ afforded labelled alcohol ( $4.26 \mathrm{~g}, 35.8 \mathrm{mmol}, 100 \%$ ) as a yellow-tinted oil. The material contained trace EtOAc but was carried forward without further purification. ${ }^{1} \mathrm{H}$ NMR suggests $\sim 90 \% \mathrm{D}$ incorporation at the alkene: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.52-7.18(\mathrm{~m}, 5 \mathrm{H}), 5.46(\mathrm{~s}, 1 \mathrm{H}), 4.55(\mathrm{~s}, 2 \mathrm{H})$.


2,2-Dibromo-1-phenylcyclopropyl-3-d)methanol (3-105). A solution of alcohol 3-104 (4.26 g, $31.5 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(60 \mathrm{~mL})$ was treated with 3,4-dihydro-2H-pyran ( $2.9 \mathrm{~mL}, 32 \mathrm{mmol}, 1.2 \mathrm{eq}$ ) followed by PPTS $(0.791 \mathrm{~g}, 3.15 \mathrm{mmol}, 0.10 \mathrm{eq})$. The resulting solution was stirred at rt for 4 h . The reaction mixture was then concentrated on the rotovap. The residue was partitioned between ether and water ( 75 mL each). The layers were separated, and the organic layer was washed with
sat aq $\mathrm{NaHCO}_{3}(75 \mathrm{~mL})$. The organic layer was then dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated to afford the protected alcohol $(7.304 \mathrm{~g})$. The material was used without further purification, and a $100 \%$ yield was assumed.

A solution of protected alcohol $(6.91 \mathrm{~g}, 31.5 \mathrm{mmol})$ and cetyltrimethyl ammonium bromide $(0.116 \mathrm{~g}, 0.315 \mathrm{mmol}, 1 \mathrm{~mol} \%)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(10 \mathrm{~mL})$ in a 250 mL 3-neck flask was affixed with an $\mathrm{N}_{2}$ inlet, mechanical stirrer, and septum. Bromoform ( $8.3 \mathrm{~mL}, 95 \mathrm{mmol}, 3.00 \mathrm{eq}$ ) was added, and the solution was cooled to $0^{\circ} \mathrm{C}$. Sodium hydroxide $(15.1 \mathrm{~g}, 0.378 \mathrm{~mol}, 12.0 \mathrm{eq})$ was added as a solution in water ( $30 \mathrm{~mL}, \sim 50 \mathrm{~g} / 100 \mathrm{~mL}$ ) via syringe over 10 min . The ice bath was removed and the flask was stirred for 25 h at 300 rpm . The reaction mixture became very dark and thick. The material was partitioned between water $(50 \mathrm{~mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$, filtered through Celite, and the layers were separated. The aqueous phase was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$. The combined extracts were washed with water $(25 \mathrm{~mL}), 1: 1$ water/brine $(25 \mathrm{~mL})$, then was dried $\left(\mathrm{MgSO}_{4}\right)$, filtered through a pad of $\mathrm{SiO}_{2}$, and concentrated.

This crude dark red oil was dissolved in $\mathrm{MeOH}(30 \mathrm{~mL})$ and treated with $p$-toluenesulfonic acid monohydrate ( $0.599 \mathrm{~g}, 3.15 \mathrm{mmol}, 10 \mathrm{~mol} \%$ ). After stirring at rt for 5 h , the reaction mixture was concentrated. The residue was partitioned between water and ether ( 75 mL each ). The layers were separated and the organic layer was washed with water ( 75 mL ), sat aq $\mathrm{NaHCO}_{3}(75 \mathrm{~mL})$, and brine $(75 \mathrm{~mL})$. The organic layer was then dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Purification by chromatography on $\mathrm{SiO}_{2}$ (15-50\% EtOAc/hexanes) afforded alcohol 3-105 (3.14 g, 10.2 mmol , $32 \%$ over 4 steps $)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.52-7.28(\mathrm{~m}, 5 \mathrm{H}), 4.05,3.95$ $\left(\mathrm{AB} \mathrm{q}, 2 \mathrm{H}, J_{\mathrm{AB}}=11.6 \mathrm{~Hz}\right), 2.09(\operatorname{app~s}, 1 \mathrm{H}), 1.72(\mathrm{br} \mathrm{s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 138.3,129.7,128.6,128.0,126.1,70.3,40.8,32.4,31.5\left({ }^{1} J_{\mathrm{CD}}=27 \mathrm{~Hz}\right)$.


3-106

2,2-Dibromo-1-(bromomethyl)cyclopropyl-3-d)benzene (3-106). A solution of alcohol 3-105 (3.14 g, 10.2 mmol ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ was treated with triphenylphosphine $(2.95 \mathrm{~g}, 11.3$ mmol, 1.10 eq ) followed by carbon tetrabromide ( $3.73 \mathrm{~g}, 11.3 \mathrm{mmol}, 1.10 \mathrm{eq}$ ). The reaction was warmed to rt for 4 h . The solution was then concentrated, and the residue was purified by chromatography on $\mathrm{SiO}_{2}$ to afford tribromide 3-106 ( $2.62 \mathrm{~g}, 7.08 \mathrm{mmol}, 69 \%$ ) as a colorless oil. The material contained $\sim 20 \%$ of allylic bromide (from unreacted cyclopropane in the previous step). After standing overnight, the material solidified. This solid was recrystallized from hexanes to afford $0.962 \mathrm{~g}(0.962 \mathrm{~g}, 25 \%)$ of material as a colorless solid: ${ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 7.44-7.31(m, 5 H), 3.97, $3.84\left(\mathrm{AB} \mathrm{q}, 2 \mathrm{H}, J_{\mathrm{AB}}=10.4 \mathrm{~Hz}\right), 2.25(\mathrm{~s}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 138.2,129.8,128.4,128.2,42.5,40.0,34.7,34.4\left({ }^{1} J_{\mathrm{CD}}=25 \mathrm{~Hz}\right)$.


N, 3-Diphenylbicyclo[1.1.0]butane-2-d-1-carboxamide (3-107). Prepared according to General Procedure G from tribromide 3-106 ( $0.898 \mathrm{~g}, 2.43 \mathrm{mmol}, 2.00 \mathrm{eq})$, $\mathrm{MeLi}(1.5 \mathrm{~mL}, 2.4 \mathrm{mmol}, 1.6$ M in $\mathrm{Et}_{2} \mathrm{O}$ ), $t$ - $\mathrm{BuLi}(1.4 \mathrm{~mL}, 2.4 \mathrm{mmol}, 1.7 \mathrm{M}$ in pentane), and phenyl isocyanate ( $0.13 \mathrm{~mL}, 1.2$ $\mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(10 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(15-40 \% \mathrm{EtOAc} /$ hexanes $)$ afforded amide 3-107 (0.241 g, $0.963 \mathrm{mmol}, 79 \%$ ) as a colorless solid: ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , acetone-D ${ }_{6}$ ) $\delta 8.95(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.50(\mathrm{~d}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.38(\mathrm{~d}, 2 \mathrm{H}, J=7.2 \mathrm{~Hz}), 7.27(\mathrm{t}, 2 \mathrm{H}, J$
$=7.6 \mathrm{~Hz}), 7.18(\mathrm{t}, 3 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.94(\mathrm{t}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 3.06(\mathrm{~d}, 2 \mathrm{H}, J=2.4 \mathrm{~Hz}), 1.53(\mathrm{~d}, 1$ $\mathrm{H}, J=1.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , acetone- $\mathrm{D}_{6}$ ) $\delta 166.9,140.3,135.4,129.3,129.1,127.3$, 126.9, 123.9, 120.4, 35.0, $34.7\left({ }^{1} J_{\mathrm{CD}}=25 \mathrm{~Hz}\right), 31.3$, 26.1.

cis 3-108, 24\%

trans 3-108, 18\%

3-Fluoro- $\boldsymbol{N}$,3-diphenylcyclobutane-2-d-1-carboxamide (3-108). Prepared according to General Procedure I from $\mathrm{KHF}_{2}(0.044 \mathrm{~g}, 0.57 \mathrm{mmol})$, diisopropylamine $(0.32 \mathrm{~mL}, 2.3 \mathrm{mmol}, 4.0 \mathrm{eq})$, pyridine• $9 \mathrm{HF}(1.0 \mathrm{~mL}$, 39 mmol HF , 68 eq HF$)$ and amide 3-107 ( $0.142 \mathrm{~g}, 0.567 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$. Chromatography on $\mathrm{SiO}_{2}(25 \% \mathrm{EtOAc} /$ hexanes $)$ afforded less polar cis 3-108 (38 $\mathrm{mg}, 0.14 \mathrm{mmol}, 25 \%)$ as a colorless solid and the more polar trans $\mathbf{3 - 1 0 8}(28 \mathrm{mg}, 0.15 \mathrm{mmol}, 18 \%)$ as a colorless solid. cis 3-108: Mp 107-110 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.53(\mathrm{~d}, 2 \mathrm{H}, J=$ $7.6 \mathrm{~Hz}), 7.50(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.40(\mathrm{t}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.36-7.29(\mathrm{~m}, 3 \mathrm{H}), 7.24(\mathrm{br} \mathrm{s}, 1 \mathrm{H})$, $7.12(\mathrm{t}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.46(\mathrm{AB} X(\operatorname{app~q}), 1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 3.12-2.93(\mathrm{ABX}(\mathrm{m}), 2 \mathrm{H}), 2.88-$ $2.75(A \mathrm{BX}(\mathrm{m}), 1 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 172.1,141.4\left({ }^{2} J_{\mathrm{CF}}=23 \mathrm{~Hz}\right), 137.7,129.1$, $128.4,128.2,124.64,124.56,124.51,119.8,97.4\left({ }^{1} J_{\mathrm{CF}}=192 \mathrm{~Hz}\right), 38.7\left({ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 34.2\left({ }^{1} J_{\mathrm{CD}}\right.$ not resolved); ${ }^{19} \mathrm{~F}$ NMR ( $470 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-146.3$; trans 3-108: $\mathrm{Mp} 146-148{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.62(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 7.54(\mathrm{~d}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.49(\mathrm{~d}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.45-7.34(\mathrm{~m}$, $3 \mathrm{H}), 7.31(\mathrm{t}, 2 \mathrm{H}, J=7.6 \mathrm{~Hz}), 7.11(\mathrm{t}, 1 \mathrm{H}, J=7.6 \mathrm{~Hz}), 3.12-2.93(\mathrm{~m}, 2 \mathrm{H}), 2.88-2.72(\mathrm{~m}, 2 \mathrm{H}) ;$ ${ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 171.6,140.8\left({ }^{2} J_{\mathrm{CF}}=22 \mathrm{~Hz}\right), 137.8,129.1,128.7,128.5,124.9\left({ }^{3} J_{\mathrm{CF}}\right.$ $=6 \mathrm{~Hz}), 124.5,120.0,91.3\left({ }^{1} J_{\mathrm{CF}}=211 \mathrm{~Hz}\right), 38.7\left({ }^{1} J_{\mathrm{CD}}=21 \mathrm{~Hz},{ }^{2} J_{\mathrm{CF}}=25 \mathrm{~Hz}\right), 30.9\left({ }^{3} J_{\mathrm{CF}}=13 \mathrm{~Hz}\right)$; ${ }^{19}$ F NMR ( $470 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$-119.2; MS (ESI') m/z 251 (100), 158 (5), 130 (5); HRMS (ESI ${ }^{+}$) $m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{NOD}[\mathrm{M}+\mathrm{H}-\mathrm{HF}]$ 251.1295, found 251.1285.

### 5.5 CHAPTER 4 EXPERIMENTAL PART



1-(3-Bromoprop-1-en-2-yl)-4-fluorobenzene (4-5). ${ }^{171} \mathrm{~A}$ solution of methylstyrene 4-4 (4.78 g, 35.1 mmol ) in $\mathrm{CHCl}_{3}(7 \mathrm{~mL})$ was treated with NBS ( $\left.8.50 \mathrm{~g}, 47.7 \mathrm{mmol}, 1.36 \mathrm{eq}\right)$. The reaction was heated to reflux $\left(\sim 90^{\circ} \mathrm{C}\right)$ for 6 h then it was cooled and filtered through Celite. The filter cake was washed with hexane $(50 \mathrm{~mL})$ and the eluent was concentrated. The residue was purified by chromatography on $\mathrm{SiO}_{2}$ (hexane) to afford bromide $4-5(3.38 \mathrm{~g}, 15.7 \mathrm{mmol}, 45 \%)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.48-7.43(\mathrm{~m}, 2 \mathrm{H}), 7.13-7.01(\mathrm{~m}, 2 \mathrm{H}), 5.50(\mathrm{~s}, 1 \mathrm{H}), 5.48(\mathrm{~s}$, $1 \mathrm{H}), 4.36(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-113.7$.


1-(2,2-Dibromo-1-(bromomethyl)cyclopropyl)-4-fluorobenzene (4-6). A solution of allylic bromide 4-5 (3.17 g, 14.7 mmol$)$ and cetyltrimethyl ammonium bromide $(0.054 \mathrm{~g}, 0.15 \mathrm{mmol}, 1$ mol \%) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$ was charged to a 100 mL 3-neck flask affixed with an $\mathrm{N}_{2}$ inlet, mechanical stirrer, and septum. Bromoform ( $3.9 \mathrm{~mL}, 44 \mathrm{mmol}, 3.0 \mathrm{eq}$ ) was added by syringe, and the solution was cooled to $0^{\circ} \mathrm{C}$. Sodium hydroxide $(7.07 \mathrm{~g}, 0.177 \mathrm{~mol})$ was added as a solution in water ( $15 \mathrm{~mL}, \sim 50 \mathrm{~g} / 100 \mathrm{~mL}$ ) via syringe over 5 min . The ice bath was removed and the flask stirred for 26 h at 300 rpm (color changed from clear to a dark, opaque brown). The reaction mixture was then partitioned between water $(50 \mathrm{~mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$. The layers were separated, and the aqueous phase was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$. The combined extracts were
washed with water $(25 \mathrm{~mL}), 1: 1$ water $/$ brine $(25 \mathrm{~mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. The residue was purified by column chromatography on $\mathrm{SiO}_{2}$ ( $10 \% \mathrm{EtOAc}$ in hexane). Upon concentration of the product-containing fractions, a solid mass formed. This material was then recrystallized from $\sim 1: 1 \mathrm{EtOAc} /$ hexanes $(\sim 15 \mathrm{~mL})$, and the crystals were washed with cold hexanes ( 30 mL ). After drying under vacuum, tribromide 4-6 ( $1.68 \mathrm{~g}, 4.35 \mathrm{mmol}, 30 \%$ ) was obtained as a colorless crystalline solid: $\mathrm{Mp} 101-103{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.38-7.30$ (m, 2 H), 7.08 (app t, $2 H, J=8.8 \mathrm{~Hz}$ ), $3.91(\mathrm{dd}, 2 \mathrm{H}, J=10.8 \mathrm{~Hz}, 1.6 \mathrm{~Hz}), 3.82(\mathrm{~d}, 1 \mathrm{H}, J=10.4$ $\mathrm{Hz}), 2.21(\mathrm{dd}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 2.01(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}) ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-$ 113.2.


2-(4-(Trifluoromethyl)phenyl)prop-2-en-1-ol (4-8). ${ }^{181}$ Magnesium turnings ( $2.71 \mathrm{~g}, 0.111 \mathrm{~mol}$ ) were charged to a 500 mL 3-neck round bottomed flask fitted with a reflux condenser, septum, and internal thermometer. After flushing with $\mathrm{N}_{2}$, the turnings were suspended in ether ( 120 mL ). Neat aryl bromide 4-7 ( $15.8 \mathrm{~mL}, 0.111 \mathrm{~mol}$ ) was added by syringe in $\sim 1 \mathrm{~mL}$ portions over $\sim 15$ min, which caused the formation of a dark brown Grignard solution and spontaneous heating of the reaction to reflux. After $\sim 1 \mathrm{~h}$, the reaction has cooled from reflux to rt. To the freshly prepared Grignard reagent was added $\mathrm{CuI}(1.27 \mathrm{~g}, 6.67 \mathrm{mmol})$ and the black suspension was stirred for 15 min. A solution of propargyl alcohol ( $2.60 \mathrm{~mL}, 44.6 \mathrm{mmol}$ ) in ether $(50 \mathrm{~mL})$ was added via cannula over 1.75 h , causing an exotherm to $29^{\circ} \mathrm{C}$. The reaction was stirred at rt for 18 h , at which point a quenched aliquot showed consumption of propargyl alcohol by ${ }^{1} \mathrm{H}$ NMR. The reaction was then cooled to $0{ }^{\circ} \mathrm{C}$ and quenched with sat aq $\mathrm{NH}_{4} \mathrm{Cl}(100 \mathrm{~mL})$. The layers were separated and the aqueous layer was extracted with ether ( $2 \times 100 \mathrm{~mL}$ ). The organic extracts were washed with water
$(100 \mathrm{~mL})$ and brine $(100 \mathrm{~mL})$, then they were dried $\left(\mathrm{MgSO}_{4}\right)$, filtered, and concentrated. The crude product was purified by chromatography on $\mathrm{SiO}_{2}(25 \% \mathrm{EtOAc} /$ hexanes $)$ to afford alcohol 4-8 $(4.86 \mathrm{~g}, 24.0 \mathrm{mmol}, 54 \%)$ as a red-tinted oil: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.58(\mathrm{AB} \mathrm{q}, 4 \mathrm{H}, J=$ $\left.8.4 \mathrm{~Hz}, \partial_{\mathrm{AB}}=0.05 \mathrm{ppm}\right), 5.55(\mathrm{~s}, 1 \mathrm{H}), 5.46(\mathrm{~s}, 1 \mathrm{H}), 4.56(\mathrm{~s}, 2 \mathrm{H}), 1.62(\mathrm{~s}, 1 \mathrm{H})$.

(2,2-Dibromo-1-(4-(trifluoromethyl)phenyl)cyclopropyl)methanol (4-9). A solution of alcohol 4-8 (4.86 g, 24.0 mmol) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(40 \mathrm{~mL})$ was treated with 3,4-dihydro-2H-pyran ( $3.1 \mathrm{~mL}, 34$ mmol, 1.4 eq$)$ followed by PPTS $(0.604 \mathrm{~g}, 2.40 \mathrm{mmol}, 0.10 \mathrm{eq})$. The resulting solution was stirred at rt for 2.5 h . The reaction mixture was then concentrated on the rotovap. The residue was partitioned between ether and water ( 75 mL each). The layers were separated, and the organic layer was washed with sat aq $\mathrm{NaHCO}_{3}(75 \mathrm{~mL})$. The organic layer was then dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated to afford the protected alcohol $(7.11 \mathrm{~g})$ as a red-tinted oil. The material contained residual solvents and was carried on without further purification.

A solution of protected alcohol $(7.11 \mathrm{~g}, 24.8 \mathrm{mmol})$ and cetyltrimethyl ammonium bromide ( $0.091 \mathrm{~g}, 0.25 \mathrm{mmol}, 1 \mathrm{~mol} \%$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(12 \mathrm{~mL})$ in a 250 mL 3-neck flask was affixed with an $\mathrm{N}_{2}$ inlet, mechanical stirrer, and septum. Bromoform ( $6.5 \mathrm{~mL}, 75 \mathrm{mmol}, 3.0 \mathrm{eq}$ ) was added, and the solution was cooled to $0^{\circ} \mathrm{C}$. Sodium hydroxide $(11.9 \mathrm{~g}, 298 \mathrm{mmol}, 12.0 \mathrm{eq})$ was added as a solution in water ( $24 \mathrm{~mL}, \sim 50 \mathrm{~g} / 100 \mathrm{~mL}$ ) via syringe over 10 min . The ice bath was removed and the flask stirred for 24 h at $300 \mathrm{rpm} .{ }^{1} \mathrm{H}$ NMR of an aliquot showed consumption of the alkene. The reaction mixture was partitioned between water $(50 \mathrm{~mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$, the layers were separated, and the aqueous phase was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$. The combined extracts were
washed with water $(25 \mathrm{~mL}), 1: 1$ water $/$ brine $(25 \mathrm{~mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated.
This crude dark red oil was dissolved in $\mathrm{MeOH}(25 \mathrm{~mL})$ and treated with $p$-toluenesulfonic acid monohydrate ( $0.472 \mathrm{~g}, 2.48 \mathrm{mmol}, 10 \mathrm{~mol} \%)$. After stirring at rt for 3 h , the reaction mixture was concentrated. The residue was partitioned between water and ether ( 75 mL each ). The layers were separated and the organic layer was washed with water ( 75 mL ), sat aq $\mathrm{NaHCO}_{3}(75 \mathrm{~mL})$, and brine $(75 \mathrm{~mL})$. The organic layer was then dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Purification by chromatography on $\mathrm{SiO}_{2}$ afforded $\mathbf{4 - 9}(4.45 \mathrm{~g}, 11.9 \mathrm{mmol}, 48 \%$ over 3 steps $)$ as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.65(\operatorname{app~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.52(\operatorname{app~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 4.01(\mathrm{AB}$ $\left.\mathrm{q}, J=12.0 \mathrm{~Hz}, \partial_{\mathrm{AB}}=0.04 \mathrm{ppm}\right), 2.11\left(\mathrm{AB} \mathrm{q}, 4 \mathrm{H}, J=7.6 \mathrm{~Hz}, \partial_{\mathrm{AB}}=0.03 \mathrm{ppm}\right) ;{ }^{13} \mathrm{C}$ NMR $(100$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.4,130.2,125.4\left(\mathrm{q},{ }^{3} \mathrm{~J}_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 70.1,40.7,31.9,31.3,21.1 ;{ }^{19} \mathrm{~F}$ NMR $(376$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-62.6$.


1-(2,2-Dibromo-1-(bromomethyl)cyclopropyl)-4-(trifluoromethyl)benzene (4-10). A solution of alcohol 4-9 (4.45 g, 11.9 mmol$)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(35 \mathrm{~mL})$ was cooled to $0{ }^{\circ} \mathrm{C}$ under $\mathrm{N}_{2} . \mathrm{PPh}_{3}(3.75 \mathrm{~g}$, $13.1 \mathrm{mmol}, 1.20 \mathrm{eq})$ was then charged, followed by $\mathrm{CBr}_{4}(4.34 \mathrm{~g}, 13.1 \mathrm{mmol}, 1.10 \mathrm{eq})$. The reaction was warmed to rt and stirred for 20 h , then it was concentrated. Purification by chromatography on $\mathrm{SiO}_{2}$ (0-15\% EtOAc in hexanes) afforded bromide 4-10 (4.31 g, 9.85 mmol , $83 \%$ ) as a colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.66(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.49(\mathrm{~d}, 2 \mathrm{H}, J=$ $8.0 \mathrm{~Hz}), 3.93(\mathrm{dd}, 1 \mathrm{H}, J=10.8 \mathrm{~Hz}, 1.2 \mathrm{~Hz}), 3.83(\mathrm{~d}, 1 \mathrm{H}, J=10.8 \mathrm{~Hz}), 2.27(\mathrm{dd}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}$, $1.2 \mathrm{~Hz}), 2.06(\mathrm{~d}, 1 \mathrm{H}, J=8.0 \mathrm{~Hz}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 142.0,130.6,130.3,125.4(\mathrm{q}$, $\left.{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 41.7,39.8,34.8,34.7,33.4,31.6,14.1 ;{ }^{19} \mathrm{~F}$ NMR $\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-62.6$.


General Procedure J (Bicyclo[1.1.0]butyl carboxylic acid synthesis): 3-(4-Fluorophenyl)bicyclo[1.1.0]butane-1-carboxylic acid (4-11). A solution of dibromocyclopropane $\mathbf{4 - 8}(0.880 \mathrm{~g}, 2.57 \mathrm{mmol})$ in $\mathrm{Et}_{2} \mathrm{O}(7 \mathrm{~mL})$ and THF ( 3 mL ) in a 50 mL round bottom flask was placed under $\mathrm{N}_{2}$ and cooled to $-78^{\circ} \mathrm{C}$. $\mathrm{MeLi}(1.6 \mathrm{~mL}, 2.6 \mathrm{mmol}, 1.0 \mathrm{eq})$ was added via syringe over 10 min . After 1.75 h at $-78^{\circ} \mathrm{C}, t-\mathrm{BuLi}(1.5 \mathrm{~mL}, 2.6 \mathrm{mmol}, 1.0 \mathrm{eq})$ was added via syringe over 10 min . After an additional 1 h at $-78^{\circ} \mathrm{C}$, a balloon of dry $\mathrm{CO}_{2}$ was bubbled through the solution via a needle for 10 min (an external bubbler was used, providing a continuous stream of $\mathrm{CO}_{2}$ through the flask). The cooling bath was then removed, and the opaque grey reaction mixture was warmed to $0{ }^{\circ} \mathrm{C}$ over 20 min . The reaction mixture was then quenched with 1 M $\mathrm{NaOH}(10 \mathrm{~mL})$, then it was partitioned between ether and $1 \mathrm{M} \mathrm{NaOH}(50 \mathrm{~mL}$ each $)$. The light brown aqueous layer was acidified to litmus with conc HCl , which caused formation of a light brown precipitate. This suspension was then extracted with ether $(40 \mathrm{~mL})$. The organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated to afford acid 4-11 ( $\left.0.228 \mathrm{~g}, 1.19 \mathrm{mmol}, 57 \%\right)$ as a buff solid: Mp 158-160 ${ }^{\circ} \mathrm{C}$; IR (ATR) 1646, 1527, 1226, $841 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (300 MHz, $\mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 7.41-7.32$ (m, 2 H), $7.05(\mathrm{appt}, 2 \mathrm{H}, J=9.0 \mathrm{~Hz}), 2.88(\mathrm{t}, 2 \mathrm{H}, J=1.2 \mathrm{~Hz}), 1.60(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CD}_{3} \mathrm{OD}\right)\left({ }^{1} J_{\mathrm{CF}}\right.$ not resolved) $\delta 172.2,127.5\left(\mathrm{~d},{ }^{3} J_{\mathrm{CF}}=8 \mathrm{~Hz}\right), 114.9\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=21 \mathrm{~Hz}\right), 34.9,31.8$, 22.4; ${ }^{19}$ F NMR ( $376 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta-117.7$; HRMS $m / z$ calcd for $\mathrm{C}_{11} \mathrm{H}_{10} \mathrm{FO}_{2}[\mathrm{M}+\mathrm{H}]$ 193.0665, found 193.0660.


3-(4-(Trifluoromethyl)phenyl)bicyclo[1.1.0]butane-1-carboxylic acid (4-12). Prepared according to General Procedure J from tribromocyclopropane 4-12 (1.50 g, 3.43 mmol), $\mathrm{MeLi}(2.1$ $\mathrm{mL}, 3.4 \mathrm{mmol}, 1.6 \mathrm{M}$ in $\left.\mathrm{Et}_{2} \mathrm{O}\right)$, and $t-\mathrm{BuLi}(2.0 \mathrm{~mL}, 3.4 \mathrm{mmol}, 1.7 \mathrm{M}$ in pentane, 1.0 eq$)$ to afford acid 4-12 ( $0.416 \mathrm{~g}, 1.72 \mathrm{mmol}, 50 \%$ ) as a colorless solid: Mp 146-148 ${ }^{\circ} \mathrm{C}$; IR (ATR) 2980, 1648, 1616, 1463, 1318, 1167, 1116, 1061, 907, 842, $689 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}$ ) $\delta 7.61(\mathrm{~d}$, $2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.52(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 4.88(\mathrm{~s}, 1 \mathrm{H}), 2.99(\mathrm{t}, 2 \mathrm{H}, J=1.2 \mathrm{~Hz}), 1.68(\mathrm{~s}, 2 \mathrm{H}) ;$ ${ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CD}_{3} \mathrm{OD}\right) \delta 171.5,140.0,128.4\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 126.2,124.9\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4\right.$ $\mathrm{Hz}), 124.4\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=269 \mathrm{~Hz}\right), 35.0,31.0,24.2 ;{ }^{19} \mathrm{~F} \operatorname{NMR}\left(376 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-64.0 ; \mathrm{MS}(\mathrm{ESI})^{+}$ $m / z 243$ (80), 233 (50), 231 (50), 197 (60), 177 (40), 155 (100); HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~F}_{3} \mathrm{O}_{2}[\mathrm{M}+\mathrm{H}] 243.0633$, found 243.0626.


## (4-(5-chloro-2-(trifluoromethyl)phenyl)piperazin-1-yl)(3-(4-fluorophenyl)bicyclo[1.1.0]

butan-1-yl)methanone (4-13). A solution of bicyclobutane acid 4-11 (0.018 g, $0.096 \mathrm{mmol}, 1.1$ eq) and piperazine free base $(0.023 \mathrm{~g}, 0.087 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ was cooled to $0^{\circ} \mathrm{C}$ and treated with $\mathrm{Et}_{3} \mathrm{~N}(0.02 \mathrm{~mL}, 0.2 \mathrm{mmol}, 2 \mathrm{eq})$. The transparent beige solution was then treated with T3P ( $50 \%$ solution in EtOAc) ( $0.09 \mathrm{~mL}, 0.13 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) dropwise via syringe over $\sim 1 \mathrm{~min}$. The reaction was stirred at $0^{\circ} \mathrm{C}$ for 30 min , then the ice bath was removed and the reaction was warmed to rt for 19.5 h . The reaction mixture was then directly purified by chromatography on $\mathrm{SiO}_{2}$ (15-
$25 \% \mathrm{EtOAc} /$ hexanes $)$ without workup to afford amide $4-13(0.015 \mathrm{~g}, 0.038 \mathrm{mmol}, 39 \%)$ as a colorless oil: IR (ATR) 2923, 1620, 1596, 1434, 1307, 1221, 1120, 1027, 906, 933, $725 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.56(\mathrm{~d}, 1 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.34-7.27(\mathrm{~m}, 2 \mathrm{H}), 7.25-7.19(\mathrm{~m}, 2 \mathrm{H}), 7.01$ (app t, $2 \mathrm{H}, J=8.8 \mathrm{~Hz}$ ), 4.1-3.5 (br m, 4 H ), 2.9-2.7 (br m, 4 H ), $2.75(\mathrm{~s}, 2 \mathrm{H}), 1.63(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ) (piperazine carbons broadened and are not clearly visible) $\delta$ 167.7, 161.9 $\left(\mathrm{d},{ }^{1} J_{\mathrm{CF}}=243 \mathrm{~Hz}\right), 152.9,138.8,135.1,129.7\left(\mathrm{~d},{ }^{4} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right), 128.6\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=5 \mathrm{~Hz}\right), 128.0(\mathrm{~d}$, $\left.{ }^{3} J_{\mathrm{CF}}=8 \mathrm{~Hz}\right), 126.3\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=272 \mathrm{~Hz}\right), 125.4\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=30 \mathrm{~Hz}\right), 115.4\left(\mathrm{~d},{ }^{2} J_{\mathrm{CF}}=21 \mathrm{~Hz}\right), 36.9,30.6$, 30.0; ${ }^{19}$ F NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-60.3,-115.9 ; \mathrm{MS}\left(\mathrm{ESI}{ }^{+} m / z 439\right.$ (100), 265 (3), 175 (5), 147 (45); HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{22} \mathrm{H}_{20} \mathrm{ClF}_{4} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}] 439.1195$, found 439.1193 .


## (4-(5-Chloro-2-(trifluoromethyl)phenyl)piperazin-1-yl)(3-(4(trifluoromethyl)phenyl)bicyclo

 [1.1.0]butan-1-yl)methanone (4-14). A solution of bicyclobutane acid 4-12 ( $0.046 \mathrm{~g}, 0.19 \mathrm{mmol}$, $1.0 \mathrm{eq})$ and piperazine $\mathrm{HCl}(0.057 \mathrm{~g}, 0.19 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{~mL})$ cooled to $0{ }^{\circ} \mathrm{C}$ was treated $\mathrm{Et}_{3} \mathrm{~N}(0.08 \mathrm{~mL}, 0.6 \mathrm{mmol}, 3 \mathrm{eq})$. The transparent beige solution was then treated with T3P (50\% solution in EtOAc) ( $0.2 \mathrm{~mL}, 0.28 \mathrm{mmol}, 1.5 \mathrm{eq}$ ) dropwise via syringe over $\sim 1 \mathrm{~min}$. The reaction was stirred at $0^{\circ} \mathrm{C}$ for 30 min , then the ice bath was removed and the reaction was warmed to rt for 22 h . The reaction mixture was then partitioned between water and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 30 mL each). The layers were separated, and the aqueous layer was extracted with additional $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The combined extracts were washed with water $(30 \mathrm{~mL})$ and sat aq $\mathrm{NaHCO}_{3}(30 \mathrm{~mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Purification by chromatography on $\mathrm{SiO}_{2}(15-25 \% \mathrm{EtOAc} /$ hexanes $)$afforded amide $\mathbf{4 - 1 4}(0.051 \mathrm{~g}, 0.10 \mathrm{mmol}, 55 \%)$ as a colorless solid: Mp $137-139^{\circ} \mathrm{C}$; IR (ATR) 2902, 1604, 1466, 1438, 1312, 1100, 1027, 924, 838, $825 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.56$ (app d, $3 \mathrm{H}, J=8.4 \mathrm{~Hz}$ ), $7.43(\operatorname{app~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.23(\operatorname{app~q}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 4.1-3.5(\mathrm{br}$, $4 \mathrm{H}), 3.0-2.7(\mathrm{br}, 4 \mathrm{H}), 2.83(\mathrm{~s}, 2 \mathrm{H}), 1.69(\mathrm{~s}, 2 \mathrm{H}),{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (piperazine carbons broadened and are not clearly visible) $\delta 167.1,152.8,138.8,138.6,128.5\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32\right.$ $\mathrm{Hz}), 128.5\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 126.6,125.7\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=29 \mathrm{~Hz}\right), 125.5,125.3\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right), 124.6$, $124.3\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=270 \mathrm{~Hz}\right), 123.6\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=272 \mathrm{~Hz}\right), 37.0,30.0,29.7,22.7 ;{ }^{19} \mathrm{~F}$ NMR $(376 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta-60.3,-62.4 ; \mathrm{MS}(\mathrm{ESI})^{+} m / z 489,256(5) ; \mathrm{HRMS}(\mathrm{ESI})^{+} \mathrm{m} / \mathrm{z}$ calcd for $\mathrm{C}_{23} \mathrm{H}_{20} \mathrm{~F}_{6} \mathrm{ClN}_{2} \mathrm{O}$ [M+H] 489.1168, found 489.1158.


## (4-(2-Methyl-5-(trifluoromethyl)phenyl)piperazin-1-yl)(3-(4-(trifluoromethyl)phenyl)

bicyclo[1.1.0]butan-1-yl)methanone (4-15). A solution of bicyclobutane acid 4-12 ( $0.052 \mathrm{~g}, 0.21$ mmol) and piperazine free base ( $1.1 \mathrm{eq}, \mathrm{HCl}$ salt washed with NaOH , dried and concentrated before use) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$ cooled to $0{ }^{\circ} \mathrm{C}$ and treated with $\mathrm{Et}_{3} \mathrm{~N}(0.06 \mathrm{~mL}, 0.43 \mathrm{mmol}, 2.0 \mathrm{eq})$. The transparent beige solution was treated with T3P $(0.23 \mathrm{~mL}, 0.32 \mathrm{mmol}, 1.5 \mathrm{eq}, 50 \%$ solution in EtOAc) dropwise via syringe. The reaction was stirred at $0^{\circ} \mathrm{C}$ for 30 min , then the ice bath was removed and the reaction was warmed to rt for 17 h . The reaction mixture was partitioned between water and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 50 mL each). The layers were separated, and the aqueous layer was extracted with additional $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The combined extracts were washed with water $(50 \mathrm{~mL})$ and sat aq $\mathrm{NaHCO}_{3}(40 \mathrm{~mL})$, then were dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Purification by chromatography on $\mathrm{SiO}_{2}$ (15-25\% EtOAc/hexanes) afforded amide 4-15 (0.072 g, $\left.0.154 \mathrm{mmol}, 72 \%\right)$ as a colorless
oil: IR (ATR) 2821, 1618, 1435, 1418, 1323, 1162, 1114, 1030, 840, $732 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.56(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.43(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.28(\operatorname{app} \mathrm{q}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz})$, $7.18(\mathrm{~s}, 1 \mathrm{H}), 4.1-3.5(\mathrm{br}, 4 \mathrm{H}), 3.0-2.7(\mathrm{br}, 4 \mathrm{H}), 2.84(\mathrm{~s}, 2 \mathrm{H}), 2.36(\mathrm{~s}, 3 \mathrm{H}), 1.70(\mathrm{~s}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) (piperazine carbons broadened and are not clearly visible) $\delta$ 167.1, $151.2,138.6,136.8,131.6,129.1\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 128.5\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 126.6,125.3\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=\right.$ $4 \mathrm{~Hz}), 124.3\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=270 \mathrm{~Hz}\right), 124.2\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=271 \mathrm{~Hz}\right), 120.4\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=3 \mathrm{~Hz}\right), 116.0\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4\right.$ $\mathrm{Hz}), 37.0,30.0,22.8,18.0 ;{ }^{19} \mathrm{~F}$ NMR (376 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta-62.2,-62.4 ; \mathrm{MS}(\mathrm{ESI})^{+} m / z 469$ (100), 197 (5); HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{24} \mathrm{H}_{23} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}] 469.1709$, found 469.1705 .


General Procedure K (Rearrangement of Bicyclo[1.1.0[butyl amides): ((E)-(4-(2-methyl-5-(trifluoromethyl)phenyl)piperazin-1-yl)(2-(4-(trifluoromethyl)
cyclopropyl)methanone (4-16). A solution of amide 4-13 (0.680 g, 1.45 mmol$)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(8 \mathrm{~mL})$ was treated with camphorsulfonic acid $(0.050 \mathrm{~g}, 0.22 \mathrm{mmol}, 15 \mathrm{~mol} \%)$. The vial was sealed and heated at $50{ }^{\circ} \mathrm{C}$ for 18 h . After cooling, the solution was partitioned between $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at sat aq $\mathrm{NaHCO}_{3}(20 \mathrm{~mL}$ each $)$. The organic layer was dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and concentrated. Purification by chromatography on $\mathrm{SiO}_{2}$ afforded methylenecyclopropane $\mathbf{4 - 1 6}(0.484 \mathrm{~g}, 1.03 \mathrm{mmol}, 71 \%)$ as an off-white foam: IR (ATR) 2921, 1638, 1417, 1324, 1116, 1069, 825, $729 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 400 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 7.58(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.46(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.33-7.27(\mathrm{~m}, 2 \mathrm{H}), 7.21(\mathrm{~s}, 1$ H), $6.53(\mathrm{~d}, 1 \mathrm{H}, J=0.8 \mathrm{~Hz}), 3.87(\operatorname{app~s}, 1 \mathrm{H}), 3.80(\mathrm{q}, 2 \mathrm{H}, J=5.2 \mathrm{~Hz}), 3.69(\mathrm{t}, 2 \mathrm{H}, J=4.8 \mathrm{~Hz})$, 3.17-3.07(m, 2 H$), 2.99(\mathrm{t}, 2 \mathrm{H}, J=4.8 \mathrm{~Hz}), 2.91(\mathrm{t}, 2 \mathrm{H}, J=4.4 \mathrm{~Hz}), 2.39(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (100
$\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 170.7,151.2,145.7,136.94,136.88,131.6,129.9\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=33 \mathrm{~Hz}\right), 129.1\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}\right.$ $=32 \mathrm{~Hz}), 128.1,126.6,126.4,125.4\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 124.9,124.2\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=271 \mathrm{~Hz}\right), 124.1(\mathrm{q}$, $\left.{ }^{1} J_{\mathrm{CF}}=271 \mathrm{~Hz}\right), 120.4\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 116.0\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 51.9,51.7,45.6,42.1,40.8,32.7$, 29.7, 21.0, 18.0; ${ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-62.2,-62.5$; MS (ESI) ${ }^{+} m / z 469$ (100), 371 (1), 225 (2); HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{24} \mathrm{H}_{23} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}] 469.1709$, found 469.1715.


## (4-(2-Methyl-5-(trifluoromethyl)phenyl)piperazin-1-yl)(3-(4-(trifluoromethyl)phenyl)

cyclobutyl)methanone (4-17). A solution of amide $4-13$ ( $0.072 \mathrm{~g}, 0.153 \mathrm{mmol}$ ) in MeOH ( 3 mL ) was placed under $\mathrm{N}_{2}$ and treated with $\mathrm{Pd} / \mathrm{C}(3 \mathrm{mg}, \sim 20 \mathrm{~mol} \%)$. A hydrogen atmosphere of 6.43 bar was then established on the Parr hydrogenator. The reaction was stirred at rt for 18 h , at which point the hydrogen was vented and the solution was purified by chromatography on $\mathrm{SiO}_{2}$ (25-40\% EtOAc/hexanes) afforded cyclobutane 4-17 ( $0.037 \mathrm{~g}, 0.079 \mathrm{mmol}, 51 \%)$ as a colorless oil.

Alternatively, 4-17 was prepared by dissolving a solution of amide 4-16 ( $0.031 \mathrm{~g}, 0.080$ $\mathrm{mmol}, 1.0 \mathrm{eq})$ in $\mathrm{MeOH}(2 \mathrm{~mL})$ and treating with $\mathrm{Pd} / \mathrm{C}(2 \mathrm{mg}, 20 \mathrm{~mol} \%)$. A hydrogen atmosphere was then established with a balloon. After stirring at rt for 2 h , the reaction mixture was filtered through Celite and concentrated to afford cyclobutane $4-17(0.015 \mathrm{~g}, 0.039 \mathrm{mmol}, 48 \%)$ as a colorless oil: IR (ATR) 2939, 1641, 1618, 1417, 1324, 1308, 1161, 1112, 1067, 834, $732 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.55(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.36(\mathrm{~d}, 2 \mathrm{H}, J=8.4 \mathrm{~Hz}), 7.32-7.24(\mathrm{~m}, 2 \mathrm{H}$, $J=8.0 \mathrm{~Hz}), 7.19(\mathrm{~s}, 1 \mathrm{H}), 3.79(\mathrm{t}, 2 \mathrm{H}, J=4.8 \mathrm{~Hz}), 3.60(\mathrm{t}, 2 \mathrm{H}, J=5.2 \mathrm{~Hz}), 3.57-3.48(\mathrm{~m}, 1 \mathrm{H})$, 3.37-3.26 (m, 1 H ), 2.95-2.86 (m, 4 H ), 2.68-2.59 (m, 2 H ), 2.53 (ddd, $2 \mathrm{H}, J=19.2 \mathrm{~Hz}, 9.6 \mathrm{~Hz}$,
$2.4 \mathrm{~Hz}), 2.37(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 172.4,151.2,148.7,136.8,131.6,129.1(\mathrm{q}$, $\left.{ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 128.5\left(\mathrm{q},{ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 126.9,125.3\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 124.3\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=270 \mathrm{~Hz}\right), 124.2$ $\left(\mathrm{q},{ }^{1} J_{\mathrm{CF}}=270 \mathrm{~Hz}\right), 120.4\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 116.0\left(\mathrm{q},{ }^{3} J_{\mathrm{CF}}=5 \mathrm{~Hz}\right), 51.9,51.7,45.5,42.2,35.5,33.1$, 32.6, 18.0; ${ }^{19} \mathrm{~F}$ NMR ( $376 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta-62.3$; MS (ESI) ${ }^{+} m / z 471$ (100), 355 (1), 246 (3), 179 (3); HRMS (ESI) ${ }^{+} m / z$ calcd for $\mathrm{C}_{24} \mathrm{H}_{25} \mathrm{~F}_{6} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}] 471.1871$, found 471.1865 .

( $E$ )-2-Benzylidene- $N$-phenylcyclopropane-1-carboxamide (4-18). Prepared according to General Procedure K from amide 3-96 ( $0.041 \mathrm{~g}, 0.16 \mathrm{mmol}$ ) and camphorsulfonic acid ( 6 mg , $0.025 \mathrm{mmol}, 15 \mathrm{~mol} \%)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$. Purified by chromatography on $\mathrm{SiO}_{2}$ to afford amide 4-18 ( $0.027 \mathrm{~g}, 0.011 \mathrm{mmol}, 66 \%$ ) as a colorless solid: Mp $149-152{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 7.53(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.46-7.28(\mathrm{~m}, 8 \mathrm{H}), 7.10(\mathrm{t}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.41(\mathrm{~s}, 1 \mathrm{H}), 3.71$ $(\mathrm{d}, 1 \mathrm{H}, J=4.8 \mathrm{~Hz}), 3.24(\mathrm{dd}, 1 \mathrm{H}, J=13.6 \mathrm{~Hz}, 4.8 \mathrm{~Hz}), 3.00(\mathrm{dd}, 1 \mathrm{H}, J=13.6 \mathrm{~Hz}, 1.2 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 171.5,150.9,137.8,133.4,129.0,128.9,128.5,124.9,124.3,123.8$, 119.8, 44.5, 34.1; MS (ESI') $m / z 250$ (100), 185 (20), 129 (5); HRMS (ESI ${ }^{+}$) $m / z$ calcd for $\mathrm{C}_{17} \mathrm{H}_{16} \mathrm{NO}[\mathrm{M}+\mathrm{H}] 250.1226$, found 250.1224 .


## (E)-2-(4-Chlorobenzylidene)- N -phenylcyclopropane-1-carboxamide

(4-19). Prepared according to General Procedure K from the corresponding bicyclobutane ( $0.043 \mathrm{~g}, 0.15 \mathrm{mmol}$, crystallized material) and camphorsulfonic acid ( $5 \mathrm{mg}, 0.023 \mathrm{mmol}, 15 \mathrm{~mol} \%$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ afforded amide $\mathbf{4 - 1 9}(0.036 \mathrm{~g}, 0.13 \mathrm{mmol}, 84 \%)$ as a
colorless semi-solid: ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.45(\mathrm{~d}, 2 \mathrm{H}, J=8.0 \mathrm{~Hz}), 7.33-7.15(\mathrm{~m}, 7 \mathrm{H})$, $7.02(\mathrm{t}, 1 \mathrm{H}, J=7.2 \mathrm{~Hz}), 6.32(\mathrm{~s}, 1 \mathrm{H}), 3.61(\mathrm{~d}, 1 \mathrm{H}, J=3.6 \mathrm{~Hz}), 3.12(\mathrm{dd}, 1 \mathrm{H}, J=12.8 \mathrm{~Hz}, 4.8$ $\mathrm{Hz}), 2.91(\mathrm{~d}, 1 \mathrm{H}, J=13.6 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR (100 MHz, $\left.\mathrm{CDCl}_{3}\right) \delta 171.1,149.5,137.7,129.1,128.8$, 126.2, 124.6, 124.4, 119.8, 44.4, 33.9; MS $\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 284$ (100), 195 (10); $\mathrm{HRMS}_{\left(\mathrm{ESI}^{+}\right) ~}^{\mathrm{m} / \mathrm{z} \text { calcd }}$ for $\mathrm{C}_{17} \mathrm{H}_{15} \mathrm{ClNO}[\mathrm{M}+\mathrm{H}]$ 284.0842, found 284.0836.

(E)-(2-Benzylidenecyclopropyl)(4-phenylpiperazin-1-yl)methanone
(4-20). Prepared according to General Procedure K from the corresponding bicyclobutane ( $0.057 \mathrm{~g}, 0.18 \mathrm{mmol}$ ) and camphorsulfonic acid ( $0.006 \mathrm{~g}, 0.03 \mathrm{mmol}, 15 \mathrm{~mol} \%$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}(25-60 \% \mathrm{EtOAc} /$ hexanes $)$ afforded $\mathbf{4 - 2 0}(0.043 \mathrm{~g}, 0.14 \mathrm{mmol}, 75 \%)$ as a colorless solid: Mp 138-142 ${ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.43-7.26(\mathrm{~m}, 7 \mathrm{H}), 7.10-6.92(\mathrm{~m}$, $3 \mathrm{H}), 6.42(\mathrm{~d}, 1 \mathrm{H}, J=1.2 \mathrm{~Hz}), 3.88-3.77(\mathrm{~m}, 3 \mathrm{H}), 3.71(\mathrm{dd}, 2 \mathrm{H}, J=4.8 \mathrm{~Hz}, 4.8 \mathrm{~Hz}), 3.28-3.17$ (m, 4 H ), 3.14-3.11 (m, 2 H ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 171.0,151.0,147.1,133.8,129.3$, $128.4,128.2,125.0,124.7,120.6,116.8,50.0,49.5,45.1,41.6,40.7,32.6 ; \mathrm{MS}^{\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 319}$ (100), 157 (10); HRMS $\left(\mathrm{ESI}^{+}\right) m / z$ calcd for $\mathrm{C}_{21} \mathrm{H}_{23} \mathrm{~N}_{2} \mathrm{O}[\mathrm{M}+\mathrm{H}] 319.1805$, found 319.1802.


4-21
(E)-(4-Phenylpiperazin-1-yl)(2-(4-(trifluoromethyl)benzylidene)cyclopropyl)methanone (421). Prepared according to General Procedure $K$ from the corresponding amide ( $0.067 \mathrm{~g}, 0.17$ $\mathrm{mmol})$ and camphorsulfonic acid $(0.006 \mathrm{~g}, 0.03 \mathrm{mmol}, 15 \mathrm{~mol} \%)$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$. Purification
by chromatography on $\mathrm{SiO}_{2}$ afforded 4-21 ( $0.051 \mathrm{~g}, 0.013 \mathrm{mmol}, 76 \%$ ) as a colorless solid: Mp $152-155{ }^{\circ} \mathrm{C},{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.58,7.45\left(\mathrm{ABq}, 4 \mathrm{H}, J_{\mathrm{AB}}=8.0 \mathrm{~Hz}\right), 7.30(\mathrm{app} \mathrm{t}, 2 \mathrm{H}$, $J=8.0 \mathrm{~Hz}), 6.99-6.89(\mathrm{~m}, 3 \mathrm{H}), 6.53(\mathrm{~s}, 1 \mathrm{H}), 3.90-3.86(\mathrm{~m}, 1 \mathrm{H}), 3.80(\mathrm{dd}, 2 \mathrm{H}, J=9.6 \mathrm{~Hz}, 4.0$ Hz ), 3.72-3.65 (m, 2 H ), 3.27-3.15 (m, 4 H$), 3.12(\mathrm{~d}, 2 \mathrm{H}, J=3.6 \mathrm{~Hz}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 170.5,151.0,145.8,136.9,129.9\left({ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 129.3,125.4\left({ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 124.9,124.1\left({ }^{1} J_{\mathrm{CF}}\right.$ $=271 \mathrm{~Hz}), 120.7,116.8,50.0,49.5,45.1,41.7,40.8,32.6 ;$ MS (ESI $\left.{ }^{+}\right) m / z 387$ (100), 225 (5); HRMS ( $\mathrm{ESI}^{+}$) $m / z$ calcd for $\mathrm{C}_{22} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{OF}_{3}[\mathrm{M}+\mathrm{H}]$ 387.1684, found 387.1674.

(E)-Phenyl(2-(4-(trifluoromethyl)benzylidene)cyclopropyl)methanone (4-22). Prepared according to General Procedure K from the corresponding amide $(0.027 \mathrm{~g}, 0.089 \mathrm{mmol})$ and camphorsulfonic acid ( $3 \mathrm{mg}, 0.013 \mathrm{mmol} \sim 15 \mathrm{~mol} \%$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1.5 \mathrm{~mL})$. Purification by chromatography on $\mathrm{SiO}_{2}$ ( $30 \% \mathrm{EtOAc} /$ hexanes) afforded amide $\mathbf{4 - 2 2}$ ( $0.008 \mathrm{~g}, 0.026 \mathrm{~mol}, 30 \%$ ) as a colorless semi-solid: ${ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 8.01(\mathrm{~d}, 2 \mathrm{H}, J=7.2 \mathrm{~Hz}), 7.65-7.56(\mathrm{~m}, 3$ H), 7.56-7.41 (m, 4 H$), 6.62(\mathrm{~s}, 1 \mathrm{H}), 4.53(\mathrm{app} \mathrm{s}, 1 \mathrm{H}), 3.28-3.13(\mathrm{~m}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right)\left({ }^{1} J_{\mathrm{CF}}\right.$ not resolved) $\delta 198.7,146.3,136.9,135.9,133.3,130.0\left({ }^{2} J_{\mathrm{CF}}=32 \mathrm{~Hz}\right), 128.8,128.2$, $125.4\left({ }^{3} J_{\mathrm{CF}}=4 \mathrm{~Hz}\right), 124.9,46.0,32.6 ; \mathrm{MS}\left(\mathrm{ESI}^{+}\right) \mathrm{m} / \mathrm{z} 303(100), 225(10), 167(30) ; \mathrm{HRMS}^{\left(\mathrm{ESI}^{+}\right)}$ $m / z$ calcd for $\mathrm{C}_{18} \mathrm{H}_{14} \mathrm{OF}_{3}[\mathrm{M}+\mathrm{H}]$ 303.0997, found 303.0989.


3-Methylene- $N$-phenylcyclobutane-1-carboxamide (4-23). A solution of amide 3-92 (0.042 g, $0.22 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})$ was treated with camphorsulfonic acid $(0.008 \mathrm{~g}, 0.034 \mathrm{mmol}, 15$ $\mathrm{mol} \%$ ). The reaction was stirred at rt for 3 h , then at $60^{\circ} \mathrm{C}$ for 14 h , and then $120^{\circ} \mathrm{C}$ for 6 h . The reaction mixture was directly purified by chromatography on $\mathrm{SiO}_{2}$ to afford methylenecyclobutane 4-23 ( $7 \mathrm{mg}, 0.04 \mathrm{mmol}, 17 \%$ ) as a colorless semi-solid: ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 7.53(\mathrm{~d}, 2$ $\mathrm{H}, J=8.1 \mathrm{~Hz}), 7.32(\mathrm{app} \mathrm{t}, 2 \mathrm{H}, J=7.8 \mathrm{~Hz}), 7.21-7.04(\mathrm{~m}, 2 \mathrm{H}), 4.84(\mathrm{~s}, 2 \mathrm{H}), 3.22-3.04(\mathrm{~m}, 3 \mathrm{H})$, 3.02-2.84 (m, 2 H ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 172.6$ (s), 143.9 (s), 137.9 (s), 129.0 (d), 124.3 (d), 119.7 (d), 106.8 ( $t$ ), 35.9 (d), 35.6 (t).

## APPENDIX A

${ }^{1} \mathrm{H},{ }^{13} \mathrm{C},{ }^{19} \mathrm{~F}$, AND ${ }^{31} \mathrm{P}$ NMR SPECTRA FOR NEW COMPOUNDS


















555-37.2, 400, CDCl 3















vi










$\left.\right|_{i} ^{\infty}$




















anti isomer

NC,

trans-2-41














## 



















anti isome


V






trans-2-55


$s-2-55$










| $\underset{\sim}{\sim}$ | $\stackrel{\circ}{\text { oे }}$ |
| :---: | :---: |
| $i$ | $\infty$ |












more polar isomer











cis-2-70
















trans-2-76





























$\underset{1}{\substack{\pi \\ ~}}$



























 pm
























\%idy io i
VVi




















































## APPENDIX B

# ATOMIC COORDINATES, BOND LENGTHS, AND BOND ANGLES FOR X-RAY CRYSTAL STRCUCTURES 



The coordinates for $\mathbf{1 - 3 6}$ were deposited in the CCDC (\#997593)
Sample and crystal data for 1-36:

Identification code
Chemical formula
Formula weight
Temperature
Wavelength
Crystal size
Crystal habit
Crystal system
Space group
milligan21314
$\mathrm{C}_{25} \mathrm{H}_{49} \mathrm{~B}_{2} \mathrm{~F}_{4} \mathrm{~N}_{2} \mathrm{P}$
506.25

180(2) K
1.54178 Å
$0.120 \times 0.150 \times 0.190 \mathrm{~mm}$
translucent colorless chunk
monoclinic
P 1 21/n 1
$\mathrm{a}=12.0541(3) \quad \alpha=90^{\circ}$
$\mathrm{b}=16.1202(4) \quad \beta=92.0953(12)^{\circ}$
$\mathrm{c}=15.0576(3) \quad \gamma=90^{\circ}$
2923.95(12)

4
$1.150 \mathrm{~g} / \mathrm{cm}^{3}$
$1.174 \mathrm{~mm}^{-1}$

Data collection and structure refinement for 1-36:

Diffractometer
Radiation source Theta range for data collection
Reflections collected
Independent reflections
Coverage of independent reflections
Absorption correction
Max. and min. transmission
Structure solution technique
Structure solution program
Refinement method
Refinement program
Function minimized
Data / restraints / parameters
Goodness-of-fit on $\mathbf{F}^{\mathbf{2}}$
$\Delta / \sigma_{\text {max }}$
Final R indices

Weighting scheme
Largest diff. peak and hole
R.M.S. deviation from mean

Bruker Apex II CCD
IMuS micro-focus source, Cu
4.02 to $68.42^{\circ}$

9574
$9574[\mathrm{R}(\mathrm{int})=0.0531]$
99.9\%
multi-scan
0.8720 and 0.8080
direct methods
SHELXS-97 (Sheldrick, 2008)
Full-matrix least-squares on $\mathrm{F}^{2}$
SHELXL-2013 (Sheldrick, 2013)
$\Sigma \mathrm{w}\left(\mathrm{F}_{\mathrm{o}}{ }^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2}$
9574/0/328
1.381
0.001

8008 data; $\quad \mathrm{R} 1=0.0468$, wR2
$\mathrm{I}>2 \sigma(\mathrm{I}) \quad=0.1300$
all data $\quad$ R1 $=0.0601$, wR2 $=0.1374$
$\mathrm{w}=1 /\left[\sigma^{2}\left(\mathrm{~F}_{\mathrm{o}}{ }^{2}\right)+(0.0680 \mathrm{P})^{2}\right]$
where $\mathrm{P}=\left(\mathrm{F}_{\mathrm{o}}{ }^{2}+2 \mathrm{~F}_{\mathrm{c}}{ }^{2}\right) / 3$
0.414 and $-0.281 \mathrm{e}^{-3}$
$0.044 \mathrm{e}^{-3}$

Atomic coordinates and equivalent isotropic atomic displacement parameters ( $\AA^{2}$ ) for 136:
$\mathrm{U}(\mathrm{eq})$ is defined as one third of the trace of the orthogonalized $\mathrm{U}_{\mathrm{ij}}$ tensor.

|  | x/a | $\mathbf{y} / \mathbf{b}$ | $\mathbf{z / c}$ | $\mathbf{~} \mathbf{U ( \mathbf { e q } )}$ |
| :--- | :---: | :---: | :--- | :--- |
| P1 | $0.73288(4)$ | $0.20888(3)$ | $0.53514(3)$ | $0.02680(15)$ |
| N1 | $0.75376(14)$ | $0.37049(11)$ | $0.60853(10)$ | $0.0330(4)$ |
| C1 | $0.96303(17)$ | $0.22337(14)$ | $0.55496(18)$ | $0.0452(6)$ |
| F1 | $0.76990(15)$ | $0.13600(10)$ | $0.78558(10)$ | $0.0766(5)$ |
| B1 | $0.7782(3)$ | $0.09909(17)$ | $0.86832(18)$ | $0.0542(8)$ |
| N2 | $0.72945(14)$ | $0.35777(10)$ | $0.45672(11)$ | $0.0325(4)$ |


|  | $\mathbf{y} / \mathbf{a}$ | $\mathbf{y} / \mathbf{b}$ | $\mathbf{z / c}$ | $\mathbf{C}(\mathbf{e q})$ |
| :--- | :--- | :--- | :--- | :--- |
| C2 | $0.07055(19)$ | $0.19561(16)$ | $0.6021(2)$ | $0.0595(7)$ |
| F2 | $0.69015(14)$ | $0.04416(9)$ | $0.87708(11)$ | $0.0687(4)$ |
| C3 | $0.08718(18)$ | $0.10302(15)$ | $0.59521(17)$ | $0.0458(6)$ |
| F3 | $0.77869(16)$ | $0.15727(9)$ | $0.93442(11)$ | $0.0794(5)$ |
| C4 | $0.98875(18)$ | $0.05588(16)$ | $0.62831(19)$ | $0.0506(6)$ |
| F4 | $0.87055(18)$ | $0.04995(13)$ | $0.87643(16)$ | $0.1005(7)$ |
| C5 | $0.88235(17)$ | $0.08340(13)$ | $0.57994(19)$ | $0.0459(6)$ |
| C6 | $0.86364(15)$ | $0.17631(12)$ | $0.59062(13)$ | $0.0293(4)$ |
| C7 | $0.5883(2)$ | $0.09087(15)$ | $0.6026(2)$ | $0.0587(8)$ |
| C8 | $0.4942(2)$ | $0.07218(18)$ | $0.6640(2)$ | $0.0656(9)$ |
| C9 | $0.39146(19)$ | $0.12236(18)$ | $0.63972(19)$ | $0.0563(7)$ |
| C10 | $0.41609(19)$ | $0.21424(17)$ | $0.6345(2)$ | $0.0552(7)$ |
| C11 | $0.51084(16)$ | $0.23259(14)$ | $0.57355(16)$ | $0.0397(5)$ |
| C12 | $0.61460(16)$ | $0.18340(12)$ | $0.60323(13)$ | $0.0309(4)$ |
| C13 | $0.73887(15)$ | $0.32474(11)$ | $0.53658(12)$ | $0.0279(4)$ |
| C14 | $0.6671(2)$ | $0.46339(14)$ | $0.34972(16)$ | $0.0455(6)$ |
| C15 | $0.6780(2)$ | $0.55434(15)$ | $0.32248(18)$ | $0.0589(7)$ |
| C16 | $0.7974(2)$ | $0.57870(16)$ | $0.31101(16)$ | $0.0569(7)$ |
| C17 | $0.8682(2)$ | $0.56057(15)$ | $0.39427(17)$ | $0.0526(6)$ |
| C18 | $0.85816(19)$ | $0.46994(14)$ | $0.42199(16)$ | $0.0438(5)$ |
| C19 | $0.73767(17)$ | $0.44611(11)$ | $0.43351(13)$ | $0.0326(4)$ |
| C20 | $0.65632(18)$ | $0.37163(14)$ | $0.74970(14)$ | $0.0389(5)$ |
| C21 | $0.66051(19)$ | $0.33682(15)$ | $0.84361(14)$ | $0.0440(5)$ |
| C22 | $0.7661(2)$ | $0.36165(14)$ | $0.89481(14)$ | $0.0461(6)$ |
| C23 | $0.86914(19)$ | $0.34039(15)$ | $0.84469(15)$ | $0.0445(6)$ |
| C24 | $0.86404(18)$ | $0.37282(14)$ | $0.74934(14)$ | $0.0399(5)$ |
| C25 | $0.75805(16)$ | $0.34242(11)$ | $0.70144(12)$ | $0.0305(4)$ |
| B2 | $0.7180(2)$ | $0.17385(16)$ | $0.41319(17)$ | $0.0408(6)$ |

## Bond lengths (i) for 1-36:

| P1-C12 | $1.8330(19)$ | P1-C6 | $1.833(2)$ |
| :--- | :--- | :--- | :--- |
| P1-C13 | $1.8691(19)$ | P1-B2 | $1.923(2)$ |
| N1-C13 | $1.318(2)$ | N1-C25 | $1.469(2)$ |
| N1-H1N | $0.80(3)$ | C1-C2 | $1.522(4)$ |
| C1-C6 | $1.532(3)$ | C1-H1A | 0.99 |
| C1-H1B | 0.99 | F1-B1 | $1.381(3)$ |
| B1-F3 | $1.367(3)$ | B1-F4 | $1.369(4)$ |


| B1-F2 | 1.392(3) | N2-C13 | 1.316(2) |
| :---: | :---: | :---: | :---: |
| N2-C19 | 1.471(2) | N2-H2N | 0.85(3) |
| C2-C3 | 1.510(3) | C2-H2A | 0.99 |
| C2-H2B | 0.99 | C3-C4 | 1.509(3) |
| C3-H3A | 0.99 | C3-H3B | 0.99 |
| C4-C5 | 1.518(3) | C4-H4A | 0.99 |
| C4-H4B | 0.99 | C5-C6 | 1.524(3) |
| C5-H5A | 0.99 | C5-H5B | 0.99 |
| C6-H6A | 1.0 | C7-C8 | 1.519(3) |
| C7-C12 | 1.525(3) | C7-H7A | 0.99 |
| C7-H7B | 0.99 | C8-C9 | 1.512(3) |
| C8-H8A | 0.99 | C8-H8B | 0.99 |
| C9-C10 | 1.513(4) | C9-H9A | 0.99 |
| C9-H9B | 0.99 | C10-C11 | 1.520 (3) |
| C10-H10A | 0.99 | C10-H10B | 0.99 |
| C11-C12 | 1.534(3) | C11-H11A | 0.99 |
| C11-H11B | 0.99 | C12-H12A | 1.0 |
| C14-C19 | 1.521(3) | C14-C15 | 1.529(3) |
| C14-H14A | 0.99 | C14-H14B | 0.99 |
| C15-C16 | 1.508(4) | C15-H15A | 0.99 |
| C15-H15B | 0.99 | C16-C17 | 1.519(4) |
| C16-H16A | 0.99 | C16-H16B | 0.99 |
| C17-C18 | 1.525(3) | C17-H17A | 0.99 |
| C17-H17B | 0.99 | C18-C19 | 1.518(3) |
| C18-H18A | 0.99 | C18-H18B | 0.99 |
| C19-H19A | 1.0 | C20-C25 | 1.523(3) |
| C20-C21 | 1.520(3) | C20-H20A | 0.99 |
| C20-H20B | 0.99 | C21-C22 | 1.517(3) |
| C21-H21A | 0.99 | C21-H21B | 0.99 |
| C22-C23 | 1.516(3) | C22-H22A | 0.99 |
| C22-H22B | 0.99 | C23-C24 | 1.527(3) |
| C23-H23A | 0.99 | C23-H23B | 0.99 |
| C24-C25 | 1.525(3) | C24-H24A | 0.99 |
| C24-H24B | 0.99 | C25-H25A | 1.0 |
| B2-H26A | 1.06(3) | B2-H26B | 1.07(2) |
| B2-H26C | 1.09(3) |  |  |

## Bond angles $\left({ }^{\circ}\right)$ for 1-36:

| C12-P1-C6 | 110.75(9) | C12-P1-C13 | 104.36(9) |
| :---: | :---: | :---: | :---: |
| C6-P1-C13 | 104.40(9) | C12-P1-B2 | 114.88(11) |
| C6-P1-B2 | 113.56(11) | C13-P1-B2 | 107.86(10) |
| C13-N1-C25 | 127.55(16) | C13-N1-H1N | 116.3(18) |
| C25-N1-H1N | 116.1(18) | C2-C1-C6 | 110.8(2) |
| C2-C1-H1A | 109.5 | C6-C1-H1A | 109.5 |
| C2-C1-H1B | 109.5 | C6-C1-H1B | 109.5 |
| H1A-C1-H1B | 108.1 | F3-B1-F4 | 110.5(3) |
| F3-B1-F1 | 111.1(2) | F4-B1-F1 | 111.2(2) |
| F3-B1-F2 | 110.5(2) | F4-B1-F2 | 104.1(2) |
| F1-B1-F2 | 109.2(2) | C13-N2-C19 | 127.20(17) |
| C13-N2-H2N | 114.6(16) | C19-N2-H2N | 117.8(16) |
| C3-C2-C1 | 111.81(19) | C3-C2-H2A | 109.3 |
| C1-C2-H2A | 109.3 | C3-C2-H2B | 109.3 |
| C1-C2-H2B | 109.3 | H2A-C2-H2B | 107.9 |
| C2-C3-C4 | 111.54(19) | C2-C3-H3A | 109.3 |
| C4-C3-H3A | 109.3 | C2-C3-H3B | 109.3 |
| C4-C3-H3B | 109.3 | H3A-C3-H3B | 108.0 |
| C3-C4-C5 | 110.8(2) | C3-C4-H4A | 109.5 |
| C5-C4-H4A | 109.5 | C3-C4-H4B | 109.5 |
| C5-C4-H4B | 109.5 | H4A-C4-H4B | 108.1 |
| C4-C5-C6 | 111.24(19) | C4-C5-H5A | 109.4 |
| C6-C5-H5A | 109.4 | C4-C5-H5B | 109.4 |
| C6-C5-H5B | 109.4 | H5A-C5-H5B | 108.0 |
| C5-C6-C1 | 109.21(16) | C5-C6-P1 | 111.19(14) |
| C1-C6-P1 | 111.65(16) | C5-C6-H6A | 108.2 |
| C1-C6-H6A | 108.2 | P1-C6-H6A | 108.2 |
| C8-C7-C12 | 110.5(2) | C8-C7-H7A | 109.6 |
| C12-C7-H7A | 109.6 | C8-C7-H7B | 109.6 |
| C12-C7-H7B | 109.6 | H7A-C7-H7B | 108.1 |
| C9-C8-C7 | 111.7(2) | C9-C8-H8A | 109.3 |
| C7-C8-H8A | 109.3 | C9-C8-H8B | 109.3 |
| C7-C8-H8B | 109.3 | H8A-C8-H8B | 107.9 |
| C8-C9-C10 | 112.1(2) | C8-C9-H9A | 109.2 |
| C10-C9-H9A | 109.2 | C8-C9-H9B | 109.2 |
| C10-C9-H9B | 109.2 | H9A-C9-H9B | 107.9 |
| C9-C10-C11 | 112.0(2) | C9-C10-H10A | 109.2 |
| C11-C10-H10A | 109.2 | C9-C10-H10B | 109.2 |


| C11-C10-H10B | 109.2 | H10A-C10-H10B | 107.9 |
| :---: | :---: | :---: | :---: |
| C10-C11-C12 | 110.3(2) | C10-C11-H11A | 109.6 |
| C12-C11-H11A | 109.6 | C10-C11-H11B | 109.6 |
| C12-C11-H11B | 109.6 | H11A-C11-H11B | 108.1 |
| C7-C12-C11 | 109.67(17) | C7-C12-P1 | 112.44(15) |
| C11-C12-P1 | 111.34(15) | C7-C12-H12A | 107.7 |
| C11-C12-H12A | 107.7 | P1-C12-H12A | 107.7 |
| N1-C13-N2 | 121.96(17) | N1-C13-P1 | 124.96(14) |
| N2-C13-P1 | 113.05(13) | C19-C14-C15 | 110.30(19) |
| C19-C14-H14A | 109.6 | C15-C14-H14A | 109.6 |
| C19-C14-H14B | 109.6 | C15-C14-H14B | 109.6 |
| H14A-C14-H14B | 108.1 | C16-C15-C14 | 111.9(2) |
| C16-C15-H15A | 109.2 | C14-C15-H15A | 109.2 |
| C16-C15-H15B | 109.2 | C14-C15-H15B | 109.2 |
| H15A-C15-H15B | 107.9 | C15-C16-C17 | 111.4(2) |
| C15-C16-H16A | 109.3 | C17-C16-H16A | 109.3 |
| C15-C16-H16B | 109.3 | C17-C16-H16B | 109.3 |
| H16A-C16-H16B | 108.0 | C16-C17-C18 | 111.2(2) |
| C16-C17-H17A | 109.4 | C18-C17-H17A | 109.4 |
| C16-C17-H17B | 109.4 | C18-C17-H17B | 109.4 |
| H17A-C17-H17B | 108.0 | C19-C18-C17 | 111.07(19) |
| C19-C18-H18A | 109.4 | C17-C18-H18A | 109.4 |
| C19-C18-H18B | 109.4 | C17-C18-H18B | 109.4 |
| H18A-C18-H18B | 108.0 | N2-C19-C18 | 110.20(16) |
| N2-C19-C14 | 109.51(17) | C18-C19-C14 | 111.53(17) |
| N2-C19-H19A | 108.5 | C18-C19-H19A | 108.5 |
| C14-C19-H19A | 108.5 | C25-C20-C21 | 109.25(17) |
| C25-C20-H20A | 109.8 | C21-C20-H20A | 109.8 |
| C25-C20-H20B | 109.8 | C21-C20-H20B | 109.8 |
| H20A-C20-H20B | 108.3 | C22-C21-C20 | 111.96(19) |
| C22-C21-H21A | 109.2 | C20-C21-H21A | 109.2 |
| C22-C21-H21B | 109.2 | C20-C21-H21B | 109.2 |
| H21A-C21-H21B | 107.9 | C21-C22-C23 | 112.01(19) |
| C21-C22-H22A | 109.2 | C23-C22-H22A | 109.2 |
| C21-C22-H22B | 109.2 | C23-C22-H22B | 109.2 |
| H22A-C22-H22B | 107.9 | C22-C23-C24 | 112.64(18) |
| C22-C23-H23A | 109.1 | C24-C23-H23A | 109.1 |
| C22-C23-H23B | 109.1 | C24-C23-H23B | 109.1 |


| H23A-C23-H23B | 107.8 | C25-C24-C23 | $109.81(17)$ |
| :--- | :--- | :--- | :--- |
| C25-C24-H24A | 109.7 | C23-C24-H24A | 109.7 |
| C25-C24-H24B | 109.7 | C23-C24-H24B | 109.7 |
| H24A-C24-H24B | 108.2 | N1-C25-C20 | $110.99(16)$ |
| N1-C25-C24 | $110.55(16)$ | C20-C25-C24 | $110.50(17)$ |
| N1-C25-H25A | 108.2 | C20-C25-H25A | 108.2 |
| C24-C25-H25A | 108.2 | P1-B2-H26A | $105.2(15)$ |
| P1-B2-H26B | $107.4(14)$ | H26A-B2-H26B | $114 .(2)$ |
| P1-B2-H26C | $103.1(14)$ | H26A-B2-H26C | $110.5(19)$ |
| H26B-B2-H26C | $115.4(19)$ |  |  |



The X-ray coordinates for 1-42 have been deposited in the CCDC (\#1008171)

Sample and crystal data for 1-42

| Identification code | Milligan57743 |  |
| :--- | :--- | :--- |
| Chemical formula | $\mathrm{C}_{23} \mathrm{H}_{34} \mathrm{BN}_{2} \mathrm{P}$ |  |
| Formula weight | 380.30 |  |
| Temperature | $220(2) \mathrm{K}$ |  |
| Wavelength | $1.54178 \AA$ |  |
| Crystal size | $0.040 \times 0.120 \times 0.210 \mathrm{~mm}$ |  |
| Crystal habit | clear colourless shard |  |
| Crystal system | triclinic |  |
| Space group | $\mathrm{P}-1$ | $\alpha=93.950(3)^{\circ}$ |
| Unit cell dimensions | $\mathrm{a}=11.3394(5) \AA$ | $\beta=90.335(2)^{\circ}$ |
|  | $\mathrm{b}=13.9575(7) \AA$ |  |
|  | $\mathrm{c}=14.6789(8) \AA$ | $\gamma=104.560(2)^{\circ}$ |
| Volume | $2242.63(19) \AA^{3}$ |  |
| Z | 4 |  |
| Density (calculated) | $1.126 \mathrm{~g} / \mathrm{cm}^{3}$ |  |
| Absorption | $1.134 \mathrm{~mm}^{-1}$ |  |
| coefficient | 824 |  |
| F(000) |  |  |

Data collection and structure refinement for 1-42:

| Diffractometer | Bruker Apex II CCD |
| :---: | :---: |
| Radiation source | IMuS micro-focus, Cu |
| Theta range for data collection | 3.02 to $68.37^{\circ}$ |
| Index ranges | $\begin{aligned} & -13<=\mathrm{h}<=12, \quad-16<=\mathrm{k}<=16, \\ & 16<=\mathrm{l}<=15 \end{aligned}$ |
| Reflections collected | 30798 |
| Independent reflections | 7969 [ $\mathrm{R}(\mathrm{int})=0.0330]$ |
| Coverage independent of reflections | 96.6\% |
| Absorption correction | multi-scan |
| Max. and min. transmission | 0.9560 and 0.7970 |
| Structure solution technique | direct methods |
| Structure solution program | direct methods |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{2}$ |
| Refinement program | SHELXL-2013 (Sheldrick, 2013) |
| Function minimized | $\Sigma \mathrm{w}\left(\mathrm{F}_{\mathrm{o}}{ }^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2}$ |
| Data / restraints / parameters | 7969 / 0 / 519 |
| Goodness-of-fit on $\mathbf{F}^{\mathbf{2}}$ | 1.904 |
| $\Delta / \sigma_{\text {max }}$ | 0.012 |
| Final R indices | $\begin{array}{ll} 6625 \text { data; } & \mathrm{R} 1=0.0628 \\ \mathrm{I}>2 \sigma(\mathrm{I}) & \mathrm{wR} 2=0.1796 \end{array}$ |
|  | $\begin{array}{ll} \text { all data } & \mathrm{R} 1=0.0743, \\ & w R 2=0.1853 \end{array}$ |
| Weighting scheme | $\begin{aligned} & \mathrm{w}=1 /\left[\sigma^{2}\left(\mathrm{~F}_{\mathrm{o}}^{2}\right)+(0.0680 \mathrm{P})^{2}\right] \\ & \text { where } \mathrm{P}=\left(\mathrm{F}_{\mathrm{o}}^{2}+2 \mathrm{~F}_{\mathrm{c}}^{2}\right) / 3 \end{aligned}$ |
| Largest diff. peak and hole | 1.684 and -0.696 $\mathrm{e}^{-3}{ }^{-3}$ |
| R.M.S. deviation from mean | $0.050 \mathrm{e}^{\text {® }}{ }^{-3}$ |

Atomic coordinates and equivalent isotropic atomic displacement parameters ( $\AA^{\mathbf{2}}$ ) for 1-42:
$\mathrm{U}(\mathrm{eq})$ is defined as one third of the trace of the orthogonalized $\mathrm{U}_{\mathrm{ij}}$ tensor.

|  | x/a | y/b | z/c | U(eq) |
| :---: | :---: | :---: | :---: | :---: |
| P1 | 0.04228(5) | 0.27378(5) | 0.19476(4) | 0.03658(19) |
| B1 | 0.0807(3) | 0.2623(4) | 0.3220(2) | 0.0622(10) |
| N1 | 0.84027(17) | 0.28302(15) | 0.10241(13) | 0.0379(5) |
| C1 | 0.7229(2) | 0.29548(19) | 0.08021(15) | 0.0368(5) |
| P2 | 0.30261(6) | 0.24570(6) | 0.68762(5) | 0.0467(2) |
| B2 | 0.2561(3) | 0.2196(4) | 0.8149(2) | 0.0649(11) |
| N2 | 0.8137(2) | 0.2451(2) | 0.25756(16) | 0.0504(6) |
| C2 | 0.7094(3) | 0.3913(2) | 0.07953(18) | 0.0500(7) |
| N3 | 0.51351(18) | 0.27361 (16) | 0.59653(14) | 0.0393(5) |
| C3 | 0.6027(3) | 0.4095(3) | 0.0480(2) | 0.0659(9) |
| N4 | 0.5185(2) | 0.22477(18) | 0.74848(16) | 0.0458(5) |
| C4 | 0.5082(3) | 0.3309(3) | 0.0165(2) | 0.0707(10) |
| C5 | 0.5204(3) | 0.2360(3) | 0.01772(19) | 0.0612(9) |
| C6 | 0.6271(2) | 0.2154(2) | 0.04900(17) | 0.0461(6) |
| C7 | 0.6374(3) | 0.1114(2) | 0.0540(2) | 0.0676(9) |
| C8 | 0.6444(3) | 0.0812(2) | 0.1510(3) | 0.0739(10) |
| C9 | 0.6032(3) | 0.1471(2) | 0.2224(2) | 0.0530(7) |
| C10 | 0.4806(3) | 0.1347(3) | 0.2402(3) | 0.0711(11) |
| C11 | 0.4418(3) | 0.1955(4) | 0.3033(3) | 0.0874(14) |
| C12 | 0.5221(3) | 0.2709(4) | 0.3504(2) | 0.0801(12) |
| C13 | 0.6448(3) | 0.2877(3) | 0.33317(19) | 0.0595(8) |
| C14 | 0.6851(2) | 0.2275(2) | 0.26948(18) | 0.0458(6) |
| C15 | 0.8769(2) | 0.26534(16) | 0.18019(15) | 0.0331(5) |
| C16 | 0.0720(2) | 0.1689(2) | 0.12064(18) | 0.0463(6) |
| C17 | 0.0172(3) | 0.1576(3) | 0.0236(2) | 0.0687(9) |
| C18 | 0.0142(4) | 0.0749(3) | 0.1682(3) | 0.0854(12) |
| C19 | 0.2089(3) | 0.1806(3) | 0.1142(3) | 0.0829(12) |
| C20 | 0.1182(3) | 0.3997(2) | 0.1590(2) | 0.0573(8) |
| C21 | 0.0544(3) | 0.4723(2) | 0.2083(3) | 0.0756(10) |
| C22 | 0.1134(3) | 0.4100(3) | 0.0568(3) | 0.0818(12) |
| C23 | 0.2534(3) | $0.4264(3)$ | 0.1934(3) | $0.0835(11)$ |
| C24 | 0.6402(2) | 0.29546(17) | 0.57736(15) | 0.0343(5) |
| C25 | 0.7057(3) | 0.3939(2) | 0.58237(18) | 0.0485(6) |
| C26 | 0.8255(3) | 0.4202(2) | 0.5566(2) | 0.0609(8) |
| C27 | 0.8826(3) | 0.3477(2) | 0.52397(19) | 0.0558(8) |
| C28 | 0.8167(2) | 0.2501(2) | 0.51772(16) | 0.0439(6) |
| C29 | 0.6956(2) | 0.22242(18) | 0.54304(15) | 0.0350(5) |


|  | $\mathbf{x} / \mathbf{a}$ | $\mathbf{y} / \mathbf{b}$ | $\mathbf{z} / \mathbf{c}$ | $\mathbf{U}(\mathbf{e q})$ |
| :--- | :--- | :--- | :--- | :--- |
| C30 | $0.6267(2)$ | $0.11453(19)$ | $0.53692(17)$ | $0.0453(6)$ |
| C31 | $0.5911(2)$ | $0.07151(19)$ | $0.62884(19)$ | $0.0466(6)$ |
| C32 | $0.6682(2)$ | $0.12844(17)$ | $0.70757(16)$ | $0.0366(5)$ |
| C33 | $0.7793(2)$ | $0.1097(2)$ | $0.72948(18)$ | $0.0443(6)$ |
| C34 | $0.8534(2)$ | $0.1631(2)$ | $0.80008(19)$ | $0.0524(7)$ |
| C35 | $0.8195(3)$ | $0.2372(2)$ | $0.84994(19)$ | $0.0529(7)$ |
| C36 | $0.7109(2)$ | $0.2588(2)$ | $0.83046(17)$ | $0.0463(6)$ |
| C37 | $0.6347(2)$ | $0.20477(19)$ | $0.76007(17)$ | $0.0389(6)$ |
| C38 | $0.4661(2)$ | $0.24739(17)$ | $0.67143(16)$ | $0.0349(5)$ |
| C39 | $0.2962(3)$ | $0.3734(3)$ | $0.6635(2)$ | $0.0671(9)$ |
| C40 | $0.4026(4)$ | $0.4444(3)$ | $0.7203(3)$ | $0.0889(13)$ |
| C41 | $0.3069(4)$ | $0.3917(3)$ | $0.5619(3)$ | $0.0809(11)$ |
| C42 | $0.1752(4)$ | $0.3913(3)$ | $0.7022(3)$ | $0.1004(15)$ |
| C43 | $0.2135(2)$ | $0.1441(2)$ | $0.60545(18)$ | $0.0438(6)$ |
| C44 | $0.2623(3)$ | $0.1411(2)$ | $0.50951(19)$ | $0.0572(7)$ |
| C45 | $0.0815(3)$ | $0.1530(3)$ | $0.6005(2)$ | $0.0788(11)$ |
| C46 | $0.2177(3)$ | $0.0470(2)$ | $0.6468(2)$ | $0.0691(9)$ |

## Bond lengths ( $\AA$ ) for 1-42:

| P1-C15 | $1.860(2)$ | P1-C16 | $1.864(3)$ |
| :--- | :--- | :--- | :--- |
| P1-C20 | $1.865(3)$ | P1-B1 | $1.942(3)$ |
| B1-H3B | $1.11(4)$ | B1-H1B | $1.12(3)$ |
| B1-H1B | $1.14(4)$ | N1-C15 | $1.273(3)$ |
| N1-C1 | $1.423(3)$ | C1-C2 | $1.385(4)$ |
| C1-C6 | $1.397(3)$ | P2-C39 | $1.861(3)$ |
| P2-C38 | $1.865(2)$ | P2-C43 | $1.875(3)$ |
| P2-B2 | $1.978(4)$ | B2-H2B | $1.07(4)$ |
| B2-H2C | $1.15(4)$ | B2-H2D | $1.04(5)$ |
| N2-C15 | $1.355(3)$ | N2-C14 | $1.430(3)$ |
| N2-H2N | $0.79(4)$ | C2-C3 | $1.380(4)$ |
| C2-H2A | 0.94 | N3-C38 | $1.263(3)$ |
| N3-C24 | $1.425(3)$ | C3-C4 | $1.377(5)$ |
| C3-H3A | 0.94 | N4-C38 | $1.367(3)$ |
| N4-C37 | $1.425(3)$ | N4-H4N | $0.87(4)$ |
| C4-C5 | $1.368(5)$ | C4-H4A | 0.94 |
| C5-C6 | $1.394(4)$ | C5-H5A | 0.94 |
| C6-C7 | $1.493(5)$ | C7-C8 | $1.520(5)$ |


| C7-H7A | 0.98 | C7-H7B | 0.98 |
| :---: | :---: | :---: | :---: |
| C8-C9 | 1.502(5) | C8-H8A | 0.98 |
| C8-H8B | 0.98 | C9-C10 | 1.386(4) |
| C9-C14 | 1.400(4) | C10-C11 | 1.362(6) |
| C10-H10A | 0.94 | C11-C12 | 1.352(6) |
| C11-H11A | 0.94 | C12-C13 | 1.379(5) |
| C12-H12A | 0.94 | C13-C14 | 1.371(4) |
| C13-H13A | 0.94 | C16-C18 | 1.529(5) |
| C16-C19 | 1.524(4) | C16-C17 | 1.533(4) |
| C17-H17A | 0.97 | C17-H17B | 0.97 |
| C17-H17C | 0.97 | C18-H18A | 0.97 |
| C18-H18B | 0.97 | C18-H18C | 0.97 |
| C19-H19A | 0.97 | C19-H19B | 0.97 |
| C19-H19C | 0.97 | C20-C22 | 1.519(5) |
| C20-C21 | 1.532(5) | C20-C23 | 1.557(4) |
| C21-H21A | 0.97 | C21-H21B | 0.97 |
| C21-H21C | 0.97 | C22-H22A | 0.97 |
| C22-H22B | 0.97 | C22-H22C | 0.97 |
| C23-H23A | 0.97 | C23-H23B | 0.97 |
| C23-H23C | 0.97 | $\mathrm{C} 24-\mathrm{C} 25$ | 1.385(4) |
| C24-C29 | 1.394(3) | C25-C26 | 1.378(4) |
| C25-H25A | 0.94 | C26-C27 | 1.392(5) |
| C26-H26A | 0.94 | C27-C28 | 1.376(4) |
| C27-H27A | 0.94 | C28-C29 | 1.390(3) |
| C28-H28A | 0.94 | C29-C30 | 1.508(4) |
| C30-C31 | 1.530(4) | C30-H30A | 0.98 |
| C30-H30B | 0.98 | C31-C32 | 1.498(3) |
| C31-H31A | 0.98 | C31-H31B | 0.98 |
| C32-C33 | 1.390(3) | C32-C37 | 1.403(4) |
| C33-C34 | 1.380(4) | C33-H33A | 0.94 |
| C34-C35 | 1.361(4) | C34-H34A | 0.94 |
| C35-C36 | 1.373(4) | C35-H35A | 0.94 |
| C36-C37 | 1.392(4) | C36-H36A | 0.94 |
| C39-C41 | 1.531(5) | C39-C40 | 1.552(5) |
| C39-C42 | 1.559(5) | C40-H40A | 0.97 |
| C40-H40B | 0.97 | C40-H40C | 0.97 |
| C41-H41A | 0.97 | C41-H41B | 0.97 |
| C41-H41C | 0.97 | C42-H42A | 0.97 |


| C42-H42B | 0.97 | C42-H42C | 0.97 |
| :--- | :--- | :--- | :--- |
| C43-C44 | $1.518(4)$ | C43-C46 | $1.535(4)$ |
| C43-C45 | $1.535(4)$ | C44-H44A | 0.97 |
| C44-H44B | 0.97 | C44-H44C | 0.97 |
| C45-H45A | 0.97 | C45-H45B | 0.97 |
| C45-H45C | 0.97 | C46-H46A | 0.97 |
| C46-H46B | 0.97 | C46-H46C | 0.97 |

## Bond angles $\left({ }^{\circ}\right)$ for 1-42:

| C15-P1-C16 | $105.90(11)$ | C15-P1-C20 | $103.98(12)$ |
| :--- | :--- | :--- | :--- |
| C16-P1-C20 | $114.82(14)$ | C15-P1-B1 | $109.86(14)$ |
| C16-P1-B1 | $111.07(16)$ | C20-P1-B1 | $110.78(18)$ |
| P1-B1-H3B | $104 .(2)$ | P1-B1-H1B | $103.2(16)$ |
| H3B-B1-H1B | $118 .(3)$ | P1-B1-H1B | $107.1(18)$ |
| H3B-B1-H1B | $115 .(3)$ | H1B-B1-H1B | $108 .(3)$ |
| C15-N1-C1 | $126.5(2)$ | C2-C1-C6 | $119.7(2)$ |
| C2-C1-N1 | $118.0(2)$ | C6-C1-N1 | $121.9(2)$ |
| C39-P2-C38 | $103.44(14)$ | C39-P2-C43 | $115.25(14)$ |
| C38-P2-C43 | $105.99(11)$ | C39-P2-B2 | $110.39(19)$ |
| C38-P2-B2 | $110.43(13)$ | C43-P2-B2 | $110.94(17)$ |
| P2-B2-H2B | $103 .(2)$ | P2-B2-H2C | $110 .(2)$ |
| H2B-B2-H2C | $112 .(3)$ | P2-B2-H2D | $105 .(2)$ |
| H2B-B2-H2D | $115 .(3)$ | H2C-B2-H2D | $112 .(3)$ |
| C15-N2-C14 | $128.2(2)$ | C15-N2-H2N | $113 .(3)$ |
| C14-N2-H2N | $119 .(3)$ | C3-C2-C1 | $121.2(3)$ |
| C3-C2-H2A | 119.4 | C1-C2-H2A | 119.4 |
| C38-N3-C24 | $126.07(19)$ | C4-C3-C2 | $119.4(3)$ |
| C4-C3-H3A | 120.3 | C2-C3-H3A | 120.3 |
| C38-N4-C37 | $129.0(2)$ | C38-N4-H4N | $117 .(2)$ |
| C37-N4-H4N | $113 .(2)$ | C5-C4-C3 | $119.9(3)$ |
| C5-C4-H4A | 120.1 | C3-C4-H4A | 120.1 |
| C4-C5-C6 | $121.9(3)$ | C4-C5-H5A | 119.0 |
| C6-C5-H5A | 119.0 | C5-C6-C1 | $117.9(3)$ |
| C5-C6-C7 | $121.5(3)$ | C1-C6-C7 | $120.5(2)$ |
| C6-C7-C8 | $113.7(3)$ | C6-C7-H7A | 108.8 |
| C8-C7-H7A | 108.8 | C6-C7-H7B | 108.8 |


| C8-C7-H7B | 108.8 | H7A-C7-H7B | 107.7 |
| :---: | :---: | :---: | :---: |
| C9-C8-C7 | 114.4(3) | C9-C8-H8A | 108.7 |
| C7-C8-H8A | 108.7 | C9-C8-H8B | 108.7 |
| C7-C8-H8B | 108.7 | H8A-C8-H8B | 107.6 |
| C10-C9-C14 | 117.0(3) | C10-C9-C8 | 120.9(3) |
| C14-C9-C8 | 121.9(3) | C11-C10-C9 | 121.5(3) |
| C11-C10-H10A | 119.2 | C9-C10-H10A | 119.2 |
| C12-C11-C10 | 120.8(3) | C12-C11-H11A | 119.6 |
| C10-C11-H11A | 119.6 | C11-C12-C13 | 119.6(4) |
| C11-C12-H12A | 120.2 | C13-C12-H12A | 120.2 |
| C14-C13-C12 | 120.2(3) | C14-C13-H13A | 119.9 |
| C12-C13-H13A | 119.9 | C13-C14-C9 | 120.8(3) |
| C13-C14-N2 | 117.9(3) | C9-C14-N2 | 121.2(3) |
| N1-C15-N2 | 129.5(2) | N1-C15-P1 | 117.10(18) |
| N2-C15-P1 | 113.31(16) | C18-C16-C19 | 109.6(3) |
| C18-C16-C17 | 108.3(3) | C19-C16-C17 | 108.5(3) |
| C18-C16-P1 | 105.8(2) | C19-C16-P1 | 109.7(2) |
| C17-C16-P1 | 114.80(19) | C16-C17-H17A | 109.5 |
| C16-C17-H17B | 109.5 | H17A-C17-H17B | 109.5 |
| C16-C17-H17C | 109.5 | H17A-C17-H17C | 109.5 |
| H17B-C17-H17C | 109.5 | C16-C18-H18A | 109.5 |
| C16-C18-H18B | 109.5 | H18A-C18-H18B | 109.5 |
| C16-C18-H18C | 109.5 | H18A-C18-H18C | 109.5 |
| H18B-C18-H18C | 109.5 | C16-C19-H19A | 109.5 |
| C16-C19-H19B | 109.5 | H19A-C19-H19B | 109.5 |
| C16-C19-H19C | 109.5 | H19A-C19-H19C | 109.5 |
| H19B-C19-H19C | 109.5 | C22-C20-C21 | 109.4(3) |
| C22-C20-C23 | 109.5(3) | C21-C20-C23 | 108.3(3) |
| C22-C20-P1 | 114.6(2) | C21-C20-P1 | 106.7(2) |
| C23-C20-P1 | 108.2(2) | C20-C21-H21A | 109.5 |
| C20-C21-H21B | 109.5 | H21A-C21-H21B | 109.5 |
| C20-C21-H21C | 109.5 | H21A-C21-H21C | 109.5 |
| H21B-C21-H21C | 109.5 | C20-C22-H22A | 109.5 |
| C20-C22-H22B | 109.5 | H22A-C22-H22B | 109.5 |
| C20-C22-H22C | 109.5 | H22A-C22-H22C | 109.5 |
| H22B-C22-H22C | 109.5 | C20-C23-H23A | 109.5 |
| C20-C23-H23B | 109.5 | H23A-C23-H23B | 109.5 |
| C20-C23-H23C | 109.5 | H23A-C23-H23C | 109.5 |


| H23B-C23-H23C | 109.5 | C25-C24-C29 | 119.2(2) |
| :---: | :---: | :---: | :---: |
| C25-C24-N3 | 118.4(2) | C29-C24-N3 | 121.9(2) |
| C26-C25-C24 | 121.0(3) | C26-C25-H25A | 119.5 |
| C24-C25-H25A | 119.5 | C25-C26-C27 | 120.2(3) |
| C25-C26-H26A | 119.9 | C27-C26-H26A | 119.9 |
| C28-C27-C26 | 118.6(3) | C28-C27-H27A | 120.7 |
| C26-C27-H27A | 120.7 | C27-C28-C29 | 121.8(3) |
| C27-C28-H28A | 119.1 | C29-C28-H28A | 119.1 |
| C28-C29-C24 | 119.0(2) | C28-C29-C30 | 120.5(2) |
| C24-C29-C30 | 120.5(2) | C29-C30-C31 | 114.9(2) |
| C29-C30-H30A | 108.6 | C31-C30-H30A | 108.6 |
| C29-C30-H30B | 108.6 | C31-C30-H30B | 108.6 |
| H30A-C30-H30B | 107.5 | C32-C31-C30 | 113.4(2) |
| C32-C31-H31A | 108.9 | C30-C31-H31A | 108.9 |
| C32-C31-H31B | 108.9 | C30-C31-H31B | 108.9 |
| H31A-C31-H31B | 107.7 | C33-C32-C37 | 117.1(2) |
| C33-C32-C31 | 120.8(2) | C37-C32-C31 | 122.1(2) |
| C34-C33-C32 | 121.8(2) | C34-C33-H33A | 119.1 |
| C32-C33-H33A | 119.1 | C35-C34-C33 | 120.3(2) |
| C35-C34-H34A | 119.9 | C33-C34-H34A | 119.9 |
| C34-C35-C36 | 120.1(3) | C34-C35-H35A | 120.0 |
| C36-C35-H35A | 120.0 | C35-C36-C37 | 120.2(3) |
| C35-C36-H36A | 119.9 | C37-C36-H36A | 119.9 |
| C36-C37-C32 | 120.6(2) | C36-C37-N4 | 117.5(2) |
| C32-C37-N4 | 121.7(2) | N3-C38-N4 | 129.8(2) |
| N3-C38-P2 | 118.19(17) | N4-C38-P2 | 111.91(18) |
| C41-C39-C40 | 111.1(3) | C41-C39-C42 | 111.3(3) |
| C40-C39-C42 | 107.3(3) | C41-C39-P2 | 113.0(2) |
| C40-C39-P2 | 105.8(2) | C42-C39-P2 | 108.0(3) |
| C39-C40-H40A | 109.5 | C39-C40-H40B | 109.5 |
| H40A-C40-H40B | 109.5 | C39-C40-H40C | 109.5 |
| H40A-C40-H40C | 109.5 | H40B-C40-H40C | 109.5 |
| C39-C41-H41A | 109.5 | C39-C41-H41B | 109.5 |
| H41A-C41-H41B | 109.5 | C39-C41-H41C | 109.5 |
| H41A-C41-H41C | 109.5 | H41B-C41-H41C | 109.5 |
| C39-C42-H42A | 109.5 | C39-C42-H42B | 109.5 |
| H42A-C42-H42B | 109.5 | C39-C42-H42C | 109.5 |
| H42A-C42-H42C | 109.5 | H42B-C42-H42C | 109.5 |


| C44-C43-C46 | $107.8(3)$ | C44-C43-C45 | $109.2(2)$ |
| :--- | :--- | :--- | :--- |
| C46-C43-C45 | $109.9(3)$ | C44-C43-P2 | $115.82(19)$ |
| C46-C43-P2 | $105.41(19)$ | C45-C43-P2 | $108.5(2)$ |
| C43-C44-H44A | 109.5 | C43-C44-H44B | 109.5 |
| H44A-C44-H44B | 109.5 | C43-C44-H44C | 109.5 |
| H44A-C44-H44C | 109.5 | H44B-C44-H44C | 109.5 |
| C43-C45-H45A | 109.5 | C43-C45-H45B | 109.5 |
| H45A-C45-H45B | 109.5 | C43-C45-H45C | 109.5 |
| H45A-C45-H45C | 109.5 | H45B-C45-H45C | 109.5 |
| C43-C46-H46A | 109.5 | C43-C46-H46B | 109.5 |
| H46A-C46-H46B | 109.5 | C43-C46-H46C | 109.5 |
| H46A-C46-H46C | 109.5 | H46B-C46-H46C | 109.5 |




Sample and crystal data for 1-44:

| Identification code | milligan62414 |
| :---: | :---: |
| Chemical formula | $\mathrm{C}_{29} \mathrm{H}_{30} \mathrm{BN}_{2} \mathrm{P}$ |
| Formula weight | 448.33 |
| Temperature | 230(2) K |
| Wavelength | 1.54178 A |
| Crystal size | $0.080 \times 0.120 \times 0.210 \mathrm{~mm}$ |
| Crystal system | monoclinic |
| Space group | P $121 / \mathrm{c} 1$ |
| Unit cell dimensions | $a=9.9801(4) \AA \quad \alpha=90^{\circ}$ |
|  | $\mathrm{b}=10.0187(4) \quad \beta=96.4999(14)^{\circ}$ |
|  | $\mathrm{c}=26.0554(9) \quad \gamma=90^{\circ}$ |
| Volume | 2588.47(17) $\AA$ |
| Z | 4 |
| Density (calculated) | $1.150 \mathrm{~g} / \mathrm{cm}^{3}$ |
| Absorption coefficient | $1.065 \mathrm{~mm}^{-1}$ |
| F(000) | 952 |

Data collection and structure refinement for 1-44:

| Diffractometer | Bruker Apex II CCD |
| :---: | :---: |
| Radiation source | IMuS micro-focus, Cu |
| Theta range for data collection | 3.41 to $68.18^{\circ}$ |
| Index ranges | $-11<=\mathrm{h}<=11,-12<=\mathrm{k}<=11,-30<=1<=31$ |
| Reflections collected | 27146 |
| Independent reflections | $4683[\mathrm{R}(\mathrm{int})=0.0273]$ |
| Max. and min. transmission | 0.9200 and 0.8070 |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{2}$ |
| Refinement program | SHELXL-2013 (Sheldrick, 2013) |
| Function minimized | $\Sigma \mathrm{w}\left(\mathrm{Fo}^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2}$ |
| Data / restraints / parameters | 4683 / 0 / 396 |
| Goodness-of-fit on $\mathrm{F}^{2}$ | 1.551 |
| $\Delta / \sigma_{\text {max }}$ | 0.006 |
| Final R indices | $4271 \text { data; } \mathrm{I}>2 \sigma(\mathrm{I}) \quad \mathrm{R} 1=0.0449,$ |
|  | $\begin{array}{ll} \text { all data } & \mathrm{R} 1=0.0481, \\ \mathrm{wR} 2=0.1467 \end{array}$ |
| Weighting scheme | $\begin{aligned} & \mathrm{w}=1 /\left[\sigma^{2}\left(\mathrm{~F}_{\mathrm{o}}^{2}\right)+(0.0814 \mathrm{P})^{2}\right] \\ & \mathrm{P}=\left(\mathrm{F}_{\mathrm{o}}^{2}+2 \mathrm{~F}_{\mathrm{c}}^{2}\right) / 3 \end{aligned}$ |
| Largest diff. peak and hole | 0.416 and $-0.225 \mathrm{e}^{-}{ }^{-3}$ |
| R.M.S. deviation from mean | $0.047 \mathrm{e}^{-3}{ }^{-3}$ |

Table 3. Atomic coordinates and equivalent isotropic atomic displacement parameters ( $\AA^{\mathbf{2}}$ ) for $\mathbf{1 - 4 4}$ :
$\mathrm{U}(\mathrm{eq})$ is defined as one third of the trace of the orthogonalized $\mathrm{U}_{\mathrm{ij}}$ tensor.

|  | $\mathbf{~} / \mathbf{a}$ | $\mathbf{y} / \mathbf{b}$ | $\mathbf{z} / \mathbf{c}$ | $\mathbf{~} \mathbf{U ( \mathbf { e q } )}$ |
| :--- | :--- | :--- | :--- | :--- |
| P1 | $0.78683(3)$ | $0.88794(4)$ | $0.82935(2)$ | $0.04084(16)$ |
| N1 | $0.00867(11)$ | $0.99504(12)$ | $0.87989(4)$ | $0.0412(3)$ |
| B1 | $0.67095(18)$ | $0.7413(2)$ | $0.84310(8)$ | $0.0527(4)$ |


|  | $\mathbf{x} / \mathbf{a}$ | $\mathbf{y} / \mathbf{b}$ | $\mathbf{z / c}$ | $\mathbf{U ( \mathbf { e q } )}$ |
| :--- | :--- | :--- | :--- | :--- |
| C1 | $0.94348(13)$ | $0.88605(13)$ | $0.87524(5)$ | $0.0393(3)$ |
| N2 | $0.96856(13)$ | $0.76400(13)$ | $0.89692(5)$ | $0.0469(3)$ |
| C2 | $0.12591(13)$ | $0.01970(13)$ | $0.91463(5)$ | $0.0390(3)$ |
| C3 | $0.10990(15)$ | $0.08574(15)$ | $0.96042(6)$ | $0.0448(3)$ |
| C4 | $0.22152(19)$ | $0.12870(18)$ | $0.99293(7)$ | $0.0588(4)$ |
| C5 | $0.3487(2)$ | $0.1050(2)$ | $0.97951(9)$ | $0.0713(5)$ |
| C6 | $0.36415(16)$ | $0.0386(2)$ | $0.93429(8)$ | $0.0660(5)$ |
| C7 | $0.25498(14)$ | $0.99586(16)$ | $0.90045(6)$ | $0.0483(3)$ |
| C8 | $0.2753(2)$ | $0.9232(2)$ | $0.85134(8)$ | $0.0638(5)$ |
| C9 | $0.2194(2)$ | $0.77962(19)$ | $0.84799(7)$ | $0.0621(4)$ |
| C10 | $0.21122(16)$ | $0.71648(15)$ | $0.89977(6)$ | $0.0513(4)$ |
| C11 | $0.3253(2)$ | $0.6611(2)$ | $0.92751(9)$ | $0.0683(5)$ |
| C12 | $0.3200(2)$ | $0.6024(2)$ | $0.97514(9)$ | $0.0791(6)$ |
| C13 | $0.2007(2)$ | $0.59757(19)$ | $0.99694(8)$ | $0.0647(5)$ |
| C14 | $0.08701(17)$ | $0.65459(16)$ | $0.97111(6)$ | $0.0506(4)$ |
| C15 | $0.09211(14)$ | $0.71410(13)$ | $0.92325(6)$ | $0.0440(3)$ |
| C16 | $0.84946(14)$ | $0.88640(14)$ | $0.76717(5)$ | $0.0433(3)$ |
| C17 | $0.93975(19)$ | $0.98334(19)$ | $0.75453(6)$ | $0.0595(4)$ |
| C18 | $0.9923(2)$ | $0.97799(18)$ | $0.70766(6)$ | $0.0617(4)$ |
| C19 | $0.95656(18)$ | $0.87769(15)$ | $0.67220(6)$ | $0.0521(4)$ |
| C20 | $0.86331(19)$ | $0.78416(17)$ | $0.68476(7)$ | $0.0585(4)$ |
| C21 | $0.81089(16)$ | $0.78692(16)$ | $0.73163(6)$ | $0.0512(4)$ |
| C22 | $0.0168(3)$ | $0.8711(2)$ | $0.62182(8)$ | $0.0728(5)$ |
| C23 | $0.71245(13)$ | $0.04943(15)$ | $0.83737(5)$ | $0.0442(3)$ |
| C24 | $0.6542(2)$ | $0.1214(2)$ | $0.79536(8)$ | $0.0725(6)$ |
| C25 | $0.5862(3)$ | $0.2392(2)$ | $0.80269(9)$ | $0.0836(7)$ |
| C26 | $0.57316(17)$ | $0.28717(17)$ | $0.85121(8)$ | $0.0634(5)$ |
| C27 | $0.63216(17)$ | $0.21436(19)$ | $0.89292(7)$ | $0.0601(4)$ |
| C28 | $0.70087(16)$ | $0.09696(17)$ | $0.88646(6)$ | $0.0521(4)$ |
| C29 | $0.4965(3)$ | $0.4136(2)$ | $0.85927(12)$ | $0.0932(8)$ |
|  |  |  |  |  |

## Bond lengths ( $\AA$ ) for 1-44:

| P1-C1 | 1.8578(14) | P1-B1 | 1.9279(19) |
| :---: | :---: | :---: | :---: |
| N1-C1 | $1.2699(18)$ | N1-C2 | $1.4176(17)$ |
| B1-H1BC | 1.12(2) | B1-H1BB | 1.11(2) |
| B1-H1BA | 0.99(2) | C1-N2 | 1.3582(18) |
| N2-C15 | $1.4315(19)$ | N2-H2N | 0.88(2) |
| C2-C3 | 1.389(2) | C2-C7 | 1.400(2) |
| C3-C4 | 1.389(2) | C3-H3 | 0.98(2) |
| C4-C5 | 1.375(3) | C4-H4 | 0.98(2) |
| C5-C6 | 1.377(3) | C5-H5 | 0.97(3) |
| C6-C7 | 1.389(2) | C6-H6 | 0.95(3) |
| C7-C8 | 1.506(2) | C8-C9 | 1.542(3) |
| C8-H8 | 1.03(2) | C8-H8B | 1.03(3) |
| C9-C10 | 1.501(3) | C9-H9B | 0.96(2) |
| C9-H9 | 1.07(2) | C10-C11 | 1.393(3) |
| C10-C15 | 1.397(2) | C11-C12 | 1.380(3) |
| C11-H11 | 0.94(3) | C12-C13 | 1.376(3) |
| C12-H12 | 0.98(3) | C13-C14 | 1.376(2) |
| C13-H13 | 0.98(2) | C14-C15 | 1.388(2) |
| C14-H14 | 0.98(2) | C16-C21 | 1.385(2) |
| C16-C17 | 1.390(2) | C17-C18 | 1.384(2) |
| C17-H17 | 0.92(2) | C18-C19 | 1.384(2) |
| C18-H18 | 0.98(3) | C19-C20 | 1.386(3) |
| C19-C22 | 1.506(2) | C20-C21 | 1.382(2) |
| C20-H20 | 0.95(2) | C21-H21 | 0.94(2) |
| C22-H22A | 0.97 | C22-H22B | 0.97 |
| C22-H22C | 0.97 | C23-C28 | 1.382(2) |
| C23-C24 | 1.383(2) | C24-C25 | 1.386(3) |
| C24-H24 | 1.02(2) | C25-C26 | 1.372(3) |
| C25-H25 | 0.98(3) | C26-C27 | 1.384(3) |
| C26-C29 | 1.507(3) | C27-C28 | 1.381(2) |
| C27-H27 | 1.01(2) | C28-H28 | 0.94(2) |
| C29-H29A | 0.97 | C29-H29B | 0.97 |
| C29-H29C | 0.97 |  |  |

## Bond angles $\left({ }^{\circ}\right)$ for 1-44:

| $\mathrm{C} 16-\mathrm{P} 1-\mathrm{C} 23$ | $107.54(7)$ | $\mathrm{C} 16-\mathrm{P} 1-\mathrm{C} 1$ | $103.09(6)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{C} 23-\mathrm{P} 1-\mathrm{C} 1$ | $105.20(6)$ | $\mathrm{C} 16-\mathrm{P} 1-\mathrm{B} 1$ | $115.92(8)$ |


| C23-P1-B1 | 113.50(8) | C1-P1-B1 | 110.57(7) |
| :---: | :---: | :---: | :---: |
| C1-N1-C2 | 125.99(12) | P1-B1-H1BC | 105.5(12) |
| P1-B1-H1BB | 108.6(11) | H1BC-B1-H1BB | 111.0(16) |
| P1-B1-H1BA | 109.1(13) | H1BC-B1-H1BA | 110.7(17) |
| H1BB-B1-H1BA | 111.6(16) | N1-C1-N2 | 131.77(12) |
| N1-C1-P1 | 115.93(10) | N2-C1-P1 | 112.27(10) |
| C1-N2-C15 | 128.45(12) | C1-N2-H2N | 115.2(12) |
| C15-N2-H2N | 116.3(12) | C3-C2-C7 | 120.44(13) |
| C3-C2-N1 | 117.68(12) | C7-C2-N1 | 121.28(13) |
| C4-C3-C2 | 120.59(15) | C4-C3-H3 | 120.5(11) |
| C2-C3-H3 | 118.8(11) | C5-C4-C3 | 119.42(18) |
| C5-C4-H4 | 120.2(13) | C3-C4-H4 | 120.2(13) |
| C4-C5-C6 | 119.81(16) | C4-C5-H5 | 119.3(15) |
| C6-C5-H5 | 120.9(15) | C5-C6-C7 | 122.42(16) |
| C5-C6-H6 | 121.9(15) | C7-C6-H6 | 115.6(15) |
| C6-C7-C2 | 117.31(15) | C6-C7-C8 | 121.10(16) |
| C2-C7-C8 | 121.56(14) | C7-C8-C9 | 114.60(14) |
| C7-C8-H8 | 108.0(12) | C9-C8-H8 | 108.6(12) |
| C7-C8-H8B | 103.1(13) | C9-C8-H8B | 112.0(14) |
| H8-C8-H8B | 110.4(18) | C10-C9-C8 | 113.49(15) |
| C10-C9-H9B | 108.3(13) | C8-C9-H9B | 109.1(13) |
| C10-C9-H9 | 109.6(11) | C8-C9-H9 | 109.2(11) |
| H9B-C9-H9 | 107.0(16) | C11-C10-C15 | 116.88(16) |
| C11-C10-C9 | 120.66(16) | C15-C10-C9 | 122.42(14) |
| C12-C11-C10 | 121.58(18) | C12-C11-H11 | 119.5(15) |
| C10-C11-H11 | 118.8(16) | C13-C12-C11 | 120.64(18) |
| C13-C12-H12 | 120.8(17) | C11-C12-H12 | 118.6(17) |
| C12-C13-C14 | 119.16(19) | C12-C13-H13 | 123.0(13) |
| C14-C13-H13 | 117.9(13) | C13-C14-C15 | 120.35(16) |
| C13-C14-H14 | 120.6(13) | C15-C14-H14 | 119.0(13) |
| C14-C15-C10 | 121.33(14) | C14-C15-N2 | 117.59(13) |
| C10-C15-N2 | 120.91(13) | C21-C16-C17 | 118.88(14) |
| C21-C16-P1 | 120.60(12) | C17-C16-P1 | 120.51(11) |
| C18-C17-C16 | 120.21(15) | C18-C17-H17 | 121.0(14) |
| C16-C17-H17 | 118.7(14) | C17-C18-C19 | 121.47(16) |
| C17-C18-H18 | 120.2(14) | C19-C18-H18 | 118.2(14) |
| C18-C19-C20 | 117.55(15) | C18-C19-C22 | 121.04(16) |
| C20-C19-C22 | 121.41(15) | C21-C20-C19 | 121.78(15) |


| C21-C20-H20 | $117.6(14)$ | C19-C20-H20 | $120.7(13)$ |
| :--- | :--- | :--- | :--- |
| C20-C21-C16 | $120.05(16)$ | C20-C21-H21 | $119.7(12)$ |
| C16-C21-H21 | $120.3(12)$ | C19-C22-H22A | 109.5 |
| C19-C22-H22B | 109.5 | H22A-C22-H22B | 109.5 |
| C19-C22-H22C | 109.5 | H22A-C22-H22C | 109.5 |
| H22B-C22-H22C | 109.5 | C28-C23-C24 | $118.78(15)$ |
| C28-C23-P1 | $119.70(12)$ | C24-C23-P1 | $121.20(12)$ |
| C23-C24-C25 | $120.24(18)$ | C23-C24-H24 | $117.8(13)$ |
| C25-C24-H24 | $121.8(13)$ | C26-C25-C24 | $121.62(19)$ |
| C26-C25-H25 | $123.0(19)$ | C24-C25-H25 | $115.4(19)$ |
| C25-C26-C27 | $117.56(16)$ | C25-C26-C29 | $121.7(2)$ |
| C27-C26-C29 | $120.74(19)$ | C28-C27-C26 | $121.73(17)$ |
| C28-C27-H27 | $117.8(12)$ | C26-C27-H27 | $120.4(12)$ |
| C23-C28-C27 | $120.06(16)$ | C23-C28-H28 | $119.4(12)$ |
| C27-C28-H28 | $120.4(12)$ | C26-C29-H29A | 109.5 |
| C26-C29-H29B | 109.5 | H29A-C29-H29B | 109.5 |
| C26-C29-H29C | 109.5 | H29A-C29-H29C | 109.5 |
| H29B-C29-H29C | 109.5 |  |  |



Sample and crystal data for 1-62:

Identification code
Chemical formula
Formula weight
Temperature
Wavelength
Crystal size
Crystal system
Space group
Unit cell dimensions

Volume
Z
Density (calculated)
Absorption coefficient
F(000)

JM1r
$\mathrm{C}_{63} \mathrm{H}_{60} \mathrm{Cl}_{4} \mathrm{~N}_{4} \mathrm{P}_{2} \mathrm{Pt}$
$1271.98 \mathrm{~g} / \mathrm{mol}$
240(2) K
1.54178 Å
$0.040 \times 0.080 \times 0.080 \mathrm{~mm}$
monoclinic
P 1211
$a=11.2580(16) \AA \quad \alpha=90^{\circ}$
$\mathrm{b}=12.9240(19) \AA \quad \beta=98.888(7)^{\circ}$
$\mathrm{c}=19.447(3) \AA \quad \gamma=90^{\circ}$
2795.5(7) $\AA^{3}$

2
$1.511 \mathrm{~g} / \mathrm{cm}^{3}$
$7.334 \mathrm{~mm}^{-1}$
1284

Data collection and structure refinement for 1-62:

Diffractometer
Radiation source

Bruker Apex II CCD
IMuS micro-focus source, Cu

| Theta range for data collection | 2.30 to $70.25^{\circ}$ |
| :---: | :---: |
| Index ranges | $-13<=\mathrm{h}<=13,-15<=\mathrm{k}<=15,-23<=1<=23$ |
| Reflections collected | 35162 |
| Independent reflections | 10415 [R(int) $=0.0263$ ] |
| Absorption correction | multi-scan |
| Max. and min. transmission | 0.7580 and 0.5910 |
| Structure solution technique | direct methods |
| Structure solution program | SHELXS-97 (Sheldrick, 2008) |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{2}$ |
| Refinement program | SHELXL-97 (Sheldrick, 2008) |
| Function minimized | $\Sigma \mathrm{w}\left(\mathrm{Fo}^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2}$ |
| Data / restraints / parameters | 10415 / 1 / 675 |
| Goodness-of-fit on $\mathbf{F}^{\mathbf{2}}$ | 1.003 |
| $\Delta / \sigma_{\text {max }}$ | 0.002 |
| Weighting scheme | $\mathrm{w}=1 /\left[\sigma^{2}\left(\mathrm{~F}_{0}{ }^{2}\right)+(0.0226 \mathrm{P})^{2}\right]$ |
| Absolute structure parameter | -0.0(0) |
| Largest diff. peak and hole | 0.668 and $-0.651 \mathrm{e}^{\circ}{ }^{-3}$ |
| R.M.S. deviation from mean | $0.058 \mathrm{e}^{-3}$ |

Table 3. Atomic coordinates and equivalent isotropic atomic displacement parameters ( $\AA^{\mathbf{2}}$ ) for 1-62:
$\mathrm{U}(\mathrm{eq})$ is defined as one third of the trace of the orthogonalized $\mathrm{U}_{\mathrm{ij}}$ tensor.

|  | $\mathbf{x} / \mathbf{a}$ | $\mathbf{y} / \mathbf{b}$ | $\mathbf{z} / \mathbf{c}$ | $\mathbf{U}(\mathbf{e q})$ |
| :--- | :--- | :--- | :--- | :--- |
| P1 | $0.82425(9)$ | $0.03999(8)$ | $0.81485(5)$ | $0.0319(2)$ |
| Pt1 | $0.84217(2)$ | $0.21192(2)$ | $0.78668(2)$ | $0.03098(5)$ |
| Cl1 | $0.01676(10)$ | $0.22898(11)$ | $0.86398(6)$ | $0.0489(3)$ |
| N1 | $0.6436(4)$ | $0.9058(3)$ | $0.8170(2)$ | $0.0417(8)$ |
| C1 | $0.9224(4)$ | $0.9794(4)$ | $0.8906(3)$ | $0.0435(10)$ |
| P2 | $0.85031(9)$ | $0.38228(8)$ | $0.75273(5)$ | $0.0320(2)$ |
| C12 | $0.66348(9)$ | $0.1981(2)$ | $0.71102(6)$ | $0.0476(5)$ |


|  | x/a | y/b | z/c | U(eq) |
| :---: | :---: | :---: | :---: | :---: |
| N2 | 0.6107(3) | 0.0774(3) | 0.8558(2) | 0.0386(8) |
| C2 | 0.9448(6) | 0.8703(4) | 0.8639(3) | 0.0607(14) |
| C3 | 0.9672(7) | 0.8830(5) | 0.7900(4) | 0.0651(18) |
| N3 | 0.7671(4) | 0.4938(3) | 0.6397(2) | 0.0430(8) |
| C4 | 0.8647(5) | 0.9484(4) | 0.7487(3) | 0.0455(11) |
| N4 | 0.8472(4) | 0.3273(3) | 0.6195(2) | 0.0424(8) |
| C5 | 0.8780(5) | 0.0845(5) | 0.9919(3) | 0.0550(12) |
| C6 | 0.8330(6) | 0.0943(6) | 0.0540(3) | 0.0652(15) |
| C7 | $0.7902(5)$ | 0.0119(6) | 0.0851(3) | 0.0625(16) |
| C8 | $0.7870(5)$ | 0.9167(5) | 0.0529(3) | 0.0621(15) |
| C9 | 0.8290(5) | 0.9050(4) | 0.9900(3) | 0.0536(12) |
| C10 | 0.8754(4) | 0.9888(4) | 0.9586(3) | 0.0441(10) |
| C11 | 0.0015(7) | 0.0451(5) | 0.6796(4) | 0.0672(16) |
| C12 | 0.0262(9) | 0.0886(6) | 0.6194(5) | 0.089(3) |
| C13 | 0.9405(11) | 0.0892(5) | 0.5604(5) | 0.091(3) |
| C14 | 0.8322(10) | 0.0473(7) | 0.5636(4) | 0.087(3) |
| C15 | 0.8074(7) | 0.0022(6) | 0.6248(4) | 0.0687(18) |
| C16 | 0.8922(6) | 0.9991(4) | 0.6837(3) | 0.0507(12) |
| C17 | $0.6699(5)$ | 0.0000(4) | 0.8294(3) | 0.0356(11) |
| C18 | 0.5285(4) | 0.8626(4) | 0.8195(2) | 0.0417(9) |
| C19 | 0.5224(5) | 0.7769(4) | 0.8615(3) | 0.0535(12) |
| C20 | 0.4148(5) | $0.7234(7)$ | 0.8609(3) | 0.0709(17) |
| C21 | 0.3138(6) | 0.7557(6) | 0.8168(4) | 0.0710(18) |
| C22 | 0.3196(5) | 0.8412(5) | 0.7751(3) | 0.0579(14) |
| C23 | 0.4257(5) | 0.8958(4) | 0.7751(2) | 0.0451(10) |
| C24 | $0.4259(5)$ | 0.9916(4) | 0.7307(3) | 0.0504(11) |
| C25 | 0.3806(5) | 0.0897(4) | 0.7653(3) | 0.0520(12) |
| C26 | 0.3909(4) | 0.0836(4) | 0.8427(3) | 0.0454(10) |
| C27 | 0.2900(5) | 0.0860(5) | 0.8754(3) | 0.0594(14) |
| C28 | 0.2972(5) | 0.0766(5) | 0.9461(3) | 0.0636(15) |
| C29 | 0.4070(5) | 0.0601(5) | 0.9873(3) | 0.0563(13) |
| C30 | 0.5099(4) | $0.0575(4)$ | 0.9566(3) | 0.0456(10) |
| C31 | 0.5027(5) | 0.0699(4) | 0.8853(3) | 0.0371(11) |
| C32 | 0.7516(5) | 0.4704(4) | 0.7940(2) | 0.0429(10) |
| C33 | 0.8381(6) | 0.5545(4) | 0.8293(3) | 0.0577(13) |
| C34 | 0.9399(6) | 0.5700(4) | 0.7865(3) | 0.0503(14) |
| C35 | 0.9892(4) | 0.4614(4) | 0.7758(2) | 0.0419(10) |


|  | $\mathbf{x} / \mathbf{a}$ | $\mathbf{y} / \mathbf{b}$ | $\mathbf{z} / \mathbf{c}$ | $\mathbf{U}(\mathbf{e q})$ |
| :--- | :--- | :--- | :--- | :--- |
| C36 | $0.5516(6)$ | $0.4101(5)$ | $0.8193(4)$ | $0.0636(15)$ |
| C37 | $0.4769(9)$ | $0.3630(7)$ | $0.8608(5)$ | $0.076(2)$ |
| C38 | $0.5261(8)$ | $0.3201(6)$ | $0.9216(5)$ | $0.080(2)$ |
| C39 | $0.6489(8)$ | $0.3231(5)$ | $0.9437(4)$ | $0.079(2)$ |
| C40 | $0.7242(7)$ | $0.3714(5)$ | $0.9026(3)$ | $0.0653(16)$ |
| C41 | $0.6764(7)$ | $0.4167(5)$ | $0.8397(3)$ | $0.0481(14)$ |
| C42 | $0.1016(5)$ | $0.5297(4)$ | $0.6820(3)$ | $0.0575(13)$ |
| C43 | $0.1887(6)$ | $0.5196(5)$ | $0.6393(4)$ | $0.0701(17)$ |
| C44 | $0.2568(6)$ | $0.4329(6)$ | $0.6417(4)$ | $0.0695(16)$ |
| C45 | $0.2373(6)$ | $0.3537(6)$ | $0.6860(4)$ | $0.0715(17)$ |
| C46 | $0.1496(6)$ | $0.3633(5)$ | $0.7286(3)$ | $0.0610(14)$ |
| C47 | $0.0811(4)$ | $0.4517(4)$ | $0.7271(3)$ | $0.0429(10)$ |
| C48 | $0.8118(4)$ | $0.4056(3)$ | $0.6575(2)$ | $0.0347(9)$ |
| C49 | $0.8527(5)$ | $0.3284(4)$ | $0.5464(2)$ | $0.0389(10)$ |
| C50 | $0.9633(5)$ | $0.3450(4)$ | $0.5254(3)$ | $0.0460(10)$ |
| C51 | $0.9739(5)$ | $0.3361(4)$ | $0.4553(3)$ | $0.0548(13)$ |
| C52 | $0.8740(6)$ | $0.3116(4)$ | $0.4075(3)$ | $0.0575(13)$ |
| C53 | $0.7653(5)$ | $0.2968(4)$ | $0.4283(3)$ | $0.0539(12)$ |
| C54 | $0.7506(5)$ | $0.3031(4)$ | $0.4981(3)$ | $0.0445(10)$ |
| C55 | $0.6303(5)$ | $0.2859(5)$ | $0.5209(3)$ | $0.0571(13)$ |
| C56 | $0.5735(5)$ | $0.3840(5)$ | $0.5488(3)$ | $0.0553(13)$ |
| C57 | $0.6277(4)$ | $0.4841(4)$ | $0.5291(3)$ | $0.0462(11)$ |
| C58 | $0.5818(5)$ | $0.5347(5)$ | $0.4668(3)$ | $0.0582(14)$ |
| C59 | $0.6322(6)$ | $0.6239(5)$ | $0.4460(3)$ | $0.0638(15)$ |
| C60 | $0.7317(6)$ | $0.6665(5)$ | $0.4871(3)$ | $0.0631(15)$ |
| C61 | $0.7772(5)$ | $0.6194(4)$ | $0.5491(3)$ | $0.0524(12)$ |
| C62 | $0.7265(4)$ | $0.5292(4)$ | $0.5712(2)$ | $0.0412(9)$ |
| C63 | $0.619(8)$ | $0.6943(6)$ | $0.6913(4)$ | $0.086(2)$ |
| C13 | $0.4819(2)$ | $0.6166(2)$ | $0.6754(2)$ | $0.1278(10)$ |
| C14 | $0.5954(3)$ | $0.80182(18)$ | $0.63711(14)$ | $0.1063(7)$ |
|  |  |  |  |  |

Table 4. Bond lengths (A) for 1-62.

| P1-C4 | $1.857(5)$ | $\mathrm{P} 1-\mathrm{C} 1$ | $1.871(5)$ |
| :--- | :--- | :--- | :--- |
| P1-C17 | $1.875(6)$ | $\mathrm{P} 1-\mathrm{Pt} 1$ | $2.3048(11)$ |
| Pt1-Cl1 | $2.2932(11)$ | $\mathrm{Pt} 1-\mathrm{P} 2$ | $2.3045(11)$ |
| Pt1-C12 | $2.3085(10)$ | $\mathrm{N} 1-\mathrm{C} 17$ | $1.267(7)$ |
| N1-C18 | $1.418(6)$ | $\mathrm{C} 1-\mathrm{C} 10$ | $1.504(7)$ |


| C1-C2 | $1.538(7)$ | C1-H1A | 0.99 |
| :--- | :--- | :--- | :--- |
| P2-C32 | $1.858(5)$ | P2-C48 | $1.861(5)$ |
| P2-C35 | $1.864(5)$ | N2-C17 | $1.347(7)$ |
| N2-C31 | $1.426(7)$ | N2-H2N | $0.83(6)$ |
| C2-C3 | $1.504(9)$ | C2-H2A | 0.98 |
| C2-H2B | 0.98 | C3-C4 | $1.551(9)$ |
| C3-H3A | 0.98 | C3-H3B | 0.98 |
| N3-C48 | $1.272(6)$ | N3-C62 | $1.415(6)$ |
| C4-C16 | $1.497(7)$ | C4-H4A | 0.99 |
| N4-C48 | $1.349(6)$ | N4-C49 | $1.433(6)$ |
| N4-H4N | $0.92(7)$ | C5-C6 | $1.385(8)$ |
| C5-C10 | $1.394(8)$ | C5-H5A | 0.94 |
| C6-C7 | $1.350(10)$ | C6-H6A | 0.94 |
| C7-C8 | $1.378(10)$ | C7-H7A | 0.94 |
| C8-C9 | $1.387(8)$ | C8-H8A | 0.94 |
| C9-C10 | $1.383(8)$ | C9-H9A | 0.94 |
| C11-C12 | $1.366(10)$ | C11-C16 | $1.380(9)$ |
| C11-H11A | 0.94 | C12-C13 | $1.380(15)$ |
| C12-H12A | 0.94 | C13-C14 | $1.345(14)$ |
| C13-H13A | 0.94 | C14-C15 | $1.393(11)$ |
| C14-H14A | 0.94 | C15-C16 | $1.374(10)$ |
| C15-H15A | 0.94 | C18-C19 | $1.384(7)$ |
| C18-C23 | $1.400(7)$ | C19-C20 | $1.393(8)$ |
| C19-H19A | 0.94 | C20-C21 | $1.379(9)$ |
| C20-H20A | 0.94 | C21-C22 | $1.378(9)$ |
| C21-H21A | 0.94 | C22-C23 | $1.387(7)$ |
| C22-H22A | 0.94 | C23-C24 | $1.510(7)$ |
| C24-C25 | $1.558(8)$ | C24-H24A | 0.98 |
| C24-H24B | 0.98 | C25-C26 | $1.494(8)$ |
| C25-H25A | 0.98 | C25-H25B | 0.98 |
| C26-C27 | $1.386(7)$ | C26-C31 | $1.407(7)$ |
| C27-C28 | $1.371(9)$ | C27-H27A | 0.94 |
| C28-C29 | $1.382(9)$ | C28-H28A | 0.94 |
| C29-C30 | $1.383(7)$ | C29-H29A | 0.94 |
| C30-C31 | $1.386(8)$ | 0.99 | 0.98 |
| C32-C41 | $1.490(8)$ | C32A | C33 |


| C34-C35 | 1.535(7) | C34-H34A | 0.98 |
| :---: | :---: | :---: | :---: |
| C34-H34B | 0.98 | C35-C47 | 1.512(7) |
| C35-H35A | 0.99 | C36-C37 | 1.392(10) |
| C36-C41 | 1.402(10) | C36-H36A | 0.94 |
| C37-C38 | 1.345(13) | C37-H37A | 0.94 |
| C38-C39 | 1.382(13) | C38-H38A | 0.94 |
| C39-C40 | 1.400(10) | C39-H39A | 0.94 |
| C40-C41 | 1.388(9) | C40-H40A | 0.94 |
| C42-C47 | 1.379(7) | C42-C43 | 1.387(9) |
| C42-H42A | 0.94 | C43-C44 | 1.354(10) |
| C43-H43A | 0.94 | C44-C45 | 1.377(10) |
| C44-H44A | 0.94 | C45-C46 | 1.390(9) |
| C45-H45A | 0.94 | C46-C47 | 1.376(8) |
| C46-H46A | 0.94 | C49-C50 | 1.386(7) |
| C49-C54 | 1.405(7) | C50-C51 | 1.391(8) |
| C50-H50A | 0.94 | C51-C52 | 1.381(9) |
| C51-H51A | 0.94 | C52-C53 | 1.361(9) |
| C52-H52A | 0.94 | C53-C54 | $1.395(7)$ |
| C53-H53A | 0.94 | C54-C55 | 1.506(7) |
| C55-C56 | 1.555(8) | C55-H55A | 0.98 |
| C55-H55B | 0.98 | C56-C57 | $1.505(8)$ |
| C56-H56A | 0.98 | C56-H56B | 0.98 |
| C57-C58 | 1.403(7) | C57-C62 | 1.403(7) |
| C58-C59 | 1.373(10) | C58-H58A | 0.94 |
| C59-C60 | 1.385(10) | C59-H59A | 0.94 |
| C60-C61 | 1.377(8) | C60-H60A | 0.94 |
| C61-C62 | 1.395(7) | C61-H61A | 0.94 |
| C63-Cl4 | 1.737(9) | C63-Cl3 | 1.763(9) |
| C63-H63A | 0.98 | C63-H63B | 0.98 |

Table 5. Bond angles $\left({ }^{\circ}\right)$ for 1-62:

| C4-P1-C1 | $95.8(2)$ | $\mathrm{C} 4-\mathrm{P} 1-\mathrm{C} 17$ | $105.1(2)$ |
| :--- | :--- | :--- | :--- |
| C1-P1-C17 | $102.4(2)$ | $\mathrm{C} 4-\mathrm{P} 1-\mathrm{Pt} 1$ | $114.21(16)$ |
| C1-P1-Pt1 | $121.85(17)$ | $\mathrm{C} 17-\mathrm{P} 1-\mathrm{Pt} 1$ | $114.76(17)$ |
| Cl1-Pt1-P2 | $91.60(4)$ | $\mathrm{C} 11-\mathrm{Pt} 1-\mathrm{P} 1$ | $91.98(4)$ |
| P2-Pt1-P1 | $176.37(4)$ | $\mathrm{Cl1}-\mathrm{Pt} 1-\mathrm{Cl} 2$ | $178.22(7)$ |
| $\mathrm{P} 2-\mathrm{Pt} 1-\mathrm{Cl} 2$ | $87.75(7)$ | $\mathrm{P} 1-\mathrm{Pt} 1-\mathrm{Cl} 2$ | $88.69(7)$ |


| C17-N1-C18 | 124.0(4) | C10-C1-C2 | 118.0(4) |
| :---: | :---: | :---: | :---: |
| C10-C1-P1 | 114.4(3) | C2-C1-P1 | 103.0(4) |
| C10-C1-H1A | 106.9 | C2-C1-H1A | 106.9 |
| P1-C1-H1A | 106.9 | C32-P2-C48 | 105.5(2) |
| C32-P2-C35 | 95.5(2) | C48-P2-C35 | 102.0(2) |
| C32-P2-Pt1 | 114.14(16) | C48-P2-Pt1 | 115.26(14) |
| C35-P2-Pt1 | 121.61(15) | C17-N2-C31 | 127.4(4) |
| C17-N2-H2N | 112.(4) | C31-N2-H2N | 120.(4) |
| C3-C2-C1 | 106.4(5) | C3-C2-H2A | 110.4 |
| C1-C2-H2A | 110.4 | C3-C2-H2B | 110.4 |
| C1-C2-H2B | 110.4 | H2A-C2-H2B | 108.6 |
| C2-C3-C4 | 109.5(5) | C2-C3-H3A | 109.8 |
| C4-C3-H3A | 109.8 | C2-C3-H3B | 109.8 |
| C4-C3-H3B | 109.8 | H3A-C3-H3B | 108.2 |
| C48-N3-C62 | 127.0(4) | C16-C4-C3 | 116.2(5) |
| C16-C4-P1 | 114.2(3) | C3-C4-P1 | 103.3(4) |
| C16-C4-H4A | 107.5 | C3-C4-H4A | 107.5 |
| P1-C4-H4A | 107.5 | C48-N4-C49 | 126.7(4) |
| C48-N4-H4N | 114.(4) | C49-N4-H4N | 119.(4) |
| C6-C5-C10 | 120.1(6) | C6-C5-H5A | 119.9 |
| C10-C5-H5A | 119.9 | C7-C6-C5 | 121.5(6) |
| C7-C6-H6A | 119.3 | C5-C6-H6A | 119.3 |
| C6-C7-C8 | 119.1(6) | C6-C7-H7A | 120.4 |
| C8-C7-H7A | 120.4 | C7-C8-C9 | 120.6(6) |
| C7-C8-H8A | 119.7 | C9-C8-H8A | 119.7 |
| C10-C9-C8 | 120.4(6) | C10-C9-H9A | 119.8 |
| C8-C9-H9A | 119.8 | C9-C10-C5 | 118.2(5) |
| C9-C10-C1 | 122.2(5) | C5-C10-C1 | 119.6(5) |
| C12-C11-C16 | 121.9(8) | C12-C11-H11A | 119.0 |
| C16-C11-H11A | 119.0 | C11-C12-C13 | 120.4(8) |
| C11-C12-H12A | 119.8 | C13-C12-H12A | 119.8 |
| C14-C13-C12 | 118.9(7) | C14-C13-H13A | 120.6 |
| C12-C13-H13A | 120.6 | C13-C14-C15 | 120.7(9) |
| C13-C14-H14A | 119.7 | C15-C14-H14A | 119.7 |
| C16-C15-C14 | 121.3(8) | C16-C15-H15A | 119.3 |
| C14-C15-H15A | 119.3 | C15-C16-C11 | 116.8(6) |
| C15-C16-C4 | 120.5(6) | C11-C16-C4 | 122.7(6) |
| N1-C17-N2 | 132.1(5) | N1-C17-P1 | 115.5(4) |


| N2-C17-P1 | 112.2(4) | C19-C18-C23 | 119.8(5) |
| :---: | :---: | :---: | :---: |
| C19-C18-N1 | 117.7(5) | C23-C18-N1 | 122.0(4) |
| C18-C19-C20 | 120.9(6) | C18-C19-H19A | 119.5 |
| C20-C19-H19A | 119.5 | C21-C20-C19 | 119.2(6) |
| C21-C20-H20A | 120.4 | C19-C20-H20A | 120.4 |
| C22-C21-C20 | 119.9(5) | C22-C21-H21A | 120.0 |
| C20-C21-H21A | 120.0 | C21-C22-C23 | 121.8(5) |
| C21-C22-H22A | 119.1 | C23-C22-H22A | 119.1 |
| C22-C23-C18 | 118.3(5) | C22-C23-C24 | 119.6(5) |
| C18-C23-C24 | 122.0(4) | C23-C24-C25 | 113.0(4) |
| C23-C24-H24A | 109.0 | C25-C24-H24A | 109.0 |
| C23-C24-H24B | 109.0 | C25-C24-H24B | 109.0 |
| H24A-C24-H24B | 107.8 | C26-C25-C24 | 114.3(4) |
| C26-C25-H25A | 108.7 | C24-C25-H25A | 108.7 |
| C26-C25-H25B | 108.7 | C24-C25-H25B | 108.7 |
| H25A-C25-H25B | 107.6 | C27-C26-C31 | 117.0(5) |
| C27-C26-C25 | 121.3(5) | C31-C26-C25 | 121.6(5) |
| C28-C27-C26 | 122.2(5) | C28-C27-H27A | 118.9 |
| C26-C27-H27A | 118.9 | C27-C28-C29 | 120.3(5) |
| C27-C28-H28A | 119.8 | C29-C28-H28A | 119.8 |
| C28-C29-C30 | 119.1(5) | C28-C29-H29A | 120.4 |
| C30-C29-H29A | 120.4 | C29-C30-C31 | 120.4(5) |
| C29-C30-H30A | 119.8 | C31-C30-H30A | 119.8 |
| C30-C31-C26 | 120.9(5) | C30-C31-N2 | 119.3(5) |
| C26-C31-N2 | 119.7(5) | C41-C32-C33 | 116.0(5) |
| C41-C32-P2 | 113.9(4) | C33-C32-P2 | 104.2(4) |
| C41-C32-H32A | 107.4 | C33-C32-H32A | 107.4 |
| P2-C32-H32A | 107.4 | C34-C33-C32 | 108.9(4) |
| C34-C33-H33A | 109.9 | C32-C33-H33A | 109.9 |
| C34-C33-H33B | 109.9 | C32-C33-H33B | 109.9 |
| H33A-C33-H33B | 108.3 | C33-C34-C35 | 105.7(4) |
| C33-C34-H34A | 110.6 | C35-C34-H34A | 110.6 |
| C33-C34-H34B | 110.6 | C35-C34-H34B | 110.6 |
| H34A-C34-H34B | 108.7 | C47-C35-C34 | 117.2(4) |
| C47-C35-P2 | 115.7(3) | C34-C35-P2 | 103.1(3) |
| C47-C35-H35A | 106.7 | C34-C35-H35A | 106.7 |
| P2-C35-H35A | 106.7 | C37-C36-C41 | 122.1(8) |


| C37-C36-H36A | 118.9 | C41-C36-H36A | 118.9 |
| :---: | :---: | :---: | :---: |
| C38-C37-C36 | 119.1(9) | C38-C37-H37A | 120.5 |
| C36-C37-H37A | 120.5 | C37-C38-C39 | 121.2(7) |
| C37-C38-H38A | 119.4 | C39-C38-H38A | 119.4 |
| C38-C39-C40 | 120.0(7) | C38-C39-H39A | 120.0 |
| C40-C39-H39A | 120.0 | C41-C40-C39 | 120.3(7) |
| C41-C40-H40A | 119.8 | C39-C40-H40A | 119.8 |
| C40-C41-C36 | 117.3(6) | C40-C41-C32 | 122.9(7) |
| C36-C41-C32 | 119.8(6) | C47-C42-C43 | 121.1(6) |
| C47-C42-H42A | 119.4 | C43-C42-H42A | 119.4 |
| C44-C43-C42 | 120.3(6) | C44-C43-H43A | 119.8 |
| C42-C43-H43A | 119.8 | C43-C44-C45 | 119.5(6) |
| C43-C44-H44A | 120.2 | C45-C44-H44A | 120.2 |
| C44-C45-C46 | 120.2(6) | C44-C45-H45A | 119.9 |
| C46-C45-H45A | 119.9 | C47-C46-C45 | 120.6(6) |
| C47-C46-H46A | 119.7 | C45-C46-H46A | 119.7 |
| C46-C47-C42 | 118.2(5) | C46-C47-C35 | 119.4(5) |
| C42-C47-C35 | 122.4(5) | N3-C48-N4 | 131.4(5) |
| N3-C48-P2 | 116.0(4) | N4-C48-P2 | 112.4(3) |
| C50-C49-C54 | 121.2(5) | C50-C49-N4 | 118.2(5) |
| C54-C49-N4 | 120.3(5) | C49-C50-C51 | 119.5(5) |
| C49-C50-H50A | 120.2 | C51-C50-H50A | 120.2 |
| C52-C51-C50 | 119.6(5) | C52-C51-H51A | 120.2 |
| C50-C51-H51A | 120.2 | C53-C52-C51 | 120.5(5) |
| C53-C52-H52A | 119.7 | C51-C52-H52A | 119.7 |
| C52-C53-C54 | 121.9(5) | C52-C53-H53A | 119.1 |
| C54-C53-H53A | 119.1 | C53-C54-C49 | 117.1(5) |
| C53-C54-C55 | 121.6(5) | C49-C54-C55 | 121.2(5) |
| C54-C55-C56 | 114.8(4) | C54-C55-H55A | 108.6 |
| C56-C55-H55A | 108.6 | C54-C55-H55B | 108.6 |
| C56-C55-H55B | 108.6 | H55A-C55-H55B | 107.6 |
| C57-C56-C55 | 114.0(4) | C57-C56-H56A | 108.7 |
| C55-C56-H56A | 108.7 | C57-C56-H56B | 108.7 |
| C55-C56-H56B | 108.7 | H56A-C56-H56B | 107.6 |
| C58-C57-C62 | 117.6(5) | C58-C57-C56 | 120.8(5) |
| C62-C57-C56 | 121.6(5) | C59-C58-C57 | 122.1(6) |
| C59-C58-H58A | 118.9 | C57-C58-H58A | 118.9 |
| C58-C59-C60 | 119.8(5) | C58-C59-H59A | 120.1 |


| C60-C59-H59A | 120.1 | C61-C60-C59 | $119.2(6)$ |
| :--- | :--- | :--- | :--- |
| C61-C60-H60A | 120.4 | C59-C60-H60A | 120.4 |
| C60-C61-C62 | $121.6(6)$ | C60-C61-H61A | 119.2 |
| C62-C61-H61A | 119.2 | C61-C62-C57 | $119.5(5)$ |
| C61-C62-N3 | $118.2(4)$ | C57-C62-N3 | $121.8(4)$ |
| C14-C63-Cl3 | $110.0(4)$ | C14-C63-H63A | 109.7 |
| C13-C63-H63A | 109.7 | Cl4-C63-H63B | 109.7 |
| Cl3-C63-H63B | 109.7 | H63A-C63-H63B | 108.2 |




Sample and crystal data for 2-66:

Identification code
Chemical formula
Formula weight
Temperature
Wavelength
Crystal size
Crystal system
Space group
milligan5
$\mathrm{C}_{38} \mathrm{H}_{52} \mathrm{~B}_{2} \mathrm{P}_{2}$
$592.35 \mathrm{~g} / \mathrm{mol}$
230(2) K
1.54178 A
$0.100 \times 0.140 \times 0.190 \mathrm{~mm}$
triclinic
P-1
Unit cell dimension
10.7360(2)
12.9572(3)
13.9710(3)
b
c
$\begin{array}{ll}=\begin{array}{c}\alpha \\ 109.8510(10)^{\circ}\end{array} & = \\ =\frac{\beta}{93.0060(10)^{\circ}} & = \\ = & \gamma \\ 94.6320(10)^{\circ} & =\end{array}$ 1815.41(7) $\AA$

2
$1.084 \mathrm{~g} / \mathrm{cm}^{3}$
$1.244 \mathrm{~mm}^{-1}$
640

Data collection and structure refinement for 2-66:

Diffractometer
Radiation source
Theta range for data collection

Bruker Apex II CCD
IMuS micro-focus source, Cu
3.38 to $68.26^{\circ}$

| Index ranges | $-12<=\mathrm{h}<=12,-15<=\mathrm{k}<=15,-16<=1<=16$ |
| :---: | :---: |
| Reflections collected | 30565 |
| Independent reflections | $6499[\mathrm{R}(\mathrm{int})=0.0247]$ |
| Absorption correction | multi-scan |
| Max. and min. transmission | 0.8860 and 0.7980 |
| Structure solution technique | direct methods |
| Structure solution program | SHELXS-97 (Sheldrick, 2008) |
| Refinement method | Full-matrix least-squares on $\mathrm{F}^{2}$ |
| Refinement program | SHELXL-97 (Sheldrick, 2008) |
| Function minimized | $\Sigma \mathrm{w}\left(\mathrm{F}_{\mathrm{o}}{ }^{2}-\mathrm{F}_{\mathrm{c}}{ }^{2}\right)^{2}$ |
| Data / restraints / parameters | 6499 / 0 / 412 |
| Goodness-of-fit on $\mathbf{F}^{\mathbf{2}}$ | 1.601 |
| $\Delta / \sigma_{\text {max }}$ | 0.683 |
| Largest diff. peak and hole | 0.459 and -0.316 e $\AA^{-3}$ |
| R.M.S. deviation from mean | $0.042 \mathrm{e}^{\text {® }}{ }^{-3}$ |

Atomic coordinates and equivalent isotropic atomic displacement parameters ( $\AA^{\mathbf{2}}$ ) for 2-66:
$\mathrm{U}(\mathrm{eq})$ is defined as one third of the trace of the orthogonalized $\mathrm{U}_{\mathrm{ij}}$ tensor.

|  | $\mathbf{x} / \mathbf{a}$ | $\mathbf{y} / \mathbf{b}$ | $\mathbf{z} / \mathbf{c}$ | $\mathbf{U}(\mathbf{e q})$ |
| :--- | :--- | :--- | :--- | :--- |
| P1 | $0.19966(3)$ | $0.52444(3)$ | $0.82114(3)$ | $0.03355(13$ |
|  |  |  |  |  |
| P2 | $0.32664(3)$ | $0.23653(3)$ | $0.38012(2)$ | $0.02806(12$ |
|  |  |  |  |  |
| B1 | $0.0721(2)$ | $0.40690(18)$ | $0.80845(17)$ | $0.0507(5)$ |
| B2 | $0.4100(2)$ | $0.28633(19)$ | $0.28232(15)$ | $0.0471(5)$ |
| C1 | $0.30086(14)$ | $0.48894(13)$ | $0.71670(11)$ | $0.0369(3)$ |
| C2 | $0.23159(14)$ | $0.45774(12)$ | $0.61122(11)$ | $0.0359(3)$ |
| C3 | $0.17927(14)$ | $0.33626(12)$ | $0.55564(11)$ | $0.0361(3)$ |
| C4 | $0.23208(14)$ | $0.34194(12)$ | $0.45546(11)$ | $0.0326(3)$ |

$\left.\begin{array}{lllll}\text { C5 } & 0.31503(16) & 0.44912(12) & 0.52218(11) & 0.0371(3) \\ \text { C6 } & 0.13207(18) & 0.36468(15) & 0.38489(13) & 0.0489(4) \\ \text { C7 } & 0.26700(15) & 0.53205(13) & 0.01746(12) & 0.0392(3) \\ \text { C8 } & 0.34174(16) & 0.56628(14) & 0.10964(13) & 0.0426(4) \\ \text { C9 } & 0.45289(16) & 0.63279(14) & 0.11847(13) & 0.0453(4) \\ \text { C10 } & 0.49153(15) & 0.66547(14) & 0.03932(13) & 0.0426(4) \\ \text { C11 } & 0.41629(15) & 0.63046(13) & 0.94792(12) & 0.0396(3) \\ \text { C12 } & 0.30398(14) & 0.56373(12) & 0.93672(11) & 0.0344(3) \\ \text { C13 } & 0.3023(2) & 0.5309(2) & 0.19656(15) & 0.0641(5) \\ \text { C14 } & 0.61199(19) & 0.7382(2) & 0.05089(17) & 0.0645(6) \\ \text { C15 } & 0.03494(14) & 0.68418(14) & 0.87547(12) & 0.0407(4) \\ \text { C16 } & 0.99422(16) & 0.78788(16) & 0.89116(13) & 0.0474(4) \\ \text { C17 } & 0.06097(19) & 0.85856(15) & 0.85328(14) & 0.0522(4) \\ \text { C18 } & 0.1683(2) & 0.83074(16) & 0.80348(15) & 0.0539(5) \\ \text { C19 } & 0.20836(17) & 0.72818(14) & 0.79004(13) & 0.0444(4) \\ \text { C20 } & 0.14216(14) & 0.65440(13) & 0.82583(11) & 0.0351(3) \\ \text { C21 } & 0.8820(2) & 0.8215(2) & 0.94971(19) & 0.0711(6) \\ \text { C22 } & 0.2421(3) & 0.9116(2) & 0.7666(2) & 0.0907(9) \\ \text { C23 } & 0.56510(14) & 0.24970(13) & 0.47283(12) & 0.0377(3) \\ \text { C24 } & 0.65519(15) & 0.22992(14) & 0.53742(14) & 0.0453(4) \\ \text { C25 } & 0.61727(16) & 0.17137(15) & 0.59964(13) & 0.0473(4) \\ \text { C26 } & 0.49283(16) & 0.13203(13) & 0.59825(12) & 0.0400(4) \\ \text { C27 } & 0.40416(14) & 0.15100(12) & 0.53136(11) & 0.0341(3) \\ \text { C28 } & 0.43931(13) & 0.21081(11) & 0.46885(10) & 0.0311(3) \\ \text { C29 } & 0.79204(17) & 0.2697(2) & 0.53894(19) & 0.0678(6) \\ \text { C30 } & 0.4531(2) & 0.06822(16) & 0.66562(14) & 0.0526(4) \\ \text { C31 } & 0.26691(15) & 0.03334(13) & 0.23348(12) & 0.0399(4) \\ \text { C32 } & 0.20427(18) & 0.92824(14) & 0.18698(13) & 0.0486(4) \\ \text { C33 } & 0.10107(19) & 0.89866(14) & 0.23020(15) & 0.0533(5) \\ \text { C34 } & 0.05788(17) & 0.97066(15) & 0.31808(14) & 0.0488(4) \\ \text { C35 } & 0.12303(15) & 0.07547(13) & 0.36388(12) & 0.0399(3) \\ \text { C36 } & 0.22863(14) & 0.10699(12) & 0.32230(11) & 0.0324(3) \\ \text { C37 } & 0.2503(3) & 0.84995(19) & 0.09215(17) & 0.0764(7) \\ \text { C38 } & 0.9448(2) & 0.9343(2) & 0.3628(2) & 0.0734(7) \\ & & & & \\ \text { C3) } & & 0\end{array}\right)$

## Bond lengths ( $\AA$ ) for 2-66:

| P1-C12 | 1.8113(15) | P1-C1 | 1.8200(15) |
| :---: | :---: | :---: | :---: |
| P1-C20 | $1.8216(16)$ | P1-B1 | 1.9184(19) |
| P2-C36 | 1.8116(14) | P2-C28 | 1.8164(14) |
| P2-C4 | 1.8340 (15) | P2-B2 | 1.9285(17) |
| B1-H1B1 | 1.12(3) | B1-H1B2 | 1.07(2) |
| B1-H1B3 | 1.12(2) | B2-H2B1 | 1.13(2) |
| B2-H2B2 | 1.10(2) | B2-H2B3 | 1.19(2) |
| C1-C2 | 1.523(2) | C1-H1A | 0.98 |
| C1-H1B | 0.98 | C2-C3 | 1.542(2) |
| C2-C5 | 1.550(2) | C2-H2A | 0.99 |
| C3-C4 | 1.5595(19) | C3-H3A | 0.98 |
| C3-H3B | 0.98 | C4-C6 | 1.533(2) |
| C4-C5 | 1.557(2) | C5-H5A | 0.98 |
| C5-H5B | 0.98 | C6-H6A | 0.97 |
| C6-H6B | 0.97 | C6-H6C | 0.97 |
| C7-C12 | 1.391(2) | C7-C8 | 1.395(2) |
| C7-H7A | 0.94 | C8-C9 | 1.389(3) |
| C8-C13 | 1.505(2) | C9-C10 | 1.383(2) |
| C9-H9A | 0.94 | C10-C11 | 1.389(2) |
| C10-C14 | 1.505(2) | C11-C12 | 1.395(2) |
| C11-H11A | 0.94 | C13-H13A | 0.97 |
| C13-H13B | 0.97 | C13-H13C | 0.97 |
| C14-H14A | 0.97 | C14-H14B | 0.97 |
| C14-H14C | 0.97 | C15-C20 | 1.389(2) |
| C15-C16 | 1.396(3) | C15-H15A | 0.94 |
| C16-C17 | 1.377(3) | C16-C21 | 1.502(3) |
| C17-C18 | 1.387(3) | C17-H17A | 0.94 |
| C18-C19 | 1.385(3) | C18-C22 | 1.506(3) |
| C19-C20 | 1.386(2) | C19-H19A | 0.94 |
| C21-H21A | 0.97 | C21-H21B | 0.97 |
| C21-H21C | 0.97 | C22-H22A | 0.97 |
| C22-H22B | 0.97 | C22-H22C | 0.97 |
| $\mathrm{C} 23-\mathrm{C} 24$ | 1.386(2) | C23-C28 | 1.394(2) |
| C23-H23A | 0.94 | C24-C25 | 1.390(3) |
| C24-C29 | 1.514(2) | C25-C26 | 1.388(3) |
| C25-H25A | 0.94 | C26-C27 | 1.393(2) |
| C26-C30 | 1.505(2) | C27-C28 | 1.397(2) |


| C27-H27A | 0.94 | C29-H29A | 0.97 |
| :--- | :--- | :--- | :--- |
| C29-H29B | 0.97 | C29-H29C | 0.97 |
| C30-H30A | 0.97 | C30-H30B | 0.97 |
| C30-H30C | 0.97 | C31-C32 | $1.391(2)$ |
| C31-C36 | $1.390(2)$ | C31-H31A | 0.94 |
| C32-C33 | $1.378(3)$ | C32-C37 | $1.505(3)$ |
| C33-C34 | $1.394(3)$ | C33-H33A | 0.94 |
| C34-C35 | $1.397(2)$ | C34-C38 | $1.507(3)$ |
| C35-C36 | $1.395(2)$ | C35-H35A | 0.94 |
| C37-H37A | 0.97 | C37-H37B | 0.97 |
| C37-H37C | 0.97 | C38-H38A | 0.97 |
| C38-H38B | 0.97 | C38-H38C | 0.97 |

## Bond angles $\left(^{\circ}\right.$ ) for 2-66:

| C12-P1-C1 | $105.45(7)$ | C12-P1-C20 | $102.99(7)$ |
| :--- | :--- | :--- | :--- |
| C1-P1-C20 | $106.08(7)$ | C12-P1-B1 | $113.92(9)$ |
| C1-P1-B1 | $112.26(9)$ | C20-P1-B1 | $115.18(9)$ |
| C36-P2-C28 | $104.87(6)$ | C36-P2-C4 | $109.45(7)$ |
| C28-P2-C4 | $107.39(7)$ | C36-P2-B2 | $113.35(8)$ |
| C28-P2-B2 | $111.13(8)$ | C4-P2-B2 | $110.37(8)$ |
| P1-B1-H1B1 | $102.4(13)$ | P1-B1-H1B2 | $104.9(13)$ |
| H1B1-B1-H1B2 | $114.2(18)$ | P1-B1-H1B3 | $109.5(12)$ |
| H1B1-B1-H1B3 | $111.8(17)$ | H1B2-B1-H1B3 | $113.1(17)$ |
| P2-B2-H2B1 | $104.2(11)$ | P2-B2-H2B2 | $102.3(10)$ |
| H2B1-B2-H2B2 | $111.1(16)$ | P2-B2-H2B3 | $106.6(10)$ |
| H2B1-B2-H2B3 | $115.3(15)$ | H2B2-B2-H2B3 | $115.7(15)$ |
| C2-C1-P1 | $114.11(10)$ | C2-C1-H1A | 108.7 |
| P1-C1-H1A | 108.7 | C2-C1-H1B | 108.7 |
| P1-C1-H1B | 108.7 | H1A-C1-H1B | 107.6 |
| C1-C2-C3 | $118.42(13)$ | C1-C2-C5 | $115.84(13)$ |
| C3-C2-C5 | $88.73(11)$ | C1-C2-H2A | 110.7 |
| C3-C2-H2A | 110.7 | C5-C2-H2A | 110.7 |
| C2-C3-C4 | $89.55(11)$ | C2-C3-H3A | 113.7 |
| C4-C3-H3A | 113.7 | C2-C3-H3B | 113.7 |
| C4-C3-H3B | 113.7 | H3A-C3-H3B | 111.0 |
| C6-C4-C3 | $112.14(13)$ | C6-C4-C5 | $112.06(13)$ |
| C3-C4-C5 | $87.83(10)$ | C6-C4-P2 | $109.88(11)$ |
| C3-C4-P2 | $121.35(10)$ | C5-C4-P2 | $111.82(10)$ |


| C2-C5-C4 | 89.36(11) | C2-C5-H5A | 113.8 |
| :---: | :---: | :---: | :---: |
| C4-C5-H5A | 113.8 | C2-C5-H5B | 113.8 |
| C4-C5-H5B | 113.8 | H5A-C5-H5B | 111.0 |
| C4-C6-H6A | 109.5 | C4-C6-H6B | 109.5 |
| H6A-C6-H6B | 109.5 | C4-C6-H6C | 109.5 |
| H6A-C6-H6C | 109.5 | H6B-C6-H6C | 109.5 |
| C12-C7-C8 | 120.42(15) | C12-C7-H7A | 119.8 |
| C8-C7-H7A | 119.8 | C7-C8-C9 | 118.32(15) |
| C7-C8-C13 | 120.36(17) | C9-C8-C13 | 121.32(17) |
| C10-C9-C8 | 122.31(15) | C10-C9-H9A | 118.8 |
| C8-C9-H9A | 118.8 | C9-C10-C11 | 118.69(15) |
| C9-C10-C14 | 121.36(16) | C11-C10-C14 | 119.94(16) |
| C10-C11-C12 | 120.37(15) | C10-C11-H11A | 119.8 |
| C12-C11-H11A | 119.8 | C7-C12-C11 | 119.89(14) |
| C7-C12-P1 | 118.94(12) | C11-C12-P1 | 121.05(12) |
| C8-C13-H13A | 109.5 | C8-C13-H13B | 109.5 |
| H13A-C13-H13B | 109.5 | C8-C13-H13C | 109.5 |
| H13A-C13-H13C | 109.5 | H13B-C13-H13C | 109.5 |
| C10-C14-H14A | 109.5 | C10-C14-H14B | 109.5 |
| H14A-C14-H14B | 109.5 | C10-C14-H14C | 109.5 |
| H14A-C14-H14C | 109.5 | H14B-C14-H14C | 109.5 |
| C20-C15-C16 | 120.91(16) | C20-C15-H15A | 119.5 |
| C16-C15-H15A | 119.5 | C17-C16-C15 | 118.22(16) |
| C17-C16-C21 | 121.28(19) | C15-C16-C21 | 120.49(18) |
| C16-C17-C18 | 121.91(17) | C16-C17-H17A | 119.0 |
| C18-C17-H17A | 119.0 | C17-C18-C19 | 119.03(17) |
| C17-C18-C22 | 120.64(19) | C19-C18-C22 | 120.32(19) |
| C20-C19-C18 | 120.50(17) | C20-C19-H19A | 119.7 |
| C18-C19-H19A | 119.8 | C19-C20-C15 | 119.40(16) |
| C19-C20-P1 | 121.93(12) | C15-C20-P1 | 118.39(12) |
| C16-C21-H21A | 109.5 | C16-C21-H21B | 109.5 |
| H21A-C21-H21B | 109.5 | C16-C21-H21C | 109.5 |
| H21A-C21-H21C | 109.5 | H21B-C21-H21C | 109.5 |
| C18-C22-H22A | 109.5 | C18-C22-H22B | 109.5 |
| H22A-C22-H22B | 109.5 | C18-C22-H22C | 109.5 |
| H22A-C22-H22C | 109.5 | H22B-C22-H22C | 109.5 |
| C24-C23-C28 | 121.15(14) | C24-C23-H23A | 119.4 |
| C28-C23-H23A | 119.4 | C23-C24-C25 | 118.50(15) |


| C23-C24-C29 | $120.72(17)$ | C25-C24-C29 | $120.77(17)$ |
| :--- | :--- | :--- | :--- |
| C26-C25-C24 | $121.96(15)$ | C26-C25-H25A | 119.0 |
| C24-C25-H25A | 119.0 | C25-C26-C27 | $118.61(15)$ |
| C25-C26-C30 | $121.43(16)$ | C27-C26-C30 | $119.96(16)$ |
| C26-C27-C28 | $120.68(14)$ | C26-C27-H27A | 119.7 |
| C28-C27-H27A | 119.7 | C23-C28-C27 | $119.09(14)$ |
| C23-C28-P2 | $118.92(11)$ | C27-C28-P2 | $121.98(11)$ |
| C24-C29-H29A | 109.5 | C24-C29-H29B | 109.5 |
| H29A-C29-H29B | 109.5 | C24-C29-H29C | 109.5 |
| H29A-C29-H29C | 109.5 | H29B-C29-H29C | 109.5 |
| C26-C30-H30A | 109.5 | C26-C30-H30B | 109.5 |
| H30A-C30-H30B | 109.5 | C26-C30-H30C | 109.5 |
| H30A-C30-H30C | 109.5 | H30B-C30-H30C | 109.5 |
| C32-C31-C36 | $121.45(16)$ | C32-C31-H31A | 119.3 |
| C36-C31-H31A | 119.3 | C33-C32-C31 | $118.37(16)$ |
| C33-C32-C37 | $121.76(18)$ | C31-C32-C37 | $119.9(2)$ |
| C32-C33-C34 | $122.11(16)$ | C32-C33-H33A | 118.9 |
| C34-C33-H33A | 118.9 | C33-C34-C35 | $118.45(17)$ |
| C33-C34-C38 | $120.15(18)$ | C35-C34-C38 | $121.39(19)$ |
| C36-C35-C34 | $120.56(16)$ | C36-C35-H35A | 119.7 |
| C34-C35-H35A | 119.7 | C35-C36-C31 | $119.04(14)$ |
| C35-C36-P2 | $124.30(12)$ | C31-C36-P2 | $116.57(12)$ |
| C32-C37-H37A | 109.5 | C32-C37-H37B | 109.5 |
| H37A-C37-H37B | 109.5 | C32-C37-H37C | 109.5 |
| H37A-C37-H37C | 109.5 | H37B-C37-H37C | 109.5 |
| C34-C38-H38A | 109.5 | C34-C38-H38B | 109.5 |
| H38A-C38-H38B | 109.5 | C34-C38-H38C | 109.5 |
| H38A-C38-H38C | 109.5 | H38B-C38-H38C | 109.5 |





Sample and crystal data for 3-98:

Identification code
Chemical formula
Formula weight
Temperature
Wavelength
Crystal size
Crystal system
Space group
Unit cell dimensions

Volume
Z
Density (calculated)
Absorption coefficien
F(000)
jam81878
$\mathrm{C}_{13} \mathrm{H}_{16} \mathrm{FNO}_{2}$
$237.27 \mathrm{~g} / \mathrm{mol}$
230(2) K
1.54178 Å
$0.008 \times 0.030 \times 0.160 \mathrm{~mm}$
orthorhombic
Pbca

$$
\begin{array}{ll}
\mathrm{a}=8.9568(11) \AA & \alpha=90^{\circ} \\
\mathrm{b}=11.5358(14) \AA & \beta=90^{\circ} \\
\mathrm{c}=24.472(3) \AA & \gamma=90^{\circ}
\end{array}
$$

8
$1.247 \mathrm{~g} / \mathrm{cm}^{3}$
$0.778 \mathrm{~mm}^{-1}$
1008

Data collection and structure refinement for 3-98:

Theta range for data collection
Index ranges
Reflections collected
Independent reflections

Max. and min. transmission
Structure soluti technique
3.61 to $68.24^{\circ}$
$-10<=\mathrm{h}<=10,-13<=\mathrm{k}<=13,-29<=1<=24$
19125
$2309[\mathrm{R}(\mathrm{int})=0.1553]$
0.7500 and 0.3400
direct methods


Atomic coordinates and equivalent isotropic atomic displacement parameters ( $\AA^{\mathbf{2}}$ ) for 3-
$U(e q)$ is defined as one third of the trace of the orthogonalized $\mathrm{U}_{\mathrm{ij}}$ tensor.

|  | $\mathbf{x} / \mathbf{a}$ | $\mathbf{y} / \mathbf{b}$ | $\mathbf{z} / \mathbf{c}$ | $\mathbf{U}(\mathbf{e q})$ |
| :--- | :--- | :--- | :--- | :--- |
| O1 | $0.1658(2)$ | $0.2099(2)$ | $0.20833(11)$ | $0.0881(9)$ |
| N1 | $0.3726(3)$ | $0.2658(3)$ | $0.25405(13)$ | $0.0743(9)$ |
| F1 | $0.4198(3)$ | $0.1106(3)$ | $0.04365(11)$ | $0.1451(12)$ |
| C1 | $0.2177(5)$ | $0.2229(6)$ | $0.0737(2)$ | $0.145(2)$ |
| C2 | $0.3527(4)$ | $0.1543(4)$ | $0.09125(18)$ | $0.0969(13)$ |
| O2 | $0.1180(3)$ | $0.5585(2)$ | $0.40871(14)$ | $0.1059(10)$ |
| C3 | $0.4619(5)$ | $0.2137(4)$ | $0.12880(19)$ | $0.1062(15)$ |
| C4 | $0.4023(4)$ | $0.1408(3)$ | $0.17685(16)$ | $0.0810(11)$ |
| C5 | $0.3238(4)$ | $0.0642(3)$ | $0.13499(19)$ | $0.0930(13)$ |
| C6 | $0.3011(3)$ | $0.2069(3)$ | $0.21450(16)$ | $0.0729(10)$ |
| C7 | $0.3060(3)$ | $0.3389(3)$ | $0.29391(16)$ | $0.0668(9)$ |
| C8 | $0.3536(4)$ | $0.3338(3)$ | $0.34713(17)$ | $0.0789(11)$ |
| C9 | $0.2972(4)$ | $0.4060(3)$ | $0.38702(17)$ | $0.0819(11)$ |
| C10 | $0.1850(4)$ | $0.4842(3)$ | $0.37262(18)$ | $0.0797(11)$ |


|  | $\mathbf{x} / \mathbf{a}$ | $\mathbf{y} / \mathbf{b}$ | $\mathbf{z} / \mathbf{c}$ | $\mathbf{U}(\mathbf{e q})$ |
| :--- | :--- | :--- | :--- | :--- |
| C11 | $0.1361(4)$ | $0.4887(3)$ | $0.31983(19)$ | $0.0824(11)$ |
| C12 | $0.1951(4)$ | $0.4179(3)$ | $0.28017(16)$ | $0.0745(10)$ |
| C13 | $0.1727(5)$ | $0.5652(5)$ | $0.4624(2)$ | $0.1249(18)$ |

Bond lengths ( $\AA$ ) for 3-98:

| O1-C6 | $1.221(3)$ | N1-C6 | $1.345(4)$ |
| :--- | :--- | :--- | :--- |
| N1-C7 | $1.421(4)$ | N1-H1N | $0.89(3)$ |
| F1-C2 | $1.405(4)$ | C1-C2 | $1.507(6)$ |
| C1-H1A | 0.97 | C1-H1B | 0.97 |
| C1-H1C | 0.97 | C2-C3 | $1.507(6)$ |
| C2-C5 | $1.515(6)$ | O2-C10 | $1.369(4)$ |
| O2-C13 | $1.404(5)$ | C3-C4 | $1.540(5)$ |
| C3-H3A | 0.98 | C3-H3B | 0.98 |
| C4-C6 | $1.501(5)$ | C4-C5 | $1.525(5)$ |
| C4-H4A | 0.99 | C5-H5A | 0.98 |
| C5-H5B | 0.98 | C7-C8 | $1.372(5)$ |
| C7-C12 | $1.390(4)$ | C8-C9 | $1.379(5)$ |
| C8-H8A | 0.94 | C9-C10 | $1.396(5)$ |
| C9-H9A | 0.94 | C10-C11 | $1.365(5)$ |
| C11-C12 | $1.373(5)$ | C11-H11A | 0.94 |
| C12-H12A | 0.94 | C13-H13A | 0.97 |
| C13-H13B | 0.97 | C13-H13C | 0.97 |

Bond angles $\left({ }^{\circ}\right)$ for 3-98:

| C6-N1-C7 | $126.4(3)$ | C6-N1-H1N | $119 .(2)$ |
| :--- | :--- | :--- | :--- |
| C7-N1-H1N | $115 .(2)$ | C2-C1-H1A | 109.5 |
| C2-C1-H1B | 109.5 | H1A-C1-H1B | 109.5 |
| C2-C1-H1C | 109.5 | H1A-C1-H1C | 109.5 |
| H1B-C1-H1C | 109.5 | F1-C2-C1 | $107.2(4)$ |
| F1-C2-C3 | $113.0(3)$ | C1-C2-C3 | $117.1(4)$ |
| F1-C2-C5 | $114.4(4)$ | C1-C2-C5 | $115.1(4)$ |
| C3-C2-C5 | $89.5(3)$ | C10-O2-C13 | $119.0(4)$ |
| C2-C3-C4 | $89.6(3)$ | C2-C3-H3A | 113.7 |
| C4-C3-H3A | 113.7 | C2-C3-H3B | 113.7 |
| C4-C3-H3B | 113.7 | H3A-C3-H3B | 111.0 |


| C6-C4-C5 | $115.3(3)$ | C6-C4-C3 | $113.6(3)$ |
| :--- | :--- | :--- | :--- |
| C5-C4-C3 | $87.9(3)$ | C6-C4-H4A | 112.6 |
| C5-C4-H4A | 112.6 | C3-C4-H4A | 112.6 |
| C2-C5-C4 | $89.9(3)$ | C2-C5-H5A | 113.7 |
| C4-C5-H5A | 113.7 | C2-C5-H5B | 113.7 |
| C4-C5-H5B | 113.7 | H5A-C5-H5B | 110.9 |
| O1-C6-N1 | $123.2(3)$ | O1-C6-C4 | $122.5(3)$ |
| N1-C6-C4 | $114.2(3)$ | C8-C7-C12 | $118.7(3)$ |
| C8-C7-N1 | $119.8(3)$ | C12-C7-N1 | $121.6(3)$ |
| C7-C8-C9 | $122.2(4)$ | C7-C8-H8A | 118.9 |
| C9-C8-H8A | 118.9 | C8-C9-C10 | $118.4(4)$ |
| C8-C9-H9A | 120.8 | C10-C9-H9A | 120.8 |
| C11-C10-O2 | $116.5(4)$ | C11-C10-C9 | $119.6(4)$ |
| O2-C10-C9 | $123.9(4)$ | C10-C11-C12 | $121.5(4)$ |
| C10-C11-H11A | 119.2 | C12-C11-H11A | 119.2 |
| C11-C12-C7 | $119.6(4)$ | C11-C12-H12A | 120.2 |
| C7-C12-H12A | 120.2 | O2-C13-H13A | 109.5 |
| O2-C13-H13B | 109.5 | H13A-C13-H13B | 109.5 |
| O2-C13-H13C | 109.5 | H13A-C13-H13C | 109.5 |
| H13B-C13-H13C | 109.5 |  |  |

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